Age and Geochemistry of the Mafic Sills, ODP Site 1276, Newfoundland Margin

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Keywords: Site 1276, rift magmatism, continental rifting, Newfoundland margin, Leg 210.

Abstract

Site 1276, Leg 210 of the Ocean Drilling Program, was located on the Newfoundland margin in seismically-defined ~128 Ma "transitional" crust just west of presumed oceanic crust, and the M3 magnetic anomaly. The goal of drilling on this non-volcanic margin was to study the rifting, nature of basement, and post-rift sedimentation in the Newfoundland-Iberia rift. Drilling of this 1739m hole was terminated 90-160 meters above basement, in the lower of a doublet of alkaline diabase sills. We have carried out geochemical studies of the sill complex, in the hopes that they will provide proxy information regarding the nature of the underlying basement. Excellent 40 Ar/ 39 Ar plateau ages were obtained for the two sills: upper sill ~ 105.3 Ma; lower sill ~ 97.8 Ma. Thus the sills are substantially younger than the presumed age of the seafloor at site 1276 (\sim 128 Ma), and were intruded beneath substantial sediment overburden (250 m for the upper, older sill, and 575 m for the lower younger sill). While some of the geochemistry of the sills has been compromised by alteration, the "immobile" trace elements show these sills to be hawaiites, differentiated from an enriched alkaline or basanitic parentage. Sr, Nd and Pb isotopes are suggestive of an enriched hotspot/plume mantle source, with a possible "added" component of continental material. These sills unequivocally were not derived from typical MORB (asthenospheric) upper mantle.

Introduction

The processes by which new oceanic crust is formed during continental rifting have been debated for decades. Rifted margins are classified as volcanic or non-volcanic (White et al., 1987; Mutter, 1993), but the source and nature of the magmatism that can accompany either type of margin remains enigmatic. Are the first magmatic products simply derived from the sub-continental lithospheric mantle that abuts the rifting (Hawkesworth et al., 1999, 2000), or is melting only possible once active upwelling of asthenospheric (MORB) mantle becomes energetic (Mutter et al., 1988; Keen et al., 1994; Rohrman and van der Beek, 1996)? Is such small-scale upwelling even dynamically possible (King and Anderson, 1998)? Does rift-related magmatism require local mantle thermal anomalies, thus implicating mantle plumes as sources for early rift magmatism (White and McKenzie, 1989; Leitch et al., 1998)? These three mantle sources (sub-continental lithosphere, depleted MORB asthenosphere, mantle plumes) have different trace element and isotopic fingerprints, so in principle are distinguishable.

The abyssal Newfoundland basin spans the transition from known continental crust to known oceanic crust (Hopper et al., 2004). Ocean Drilling Program Leg 210, Site 1276, was drilled in ~128 my transitional crust on this non-volcanic margin, to constrain the rifting, nature of basement, and post-rift sedimentation in the Newfoundland-Iberia rift; Figure 1 (Tucholke, B.E., Sibuet, J.-C., Klaus, A., et al., 2004; Shipboard Scientific Party, 2004). A downhole lithology log may be found in the latter reference. While drilling terminated 90-160 meters above true basement, several diabasic sills were cut near the bottom of the 1739 m hole, and these may provide a proxy for basement magmatism in this rift (identification of these bodies as sills rests on the recovery of a baked upper contact for the upper body, and the ages reported below that show both bodies are substantially younger than the enclosing sediment). We show that these sills post-date the rifting and formation of basement by some 25 my years, and are definitely not of MORB pedigree, but are rather of "plume-like" geochemistry.

Methods

Ten core samples from the upper sill and 7 from the lower sill were thin-sectioned and studied petrographically; sample numbers, depths and a brief petrography are given in Table 1. From these samples, three core sections were selected from each sill (weighing from 47 to 120 grams; these six samples are hereafter referred to only by their core and section number, bold

font in Table 1) and fragmented into sub-cm size chips. The degree of alteration is moderate in these samples, ranging from perhaps 25-40%. The freshest available material was hand-picked under a binocular microscope; several grams were sent to Bob Duncan's lab at Oregon State University for ⁴⁰Ar/³⁹Ar analysis, and the remainder was powdered in an agate mill. A split of this powder was analyzed for major and trace elements (XRF and ICP) at the GeoAnalytical Lab, Washington State University. A sub-sample of this powder was leached for 1 hour in hot 6N HCl, repeatedly rinsed in pure water, and used for Sr, Nd and Pb isotopic analysis (Taras and Hart, 1987). In addition, a split of this leached powder was also analyzed for trace elements at the GeoAnalytical Lab, WSU.

The precision (1σ) for major elements in basalts by XRF is 0.11 - 0.33% of the amount present (SiO₂, Al₂O₃, TiO₂, P₂O₅) and 0.38 - 0.71% (other elements). For Ba, Rb, Sr, Zr and Y by XRF, the precision is 0.5 – 3.1% (1 σ); see Johnson et al. (1999) for the XRF techniques, the precision of other trace elements, and for evaluation of accuracy for majors and traces based on analyses of international standards. For trace element analyses by ICP-MS, the precision for basalts is 0.77 – 3.2% (1 σ) for all elements except Th (9.5%) and U (9.3%); see Knaack et al. (1994) for the ICP-MS techniques, and evaluation of accuracy based on analyses of international standards.

Sr and Nd chemistry was done with conventional ion chromatography, using DOWEX 50 cation resin, and HDEHP-treated teflon for Nd separation (Taras and Hart, 1987). Pb chemistry utilized the HBr-HNO₃ procedure of Galer (1986) and Abouchami et al. (1999), with a single column pass. Sr, Nd and Pb analyses were done on the NEPTUNE multi-collector ICP-MS at WHOI. For Sr and Nd, the internal precision is 5-10 ppm (2σ); external precision, after adjusting to 0.710240 and 0.511847 for the SRM987 and La Jolla Nd standards respectively, is estimated to be 15-25 ppm (2σ). Further details may be found in Jackson and Hart (2006) and Hart et al (2005). Pb analyses carry internal precisions on XXX/204 ratios of 15-30 ppm; SRM997 Tl was used as an internal standard, and external reproducibility (including full chemistry) ranges from 17 ppm (2σ) for ²⁰⁷Pb/²⁰⁶Pb, to 117 ppm (2σ) for ²⁰⁸Pb/²⁰⁴Pb (Hart et al., 2002; 2004). Pb ratios were adjusted to the SRM981 values of Todt et al. (1996); two separate lots of the SRM981 standard were inter-compared and any possible isotopic heterogeneity between these lots was <10 ppm for XXX/204 ratios, and <1.5 ppm for ²⁰⁸Pb/²⁰⁶Pb ratios.

Classification and Alteration

Major element, trace element and isotope data are given in Table 2. If the major element chemistry is taken at face value, both sills are clearly alkalic, and would be classified as alkali basalt-hawaiite on a conventional alkali-silica plot; see Fig. 2. Only sample 88-5 plots below the "main cluster"; this sample has high Ba and volatiles and is more altered than the others (see below). The high TiO_2 (3.5%-3.6%) also supports a classification as a differentiated basanite. Mg-Numbers range from 0.42-0.44 for the lower sill, and 0.52-0.53 for the upper sill, consistent with significant differentiation, and location in the hawaiite field.

Primary Cpx phenocrysts from samples 88-5 and 99-6 were analyzed by electron probe to further constrain the composition of the sills prior to any alteration (Supplementary Data, Table 1). The Al_2O_3 and TiO_2 contents show systematic co-variations, Fig. 3, with the clinopyroxenes from the upper sill sample containing a slightly lower CaTiAl₂O₆ component. More importantly, this plot serves as a robust discriminator between alkali basalts and N-MORBS, as the TiO_2 contents of clinopyroxenes from N-MORBS are distinctly lower than the clinopyroxenes from the sill samples, Fig. 3. This is a simple reflection of the lower TiO_2 contents of N-MORB melts (~1.5%) compared to the sill whole rocks (3.5%-3.6%).

However, there is also clear evidence of open-system behavior for some elements during the extensive alteration that these mafic sills have undergone. For example, the Ba contents in the upper sill can be very high (up to 4900 ppm!), and this Ba is located in the abundant acicular secondary apatite that is especially prevalent in the upper sill (Table 1). This high Ba is correlated with low Na₂O, and the low Na₂O is itself correlated with high volatile contents (up to 3%); see Fig. 4. We suggest that the primary Na₂O contents were all similar (~ 3.8%), and that loss of Na has occurred during plagioclase alteration. "Restoring" this Na would shift the upper sill data on Fig. 2 upward by 0.5-1%, making these samples even more alkalic. There is no obvious correlation of K₂O with Ba, Na or volatiles, suggesting that K₂O has been less affected by alteration. This obviously is important in understanding the ⁴⁰Ar/³⁹Ar geochronology.

