

**STRUCTURAL, LUMINESCENCE AND ELECTRICAL CONDUCTIVITY OF
STRONTIUM TITANATE THIN FILMS PREPARED BY DIP COATING
METHOD**

SALEHA BINTI MAAROF

A thesis submitted in fulfillment of the
requirements for the award of the degree of
Master of Science (Physics)

**Faculty of Science
University Teknologi Malaysia**

JUNE 2014

Dedicated to

my beloved husband, parents, family, and friends.

ACKNOWLEDGEMENT

First and foremost, I would like to acknowledge my supervisor, Associate Professor Dr. Karim Deraman for his guidances during this study. I would like to thank our research group colleagues, Associate Professor Dr. Rosli Husin, Associate Dr. Mohd Nor Yusof, Dr. Wan Nurulhuda and others research group student for their assistance and relationship.

I am also thankful to the Department of Physics, Faculty of Science, Universiti Teknologi Malaysia, Johor, the financial support from Ministry of Higher Education (MOHE) under the vote 4F005 for funding the project.

ABSTRACT

Strontium titanate (SrTiO_3) sol-gel was deposited onto glass substrates by dip coating method. The effects of deposition parameters such as different annealing temperatures and number of coating layers were studied. Strontium nitrate anhydrous ($\text{Sr}(\text{NO}_3)_2$), titanium isopropoxide ($\text{C}_{12}\text{H}_{28}\text{O}_4\text{Ti}$), ethylene glycol ($\text{HOCH}_2\text{CH}_2\text{OH}$) and nitric acid (HNO_3) were used to prepare sol-gel of strontium titanate. Ethylene glycol was added to promote polymerization between $\text{C}_{12}\text{H}_{28}\text{O}_4\text{Ti}$ and $\text{Sr}(\text{NO}_3)_2$. The samples structure were characterized by X-ray diffraction (XRD), atomic force microscopy (AFM), Fourier transform infra red (FT-IR) and photoluminescence (PL) spectroscopy. XRD analysis showed the process of the intermediate crystalline phase of $\text{Sr}(\text{NO}_3)_2$ with lattice parameter $a = b = c = 0.78$ nm for as-prepared sample. An amorphous inorganic phase was found for samples annealed at 300 and 400 °C. The samples began to change to SrTiO_3 phase after being annealed at 500 °C with lattice parameter $a = b = c = 0.39$ nm and peak orientation at planes (110), (111), (200) and (211). Thin films with double coating layers showed clearer peaks orientation. The microstructure study using AFM revealed the grain size in the range 220 – 460 nm and surface roughness in the range 33 – 69 nm at 500 °C for different number of coating layers. FT-IR analysis shows that SrTiO_3 crystallization phase was completely formed at higher temperatures of 500, 600 and 700 °C. PL spectroscopy of SrTiO_3 thin films appeared to have spectra emission in the region of 310 - 510 nm wavelength. The spectra also showed the different peaks appearing at different annealed temperatures. Van der Pauw technique was used to measure the electrical conductivity and activation energy for single and four coating layers samples at 500 °C annealing temperature. The results showed that the electrical conductivity for single layer was higher than the four coating layers. Activation energy in high temperature region were in the range of 0.656 meV and 0.446 meV whereas activation energy in low temperature region were in the range of 0.162 meV and 0.105 meV for single and four coating layers samples, respectively.

