- 1 Identification of free amino acids in several crude extracts of two legumes
- 2 using Thin Layer Chromatography
- 3 Authors
- 4 Taghread Hudaib
- 5 Sarah Brown
- 6 Daniel Wilson
- 7 Paul E. Eady
- 8
- 9 Key words
- 10 Amino acids; Aqueous-ethanolic extract; Thin layer chromatography; Organic
- 11 solvents; Vigna unguiculata; Phaseolus vulgaris.
- 12

## 13 **1. Introduction**

- 14 Amino acids are primary metabolites and nutritional organic compounds that are the
- 15 building blocks of proteins [1]. The standard (proteinogenic) amino acids consist of
- 16 an amine group (NH2), a carboxylic acid (COOH), alpha hydrogen and a side chain
- 17 [2]. Thin layer chromatography (TLC) is a simple, qualitative, sensitive and widely
- 18 used method for the separation and identification of amino acids in plant extracts.
- 19 Amino acids vary in their solubility in water and organic solvents depending on the
- 20 nature of their side chains; water soluble amino acids have polar side chains [3].

Taghread Hudaib \* University of Lincoln, School of Life Sciences, Brayford Pool, Lincoln, LN6 7TS, UK. Telephone number +44(0)1522 83 5326 Email: <u>thudaib@lincoln.ac.uk</u> Fax: +44 1522 886974

Sarah Brown Applied Materials Technology Ltd, Lyndon Business Park, Farrier Road, Lincoln, LN6 3RU, UK. Email: sarah.brown@appmat.co.uk

Daniel Wilson University of Lincoln. School of pharmacy, Brayford Pool, Lincoln, LN6 7TS Telephone number +44 (0)1522 886825 Email: dwilson@lincoln.ac.uk

In general, the 70% aqueous ethanol is the most preferable solvent to prepare crude
extracts prior to amino acid TLC analysis [4].

23	In this study, the amino acid profiles of seed cotyledons and seed coats from two
24	types of legumes; black-eyed beans (Vigna unguiculata) and red kidney beans
25	(Phaseolus vulgaris) were determined and compared in order to evaluate their
26	nutritional value as major source of plant amino acids. To improve the separation
27	and detection of free amino acids, the TLC technique was applied on several crude
28	extracts obtained at the primary extraction stage using a range of polar and nonpolar
29	solvents. The obtained crude extracts are:
30	1. Kidney bean seeds 70% Ethanol (KBS 70% Ethanol)
31	2. Kidney bean seeds 100% Methanol (KBS100% Methanol)
32	3. Kidney bean seeds 100% Acetone (KBS 100% Acetone)
33	4. Kidney bean seed coat 70% Ethanol (KBSC 70% Ethanol)
34	5. Kidney bean seed coat 100% Methanol (KBSC 100% Methanol)
35	6. Black-eyed bean seeds 70% Ethanol (BEBS 70% Ethanol)
36	7. Black-eyed bean seeds 100% Methanol (BEBS100% Methanol)
37	8. Black-eyed bean seeds 100% Acetone (BEBS 100% Acetone)
38	9. Black-eyed bean seed coat 70% Ethanol (BEBSC 70% Ethanol)
39	10.Black-eyed bean seed coat 100% Methanol (BEBSC100% Methanol)
40	11. Kidney bean seeds 100% Chloroform (KBS 100% Chloroform)

Taghread Hudaib \* University of Lincoln, School of Life Sciences, Brayford Pool, Lincoln, LN6 7TS, UK. Telephone number +44(0)1522 83 5326 Email: <u>thudaib@lincoln.ac.uk</u> Fax: +44 1522 886974

Sarah Brown Applied Materials Technology Ltd, Lyndon Business Park, Farrier Road, Lincoln, LN6 3RU, UK. Email: <u>sarah.brown@appmat.co.uk</u>

Daniel Wilson University of Lincoln. School of pharmacy, Brayford Pool, Lincoln, LN6 7TS Telephone number +44 (0)1522 886825 Email: <u>dwilson@lincoln.ac.uk</u>

