

Supporting Information

Boosting the Performance of One-Step Solution-Processed Perovskite Solar Cells using a Natural Monoterpene Alcohol as Green Solvent Additive

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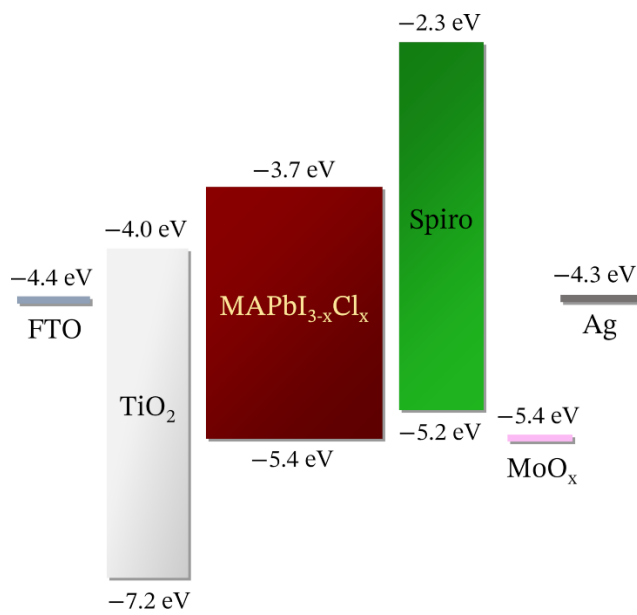


Figure S1. Energy band diagram of the complete planar n-i-p perovskite solar cell. Literature reports for band edges and Fermi levels were used.¹⁻³

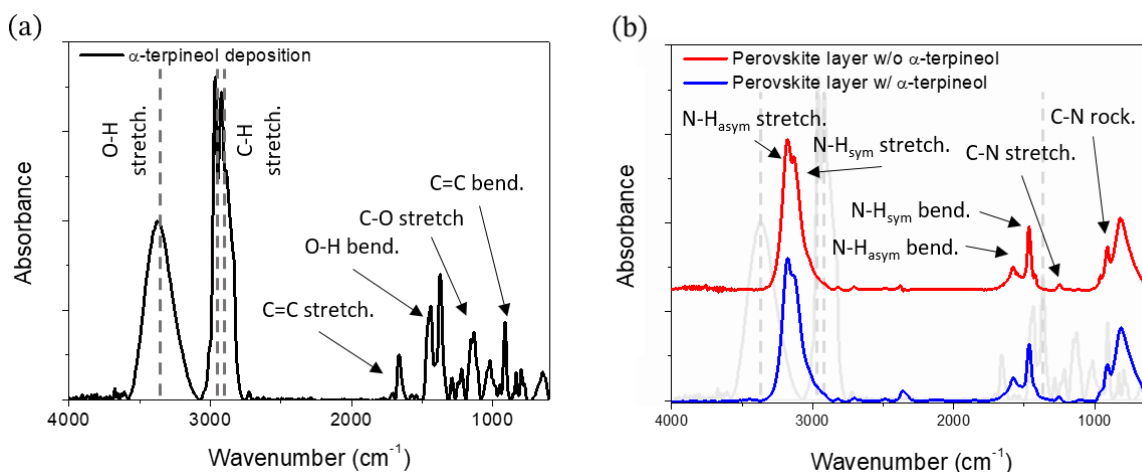


Figure S2. FT-IR spectra registered for: (a) a deposition of a 10 vol% solution of α -terpineol in chloroform over a glass support, left to dry under air at room temperature, and (b) a perovskite layer grown in absence (red line) and presence (blue line) of α -terpineol prepared in accordance with the general procedure reported in Experimental Section in the main text. For the latter spectrum, the absence of pure α -terpineol band (in the background) supports the hypothesis of a

successful removal of the additive when the perovskite layer has been grown in presence of the additive.