ABSTRAK

Sol-gel strontium titanat (SrTiO_3) telah dimendapkan ke atas substrat kaca dengan menggunakan kaedah salutan celupan. Kesan parameter pemendapan seperti perbezaan suhu penyepuhlindapan dan bilangan salutan lapisan telah dikaji. Anhitrat strontium nitrat ($\text{Sr}(\text{NO}_3)_2$), titanium isoproposida ($\text{C}_{12}\text{H}_{28}\text{O}_4\text{Ti}$), etilena glikol ($\text{HOCH}_2\text{CH}_2\text{OH}$) dan asid nitrik (HNO_3) telah digunakan untuk menghasilkan sol-gel strontium titanat. Etilena glikol telah ditambah untuk menggalakkan pempolimeran diantara $\text{C}_{12}\text{H}_{28}\text{O}_4\text{Ti}$ dan $\text{Sr}(\text{NO}_3)_2$. Struktur sampel telah dicirikan dengan menggunakan pembelauan X-ray (XRD), mikroskopi daya atom (AFM), transformasi Fourier infra merah (FT-IR) dan spektroskopi fotopendarcahaya (PL). Analisis XRD telah menunjukkan proses fasa pertengahan kristal bagi $\text{Sr}(\text{NO}_3)_2$ dengan parameter kekisi $a = b = c = 0.78 \text{ nm}$ untuk sampel pra-persediaan. Fasa bukan organik amorfus telah diperolehi bagi sampel sepuhlindapan pada 300 and 400 °C. Sampel mula berubah kepada fasa SrTiO_3 selepas sepuhlindapan pada 500 °C dengan parameter kekisi $a = b = c = 0.39 \text{ nm}$ dan orientasi puncak pada satah (110), (111), (200) dan (211). Saput tipis dengan dwi lapisan salutan menunjukkan puncak-puncak orientasi yang paling jelas. Kajian mikrostruktur melalui AFM menunjukkan saiz butiran pada julat 220 – 460 nm dan permukaan kasar pada julat 33 – 69 nm pada suhu 500 °C untuk bilangan lapisan salutan yang berbeza-beza. Analisis FT-IR menunjukkan fasa kristal SrTiO_3 telah terhasil secara menyeluruh pada suhu lebih tinggi 500, 600 and 700 °C. Spektroskopi PL untuk saput tipis SrTiO_3 menunjukkan pancaran spektra pada panjang gelombang dalam lingkungan 310 – 510 nm. Spektra yang dihasilkan juga menunjukkan kemunculan puncak yang berbeza pada suhu sepuhlindapan yang berbeza. Teknik Van der Pauw telah digunakan untuk mengukur konduktiviti elektrik dan tenaga aktif untuk sampel satu dan empat lapisan salutan pada suhu penyepuhlindapan 500 °C. Keputusan menunjukkan konduktiviti pada sampel satu lapisan adalah lebih tinggi berbanding sampel empat lapisan. Tenaga aktif pada julat suhu tinggi berada dalam lingkungan 0.656 meV dan 0.446 meV manakala tenaga aktif pada julat suhu rendah berada dalam lingkungan 0.162 meV dan 0.105 meV masing-masing untuk sampel satu dan empat lapisan salutan.

TABLE OF CONTENTS

CHAPTER	TITLE	PAGE
	DECLARATION	ii
	DEDICATION	iii
	ACKNOWLEDGEMENTS	iv
	ABSTRACT	v
	ABSTRAK	vi
	TABLE OF CONTENTS	vii
	LIST OF TABLES	ix
	LIST OF FIGURES	x
	LIST OF SYMBOLS	xii
	LIST OF APPENDICES	xiv
1	INTRODUCTION	
1.1	Introduction	1
1.2	Problem Statement	3
1.3	Research Objectives	3
1.4	Scope of Studies	4
2	LITERATURE REVIEW	
2.1	Introduction	5
2.2	Strontium Titanate (SrTiO_3)	6
2.3	Strontium Titanate Thin Films	8
2.4	Behavior of Strontium Titanate Thin Films	11
2.4.1	X-ray Identification	12

2.4.2	Microstructure	15
2.4.3	Infrared (FT-IR) Spectroscopy	18
2.4.4	Photoluminescence (PL) Spectroscopy	23
2.4.5	Electrical Conductivity	26
3	INSTRUMENTATION AND METHODOLOGY	
3.1	Introduction	29
3.2	Dip Coater Machine	30
3.3	Synthesis of Strontium Titanate Sol-gel Solution	32
3.4	Substrate Preparation	34
3.5	Sample Preparation	35
3.6	Sample Characterizations	36
3.6.1	X-ray Diffraction (XRD)	36
3.6.2	Atomic Force Microscopy (AFM)	36
3.6.3	Infrared Spectroscopy (FT-IR)	37
3.6.4	Photoluminescence Spectroscopy (PL)	38
3.6.5	Electrical conductivity measurement	38
4	RESULT AND DISCUSSTION	
4.1	Introduction	40
4.2	X-ray Phase Identification	41
4.3	Microstructure Evaluation	45
4.4	Infrared Spectra	55
4.5	Photoluminescence Spectra	60
4.6	Electrical Conductivity	66
5	CONCLUSIONS AND FUTURE WORK	
5.1	Conclusions	70
5.2	Future Works	
REFERENCES		72
APPENDICES A-H		80-87

