# Enhanced growth of the red alga *Porphyra yezoensis* Ueda in high $CO_2$ concentrations

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Received 1 July 1991; revised 12 October 1991; accepted 18 October 1991

Key words: CO<sub>2</sub>, growth, pH, photosynthesis, Porphyra yezoensis, red alga

## Abstract

Leafy thalli of the red alga *Porphyra yezoensis* Ueda, initiated from conchospores released from free-living conchocelis, were cultured using aeration with high  $CO_2$ . It was found that the higher the  $CO_2$  concentration, the faster the growth of the thalli. Aeration with elevated  $CO_2$  lowered pH in dark, but raised pH remarkably in light with the thalli, because the photosynthetic conversion of  $HCO_3^-$  to  $OH^-$  and  $CO_2$  proceeded much faster than the dissociation of hydrated  $CO_2$  releasing H<sup>+</sup>. Photosynthesis of the alga was found to be enhanced in the seawater of elevated dissolved inorganic carbon (DIC,  $CO_2 + HCO_3^- + CO_3^-$ ). It is concluded that the increased pH in the light resulted in the increase of DIC in the culture media, thus enhancing photosynthesis and growth. The relevance of the results to removal of atmospheric  $CO_2$  by marine algae is discussed.

# Introduction

With industrial activities, the increasing  $CO_2$  concentration in the atmosphere is considered to trigger the 'greenhouse effect'. Studies are needed both to investigate the ecological impact of increasing  $CO_2$  and to make effective use of this  $CO_2$ . There have been hardly any reports on the effects of elevated  $CO_2$  concentration on marine macroalgae. However, the brown alga *Fucus serratus* has recently been reported to survive in 5%  $CO_2$  for three weeks, with its photosynthetic physiology being affected (Johnston & Raven, 1990).

The red alga Porphyra yezoensis Ueda has been

cultivated in Japan, and several studies have been made on its ecology and physiology (Oohusa, 1980; Merrill *et al.*, 1983; Tajiri & Aruga, 1984; Kato & Aruga, 1984; Gao & Aruga, 1987). However, nothing is known about its responses to elevated  $CO_2$  concentration. This paper presents findings on the impact of elevated  $CO_2$  concentrations on the growth of this alga.

# Material and method

From the stock of free-living conchocelis stage of *Porphyra yezoensis* (ZGRW, Miura 1990) maintained in the laboratory a cluster of conchocelis filaments was collected and cultured in an incubator under 10:14 LD cycle at 15 °C (light period: 0800–1800 at 300  $\mu$ mol photon m<sup>-2</sup>s<sup>-1</sup>). Synthetic fibres (vinylon monofilaments) were used for the conchospores to attach. Zero day was assigned to the age of the leafy thalli on the day of attachment. The thalli kept in batch culture were detached from the synthetic fibre at about 20 days old. Then 150 individual thalli (5 mm in length) were separated into three groups, one for the culture aerated with air (350 ppm CO<sub>2</sub>), the other two for the cultures aerated with air containing 1000 and 1600 ppm CO<sub>2</sub>, respectively.

The flasks with culture medium and thalli were placed in the CO<sub>2</sub> incubators (Shimadzu BEC-II-250) of 250 l capacity, in which CO<sub>2</sub> concentration was adjusted by mixing 99.9% CO<sub>2</sub> from a CO<sub>2</sub> cylinder with air transported from outdoor by an air compressor (Fig. 1). CO<sub>2</sub> concentrations of air in the incubators were monitored automatically every 10 min by infrared gas analysis (IRGA, Shimadzu URA-107, including a continuous measuring unit, Shimadzu IRA-107). Aeration was carried out by using a mini-pump (Shibata MP-2N) at a flow rate of 0.3–0.51 min<sup>-1</sup>. For culture medium, seawater collected from the sea off Miyazu facing the Sea of Japan (0.03 mg



*Fig. 1.* Illustration for indoor culture of *Porphyra yezoensis* under elevated  $CO_2$ . A, air compressor; B,  $CO_2$  cylinder; F, flow meter; M,  $CO_2$  monitor; a,  $CO_2$  inlet needle valve; c, air inlet needle valve; P, mini-pump; BF, culture flask; S, *P. yezoensis* thallus; C,  $CO_2$  cylinder for calibration; D,  $N_2$  cylinder for calibration; IRGA, infrared gas analyzer.

