# On the synthesis of fused thiazolo[5,4-d] isoxazoles and a novel rearrangement involving conversion of 5(4H)-isoxazolones to 4(5H)-isoxazolones

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The interaction of 4-bromo-3-substituted-(4H)-isoxazol-5-ones (1) with alkyl/aryl thiocarbamides (2), 4-alkyl/aryl thiosemicarbazides (7) and 2, 4-disubstituted thiosemicarbazides (12) afford 5-alkyl/ arylamino-3-substituted-thiazolo [5, 4-d]isoxazoles (3) and 6-amino/anilino-5-alkyl/arylimino-3-substituted-thiazolo[5, 4-d]isoxazoles respectively. An alternate unambiguous one step synthesis is described. The compounds have been characterised by chemical reactions and spectral data.

The compounds containing fused thiazolo[5, 4-d]isoxazoles have been reported earlier as one of the derivatives prepared from 4-chloro-2-phenyliminothiazole-5-carboxaldehyde<sup>1</sup>. We now report an alternate synthesis through readily accessible 4-bromo-3-substituted-(4H)-isoxazol-5-ones with thiocarbamides and thiosemicarbazides.

On refluxing 1a and 2a in ethanol in the presence of pyridine 5-amino-3-phenyl-thiazolo[5, 4-d]isoxazole (3a) was obtained. It showed characteristic IR absorption bands in KBr at 3393 and 3239 (NH<sub>2</sub>), 1649 cm<sup>-1</sup> (vC=N). Its <sup>1</sup>H NMR spectrum (300 MHz in DMSO- $d_6$ ) revealed tautomeric form 3a showing peaks  $\delta$ 7.50-7.55 (m, 5H, Ar-H), 8.45 and 8.68 (2 singlets, 1H, 2 × -NH; D<sub>2</sub>O exchangeable). <sup>13</sup>C NMR (300 MHz in DMSO- $d_6$ ) showed peaks at 126.4 to 132.75 (6 aromatic carbons), 164 and 159 (2×C=N carbons) and 139.5 (>C=C< carbons).

Compound **3a** on treatment with alkyl/aryl isothiocyanates (**4**) in alcoholic alkali afforded 5-(3'alkyl/aryl-thioureido)-3-substituted-thiazolo[5,  $\cdot$ 4-*d*]isoxazoles (**5**) [Scheme-I]. For **5b** IR (KBr) bands were observed at 3179, 3042 and 1610 cm<sup>-1</sup>. Its <sup>1</sup>H NMR spectrum showed peaks at  $\delta 3.76$  (s, 3H, - OCH<sub>3</sub>), 6.9-7.5 [m, 9H, aromatic H containing two doublets at 6.96 and 7.46 (*J*=6.8 Hz each)], 9.16 and 9.34 (s, 1H, 2× -NH; D<sub>2</sub>O exchangeable). <sup>13</sup>C NMR for **5c** showed peaks at 21 (-*C*H<sub>3</sub> carbon), 125.5-135.0 (12 aromatic carbons), 164 and 158 (2× - *C*=N), 138 (> *C*=C <) and 180 ppm (-*C*=S carbon). An alternate unambiguous one-step synthesis of 5 was achieved by the interaction of 1 with dithiobiurets (6).

Similarly 1 and 4-alkyl/aryl thiosemicarbazides (7) in refluxing ethanol and pyridine afforded 6-amino-5-alkyl/arylimino-3-substituted-