Splits of the powdered samples were strongly leached, in an attempt to remove alteration phases prior to isotope analysis. This is a common procedure in the isotopic analysis of old and altered basalts (Cheng et al., 1987; Mahoney et al., 1987; Taras and Hart, 1987). A comparison of the leached and unleached trace element concentrations may provide clues as to the nature of the alteration phases, and their impact on the geochemistry of these sills. Fig. 5 is a plot of the

ratio of the leached concentration of the trace elements to the concentrations in the whole rock (unleached) powder. Elements plotting on or above the 1.0 line are those unaffected by leaching, or "concentrated" in the residual powder. These include the elements normally classified as "immobile" (Th, Nb, Ta, Zr, Hf). It is interesting that Rb and U, elements that are usually susceptible to alteration, are also relatively immobile. However, the Ba that was added to the upper sill during alteration is not significantly "removable" by leaching (whereas the Cs is). In other words, lack of "leachability" does not infer lack of alteration. Pb, Sr and Eu in the upper sill samples are less affected by leaching than in the lower sill samples, perhaps reflecting their hosting in plagioclase. While the abundance of groundmass plagioclase is comparable in both sills, the upper sill does have small quantities of distinctly larger plagioclase phenocrysts that may be more resistant to the leaching procedure.

Also note that the REE, normally considered immobile elements, are strongly leached in all cases, and comparable to Cs and K in this regard. The inference then is that the REE, Cs and K are contained in phases that are leachable, but that are not hosts for the "unleached" elements such as Rb, Ba, Th, U, Nb, Ta, Zr and Hf. While a glassy or fine-grained mesostasis might be considered as a candidate for a leachable REE-containing phase, this should also provide leachable Rb, Ba, Th, etc., whereas these elements do not appear to be leachable. It might be inferred that a large fraction of the REE have been added during alteration, and are contained in leachable REE-specific phases such as rhabdophane. This REE-phosphate is common as an alteration product in subaerial basalts (Fodor et al., 1989; Cotten et al., 1995), and is acid-soluble. The presence of secondary apatite in the upper sill is evidence that phosphorous was available in the alteration fluids. To our knowledge, however, rhabdophane has not been reported as an alteration product in submarine settings. Also, the whole-rock spidergrams, to be discussed below, give no indication that the REE are anything but primary magmatic in nature.

Geochronology

Ages on two samples from each sill were determined by step-release ⁴⁰Ar/³⁹Ar dating. Despite the moderate alteration state of these samples, the Ar plateaus are excellent, Fig. 6 (data given in Table 2 supplementary material), and the ages meet the four statistically testable criteria given by Duncan and Keller (2004) that are necessary for an age to define an accurate estimate of the time of sample crystallization. The low-temperature Ar fractions (15-20%) are younger

than the high-temperature plateau ages, perhaps indicating time-integrated (open system) Ar loss from a potassic mineral or glass that formed contemporaneously with sill emplacement, or a younger addition of potassic clay minerals from which the Ar was released during the lower temperature steps of the lab heating. Note that 60-80% of the K in the sill samples was leachable, as shown in Fig. 5, suggesting that at least some of the radiogenic Ar might also be loosely-bound.

The two ages for the upper sill agree within ~1% (104.7 \pm 1.7 Ma, 105.9 \pm 1.8 Ma, 2 σ errors), averaging 105.3 Ma. The two ages for the lower sill are younger, and are spread by 3.8 Ma, just at 2σ analytical precision (99.7 ± 1.8 Ma, 95.9 ± 2.0 Ma), averaging 97.8 Ma; the ~ 7.5 Ma difference in measured age between the two sills is highly significant on a statistical basis. The harder question to answer is whether this difference in measured ages also represents a difference in crystallization ages. It could be argued, for example, that the younger age for the lower sill is an "alteration age", caused by a late addition of K. There is little support for this in the trace element geochemistry, however. For example, both the content of K, and the ratio of K to immobile elements such as Nb and Ta is lower in the lower (younger) sill by 35-45% (table 2), as is the K/La ratio (whereas the absolute Nb, Ta and La contents are very similar in the two sills). This is the opposite of what would be expected if the younger lower sill ages are due to late addition of K. Also, various trace element ratios in the lower sill are closer to "canonical" OIB values than those in the upper sill. This is particularly true for the youngest sample (99-6), that has K/La, Rb/Cs, Ba/Rb, K/Rb and Nb/U of 407 (306), 84 (95), 21.5 (11.3), 406 (480) and 51 (52); canonical values are given in parentheses and are from Hofmann and White (1983), Morris and Hart (1983) and Hofmann (2003). While these geochemical considerations may not constitute compelling evidence that the measured ⁴⁰Ar/³⁹Ar ages are in fact crystallization ages, there is little in the geochemistry that argues otherwise. This, along with the excellent 40 Ar/ 39 Ar plateaus, leads us to accept the measured ages as crystallization ages.

It follows then that the sills are not related to the same magmatic event. This is substantiated by the isotopic and trace element data (see below) that also show that these two sills, while familial, are not strictly co-genetic.

The sills are substantially younger than the presumed age of the seafloor at Site 1276 (just west of M3 anomaly, ~ 128 Ma; Tucholke, Sibuet, Klaus et al., 2004), and they are also younger than the host sediments (~ 111 Ma at upper sill depth, > 115 Ma (?) at lower sill depth). There

was ~ 250 m of sedimentary overburden when the upper (older) sill was intruded, and ~ 575 m when the lower sill was intruded (using the nannofossil-based age-depth curve, Figure 141 in: Shipboard Scientific Party, 2004).

Trace element geochemistry

The easiest way to illustrate the overall chemistry of the two sills is with a trace element abundance plot (or "spidergram"), Fig. 7. On this plot, the two sills appear very similar in geochemistry, with the exception of Cs, Ba, Pb and Sr. Cs is very susceptible to alteration, and we have already commented on the Ba alteration anomaly. The Pb isotopes are very well behaved (i.e. Pb is not significantly affected by alteration; see below), thus we suggest the difference in Pb between the two sills may relate to variable control of Pb by sulfide during melting and/or differentiation (Hart and Gaetani, 2006). The Sr is variable within each sill, as well as between the sills; the high Sr in the lower sill is also very leachable (Fig. 5), suggesting containment in carbonate. However, carbonate is petrographically much more apparent in the upper sill.

Also shown on the spidergram plot, for comparison, is a pattern for oceanic EM2 basalts, as representative of enriched oceanic "hotspot" volcanism. From La to Lu, the EM2 pattern is very similar to the patterns of the sills. For the more incompatible elements, the sills lack the negative K anomaly (ubiquitous in all oceanic basalts, including MORB), and the peak at Nb-Ta. Except for the Cs and Ba alteration anomalies, the sills also appear generally more depleted than EM2 basalts in U and Th. While the lower Nb and Ta in the sills is suggestive of basalts that have experienced some contamination by continental materials, the low U and Th (and Rb) are inconsistent with crustal components. It is more likely that the sills simply reflect a somewhat higher degree of partial melting of a fairly "garden variety" OIB-type source. What is unambiguously clear is that these basalts are not related to MORB.

To further quantify this geochemical comparison, Fig. 8 compares several specific trace element indicators in the sills with a sampling of "end-member" oceanic basalts, and various continental materials. La/Sm is an index of the degree of incompatible-element enrichment (mantle source composition, plus modification during melting), showing that the sills are much more "enriched" than N-MORB, but less enriched than oceanic hotspot (plume) basalts from EM1, EM2 or HIMU islands. Nb/Nb* measures the abundance of Nb relative to the normalized

average of Th and La (Nb*). The well-known deficiency of high-field-strength elements, such as Nb, in continental materials, is indicated by low Nb/Nb* values. The Site 1276 sills are in fact quite normal in Nb/Nb* compared to oceanic basalts, and in this respect do not have a discernable "continental" signature.

So far we have shown that the Site 1276 sills are clearly not MORB and, while broadly similar to oceanic hotspot basalts, they are also different from these in detail. A more relevant comparison would be with Mesozoic rift-related basalts and diabases from the coastal region of northeastern North America (Jansa and Pe-Piper, 1988, McHone, 1992, Pe-Piper et al., 1990,1992, 1994). Unfortunately, most studies of these rocks do not contain comprehensive high-quality trace element data, or the state of preservation of the rocks is poor. Elements such as Sr, Ba and Zr are the most commonly reported, and are typically fair to good in resistance to alteration. Fig. 9 shows a sampling of this data, in comparison to the Site 1276 sills, from: N-MORB, Mesozoic Appalachian Tholeiites (MAT), and other mafic rift-related volcanics from the Grand Banks, Scotia Shelf, NE Seamounts, Newfoundland Seamounts, Georges Bank, Baltimore Canyon, and average lamprophyres from New England and Quebec. The anomalous Ba in the upper sill is quite obvious on this plot. Otherwise, the sills fall along the crude Zr-Sr and Ba-Sr arrays defined by the other data. The sills are clearly more enriched than the Mesozoic tholeiitic diabases (MAT), though not as enriched as some of the lamprophyres from New England, or those drilled on Georges Bank and Baltimore Canyon. The sills, particularly the upper sill, are quite similar in chemistry to basalts dredged from several of the Newfoundland Seamounts (Sullivan and Keen, 1977); these seamounts lie only 190 km SSW of Site 1276, and they are also close in age (97.7 my; Sullivan and Keen, 1977) to the Site 1276 sills. Karner and Shillington (2005) have noted this temporal equivalence; they also propose that the migration of the Azores, Madeira and Canary plumes across this region between 80-120 my ago is responsible for widespread magmatic activity along this rifted margin. It might also be noted that the Site 1276 locality is directly "downstream" from the Milne Seamounts (see Fig. 1), but the relevance of this possible "connection" must await geochemical work on this hot spot chain.