LIST OF TABLES

TABLE NO.	TITLE	PAGE
2.1	Author, year and methodology from previous research for strontium titanate material.	8
2.2	The raw material used and objective of sol-gel thin films	10
2.3	Element stretch absorption bands from previous research.	22
3.1	Raw materials used in the preparation strontium titanate precursor solution.	34
3.2	Thin films parameters for dip coating process.	35
4.1	Average grain size and surface roughness.	44
4.2	Experimental and theoretical wavelength value of Sr^{2+} ions for two layers coated thin films.	61
4.3	Experimental and theoretical wavelength value of Ti^{4+} ions for two layers coated thin films.	62
4.4	Experimental and theoretical wavelength value of O^{2+} ions for two layers coated thin films.	63
4.5	Activation Energy at high and lower temperature regions.	66

LIST OF FIGURES

FIGURE NO.	TITLE	PAGE
2.1	Strontium titanate perovskite cubic structure.	7
2.2	Bragg diffraction.	13
2.3	XRD profile of strontium titanate thin films for: (a) as-deposited, (b) 100, (c) 200, (d) 300, (e) 400, (f) 500 and (g) 600 °C annealed temperatures (Gao <i>et al.</i> , 2002).	14
2.4	XRD profile of strontium titanate thin films for: (a) as-deposited, (b) 400 and 500 °C annealed temperatures (Gao <i>et al.</i> , 2003).	15
2.5	3D AFM image of strontium titanate thin films (Pontes <i>et al.</i> , 2001).	16
2.6	2D AFM image of strontium titanate thin films (Pontes <i>et al.</i> , 2001).	17
2.7	Infrared spectroscopy diagram.	18
2.8	FT-IR spectra of a thin film annealed at different temperatures (Leite <i>et al.</i> , 1999).	20
2.9	FT-IR spectra of: (a) amorphous precipitate collected after depositing thin films and (b) after annealed at 600 °C for 2 hours in air (Goa <i>et al.</i> , 2002).	21
2.10	Illustration of different recombination.	24
3.1	Dip coating chamber.	30
3.2	Schematic diagram of dip coating chamber.	31
3.3	Flow chart for preparation strontium titanate polymeric precursor method.	33
3.4	Experiment setup for Van der Pauw technique.	39
4.1	XRD pattern films for: (a) as-prepared (b) 300 (c) 400 (d) 500 °C.	40
4.2	XRD pattern of films synthesized at 500 °C annealed temperature for: (a) single layer, (b) two layers, (c) three layers, (d) four layers and (e) five layers.	41

4.3	Intensity of dominant peak (32.4°) with number of thin film coated layer.	42
4.4	Polymeric precursor solution without precipitate.	44
4.5	3D sample surfaces obtain by AFM for single layer coated film.	46
4.6	3D sample surfaces obtain by AFM for two layers coated film.	47
4.7	3D sample surfaces obtain by AFM for three layers coated film.	48
4.8	3D sample surfaces obtain by AFM for four layers coated film.	49
4.9	3D sample surfaces obtain by AFM for five layers coated film.	50
4.10	Surface roughness with number of thin film coating layer.	51
4.11	Average grain size with number of thin film coating layer.	52
4.12	FT-IR spectra of: (a) strontium titanate polymeric precursor solution and (b) thin film annealed at $500\text{ }^\circ\text{C}$.	55
4.13	FT-IR spectra of strontium titanate thin films annealed at different temperatures for double layers thin films. Films annealed at: (a) 300, (b) 400, (c) 500, (d) 600 and (e) 700 $^\circ\text{C}$.	57
4.14	Photoluminescence spectra for double layers thin films; as-prepared, (b) 300, (c) 400, (d), 500 and 600 $^\circ\text{C}$.	59
4.15	Peak intensity for different annealing temperature.	60
4.16	Dependence of film conductivity on inverse temperature for $500\text{ }^\circ\text{C}$ annealed temperature at different coated layer; single layer ■ and four layers ♦.	65

