

Cite this: DOI: 10.1039/c0xx00000x

www.rsc.org/xxxxxx

ARTICLE TYPE

High-resolution isolation of protein corona nanoparticles from complex physiological fluids

Desirè Di Silvio,^{a,b} Neil Rigby,^b Balazs Bajka^b, Andrew Mayes,^c Alan Mackie^b and Francesca Baldelli Bombelli^{*a,d}

⁵ Received (in XXX, XXX) Xth XXXXXXXXX 20XX, Accepted Xth XXXXXXXXX 20XX

DOI: 10.1039/b000000x

Nanoparticles (NPs) in contact with biological fluids are generally coated with environmental proteins, forming a stronger layer of proteins around the NP surface called the hard corona. Protein corona complexes provide the biological identity of the NPs and their isolation and characterization are essential to understand their *in vitro* and *in vivo* behaviour. Here we present a one-step methodology to recover NPs from complex biological media in a stable non-aggregated form without affecting the structure or composition of the corona. This method allows NPs to be separated from complex fluids containing biological particulates and in a form suitable for use in further experiments.

The study has been performed systematically comparing the new proposed methodology to standard approaches for a wide panel of NPs. NPs were first incubated in the biological fluid and successively recovered by sucrose gradient ultracentrifugation in order to separate NPs and their protein corona from the loosely bound proteins. The isolated NP-protein complexes were characterized by size and protein composition through Dynamic Light Scattering, Nanoparticle Tracking Analysis, SDS-PAGE and LC-MS. The protocol described is versatile and can be applied to diverse nanomaterials and complex fluids. It is shown to have higher resolution in separating the multiple protein corona complexes from a biological environment with a much lower impact on their *in situ* structure compared to conventional centrifugal approaches.

Introduction

In last decades, nanoparticles (NPs) have found several applications in a wide range of fields, such as medicine,^{1, 2} cosmetics,³ paints,⁴ high-tech,⁵⁻⁷ and food industries⁸ and by 2020 nanotechnology is forecast to produce about 60000 tons of NPs per year.⁹ The wide use of NPs in several areas has exponentially increased their exposure to humans, both intentionally and unintentionally. In order to understand the potential impact of engineered nanomaterials (ENMs) on human health, it is fundamental to fully characterize them in relevant biological fluids.¹⁰ In fact, when in contact with biological fluids, NPs spontaneously interact and adsorb proteins on their surface forming what is known as protein corona (PC).^{11, 12} This corona provides the biological identity of the NP and determines its interactions with the surrounding biological matter.^{13, 14}

The PC has a dynamic structure formed by external layers of proteins that quickly exchange with the surrounding environment (soft corona) and an inner layer of proteins strongly bound to the NP surface (hard corona).¹⁵⁻¹⁸ The PC composition depends on NP physical-chemical properties and size, but also media composition and incubation time has been found to have strong effects.^{19, 20, 21} Generally, the hard corona is very stable and almost irreversibly bound to the NP surface when the experimental conditions are kept constant,¹² although it has recently been shown that small alterations in the composition of the biological media can strongly modify the hard corona.²² Moreover, preliminary studies on the evolution of the PC, where

NPs were sequentially incubated in different biological fluids, have shown that, even if changes occur in the PC upon incubation in the second biological fluid, a sort of fingerprint of the “history” of the NP is kept.²³ Thus, isolating HC complexes is crucial to independently study their composition and to be able to relate possible biological responses to it.

One of the biggest concerns in this scenario is to isolate HC complexes that preserve the features of those *in situ* in the biological fluid. Commonly, an *ex situ* approach is used to separate protein-NP complexes from the excess of fluid and isolate HC (HC) complexes: sequential cycles of centrifugation/washing are carefully optimized according to NP and media properties.^{24, 25} In many cases this approach is suitable and gives reliable results, but multiple purification steps can alter the equilibrium of the system and lead to corona modifications due to the time-scales that characterize the dynamic nature²⁶ of the complexes. Application of this approach to NPs characterized by small diameters (about 5-20 nm) and/or low densities (close to 1 g/cm³) may not achieve a good separation between unbound proteins and PC complexes. High speeds and long times are often necessary with promotion of extensive aggregation with respect to what occurs *in situ*. These effects are even more accentuated in the isolation of HC complexes for NPs that do not form rich coronas (for example pegylated NPs). For this reason it becomes important to develop methodologies that minimize the number of steps of the *ex-situ* purification of HC complexes to affect as less as possible their properties from *in situ*.

Furthermore, PC complexes are often very heterogeneous

presenting simultaneously monomers, dimers, trimers, etc. for which the actual composition is unknown. In this regard, the biological impact of these co-existing complexes formed by the same NP might be different as they have different sizes and are likely to carry different proteins. Thus, it is important to be able to separate those different complexes and independently study their effect on the biological matter. Recently, the use of differential sedimentation centrifugation (DCS) permitted the analytical separation of different populations of HC complexes for different NPs and demonstrated that they were representative of those *in situ*. However, this technique does not allow the recovery of the different populations for further studies.^{12, 27} In particular, their recovery becomes important to determine the physical and biological properties of these complexes. Conventional procedures allow the isolation and recovery of mixed PC populations, which are also aggregated with respect to their *in situ* cognate PC complexes.

Many techniques have been applied to separate and study PC complexes such as size-exclusion chromatography (SEC), magnetic separation through the use of magnetic columns MACS able to separate PC complexes of magnetic NPs,²⁸ and field-flow-fractionation.^{29, 30} However, none of these methodologies are preparative and permit full recovery of the HC complexes for further studies. Preparative and analytical ultracentrifugation (UC) are widely exploited in biology to isolate cell components and explore protein thermodynamics.³¹ Density gradient UC has been extended to purify NPs from the excess of coating agents,³² and obtain narrow size distributions.^{33, 34} Recently, Docter, Tenzer and co-workers^{35, 36} and Werwie and co-workers³⁷ used a sucrose cushion as first step to separate unbound proteins from corona complexes followed by centrifugal washing to obtain HC complexes.

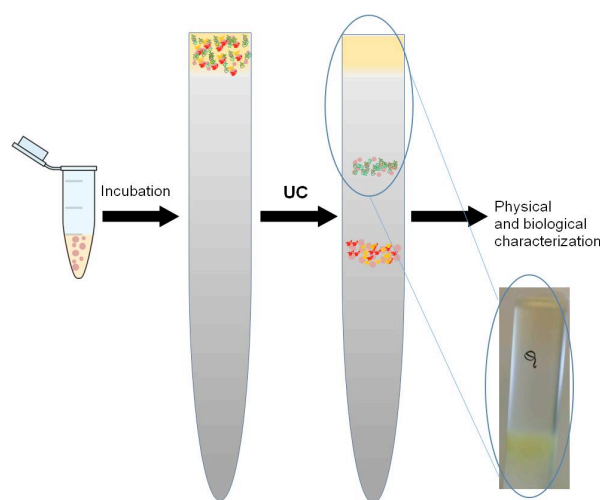
Here we propose the use of sucrose gradient UC as a one-step methodology to gently separate HC complexes from *in situ* complex physiological fluids such as bovine serum and digestive fluids. We have shown that this procedure has a much lower impact on the structure of the complexes and a much higher resolution in separating different complexes with respect to conventional protocols. In fact, this methodology permitted full recovery of HC complexes *ex-situ* minimizing modifications with respect to those *in situ* and allowing separation of the different populations co-existing *in situ* (see Scheme 1).