Isotope geochemistry

Here we use the Sr, Nd and Pb isotope data on the sills to "fingerprint" the nature of the mantle source supplying the magmas, and compare the isotopic signature with that of a selection

of North Atlantic hotspots (OIBs). Because of the age of the sills, neither the present day isotope ratios nor the initial (age-corrected) ratios can be directly compared with Sr, Pb and Nd isotopes from the hotspots (most of which are < 5 Ma). Instead we utilize the following scheme (similar to that described in Taras and Hart, 1987): the isotope data for the sills are first corrected back to initial ratios, using the measured 40 Ar/ 39 Ar plateau ages, and the parent/daughter ratios as measured on the leached powders. It is important to use the leached powder data for this agecorrection, because the leached residue that the isotopes were measured on has different parent/daughter ratios than the bulk rock (see Fig. 5). This effect is particularly large for the lower sill samples, due to the very high Th/Pb and U/Pb ratios of these leached powders. These "initial ratios" are then modeled as representing mantle domains that were differentiated at 1.9 Ga from a "bulk earth" mantle (the present day values for our "bulk earth" mantle are fairly conventional: ${}^{206}Pb/{}^{204}Pb = 17.879$; ${}^{207}Pb/{}^{204}Pb = 15.530$; ${}^{208}Pb/{}^{204}Pb = 37.925$; ${}^{87}Sr/{}^{86}Sr = 10.530$; ${}^{86}Sr = 10.530$; ${}^{$ 0.70470; ¹⁴³Nd/¹⁴⁴Nd = 0.512638). For each sample, a model parent/daughter (mu) ratio was calculated for this evolution from 1.9 Ga to 100 Ma. The initial ratios for each sample were then projected ahead to present using these model parent/daughter ratios. This forward projection is not very sensitive to the assumed age of mantle differentiation.

Table 3 shows a comparison of the Sr, Nd and Pb isotope data as measured on the leached powders, with the calculated initial ratios, and the forward projected model values. For Sr and Nd, the differences between measured and model ratios are minor (50-500 ppm for ⁸⁷Sr/⁸⁶Sr, 27-56 ppm for ¹⁴³Nd/¹⁴⁴Nd). For the Pb isotopes, the initial ratio correction is large, due to the high U/Pb and Th/Pb ratios of the Core 99 leached powders; this propagates to a significant difference between measured and present model ratios. It is difficult to assess the extent to which these calculated ratios may be affected by alteration effects, either directly in the Pb isotope component, or in the parent/daughter ratios. We note however that the Pb data are remarkably linear in all Pb-Pb plots (measured, initial or model). In addition, the data are well-aligned in U/Pb and Th/Pb isochron plots of the raw measured data (not shown); while the isochron "ages" derived from these arrays do not agree (185 Ma for U/Pb versus 135 Ma for Th/Pb), this simply suggests that the various samples did not have a common Pb isotope composition when the sills formed. This observation is further emphasized by the apparent Pb-Pb isochron age of 395 Ma (for the measured data).

As a further test of possible alteration effects on the isotope data, we analyzed both the leached and unleached powders for ⁸⁷Sr/⁸⁶Sr. We compare these data in terms of initial ratios, as the Rb/Sr ratios in the leached and unleached powders are quite different. This data is reported in Table 3. In 4 of the 6 samples, the initial ⁸⁷Sr/⁸⁶Sr in the unleached powder is higher, as might be expected for altered basalts in a seawater environment (i. e. sediment pore waters). In one case, the ⁸⁷Sr/⁸⁶Sr is slightly lower (71 ppm), and in one case it is the same. However, the differences are surprisingly small, the largest being only 540 ppm, and the average difference being only 275 ppm. If the ⁸⁷Sr/⁸⁶Sr in these sills has been affected by alteration, the process has affected both sills (and all samples) in a very similar way, and the alteration has been so pervasive that the leachable and unleachable ⁸⁷Sr/⁸⁶Sr is not distinguishable. While it would be reckless to draw apodictic conclusions from these highly altered sills, the most parsimonious interpretation is that the ⁸⁷Sr/⁸⁶Sr ratios are close to primary magmatic values.

Fig. 10 shows the model present-day isotope data for Sr, Nd and Pb. First, it is clear from the Nd and Pb isotope data that the upper and lower sills are not strictly co-genetic, but have distinctly different isotopic signatures. It is also clear (as with the trace elements) that the sills are not even remotely MORB-like, so are not related to normal ocean crust production from the depleted upper mantle. In comparison with a sampling of isotopic data from North Atlantic hotspot basalts (Iceland, Azores, Canaries, Madeira), the Pb isotope arrays fall close to the hotspot fields, but with a clear divergence toward higher ²⁰⁷Pb/²⁰⁴Pb at lower ²⁰⁶Pb/²⁰⁴Pb; this is a hallmark of both the oceanic EM1 mantle end-member, and of old continental lithosphere. This divergence is even more obvious in the Nd-Pb isotope plot, where the data for the Site 1276 sills ranges orthogonally away from the field of North Atlantic hotspots. The mixing end-member for this array is very similar to the EM1 hotspot mantle end-member (though the closest such hotspot is Gough Island, S. Atlantic!). In Nd-Sr isotope space, the sill data fall at the high ⁸⁷Sr/⁸⁶Sr end of the North Atlantic hotspot field – but well within the domain of EM2 hotspots (e.g. Samoa or the Societies). Our conclusion is that the sills are clearly not MORB-like, but fall within the general global domain of plumes or hotspots, though none of the known North Atlantic "plumes" are a close match to the sills. As yet, the nearest hotspots (Newfoundland and Milne Seamounts) are isotopically *incognitus* in the published literature.

Summary

The Site 1276 basaltic sills were intruded into a significant overburden of syn-rift sediments, and are some 25 Ma younger than the underlying basement. Their enriched geochemistry identifies them as likely being of hotspot or plume derivation, but they are definitely not of MORB spreading ridge character. Several scenarios are possible: 1). The sills were intruded off-axis through typical MORB ocean crust, several hundred km west of an active spreading center. Their alkaline character reflects preferential melting of enriched enclaves embedded in depleted MORB mantle (similar to the model proposed by Zindler et al. (1984) and Castillo and Batiza (1989) for off-axis EPR and MAR seamounts). 2). The sills, and the older underlying basement, are both of alkaline character, formed from plume-type mantle during the transitional stages of rifting. The Milne Seamount chain to the east may be implicated in this magmatism. Thus the enduring enigma of early-rift magmatism persists, and a final resolution must await a full penetration to basement at a drill site such as Site 1276.

Acknowledgments

We thank Brian Tucholke for piquing our interest in Site 1276, and for his helpful tutorials on same. Lary Ball performed his usual miracles with the NEPTUNE. Matt Jackson carried out the electron probe work and did the Ba sleuthing. Helpful reviews were provided by Stephanie Ingle and Georgia Pe-Piper. This research used samples provided by the Ocean Drilling Program (ODP is sponsored by the U.S. National Science Foundation (NSF) and participating countries under management of Joint Oceanographic Institutions (JOI), Inc.). Funding for this research was provided by JOI/USSSP 261855 and NSF-EAR0509891.

References

- Abouchami, W., Galer, S. J. G., Koschinsky, A., 1999. Pb and Nd isotopes in NE Atlantic Fe-Mn crusts: proxies for trace metal paleosources and paleocean circulation. Geochim. Cosmochim. Acta 63, 1489-1505.
- Castillo, P. and Batiza, R., 1989. Strontium, neodymium and lead isotope constraints on nearridge seamount production beneath the South Atlantic. Nature 342, 262-265.
- Cheng, Q., Park, K.-H., Macdougall, J. D., Zindler, A., Lugmair, G. W., Hawkins, J. Lonsdale, P. and Staudigel, H., 1987. Isotopic evidence for a hot spot origin of the Louisville seamount

chain. In: Seamounts, Islands and Atolls, B. Keating, P. Fryer, R. Batiza and G. W. Boehlert, eds., Am. Geophys. Union Monogr. 43, 283-296.

