

Leaf Respiratory CO<sub>2</sub> Is <sup>13</sup> C-Enriched Relative to Leaf Organic Components in Five Species of C<sub>3</sub> Plants Author(s): Cheng-yuan Xu, Guang-hui Lin, Kevin L. Griffin and Raymond N. Sambrotto Source: *New Phytologist*, Vol. 163, No. 3 (Sep., 2004), pp. 499-505 Published by: <u>Wiley</u> on behalf of the <u>New Phytologist Trust</u> Stable URL: <u>http://www.jstor.org/stable/1514453</u> Accessed: 06/11/2014 03:04

New Phytologist

Your use of the JSTOR archive indicates your acceptance of the Terms & Conditions of Use, available at http://www.jstor.org/page/info/about/policies/terms.jsp

JSTOR is a not-for-profit service that helps scholars, researchers, and students discover, use, and build upon a wide range of content in a trusted digital archive. We use information technology and tools to increase productivity and facilitate new forms of scholarship. For more information about JSTOR, please contact support@jstor.org.



Wiley and New Phytologist Trust are collaborating with JSTOR to digitize, preserve and extend access to New Phytologist.

http://www.jstor.org

Author for correspondence:

Email: xucy@ldeo.columbia.edu

Key words: C<sub>3</sub> plants, carbon isotope,

dark respiration, fractionation, Keeling

Cheng-yuan Xu

plot.

Tel: +(845) 3658995

Fax: +(845) 3658150

Received: 18 May 2004

Accepted: 14 June 2004

Research

# Rapid report

Leaf respiratory  $CO_2$  is <sup>13</sup>C-enriched relative to leaf organic components in five species of  $C_3$  plants

Cheng-yuan Xu<sup>1,2</sup>, Guang-hui Lin<sup>2,3\*</sup>, Kevin L. Griffin<sup>1</sup> and Raymond N. Sambrotto<sup>1</sup>

<sup>1</sup>Lamont Doherty Earth Observatory, Department of Earth and Environmental Sciences, Columbia University, Palisades, NY 10964, USA, <sup>2</sup>Biosphere 2 Center, Columbia University. Oracle, AZ 85623, USA, <sup>3</sup>Laboratory of Quantitative Vegetation Ecology and Research Center of Plant Ecology and Biodiversity Conservation, Institute of Botany, The Chinese Academy of Sciences, Xiangshan, Beijing 100093, China \*Present address: Department of Global Ecology, Carnegie Institutution of Washington, 260 Panama St., Stanford, CA 94305–1297, USA

## Summary

• Here, we compared the carbon isotope ratios of leaf respiratory  $CO_2$  ( $\delta^{13}C_R$ ) and leaf organic components (soluble sugar, water soluble fraction, starch, protein and bulk organic matter) in five  $C_3$  plants grown in a glasshouse and inside Biosphere 2. One species, *Populus deltoides*, was grown under three different  $CO_2$  concentrations. • The Keeling plot approach was applied to the leaf scale to measure leaf  $\delta^{13}C_R$  and

these results were compared with the  $\delta^{13}$ C of leaf organic components.

• In all cases, leaf respiratory  $CO_2$  was more <sup>13</sup>C-enriched than leaf organic components. The amount of <sup>13</sup>C enrichment displayed a significant species-specific pattern, but the effect of  $CO_2$  treatment was not significant on *P. deltoides*.

• In C<sub>3</sub> plant leaves, <sup>13</sup>C-enriched respiratory CO<sub>2</sub> appears widespread. Among currently hypothesized mechanisms contributing to this phenomenon, non-statistical carbon isotope distribution within the sugar substrates seems most likely. However, caution should be taken when attempting to predict the  $\delta^{13}$ C of leaf respiratory CO<sub>2</sub> at the ecosystem scale by upscaling the relationship between leaf  $\delta^{13}$ C<sub>R</sub> and  $\delta^{13}$ C of leaf organic components.

New Phytologist (2004) 163: 499-505

© New Phytologist (2004) doi: 10.1111/j.1469-8137.2004.01153.x

# Introduction

It is well known that carbon isotope discrimination takes place during plant photosynthetic CO<sub>2</sub> fixation, resulting in all higher plants being depleted in <sup>13</sup>C in organic carbon relative to atmospheric CO<sub>2</sub>. The models of <sup>13</sup>C fractionation in photosynthesis have been well established (Farquhar *et al.*, 1982). By contrast, studies on the carbon isotope ratio of CO<sub>2</sub> generated by dark respiration ( $\delta^{13}C_R$ ) are limited. Although a possible isotope effect during dark respiration might significantly influence the carbon isotope signature of plants and other components of an ecosystem, fewer studies have focused on determining the magnitude of this potential effect and the results appear contradictory (O'Leary, 1981; Lin & Ehleringer, 1997; Duranceau *et al.*, 1999, 2001).