NPs of different size, material and surface coating were tested in serum and in simulated digestive fluids. The PC complexes isolated by conventional centrifugation methods and UC were analysed by size through Nanoparticle Tracking Analysis (NTA) and Dynamic Light Scattering (DLS). PC composition was investigated by SDS-PAGE and for PC complexes obtained from digestive fluids, Mass Spec analysis was also performed to confirm the identities of the recovered proteins. Moreover, to prove how isolation procedure influenced the interaction of PC complexes with biological matter, PC complexes of magnetite NPs recovered from serum by both UC and conventional centrifugation methods were incubated on cells in serum free conditions to compare their NP cellular uptake.

Experimental section

Material

Yellow-green carboxylate-modified polystyrene NPs of 100 nm and 20 nm nominal diameter were purchased from Invitrogen (PS-COOH100, PS-COOH20). PS-COOH100 NPs were pegylated through EDAC chemistry using Jeffamine M1000 (Huntsman) and purified by centrifugal washings (3x30 minutes at 17000 rcf, 20°C) (PS-PEG). Bare 50 nm silica NPs were purchased from Kisker (SiO₂). Carboxylated Fe₃O₄ NPs were synthesized following the protocol of Sun et al.³⁸ and coated by poly(maleic)-alt-1-octadecene (Sigma) according to Lin et al.³⁹ obtaining NPs of about 50 nm of hydrodynamic diameter. For cell uptake studies Fe₃O₄ NPs were fluorescently labelled with BODIPY FLEDA (Lifetechnologies) that was attached to the NP surface by EDAC chemistry. Bovine Serum Hyclone (FBS) was purchased from Fisher. Sucrose, sodium phosphate dibasic, potassium phosphate monobasic, sodium chloride and potassium chloride are from SIGMA.



Scheme 1 Scheme of the methodology to isolate and study HC NPs. NPs are incubated in biological fluid and then subjected to ultracentrifugation (UC). In the image of the vial on the far left pink dots are NPs and the yellow background is the biological medium. In the images of the UC vials (middle images) green, red and yellow dots represent some proteins that form diverse coronas around the NPs and are separated by UC.

Preparation of protein corona samples

NPs in PBS were incubated in 10%, 55% and 90% v/v FBS for one hour at 37°C. NPs dispersed in FBS solution are called *in situ*, HC NPs isolated through ultracentrifugation are labelled as UC; HC NPs isolated by conventional methodology (three centrifugations at 15500 rcf at 4°C and re-suspensions in 500µl PBS pH 7.4) are labelled as HC. Samples were characterized by NTA and DLS before and after incubation.

Simulated Salivary, Gastric and Duodenal Digestions

An AT-700 pH Stat Kyoto Electronic Manufacturing Company was used to run simulated digestions. Digestion mixture compositions, time points at which samples were taken and timescales for digestions varied. Buffer solutions were used in all digestions but whilst desirable pH was set at the beginning of each phase of digestion, it was not actively controlled throughout (instead monitored and recorded during the sample taking). The chosen food material was skimmed milk powder (SMP, 34 mg/ml) and individual enzymes in the different digestion phases

were used. Pepsin (from porcine gastric mucosa, Sigma-Aldrich, Lot. 091M7020V) was used in the gastric phase, trypsin (from porcine pancreas, Sigma-Aldrich, Lot. 045K7775) and chymotrypsin (from bovine pancreas, Sigma-Aldrich, Lot. 060M7007V) for the duodenal phase, as well as bile. Fe₃O₄ NPs at a final concentration of 1.5x10¹³ NPs/ml were incubated with 1ml of fluids collected at time points corresponding to the oral phase (2 min, pH 7, amylase), before the start of gastric phase (60 min), after gastric phase was finished (120 min, pH 3) and after duodenal digestion (240 min, pH 7.0). The NPs were incubated at 37°C for one hour in Brunswick Scientific Excella E24 Incubator Shaker. Equal volumes of purified water were incubated with the fluids to act as controls.

Ultracentrifugation

Solutions of sucrose were prepared at increasing concentration and 11 ml of linear sucrose gradient were layered in 13 ml tubes and left to equilibrate overnight before being subjected to ultracentrifugation using a SW41 Ti rotor (Beckman Coulter) at 20°C. 0.7 ml of samples was loaded and different protocols were used according to the physical properties of the NP and the biological media (details of the gradient are given in Table 1). After the run, aliquots of 1 ml were collected sucking up sucrose from the top of tubes to the bottom and analysed to identify protein corona NPs either screening by dimension or UV absorption at 260 nm.

Table 1 Ultracentrifugation experimental conditions used to separate the NPs from the biological fluids. Particles were in PBS pH 7.4.

Type of NP	Nominal size [nm]	NPs/ml ^a	Sucrose density Δ [% w/w]	Speed [rcf]	Time [min]
PS-COOH100	100	3.6x10 ¹¹	5-30	77k	60
PS-COOH20	20	4.5x10 ¹³	4-40	110k	120
PS-PEG	100	3.6x10 ¹¹	3-30	60k	60
SiO ₂	50	1.9x10 ¹³	3-30	12k	20
Fe ₃ O ₄	50	1.5x10 ¹³	35-60/70	187k	120

^a Concentration of the NPs in the *in situ* samples before being loaded in the ultracentrifuge.

Dynamic Light scattering (DLS)

Hydrodynamic diameters were measured by Zetasizer SZ (Malvern). 50µl of the samples were diluted in 400µl of PBS in 1ml cuvette to obtain attenuation values in the 7-9 range and measured at 25°C equilibrating samples for 120 seconds prior to measurement. Data were presented as an average of three measurements. UC samples were dialysed against PBS (2000 MWCO, Spectrum labs) for at least 24 hours at 4°C before analysis. The Z-averaged sizes (Z-ave) and the polydispersity index (PdI) were obtained by cumulant analysis of the auto-correlation function.

Nanoparticle Tracking Analysis (NTA)

Samples were analysed by NTA diluting them with MQ water to reach an ideal concentration of 1-3x10⁸ NPs/ml. For fluorescently-labelled particles a fluorescent filter (λ of 488 nm) was used. Three videos of sixty seconds were collected for each sample and analysed by NTA software. The software is able to

track NPs individually and calculates the diffusion coefficient for each one. In this way, a mean value for the hydrodynamic diameter (Mean) is obtained with the relative standard deviation (SD) respect to all tracks. Results are presented as an average of three independent measurements.

SDS-PAGE electrophoresis

Dialysed samples were concentrated by Amicon centrifugal filters (MWCO 100kDa, Millipore) at 405 rcf to a final volume of 200 µl. 30 µl were added to 15 µl of SDS-PAGE loading buffer 3x (10% DTT, Thermo Scientific) and kept at 98°C for 5 minutes. Digested samples that did not contain NPs were mixed to the loading buffer without any previous treatment and denatured as above. HC samples were prepared re-suspending the pellet in 60 µl of PBS to which 30 µl of loading buffer was added before denaturation. 20 µl of samples and 5 µl of molecular ladder (PAGERuler Broad Range, Biolabs) were loaded in the wells of 12% Precast Gel NuPAGE (Life Technology). Samples from simulated digestion were loaded on 10% Precast Gel NuPAGE (Life Technology) and the molecular ladder used was Mark12 Unstained standard from Invitrogen (5 µl). The running buffer used was MES buffer (NuPAGE 20x, Invitrogen). Gels were run at 200V for 35 minutes. Gels were developed by Sypro Ruby Protein Stain (Biorad) and imaged by Biorad Pharos FX+, the software used to elaborate images was Image Lab (Biorad).