- Cotten J., Le Dez, A., Bau, M., Caroff, M., Maury, R. C., Dulski, P., Fourcade, S., Bohn, M., Brousse, R., 1995. Origin of anomalous rare-earth element and Yttrium enrichments in subaerially exposed basalts: Evidence from French Polynesia. Chemical Geology 119, 115-138.
- Dick, H. J. B. and Johnson K. T. M., 1995. REE and Trace Element Composition of Clinopyroxene Megacrysts, Xenocrysts, and Phenocrysts in Two Diabase Dikes from Leg 140, Hole 504B. Proc. ODP 137/140, 121-130.
- Duncan, R. A. and Keller, R. A., 2004. Radiometric ages for basement rocks from the Emperor Seamounts, ODP Leg 197. Geochemistry, Geophysics, Geosystems 5(8): Q08L03, doi:10.1029/2004GC000704.
- Eisele, J., Sharma, M., Galer, S.J.G., Blichert-Toft, J., Devey, C.W., Hofmann, A.W., 2002. The role of sediment recycling in EM-1 inferred from Os, Pb, Hf, Nd, Sr isotope and trace element systematics of the Pitcairn hotspot. Earth and Planetary Science Letters, 196, 197-212.
- Fodor R. V., Malta, D. P., Bauer, G. R., Jacobs, R. S., 1989. Microbeam Analyses of Rare-Earth Element Phosphate in Basalt from Kahoolawe Island, Hawaii. In: Proceedings of the 24th Annual Conference of the Microbeam Analytical Society, 554-558.
- Galer, S. J. G., 1986. Chemical and isotopic studies of crust-mantle differentiation and the generation of mantle heterogeneity. Unpubl. PhD., University of Cambridge.
- Gao, S., Luo, T-C., Zhang, B-R., Zhang, H-F., Han, Y-W., Zhao, Z-D., Hu, Y-K., 1998. Chemical composition of the continental crust as revealed by studies in East China. Geochimica et Cosmochimica Acta 62, 1959-1975.
- Hart, S.R. and Gaetani, G.A., 2006. Mantle Pb paradoxes: the sulfide solution. Contributions to Mineralogy and Petrology 152, xxx.
- Hart, S. R., Workman, R. K., Coetzee, M., Blusztajn, J., Ball, L., Johnson, K. T. M., 2002. The Pb Isotope Pedigree of Western Samoan Volcanics: New Insights from High-Precision Analysis by NEPTUNE ICP/MS, EOS 83, F20.
- Hart, S. R., Ball, L., and Jackson, M., 2005. Sr Isotopes by Laser Ablation PIMMS: Application to Cpx from Samoan Peridotite Xenoliths, WHOI Plasma Facility Open File Technical Report 11. (<u>http://www.whoi.edu/science/GG/people/shart/open_file.htm</u>)
- Hart, S. R., Workman, R. K., Ball, L., and Blusztajn, J., 2004. High Precision Pb Isotope Techniques from the WHOI NEPTUNE PIMMS, *WHOI Plasma Facility Open File Technical Report 10.* (http://www.whoi.edu/science/GG/people/shart/open_file.htm)

Hauri, E.H. and Hart, S.R., 1997. Rhenium abundances and systematics in oceanic basalts. Chemical

Geology 139,185-205.

- Hawkesworth, C.J., Gallagher, K., Kirstein, L., Montovani, M.S.M., Peate, D.W., and Turner, S.P., 2000. Tectonic controls on magmatism associated with continental break-up: an example from the Paraná-Etendeka Province. Earth and Planetary Science Letters 179, 335-349.
- Hawkesworth, C.J., Kelley, S., Turner, S.P., le Roex, A., and Storey, B., 1999. Mantle processes during Gondwana breakup and dispersal. J. Afr. Earth Sci. 28, 239-261.
- Hofmann, A., 2003. Sampling Mantle Heterogeneity through Oceanic Basalts: Isotopes and Trace Elements. In: H.D. Holland and K. K. (Editors), Treatise on Geochemistry, 2.03, 61-101.
- Hofmann, A.W. and White, W.M., 1983. Ba, Rb and Cs in the Earth's Mantle. Z. Naturforsch 38a, 256-266.
- Hopper, J.R., Funck, T., Tucholke, B.E., Larsen, H.C., Holbrook, W.S., Louden, K.E., Shillington, D.J., and Lau, H., 2004. Continental breakup and the onset of ultraslow seafloor spreading off Flemish Cap on the Newfoundland rifted margin. Geology 32, 93-96.
- Hurtubise, D.O., Puffer, J.H., and Cousminer, H.L., 1987. An offshore Mesozoic igneous sequence, Georges Bank basin, North Atlantic. Geological Society of America Bulletin 98, 430-438.
- Jackson, M. G. and S. R. Hart, 2006. Strontium Isotopes in melt inclusions from Samoan basalts: Implications for the heterogeneity in the Samoan plume. Earth and Planetary Science Letters 245, 260-277.
- Jansa, L.F., and Pe-Piper, G., 1988. Middle Jurassic to Early Cretaceous Igneous Rocks Along Eastern North American Continental Margin. American Association of Petroleum Geologists Bulletin 72, 347-366.
- Johnson, D.M., Hooper, P.R., Conrey, R.M., 1999. XRF Analysis of Rocks and Minerals for Major and Trace Elements in a Single Low Dilution Li-tetraborate Fused Bead. Advances in X-Ray Analysis 41, 843-867.
- Karner, G.D., and Shillington, D.J., 2005. Basalt sills of the U reflector, Newfoundland Basin: A serendipitous dating technique. Geology 33, 985-988.
- Keen, C. E., Courtney, R. C., Dehler, S. A., Williamson, M.-C., 1994. Decompression melting at rifted margins: comparison of model predictions with the distribution of igneous rocks on the eastern Canadian margin. Earth and Planetary Letters 121, 403-416.
- King, S.D., and Anderson, D.L., 1998. Edge-driven convection. Earth and Planetary Science Letters 160, 289-296.

- Knaack, C.M., Cornelius, S.B., Hooper, P.R., 1994. Trace element Analysis of rocks and minerals by ICP-MS. In: Open File Report, GeoAnalytical Lab, Washington State University, p. 10.
- Koppers, A.A.P., Staudigel, H., Duncan, R.A., 2003. High-Resolution ⁴⁰Ar/³⁹Ar dating of the oldest oceanic basement basalts in the western Pacific basin. Geochemistry, Geophysics, Geosystems 4(11):8914, doi:8910.1029/2003GC000574.
- Leitch, A. M., Davies, G. F., Wells, M., 1998. A plume head melting under a rifting margin, Earth and Planetary Science Letters 161, 161-177.
- Mahoney, J. J., 1987. An isotopic survey of Pacific oceanic plateaus: implications for their nature and origin, in: Keating, B. H., Fryer, P., Batiza, R. and Boehlert, G. W. (eds), Seamounts, Islands and Atolls, Am. Geophys. Union Monogr. 42, 207-220.
- McCulloch, M.T., Bennett, V.C., 1994. Progressive growth of the Earth's continental crust and depleted mantle. Geochimica et Cosmochimica Acta 58, 4717-4738.
- McDonough, W.F., and Sun, S-s, 1995. The composition of the Earth. Chemical Geology 120, 223-253.
- McHone, J.G., 1992. Mafic dike suites within Mesozoic igneous provinces of New England and Atlantic Canada, p. 1-11 *in*: Eastern North American Mesozoic Magmatism, , Puffer, J.H., and Ragland, P.C., eds., Geological Society of America Special Paper 268, 406 p.
- Morris, J.D. and Hart, S.R., 1983. Isotopic and incompatible element constraints on the genesis of island arc volcanics from Cold Bay and Amak Island, Aleutians, and implications for mantle structure. Geochim. Cosmochim. Acta 47, 2015-2030.
- Mutter, J.C., 1993. Margins declassified. Nature 364, 393–394.
- Mutter, J.C., Buck, W.R., and Zehnder, C.M., 1988. Convective partial melting, 1. A model for the formation of thick basaltic sequences during the initiation of spreading. J. Geophys. Res. 93, 1031–1048.
- Pan, Y. and Batiza, R., 2003. Magmatic processes under mid-ocean ridges: A detailed mineralogic study of lavas from East Pacific Rise 9°30'N 10°30'N and 11°20'N. Geochemistry, Geophysics, Geosystems, 4(11) 8623, doi:10.1029/2003GC000309.
- Pegram, W. J., 1990. Development of continental lithospheric mantle as reflected in the chemistry of the Mesozoic Appalachian tholeiites, U. S. A. Earth Planetary Science Letters 97, 316-331.
- Pe-Piper, G., Jansa, L.F., Palacz, Z. 1994. Geochemistry and regional significance of the Early Cretaceous bimodal basalt-felsic associations on the Grand Banks, eastern Canada. Geological Society of America Bulletin 106, 1319-1331.

- Pe-Piper, G., and Jansa, L.F., 1987. Geochemistry of the late Middle Jurassic—Early Cretaceous igneous rocks on the eastern North American margin. Geological Society of America Bulletin 99, 803-813.
- Pe-Piper, G., Jansa, L.F., and Lambert, R.S.J., 1992. Early Mesozoic magmatism on the Eastern Canadian margin: Petrogenic and tectonic significance, p. 13-36 in: Eastern North American Mesozoic Magmatism, , Puffer, J.H., and Ragland, P.C., eds., Geological Society of America Special Paper 268, 406 p.
- Pe-Piper, G., Piper, D.J.W., Keen, M.J., and McMillan, N.J., 1990. Igneous rocks of the continental margin: tectonic and geophysical overview. p.75-84 *in*: Keen, M.J., and Williams, G.L., Eds., Geology of the Continental Margin of Eastern Canada. Geol. Soc. Am. Memoir 2, 855 p.
- Plank, T., and Langmuir, C.H., 1998. The chemical compositions of subducting sediments and its consequences for the crust and mantle. Chemical Geology 145, 325-394.
- Rohrman, M., and van der Beek, P., 1996. Cenozoic postrift domal uplift of North Atlantic margins: An asthenospheric diapirism model. Geology 24, 901-904.
- Ross, M.E., 1992, Petrology and tectonic significance of Mesozoic mafic dikes of the coastal New England igneous province, Massachusetts. *in* Puffer, J.H., and Ragland, P.C., eds., Eastern North American Mesozoic Magmatism. Geological Society of America Special Paper 268, 406 p.
- Rudnick, R.L., Fountain, D.M., 1995. Nature and composition of the continental crust: a lower crustal perspective. Review of Geophysics 33, 267-309.
- Shipboard Scientific Party, 2004. Leg 210 summary, In: Tucholke, B.E., Sibuet, J.-C., Klaus, A., et al., Proc. ODP Init. Repts., 210: College Station TX (Ocean Drilling Program), 1–78.
- Su, Y., and Langmuir, C. H., 2003, Global MORB chemistry compilation at the segment scale, MS Thesis, Department of Earth and Environmental Sciences, Columbia University. Available at: <u>http://petdb.ldeo.columbia.edu/documentation/morbcompilation/</u>.
- Sullivan, K. D. and Keen, C. E., 1977. Newfoundland Seamounts Petrology and Geochemistry, in Baragar, W.R.A., Coleman, L.C., Hall, J.M., eds., Volcanic regimes of Canada, Geol. Assoc. Canada Special Paper 16, 461-476.
- Taras, B. D., Hart, S.R., 1987. Geochemical evolution of the New England seamount chain: isotopic and trace element constraints. Chemical Geology 64, 35-54.
- Tucholke, B.E., Sibuet, J.-C., Klaus, A., et al., 2004. Proc. ODP, Init. Repts., 210. Ocean Drilling Program, College Station TX.
- Todt, W., Cliff, R.A., Hanser, A., Hofmann, A. W., 1996. Evaluation of a 202Pb-205Pb double spike for high-precision lead isotope analysis, *In*: Basu, A. and Hart, S.R., (Eds.) Earth

Processes: Reading the Isotopic Code. Am. Geophys. Union Geophysical Monograph 95, 429-437.