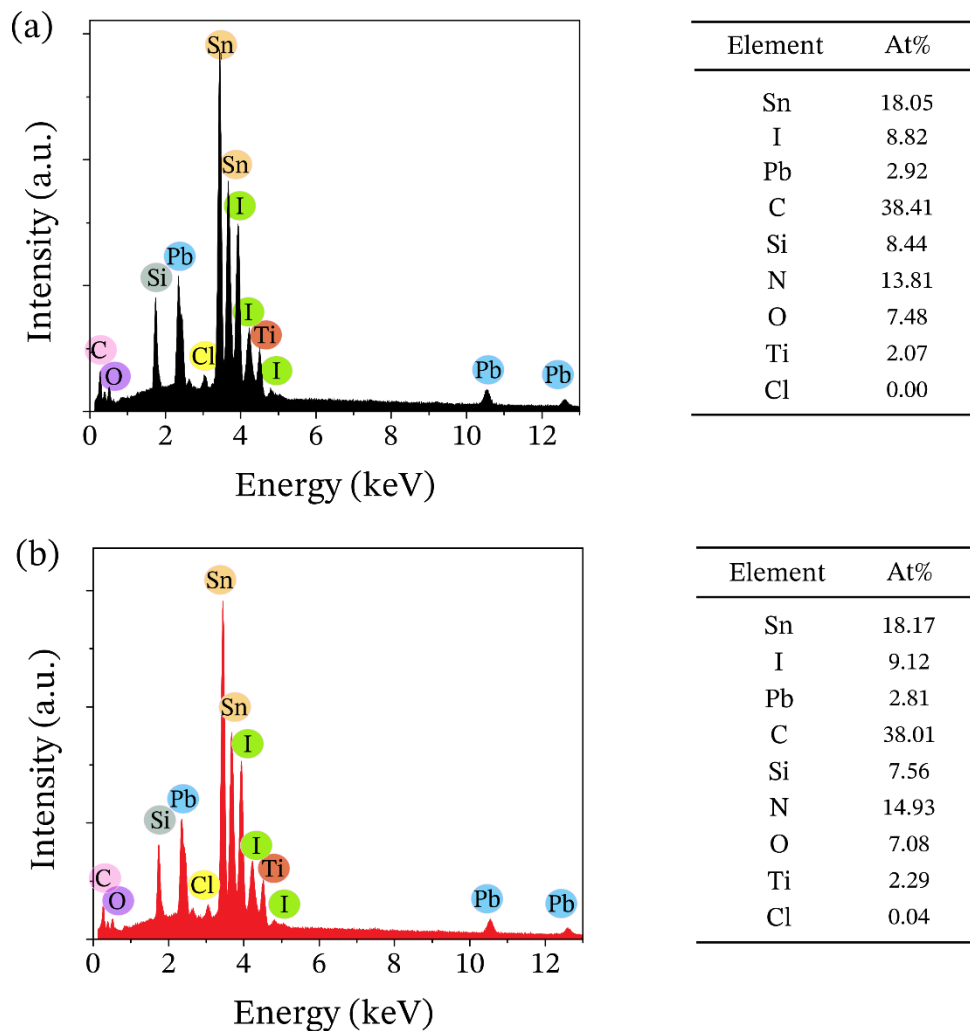


Figure S3. EDX spectra and atomic percentages obtained for $\text{CH}_3\text{NH}_3\text{PbI}_{3-x}\text{Cl}_x$ films deposited onto *c*- TiO_2 -coated FTO/glass substrates from (a) pristine precursor solution and from (b) a precursor solution containing α -terpineol.