LIST OF SYMBOLS

FT-IR	- Fourier Transform Infrared Spectroscopy
XRD	- X-ray Diffraction
AFM	- Atomic Force Microscopy
PL	- Photoluminescence
VDP	- Van der Pauw
SrTiO_3	- Strontium Titanate
$\text{Sr}(\text{NO}_3)_2$	- Strontium Nitrate
$\text{OHCH}_2\text{CH}_2\text{OH}$	- Ethylene Glycol
$\text{C}_{12}\text{H}_{28}\text{O}_4\text{Ti}$	- Titanium Isopropoxide
JCPDS	- Joint Center Power Diffraction Standard card
σ	- Electrical Conductivity
σ_0	- Pre-exponential of Electrical Conductivity
E_a	- Activation Energy
k	- Boltzmann constant
d	- Spacing between the planes in the atomic lattice
d_{hkl}	- Interplanar spacing for planes with Miller indicates (hkl)
λ	- Wavelength
θ	- Angle between the incident ray and the scattering planes
v	- Frequency (Hz)
k	- Force constant of the bond (Nm^{-1})
μ	- Reduce mass (kg)
m_1	- Relative atomic mass of $M_1 \times u$
m_2	- Relative atomic mass of $M_2 \times u$
u	- atomic mass unit = 1.66×10^{-27} kg
h	- Plank's constant
c	- Speed of light

LIST OF APPENDICES

APPENDIX	TITLE	PAGE
A	Bruker D8 diffractometer with CuK _α X-ray radiation at Universiti Kebangsaan Malaysia.	80
B	Scanning Probe Microscopy (SPM) at Universiti Teknologi Malaysia.	81
C	Image Analysis P9 Program Interface, AFM software. (Mironov., 2004).	82
D	Perkin Elmer Spectrum GX FT-IR Microscope at Universiti Teknologi Malaysia.	83
E	Luminescence LS 55 model (Perkin Elmer) at Universiti Teknologi Malaysia.	84
F	List of Infrared Absorption Frequency.	85
G	JCPDS Card for Strontium Titanate.	86
H	Energy Levels for Sr ²⁺ , Ti ⁴⁺ and O ²⁺	87

CHAPTER 1

INTRODUCTION

1.1 Introduction

Phenomenon of thin film can be describe when a layer of oxide appeared on a material. Thus, scientist used thin as a term to explain about a very thin layer appeared on a material that can be characterize and applied in technology development. This phenomenon has been discovered by Faraday last 1838 using an electrolysis process, followed by Bunsen & Grove on 1852 using chemical reaction, in 1857 Faraday have discovered thermal evaporation process and further studied have been carried out intensively after world war two (Samsudi Sakrani, 2002).

Thin film is a layer of solid material such as metal, semiconductor or insulator deposited on a substrate with appropriate thickness in range from 10-1000 nm (Suriani, 2005). For layer greater than 1000 nm, it will be grouped as a thick film or film. It will be known as ultra film for layer thinner than 100 nm. In thin films study, variable of thickness shown different characterization result. A substrate is use as a medium to deposit thin film. To produce a good quality and

interact with film in hygiene surface. The main applications in thin film development are in electronic devices and optical coatings.

Generally thin film can be prepared either in chemical (Desai *et al.*, 2006) or physical method (Dantus *et al.*, 2011). Examples of chemical method are dip coating and spray pyrolysis techniques. Other examples of physical method are sputtering and vacuum evaporation techniques. There are various methods of preparing nanostructure thin films such as sol-gel process (known as dip coating method and spin coating method), metal organic decomposition (MOD), liquid phase deposition (LPD), spray pyrolysis and thermal evaporator under vacuum. A lot of researchers are competing in producing thin films with a simple technique, large surface area, inexpensive and non-pollution for industrial user. Dip coating with sol-gel process have been widely used to prepared thin films and ceramic powder. This type of method has an capability to control the stoichiometric literally at the molecular level.

ABO_3 family is known as cations elements for A and B and anion element for O or oxygen. Recently a lot of ABO_3 family for examples barium titanate, barium strontium titanate and strontium titanate have been attracted due to their ferroelectric and electro-optic properties (Soledade *et al.*, 2002 and Goa *et al.*, 2002). Strontium titanate is a paraelectric material with cubic perovskite structure (Zanetti *et al.*, 1997 and Benthem *et al.*, 2001). Theoretically, strontium titanate material is known to has properties such as high dielectric constant, good thermal stability, second order phase transformation and stress induce phase transition photo-activity (Kim *et al.*, 2007). These advantages makes strontium titanate as the best material and widely used for application in memory cell capacitors for non-volatile memories or dynamic random access application (DRAM), microelectronic, solar and sensor technology in recent years usually in bulk material or thin films technology development (Brankovic *et al.*, 2004).

1.2 Problem Statement

Nowadays, strontium titanate has attracted many researchers because of its wide application and favorable performance in industry. With this purpose, numerous of work have been done in different environment and condition to develop a good quality of strontium titanate thin films with a simple step, low cost and environmental friendly.