41 12. Black-eyed bean seeds 100% Chloroform (BEBS 100% Chloroform)

### 42 **2. Experimental**

#### 43 2.1. Crude extracts preparation

44 Two kilograms of black-eyed beans and two kilograms of red kidney-beans were 45 soaked in distilled water for 30 and 45 min respectively to remove the seed coat. Then, seeds and seed coats were left to dry at room temperature. After that, the 46 47 seeds and seed coats were ground to flour using a coffee grinder and de-fatted using a nonpolar solvent (hexane). The ground seeds and seed coats were extracted 48 49 using solvent extraction methods [5] in which solvents with different polarity were 50 used in Soxhlet extraction [6]. For both the decorticated seeds and seed coats, 500 ml of solvent were needed for extraction of 100g of flour along the extraction 51 52 process. The process continued until the resulting filtrate is colourless, which 53 indicates that most of the soluble constituents have been extracted [7]. The following 54 selective aqueous and organic solvents were used to prepare the crude extracts: i. Aqueous-ethanol (70% ethanol), Polarity = 1.0 for water and 0.65 for ethanol. 55 ii. Methanol (100%), Polarity = 0.7656 Acetone (100%), Polarity = 0.3657 iii. Chloroform (100%), Polarity = 0.2658 iv.

Sarah Brown Applied Materials Technology Ltd, Lyndon Business Park, Farrier Road, Lincoln, LN6 3RU, UK. Email: sarah.brown@apomat.co.uk

Daniel Wilson University of Lincoln. School of pharmacy, Brayford Pool, Lincoln, LN6 7TS Telephone number +44 (0)1522 886825 Email: dwilson@lincoln.ac.uk

Taghread Hudaib \* University of Lincoln, School of Life Sciences, Brayford Pool, Lincoln, LN6 7TS, UK. Telephone number +44(0)1522 83 5326 Email: <u>thudaib@lincoln.ac.uk</u> Fax: +44 1522 886974

59	Extra	cts were then suction-filtered and evaporated to dryness under vacuum in a						
60	rotary evaporator (Rotavapor $^{\mbox{\scriptsize R}}$ R-210/R-215) at the boiling point of the solvent. The							
61	dry e	xtracts were preserved at 4°C prior to analysis for amino acid [8].						
62	2.2.	Thin Layer Chromatography						
63	The a	amino acid profiles of all the extracts were determined using the Thin Layer						
64	Chro	matography technique (TLC). The following fresh solutions were prepared to						
65	run tł	ne TLC:						
66	i.	Mobile phase: Butanol: Acetic acid glacial: Water, mixed in a ratio of 12: 3: 5						
67		respectively [9]						
68	ii.	Visualization reagent: Ninhydrin reagent: 0.2g/100ml ethanol.						
69	iii.	Amino acid standards: The following Amino acid standards were dissolved in						
70		water to a concentration of $2\%$ (w/v) and used for TLC spotting: Glycine,						
71		Serine, Leucine, Cysteine, Valine, Aspartic Acid, Tryptophan, Tyrosine,						
72		Threonine, Histidine, Proline, Glutamic Acid, Cystine, Arginine, Alanine,						
73		Glutamine, Isoleucine, Asparagine, Methionine, Phenylalanine, Hydroxyproline,						
74		and Lysine.						
75	Using	capillary tubes, small spots of amino acid standards and bean extract solutions						
76	were	applied 2cm above one edge of silica coated TLC plates (Whatman silica gel 60						
77	A, 20	$\times$ 20 cm, 250 $\mu m$ thickness with fluorescent indicator) with a 1cm space						
78	betwe	een the spots. The spots were then labeled with a pencil and left to dry at room						

Taghread Hudaib \* University of Lincoln, School of Life Sciences, Brayford Pool, Lincoln, LN6 7TS, UK. Telephone number +44(0)1522 83 5326 Email: <u>thudaib@lincoln.ac.uk</u> Fax: +44 1522 886974