- White, R.S., and McKenzie, D., 1989. Magmatism at rift zones: the generation of volcanic continental margins and flood basalts. J. Geophys. Res. 94, 7685–7729.
- White, R.S., Spence, G.D., Fowler, S.R., McKenzie, D.P., Westbrook, G.K., and Bowen, A.N., 1987. Magmatism at rifted continental margins. Nature 330, 439–444.
- Woodhead J.D., 1996. Extreme HIMU in an oceanic setting: the geochemistry of Mangaia Island (Polynesia), and temporal evolution of the Cook-Austral hotspot. J. Volcanology and Geothermal Research 72, 1-19.
- Workman, R. K., Hart, S. R., Jackson, M., Regelous, M, Blusztajn, J., Kurz, M., Farley, K., and Staudigel, H., 2004. Recycled Metasomatized Lithosphere as the Origin of the Enriched Mantle II (EM2) End-member: Evidence from the Samoan Volcanic Chain. G-Cubed 5, 2003GC000623.
- Zindler, A., Staudigel, H., and Batiza, R., 1984. Isotope and trace element geochemistry of young Pacific seamounts: implications for the scale of upper mantle heterogeneity. Earth and Planetary Science Letters 70, 175-195.

Figure Captions

- Fig. 1. Bathymetry of the northern Newfoundland Basin, showing the location of Site 1276 at the foot of the continental rise south of Flemish Cap. The Newfoundland and Milne seamount chains are also shown. Site 1276 was drilled in 4560 m of water, to a total depth of 1739 m below sea floor, at 45° 24.3198'N, 44° 47.1496'W.
- Fig. 2. Alkali-Silica classification plot showing that both lower (blue diamonds) and upper (red circles) sills from Site 1276 lie in the hawaiite field. A global average (Su and Langmuir, 2003) of normal mid-ocean ridge basalts (N-MORB, filled square) lies well below the alkali basalt-tholeiite dividing line. Data are for unleached powders; Table 2.
- Fig. 3. Al₂O₃-TiO₂ contents of clinopyroxenes from two samples, Site 1276 sills. Data for the clinopyroxene compositions is given in Table 1, Supplementary material. Representative N-MORB clinopyroxene compositions are from Dick and Johnson (1995) and Pan and Batiza (2003).

- Fig. 4. Na₂O, Ba and volatile content of lower (blue diamonds) and upper (red circles) sills from Site 1276. Note log scale for Ba, and the general correlation of high Ba and low Na₂O with increasing alteration (high volatile content). Data are for unleached powders; Table 2.
- Fig. 5. Trace element abundances (spidergram) for the ratio of concentration in leached rock powder divided by concentration in unleached rock powder (Table 2); average of three upper sill samples – red circles; average of three lower sill samples – blue diamonds.
- Fig. 6. ⁴⁰Ar/³⁹Ar stepwise heating profiles for two un-leached upper sill samples (88-2, 88-3) and two un-leached lower sill samples (99-3, 99-6). Data for these ages is given in Table 2, Supplementary material. Step-release heating from 500°-1400°C; 2-sigma errors include measurement uncertainties, and uncertainty in J-value (flux gradient from FCT-3 biotite monitor), but not uncertainty in monitor age. Ages determined in Bob Duncan's Noble Gas Mass Spectrometry Lab, Oregon State University. Details of the techniques are given in Duncan and Keller, 2004, and Koppers et al., 2003.
- Fig. 7. Trace element abundances for the average of three upper sill (red circles) and three lower sill (blue diamonds) samples, normalized to bulk silicate earth abundances (McDonough and Sun, 1995; Cs value was adjusted down to 0.010 ppm to provide a more realistic Rb/Cs = 60 for BSE). An average EM2 basalt pattern, dashed line, is shown for comparison (average of 5 highest ⁸⁷Sr/⁸⁶Sr basalts from Malumalu seamount, Samoa; Workman et al., 2004). Data are for un-leached powders; Table 2.
- Fig. 8. La/Sm-Nb/Nb* for Site 1276 sills (lower sill shaded diamonds, upper sill shaded circles. Data are for unleached powders; Table 2), in comparison with various estimates of continental crust (red squares), and the mantle end-members N-MORB, EM1, EM2 and HIMU (yellow triangles): (Continental crust estimates Rudnick and Fountain, 1995, McCulloch and Bennett, 1994, Gao et al., 1998, Plank and Langmuir, 1998. HIMU average of 21 samples from Mangaia and Tubuai Islands: Woodhead, 1996; Hauri and Hart, 1997. EM1 average of 18 samples of Tedside Series, Pitcairn Island: Eisele et al., 2002; Hauri, Hart and Farley, unpublished. EM2 average of 6 samples from Malumalu seamount, Samoa: Workman et al., 2004. N-MORB global average of Su and Langmuir, 2003. Nb/Nb* = (Nb)n/[(Th)n*(La)n]^0.5.

Fig. 9. Zr, Ba and Sr contents of the Site 1276 sills (shaded circles - upper sill; shaded diamonds

– lower sill. Data are for unleached powders; Table 2) in comparison to: N-MORB (filled triangle), Mesozoic Appalachian Tholeiites (MAT, open circle) and other mafic rift-related volcanics from the Newfoundland Seamounts (small open squares) and Grand Banks, Scotia Shelf, NE Seamounts, Georges Bank, Baltimore Canyon, and average lamprophyres from New England and Quebec (shaded squares); Data from Hurtibise et al.,1987; McHone, 1992; Pegram, 1990; Pe-Piper and Jansa, 1987; Pe-Piper et al., 1990, 1992, 1994; Ross, 1992; Sullivan and Keen, 1977.

Fig. 10. Sr, Nd and Pb isotope ratios of Site 1276 sills (upper sill-large shaded circles; lower sill-large shaded diamonds) compared with North Atlantic MORB (small circles; PETDB data base) and basalts from North Atlantic hotspots/plumes (Azores, Canaries, Iceland, Madeira; GEOROC Data Base). Sill data are for leached powders, Table 2, corrected back to initial ratios and forward to present using a mantle evolution model; see text and Table 3.















Figure 6







Figure 9



			Piece 1	Mid-interva Dopth	l Woight			
Site	Core	Section	cm	mbsf	grams	Alteration	Grain Size	e Comments
Upper co	ntact			1612.83	-			
1276	87	6	102-104	1613.14	31	high	fine	up to 5% apatite, abundant carbonate
1276	88	1	83-86	1614.95	48	moderate to high	fine	Moderate clay and carbonate, some apatite
1276	88	2	88-92	1616.02	59	moderate	fine	Some carbonate patches, trace of apatite and biotite
1276	88	3	29-33	1616.87	50	high	fine	Abundant clay, carbonate and apatite
1276	88	3	48-51	1617.06	49	moderate	fine	Some oxide exsolution, little carbonate or apatite.
1276	88	3	76-80	1617.34	54	moderate to low	fine	Clean plagioclase and oxides, 20% clay, rare carbonate; best of core 88
1276	88	4	39-41	1618.46	63	moderate	fine	Clean plagioclase and oxides, 25% clay, some apatite and biotite
1276	88	5	67-71	1620.22	51	moderate	fine	Clean plagioclase and oxides, 25% clay, 3% apatite and some biotite
1276	88	6	48-51	1621.12	57	moderate	aphanitic	Abundant clean cpx and oxides, 25% clay, some carbonate
1276	88	7	81-85	1622.66	51	high	aphanitic	Abundant clay and carbonate, exsolved oxides
Lower co	ntact			1623.00	-			
Upper co	ntact not	t recovere	d	<1715.33	-			
1276	99	1	46-49	1719.88	39	moderate to high	fine	Some oxide exsolution, very altered groundmass
1276	99	2	0-4	1720.85	42	moderate	fine	20% large clean cpx phenocrysts, altered plagioclase, minimal carbonate
1276	99	3	19-22	1722.49	47	moderate to low	medium	35% large clean cpx phenocrysts, minimal carbonate; best of core 99
1276	99	5*	10-13	1724.07	47	moderate	medium	30% large clean cpx phenocrysts, minimal carbonate, altered groundmass
1276	99	5*	30-34	1724.28	61	moderate	medium	25% large clean cpx phenocrysts, minimal carbonate, altered groundmass
1276	99	6*	24-27	1725.42	62	moderate	medium	25% large clean cpx phenocrysts, minimal carbonate, altered groundmass
1276	99	6*	27-30	1725.44	58	moderate	medium	25% large clean cpx phenocrysts, minimal carbonate, altered groundmass
Lower co	ntact no	t recovere	ed	>1725.59	-			
* Section Samples Site 1276	out of o with Cor was dri	rder & pi e and Sec lled to a d	eces may 1 ction in Bo lepth of 17	not be in cor old font were '39m bsf in 4	rect order used for 4560m wa	chemical and isoto ter depth at 45° 24	pic analysis .3198'N, 44	(the two 99-6 samples were combined) ° 47.1496' W.