Studies on the carbon isotopic effects during respiration trace back half a century. Historically, carbon isotope discrimination during respiration was considered to be negeligible

www.newphytologist.org

(O'Leary, 1981; Farquhar *et al.*, 1982, 1989; Flanagan & Ehleringer, 1998). Early experimental studies observed that  $\delta^{13}C_R$  is very close (approx.  $\pm 1$ ‰) to bulk carbon in some geminating crop seedlings (Baertschi, 1953; Smith, 1971). More recently (Lin & Ehleringer, 1997) cultured mesophyll protoplasts of bean and corn leaves with carbohydrates of known isotopic ratios as the only carbon source and found no significant differences between  $\delta^{13}C_R$  and  $\delta^{13}C$  of the substrate in either species.

Still, other studies suggest that respiratory CO<sub>2</sub> of plants can be remarkably <sup>13</sup>C-enriched or <sup>13</sup>C depleted (4–5‰ more positive or -4-8% more negative) in comparison with whole plant or leaf  $\delta^{13}$ C (Smith, 1971; Troughton *et al.*, 1974). Recently, Duranceau et al. (1999, 2001) and Ghashghaie et al. (2001) compared  $\delta^{13}C$  of leaf respiration and leaf organic components in beans, tobacco, and sunflower. They report a 3-6<sup>13</sup>C-enrichment in respiratory CO<sub>2</sub> compared to sucrose, the assumed substrate of dark respiration. Although this hypothesis was not tested in other species, they concluded that carbon isotope fractionation during dark respiration was widespread in C3 plants. In C4 plants, Henderson *et al.* (1992) found that the  $\delta^{13}$ C of dry matter was more negative than that predicted by the discrimination occurring during CO<sub>2</sub> uptake and partly attribute the difference to an isotope effect during dark respiration. Furthermore,  $\delta^{13}C_{R}$  can change daily or seasonally (Park & Epstein, 1961; Jacobson et al., 1970; Damesin & Lelarge, 2003) and can be influenced by environmental or physiological factors (temperature, respiratory quotient, etc., Tcherkez et al., 2003).

Despite the growing contradictory evidence, the assumption that carbon fractionation in dark respiration is negligible is widely applied in ecological and physiological studies (Flanagan & Ehleringer, 1998; Yakir & Sternberg, 2000; Ehleringer et al., 2002). At the ecosystem scale, the concept of ecosystem <sup>13</sup>C discrimination ( $\Delta^{13}C_e = (\delta^{13}C_{trop} - \delta^{13}C_R)/(1 + \delta^{13}C_R)$ , or  $\Delta^{13}C_e = (\delta^{13}C_{atm} - \delta^{13}C_R)$ ), has recently been used to partition NEE (Net Ecosystme Exchange) into photosynthesynthetic and respiratory components (Bowling et al., 2001), by assuming that the  $\delta^{13}C_R$  should reflect the <sup>13</sup>C signature of total organic carbon in the ecosystem (Buchmann et al., 1997; Yakir & Sternberg, 2000). Likewise, the  $\delta^{13}$ C of organic carbon in leaf, soil, and litter were used to estimate the  $\delta^{13}C_{R}$  generated by each components (Lin *et al.*, 1999, 2001). If the  $\delta^{13}C_{\rm R}$  does not correctly reflect the  $\delta^{13}C$ of the respiration substrate pool, the conclusions of these studies will need to be reconsidered and modified accordingly.

Additional uncertainties regarding the use of  $\delta^{13}C_R$  as a tool for understanding ecosystem scale processess arise from several other factors. For example, initial studies focused on seedlings or tubers and subsequent leaf scale studies were conducted in only a few crop species. In addition, plant materials were subjected to a CO<sub>2</sub>-free environment in all previous studies, which may itself influence leaf  $\delta^{13}C_R$  (O'Leary, 1981). Clearly, much more detailed information on the species effects, and ultimately the mechanisms are needed to gain insight into ecosystem level processes. In this study, we applied a Keeling plot approach on the leaf scale to measure leaf  $\delta^{13}C_R$  (Fessenden & Ehleringer, 2003) and studied its relationship with  $\delta^{13}C$  of leaf soluble sugar, water soluble fraction, starch, protein, and bulk organic carbon of five C<sub>3</sub> plants. Our primary goal was to test the hypothesis (1) that  $\delta^{13}C_R$  will be the same as major substrates (assumed to be soluble sugar), an assumption that underlies many current ecological studies, neglecting the carbon isotope effect in respiration. If hypothesis (1) is rejected, we further hypothesize (2) the differences between  $\delta^{13}C_R$  and  $\delta^{13}C$  leaf organic components is not species-specific and (3) will not be significantly influenced by growth CO<sub>2</sub> level (in *Populus deltoides*).