 $1^{-1}$  inorganic-N, 0.006 mg  $1^{-1}$  inorganic-P) was filtered (Whatman GF/C) and enriched with PES medium (Provasoli, 1966) after being autoclaved. Culture medium was renewed every other day. Illumination was provided by fluorescent lamps (National, 36 W). Photosynthesis was measured at 15 °C and 600  $\mu$ mol photon m<sup>-2</sup> s<sup>-1</sup> in a flow-water photosynthesis measurement system (Gao, 1989) jointed with a permeable membrane mass spectrometer (Ametek M200), and its rate was compared in the seawater with 2 mM and 10 mM DIC. The photosynthetic rate was determined from increased  $O_2$  in the closed system. Each comparison was completed with the same 10-15 thalli on the same day within 20 min in order to avoid effects caused by diurnal photosynthetic variation (Oohusa, 1980; Gao & Umezaki, 1989; Gao, 1990). Seawater with 10 mM DIC was prepared by adding NaHCO<sub>3</sub> into the usual seawater of 2 mM DIC, and the pH was adjusted to that of the usual seawater (pH 8.2) with NaOH and HCl.

#### Results

Significant differences (t-test, P < 0.01) in both length and width of Porphyra yezoensis thalli were found 10 days later in the cultures aerated with different concentrations of  $CO_2$  (Fig. 2). Growth of the thalli was fastest in 1600 ppm CO<sub>2</sub> culture, intermediate in 1000 ppm CO<sub>2</sub> culture and slowest in 350 ppm  $CO_2$  culture, i.e. it was enhanced with an increase in  $CO_2$  concentrations from 350 to 1600 ppm. Twenty days later, the thallus length or width in the cultures with 1000 and 1600 ppm CO<sub>2</sub> was about 1.4 and 1.9 times that in 350 ppm  $CO_2$  culture. Clear differences can be seen among the biomass of 50 thalli in respective cultures, and no extraordinary morphological features were observed (photo in Fig. 2). No significant difference (t-test, P > 0.05) in the length/width ratio of thalli (average 3.5) was found under various  $CO_2$  concentrations throughout culture period.

Aeration with air of 1000 and 1600 ppm  $CO_2$ reduced pH from 8.2 (350 ppm  $CO_2$ ) to 7.8 and 7.2, respectively, in the culture media without algal



Fig. 2. Growth of Porphyra yezoensis thalli under various  $CO_2$  concentrations. The mean length and width of 50 thalli are indicated with SD (n = 50). (Photo, biomass of the respective 50 thalli after 20 days in culture).

thalli or with algal thalli in dark, but not much with algal thalli in light. It was found that pH of the culture media with thalli lowered to reach the lowest value during dark period but rose to reach the highest value during the light period (Fig. 3A), and the higher the  $CO_2$  concentration, the bigger the pH rise in light (Fig. 3B).

Figure 4 shows changes of pH with flow rates



Fig. 3. Diurnal variations of pH (A) and its extends ( $\triangle$  pH, B) in culture media with the thalli of *Porphyra yezoensis* aerated with air of 350, 1000 and 1600 ppm CO<sub>2</sub>. The black bar indicates dark period, and the white bar light period. Measured with 18 to 21 days culture, one day after the culture medium was renewed.

of air and addition of carbonic anhydrase (CA) into the culture medium with the thalli in the light. When CA was added into the culture media with a saturating aeration rate (about  $1 \ln 1^{-1}$ ), the rise of pH was suppressed. It was also found that pH dropped abruptly to reach a constant value during the dark period when CA was added into the culture medium.

Light-saturated net photosynthesis of the alga was found to be much higher in seawater with 10 mM DIC than in usual seawater with 2 mM DIC, independent of culture background of  $CO_2$ concentrations (Fig. 5). The photosynthetic rate in the seawater with 10 mM DIC was enhanced by 31%, 17% and 40% in the thalli cultured with 350, 1000 and 1600 ppm  $CO_2$ , respectively.