In recent years, works on sol-gel preparation with dip coating technique to deposit strontium titanate thin films have been actively considered. Even though there are several report on strontium titanate thin films, there is no clear-cut understanding on the relation between numbers of coating layer thin films and annealing temperatures with several characterization. Thus, this experiment will be done by various deposition parameters such as number of coating layer on corning glass substrate and at different annealed temperatures. By changing the parameters in strontium titanate thin films, one can expect to obtain new characterization result and may give better understanding on structural and electrical conductivity. Thus, the change in behavior of these films can be used in specific applications and development.

1.3 Research Objectives

Research objective of this study are:

- i. To prepare strontium titanate thin films by using sol-gel dip coating method at different numbers of coating layer and annealed temperatures.
- ii. To characterize the structural behavior of strontium titanate thin films in coating layer and at annealing temperatures by using X-ray diffraction

- (XRD) analysis, atomic force microscopy (AFM), Fourier Transformation Infrared Spectroscopy (FT-IR) and photoluminescence spectrometer (PL).
- iii. To investigate the electrical conductivity of coating layer thin films by using van der Pauw (VDP) technique.

1.4 Scope of Studies

The structural and chemical bonding of strontium titanate was characterized by using x-ray diffraction (XRD) analysis while infrared spectroscopy (FT-IR) was used to confirm the existence of strontium titanate crystallization. Atomic force microscopy (AFM) was used to characterize the morphology of the films. Meanwhile, emission and electrical conductivity behavior of the films were investigated by using photoluminescence spectroscopy (PL) and Van der Pauw (VDP), respectively.

This thesis has covered five chapter. Chapter 1 includes background, problem statement, objective and scope of studies. Chapter 2 discussed about literature review regarding the physical properties of strontium titanate, thin films methodology in previous work and properties of strontium titanate thin films. This covers theory of sample characterization of strontium titanate thin films, on the structural and electrical conductivity properties. The methodology and instrumentation in chapter 3 covered all the experimental and characterizations process. Next, chapter 4 discussed the results of experimental conducted in the research. Lastly, chapter 5 summarizes the studies and some suggestion are presented for future works.

REFERENCES

- Baba, S., Numata, K., and Miyake, S. (2000). Characteristics of Post-annealed SrTiO₃ Thin Films Prepared by Mirror-confinement-type ECR Plasma Sputtering. *Science and Technology of Advanced Materials*. 1, 211-217.
- Bae, D. S., Kim, E. J., Park, S. W. and Han, K. S. (2006). Fabrication and Microstructure of the Composite Membranes by Sol-Gel Process. *Journal Metal*. 41, 6162-6164.
- Benthem, K., Elsasser, C., and French, R. H. (2001). Bulk Electronic Structure of SrTiO₃: Experiment and Theory. *Journal of Applied Physics*. 90(12), 6156-6164.
- Brankovic, G., Brankovic, Z., Varela, J. A., and Longo, E. (2004). Strontium Titanate Films Prepared by Spray Pyrolysis. *Journal of the European Ceramic Society*. 24, 989-991.
- Brankovic, G., Brankovic, Z., Goes, M. S., Santos, C. O. P., Cilense, M., Varela, J. A., and Longo, E. (2005). Barium Strontium Titanate Powders Prepared by Spray Pyrolysis. *Materials Science and Engineering*. B122, 140-144.
- Cai, Q. J., Gan, Y., Park, M. B. C., Yang, H. B., Lu, Z. S., Li, C. M., Guo, J., and Dong, Z. L. (2009). Solution- Processable Barium Titanate and Strontium Titanate Nanoparticles Dielectric for Low-Voltage Organic Thin-Film