LC-MS

ProPick instrument was used to locate and cut bands from the gel. Bands were digested at 37°C for three hours by 10mM Ammonium Bicarbonate 10 ng/µl Trypsin Gold (Promega, V528A in 50mM Acetic Acid) (prepared 01-May-14). 1% formic acid (Sigma) was added prior freezing samples and storing at -80°C. Samples were washed in 50% acetonitrile (Fisher), vortexed and dried out at the Low Drying setting (no heat) on a Speed Vac SC110 (Savant) fitted with a Refrigerated Condensation Trap and a Vac V-500 (Buchi). Samples were again stored at -80°C until ready for Orbitrap analysis. Protein identification was achieved by combining spectrum quality scoring obtained from a conventional database search program MASCOT (Matrix Science, London, England). Search parameters were: peptide mass and fragment mass tolerances of 5ppm and 0.5 Da respectively, variable modification was oxidation (M), fixed modification carbamidomethyl (C), enzyme specificity was trypsin, two missed cleavage was allowed. All taxonomy database was searched.

Cell Uptake experiments.

M2O2 cell line were incubated at 37°C, 5% CO₂ / 95% air and were cultured in RPMI-1640 (Life Technologies, Carlsbad, CA) supplemented with 10% heat-inactivated foetal bovine serum (FBS), 2 mM L-glutamine, 100 µg/ml penicillin and 100 µg/ml streptomycin all purchased Life Technologies, Carlsbad, CA. At P12 cells were seeded at a density of 4x10⁴ cells/ml on glass coverslips and left to adhere overnight. Cells were incubated with Fe₃O₄ NPs (10¹² NPs/ml) for 2, 4 and 24 hours. In particular cells were treated with fluorescently labelled Fe₃O₄ NPs bare, HC,

UC1 and UC2. Cells were washed with PBS (x3) and fixed (5% formaldehyde, 2% sucrose, 0.02% w/V sodium azide in PBS) then permeabilized (5 min with 0.1% Triton X-100 in PBS). Texas Red®-X Phalloidin (Life Technologies) (6.6 μM) was employed to stain actin filaments and Hoechst 33342 dye to stain nuclei (2 μg/ml, Sigma). The coverslip was mounted face down on a slide for microscope with 12 μl of mounting media (Vectashield, Vector), left to dry overnight and then sealed with nail varnish. Zeiss LSM510-META confocal microscope (UEA) was used to image cell in multi-track mode. ImagePro software was used to elaborate the z-stack recorded. ImagePro software was used to process images.

Results and Discussion

Isolation of HC (HC) complexes from FBS.

NPs of different size, material and surface coating (see Table 1) were incubated at 37°C in FBS and separated by the excess of proteins by UC on sucrose gradient. The recovered HC complexes were further characterized and compared with those *in situ* and those isolated by conventional centrifugation protocols. We chose five different types of NPs that spanned a variety of materials and presented different PC complexes to prove the suitability of this protocol to successfully separate HC complexes of diverse properties. In particular, PS-COOH100 NPs were chosen as control sample for comparing PC complexes obtained with our novel procedure with those obtained using the conventional approach,¹² and PS-COOH20 NPs were chosen as an example of NP-protein complexes that cannot be isolated by normal centrifugation, which promoted extensive aggregation and loss of material in the recovery of HC complexes.¹⁹ PS-PEG NPs were chosen as an example of NPs with a reduced tendency to form a PC in biological environments and thus very difficult to isolate and recover from the biological fluid,⁴⁰⁻⁴² while SiO₂ NPs are NPs able to form two populations of HC complexes that could be successfully separated and recovered by UC.²⁷ As last, Fe₃O₄ NPs are an example of engineered NPs designed for biomedical applications, for which the conventional approaches promoted extensive aggregation.⁴³ For all the samples the excess of proteins from the media were enriched in the lower density sucrose layer at the top of the tube as shown in Scheme 1.

Table 2 DLS and NTA characterization of PS-COOH100 NPs in PBS, 90% FBS and of HC complexes isolated by centrifugation (HC) and ultracentrifugation (UC 1-2).

	DLS		NTA		
	D _h ^b [nm]	PdI ^b	Mean[nm] ^c	SD[nm] ^c	NPs/ml
PS-COOH ^a	100.8±0.7	0.01	78±3	29±2	7x10 ¹²
PS-COOH <i>in situ</i>	128.1±0.9	0.03	130±1	31±1	3x10 ¹²
PS-COOH HC	168.0±0.9	0.26	154±3	52±7	7x10 ¹²
PS-COOH UC1	122.2±1.7	0.01	117±1	34±1	5x10 ¹²
PS-COOH UC2	131.4±1.6	0.01	126±2	35±1	1x10 ¹²

^a in PBS pH 7.4. ^b Hydrodynamic diameters (D_h) and polydispersity indexes (PdI) obtained by cumulant analysis of the autocorrelation functions. ^c Mean is the averaged size of all tracked particles and SD is the related Standard Deviation. All the data are presented as the average of three independent measurements.

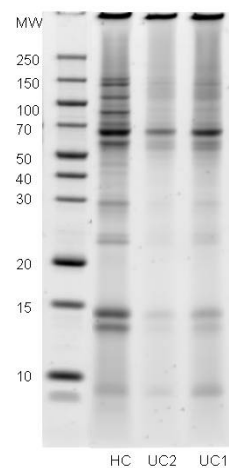


Fig. 1 HC complexes of PS-COOH100. SDS-PAGE gel of the HC complexes of PS-COOH100 NPs in 90% FBS isolated by conventional centrifugation methods (hard corona, HC) and sucrose ultracentrifugation (UC) respectively, as indicated by the label below the tracks of the gel.

The validation of the UC procedure for isolating HC complexes was first performed by studying 100 nm PSCOOH NPs dispersed in 90% FBS, whose PC complexes have already been extensively characterized in the literature.^{19, 44} PS-COOH100 *in situ* was shown to form a rather monodisperse PC complexes with hydrodynamic diameter of about 130 nm (see Table 2). The isolation of such complexes by normal centrifugation promoted limited aggregation leading to the formation of HC NPs with hydrodynamic diameters of about 170 nm and characterized by higher PdI. The same sample subjected to UC produced three separated bands at different concentration in the sucrose gradient, two of which contained similar complexes and were pooled together. This indicated that with this procedure it was possible to separate in one-step process HC complexes with a structure similar to those *in situ* and a high resolution in separation by size. In fact, complexes differing by only of 7% in size (122 nm and 131 nm) were isolated in two well-separated sucrose bands (see Figure 1 UC1 and UC2). The strength of this methodology resides also in the full recovery of the material that could be further analysed to determine the composition of the PC by SDS-PAGE (see Fig. 1) and size (Table 2).

A more challenging task was to recover HC complexes of 20 nm PSCOOH NPs, for which the conventional centrifugation methodology did not work effectively as it promoted extensive aggregation and loss of material as indicated by the hydrodynamic diameter of about 250 nm with PdI over 0.5 (see Table S3 and Fig. S1 in SI). In fact, the incubation in 90% FBS caused the formation of aggregates with an averaged hydrodynamic diameter of 65 nm compared with 30 nm of the bare NPs in PBS, and long times and high speed necessary to sediment the HC complexes clearly amplified this effect. The use of UC allowed us not only to avoid this aggregation, but also to separate the different populations of complexes. In particular, the presence of a small population of large aggregates of about 240 nm was highlighted. This population was mainly formed by proteins, as indicated by the low density of the sucrose layer (UC1) where they accumulated compared to that of UC2 and UC3, containing most of the complexes. This was also confirmed by the concentrations revealed by NTA analysis for NPs in the

UC2 and UC3 sucrose layers, which was comparable to that of the NPs *in situ*. Moreover, the hydrodynamic sizes of the complexes recovered from UC2 and UC3 layers were comparable to those of the *in situ* samples and also characterised by a lower PDI related to the separation from the larger protein aggregates as shown in Fig. 2a. The lower sizes detected by NTA for the HC and UC1 samples confirmed the presence of big protein

aggregates in these samples, whose lower sizes were likely due to a partial disaggregation by dilution (required for measuring NTA). SDS-PAGE analysis, reported in Fig. 2b, showed that fractions UC2 and UC3 were very similar in protein composition, while UC1 was enriched with proteins of Mw=60-70 kDa and 150 kDa (probably BSA and IgG).