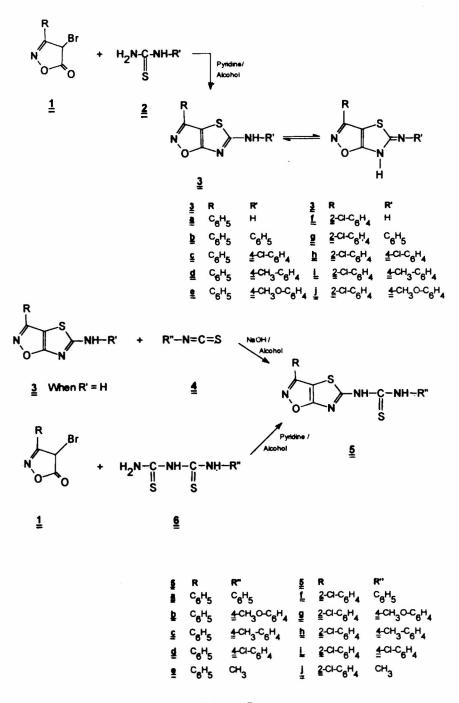
thiazolo[5,4-d]isoxazoles (8) which on reaction with aldehydes (9) in the presence of sodium acetate afforded Schiff's bases (10) (Scheme-II). The relevant spectral data and physical characteristic are listed in Table I.

The structure of 10 has been confirmed by an alternate unambiguous synthesis by the interaction of 1 with thiosemicarbazones (11). The compounds thus obtained were identified through undepressed m.m.p., superimposible IR spectra and co-TLC.

2, 4-Diaryl-thiosemicarbazides (12) and 1 under similar conditions gave 6-anilino-5-arylimino-3-substituted-thiazolo [5, 4-d]isoxazoles (13) (Scheme-II). Compounds 13 could also be obtained by the reaction of 1 with 1, 4-diaryl-thiosemicarbazides (14). The formation of 13 from 12 and 14 could be rationalised as arising by isomerisation of 12 to 14 during heating as reported earlier by Bose<sup>2</sup>.

Compound 3 (0.025 *M*, R=Ph, R'=H) and aqueous sodium hydroxide (0.115 M) were stirred together at ambidient temperature for 4 hr. On acidification with dil. HCl in cold, 3-phenyl-4(5H)-isoxazolone<sup>3</sup> (15a) was obtained in 78% yield. For 15a IR (KBr) spectrum showed bands at 3058, 2962, 1806, 1376, 1162, 1078 and 877 cm<sup>-1</sup>. Its <sup>1</sup>H NMR (300 MHz in CDCl<sub>3</sub>) spectrum showed peaks at  $\delta$ 3.8 (s, 2H, -CH<sub>2</sub>),7.20-

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Scheme I

7.55 (m, 5H, Ar-H). <sup>13</sup>C NMR (300 MHz in CDCl<sub>3</sub>) spectrum showed peaks at 33.98 ( $-CH_2$ ), 126.58-132.15 (6 aromatic carbons), 163.01 (-C=N) and 174.63 (-C=O). In the mass spectrum (EI), m/z peaks were observed at 161 (M<sup>+</sup>), 133, 119, 116, 103, 93, 89 and 77.

The conversion of 5(4H)-isoxazolones into 4(5H)-isoxazolones opens up a novel approach to the synthesis of the almost inaccessible 4(5H)-isoxazolones. The probable mechanism for the

formation of **15** could be rationalised as shown in Scheme III. The formation of thiourea could be explained as arising by the thiohydrolysis of the cyanamide formed during the course of the reaction, as hydrogen sulphide was detected in the reaction. The identity of the thiourea was confirmed by TLC and chemical analysis. Small quantities of elemental sulphur were detected in the reaction mixture. It may have probably reacted with the alkali affording sodium sulphide.