## Materials and Methods

#### Plant materials

We studied five C3 plants, which are among the most abundant species in the Tropical Rain Forest (TRF) and Intensive Forest Biome (IFB) of Biosphere 2 (a 1.29 hectare glass enclosed research facility in Oracle, AZ, USA). This leaf scale study was also designed to provide background information for further investigation on the respiratory isotope effect at mesocosom scale within Biosphere 2. Among 4 tropical species we studied, Musa paradisiacal (tree), Coffea arabica (shrub), and Epipremumn pinnatum (vine) were grown in a glasshouse (the demonstrate lab or DL), while Clitoria racemosa (tree), was grown in the TRF biome. Populus deltoides (IFB monoculture), a temperate tree species, was grown in three  $CO_2$  concentrations: close to ambient in the DL, 800 p.p.m. in the IFB mid bay (MB) and 1200 p.p.m. in west bay (WB). In the IFB, tank supplied CO<sub>2</sub> with a very low  $\delta^{13}$ C (c. -28‰) was used to maintain elevated CO<sub>2</sub> concentrations. All plants grew under a natural photoperiod and night time temperatures of 23-28°C depending on the location.

#### Air sampling

All air samples were collected between 20:00 h to 00:30 h, when plants were in natural darkness. One to several healthy, intact, visually mature (well expanded and with developed cuticle) leaves were sealed in an opaque respiration chamber (modified from a mylar balloon) with a small fan to ensure air mixing. The chamber was connected to a closed loop gas exchange system including a pump, a CO<sub>2</sub> infrared gas analyzer (LI-6200, Licor, Inc., Lincoln, NE, USA), a desiccant tube containing magnesium perchlorate and six 100 mL flasks (Fig. 1a). The entire system was 10–15 L to hold the largest leaf (*Epipremumn pinnatum*) in our study and was checked for leakage before each sampling by exhaling on all connections. The airflow rate was approx. 1 L<sup>-1</sup> per min. Ambient air was



**Fig. 1** Leaf scale Keeling plot apparatus and result. (a) Diagram of sampling apparatus used in this study. (b) The result of one *Musa paradisiacal* leaf ( $R^2 = 0.9986y = 5822.9x - 23.76$ , geometric mean regression). The scale of x-axis (0.0023 - 0.0018) is equivalent to  $[CO_2]$  of 435–555 p.p.m. In all measurements, R2 is > 0.95 and in most cases close to 0.99.

pumped through the entire system before closure and then allowed to circulate for 10-15 min to ensure adequate mixing before sampling. The air samples were collected in sequence by closing both stopcocks on a flask for each 15-20 p.p.m.  $CO_2$  increment. Humidity was not strictly controled in our study.

We compared our leaf-level Keeling plot method with a traditional CO<sub>2</sub> free chamber connected to the Li-6200 photosynthesis system and found that two methods yielded similar results in leaf  $\delta^{13}C_R$  (±0.5%) when the incubation chamber was well sealed. However, the incubation method often gave much more scatter results with same plant leaf than leaf–level Keeling plot approach.

#### Leaf sampling and chemical extraction

After air sampling, half of the leaf material contained within the cuvette was immediately frozen in liquid nitrogen and then stored in  $-20^{\circ}$ C freezer for subsequent extraction of carbohydrate and protein. The remaining leaf material was dried in a 60°C oven for carbon isotope analysis of bulk leaf organic matter.

A subsample of 0.1–1 g of leaf material was used for soluble sugar and starch extraction. For each 0.1 g of sample, 1 mL of deionized water was added and the mixture was ground in a chilled mortar and pestle. The resulting extract was kept at 0°C for 20 min before centrifugation at 12 000 g for 10 min. The supernatant containing the soluble fraction was then boiled for 3 min and centrifuged again as already described (Duranceau et al., 1999). The water soluble fraction was then mixed with Dowex-50 (H<sup>+</sup>) and Dowex-1 (Cl<sup>-</sup>) resins in sequence to remove amino acids and organic acids, respectively. The eluate has been showen to have a carbon isotope composition representative of leaf soluble sugars (Brugnoli et al., 1988). The pellets were washed in ethanol (80% v:v) at 80°C to eliminate chlorophyll and then suspended twice in 6 mol/L HCl at 5°C (1 h each) to solubilize the starch. After adding methanol (4× by volume), the supernanent was kept at 5°C overnight and starch precipited was desiccated in a freeze dryer (Damesin & Lelarge, 2003, with a few modifications). Proteins were extracted by boiling the supernanent of grounded leaf tissue (in 2% NaCl; 10 000 g 15 min) for 30 min (Jacobson et al., 1970). The precipitant was dried overnight in a desiccator at room temperature. All products from these extractions were kept at -20°C until carbon isotope analyses were performed. According to the references mentioned above, fractionation of carbon isotopes did not occur during the extraction processes.