Table 3 DLS and NTA characterization of PS-COOH20 NPs in PBS and 90% FBS.

	DLS			NTA		
	D_H^b [nm]	PDI ^b	Mean ^c [nm]	SD ^c [nm]	NPs/ml	
PS-COOH ^b	32.5±0.1	0.10	64±21	28±5	3×10 ¹²	
PS-COOH <i>in situ</i>	67.6±0.7	0.31	81±3	38±6	8×10 ¹²	
PS-COOH HC	247.3±7.6	0.65	186±18	66±6	6×10 ¹¹	
PS-COOH UC1	259.5±4.6	0.43	86±5	59±4	7×10 ¹¹	
PS-COOH UC2	68.2±1.9	0.16	58±1	26±1	1×10 ¹³	
PS-COOH UC3	91.1±1.9	0.18	73±3	35±3	9×10 ¹²	

^a in PBS pH 7.4. ^b Hydrodynamic diameter (D_H) and polydispersity index (PDI) obtained by cumulant analysis of the autocorrelation functions. ^c Mean is the averaged size of all tracked particles and SD is the related Standard Deviation. All the data are presented as the average of three independent measurements.

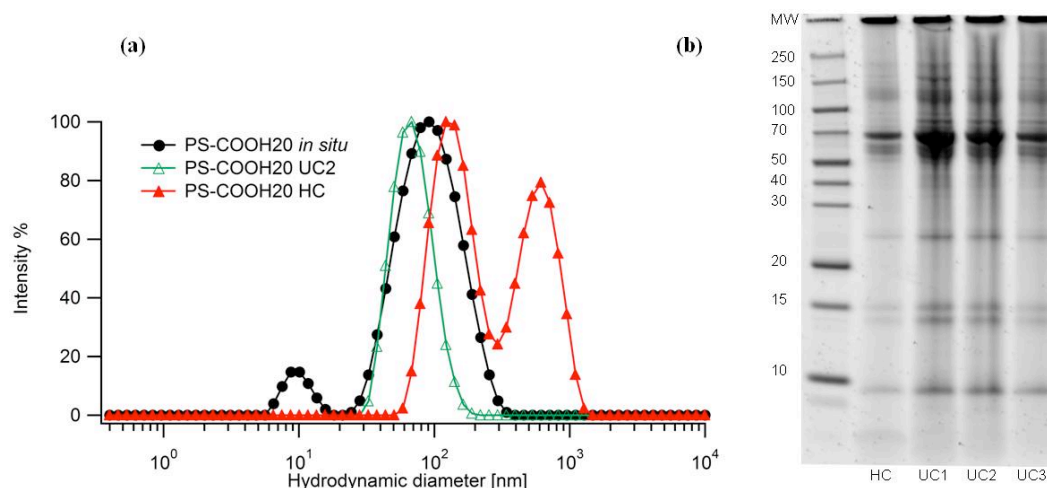


Fig. 2 HC complexes of PS-COOH20 in 90% FBS. a) Size distribution by intensity percentage of NPs *in situ* (black dots), HC complexes isolated by centrifugation (HC, red triangles) and ultracentrifugation (UC2, green empty triangles). b) SDS-PAGE gel of the complexes in 90% FBS isolated by conventional centrifugation methods (hard corona, HC) and ultracentrifuge (UC1-2), respectively, as indicated by the label below the tracks of the gel.

Another case where conventional centrifugation methods have not been very effective in isolating HC complexes is for NPs that do not show a high tendency to adsorb proteins, *e. g.* pegylated NPs. The difficulty in isolation of hard corona becomes even higher when these NPs have low density such as polymeric NPs and liposomes. Ultracentrifugation was used to separate liposomes by size⁴⁵ and recently Pozzi and co-workers⁴⁶ showed that liposome pegylation could be exploited both to reduce protein adsorption to limit macrophages uptake and to enrich the residual corona with apolipoproteins that bind specifically some receptors of prostate cancer cells. Pegylated polystyrene NPs (100 nm size) were incubated in 90% FBS and characterized by DLS and NTA (Table 4). Their dynamic properties did not change significantly with respect to those in PBS indicating a weak interaction with the environmental proteins and no significant changes in the structure of the bare NPs, although protein adsorption could not be completely ruled out. HC complexes isolated by centrifugation were extensively aggregated with respect to those *in situ* and some loss of material occurred

(see Fig. S1 in SI). The same sample was separated by UC and a well-defined single band enriched in NPs was isolated. The recovered NPs were analysed by DLS and NTA and results were very similar to those *in situ*. DLS, unlike NTA, showed an increase in diameter and PDI with respect to *in situ* sample (Fig. 3a) and a slight loss of NPs likely due to the very similar density to the proteins. The corona composition of the complexes isolated by the two methods was compared by SDS-PAGE (see Fig. 3b) and differences could be seen especially for high molecular weight proteins indicating that aggregation and loss of material can ultimately affect the properties of the hard corona of the *in situ*. Thus, UC was shown to be a promising method for recovering HC complexes of pegylated nanomaterials and allows their further characterization and biological response.

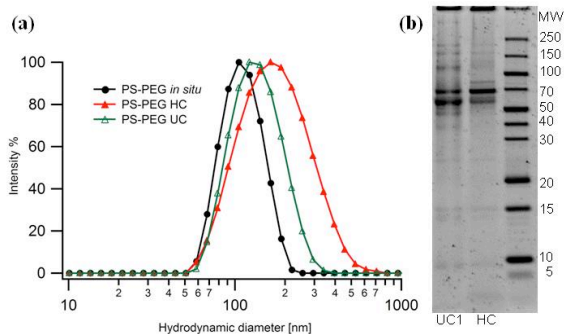


Fig. 3 HC complexes of PS-PEG. **a)** Size distribution by intensity percentage of PS-PEG NPs *in situ* 90% FBS (black dots), HC complexes isolated by centrifugation (HC, red triangles) and ultracentrifugation (UC, green empty triangles). **b)** SDS-PAGE gel of the HC complexes of PS-PEG NPs in 90% FBS isolated by ultracentrifugation (UC) and conventional centrifugation methods (HC), respectively, as indicated by the label below the tracks of the gel.

complexes isolated by UC had a structure more representative of that *in situ*.

The possibility of isolating those complexes without altering their physical properties is clearly important for studying their biological response in more detail. Fluorescently labelled water soluble oleic acid coated Fe_3O_4 NPs (see Fig.S2-S4 and Table S1 in SI) coated by PMAO polymer were incubated in 55% FBS. Larger NP-protein complexes were observed *in situ* by DLS together with the protein background (peak at 10-15 nm) as shown by the size distribution reported in Fig. 5a. HC samples showed extensive aggregation although and the pellet obtained by centrifugation could not be completely re-suspended. In contrast, UC samples were successfully isolated and recovered in two NP-protein fractions containing structures of about 77 nm and 140 nm, respectively and that seem to correspond to the species present in the *in situ* sample. The protein patterns were analysed by SDS-PAGE and qualitatively they seemed to be very similar indicating that in this case NP-protein complexes of different diameters (monomer and dimers) were enriched with similar proteins.