- transistors. *Chemistry Materials Article.* 21, 3135-3161.
- Cheng, Z. Y., Wang, H. F., Quan, Z. W., Lin, C. K., Lin, J., and Han, Y. C. (2005). Layered Organic-Inorganic Perovskite-Type Hybrid Materials Fabrication by Spray Pyrolysis Route. *Journal of Crystal Growth.* 285, 352-357.
- Dantus, C., Rusu, R. S., and Rusu, G. I. (2011). On the Mechanism of Electronic transport in Polycrystalline CdO Thin Films. *Superlattices and Microstructures.* 50, 303-310.
- Deraman, K., Sakrani, S., Ismail, B.B., Wahab, Y., and Gould, R. D. (1994). Electrical Conductivity Measurements in Evaporated Tin Sulphide Thin Films. *Journal Electronics.* 76(5), 917-922.
- Desai, J. D., Min, S. K., Jung, K. D., and Joo, O. S. (2006). Spray Pyrolysis Synthesis of Large Area NiO_x Thin Films from Aqueous Nickel Acetate Solution. *Applied Surface Science.* 253, 1781-1786.
- Djermouni, M., Zaoui, A., Kacimi, S., and Bouhafs, B. (2010). Vacancy defects in Strontium Titanate: Ab Initio Calculation. *Computational Materials Science.* 49, 904-909.
- Gao, Y., Masuda, Y., Yonezawa, T., and Koumoto, K. (2002). Site-Selective Deposition and Micropatterning of SrTiO_3 Thin Film on Self-Assembled Monolayers by the Liquid Phase Deposition Method. *Chemistry Materials.* 14, (12), 5006-5014.
- Gao, Y., Masuda, Y., Yonezawa, T., and Koumoto, K. (2003). Preparation of SrTiO_3 Thin Films by the Liquid Phase Deposition Method. *Materials Science & Engineering.* B99, 290-293.
- Han, J., Wan, F., Zhu, Z., and Zhang, W. (2007). Dielectric Response of Soft

- Mode in Ferroelectric SrTiO₃. *Applide Physics Latters.* 90: 1-3.
- Hod, I., Shalom, M., Tachan, Z., Ruhle, S., and Zaban, A. (2010). SrTiO₃ Recombination-Inhibiting Barrier Layer for Type II Dye-Sensitized Solar Cells. *Journal Physics Chemistry,* 114, 10015-10018.
- Hyun, S., and Char, K. (2001). Effect of Strain on the Dielectric Properties of Tunable Dielectric SrTiO₃ Thin Films. *Applied Physics Letters.* 79(2), 254 – 256.
- Iwatsuki, S., Kera, M., Nakashima, K., Fujii, I., Takei, T., Kumada, N., and Wada, S. (2011). Preparation of Strontium Titanate Nanocubes Using Titanium Alkoxide and Their Accumulations by Capillary Force. *Materials Sciences and Engineering.* 18, 1-4.
- Josef, G., and Hans, Meixner. (1990). Electrical Conductivity of Sputtered Films of Strontium Titanate. *Journal Applied Physics.* 67(12) 7453 – 7459.
- Kamalasan, M. N., Kumar, N. D., and Chandra, S. (1996). Fourier Transform-Infrared and Optical Studies on Sol-Gel Synthesized SrTiO₃ Precursor Films. *Journal of materials Science.* 31, 2741-2745.
- Kang, Y. C., Seo, D. J., Park, S. B., and Park., H. D. (2002). Direct Synthesis of Strontium Titanate Phosphor Particles with High Luminescence by Flame Spray Pyrolysis. *Materials Research Bulletin.* 37, 263-269.
- Kao, C. F., and Yang, W. D. (1999). Preparation of Barium Strontium Titanate Powder from Citrate Precursors. *Applied Organometallic Chemistry.* 13, (383-397)
- Kim, K. H., Park, J. K., Kim, C. H., Park, H. D., Chang, H., and Choi, S. Y. (2002). Synthesis of SrTiO₃:Pr, Al by Ultrasonic Spray Pyrolysis. *Ceramic*

International. 28, 29-36.

- Kim, K. J., Kim, J. H., Park, M. S., Kwon, H. K., Kim, H., and Kim, Y. J. (2012). Enhancement of electrochemical and Thermal Properties of Polycrystalline Separators Coated with Polyvinylidene Fluoride-Hexafluoropropylene Co-Polymer for Li-Ion Batteries. *Journal of Power Sources.* 198, 298-302.
- Kim, K. J., Suzuki, M., Yoshikawa, A., Fukuda, T., Auh, K. H., and Shin, K. C. (2007). Films Synthesis of SrTiO₃ by the Spray-ICP Technique. *Journal of Alloys and Compounds.* 440, 216-219.
- Kirk, K. J., Elgoyhen, J., Hood, J. P., and Hutson, D. (2011). Use of Mass Loading to Operate a Thin Film Ultrasonic Transducer in the 300 kHz-10 MHz Frequency Range. *Journal Electroceram.* 27, 29-32.
- Krzysztof, S., Jan, L., Cieslinski., and Kamil, L. (2012). Van der Pauw Method on a Sample with an Isolated Hole. *Physics Letters A.*
- Leite, E. R., Mastlaro, V. R., Zanetti, S. M., and Longo, E. (1999) Crystallization Study of SrTiO₃ Thin Films Prepared by Dip Coating. *Materials Research.* 2(2), 93-97.
- Longo, V. M., Figueiredo. A. T., Lazaro. S., Gulgel. M. F., Costa. M. G. S., Paiva-Santos. C. O., and Franco. R. W. A. (2008) Structural Condition That Leads to Photoluminescence Emission in : An Experimental and Theoretical Approach. *Journal of Applied Physics.* 104, 023515.
- Mar, M. N., Ratner, B. D., and Yee, S. S. (1999). An Intrinsically Protein-Resistant Surface Plasmon Resonance Biosensor Based Upon a RF-Plasma-Deposited Thin Film. *Sensor and Actuators B Chemical.* 54, 125-131.
- Marques, M., Teles, L. K., Anjos, V., Scolfarro, L. M. R., Leite, J. R., Freire, V. N.,

