CORE

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## Structure Reports

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## Key indicators

Single-crystal X-ray study
$T=93 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.001 \AA$
$R$ factor $=0.046$
$w R$ factor $=0.142$
Data-to-parameter ratio $=21.6$
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

[^0]
## 1-(Methoxymethyl)pyrene

The title compound, $\mathrm{C}_{18} \mathrm{H}_{14} \mathrm{O}$, crystallizes with aromatic $\pi-\pi$ stacking interactions.

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## Comment

With their particular electronic, optical and geometric properties, pyrene and its derivatives (Garcia-Cruz et al., 2004; Takahashi et al., 2004; Borisevich et al., 1992) have attracted considerable recent interest. This is demonstrated in a great number of applications including host-guest (Vögtle, 1996) and metal coordination chemistry (Arrais et al., 2004), as well as fluorescent sensor development (Bren, 2001). Although pyrenes with a functional side arm are important building blocks in this respect, reports on corresponding crystal structures are limited (Foss \& Stevens, 1985; Olszak et al., 1989). Here, we describe the structure of 1-(methoxymethyl)pyrene, (I), a new example of this type of compound.

(I)

The pyrene part of the molecule shows no significant deviations of bond lengths and angles compared with those of the unsubstituted analogue (Camerman \& Trotter, 1965; Allmann, 1970; Hazell et al., 1972; Kai et al., 1978), and is almost planar. The largest deviation from the mean plane through the carbon framework of the pyrene unit is 0.042 (1) $\AA$ for atom C 1 . The torsion angle $\mathrm{C} 2-\mathrm{C} 1-\mathrm{C} 17-\mathrm{O} 1$ is $15.1(1)^{\circ}$, indicating that the methoxymethyl group is turned away from the aromatic plane, while the $\mathrm{C}-\mathrm{C}-\mathrm{O}-\mathrm{C}$ fragment itself exhibits a nearly ideal anti-periplanar conformation [179.7 (1) ${ }^{\circ}$ ].

Due to the aromatic nature of the molecule, the crystal packing of (I) is dominated by arene-arene supramolecular contacts. In contrast to the unsubstituted pyrene, which forms face-to-face dimers packed in a herringbone fashion, the crystal structure of (I) is characterized by molecular stacks which are stabilized by offset face-to-face interactions. Within the stacks, the mean distance between consecutive molecules is $3.5 \AA$. The molecules of neighboring stacks are inclined to each other at an angle of 86.47 (3) $)^{\circ}$. Hence the compound has an overall packing structure resembling the pattern found in
the solid state structures of many aromatic compounds (Desiraju, 1989).

## Experimental

The title compound, (I), was synthesized from commercially available pyrene-1-carbaldehyde, which was initially reduced with sodium borohydride in boiling methanol, following an analogous procedure described for the reduction of anthracene-9-carbaldehyde (Steward, 1960), to yield the intermediate compound 1-(hydroxymethyl)pyrene. This was transformed into 1-(bromomethyl)pyrene by the usual bromination with phosphorus tribromide in chloroform (Okamoto et al., 1990). Subsequent treatment with boiling methanol and recrystallization of the product from the same solvent yielded $60 \%$ of compound (I) as colorless needles (m.p. $322-323 \mathrm{~K}$ ). ${ }^{1} \mathrm{H}$ NMR spectroscopic data ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 8.32(d, \mathrm{ArH}, 1 \mathrm{H}), 8.13(m$, $\mathrm{ArH}, 4 \mathrm{H}), 8.02(m, \mathrm{ArH}, 4 \mathrm{H}), 5.14\left(s, \mathrm{CH}_{2} \mathrm{OH}, 2 \mathrm{H}\right), 3.49\left(s, \mathrm{CH}_{3}, 3 \mathrm{H}\right)$.

## Crystal data

$$
\begin{aligned}
& \mathrm{C}_{18} \mathrm{H}_{14} \mathrm{O} \\
& M_{r}=246.29 \\
& \text { Monoclinic, } P 2_{1} / n \\
& a=4.7220(10) \AA \\
& b=20.087(4) \AA \\
& c=12.824(3) \AA \\
& \beta=91.13(3)^{\circ} \\
& V=1216.1(5) \AA^{3}
\end{aligned}
$$

$$
\begin{aligned}
& Z=4 \\
& D_{x}=1.345 \mathrm{Mg} \mathrm{~m}^{-3} \\
& \text { Mo K } \text { radiation } \\
& \mu=0.08 \mathrm{~mm}^{-1} \\
& T=93(2) \mathrm{K} \\
& \text { Prism, colorless } \\
& 0.41 \times 0.20 \times 0.17 \mathrm{~mm}
\end{aligned}
$$

## Data collection

Bruker SMART CCD area-detector diffractometer
$\varphi$ and $\omega$ scans
Absorption correction: multi-scan (SADABS; Sheldrick, 2000) $T_{\text {min }}=0.906, T_{\text {max }}=0.986$

## Refinement

Refinement on $F^{2}$

$$
\begin{aligned}
& w=1 /[ \sigma^{2}\left(F_{\mathrm{o}}{ }^{2}\right)+(0.0988 P)^{2} \\
&+0.1716 P] \\
& \text { where } P=\left(F_{\mathrm{o}}^{2}+2 F_{\mathrm{c}}^{2}\right) / 3 \\
&(\Delta / \sigma)_{\max }=0.001 \\
& \Delta \rho_{\max }=0.48 \mathrm{e} \AA^{-3} \\
& \Delta \rho_{\min }=-0.20 \mathrm{e}^{-3}
\end{aligned}
$$



Figure 1
Perspective view of (I), showing $50 \%$ probability displacement ellipsoids for the non-H atoms.


Figure 2
Packing diagram of (I), viewed down the $c$ axis. $H$ atoms have been omitted.

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