

# Flame-retardant thermoplastics derived from plant cell wall polymers by single ionic liquid substitution

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# Electronic Supplementary Information

Fire-retardant thermoplastics derived from plant cell wall

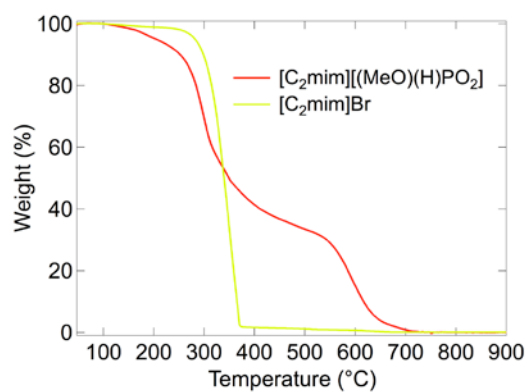
by single ionic liquid substitution

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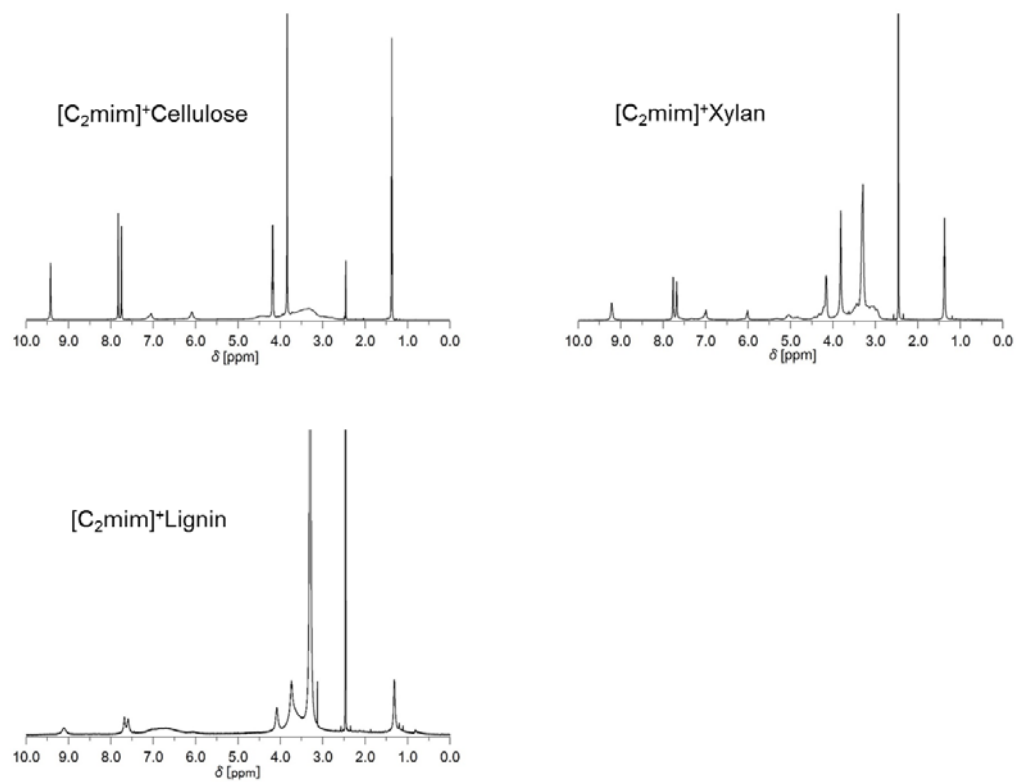
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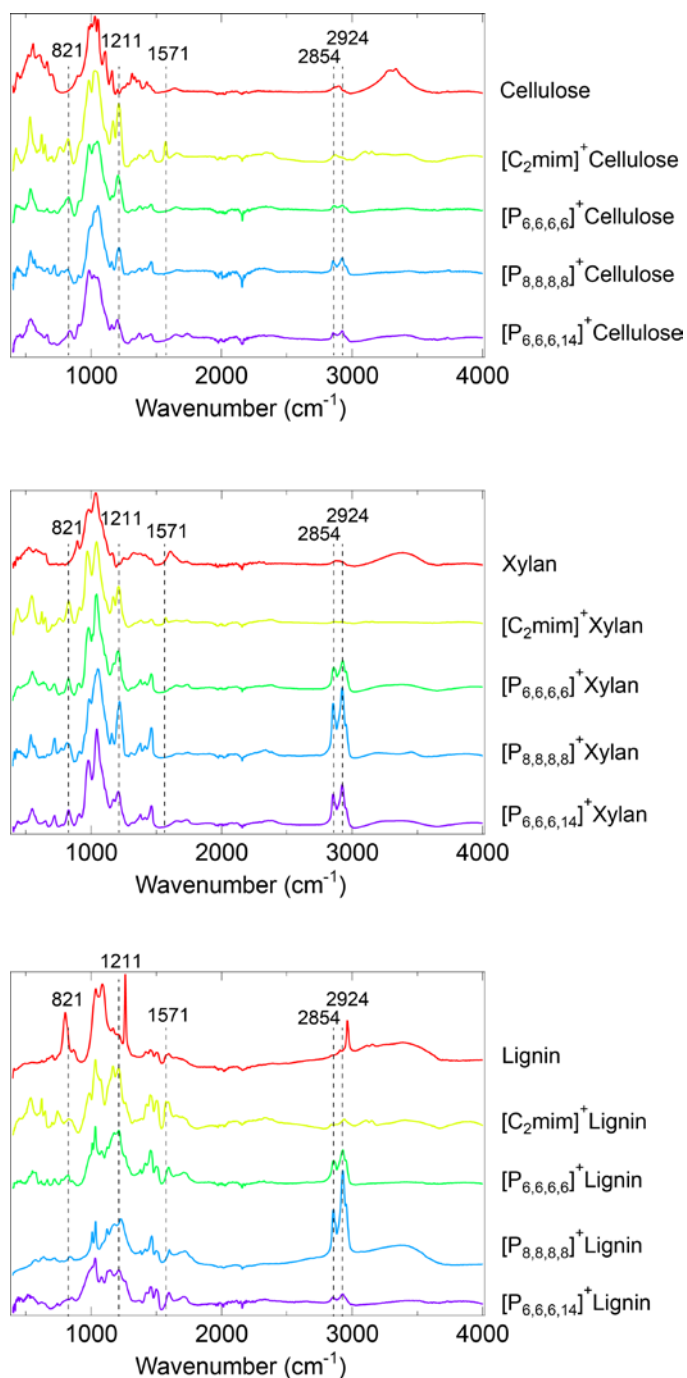
## Supporting Figures