#### Carbon isotope analysis

The carbon isotope ratios in delta notation were expressed as  $\delta^{13}C$  (‰) = [R<sub>sample</sub>/R<sub>standard</sub> - 1] × 1000, where R is the molar ratio of  $^{13}C/^{12}C$ .  $\delta^{13}C_R$  was measured in an Isochrom isotoperatio mass spectrometer (Fison Instrument Inc., Manchester, UK) at Biosphere 2 Center (B2C).  $\delta^{13}C$  of the leaf organic components was analyzed either at B2C or with an Europa 20/20 continuous flow (CF) isotope ratio mass spectrometer (IRMS) coupled with an ANCA NT combustion system at Lamont-Doherty Earth Observatory (PDZ-Europa, Cheshire, UK). NIST sucrose was used as the standard for intermachine calibration. All  $\delta^{13}C$  values are expressed relative to Pee Dee Belemnite (PDB).

The mixing model of Keeling (1958, 1961) was used to calculate the isotope ratio of  $CO_2$  respired by a leaf:

$$\delta^{13}C_{cham} = [CO_{2 atm}] \times (\delta^{13}C_{atm} - \delta^{13}C_R) / [CO_{2 cham}] + \delta^{13}C_R$$

where  $[CO_2]$  is the concentration of  $CO_2$  and  $\delta$  is the stable isotope ratio of  $CO_2$ . The subscripts cham, atm and R represent the air within the chamber, the air in experimental atmosphere and respiratory  $CO_2$ , respectively. Geometric mean regressions were used to establish the linear relationship between  $\delta^{13}C_{cham}$  and  $1/[CO_{2 cham}]$  (Pataki *et al.*, 2003) and the intercept at the Yaxis is the  $\delta^{13}C$  value of leaf respiratory  $CO_2$  (Fig. 1b).

#### Statistical analysis

A one-way analysis of variance (anova) was used to test the species effects on the possible differences of  $\delta^{13}C$  between the respiratory CO<sub>2</sub> and leaf organic components. Effects were

New Phytologist (2004) 163: 499-505

<sup>©</sup> New Phytologist (2004) www.newphytologist.org

considered to be significant at the 0.05 probability level. In addition, A student's *t*-test was used for multiple comparisons among *P. deltioides* grown in three  $CO_2$  concentrations to evaluate the effects of  $CO_2$  treatments.

Linear regressions were used to analyze the relationship between  $\delta^{13}C_R$  and  $\delta^{13}C$  of the leaf organic components of all samples or averages of each species/treatment combination. Data from similar leaf-level studies (Duranceau *et al.*, 1999, 2001; Ghashghaie *et al.*, 2001; Tcherkez *et al.*, 2003), which had three  $C_3$  crop species in three sets of environmental or genetic treatments, were included in the regression. We assume that the average  $\delta^{13}C$  of all sugars and water soluble materials (soluble sugar and organic acids) analyzed in those studies are equivalent to the 'soluble sugar' and 'water soluble fraction' in our study.

## Results

The isotopic signatures of the measured pools exhibited a similar pattern for all five species and the three CO<sub>2</sub> treatments for *P. Deltoides*. In each case, leaf  $\delta^{13}C_R$  was the most positive, followed by the  $\delta^{13}C$  of starch (except *M. paradisiacal*), while the  $\delta^{13}C$  of the bulk organic matter was the lightest (Fig. 2). The amount of <sup>13</sup>C-enrichment in leaf  $\delta^{13}C_R$  was 3.5–5.9‰ relative to soluble sugar (Table 1), the assumed major substrate for dark respiration. Compared with the water soluble fraction, starch, and bulk organic matter, the amount of <sup>13</sup>C enrichment in respiratory CO<sub>2</sub> was 2.7–5.2‰, 1.4–4.2‰, and 4.1–6.9‰, respectively, depending on species (Table 1). During the sampling period, leaves released < 0.001 g carbon, which should not



Fig. 2 Leaf  $\delta^{13}C_R$  and leaf organic components in five species; (a) *Epipremumn pinnatu* (b) *Musa paradisiacal* (c) *Populus deltoides* Bartr (in 3 growth [CO<sub>2</sub>]) (d) Coffea *arabica* (e) *Clitoria racemosa*. Values are mean ± SEM among leaves (n = 3-6).