Table I-Characterisation data of Schiff's bases 10						
С	Compd	R	R <sub>1</sub>	R <sub>2</sub>	m.p. (°C)	Yield (%)
1	0a	C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>5</sub>	182	80
1		C <sub>6</sub> H,	C <sub>6</sub> H <sub>5</sub>	$2 - OH - C_6H_4$	178	82
1		C <sub>6</sub> H <sub>5</sub>	C,H,	$4 - CH_3O - C_6H_4$	198	84
		C <sub>6</sub> H <sub>5</sub>	$4 - CH_3 - C_6H_4$	C <sub>6</sub> H <sub>5</sub>	106	85
1		C <sub>6</sub> H <sub>5</sub>	$4 - CH_3 - C_6H_4$	$2 - OH - C_6H_4$	102	82
1		C <sub>6</sub> H <sub>5</sub>	$4 - CH_3 - C_6H_4$	$4 - CH_3O - C_6H_4$	162	81
1		C <sub>6</sub> H <sub>5</sub>	$4 - CH_3O - C_6H_4$		101	84
		C <sub>6</sub> H <sub>5</sub>	$4 - CH_{3}O - C_{6}H_{4}$		105	82
1		C <sub>6</sub> H <sub>5</sub>	$4 - CH_{3}O - C_{6}H_{4}$		98	84
1		C <sub>6</sub> H <sub>5</sub>	$(CH_3)_3 - C$	C,H,	155	79
		C <sub>6</sub> H <sub>5</sub>	$(CH_3)_3 - C$	$2 - OH - C_6H_4$	110	78
1		C <sub>6</sub> H <sub>5</sub>	$(CH_3)_3 - C$	$4 - CH_3O - C_6H_4$	108	80
1	Om	C <sub>6</sub> H <sub>5</sub>	Н	C <sub>6</sub> H <sub>5</sub>	176	76
1	On	C <sub>6</sub> H <sub>5</sub>	Н	$2 - OH - C_6H_4$	156	78
1		C <sub>6</sub> H <sub>5</sub>	Н	$4 - CH_3O - C_6H_4$	172	79
1	0p	$2 - Cl - C_6H_4$	C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>5</sub>	144	82
1	lOq	$2 - Cl - C_6H_4$	C <sub>6</sub> H <sub>5</sub>	$2 - OH - C_6H_4$	141	84
1	lOr	$2 - Cl - C_6H_4$	C <sub>6</sub> H <sub>5</sub>	$4 - CH_{3}O - C_{6}H_{4}$	136	86
1	lOs	$2 - Cl - C_6H_4$	$4 - CH_3 - C_6H_4$	C <sub>6</sub> H <sub>5</sub>	133	85
1	lOt	$2 - Cl - C_6H_4$	$4 - CH_3 - C_6H_4$	$2 - OH - C_6H_4$	146	84
1	lOu	$2 - Cl - C_6H_4$	$4 - CH_3 - C_6H_4$	$4 - CH_{3}O - C_{6}H_{4}$	132	82
	lOv	$2 - Cl - C_6H_4$	$4 - CH_3O - C_6H_4$	C <sub>6</sub> H <sub>5</sub>	148	84
	lOw	$2 - Cl - C_6H_4$	$4 - CH_3O - C_6H_4$	$2 - OH - C_6H_4$	144	85
	lOx	$2 - Cl - C_6H_4$	$4 - CH_3O - C_6H_4$	$4 - CH_3O - C_6H_4$	135	85
	lOy	$2 - Cl - C_6H_4$	$(CH_3)_3 - C$	$C_6H_5$	126	81
	lOz	$2 - Cl - C_6H_4$	$(CH_3)_3 - C$	$2 - OH - C_6H_4$	140	80
	lOaa	$2 - Cl - C_6H_4$	$(CH_3)_3 - C$	$4 - CH_3O - C_6H_4$	137	79
	lObb	$2 - Cl - C_6H_4$	Н	C <sub>6</sub> H <sub>5</sub>	188	77
	locc	$2 - Cl - C_6 H_4$	н	$2 - OH - C_6H_4$	184	76
	10dd**	$2 - Cl - C_6 H_4$	Н	$4 - CH_3O - C_6H_4$	198	78
factory C, H, N and S analyses were obtained within 0.3% error						