**Fig. S1** TGA curves of [C<sub>2</sub>mim][(MeO)(H)PO<sub>2</sub>] which were used in this study under air condition.

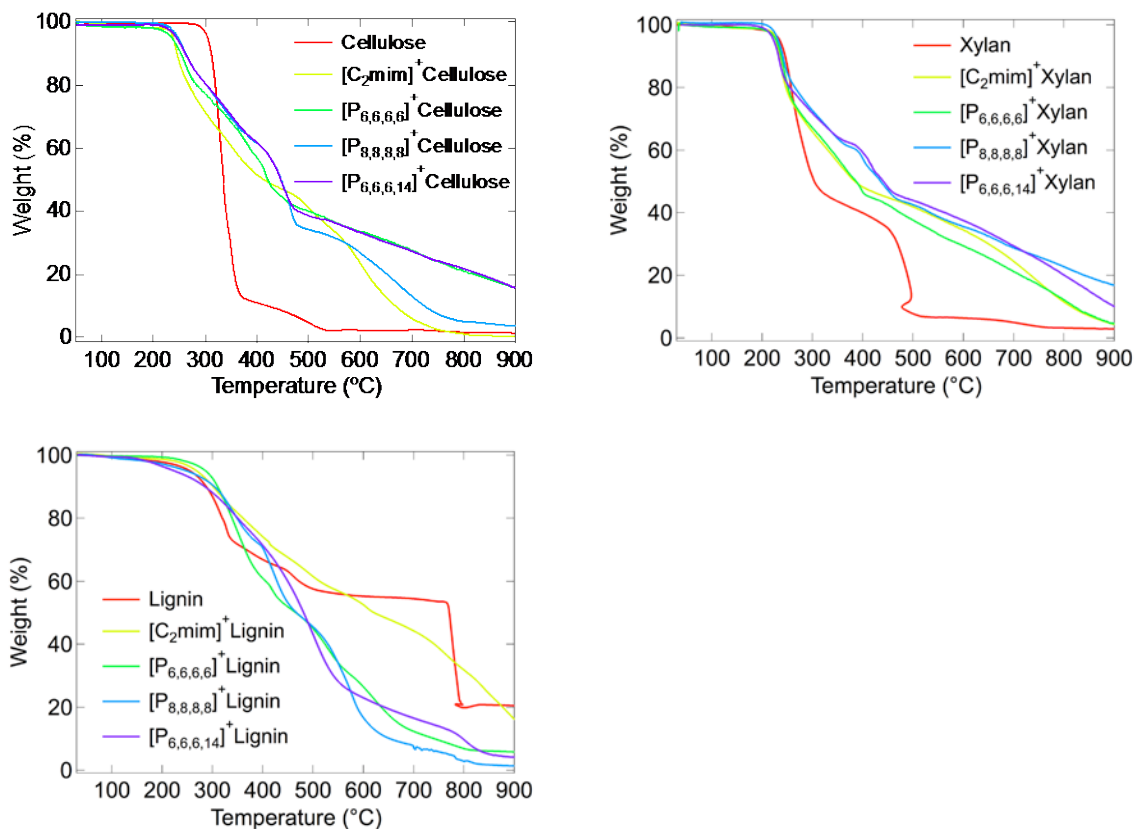


**Fig. S2**  $^1\text{H}$  NMR spectra of [C<sub>2</sub>mim]<sup>+</sup>cellulose, [C<sub>2</sub>mim]<sup>+</sup>xylan and [C<sub>2</sub>mim]<sup>+</sup>lignin in DMSO-*d*<sub>6</sub>.



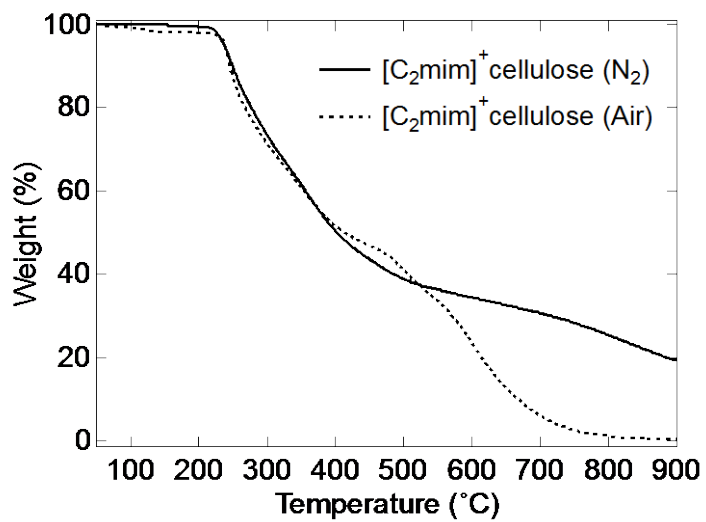
**Fig. S3** FT-IR spectra of the underivatized and derivatized cellulose, xylan, and lignin.

The peaks derived from the P-O-C bond at  $821\text{ cm}^{-1}$ , P=O bond at  $1211\text{ cm}^{-1}$ , and C=N bond at  $1571\text{ cm}^{-1}$  were detected in  $[\text{C}_2\text{mim}]^+$ cellulose,  $[\text{C}_2\text{mim}]^+$ xylan and  $[\text{C}_2\text{mim}]^+$ lignin. The peaks derived from the C-H bond of phosphonium cation at  $2854$  and  $2924\text{ cm}^{-1}$  were also detected in the spectrum of phosphonium-type derivatives. In the case of lignin derivatives, the signals are not strong due to the low derivatization ratio.

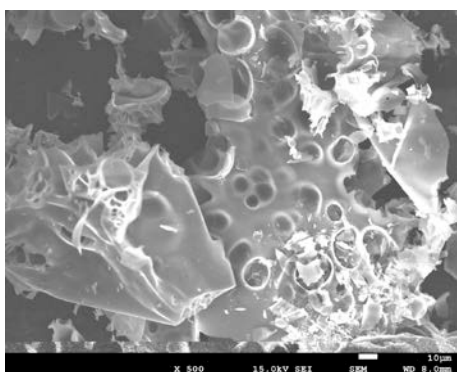
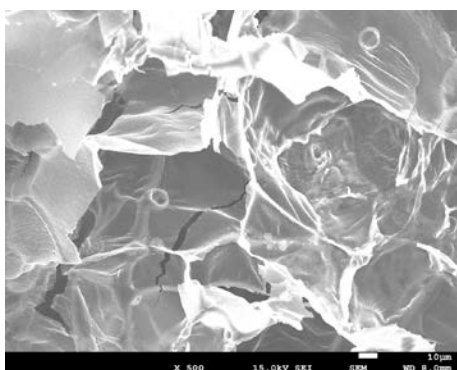
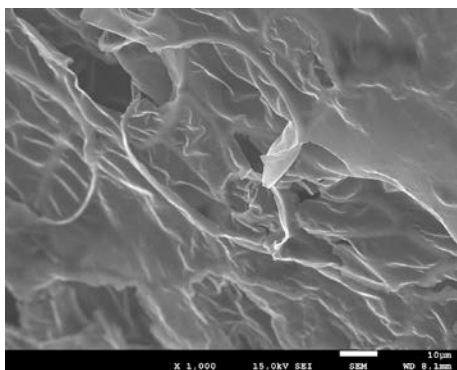