HC complexes recovered by centrifugation and UC were then incubated at a comparable concentration, determined by NTA and fluorescence, in serum-free conditions with M202 cells

Fe_3O_4 NPs were covalently labelled with a fluorescent dye and cells were stained for actin filaments and nuclei. In Figure 6 representative images of cells incubated for two hours with PC samples in serum free conditions are reported. A different fluorescent pattern between PC NPs isolated by conventional centrifugation methods (Figure 6a) and those obtained by UC (Fig. 6b and 6c) can be observed. In fact, the images show that samples treated with HC NPs are characterized by large fluorescent aggregates hardly taken up by the cells, while those treated with the PC complexes from UC fractions are more monodispersed in size and showed a much higher cell uptake. No significant differences are instead observed in NP cell uptake from the two UC fractions indicating that, in this case, different structured PC complexes are “seen” similarly by the cells. This is also in agreement with the protein corona pattern observed in the SDS-PAGE (see Figure 5). Uptake experiments done at different time of incubation, 4 hr and 24 hr did not show major changes (data not shown).

Protein coronas of 50 nm SiO_2 NPs have been shown to be very different depending on the protein concentration in the biological fluids²⁷, in particular they form larger aggregates at lower protein concentrations (dimer, trimer, etc.), while smaller and more monodispersed protein-NP complexes form at higher protein concentrations. Here we tried to separate the NP-protein complexes in 10% FBS (protein concentration 0.0036 g/ml) in more resolved fractions containing the different species present *in situ* and compare them with the analogue complexes in 90% FBS (protein concentration 0.032 g/ml). Also in this case the formation of protein-NP agglomerates of different size and composition in 10% FBS made their isolation impractical by conventional centrifugation as extensive aggregation occurred as shown in Table 5. UC resulted in successful separation and recovery of two equivalent (in number of NPs) fractions of NP-protein complexes characterized by hydrodynamic diameters of 110 and 180 nm, respectively. SDS-PAGE showed different pattern of proteins for the two fractions highlighting the different nature of these complexes that are likely to interact differently with biological matter (Fig. 4a). For the sample incubated in 90% FBS, HC and UC samples showed a very similar protein pattern but DLS and NTA results clearly showed that NP-protein

Table 4 DLS and NTA characterization of PS-PEG NPs in different media.

	DLS			NTA		NPs/ml
	D_H^b [nm]	Pdl ^b	Mean ^c [nm]	SD ^c [nm]		
PS-PEG ^a	106.3±0.7	0.04	125±1	38±1	3x10 ¹²	
PS-PEG <i>in situ</i>	105.9±0.5	0.10	136±1	42±2	4x10 ¹¹	
PS-PEG HC	155.7±2.5	0.22	123±10	36±8	1x10 ¹¹	
PS-PEG UC	137.1±0.1	0.22	118±6	34±1	7x10 ¹⁰	

^a in PBS pH 7.4. ^b Hydrodynamic diameter (D_H) and polydispersity index (Pdl) obtained by cumulant analysis of the autocorrelation functions. ^c Mean is the averaged size of all tracked particles and SD is the related Standard Deviation. All the data are presented as the average of three independent measurements.

Cite this: DOI: 10.1039/c0xx00000x

www.rsc.org/xxxxxx

ARTICLE TYPE

Table 5 DLS and NTA characterization of silica NPs in 10% and 90% FBS.

SiO ₂ ^a	DLS		Mean ^c [nm]	NTA		NPs/ml
	D _H ^b [nm]	PdI ^b		SD ^c [nm]	NPs/ml	
SiO ₂ 10% <i>in situ</i>	46.7±0.5	0.02	37±4	18±4	6x10 ¹²	
SiO ₂ 10% HC	141.5±0.2	0.19	211±13	65±8	1x10 ¹³	
SiO ₂ 10% UC1	291.0±3.9	0.36	157±2	73±1	1x10 ¹¹	
SiO ₂ 10% UC2	123.5±1.7	0.23	98±3	43±8	3x10 ¹⁰	
SiO ₂ 10% UC2	179.1±0.9	0.30	153±21	41±9	4x10 ¹⁰	
SiO ₂ 90% <i>in situ</i>	81.1±0.3	0.20	108±5	36±4	1x10 ¹³	
SiO ₂ 90% HC	129.7±4.4	0.27	174±13	75±15	1x10 ¹²	
SiO ₂ 90% UC	80.2±1.1	0.22	101±4	38±1	9x10 ¹⁰	

^a in PBS pH 7.4. ^b Hydrodynamic diameter (D_H) and polydispersity index (PdI) obtained by cumulant analysis of the autocorrelation functions. ^c Mean is the averaged size of all tracked particles and SD is the related Standard Deviation. All the data are presented as the average of three independent measurements

Fig. 4 Characterization of SiO₂ NPs. In graph a) and b) are reported size distributions by intensity percentage of SiO₂ NPs in 10% and 90% FBS respectively, compared to NPs in PBS (dashed line). c) SDS-PAGE gel of the hard corona complexes of SiO₂ NPs in 10% and 90% FBS isolated by ultracentrifugation (UC) and conventional centrifugation methods (HC), respectively, as indicated by the labels below the tracks of the gel.

Table 6 DLS and NTA characterization of Fe₃O₄ NPs in 55% FBS.

Fe ₃ O ₄ ^a	DLS		NTA	
	D _H ^b [nm]	PdI ^b	Mean ^c [nm]	SD ^c [nm]
Fe ₃ O ₄ <i>in situ</i>	51.4±0.3	0.17	93±7	42±12
Fe ₃ O ₄ HC ^d	63.1±1.8	0.50	157±3	51±4
Fe ₃ O ₄ UC1	251.8±9.5	0.55	n.d.	n.d.
Fe ₃ O ₄ UC2	77.4±0.2	0.26	90±1	33±1
Fe ₃ O ₄ UC2	147.1±4.3	0.20	139±3	47±2

^a in PBS pH 7.4. ^b Hydrodynamic diameter (D_H) and polydispersity index (PdI) obtained by cumulant analysis of the autocorrelation functions. ^c Mean is the averaged size of all tracked particles and SD is the related Standard Deviation. All the data are presented as the average of three independent measurements. ^d sample too polydispersed to be measured.

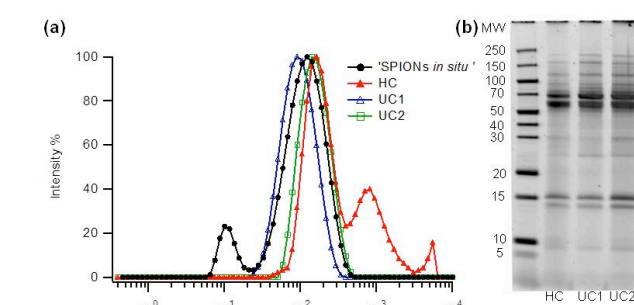


Fig. 5 Fe₃O₄ in 55% FBS. a) Size distributions from DLS measures of Fe₃O₄ NPs *in situ* and HC complexes recovered by ultracentrifugation (UC1-2) and by centrifugal washings (HC) b) SDS-PAGE of hard corona complexes isolated by conventional centrifugation (HC) and by density gradient ultracentrifugation (UC1-2).