Satisfactory C, H, N and S analyses were obtained within 0.3% error

\*10c: PMR :  $\delta 3.76$  (s, 3H, -OCH<sub>3</sub>), 6.84-7.93 (m, 14H, ArH), 8.14 (s, 1H, -CH); <sup>13</sup>C NMR : 55.71 (-OCH<sub>3</sub> carbon), 114.93 (> C = C < carbon), 126.38-134.56 (18 aromatic carbons), 143.2, 152.5 and 157.4 ppm (-C=N carbons) \*10dd : PMR :  $\delta$  3.81 (s, 3H, -OCH<sub>3</sub>), 7.00-7.87 (m, 8H, aromatic H), 8.28 (s, 1H, -CH), 9.44 (s, 1H, -NH); <sup>13</sup>C NMR : 55.83 (-OCH<sub>3</sub> carbon), 114.73 (> C = C < carbon), 125.69-132.98 (12 aromatic carbons), 152.0, 162.37 & 164.29 (3 × -C=N carbons)

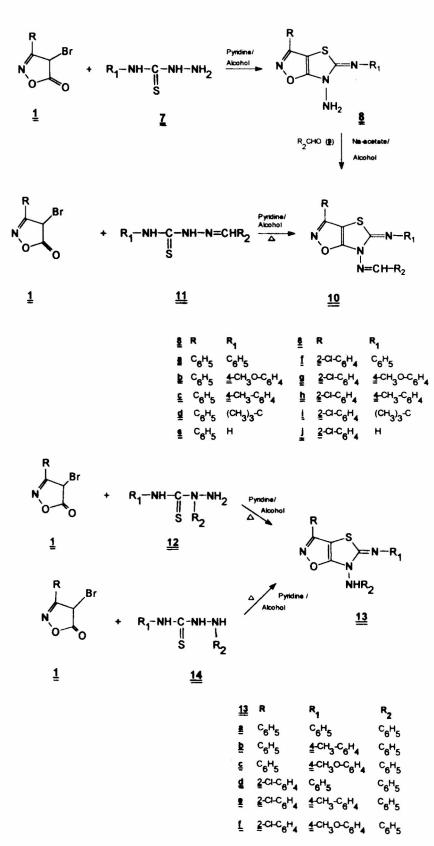
Compound **15a** reacted with bromine in acetic acid in presence of UV light to afford 5-bromo derivative **16a**. Its IR (KBr) spectrum showed bands at 3025, 1813, 1380, 1159, 1080 and 874 cm<sup>-1</sup>. On reaction with thiocarbamide it gave 5-amino-3-phenyl-thiazolo [4, 5-*d*]isoxazole (**17a**). Its <sup>1</sup>H NMR spectrum revealed tautomeric form of **17a** by showing peaks at  $\delta$ 7.45-7.64 (m, 5H, Ar-H), 8.65 & 8.18 (s, 1H 2× - NH; D<sub>2</sub>O exchangeable). <sup>13</sup>C NMR spectrum showed peaks at 125.65-132.50 (6 aromatic carbons), 163 and 158 (2×C=N), 137.50 (>C=C< carbons).

## **Experimental Section**

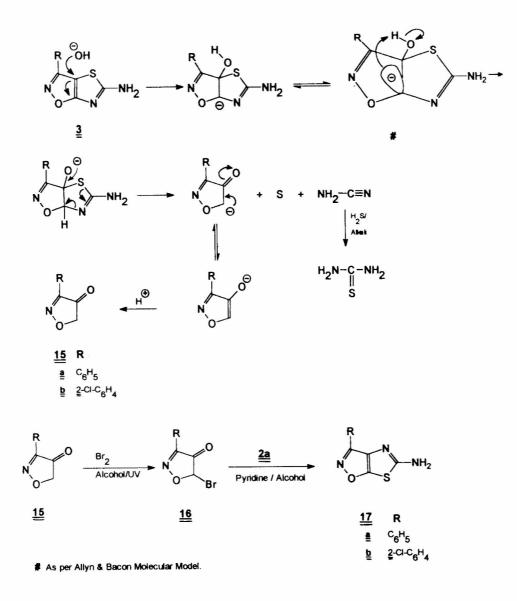
Melting points were determined in open glass capillaries and are uncorrected. IR were recorded on Perkin-Elmer 1600 series FTIR and PMR and <sup>13</sup>C NMR on 300 MHz VXRO using DMSO-

deuterated solvent. TLC in various solvents showed the compounds to be homogeneous. Compounds 1, 2, 6, 7, 11, 12, 14 were prepared by literature methods<sup>4-9</sup>.