New Phytologist (2004) 163: 499–505

www.newphytologist.org © New Phytologist (2004)

# New Phytologist

Species	Soluble Sugar	Water Soluble Fraction	Starch	Protein	Organic Matter
P. deltoides Bartr. (Ambient CO <sub>2</sub> )	3.83 ± 0.37	3.81 ± 0.52	2.32 ± 0.37	3.30 ± 0.98	4.27 ± 0.53
P. deltoides Bartr. (800 p.p.m.)	5.24 ± 1.68	4.90 ± 1.73	4.39 ± 1.37	-	6.46 ± 1.73
P. deltoides Bartr. (1200 p.p.m.)	4.62 ± 0.86	4.25 ± 1.24	3.70 ± 1.80	-	6.83 ± 1.8
M. paradisiaca	5.86 ± 0.16	3.71 ± 0.29	$4.06 \pm 0.46$	-	5.55 ± 0.47
C. arabica	5.12 ± 0.67	5.11 ± 0.68	2.77 ± 0.52	-	$6.14 \pm 0.57$
E. pinnatum	3.74 ± 0.51	2.67 ± 0.16	1.91 ± 0.83	$2.88 \pm 0.33$	$4.05 \pm 0.55$
C. racemosa	$3.40 \pm 0.39$	3.49 ± 0.46	1.28 ± 0.24	- ',	$5.16 \pm 0.36$
ANOVA of Species Effect	P = 0.008	<i>P</i> = 0.049	P = 0.020	-	P = 0.066

**Table 1** The amount of <sup>13</sup>C enrichment in respiratory CO<sub>2</sub> (‰) in comparison with leaf organic components, shown by mean  $\pm$  SEM (n = 3-6).

One way ANOVA results of the species effect of ambient grown plants are listed in the last row (P < 0.05 are considered significant). In *P. deltoides*, <sup>13</sup>C enrichments in respiratory CO<sub>2</sub>, relative to all organic components are not significantly influenced by the growth [CO<sub>2</sub>] (multicomparison with student's *t*-tests, P = 0.10-0.87).



Fig. 3 Correlation between leaf  $\delta^{13}C_R(y)$  and  $\delta^{13}C$  of leaf organic components (a, soluble sugar; b, water soluble fraction; c, starch; d, organic matter). Data from previous similar studies at leaf scale are also included.  $R^2$  of the regression line (in dashes) and the mean of <sup>13</sup>C-enrichment are shown. For all organic components compared, the slopes of linear regression are close to 1 (solid line, P = 0.08–0.96). Each point represents a data set from a single species or treatment, shown as mean ± SEM among leaves.

influence the  $\delta^{13}$ C of remaining organic component pools significantly.

The amount of <sup>13</sup>C enrichment in respiratory CO<sub>2</sub> relative to soluble sugar, water soluble fraction and starch each had a significant species effect (Table 1). This effect was apparent but not so significant in the bulk leaf organic matter (P = 0.066). The effect of CO<sub>2</sub> treatment in *P. deltoides* was not significant. Although leaf respiratory <sup>13</sup>C-enrichment in the DL was smaller than that in the IFB in average, the differences were not statistically significant (Table 1), partly due to the large variation of leaf  $\delta^{13}C_R$  in the MB and WB of the IFB.

The correlation between leaf  $\delta^{13}C_R$  and  $\delta^{13}C$  of the leaf organic components was highly significant (P < 0.01). On average across all species and treatments, the leaf respiratory CO<sub>2</sub> was 3.8‰ to 5.8‰ more positive than the four leaf organic components (Fig. 3) and in all cases, the slope of

© New Phytologist (2004) www.newphytologist.org

New Phytologist (2004) 163: 499-505

regression line was close to 1 (*F*-test to compare actual slope and 1, P = 0.08 - 0.98).

#### Discussion

Our results obtained with the leaf scale Keeling plot method are comparable to previous studies using a  $CO_2$  free respiration chamber (Park & Epstein, 1961; Jacobson et al., 1970; Smith, 1971; Duranceau et al., 1999; Ghashghaie et al., 2001; Damesin & Lelarge, 2003; Tcherkez et al., 2003). O'Leary (1981) pointed out that the CO<sub>2</sub> free environment might influence stomatal conductance and the extent of anapleurotic respiratory CO<sub>2</sub> refixation, but the effect on leaf  $\delta^{13}C_{R}$  has not been experimental evaluated. The 'leaf Keeling plot' approach maintains the leaf under CO<sub>2</sub> concentrations closer to their initial growth conditions (not more than 150 p.p.m. above background level, which is higher than the natural atmosphere in our study because of the plant and soil respiration in the glasshouse). In addition, the small air sampling flask (100cc) used is significantly more convenient for field measurement in remote areas. However, because leaf  $\delta^{13}C_{R}$  of C<sub>4</sub> plant is similar to surrounding ambient CO<sub>2</sub> (-7% to -15% vs. -8%), the change in the  $\delta^{13}$ C within the respiration chamber CO<sub>2</sub> is not large enough (Pataki et al., 2003) to use the leaf-scale Keeling plot method with C4 plants directly (data not shown).