- Farias, G. A., and Silva, E. F. (2003). Full-Relativistic Calculation of the SrTiO₃ Carrier Effective Masses and Complex Dielectric Function. *Applied Physics Letters*. 82(18), 3074-3076.
- Mironov, V. L., (2004). *Fundamentals of Scanning Probe Mictoscopy*. The Russian Academy Sciences Institute of Physics of Microstructures.
- Moos, R., and Hardtl, K. H. (1997). Defect Chemistry of Donor-Doped and Undoped Strontium Titanate Ceramics Between 1000° and 1400 °C . *Journal of the American Ceramic Society*. 80(10), 2549-2562.
- Ohly, C., Eifert, S. H., Guo, X., Schubert, J., and Waser, R. (2006). Electrical Conductivity of Epitaxial SrTiO₃ Thin Films as a Function of Oxygen Partial Pressure and Temperature. *Journal of the American Ceramic Society*. 89(9), 2845-2852.
- Paik, K. W., Hyun, J. G., Lee, S., and Jang, K. W. (2006). Epoxy/BaTiO₃(SrTiO₃) Composite Films and Pastes for High Dielectric Constant and Low Tolerance Embedded Capacitors in Organic Substrates. *Electronic System integration Technology Conference*. Dresden, Germany: IEEE, 794 - 801.
- Patidar, D., Rathore, K. S., Saxena, N. S., and Sharma, T. P. (2008). Energy Band Gap and Conductivity Measurement of CdSe.Thin Films. *Chalcogenide Letters*. 5(2), 21-25.
- Pontes, F. M., Lee, E. J. H., Leite, E. R., and Longo E. (2000) High Dielectric Constant of SrTiO₃ Thin Films Prepared by Chemical Process. *Journal of Materials Science*. 38, 4783-4787.
- Pontes, F. M., Leite, E. R., Lee, E. J. H., Longo, E., and Varela, J. A. (2001). Preparation Microstructural and Electrical Characterization of SrTiO₃ thin

Films Prepared by Chemical Route. *Journal of the European Ceramic Society*. 21, 419-426.

Pookmanee, P., and Phanicphant, S. (2009). Titanium Dioxide Powder Prepared by a Sol-Gel Method. *Journal of Ceramic Processing Research*. 10(2), 167-170.

Raut, N. C., Mathews, T., Chandramohan, P., Srinivasan, M. P., Dash, S., and Tyagi, A. K. (2011). Effect of Temperature on the Growth of TiO₂ Thin Films Synthesized by Spray Pyrolysis: Structural Compositional and Optical Properties. *Materials Research Bulletin*. 46, 2057-2063.

Ren, Y., Zhao, G., and Chen, Y. (2011). Fabrication of Textured SnO₂ : F Thin Films by Spray Pyrolysis. *Applied Surface Science*. 258, 914-918.

Rho, J., Jang, S., Ko, Y. D., Kang, S., Kim, D., Chung, J. S., Kim, M., Han, M., and Choi, E. (2009). Photoluminescence Induced by Thermal Annealing in SrTiO₃ Thin Film. *Applied Physics Letters*. 95, 241906.

Romero, R., Martin, F., Ramos-Barrado, J. R., and Leinen, D. (2010). Synthesis and Characterization of Nanostructured Nickel Oxide Thin Films Prepared with Chemical Spray Pyrolysis. *Thin Solid Films*. 518, 4499-4502.