Table 1 - Leg 210 Site 1276 Sample Inventory and Petrographic Notes

	88-2	88-2L	88-3	88-3L	88-5	88-5L	99-3	99-3L	99-5	99-5L	99-6	99-6L
siO2	46.62		46.43		46.14		45.77		47.67		47.14	
TiO2	3.544		3.478		3.531		3.612		3.465		3.591	
1203	15.33		14.92		14.68		14.87		15.38		15.18	
FeO*	14.65		14.56		14.73		15.42		13.32		14.29	
MnO	0.320		0.347		0.331		0.236		0.201		0.205	
MσO	8.06		7 84		8 30		6.12		4 87		5 1 5	
CaO	5.63		6.38		7.50		8.52		0.14		8.60	
Na2O	3.17		3.46		2 70		3.55		3.77		3.80	
Na20	1.070	0.202	1 075	0.265	1 409	0.202	1.260	0.617	1 6 4 9	0.426	1 402	0 562
N20	1.970	0.382	1.875	0.565	1.408	0.582	1.500	0.017	1.048	0.430	1.495	0.365
P2O5	0.708		0.700		0.672		0.538		0.535		0.550	
Sum Dann Sama	100.00		100.00		100.00		100.00		100.00		100.00	
taw Sum	97.10		97.70		97.01		97.02		96.15		90.44	
CRF Trace Elem	ents (ppm)		20.0		25.0		52.2		28.6		26.2	
N1	20.0		50.9 40.7		55.9		33.3		28.0		30.5	
Cr	32.6		42.7		52.2		45.3		9.0		17.3	
Sc.	27.3		26.9		28.6		26.7		27.2		26.3	
v	360		355		375		408		363		383	
Ba	1887		1345		4855		748		744		678	
Rb	28.8		28.0		21.9		31.6		36.0		32.8	
Sr	387		294		440		558		912		756	
Zr	180		186		176		229		242		240	
Y	30.8		31.7		29.7		32.1		32.2		32.0	
Nb	28.0		28.7		27.6		28.0		29.8		29.9	
Ga	20		22		21		21		22		22	
Cu	36		37		36		10		107		60	
- Zn	138		127		127		102		111		115	
	24.4		27.1		26.1		22.8		27.1		26.0	
Ce	61.6		61.1		52.3		67.2		63.0		63.3	
CP Trace Flor	nents (nnw	n)										
Cs	4.39	0.323	6.46	0.170	1.024	0.657	4.50	0.351	0.313	0.105	0.363	0.097
Rb	26.4	30.9	26.2	20.6	20.5	28.7	30.0	33.7	33.2	36.3	30.5	38.8
Ra	1780	2509	1272	1087	4685	1746	726	745	728	745	655	843
Ба Гь	2 102	1.672	2 269	2,009	2 2 4 9	2 2 2 2	1 507	1 /22	2 152	1 770	1.062	1 940
111	2.195	0.747	2.306	2.098	2.240	2.230	1.307	1.455	2.155	1.779	0.500	1.042
	0.692	0.747	0.715	0.775	0.099	0.879	0.450	0.407	0.015	0.550	0.589	0.005
ND	26.57	18.44	27.18	26.97	26.29	28.26	27.71	28.50	29.69	34.14	30.20	29.07
ľa	1.853	1.251	1.901	1.917	1.809	1.987	1.959	2.255	2.092	2.461	2.097	2.275
La	29.14	9.63	29.18	11.86	27.73	9.90	29.15	18.28	31.92	12.49	30.45	10.06
Ce	62.39	17.46	62.48	22.10	59.43	18.86	62.69	35.83	67.78	25.86	65.18	20.91
Pb	3.683	3.804	3.502	2.810	3.356	3.005	1.177	0.459	1.808	0.465	2.228	0.424
Pr	8.01	2.15	7.98	2.65	7.61	2.31	8.02	4.27	8.55	3.27	8.32	2.70
Nd	36.25	9.67	36.35	11.67	34.50	10.33	35.98	18.09	37.98	14.54	37.43	12.22
Sr	372	356	288	391	440	260	544	278	934	395	775	364
Zr	163.1	167.2	169.5	197.6	159.7	228.1	109.9	94.6	185.9	195.9	175.7	171.4
Hf	4.06	4.67	4.23	5.09	3.98	5.84	2.87	2.98	4.59	4.97	4.31	4.64
Sm	8.74	2.78	8.81	3.21	8.33	2.85	8.61	4.43	8.98	3.89	8.85	3.38
En	3 474	2 277	3 484	2 264	3 331	2.061	2 939	1 828	3 198	1 281	3.028	1 147
Gd	7 98	2.27	8 24	3 10	7 01	2.001	8 02	4 10	8 30	3 76	8 26	3 20
	1 160	0.470	1 210	0.538	1 140	0.405	1 152	0.658	1 216	0.606	1 215	0.549
1.0 Der	6 5 4	0.4/9	6.60	2 20	6.24	2.05	6.20	2 70	6 70	2 5 1	6.65	2.240
	0.54	2.90	0.09	3.28 0.655	0.34	3.05	0.30	5./8 0.714	0.78	3.31	0.00	3.33
ELU 67	1.213	0.380	1.245	0.000	1.180	0.015	1.140	0./14	1.203	0.082	1.233	17.03/
Y	35.30	15.53	37.48	17.89	35.55	17.04	33.17	19.77	37.41	18.94	37.08	17.94
Er	2.944	1.603	3.131	1.718	2.900	1.685	2.655	1.753	3.032	1.710	2.963	1.662
ſm	0.398	0.238	0.407	0.254	0.379	0.246	0.346	0.237	0.390	0.238	0.386	0.237
Yb	2.306	1.593	2.373	1.640	2.256	1.620	1.989	1.445	2.223	1.470	2.178	1.443
Ĺu	0.355	0.253	0.361	0.262	0.336	0.266	0.294	0.222	0.336	0.234	0.332	0.232
Sc	27.4	27.8	27.6	31.2	29.1	26.1	26.9	33.5	27.8	30.5	26.9	27.2
875×1865-	0 706227	0 706100	0 706445	0 705000	0 706122	0 706250	0.705000	0 705000	0 705004	0 706194	0 705702	0 70606
575F/808F 143Nd/144N3	0.706337	0.700100	0.706445	0.703898	0.700133	0.700239	0.705900	0.703899	0.703906	0.700184	0.705792	0.70000
1431NU/1441NG		10.012098		18 4666		18 5007		10.0070		20.2520		0.51273
200PD/204PD		10.48/0		16.4000		16.5096		19.98/9		20.3580		20.5031
20/PD/204Pb		15.5586		15.5536		15.5596		15.6396		15.6576		15.6686
208Pb/204Pb		38.1543		38.1243		38.1363		39.4972		39.5603		39.6913