Sarah Brown Applied Materials Technology Ltd, Lyndon Business Park, Farrier Road, Lincoln, LN6 3RU, UK. Email: <u>sarah.brown@appmat.co.uk</u>

Daniel Wilson University of Lincoln. School of pharmacy, Brayford Pool, Lincoln, LN6 7TS Telephone number +44 (0)1522 886825 Email: <u>dwilson@lincoln.ac.uk</u>

79 temperature before being located vertically in a TLC chamber containing the developing solution (the mobile phase). The plate edge beneath the spots was 80 81 immersed in the developing solution and the TLC chamber sealed. Then, the plates 82 were left for approximately three hours. The highest point reached by the mobile 83 phase was marked. When the plates were completely dry they were sprayed with the Ninhydrin reagent and left at 80-100°C for 5min to allow the visualizing reagent to 84 85 react and stain the spots. The distances from the starting point to the spot centers were measured and the Retardation Factor ( $R_{\rm f}$ ) value for each standard and unknown 86 87 spot was calculated ( $R_{\rm f}$  = the distance traveled by the spot / the distance traveled by 88 the mobile phase front).

89 The following symbols were used to describe the results of the detected amino acids:

90 (+) observed; (++) bright; (<u>+</u>) faint [10].

# 91 **3. Results and Discussion**

The organic solvents used to produce the crude extracts varied in their ability to extract amino acids; no amino acids were detected in Chloroform fractions. Sixteen amino acids were extracted by different polar solvents. The calculated  $R_{\rm f}$  value for each standard is shown in Table 1. These  $R_{\rm f}$  values were used as markers to identify the amino acids in each extract (Table 2).

97

Sarah Brown Applied Materials Technology Ltd, Lyndon Business Park, Farrier Road, Lincoln, LN6 3RU, UK. Email: sarah.brown@apomat.co.uk

Daniel Wilson University of Lincoln. School of pharmacy, Brayford Pool, Lincoln, LN6 7TS Telephone number +44 (0)1522 886825 Email: dwilson@lincoln.ac.uk

Taghread Hudaib \* University of Lincoln, School of Life Sciences, Brayford Pool, Lincoln, LN6 7TS, UK. Telephone number +44(0)1522 83 5326 Email: <u>thudaib@lincoln.ac.uk</u> Fax: +44 1522 886974

- 98 Table1. Essential and nonessential amino acid standards; abbreviation, polarity and
- *R*<sub>f</sub> values obtained from the TLC on silica plates, eluted with n-butanol: acetic acid: 99
- water mixture 12:3:5 (by volume) respectively and visualized by Ninhydrin stain. 100

Amino acids Abbreviations		Side-chain polarity	Rf
Histidine	His	polar	0.162
Isoleucine	lle	nonpolar	0.515
Leucine	Leu	nonpolar	0.562
Lysine	Lys	polar	0.131
Methionine	Met	nonpolar	0.462
Phenylalanine	Phe	nonpolar	0.615
Threonine	Thr	polar	0.262
Tryptophan	Try	nonpolar	0.615
Valine	Val	nonpolar	0.439
Alanine	Ala	nonpolar	0.277
Arginine	Arg	polar	0.162
Asparagine	Asn	polar	0.215
Aspartic acid	Asp	polar	0.24
Cysteine	Cys	polar	0.362
L-Cystine	L-Cys	polar	0.139
Glutamic acid	Glu	polar	0.277
Glutamine	Gln	polar	0.223
Glycine	Gly	nonpolar	0.246
Proline	Pro	nonpolar	0.262
Serine	Ser	polar	0.254
Tyrosine	Tyr	polar	0.539
Hydroxyproline	Нур	polar	0.426

101

- Table2. Amino acids detected in extracts of seed cotyledon and seed coat of kidney-102
- beans and black-eyed beans identified by their *R*<sup>f</sup> values referring to the amino acid 103
- 104 standards' Rf values.