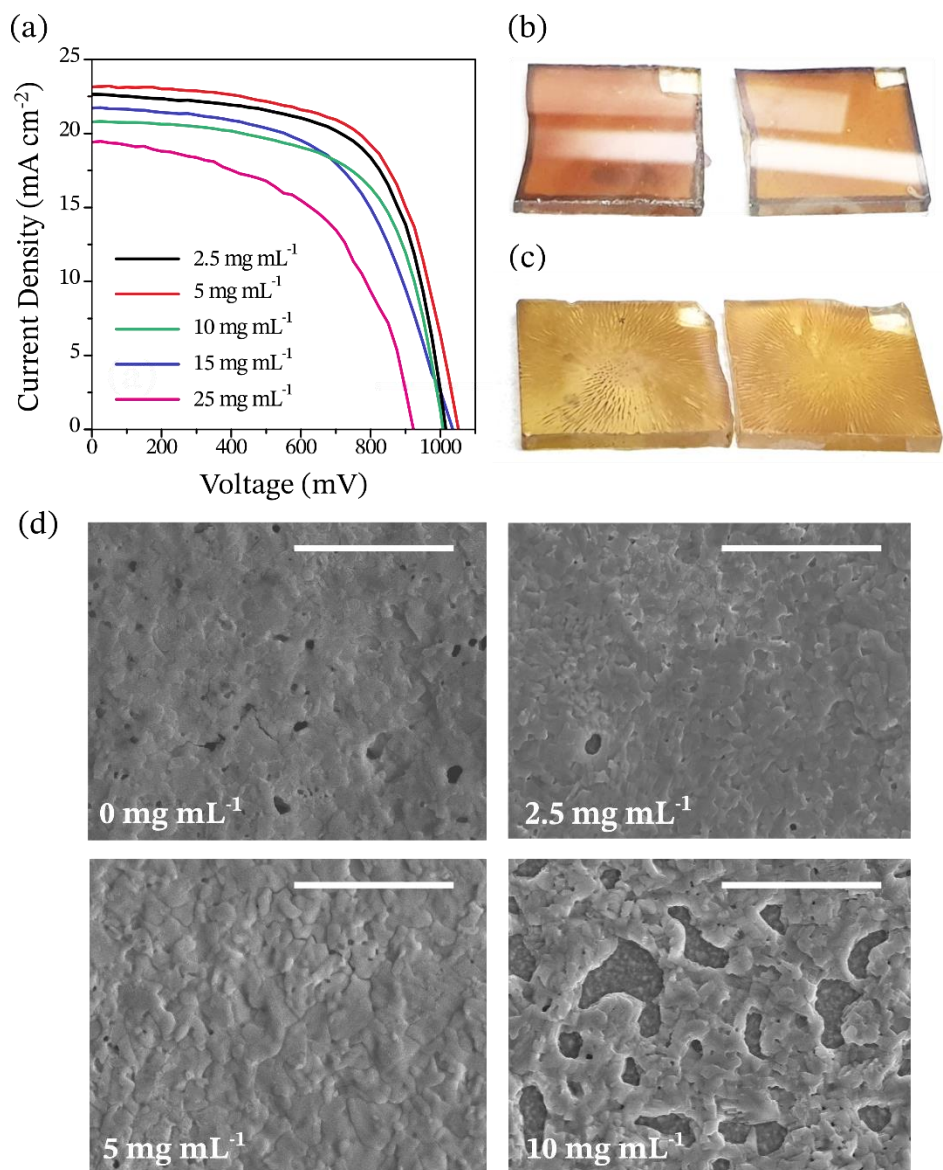


Figure S4. (a) Effect of the α -terpineol additive concentration on the J-V curves of planar n-i-p PSCs. (b) Perovskite precursor films (prior to thermal treatment at 105 °C) obtained by adding 5 mg mL⁻¹ of α -terpineol to the one-step precursor solution. (c) Striation defects in perovskite precursor films obtained by adding 25 mg mL⁻¹ of α -terpineol to the one-step precursor solution. (d) Top-view SEM images of CH₃NH₃PbI_{3-x}Cl_x perovskite films spin-coated on top of *c*-TiO₂/FTO-coated glass substrates from precursor solutions with different concentration of α -terpineol, namely 0 mg mL⁻¹, 2.5 mg mL⁻¹, 5 mg mL⁻¹, and 10 mg mL⁻¹. Scale bar: 5 μ m.

Table S1. Overview on PSC performances (short-circuit current, J_{sc} ; open-circuit voltage, V_{oc} ; fill factor, FF; power conversion efficiency, PCE) obtained under AM1.5G and reverse voltage scanning direction as a function of the α -terpineol additive concentration in the precursor solution.