In all five  $C_3$  species we studied, respiratory  $CO_2$  was more <sup>13</sup>C-enriched than the leaf organic components. Compared with soluble sugar, leaf  $\delta^{13}C_R$  was 3.5‰ to 5.9‰ more positive, which is consistent with the observation of Ghashghaie et al. (2001) and Duranceau et al. (1999, 2001) in crop plants. Therefore, the results led us to reject hypothesis (1) and conclude that a 3% to 6%  $^{13}$ C-enrichment relative to major respiratory substrate is widespread in leaf respiratory CO<sub>2</sub> from C<sub>3</sub> plants. The statistical significance found in the species is a basis for rejecting hypothesis (2) as well. Interestingly, in P. Detoides, the growth CO2 concentration did not influence the pattern of <sup>13</sup>C enrichment to a signifcant extent, supporting our hypothesis (3). We attribute the large variation in the amount of <sup>13</sup>C enrichment in 800 p.p.m. and 1200 p.p.m. CO<sub>2</sub> concentration to the variable CO<sub>2</sub> environment in the IFB, which had diurnal  $[CO_2]$  fluctuation of up to 300 p.p.m. and varible tank CO<sub>2</sub> injections. The variable background air CO<sub>2</sub> isotope signature could increase the isotopic heterogeneity in the substrate pool and ultimately, the variation in leaf  $\delta^{13}C_{R}$ .

Tcherkez *et al.* (2003) concluded that leaf  $\delta^{13}C_R$  in  $C_3$  plants is determined by (1) the carbon source used for respiration (2) possible isotope effects of respiratory enzymes, and (3) non-statistical distribution of <sup>13</sup>C in glucose. It is difficult to justify that an isotopically heavier respiratory substrate was used to any significant extent in addition to the pools measured here (particularly in light of the good correlations between putative substrates and  $\delta^{13}C_R$ ; Fig. 3). Also, previous

studies on mesophyll protoplasts (Lin & Ehleringer, 1997) indicated that fractionation likely does not occur in the main stream of respiratory enzyme reactions (glycolysis and TCA cycle). Instead, our synthesis of current results suggest that the non-statistical distribution of <sup>13</sup>C in sugars (Rossmann et al., 1991) is the most reasonable explanation for <sup>13</sup>C-enriched respiratory CO2. In dark respiration, the C-3 and C-4 of carbon atoms of glucose are <sup>13</sup>C-enriched (-20.9‰ in average), and are released early in glycolysis. The other 4 carbon atoms are isotopically lighter (-27.1‰ in average) and can enter secondary metabolisms through the TCA cycle. Thus a greater contribution of the C-3 and C-4 carbon atoms to respired  $CO_2$  would result in the isotopically heavier  $\delta^{13}C_R$  we observed. Based on the results of Rossmann et al. (1991), we estimate that a 3‰ enrichment in <sup>13</sup>C of respiratory CO<sub>2</sub> requires that 82% of the respired carbon be drived from C-3 and C-4. Although this number is possible, it is critical to conduct a more complete carbon budget for the entire leaf in future studies. Experiments similar to those of Lin and Ehleringer (1997), culturing protoplast in substrates labeled with  $^{13}C$  on certain carbon atoms, may directly illustrate the origin of respiratory CO<sub>2</sub>.

The concept of discrimination, defined as  $\Delta = (\delta_{source} - \delta_{product})/\delta_{source} - 1$ , applies to reactions with distinguishable source and product. However, in dark respiration, diverse substrates can be oxidized and a variety of compounds can be produced. Generation of CO<sub>2</sub> is only one branch of the overall metabolic network and it is not clear how an enzyme isotope effect (e.g. pyruvate dehydrogenase, Deniro & Epstein, 1977) may have influence the overall  $\delta^{13}C_R$ . Therefore we suggest that more suitable terminology is needed the <sup>13</sup>C enrichment of leaf respiratory CO<sub>2</sub> in C<sub>3</sub> plants.