Taghread Hudaib \* University of Lincoln, School of Life Sciences, Brayford Pool, Lincoln, LN6 7TS, UK. Telephone number +44(0)1522 83 5326 Email: <u>thudaib@lincoln.ac.uk</u> Fax: +44 1522 886974 Applied Materials Technology Ltd, Lyndon Business Park, Farrier Road, Lincoln, LN6 3RU, UK. Email: sarah.brown@appmat.co.uk Sarah Brown

Daniel Wilson University of Lincoln. School of pharmacy, Brayford Pool, Lincoln, LN6 7TS Telephone number +44 (0)1522 886825 Email: <u>dwilson@lincoln.ac.uk</u>

Extracts► Amino acids▼	KBS 70% Ethanol	KBS 100% Methanol	KBS 100% Acetone	KBSC 70% Ethanol	KBSC 100% Methanol	BEBS 70% Ethanol	BEBS 100% Methanol	BEBS 100% Acetone	BEBSC 70% Ethanol	BEBSC 100% Methanol
Ala		++	+				+	<u>+</u>	<u>+</u>	+
Arg	+		+							
Asn	<u>+</u>									
Asp			<u>+</u>			+				
Cys				+					+	
Glu			+					<u>+</u>	<u>+</u>	+
Gly	+	+								
Pro			+				+			
Tyr	<u>+</u>	+	+				++	++	++	+
His	<u>+</u>	+	<u>+</u>							
lle	+	+	+			+				
Leu		++				+	<u>+</u>			
Lys	+	+	<u>+</u>			+	<u>+</u>			
Phe		+		+				<u>+</u>		
Thr	+	<u>+</u>		+			+			
Val	+	++	+			+	+	<u>+</u>	+	+

105

106

107 The results obtained from the TLC (Table 1 and Table 2) indicate that the amino acid

108 profile of the kidney-beans and the black-eyed beans varied with different amino acids

109 detected in different crude extracts. Some amino acids such as Ala, Asp, Glu, Pro, Leu

and Val were poorly or not detected in the crude extracts prepared with the 70%

aqueous ethanol. All of the essential amino acids with the exception of Met were found

in the kidney-bean extracts whilst Arg, Asn, Met, Gly and His were absent from the

113 black-eyed bean extracts. Most of the amino acids were detected in the seed

Sarah Brown Applied Materials Technology Ltd, Lyndon Business Park, Farrier Road, Lincoln, LN6 3RU, UK. Email: sarah.brown@appmat.co.uk

Daniel Wilson University of Lincoln. School of pharmacy, Brayford Pool, Lincoln, LN6 7TS Telephone number +44 (0)1522 886825 Email: <u>dwilson@lincoln.ac.uk</u>

Taghread Hudaib \* University of Lincoln, School of Life Sciences, Brayford Pool, Lincoln, LN6 7TS, UK. Telephone number +44(0)1522 83 5326 Email: <u>thudaib@lincoln.ac.uk</u> Fax: +44 1522 886974

114 cotyledon fractions, with only a few detected in the seed coat extracts. More amino acids were detected in the black-eyed bean seed coat than the kidney bean seed coat; 115 the 70% Ethanol fraction of black-eyed bean seed was found to contain Ala, Cys, Glu, 116 117 Tyr, Val and the100% Methanol fraction contains Ala, Glu, Tyr and Val. Only one 118 fraction of the kidney bean seed coat was found to have amino acids; the 70% Ethanol that found to contain Cys, Phe, Thr. Amino acids have different functions in different 119 120 legume seeds; Cys is a sulfur containing amino acid that plays an important role in plant defense mechanisms against predators and is found to be toxic to cowpea 121 122 beetles Callosobruchus maculatus [11]. This may explain why seed coats help protect 123 legumes against stored seed pests and minimize losses of stored products. 124 Moreover, both types of beans seem to lack the following amino acids: Hyp, Ser, Gln, 125 L-Cys, Try, and Met which reduces the nutritional value of these beans. Combining 126 beans with other legumes or grains is recommended to provide a meal that is rich in all essential amino acids. 127