**Fig. S4** TGA curves of the underivatised and derivitised cellulose, xylan, and lignin measured under air condition.

TG signals sometimes behaved abnormally (for example in the case of underivatised xylan) and the abnormal behaviour is attributed to the combustion of the samples because the samples were measured under air condition.

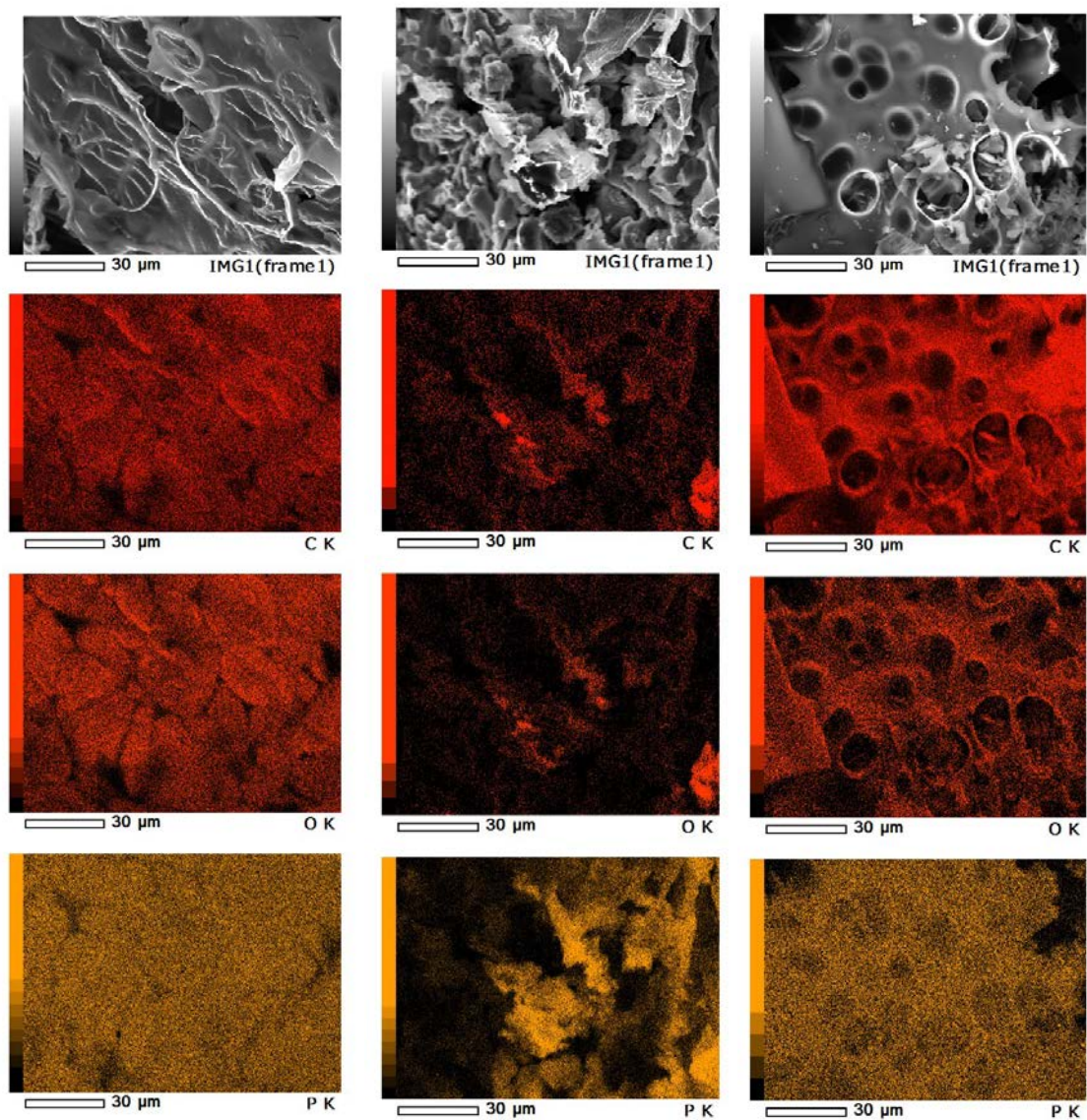