5-Alkyl/aryl amino-3-substituted-thiazolo [5, 4-*d*]isoxazoles (3): General procedure. A mixture of 1 (0.01 mole), 2 (0.01 mole) and pyridine (0.02 mole) in ethanol (25 mL) was refluxed for 3 hr, cooled and poured onto crushed ice when compound 3 was obtained. It was recrystallised from ethanol. The compounds **3a-j** were similarly synthesized. **3a**: (m.p. 230°)75%); **3b**: m.p. 96° (85%); **3c**: m.p. 105° (86%); <sup>1</sup>H NMR :  $\delta$ 8.4 (s, 1H, -NH), 6.8-7.8 (m, 9H, ArH); **3d**: m.p. 174° (85%); **3e** : m.p. 124° (85%); **3f**: m.p. 215° (75%); <sup>1</sup>H NMR :  $\delta$  8.32 (s, 1H, -NH), 8.58 (s, 1H, --NH), 7.38-7.53 (m, 4H, ArH); <sup>13</sup>C NMR :



Scheme II





127.36-133.0 (6 aromatic carbons), 164 and 159  $(2 \times -C = N \text{ carbons})$ , 137 ppm (>C = C < carbons); 3 g, m.p. 106° (85%); **3h**: m.p. 152° (87%); **3i**: m.p. 130° (85%); **3j**: m.p. 102° (85%).

5-(3'-Alkyl/aryl thioureido)-3-substituted-thiazolo [5, 4-*d*]isoxazoles (5): General procedure. A mixture of 3 (R'=H, 0.01 mole), 4 (0.01 mole) and NaOH (0.01 mole) in ethanol (25 mL) was refluxed for 3.5 hr, cooled and poured onto crushed ice. It was further brought to *p*H 6 by adding cold dil. HCl when compound 5 separated out. It was recrystallised from dil. ethanol. Compounds 5a-d were similarly synthesised. 5a: m.p. 102° (88%); <sup>13</sup>C NMR : 124.50-129.20 (12 aromatic carbons), 164.50 and 157.88 (2×*C*=N carbons), 137.90 (>*C*=*C*< carbon), 179.71 ppm (-*C*=S carbon); **5b**: m.p. 158° (89%); **5c**: m.p. 145° (88%); **5d** : m.p. 140° (86%); <sup>1</sup>H NMR :  $\delta$  6.82 - 8.10 (m, 9H, ArH), 9.24 and 9.35 (s, 1H, 2× - NH); **5e** : m.p. 149° (70%); **5f** : m.p. 108° (68%); **5g**: m.p. 145° (88%); **5h** : m.p. 141° (87%); **5i** : m.p. 135° (85%); **5j** : m.p. 138°, 72%).

An alternate unambiguous synthesis of 5: General procedure A mixture of 1 (0.01 mole), 6 (0.01 mole) and pyridine (0.02 mole) in ethanol (25 mL) was refluxed for 3 hr, cooled and poured onto crushed ice, when compound 5 was obtained. It was recrystallised from dil. ethanol. Compounds 5a-j were synthesised and characterised by co-TLC, superimposable IR with authentic samples prepared as described earlier.