#### Discussion

 $CO_2$  physically dissolved in culture media ([ $CO_2$ ]aq) can reach an equilibrium with that in the bubbling gas (PCO<sub>2</sub>), being hydrated to form carbonic acid, which dissociates to bicarbonate, then to carbonate as follows (Stum & Morgan, 1981):

$$cPCO_2 \rightleftharpoons [CO_2]aq$$
 (1)

$$[CO_2]aq + H_2O \rightleftharpoons H_2CO_3$$
(2)



Fig. 4. Changes of pH with flow rates of air and the presence of carbonic anhydrase (CA) in the culture medium with thalli of *Porphyra yezoensis*. The arrows indicate the time for change of flow rate of air and the addition of CA.



Fig. 5. Comparisons of light-saturated net photosynthetic rates of *Porphyra yezoensis* thalli cultured under various  $CO_2$  concentrations in seawater with 2 mM and 10 mM DIC.

$$H_2CO_3 \stackrel{kl}{\leftarrow} H^+ + HCO_3^-$$
(3)

$$HCO_{3}^{-} \stackrel{k2}{\leftarrow} H^{+} + CO_{3}^{-} , \qquad (4)$$

where c and k respectively indicate the solubility of  $CO_2$  and dissociation constant whose values depend on the temperature and chlorinity of seawater.

When  $CO_2$  concentration in air increases, the above reactions proceed toward the right-hand side to reach a new equilibrium, which gives rise to an increase of H<sup>+</sup> ions in seawater. Therefore, aeration with high  $CO_2$  can reduce pH of the culture medium. In the present study, aeration with elevated  $CO_2$  lowered pH in the dark but raised pH in the light in the culture medium with the thalli.

It has been shown that ribulose-bisphosphate carboxylase/oxygenase (Rubisco) is the predominant initial carboxylating enzyme in seaweeds (Kremer & Kuppers, 1977; Cook & Colman, 1987; Reiskind et al., 1988; Beer et al., 1990), which is known to fix  $CO_2$  not  $HCO_3^-$  (Cooper et al., 1969). Nevertheless, a number of seaweeds have been reported to take up  $HCO_3^-$  (Thomas & Tregunna, 1968; Jollife & Tregunna, 1970; Sand-Jensen & Gordon, 1984; Bidwell & McLachlan, 1985; Brechignac & Andre, 1985; Cook et al., 1986; Johnston & Raven, 1986; Smith & Bidwell, 1987; Axelsson & Uusitalo, 1988; Holbrook et al., 1988; Lignell & Pedersen, 1989; Surif & Raven, 1989; Madsen & Maberly, 1990; Beer et al., 1990). The rise of pH in the light must be attributable to photosynthetic carbon fixation in that the tremendous amount of  $HCO_3^-$  in the culture media or within the algal cells is converted to  $OH^-$  and  $CO_2$  with catalysis of CA (Asada, 1981; Silverman, 1991), the former raising pH, and the latter being assimilated in photosynthesis:

$$HCO_{3}^{-} \stackrel{CA}{\leftarrow} CO_{2} + OH^{-}$$
(5)

$$CO_2 + H_2O \xrightarrow{\text{light}} [H_2CO] + O_2.$$
 (6)

However, if the production rate of OH<sup>-</sup> ions in the reaction (5) is lower than that of  $H^+$  ions from dissociation of hydrated CO<sub>2</sub> in the reactions (3) and (4), the pH rise could not be attributable to the photosynthetic utilization of HC  $O_3^-$ . Addition of CA into the culture media suppressed the rise of pH at the saturating aeration rate. This indicates that the pH change disappeared when the reactions (3) and (5) are balanced with the catalysis of CA (Fig. 4). Accordingly, pH dropped gradually during the dark period and rose quickly during the light period (Fig. 3) in the culture media without CA added, just because the reaction (3) proceeded much more slowly than the reaction (5). The faster the photosynthetic carbon fixation, the more the OH<sup>-</sup> and the quicker the rise of pH. Because greater diurnal variations in pH were observed in the culture media aerated with high CO<sub>2</sub> concentration, photosynthetic carbon fixation must be much more actively advanced in such cultures.