**Fig. S5** TGA curves of [C<sub>2</sub>mim]<sup>+</sup>cellulose measured under air and nitrogen gas conditions.

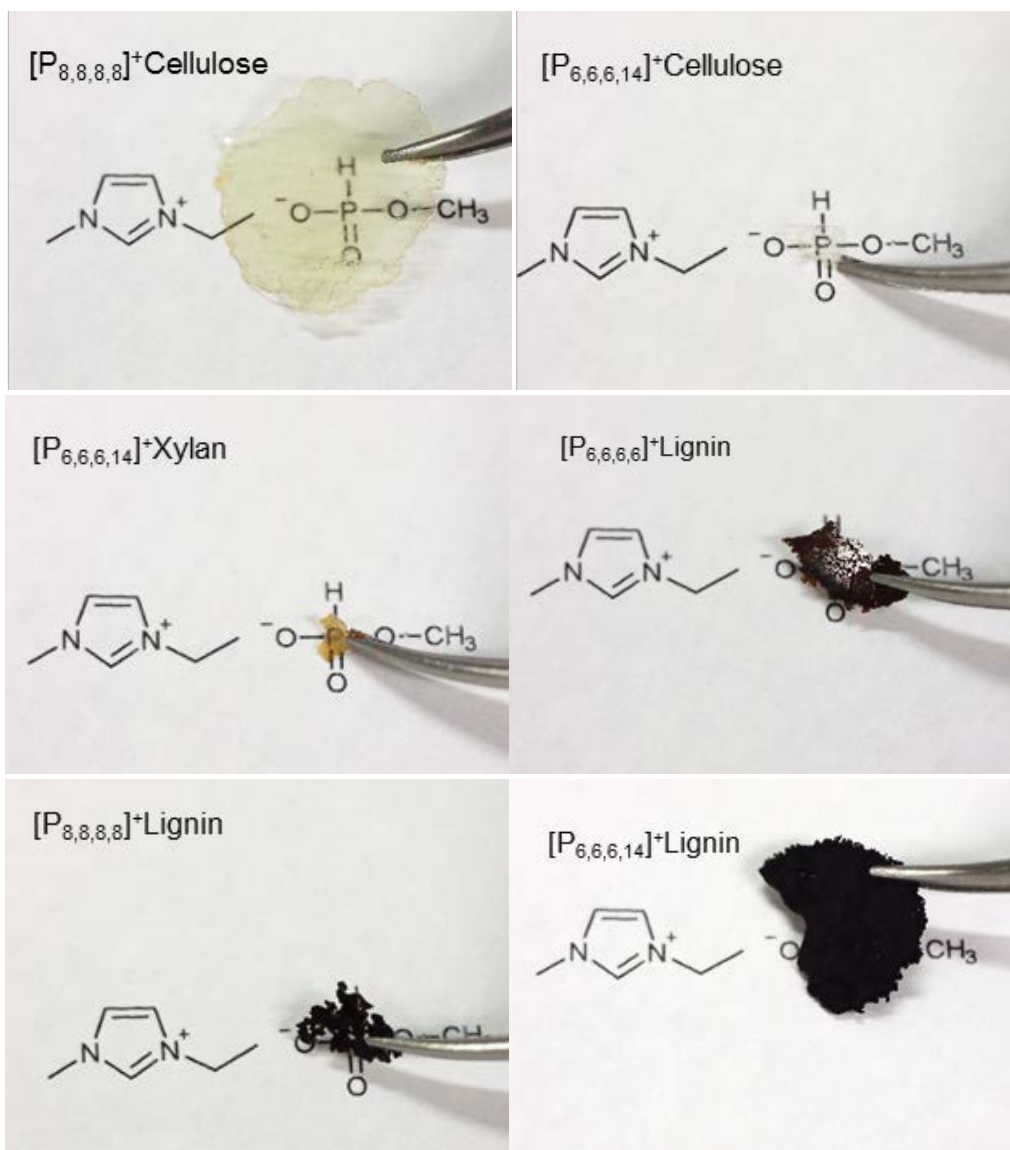


**Fig. S6** SEM images of the char layers of [C<sub>2</sub>mim]<sup>+</sup>cellulose (top), [C<sub>2</sub>mim]<sup>+</sup>xylan (middle), [C<sub>2</sub>mim]<sup>+</sup>lignin (bottom).