Additive Content	J_{sc} [mA cm ⁻²]	V_{oc} [mV]	FF [%]	PCE [%]
0 mg mL ⁻¹	23.0	1039	58.2	13.9
2.5 mg mL ⁻¹	22.7	1015	64.0	14.7
5 mg mL ⁻¹	23.2	1053	63.1	15.4
10 mg mL ⁻¹	20.8	1007	62.7	13.1
15 mg mL ⁻¹	21.7	1035	56.1	12.6
25 mg mL ⁻¹	19.4	922	53.1	9.5

Table S2. Recently reported performances of PSCs based on perovskite films prepared by additive-assisted solution processes, and comparison with our best-efficiency PSCs featuring Ag or Au top electrodes. The power conversion efficiencies (PCEs) of the top pristine and additive-treated PSCs are tabulated basing on data available in each reference. Note that the PSCs fabricated in this work exhibit several advantages over the other devices, including simple planar device architecture, simple one-step spin-coating deposition without the need of antisolvents, and non-toxic, low-cost, and eco-friendly nature of the additive employed.

Additive	Device Architecture	Deposition Method	Additive Properties	PCE <i>pristine</i> [%]	PCE <i>treated</i> [%]	Ref.
DIO	FTO/PEDOT:PSS/ MAPbI ₃ /PCBM/Ag (planar)	One-step spin-coating + Antisolvent	Toxic (halogenated) High boiling point Interaction through I-atoms	10.77	12.82	4

DMSO	FTO/c-TiO ₂ /mp-TiO ₂ /MAPbI ₃ /Spiro-OMeTAD/Au (mesoporous)	Two-step spin-dip method (DMSO in Pbl ₂ solution)	Percutaneous absorption High boiling point O-donor Lewis base	12.79	17.26	5
NMP	FTO/c-TiO ₂ /mp-TiO ₂ /MAPbI ₃ /Spiro-OMeTAD/Au (mesoporous)	Two-step spin-dip method (NMP in Pbl ₂ solution)	Reproductively toxic High boiling point O-donor Lewis base	13.3	17.5	6
HMPA	FTO/c-TiO ₂ /mp-TiO ₂ /MAPbI ₃ /PTAA/Au (mesoporous)	Two-step spin-coating (HMPA in Pbl ₂ solution) + Antisolvent	Carcinogen High boiling point O-donor Lewis base	18.02	19.5	7
Dimethyl-acetamide (DMA)	FTO/c-TiO ₂ /mp-TiO ₂ /MAPbI ₃ /Spiro-OMeTAD/Au (mesoporous)	Two-step spin-coating (DMA in Pbl ₂ solution)	Reproductively toxic High boiling point O-donor Lewis base	12.32	16.05	8
ACN	FTO/c-TiO ₂ /mp-TiO ₂ /Cs _x FA _{1-x-y} MA _y Pb(I _{1-z} Br _z) ₃ /Spiro-OMeTAD/Au (mesoporous)	Two-step spin-coating (ACN in MA ⁺ /FA ⁺ /Cs ⁺ solution)	Modest toxicity Low boiling point Weak coordination	13.06	15.64	9
PEG	FTO/c-TiO ₂ /MAPbI _{3-x} Cl _x /Spiro-OMeTAD/Au (planar)	One-step spin-coating No antisolvent	Non-toxic Eco-friendly Low-cost Hydrogen bonding	10.58	13.20	10
DMI	FTO/c-TiO ₂ /mp-TiO ₂ /MAPbI ₃ /Spiro-OMeTAD/Au (mesoporous)	Two-step spin-coating (DMI in Pbl ₂ solution)	Modest toxicity Eco-friendly High boiling point O-donor Lewis base	10.72	14.54	11
α-Terpineol	FTO/c-TiO ₂ /MAPbI _{3-x} Cl _x /Spiro-OMeTAD/MoO _x /Ag (planar)	One-step spin-coating No antisolvent	Non-toxic Eco-friendly Low-cost	13.9	15.4	This work
	FTO/c-TiO ₂ /MAPbI _{3-x} Cl _x /Spiro-OMeTAD/MoO _x /Au (planar)		High boiling point Hydrogen bonding	16.1	17.5	

Table S3. Degradation of PV parameters (short-circuit current, J_{sc} ; open-circuit voltage, V_{oc} ; fill factor, FF; power conversion efficiency, PCE) for solar cells processed from solutions with/without α -terpineol and stored in a N_2 -filled glove box.