The strong correlation between leaf  $\delta^{13}C_R$  and  $\delta^{13}C$  of leaf organic components suggest that in  $C_3$  plants the amount of <sup>13</sup>C enrichment is limited within a narrow range based on the similar mechanism involved. Therefore, it is possible to scale up leaf level results to predict leaf  $\delta^{13}C_R$  at the ecosystem level. For example, the average <sup>13</sup>C-enrichment in leaf  $\delta^{13}C_R$ relative to bulk organic matter (5.8‰, Table 1) can be used to approximate a whole  $C_3$  canopy. Obviously, a rigorous estimate of ecosystem scale effects would require more species specific data, as well as the effect of environmental factors (temperature, moisture, etc.) on the amount of <sup>13</sup>C enrichment in respiratory  $CO_2$ .

Synthesizing the currently available data, we found that significant inconsistency occurs in the results between the leaf and ecosystem levels. In most cases,  $\delta^{13}C$  of soil respiration, a much larger component than foliar respiration (Law *et al.*, 2001), is more positive than ecosystem respiration, indicating relatively <sup>13</sup>C-depleted vegetation  $\delta^{13}C_R$ . The scale difference may originate from the fact that, leaf respiration has fairly homogenous substrates and generates CO<sub>2</sub> relatively 'instantaneously'; while at the ecosystem level, photo-assimilated carbon is released over much longer timescales as a 'lagged

www.newphytologist.org © New Phytologist (2004)

New Phytologist

and prolonged' flux, which reflect more hetreogenous pools. Clearly, caution must be taken when predicting vegetation  $\delta^{13}C_R$  by scaling up leaf level results. At the leaf scale, further studies are required to understand (1) long-term patterns (ontogenetic, seasonal and annual) of leaf  $\delta^{13}C_R$  (2) the contribution and (3) representativeness of leaf respiration to total plant respiration. Alternatively, direct comparisons of  $\delta^{13}C$  of assimilated and respired CO<sub>2</sub> at the ecosystem level could also provide critical information to create more robust models.

## Acknowledgements

We thank Joost van Haren and Sara Green for technical assistance with the isotopic measurements. We also acknowledge Joost van Haren's effort in testing the protocol of leaf scale Keeling plot. Josslyn B. Shapiro and Dr O. R. Anderson are thanked for helpful comments on an earlier version of this manuscript. This work was supported in part by Biosphere 2 Center of Columbia University and by a grant from the Packard Foundation (DLP 998306 to G.L. & K.L.G.). G.L. was also supported in part by the 'Bairen Project' program of the Chinese Academy of Sciences.

## References

- Baertschi P. 1953. Die Fractionierung der natürlichen Kohlenstoffisotopen im Kohlendioxydstoffwechsel grüner Pflanzen. *Helvetica Chimica Acta* 36: 773–781.
- Bowling DR, Tans PP, Monson RK. 2001. Partitioning net ecosystem carbon exchange with isotopic fluxes of CO<sub>2</sub>. *Global Change Biology* 7: 127–145.
- Brugnoli E, Hubick KT, Von Caemmerer S, Wong SC, Farquhar GD. 1988. Correlation between the carbon isotope discrimination in leaf starch and sugars of C<sub>3</sub> plants and the ratio of intercellular and atmospheric partial pressures of CO<sub>2</sub>. *Plant Physiology* 88: 1418–1424.
- Buchmann N, Guehl JM, Barigah TS, Ehleringer JR. 1997. Interseasonal comparison of  $CO_2$  concentrations, isotopic composition, and carbon dynamics in an Amazonian rainforest (French Guiana). *Oecologia* 110: 120–131.
- Damesin C, Lelarge C. 2003. Carbon isotope composition of current-year shoots from *Fagus sylvatica* in relation to growth, respiration and use of reserves. *Plant, Cell & Environment* 26: 207–219.
- Deniro MJ, Epstein S. 1977. Mechanism of carbon isotope fractionation associated with lipid-synthesis. *Science* 197: 261–263.
- Duranceau M, Ghashghaie J, Badeck F, Deleens E, Cornic G. 1999.  $\delta^{13}$ C of CO<sub>2</sub> respired in the dark in relation to  $\delta^{13}$ C of leaf carbohydrates in *Phaseolus vulgaris* L. under progressive drought. *Plant, Cell & Environment* 22: 515–523.
- Duranceau M, Ghashghaie J, Brugnoli E. 2001. Carbon isotope discrimination during photosynthesis and dark respiration in intact leaves of *Nicotiana sylvestris*: comparisons between wild type and mitochondrial mutant plants. *Australian Journal of Plant Physiology* 28: 65-71.
- Ehleringer JR, Bowling DR, Flanagan LB, Fessenden J, Helliker B, Martinelli LA, Ometto JP. 2002. Stable isotopes and carbon cycle processes in forests and grasslands. *Plant Biology* 4: 181–189.
- Farquhar GD, Ehleringer JR, Hubick KT. 1989. Carbon isotope

discrimination and photosynthesis. Annual Review of Plant Physiology and Plant Molecular Biology 40: 503–537.