**Fig. S7** SEM-EDX images of the char layers of  $[C_2mim]^+$ cellulose (top),  $[C_2mim]^+$ xylan (middle),  $[C_2mim]^+$ lignin (bottom).



**Fig. S8** Thin films of [P<sub>8,8,8,8</sub>]<sup>+</sup>cellulose, [P<sub>6,6,6,14</sub>]<sup>+</sup>cellulose, [P<sub>6,6,6,14</sub>]<sup>+</sup>xylan, [P<sub>6,6,6,6</sub>]<sup>+</sup>lignin, [P<sub>8,8,8,8</sub>]<sup>+</sup>lignin, and [P<sub>6,6,6,14</sub>]<sup>+</sup>lignin after hot pressing.

[P<sub>8,8,8,8</sub>]<sup>+</sup>Cellulose



[P<sub>6,6,6,14</sub>]<sup>+</sup>Cellulose



[P<sub>6,6,6,14</sub>]<sup>+</sup>Xylan



[P<sub>6,6,6,6</sub>]<sup>+</sup>Lignin



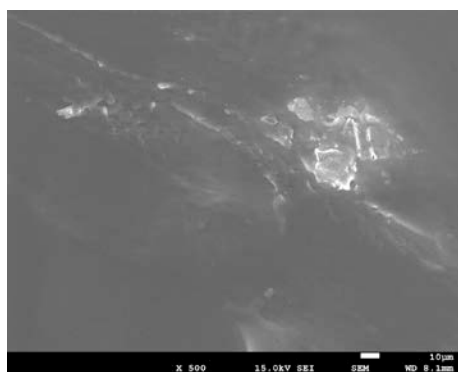
[P<sub>8,8,8,8</sub>]<sup>+</sup>Lignin



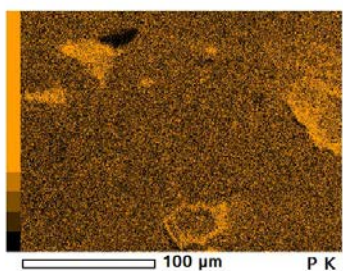
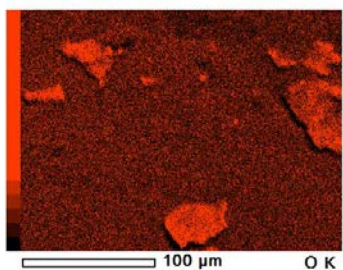
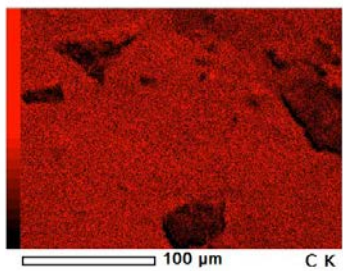
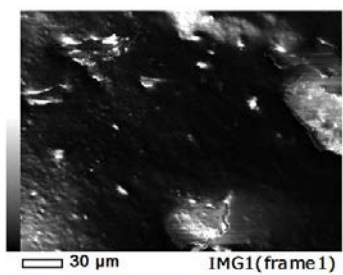
[P<sub>6,6,6,14</sub>]<sup>+</sup>Lignin