$\begin{array}{cccccccccccccccccccccccccccccccccccc$.9 105.3 51 28.66 .6 260.4 12 2.855 57 10.33 98 2.238 73 0.879 10 3.005 10 18.51 8 48.7 94 2.631 898 0.70625	99.7 33.66 278.1 4.432 18.09 1.433 0.467 0.459 66.9 212.2 3.173	97.8 36.34 394.7 3.888 14.54 1.779 0.530 0.465 75.4 261.4	95.9 38.82 364.4 3.379 12.22 1.842 0.603 0.424 94.3								
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	51 28.66 .6 260.4 .12 2.855 .67 10.33 .98 2.238 .73 0.879 .10 3.005 .10 18.51 .8 48.7 .4 2.631 .898 0.70625	33.66 278.1 4.432 18.09 1.433 0.467 0.459 66.9 212.2 3.173	36.34 394.7 3.888 14.54 1.779 0.530 0.465 75.4 261.4	38.82 364.4 3.379 12.22 1.842 0.603 0.424 94.3								
.0 390 .2 3.21 .7 11.6 .3 2.09 .7 0.77 .4 2.81 .2 17.4 .5 48.3 .4 2.80 .100 0.7053 .200 0.7055	.6 260.4 12 2.855 57 10.33 98 2.238 73 0.879 10 3.005 10 18.51 8 48.7 14 2.631 898 0.70625	278.1 4.432 18.09 1.433 0.467 0.459 66.9 212.2 3.173	394.7 3.888 14.54 1.779 0.530 0.465 75.4 261.4	364.4 3.379 12.22 1.842 0.603 0.424 94.3								
32 3.21 7 11.6 '3 2.09 7 0.77 '4 2.81 ·2 17.4 '5 48.3 4 2.80 100 0.705 726 0.705	12 2.855 57 10.33 10 2.238 10 3.005 10 3.005 10 18.51 8 48.7 14 2.631 898 0.70625	4.432 18.09 1.433 0.467 0.459 66.9 212.2 3.173	3.888 14.54 1.779 0.530 0.465 75.4 261.4	3.379 12.22 1.842 0.603 0.424 94.3								
7 11.6 7 2.09 7 0.77 14 2.81 12 17.4 15 48.4 4 2.80 100 0.705 126 0.705	57 10.33 58 2.238 73 0.879 10 3.005 40 18.51 8 48.7 94 2.631 898 0.70625	18.09 1.433 0.467 0.459 66.9 212.2 3.173	14.54 1.779 0.530 0.465 75.4 261.4	12.22 1.842 0.603 0.424 94.3								
'3 2.09 47 0.77 14 2.81 12 17.4 15 48.3 4 2.80 100 0.7055 726 0.7055	2.238 2.238 73 0.879 10 3.005 10 18.51 8 48.7 14 2.631 898 0.70625	1.433 0.467 0.459 66.9 212.2 3.173	1.779 0.530 0.465 75.4 261.4	1.842 0.603 0.424 94.3								
47 0.77 14 2.81 2 17.4 25 48.3 4 2.80 100 0.705 726 0.705	73 0.879 10 3.005 10 18.51 8 48.7 04 2.631 898 0.70625	0.467 0.459 66.9 212.2 3.173	0.530 0.465 75.4 261.4	0.603 0.424 94.3								
4 2.81 2 17.4 '5 48.3 4 2.80 100 0.705 726 0.705	10 3.005 10 18.51 8 48.7 14 2.631 898 0.70625	0.459 66.9 212.2 3.173	0.465 75.4 261.4	0.424 94.3								
42 17.4 15 48.3 4 2.80 100 0.7053 726 0.7054	40 18.51 8 48.7)4 2.631 898 0.70625	66.9 212.2 3.173	75.4 261.4	94.3								
75 48. 4 2.80 100 0.705 726 0.705	8 48.7)4 2.631 898 0.70625	212.2	261.4									
4 2.80 100 0.705 726 0.705)4 2.631 898 0.70625	3 173		297.8								
100 0.705	898 0.70625	5.175	3.468	3.157								
726 0 705		59 0.70589 <u>9</u>	Э 0.706184	0.706066								
, 20 0.705	668 0.70578	32 0.705402	2 0.705813	0.705646								
28 0.13	06 0.1350	0.1203	0.1362	0.1297								
924 0.705	865 0.70598	34 0.705573	3 0.706003	0.705822								
598 0.512	606 0.51261	0.512740	0.512728	0.512731								
479 0.512	491 0.51249	0.512643	3 0.512625	0.512626								
41 0.19	51 0.1956	5 0.2080	0.2064	0.2065								
512 0.512	626 0.51263	32 0.512779	9 0.512757	0.512756								
18.46	66 18.509	6 19.9879	20.3580	20.5031								
86 15.55	36 15.559	6 15.6396	15.6576	15.6686								
43 38.12	243 38.136	3 39.4972	39.5603	39.6913								
41 18.17	18.204	8 18.9454	19.2056	19.0894								
88 15.53	15.544	9 15.5896	15.6023	15.6009								
50 37.86	581 37.8820	0 38.4478	38.2922	38.2752								
4 9.7	2 9.81	12.06	12.85	12.50								
.0 34.7	70 34.90	41.00	39.32	39.12								
05 10.00	18.366	3 19.1334	19.4021	19.2767								
65 18.33	15.552	7 15.5986	15.6118	15.6099								
67 15.54	38.064	3 38.6505	38.4829	38.4612								
85 18.32 67 15.54 30 38.05	* 87Sr/86Sr initial, unleached 0.706031 0.706049 0.705931 0.705674 0.705763 0.705637											
18.32 67 15.54 30 38.05 031 0.706	* 87Sr/86Sr initial, unleached 0.706031 0.706049 0.705931 0.705674 0.705763 0.705637											
	18.53 18.53 167 15.54 130 38.05	67 15.5475 15.552 30 38.0503 38.064 031 0.706049 0.70593	67 15.5475 15.5527 15.5986 30 38.0503 38.0643 38.6505 031 0.706049 0.705931 0.705674 87Sr/86Sr, Table 2) 2	30 18.3394 18.3003 19.1334 19.4021 67 15.5475 15.5527 15.5986 15.6118 30 38.0503 38.0643 38.6505 38.4829 031 0.706049 0.705931 0.705674 0.705763 87Sr/86Sr, Table 2) 2 3 3 3								

Table 3. Sr, Nd and Pb isotope initial and model present day ratios

Table 1, Supplementary Material: Compositions of Clinopyroxene Phenocrysts and Micro-phenocrysts in one Upper and one Lower Sill sample

				Normali	zed Comp	ositions				Normalized	Total as	Mg
Sample Number	\underline{SiO}_2	$\underline{\text{TiO}}_2$	$\underline{Al_2O_3}$	$\underline{Cr_2O_3}$	<u>FeO</u>	<u>MnO</u>	<u>MgO</u>	<u>CaO</u>	<u>Na₂O</u>	Total	Measured	Number
88-5, 67-71	48.44	2.113	4.60	nd	8.79	0.203	13.73	21.76	0.389	100	100.25	0.736
88-5, 67-71	48.54	1.975	4.58	nd	8.82	0.167	14.16	21.38	0.371	100	100.61	0.741
88-5, 67-71	47.94	2.336	4.91	nd	9.31	0.223	13.59	21.29	0.413	100	100.84	0.722
88-5, 67-71	49.64	1.549	3.50	nd	8.24	0.166	14.67	21.82	0.400	100	100.56	0.760
88-5, 67-71	48.66	1.946	4.47	nd	8.77	0.178	14.07	21.62	0.300	100	100.50	0.741
99-6, 24-27	48.50	2.057	4.08	0.0045	9.62	0.187	13.80	21.37	0.377	100	100.95	0.719
99-6, 24-27	48.30	2.334	4.37	nd	9.39	0.201	13.51	21.55	0.361	100	100.09	0.719
99-6, 24-27	49.26	1.902	3.50	nd	10.00	0.285	13.62	21.00	0.421	100	101.19	0.708
99-6, 24-27	49.83	1.607	3.29	nd	9.02	0.221	14.39	21.27	0.364	100	100.78	0.740
99-6, 24-27	48.54	2.129	4.18	nd	9.60	0.240	13.84	21.06	0.417	100	100.42	0.720
99-6, 24-27	51.12	1.203	2.22	nd	9.29	0.228	15.35	20.30	0.288	100	100.84	0.747
99-6, 24-27	48.48	2.156	4.34	nd	9.72	0.218	13.93	20.70	0.453	100	100.85	0.719
Average 88-5	48.64	1.984	4.41	nd	8.79	0.188	14.04	21.57	0.375	100	100.55	0.740
Average 99-6	49.15	1.913	3.71	nd	9.52	0.226	14.06	21.04	0.383	100	100.73	0.725

Sample 88	-2, 88-92 cn	n									
Increme Heating	ental J		36Ar(a)	37Ar(ca)	38Ar(cl)	39Ar(k)	40Ar(r)	Age ±2σ (Ma)	40Ar(r) (%)	39Ar(k) (%)	K/Ca ±2σ
050676	500 °C		0.00649	0.00414	0.00254	0 00790	E 4504	0 62.06 1.2.54	41.06	E 94	0 222 + 0 028
050676	500°C		0.02046	0.29414	0.00254	0.22702	5.4004	02.00 ± 2.04	41.00	3.04	0.333 ± 0.020
050677	700 °C		0.00000	0.10131	0.00108	0.13727	0.5200	$0 \qquad 93.07 \pm 1.32$	71.75	5.52	0.563 ± 0.047
050678	700°C		0.00878	0.21556	0.00112	0.25565	10 0570	$17 95.00 \pm 2.30$	70.00	0.50	0.310 ± 0.047
050679	800 °C	,	0.00768	0.37878	0.00077	0.35532	10.0073	3 100.04 ± 1.05	02.01	9.11	0.403 ± 0.034
050680	875 °C		0.00460	0.36499	0.00090	0.44414	10.1307	104.03 ± 0.93	93.01	16.50	0.523 ± 0.040
050601	950 C		0.00308	0.22022	0.00103	0.04710	20.0300	105.42 ± 0.00	90.07	17.15	1.230 ± 0.100
050662	11025 C		0.00400	0.19301	0.00220	0.00925	27.4442	$\begin{array}{cccc} 4 & 105.04 \pm 0.07 \\ 4 & 104.04 \pm 1.11 \end{array}$	95.79	17.15	1.460 ± 0.135
050663	1175 °C		0.00320	0.24505	0.00119	0.54194	22.0000	$14 104.04 \pm 1.11$	95.60	14.04	0.951 ± 0.065
050664	1175 C	*	0.00200	0.99952	0.00119	0.55542	22.0900	104.22 ± 0.02	90.09	14.24	0.239 ± 0.017
050665	1400 °C	~	0.00031	0.49725	0.00016	0.06721	2.7434	104.58 ± 0.73	90.72	1.72	0.058 ± 0.004
		Σ	0.06746	3.51646	0.01278	3.90138	153.4096	5			
Informa on Ana	ation lysis			Results		40(r)/39(k)	±2σ	Age ±2σ (Ma)	MSWD	39Ar(k) (%,n)	K/Ca $\pm 2\sigma$
Sample Material	88-2, 88-9 whole rc)2cm ock	-	Weighted Plateau		40.8821 ± 0.1690 ± 0.41%		104.73 ± 1.66 ± 1.59%	1.93	74.98 6	0.078 ± 0.088
Location Analyst	ODP210- ⁻ jh	1276						External Error ± 2.35 Analytical Error ± 0.42	2.57 1.3900	Statistical T ration	on
Project Irradiation	ODP 2 ⁻ OSU3F	10 04		Total Fus	ion Age	39.3219	± 0.1432 ± 0.36%	100.84 ± 1.59 ± 1.58%		10	0.477 ± 0.014
J-value	0.00146	62						External Error ± 2.26			