On the other hand, photosynthesis is accom-

$$106 \text{ CO}_{2} + 16 \text{ NO}_{3}^{-} + \text{HPO}_{4}^{2-} + 122 \text{ H}_{2}\text{O} + 18 \text{ H}^{+} \\ \xrightarrow{\text{light}} [\text{C}_{106}\text{H}_{263}\text{O}_{110}\text{N}_{16}\text{P}_{1}] + 138 \text{ O}_{2}$$

$$106 \text{ CO}_{2} + 16 \text{ NH}_{4}^{+} + \text{HPO}_{4}^{2-} + 108 \text{ H}_{2}\text{O} \\ \xrightarrow{\text{light}} [\text{C}_{106}\text{H}_{263}\text{O}_{110}\text{N}_{16}\text{P}_{1}] + 107 \text{ O}_{2} + 14 \text{ H}^{+}.$$

Such processes are accompanied by the uptake of  $H^+$  or  $OH^-$  (or release of  $OH^-$  or  $H^+$ ), that is, by changes in alkalinity which is associated with charge balance. Subsequently, pH changes can result from the assimilation of ions. It was reported that the median C:N:P atomic ratio of marine macroalgae was 550:30:1 (Atkinson & Smith, 1983). Thereby, carbon must be assimilated much faster than nitrogen in the photosynthesis of marine macroalgae. Consequently, during the photosynthesis of *P. yezoensis*,  $NO_3^-$  assimilation would have a small effect on the pH rise in the culture medium with the thalli compared with the  $HCO_3^-$  utilization.

The rise of pH increases the dissolution of  $CO_2$ into the culture medium. DIC concentrations were calculated from the pH, CO<sub>2</sub> solubility and the dissociation constants of equations (3) and (4) provided that the influence by the alkalinity changes resulted from the uptake of ions was negligible. The estimated DIC was 1.3-21.5 mM, 3.7-61.5 mM and 5.9-98.4 mM in the pH range of 8.0-9.0 in the culture media aerated with 350, 1000 and 1600 ppm CO<sub>2</sub>, respectively. However, because both the solution of  $CO_2$  and the subsequent hydration are slower processes compared to photosynthesis (Fig. 4 shows indirectly), the actual DIC in the culture medium during the light period might be considerably lower than the above estimated values. In the present study, nevertheless, the aeration of high  $CO_2$  could bring about high DIC due to the raise pH in the culture medium.

Light-saturated net photosynthesis of the alga was enhanced in the seawater with elevated DIC (Fig. 5) i.e. photosynthesis of the alga is limited by inorganic carbon source in natural seawater as reported in some other marine macroalgae (Tseng & Sweeney, 1946; Wheeler, 1980; Holbrook *et al.*, 1988; Surif & Raven, 1989; Madsen & Maberly, 1990; Maberly, 1990). Therefore, the photosynthesis of the alga must have been accelerated in the culture media aerated with high  $CO_2$ . The accelerated photosynthesis resulted in enhanced growth of the alga.

The most important conclusion we draw from our results is that  $CO_2$  increased up to 5 times the present level enhances the photosynthesis and growth of *Porphyra yezoensis*, and that pH and DIC in seawater are correlated to each other, i.e. increased DIC with increased CO<sub>2</sub> in air accelerates the photosynthesis, and the accelerated photosynthesis raises pH, which then takes turns to elevate DIC. Stum & Morgan (1981) suggested that doubling of  $CO_2$  in the atmosphere would result, under equilibrium conditions, in a lowering of pH by 0.279 units in the mixed surface seawater of the sea. Is it possible, without understanding the above correlation, that pH of seawater would be chemically reduced with industrial emissions of  $CO_2$  and that the lowered pH would result in a release of  $CO_2$  into air from  $HCO_3^$ and  $CO_3^{2-}$  in the sea?

Our findings are applicable to effective use of  $CO_2$  with useful marine algae as well as to evaluating the effects of elevated CO<sub>2</sub> concentrations on marine ecology and air-sea CO<sub>2</sub> exchange, because our cultures were maintained in an opensystem which is considered as a simulator of the globe. Photosynthetic utilization of  $HCO_3^-$  has also been reported in marine phytoplankton (Badger & Andrews, 1982; Rees, 1984; Patel & Merrett, 1986; Munoz & Merrett, 1988, 1989). We suggest that, if the majority of marine plants raise pH of seawater under elevated CO<sub>2</sub> concentrations, the significance of marine photosynthesis should be noticed not only in carbon fixation but also in increasing the dissolution of  $CO_2$ into ocean.

## Acknowledgements

We thank Prof. A. Miura for his kindness in providing the free-living conchocelis of *P. yezoensis*.

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