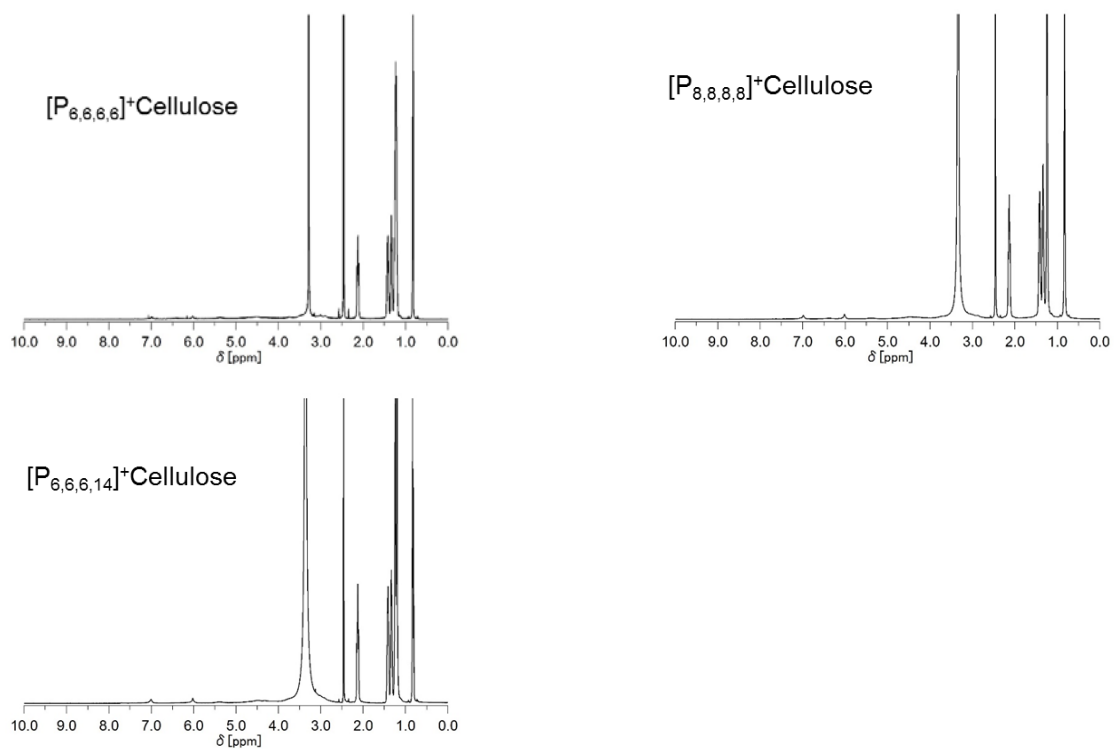
**Fig. S9** Char formation of the thin films of [P<sub>8,8,8,8</sub>]<sup>+</sup>cellulose, [P<sub>6,6,6,14</sub>]<sup>+</sup>cellulose, [P<sub>6,6,6,14</sub>]<sup>+</sup>xylan, [P<sub>6,6,6,6</sub>]<sup>+</sup>lignin, [P<sub>8,8,8,8</sub>]<sup>+</sup>lignin, and [P<sub>6,6,6,14</sub>]<sup>+</sup>lignin after burning and extinguish the fire.



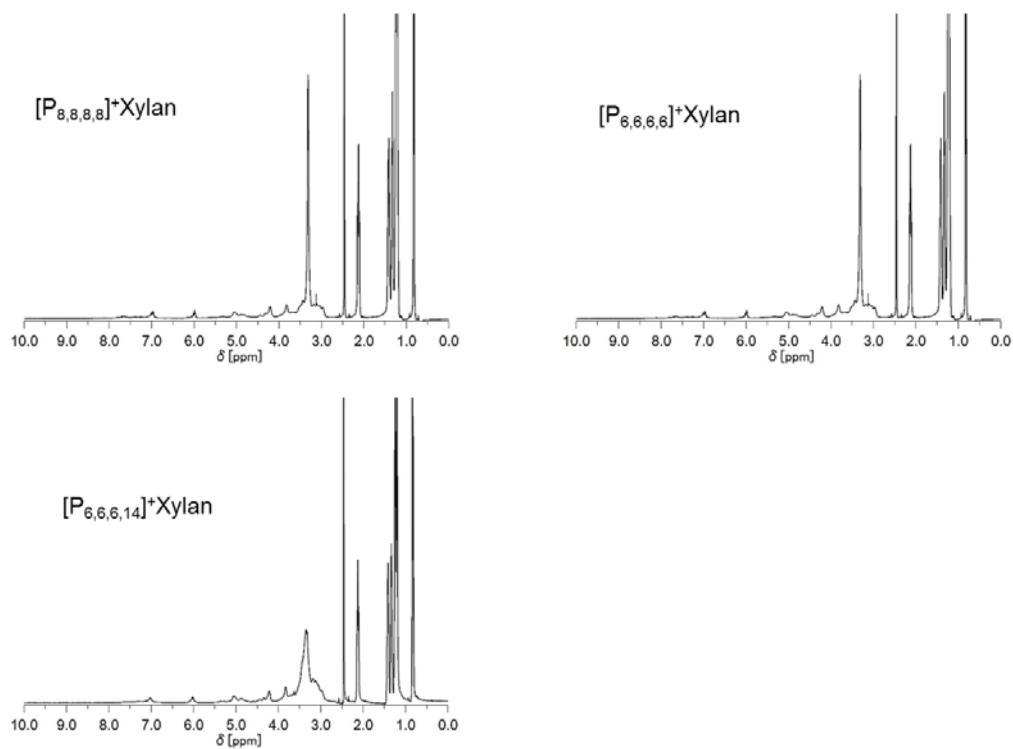
**Fig. S10** A SEM image of the char layer of [P<sub>8,8,8</sub>]<sup>+</sup>cellulose.