Sample 88-3, 76-80 cm

28.03

Standard

Incremental Heating		36Ar(a)	37Ar(ca)	38Ar(cl)	39Ar(k)	40Ar(r)	Age ±2σ (Ma)	40Ar(r) (%)	39Ar(k) (%)	K/Ca ±2σ
	0	0.04004	0.40400	0.00400	0.00744	0.00.100	54.00 4.00	00.05	7.50	0.055 0.040
05C646 500 °C	6	0.04004	0.48433	0.00426	0.28714	6.08432	54.26 ± 1.68	33.95	7.50	0.255 ± 0.018
05C647 600 °C	С	0.00801	0.13358	0.00120	0.11910	3.71017	79.22 ± 2.77	61.04	3.13	0.383 ± 0.033
05C648 700 °	С	0.00234	0.04804	0.00036	0.05276	1.88242	90.44 ± 1.22	73.16	1.39	0.472 ± 0.035
05C649 800 °C	С	0.00751	0.47835	0.00143	0.43509	17.50671	101.68 ± 1.50	88.73	11.45	0.391 ± 0.029
05C650 875 °C	с 🗸	0.00319	0.18665	0.00117	0.34873	14.55655	105.38 ± 1.30	93.90	9.18	0.803 ± 0.062
05C651 950 °C	с 🗸	0.00263	0.15669	0.00214	0.57489	24.19418	106.22 ± 0.82	96.87	15.13	1.578 ± 0.128
05C652 1025 °C	с 🗸	0.00290	0.18016	0.00253	0.67741	28.52114	106.26 ± 0.65	97.06	17.83	1.617 ± 0.142
05C653 1100 °	с 🗸	0.00205	0.28191	0.00221	0.60329	25.28643	105.80 ± 1.07	97.64	15.88	0.920 ± 0.065
05C654 1175 °	с 🗸	0.00234	1.55100	0.00221	0.67321	28.20720	105.76 ± 0.60	97.58	17.72	0.187 ± 0.013
05C655 1250 °C	с 🗸	0.00021	0.29844	0.00011	0.02598	1.08689	105.60 ± 1.89	94.46	0.68	0.037 ± 0.003
05C656 1400 °	С	0.00021	0.11816	0.00000	0.00186	0.03820	52.66 ± 21.41	38.08	0.05	0.007 ± 0.001
	Σ	0.07144	3.91732	0.01764	3.79947	151.07419				

Analytical Error ± 0.36

In or	formation n Analysis	Res	sults	$40(r)/39(k)~\pm 2\sigma$	Age ±2σ (Ma)	MSWD	39Ar(k) (%,n)	K/Ca ±2σ
Sam Mate	ple 88-3, 76-80cm rial whole rock	Wei	ghted Plateau	41.9776 ± 0.1408 ± 0.34%	105.95 ± 1.78 ± 1.68%	0.55	76.42 6	0.048 ± 0.064
Loca Anal	tion ODP210-1276 yst jh				External Error ± 2.45 Analytical Error ± 0.35	2.57 1.0000	Statistical T ratio Error Magnificatio	on
Proje Irrad	ect ODP 210 iation OSU3F04	Tota	al Fusion Age	39.7619 ± 0.1495 ± 0.38%	100.52 ± 1.70 ± 1.69%		11	0.417 ± 0.014
J-val Stan	ue 0.001441 dard 28.03				External Error ± 2.33 Analytical Error ± 0.37			

Sample 99-3, 19-22 cm

Increme Heating	ental		36Ar(a)	37Ar(ca)	38Ar(cl)	39Ar(k)	40Ar(r)	Age ±2σ (Ma)	40Ar(r) (%)	39Ar(k) (%)	K/Ca ±2σ
05C666 05C667 05C668 05C669 05C670 05C671 05C672 05C673 05C674 05C675	500 °C 600 °C 700 °C 800 °C 875 °C 950 °C 1025 °C 1100 °C 1175 °C 1400 °C	\$ \$ \$ \$	0.01984 0.00567 0.00332 0.00339 0.00166 0.00446 0.00322 0.00127 0.00287 0.00101	0.50694 0.15032 0.18517 0.37800 0.16580 0.26913 0.30632 0.33265 1.69468 0.97722	0.01851 0.00599 0.00318 0.00322 0.00266 0.01137 0.00918 0.00251 0.00432 0.00171	0.22850 0.12399 0.16058 0.20696 0.19000 0.39496 0.43144 0.23933 0.17060 0.03777	3.6482 3.4511 6.3914 8.2839 7.6042 15.9564 17.2343 9.2184 6.4993 1.4162	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	38.35 67.29 86.66 89.19 93.90 92.35 94.75 96.06 88.43 82.64	10.46 5.68 7.35 9.48 8.70 18.08 19.75 10.96 7.81 1.73	$\begin{array}{l} 0.194 \pm 0.015 \\ 0.355 \pm 0.027 \\ 0.373 \pm 0.029 \\ 0.235 \pm 0.017 \\ 0.493 \pm 0.036 \\ 0.631 \pm 0.053 \\ 0.606 \pm 0.049 \\ 0.309 \pm 0.022 \\ 0.043 \pm 0.003 \\ 0.017 \pm 0.001 \end{array}$
Σ Information on Analysis		0.04672	4.96622 Results	0.06267	2.18413 40(r)/39(k)	79.7038 ±2σ	Age ±2σ (Ma)	MSWD	39Ar (k) (%,n)	K/Ca ±2σ	
Sample Material Location Analyst	99-3, 19-2 whole rc ODP210- ⁻ jh	22cm ock 1276		Weighted	Plateau	40.0739	± 0.2097 ± 0.52%	99.71 ± 1.84 ± 1.84% External Error ± 2.43 Analytical Error ± 0.51	0.75 2.78 1.0000	63.36 5 Statistical T ratio Error Magnificatic	0.345 ± 0.138 n
Project Irradiation J-value Standard	ODP 2 ⁻ OSU3F 0.0014 ⁻ 28.03	10 04 18		Total Fus	ion Age	36.4923	± 0.1753 ± 0.48%	91.02 ± 1.67 ± 1.84% External Error ± 2.21 Analytical Error ± 0.43		10	0.189 ± 0.006

Sample 99-6, 24-30 cm

Increme Heating	ental		36Ar(a)	37Ar(ca)	38Ar(cl)	39Ar(k)	40Ar(r)	Age ±2σ (Ma)	40Ar(r) (%)	39Ar(k) (%)	K/Ca $\pm 2\sigma$
05C657 05C658 05C659 05C660 05C661 05C662 05C663 05C663 05C664 05C665	500 °C 700 °C 800 °C 875 °C 950 °C 1025 °C 1100 °C 1175 °C 1400 °C	555	0.00034 0.03756 0.00758 0.00322 0.00714 0.00950 0.01240 0.02104 0.00569	0.00621 0.52465 0.25672 0.16173 0.20625 0.20076 0.44426 2.34242 0.98991	0.00003 0.03713 0.00893 0.00361 0.01016 0.00448 0.00520 0.00643 0.00199	0.00134 0.36419 0.16394 0.14055 0.32380 0.76963 0.60422 0.37499 0.05071	0.0095 9.9279 6.1352 5.5820 12.7349 30.0881 23.2857 13.4634 1.8222	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	8.69 47.21 73.23 85.41 85.77 91.45 86.38 68.40 51.98	0.05 13.04 5.87 5.03 11.59 27.55 21.63 13.42 1.82	$\begin{array}{c} 0.093 \pm 0.012 \\ 0.298 \pm 0.024 \\ 0.275 \pm 0.020 \\ 0.374 \pm 0.028 \\ 0.675 \pm 0.054 \\ 1.648 \pm 0.144 \\ 0.585 \pm 0.043 \\ 0.069 \pm 0.005 \\ 0.022 \pm 0.002 \end{array}$
Informa	tion	Σ	0.10448	5.13290	0.07795	2.79338	103.0492	Age $\pm 2\sigma$		39Ar(k)	
on Anal Sample Material	ysis 99-6, 24-3 whole rc	0cm ock		Weighted	Plateau	39.0069	± 20 ± 0.3500 ± 0.90%	(Ma) 95.86 ±2.00 ±2.08%	2.83	(%,n) 65.81 4	0.499 ± 0.242
Location Analyst	ODP210- jh	1276						External Error ± 2.51 Analytical Error ± 0.84	3.18 1.6828	Statistical T ratio Error Magnificati	on
Project Irradiation J-value Standard	ODP 2 ⁻ OSU3F 0.00138 28.03	10 04 99		Total Fus	ion Age	36.8906	± 0.1958 ± 0.53%	90.79 ± 1.78 ± 1.96% External Error ± 2.29 Analytical Error ± 0.47		9	0.234 ± 0.009