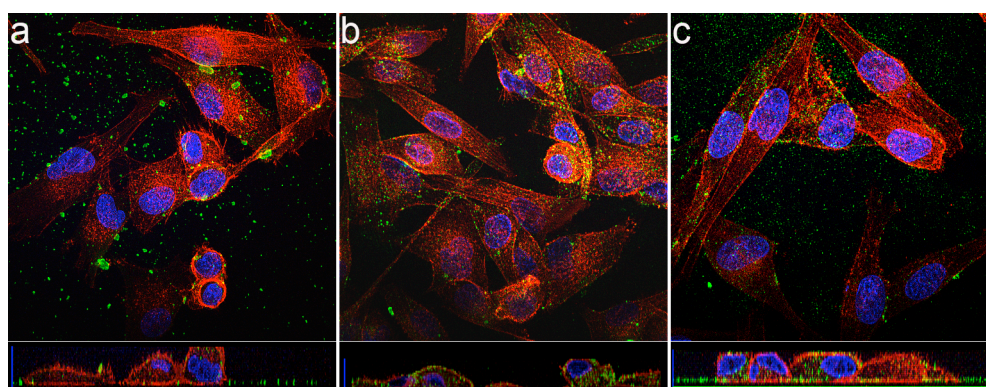


Fig. 6 Fe₃O₄ PCs uptake. Confocal images of M2O2 cells after 2 hours of incubation with Fe₃O₄ PC complexes (10¹² NPs/ml) isolated a) by centrifugation (HC) and b)-c) by ultracentrifugation. Actin filaments were stained by Texas Red@-X Phalloidin (red channel), nuclei by Hoechst 33342 (blue channel) and NPs by BODIPY FL EDA (green channel).

25

Recovery of HC complexes from gastrointestinal fluids.

The importance of this methodology has also been demonstrated in the recovery of NP-protein complexes from different non-serum biological fluids such as simulated gastrointestinal fluids.

The digestion of NPs in these fluids required a very long and complex protocol (see Materials and Methods section) for which the isolation with conventional methods was ineffective, due to NP concentration and agglomeration problems. Investigation of the protein coronas of NPs in gastrointestinal fluids is important in relation to the application of NPs in food⁸, food packaging, toxicology⁴⁷⁻⁴⁹ and medicine. It has been shown that NPs such as iron oxide and Ag NPs tend to aggregate in the gastrointestinal tract^{50, 51} because of the extreme conditions of ionic strength and pH. In a recent paper, Seung-Chul Yang and co-workers⁵² proposed a procedure to stabilize iron oxide NPs in aqueous solution and highlighted that in digestive fluids aggregation occurred, but primary particle size measurements are still possible if factors affecting colloidal stability such as enzymes, pH and electrolytes were removed from samples.

In this scenario, density gradient UC is the appropriate technique to extract PC complexes from these fluids without over-manipulating the samples. In particular, we incubated Fe₃O₄ NPs in simulated salivary, gastric and intestinal fluids prepared as described in the Material and Methods section. NPs appeared to be stable in salivary and gastric fluids forming small clusters of about 100 nm (see Fig 5-6 in SI), while extensive aggregation occurred in the intestinal conditions. In saliva and gastric fluids the recovery and analysis of PC complexes was also possible with conventional methods, although their protein composition was likely affected with respect to that *in situ*. In fact, although NTA analysis showed very similar distributions for corona complexes isolated by UC compared to the normal protocol (fig 7a), the protein patterns, shown in the SDS PAGE, were different (fig 7b) indicating a strong enrichment of proteins in the HC samples probably due to contamination from protein agglomerates that sediment together with the PC complexes. These aggregates are instead removed in the first sucrose gradient layers by UC with the recovery of pure fractions with PC complexes. Subsequent LC-MS analysis of the PC isolated by UC showed that they were mainly composed of pepsin (34Kda), some selected peptides from β -casein hydrolysis at very low molecular weight (6 kDa)^{53, 54}, α -lactalbumin (14.4 kDa) and β -lactoglobulin (18.4 kDa), while HC fractions were characterized by some stronger bands at high molecular weight and consisted of a number of hydrolysed fragments not easily identifiable⁵⁵ due probably to an extended exposure time with the biological fluid during the pellet procedure in the centrifuge (see SI for LC-MS data).^{35, 56, 57}

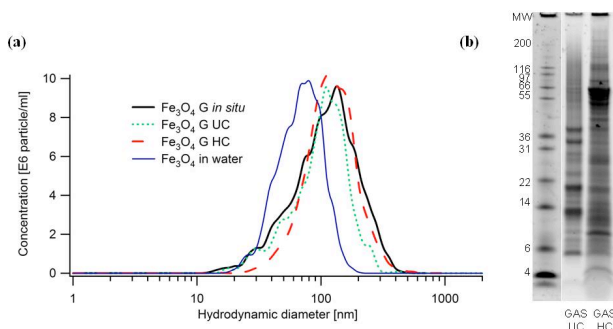


Fig. 7 Fe₃O₄ NPs in simulated gastric fluid. a) Size distributions obtained from NTA analysis for Fe₃O₄ NPs incubated with gastric fluids and relative hard coronas isolated by ultracentrifugation (UC) and centrifugation (HC). b) SDS-PAGE of the hard corona samples isolated by ultracentrifugation (UC) and conventional centrifugation (HC) methods, respectively, as indicated by the labels below the tracks of the gel.

In contrast, the extensive aggregation combined with the complex composition of the intestinal media made impossible to extract PC complexes from this environment by conventional methods. Mainly composed of enzymes (trypsin, 23kDa, and chymotrypsin, 25kDa) and some persistent peptides, intestinal environment led to strong NP aggregation but PC complexes could be isolated through UC. NTA analysis on the starting sample (*in situ*) showed a wide range of particulates ranging from 200 nm to micron size. NTA on UC fractions containing PC complexes recovered from this environment showed a small and relatively narrow population without contamination of the larger complexes that were instead found in the HC complexes isolated by centrifugal washings (see Fig. 8).

To rule out the possibility of proteins running through the sucrose gradient, control gels were performed with fluids in the absence of NPs as can be seen in Fig. 9. The first two sucrose fractions of all samples (with and without NPs) contained the unbound proteins, showing similar composition in both samples (lanes 1-2, Fig. 9 and lanes 1*-2*, Fig. 9b). In the sample with the NPs the sucrose layer labelled as 8 showed the presence of a rich protein pattern with respect to the control that did not show the presence of any protein (lane 8*, Fig. 9b). In fact, sucrose fraction 8 was also brown coloured indicating the presence of Fe₃O₄ NPs. The hard and soft corona obtained by centrifugation are very similar (lane HC and SC of Fig. 9a) while the corona isolated by UC (lane 8, Fig. 9a) presented some notable bands at 21 kDa persistent from the gastric phase and chymotrypsin at 25 kDa. Bile salts caused desorption of proteins according to their concentration and exposure time^{35, 58, 59}, therefore also in this case the ability of UC to limit the contact time between PC complexes and biological medium may affect the corona composition. Some bands at higher molecular weight could not be found anywhere else and showed quite regular spacing among them. From molecular weights estimation, we assume these to be oligomers of very small fragment (4.9 kDa) although lower weight oligomers seem to be missing.

Conclusions

In this study we demonstrated that sucrose gradient ultracentrifugation is an effective tool in the isolation of protein corona complexes from complex biological media without affecting their structure and composition with respect to those *in situ*. This approach can be applied to a wide range of nanoparticles by simply tuning the experimental conditions (centrifugation rate, time, temperature). Moreover, it proved to be a less invasive method keeping the structure and composition of PC complexes intact as well as having a much higher resolution compared to conventional approaches in terms of separation by size. In fact, not only could the NPs be fully recovered from the environment, but also the structure of the resulting NP-protein complexes was more representative of that of the complexes *in situ*. Isolated PC complexes were also incubated with cells and

those recovered by UC methodology showed much less aggregation and higher uptake with respect to those recovered by conventional methods. Moreover, it also allowed a fine separation of the different protein-NPs aggregates present simultaneously in the biological environment even from very complex matrices such as simulated digestive fluids. In fact, to best of our knowledge, this is the first example of recovery of PC complexes from digestive fluids, in which NPs are known to aggregate extensively (mostly in intestinal conditions).