6-Amino-5-alkyl/arylimino-3-substituted-thiazolo [5,4-d]isoxazoles (8): General procedure. A mixture of 1 (0.01 mole), 7 (0.01 mole) and pyridine (0.02 mole) in ethanol (25 mL) was refluxed for 3 hr, cooled and poured onto crushed ice when compound 8 was obtained. It was recrystallised from dil. ethanol. The compounds 8a-j were synthesized. 8a: m.p. 142° (88%); 8b: m.p. 168°, (89%); IR(KBr) : 3285 and 3185 (vNH<sub>2</sub>), 1606 (vC = N), 1245, 1028 and 755 cm<sup>-1</sup>; <sup>1</sup>H NMR :  $\delta$  3.71 (s, 3H, -OCH<sub>3</sub>), 9.55 (s, 2H, -NH<sub>2</sub>; D<sub>2</sub>O exchangeable), 6.84-7.54 [m, 9H, aromatic H containing two doublets for 4-methoxyphenyl; at 6.87 (J=9.08 Hz) and 7.43 (J=9.08 Hz)]; <sup>13</sup>C : 55.62  $(-OCH_3 \text{ carbon}),$ NMR 113.67 (>C=C< carbons), 114.56-128.59 (12 aromatic carbons), 154.30 and 156.29 ppm  $(2 \times -C = N)$ carbons); 8c: m.p. 171° (88%); 8d: m.p. 182° (74%); <sup>1</sup>H NMR :  $\delta 1.20$  (s, 6H,  $2 \times - CH_3$ ), 1.46 (s, 3H,  $1 \times -CH_3$ ), 9.76 (s, 2H,  $-NH_2$ ), 7.39-7.61 (m, 5H, ArH); 8e: m.p. 189° (72%); <sup>1</sup>H NMR :  $\delta 9.62$  (s, 2H,  $-NH_2$ ), 5.31 (s, 1H, -NH), 7.29-7.92 (m, 5H, ArH); 8f : m.p. 105°C (86%); 8g; 116° (86%); 8h: m.p. 122° (89%); <sup>13</sup>C NMR: 26 (-CH<sub>3</sub> carbon), 112.07 (>C=C< carbons), 117.74-134.90 (12 aromatic carbons), 156.92 and 158.99 ppm (2×-C=N carbons); 8i: m.p. 155° (75%); 8j: m.p. 175° (72%).

Schiff's bases (10). A mixture of 8 (0.01 mole), 9 (0.01 mole) and fused sodium acetate (0.01 mole) was refluxed in methanol for 3-4 hr. The reaction mixture was then poured onto crushed ice to get 10. It was further recrystallised from dil. ethanol. The compounds 10a-dd were synthesised which are incorporated in Table I.

## 6-Anilino-5-alkyl/arylimino-3-substituted-

thiazolo [5,4-d] isoxazoles (13): General procedure. A mixture of compound 1 (0.01 mole), 12 (0.01 mole) and pyridine (0.02 mole) in ethanol (25 mL) was refluxed for 3 hr, cooled and poured onto crushed ice when compound 13 was obtained. It was recrystallised from dil. ethanol. The compounds 13a-f were similarly synthesised. 13a : m.p. 102° (86%); IR(KBr) : 3282, 1594, 1454, 1315 and 753 cm<sup>-1</sup>; <sup>1</sup>H NMR : δ 8.98 (s, 1H, NHAr), 6.99 to 7.69 (m, 15H, aromatic H); <sup>13</sup>C NMR : 122.60-129.95 (18 aromatic carbons), 146.15 and 149.45  $(2 \times - C = N \text{ carbons})$ , 113.85 ppm (>C = C < carbons), **13b** : m.p. 148° (84%); **13C** : m.p.  $168^{\circ}$  (85%); **13d** : m.p.  $162^{\circ}$  (82%); **13e** : m.p.  $142^{\circ}$  (84%); <sup>1</sup>H NMR :  $\delta$  2.25 (s, 3H,  $-CH_3$ , 5.48 (s, 1H, -NH), 6.87-7.66 (m, 13H, aromatic H); <sup>13</sup>C NMR : 20.8 ( $-CH_3$  carbon), 119.36 (> C = C < carbons), 120. 17-130.85 (18 aromatic carbons), 149.85 and 145.94 ppm (2 × -C = N carbons); **13f** (134° (85%).