Sample	Storage Time [h]	J_{sc} [$mA\ cm^{-2}$]	V_{oc} [mV]	FF [%]	PCE [%]
Without α -terpineol	0	23.0	1039	58.2	13.9
	168	20.1	1027	53.0	11.0
	336	19.6	993	52.3	10.2
	504	15.5	984	52.4	8.0
With α -terpineol	0	23.2	1053	63.1	15.4
	168	23.6	1018	60.7	14.6
	336	21.8	1035	58.8	13.2
	504	21.1	1006	55.2	11.7

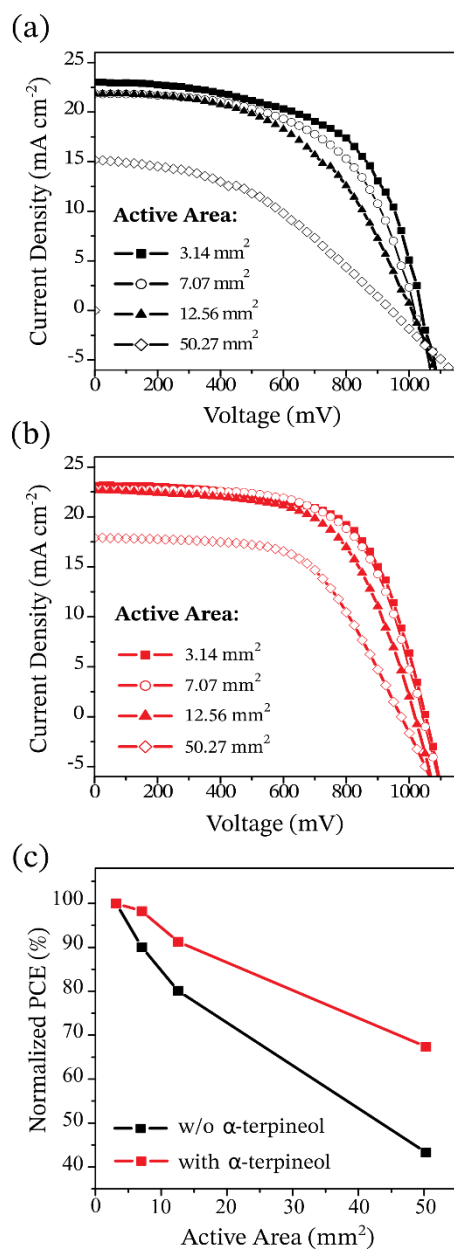


Figure S5. Reverse scanning J-V curves of the best-performance PSCs processed from solutions (a) without α -terpineol and (b) with 5 mg mL^{-1} α -terpineol with increasing of the active area, measured under standard 1 sun illumination. (c) The corresponding variation in PCE with increasing of the active area for pristine and α -terpineol-modified solar cells.

Table S4. Summary of PSC performances for solar cells processed from solutions with/without α -terpineol and featuring different active areas, measured under standard AM 1.5G illumination.

Sample	Active Area [mm ²]	J _{sc} [mA cm ⁻²]	V _{oc} [mV]	FF [%]	PCE [%]
Without α -terpineol	3.14	23.0	1039	58.2	13.9
	7.07	21.9	1018	56.3	12.5
	12.56	21.8	1009	50.6	11.1
	50.27	15.2	943	42.1	6.0
With α -terpineol	3.14	23.2	1053	63.1	15.4
	7.07	22.8	1043	63.6	15.1
	12.56	22.7	1018	60.7	14.0
	50.27	17.9	974	59.4	10.4