The ability to isolate and recover NP-protein complexes in a stable form with such a high size resolution from diverse biological media will have a significant impact on the interpretation of the role of the protein corona in the interaction with cellular mechanisms. The ability to quantitatively recover such complexes using our approach opens up the possibility of performing systematic biological studies to shed light on these interactions.

Acknowledgments. This work was supported by funds from the British Skin Foundation and the Royal Society. FBB thanks the financial support from Regione Lombardia (Fondo per lo Sviluppo e la Coesione – FAS 2007-2013) Dr Davide Carta and Procarta are acknowledged for the help with NTA measurements. We thank Francesco Rossi for the preparation of digestive fluids.

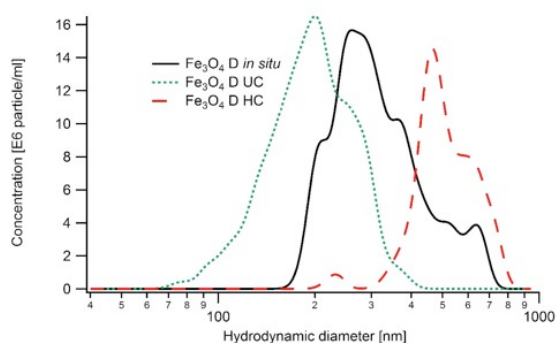


Fig. 8 Fe_3O_4 NPs in simulated intestinal fluid. Size distributions obtained from NTA analysis for Fe_3O_4 NPs incubated with intestinal fluids for one hour and relative hard coronas isolated by ultracentrifugation (UC) and centrifugation (HC).

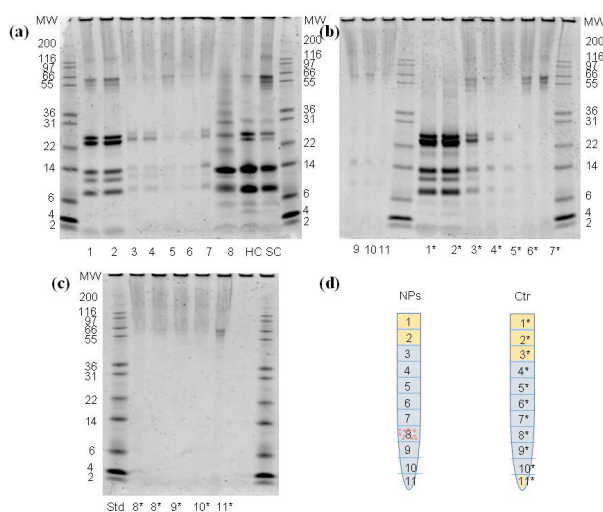


Fig. 9 SDS-PAGE of Fe_3O_4 NPs in simulated intestinal fluid. a) SDS-Gel containing UC fractions of the sample of intestinal fluid containing the NPs. Lane 8 contains the PC complexes, lanes called HC and SC contained soft corona and hard corona complexes, respectively, isolated by conventional methods. b)-c) Control samples without NPs to show no contaminations from free proteins in the lanes with the PC complexes. d) Schematic drawing of the sucrose layer arrangement in the UC tube showing where NPs and proteins were found in samples with NPs (NPs) and without NPs (Ctr).

Notes and references

^a School of Pharmacy, University of East Anglia, Norwich Research Park, NR4 7TJ, Norwich, UK. E-mail: f.baldelli-bombelli@uea.ac.uk

^b Institute of Food Research, Norwich Research Park, Colney Ln, NR4 7UA, Norwich, UK.

^c School of Chemistry, University of East Anglia, Norwich Research Park, NR4 7TJ, Norwich, UK.

^d CEN-European Centre for Nanomedicine c/o Dipartimento di Chimica, Materiali ed Ingegneria Chimica "Giulio Natta", Politecnico di Milano, Milan, Italy.

† Electronic Supplementary Information (ESI) available: [details of any supplementary information available should be included here]. See DOI: 10.1039/b000000x/

1. D. Mishra, J. R. Hubenak and A. B. Mathur, *Journal of Biomedical Materials Research Part A*, 2013, **101**, 3646-3660.
2. M. L. Etheridge, S. A. Campbell, A. G. Erdman, C. L. Haynes, S. M. Wolf and J. McCullough, *Nanomedicine: Nanotechnology, Biology and Medicine*, 2013, **9**, 1-14.
3. E. W. Group, Nano-materials: prevalence in personal care products, <http://www.ewg.org/skindeep/2007/08/25/hundreds-of-personal-care-products-contain-poorly-studied-nano-materials/>.
4. J. P. Kaiser, S. Zuin and P. Wick, *Science of the Total Environment*, 2013, **442**, 282-289.
5. W. Zheng, P. Huang, D. Tu, E. Ma, H. Zhu and X. Chen, *Chemical Society Reviews*, 2015.
6. Z. Valerio, A. Gabriele De, V. Luigi, C. Valentina, M. Claudia, G. Jacek, R. Andrea, C. Aldo Di and M. B. Thomas, *Nanotechnology*, 2013, **24**, 255401.
7. A. N. Shipway, E. Katz and I. Willner, *ChemPhysChem*, 2000, **1**, 18-52.
8. T. Borel and C. M. Sabliov, *Annual Review of Food Science and Technology*, 2014, **5**, 197-213.
9. N. Lewinski, V. Colvin and R. Drezek, *Small*, 2008, **4**, 26-49.
10. A. E. Nel, L. Madler, D. Velegol, T. Xia, E. M. V. Hoek, P. Somasundaran, F. Klaessig, V. Castranova and M. Thompson, *Nat Mater*, 2009, **8**, 543-557.
11. T. Cedervall, I. Lynch, S. Lindman, T. Berggård, E. Thulin, H. Nilsson, K. A. Dawson and S. Linse, *Proceedings of the National Academy of Sciences*, 2007, **104**, 2050-2055.
12. D. Walczyk, F. B. Bombelli, M. P. Monopoli, I. Lynch and K. A. Dawson, *Journal of the American Chemical Society*, 2010, **132**, 5761-5768.
13. M. Mahmoudi, J. Meng, X. Xue, X. J. Liang, M. Rahman, C. Pfeiffer, R. Hartmann, P. R. Gil, B. Pelaz, W. J. Parak, P. del Pino, S. Carregal-Romero, A. G. Kanaras and S. Tamil Selvan, *Biotechnology Advances*, 2014, **32**, 679-692.
14. C. D. Walkey and W. C. W. Chan, *Chemical Society Reviews*, 2012, **41**, 2780-2799.