# 128 **4.** Conclusion

TLC is an efficient method for the analysis of amino acids from plant materials.
Amino acids vary in their polarity according to their side chains. Producing several
crude extracts is a vital first step for the further separation steps. Using several
solvents, with different polarity in crude extracts preparation improves the efficacy of

Sarah Brown Applied Materials Technology Ltd, Lyndon Business Park, Farrier Road, Lincoln, LN6 3RU, UK. Email: sarah brown@appmat.co.uk

Daniel Wilson University of Lincoln. School of pharmacy, Brayford Pool, Lincoln, LN6 7TS Telephone number +44 (0)1522 886825 Email: dwilson@lincoln.ac.uk

Taghread Hudaib \* University of Lincoln, School of Life Sciences, Brayford Pool, Lincoln, LN6 7TS, UK. Telephone number +44(0)1522 83 5326 Email: <u>thudaib@lincoln.ac.uk</u> Fax: +44 1522 886974

the TLC technique, enhances the separation and simplifies the identification of polarand non-polar amino acids.

### 135 **5. References:**

- 136 [1] K. J. Jalkanen, M. Elstner, and S. Suhai, J. Molecular Structure: THEOCHEM,
- 137 **675** (2004) 61-77.
- 138 [2] A. Ambrogelly, S. Palioura, D. Söll, Nature Chemical Biology 3 (2007) 29-35.
- [3] *R. Wolfenden*, *L. Andersson*, *P. M. Cullis*, and *C. C. B. Southgate*, Biochemistry,
  20 (1981) 849-855.
- 141 [4] *B. Fried*, and *J. Sherma*, CRC Press, (1996).
- 142 [5] *J.H. Doughari*, Agricultural and Biological Sciences edition, In Tech, (2012)1-32.
- 143 [6] G. Jayaprakasha, R. Singh, and K. Sakariah, Food Chemistry, 73 (2001) 285-
- 144 **290**.
- 145 [7] S.A. Audu, I. Mohammed, and H.A. Kaita, Life Science Journal, 4 (2007) 75-79.
- 146 [8] V. A. Maikai, B. V. Maikai, and P. I. Kobo, Journal of parasitology research,
- 147 (2014) 904318-904318.
- 148 [9] G. S. Chakraborthy, IJRAP, 1 (2010) 131-134.
- 149 [10] M. Waksmundzka-Hajnos, J. Sherma, and T. Kowalska, CRC Press, (2008).

Sarah Brown Applied Materials Technology Ltd, Lyndon Business Park, Farrier Road, Lincoln, LN6 3RU, UK. Email: sarah.brown@appmat.co.uk

Daniel Wilson University of Lincoln. School of pharmacy, Brayford Pool, Lincoln, LN6 7TS Telephone number +44 (0)1522 886825 Email: dwilson@lincoln.ac.uk

Taghread Hudaib \* University of Lincoln, School of Life Sciences, Brayford Pool, Lincoln, LN6 7TS, UK. Telephone number +44(0)1522 83 5326 Email: <u>thudaib@lincoln.ac.uk</u> Fax: +44 1522 886974

150	[11] <i>D.H.</i>	Janzen, H.B.	Juster, ar	nd <i>E. Art</i>	<i>hur Bell</i> , F	Phytochemist	ry, <b>16</b>	(1977)	223-
-----	------------------	--------------	------------	------------------	---------------------	--------------	---------------	--------	------

- 151 227.
- 152
- 153
- 154
- . -
- 155

156

Taghread Hudaib \* University of Lincoln, School of Life Sciences, Brayford Pool, Lincoln, LN6 7TS, UK. Telephone number +44(0)1522 83 5326 Email: <u>thudaib@lincoln.ac.uk</u> Fax: +44 1522 886974

Sarah Brown Applied Materials Technology Ltd, Lyndon Business Park, Farrier Road, Lincoln, LN6 3RU, UK. Email: <u>sarah.brown@appmat.co.uk</u>

Daniel Wilson University of Lincoln. School of pharmacy, Brayford Pool, Lincoln, LN6 7TS Telephone number +44 (0)1522 886825 Email: <u>dwilson@lincoln.ac.uk</u>