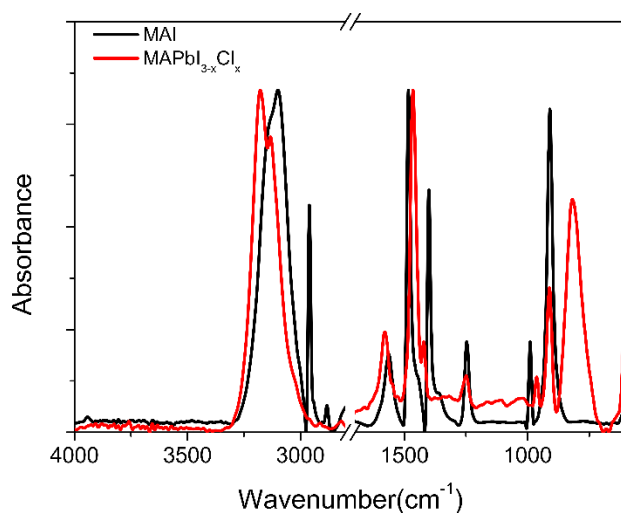


Figure S6. FT-IR spectra registered for: MAI precursor (black line) and a MAPbI_{3-x}Cl_x perovskite layer grown in presence of α -terpineol (red line), prepared in accordance with the general procedure reported in Experimental Section in the main text. For sake of clarity, the spectra have been normalized accordingly to the N-H stretching band intensity (spectral range 4000-2800 cm⁻¹) or to the N-H_{sym} bending band intensity (spectral range 1700-600 cm⁻¹).

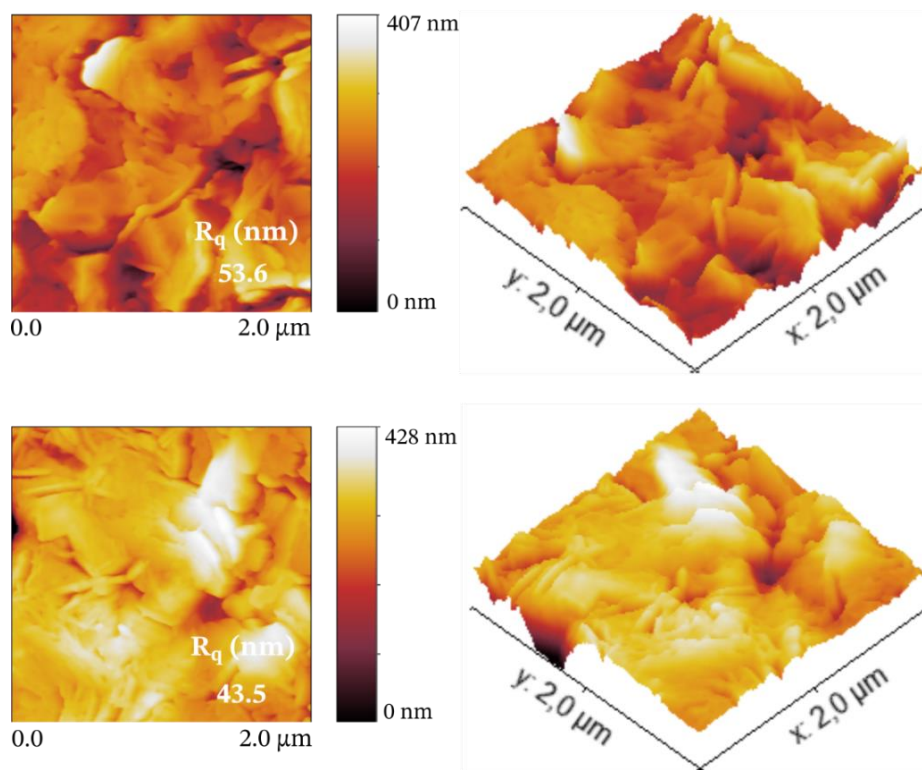


Figure S7. Top-view (left) and 3D (right) AFM images of the pristine perovskite film (above) and of the perovskite film prepared using the α -terpineol additive (below), deposited on top of c -TiO₂/FTO/glass substrates. The surface roughness is noted as the root mean-square value (R_q).