15. I. Lynch and K. A. Dawson, *Nano Today*, 2008, **3**, 40-47.
16. S. Milani, F. Baldelli Bombelli, A. S. Pitek, K. A. Dawson and J. Rädler, *ACS Nano*, 2012, **6**, 2532-2541.
17. J. Wang, U. B. Jensen, G. V. Jensen, S. Shipovskov, V. S. Balakrishnan, D. Otzen, J. S. Pedersen, F. Besenbacher and D. S. Sutherland, *Nano Letters*, 2011, **11**, 4985-4991.
18. E. A. Vogler, *Biomaterials*, 2012, **33**, 1201-1237.
19. M. Lundqvist, J. Stigler, G. Elia, I. Lynch, T. Cedervall and K. A. Dawson, *Proceedings of the National Academy of Sciences*, 2008, **105**, 14265-14270.
20. E. Casals, T. Pfaller, A. Duschl, G. J. Oostingh and V. Puentes, *ACS Nano*, 2010, **4**, 3623-3632.
21. W. G. Kreyling, S. Fertsch-Gapp, M. Schäffler, B. D. Johnston, N. Haberl, C. Pfeiffer, J. Diendorf, C. Schleh, S. Hirn, M. Semmler-Behnke, M. Eppler and W. J. Parak, *Beilstein Journal of Nanotechnology*, 2014, **5**, 1699-1711.
22. A. Albanese, C. D. Walkey, J. B. Olsen, H. Guo, A. Emili and W. C. W. Chan, *ACS Nano*, 2014, **8**, 5515-5526.
23. M. Lundqvist, J. Stigler, T. Cedervall, T. Berggård, M. B. Flanagan, I. Lynch, G. Elia and K. Dawson, *ACS Nano*, 2011, **5**, 7503-7509.
24. V. Sherwood, D. Di Silvio and F. Baldelli Bombelli, Springer Berlin Heidelberg, 2014, ch. 36, pp. 1-26.
25. S. R. Saptarshi, A. L. Duschl A Fau - Lopata and A. L. Lopata, *Journal of Nanobiotechnology*, 2013.
26. P. d. Pino, B. Pelaz, Q. Zhang, P. Maffre, G. U. Nienhaus and W. J. Parak, *Materials Horizons*, 2014, **1**, 301-313.
27. M. P. Monopoli, D. Walczyk, A. Campbell, G. Elia, I. Lynch, F. Baldelli Bombelli and K. A. Dawson, *Journal of the American Chemical Society*, 2011, **133**, 2525-2534.
28. U. Sakulku, M. Mahmoudi, L. Maurizi, J. Salaklang and H. Hofmann, *Sci. Rep.*, 2014, **4**.
29. L. Böhmert, M. Girod, U. Hansen, R. Maul, P. Knappe, B. Niemann, S. M. Weidner, A. F. Thünemann and A. Lampen, *Nanotoxicology*, 2013, **8**, 631-642.
30. J. Ashby, S. Schachermeyer, S. Pan and W. Zhong, *Analytical Chemistry*, 2013, **85**, 7494-7501.
31. G. J. Howlett, A. P. Minton and G. Rivas, *Current Opinion in Chemical Biology*, 2006, **10**, 430-436.
32. K. L. Planken and H. Colfen, *Nanoscale*, 2010, **2**, 1849-1869.
33. N. Kanokwan, R. Vinayak, A. F. Jeffrey and R. Vytas, *Nanotechnology*, 2013, **24**, 155701.
34. K. E. Moore, M. Pfohl, F. Hennrich, V. S. K. Chakradhanula, C. Kuebel, M. M. Kappes, J. G. Shapter, R. Krupke and B. S. Flavel, *ACS Nano*, 2014.
35. D. Docter, U. Distler, W. Storck, J. Kuharev, D. Wünsch, A. Hahlbrock, S. K. Knauer, S. Tenzer and R. H. Stauber, *Nat. Protocols*, 2014, **9**, 2030-2044.
36. S. Tenzer, D. Docter, J. Kuharev, A. Musyanovych, V. Fetz, R. Hecht, F. Schlenk, D. Fischer, K. Kiouptsi, C. Reinhardt, K. Landfester, H. Schild, M. Maskos, S. K. Knauer and R. H. Stauber, *Nat Nano*, 2013, **8**, 772-781.
37. M. Werwie, N. Fehr, X. Xu, T. Basché and H. Paulsen, *Biochimica et Biophysica Acta (BBA) - General Subjects*, 2014, **1840**, 1651-1656.
38. S. Sun, H. Zeng, D. B. Robinson, S. Raoux, P. M. Rice, S. X. Wang and G. Li, *Journal of the American Chemical Society*, 2003, **126**, 273-279.
39. C.-A. J. Lin, R. A. Sperling, J. K. Li, T.-Y. Yang, P.-Y. Li, M. Zanella, W. H. Chang and W. J. Parak, *Small*, 2008, **4**, 334-341.
40. S. Stolnik, S. Dunn, M. Garnett, M. Davies, A. A. Coombes, D. C. Taylor, M. P. Irving, S. C. Purkiss, T. F. Tadros, S. Davis and L. Illum, *Pharm Res*, 1994, **11**, 1800-1808.
41. J.-C. Leroux, P. Gravel, L. Balant, B. Volet, B. M. Anner, E. Allémann, E. Doelker and R. Gurny, *Journal of Biomedical Materials Research*, 1994, **28**, 471-481.
42. V. Torrisi, A. Graillot, L. Vitorazi, Q. Couzet, G. Marletta, C. Loubat and J. F. Berret, *Biomacromolecules*, 2014, **15**, 3171-3179.
43. M. Mahmoudi, M. A. Shokrgozar, S. Sardari, M. K. Moghadam, H. Vali, S. Laurent and P. Stroeve, *Nanoscale*, 2011, **3**, 1127-1138.
44. M. Jansch, P. Stumpf, C. Graf, E. Rühl and R. H. Müller, *International Journal of Pharmaceutics*, 2012, **428**, 125-133.
45. V. Sánchez-López, J. M. Fernández-Romero and A. Gómez-Hens, *Analytica Chimica Acta*, 2009, **645**, 79-85.
46. D. Pozzi, V. Colapicchioni, G. Caracciolo, S. Piovesana, A. L. Capriotti, S. Palchetti, S. De Grossi, A. Riccioli, H. Amenitsch and A. Lagana, *Nanoscale*, 2014, **6**, 2782-2792.
47. G. Maiorano, S. Sabella, B. Sorce, V. Brunetti, M. A. Malvindi, R. Cingolani and P. P. Pompa, *ACS Nano*, 2010, **4**, 7481-7491.
48. M. J. D. Clift, S. Bhattacharjee, D. M. Brown and V. Stone, *Toxicology Letters*, 2010, **198**, 358-365.
49. Q. Peng, S. Zhang, Q. Yang, T. Zhang, X.-Q. Wei, L. Jiang, C.-L. Zhang, Q.-M. Chen, Z.-R. Zhang and Y.-F. Lin, *Biomaterials*, 2013, **34**, 8521-8530.
50. E. Zimmermann and R. H. Müller, *European Journal of Pharmaceutics and Biopharmaceutics*, 2001, **52**, 203-210.
51. Y. V. Frenkel, A. D. Clark, K. Das, Y.-H. Wang, P. J. Lewi, P. A. J. Janssen and E. Arnold, *Journal of Medicinal Chemistry*, 2005, **48**, 1974-1983.
52. S.-C. Yang, S.-Y.-R. Paik, J. Ryu, K.-O. Choi, T. S. Kang, J. K. Lee, C. W. Song and S. Ko, *Food Chemistry*, 2014, **161**, 185-191.
53. A. Macierzanka, A. I. Sancho, E. N. C. Mills, N. M. Rigby and A. R. Mackie, *Soft Matter*, 2009, **5**, 538-550.
54. M. Defernez, G. Mandalari and E. N. C. Mills, *ELECTROPHORESIS*, 2010, **31**, 2838-2848.
55. N. Kitabatake and Y.-I. Kinekawa, *Journal of Agricultural and Food Chemistry*, 1998, **46**, 4917-4923.
56. M. P. Monopoli, C. Aberg, A. Salvati and K. A. Dawson, *Nat Nano*, 2012, **7**, 779-786.
57. E. Monogioudi, G. Faccio, M. Lille, K. Poutanen, J. Buchert and M.-L. Mattinen, *Food Hydrocolloids*, 2011, **25**, 71-81.
58. T. Winuprasith, S. Chantarak, M. Suphantharika, L. He and D. J. McClements, *Journal of Colloid and Interface Science*, 2014, **426**, 333-340.
59. J. Maldonado-Valderrama, N. C. Woodward, A. P. Gunning, M. J. Ridout, F. A. Husband, A. R. Mackie, V. J. Morris and P. J. Wilde, *Langmuir*, 2008, **24**, 6759-6767.