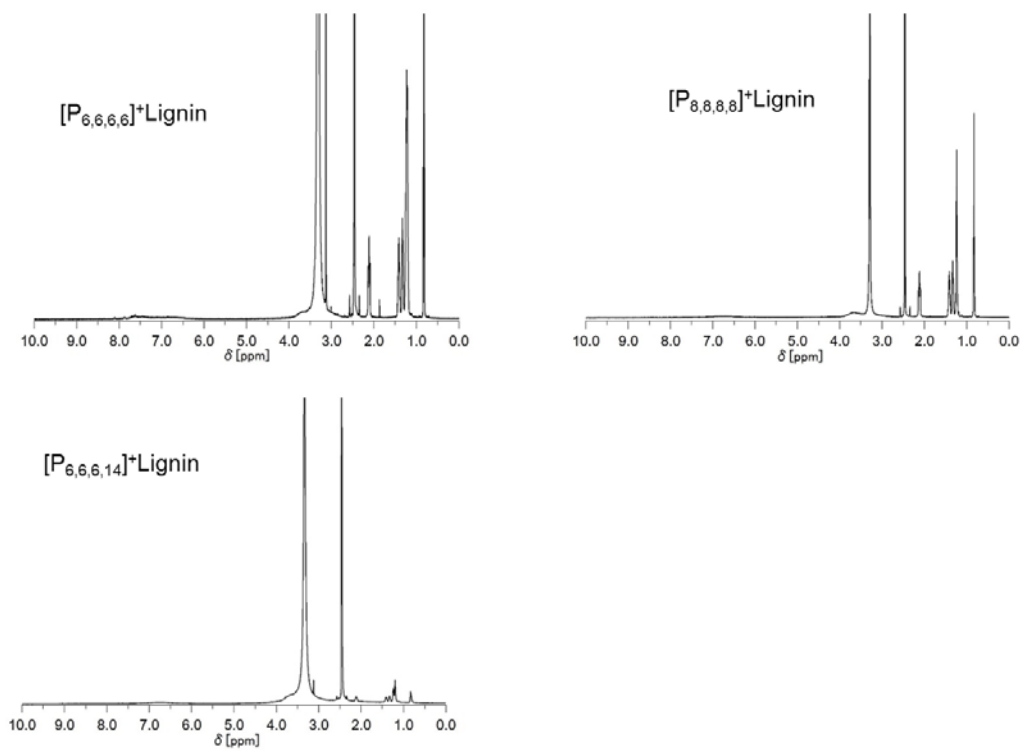
**Fig. S11** A SEM-EDX image of the char layer of  $[\text{P}_{8,8,8}]^+$ cellulose.



**Fig. S12** <sup>1</sup>H NMR spectra of [P<sub>6,6,6,6</sub>]<sup>+</sup>, [P<sub>8,8,8,8</sub>]<sup>+</sup>, and [P<sub>6,6,6,14</sub>]<sup>+</sup>cellulose in DMSO-*d*<sub>6</sub>. The signal at 3.4 ppm is water.



**Fig. S13**  $^1\text{H}$  NMR spectra of  $[\text{P}_{6,6,6,6}]^+$ ,  $[\text{P}_{8,8,8,8}]^+$ , and  $[\text{P}_{6,6,6,14}]^+$  xylan in  $\text{DMSO-}d_6$ . The signal at 3.4 ppm is water.



**Fig. S14**  $^1\text{H}$  NMR spectra of  $[\text{P}_{6,6,6}]^+$ ,  $[\text{P}_{8,8,8}]^+$ , and  $[\text{P}_{6,6,6,14}]^+$ lignin in  $\text{DMSO-}d_6$ . The signal at 3.4 ppm is water.