Samsudi Sakrani. (2002). *Kuliah: Teknologi Saput Tipis*. Universiti Teknologi Malaysia, Skudai, Johor.

Sansonetti, J. E., and Martin W. C., (2011). *Gaithersburg Patent No. MD 20899*. Retrieved on 16 August, 2013, from <http://www.nist.gov>.

Schmidt, S., Lu, J., Keane, S. P., Bregante, L. D., Klenov, D. O., and Stemmer, S. (2005). Microstructure and Dielectric Properties of Texture SrTiO₃ Thin Films. *Journal of the American Ceramic Society*. 88(4), 789-801.

- Soledade, L. E. B., Lomgo, E., Leite, E. R., Pontes, F. M., Lanciotti, Jr., Campos, C. E. M., Pizani, P. S., and Varela, J. A. (2002). Room-temperature photoluminescence in amorphous SrTiO₃ – the influence of acceptor-type dopants. *Applied Physics Materials Science & Processing*. 75, 629 – 632.
- Sur, S., Ozturk, Z., Oztas, M., Bedir, M., and Ozdemir, Y. (2011). Studies on Structural and Electrical Properties of MnO Films Prepared by the Spray Pyrolysis Method. *Physica Scripta*. 84, 1-4.
- Suriani, A.B. (2005). *The Structural And Optical Properties of Hydrogenerated Amorphous Carbon (a-C:H), Thin Films Deposited Using A DC-PECVD Technique*. Degree of Master of Science (Physics). Universiti Teknologi Malaysia.
- Sravani Gullapalli and Andrew R. Barron (2010). Physical Methods in Chemistry and Nano Science. *Creative Commons Attribution License (cc-By3.0)*. 0500.
- Swanson, H., and Fuyat. Nalt. Bur. (1954). Stand. (U.S), Circ. 539 (3), 44.
- Tumarkin, A. V., Gaidukov, M. M., Gagarin, A. G., Samoilova, T. B. and Kozyrev, A. B. (2012). Ferroelectric Strontium Titanate Thin Films for Microwave Applications. *Taylor and Francis*. 439, 49-55.
- Wang, J., Yao, X., and Zhang, L. (2004). Preparation and Dielectric Properties of Barium Strontium Titanate Glass-Ceramics Sintered from Sol-Gel Derived Powders. *Ceramic International*. 30, 1749-1752.
- Xu, C. Huang, J., Tan, X., Yu, T., Cui, Z., and Zhao, Lin. (2010). Preparation, Characteristics, and Photocatalytic Tests of Fe-Doped TiO₂ Films Prepared by a Sol-Gel Drain Coating via Homemade Devices. *Journal of*

Dispersion Science and Technology. 31, 1732-1739.

Xie, J., Ji, T., Xihui, O. Y., Xiao, Z., and Shi, H. (2008). Preparation of SrTiO₃ nanomaterial from Layered Titanate Nanotubes or Nanowires. *Solid State Communication.* 147, 226-229.

Yan, H., and Okuzaki, H. (2011). SrTiO₃ Insulator for Low-voltage Organic Field-effect Transistors. *Synthetic Metals.* 161, 1781-1786.

Zanetti, S. M., Longo, E., Varela, J. A., and Leite, E. R. (1997). Microstructure and Phase Evolution of SrTiO₃ Thin Films on Si Prepared by the use of Polymeric Precursor. *Materials Letters.* 31, 173-178.

Zanetti, S. M., Leite, E. R., Longo, E., and Varela, J. A. (1999). Crack Developed During SrTiO₃ Thin-Film Preparation from Polymeric Precursors. *Applied Organometallic Chemistry.* 13, 373-382.

Zanetti, S. M., Leite, E. R., Longo, E., and Varela, J. A. (1999). SrBi₂Nb₂O₉ Thin Films Deposited by Dip Coating Using Aqueous Solution. *Journal of the European Ceramic Society.* 19, 1409-1412.

Zhou, Y. X., Bhuiyan, S., Scruggs, S., Fang, H., Mironova, M., and Salama, K. (2003). Strontium titanate Buffer Layers Deposited on Rolled Ni Substrates with metal organic Deposited. *Superconductor Science and Technology.* 16, 901-906.

Zhou, X. Y., Miao, J. Y., Dai, Chan, H. L. W., Choy, C. L., Wang, Y., and Li, Q. (2007). Epitaxial Growth of SrTiO₃ Thin Film on Si by Laser Molecular Beam Epitaxy. *Applied Physics Letters.* 90, 012902.