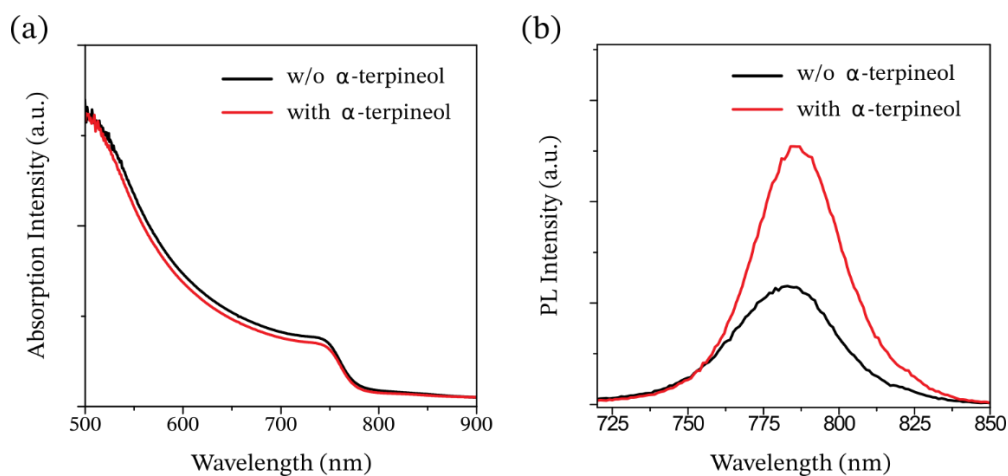


Figure S8. (a) Absorption spectra and (b) steady-state PL spectra of the pristine perovskite films with/without α -terpineol deposited onto FTO/glass substrates.

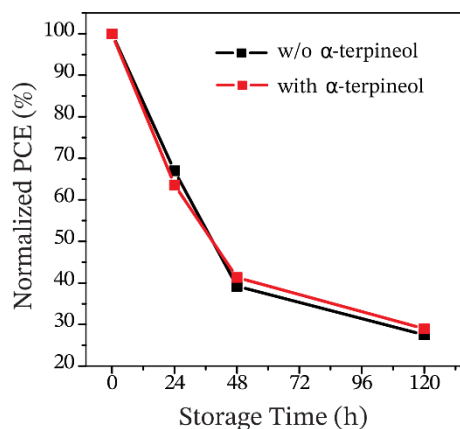


Figure S9. PCE decays of the control and α -terpineol-treated Ag-based devices exposed to ambient air under dark without encapsulation.

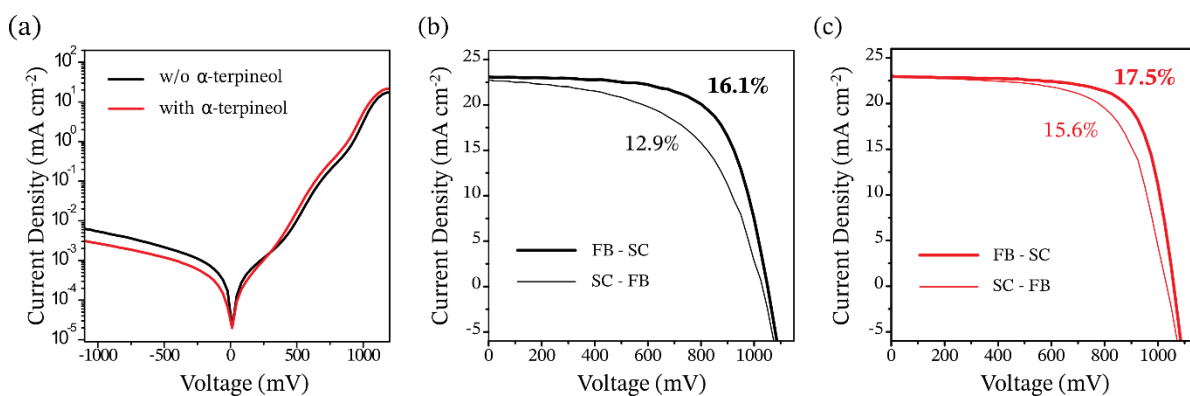


Figure S10. (a) Dark J-V curves (semilog plot) of the best-performance Au-based PSCs processed from solutions with/without α -terpineol. Hysteresis effect in the J-V curves of the top Au-based devices processed from solutions (b) without α -terpineol and (c) with α -terpineol by varying the voltage scanning direction, from forward bias to short-circuit (FB-SC) and from short-circuit to forward bias (SC-FB).

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