An alternate unambiguous synthesis of 10: General procedure. Compounds 11 were prepared by refluxing 4-alkyl/aryl thiosemicarbazides (7) with aldehydes (9) in ethanol for 2 hr. The reaction mixture in each case was left overnight, filtered and crystallised from ethanol.

A mixture of 1 (0.01 mole), 11 (0.01 mole) and pyridine (0.02 mole) was then refluxed in ethanol (25 mL) for 3-4 hr, cooled and poured onto crushed ice, when compound 10 was obtained. It was recrystallised from dil. ethanol. Further confirmation of the identity was done by superimposable IR and m.m.p. determination with an authentic sample prepared as described earlier.

An alternate synthesis of 13 : General procedure. A mixture of 14 (0.01 mole), 1 (0.01 mole) and pyridine (0.02 mole) in ethanol (25 mL) was refluxed for 3 hr, cooled and poured onto crushed ice, when compound 13 was obtained. It was recrystallised from dil. ethanol. Its identity was further confirmed with an authentic sample prepared as described earlier.

**3-substituted-4(5H)-isoxazolones (15) : General procedure.** Compound **3** (R'=H, 0.025 mole) and NaOH (0.115 mole) in distilled water (50 mL) were stirred vigorously for 4-5 hr. It was further acidified to *p*H 6 with cold dil. HCl, when white creamy compound **15** separated out. A strong smell of hydrogen sulphide was perceptible. It was washed several times with cold water and further recrystallised from dil. ethanol. **15a** : m.p. 125° (Lit. 124°) (78%); **15b:** m.p. 96 ° (77%); <sup>1</sup>H NMR :  $\delta$  4.4 (s, 1H, -CH), 5.55 (s, 1H, -OH tautomeric form), 7.20-7.80 (m, 4H, aromatic H); <sup>13</sup>C NMR : 36.5 (-CH<sub>2</sub> carbon) 128.0-134.5 (6 aromatic carbons), 163.0 (-C=N carbon) and 173.0 ppm (> C = O carbon).

#### 5-Bromo-3-substituted-4(5H)-isoxazolones

(16). Compound 15 (0.01 mole) was dissolved in minimum amount of acetic acid and to it bromine (0.01 mole) in acetic acid was added slowly in the presence of UV light during one hr. The stirring was continued for 30 min and then poured onto crushed ice when compound 16 was obtained. It was recrystallised from chloroform and immediately used for further reaction as it was found to be unstable. 16a : m.p. 114° (75%); IR (KBr) :

3025, 1815, 1380, 1159, 1080 and 874 cm<sup>-1</sup>; **16b** : m.p. 106° (77%).

5-Amino-3-substituted-thiazolo[4,5-d] isoxazoles (17) : General procedure. Compound 16 (0.01 mole), thiocarbamide (0.02 mole) and pyridine (0.02 mole) in ethanol (20 mL) were refluxed for 3 hr, cooled and poured onto crushed ice when compound 17 was obtained. It was recrystallised from dil. ethanol. 17a: m.p. 156° (72%); 17b, (m.p. 144° (71%).

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## References

- 1 Pawar R A & Rajput A P, Indian J Chem, 28B, 1989, 866.
- 2 Bose P K & Raychaudhury P C, J Indian Chem Soc, 4, 1927, 257.
- 3 Desimoni G, Ann Chim (Rome), 58, 1968, 1363; Chem Abstr, 70, 1968, 68225c.
- 4 Posner T, Ber dt Chem Ges, 39, 1906, 3521.
- 5 Joshua C P, J Org Chem, 28, 1963, 1293.
- 6 Dixit S N, J Indian Chem Soc, 38, 1961, 44.
- 7 Konher M V, Interaction of amines with 2, 5-diamino/thiosubstituted-1, 3, 4-thiadiazolidines and related system, Ph.D thesis, Nagpur University, Nagpur, 1971.
- 8 Tisler M Z, Anal Chem, 155, 1957, 86; 150, 1956, 345; 148, 1956, 164.
- 9 Holtzman B, Ber dt Chem Ges, 34, 1901, 320.