- Farquhar GD, O'Leary MH, Berry JA. 1982. On the relationship between carbon isotope discrimination and the inter-cellular carbon-dioxide concentration in leaves. *Australian Journal of Plant Physiology* 9: 121–137.
- Fessenden JE, Ehleringer JR. 2003. Temporal variation in δ<sup>13</sup>C of ecosystem respiration in the Pacific Northwest: links to moisture stress. *Oecologia* 136: 129–136.
- Flanagan LB, Ehleringer AR. 1998. Ecosystem-atmosphere CO<sub>2</sub> exchange: interpreting signals of change using stable isotope ratios. *Trends in Ecology* and Evolution 13: 10–14.
- Ghashghaie J, Duranceau M, Badeck FW, Cornic G, Adeline MT, Deleens E. 2001.  $\delta^{13}$ C of CO<sub>2</sub> respired in the dark in relation to  $\delta^{13}$ C of leaf metabolites: comparison between *Nicotiana sylvestris* and *Helianthus annuus* under drought. *Plant, Cell & Environment* 24: 505–515.
- Henderson SA, Voncaemmerer S, Farquhar GD. 1992. Short-term measurements of carbon isotope discrimination in several C<sub>4</sub> species. *Australian Journal of Plant Physiology* 19: 263–285.
- Jacobson BS, Smith BN, Epstein S, Laties GG. 1970. Prevalence of <sup>13</sup>C in respiratory carbon dioxide as an indicator of type of endogenous substrate, change from lipid to carbohydrate during respiratory rise in potato slices. *Journal of General Physiology* 55: 1–17.
- Keeling CD. 1958. The concentration and isotopic abundances of atmospheric carbon dioxide in rural areas. *Geochimica et Cosmochimica Acta* 13: 322–334.
- Keeling CD. 1961. The concentration and isotopic abundances of carbon dioxide in rural and marine air. *Geochimica et Cosmochimica Acta* 24: 277–298.
- Law BE, Thornton PE, Irvine J, Anthoni PM, Van Tuyl S. 2001. Carbon storage and fluxes in ponderosa pine forests at different developmental stages. *Global Change Biology* 7: 755–777.
- Lin GH, Ehleringer JR. 1997. Carbon isotopic fractionation does not occur during dark respiration in  $C_3$  and  $C_4$ . *Plant Physiology* 114: 391–394.
- Lin GH, Ehleringer JR, Rygiewicz PT, Johnson MG, Tingey DT. 1999. Elevated CO<sub>2</sub> and temperature impacts on different components of soil CO<sub>2</sub> efflux in Douglas-fir terracosms. *Global Change Biology* 5: 157–168.
- Lin GH, Rygiewicz PT, Ehleringer JR, Johnson MG, Tingey DT. 2001. Time-dependent responses of soil  $CO_2$  efflux components to elevated atmospheric  $CO_2$  and temperature in experimental forest mesocosms. *Plant and Soil* 229: 259–270.
- O'Leary MH. 1981. Carbon isotope fractionation in plants. *Phytochemistry* 20: 553-567.
- Park R, Epstein S. 1961. Metabolic fractionation of <sup>13</sup>C & <sup>12</sup>C in plants. *Plant Physiology* 36: 133–138.
- Pataki DE, Ehleringer JR, Flanagan LB, Yakir D, Bowling DR, Still CJ, Buchmann N, Kaplan JO, Berry JA. 2003. The application and interpretation of Keeling plots in terrestrial carbon cycle research. *Global Biogeochemical Cycles* 17: art, no. -1022.
- Rossmann A, Butzenlechner M, Schmidt HL. 1991. Evidence for a nonstatistical carbon isotope distribution in natural glucose. *Plant Physiology* 96: 609-614.
- Smith BN. 1971. Carbon isotope ratios of respired CO<sub>2</sub> from castor bean, corn, peanut, pea, radish, squash, sunflower and wheat seedlings. *Plant and Cell Physiology* 12: 451–452.
- Tcherkez G, Nogues S, Bleton J, Cornic G, Badeck F, Ghashghaie J. 2003. Metabolic origin of carbon isotope composition of leaf dark- respired CO<sub>2</sub> in French bean. *Plant Physiology* 131: 237–244.
- Troughton MJ, Card KA, Hendy CH. 1974. Photosynthetic pathways and carbon isotope discrimination by plants. *Carnegie Institute of Washington* Year Book 73: 768-780.
- Yakir D, Sternberg LDL. 2000. The use of stable isotopes to study ecosystem gas exchange. *Oecologia* 123: 297–311.

© New Phytologist (2004) www.newphytologist.org

New Phytologist (2004) 163: 499-505