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八种水果中的多酚含量及其抗氧化性

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摘 要:测定了苹果、石榴、橄榄、芒果、香蕉、菠萝、葡萄和龙眼的壳、肉及核中的多酚含量及其抗氧化性。以70% 丙酮(v/v)为提取溶剂,室温下超声波辅助浸提样品后得到提取液。采用普鲁士兰(Prussian blue)法测定了提取液中多酚和单宁的含量,利用 FRAP法测定其抗氧化性。结果表明:提取液中多酚和单宁的含量均与其抗氧化性成正相关关系;石榴、橄榄、芒果、葡萄和龙眼等水果的壳与核有望成为天然抗氧化剂的新来源。

关键词: 多酚; 单宁; 抗氧化性; 水果皮; 果肉和种子

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Polyphenol Contents in Eight Fruits and Their Antioxidant Activities

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Abstract The contents of polyphenols of peel pulp and seed fractions of apple, pomegranate, olive, mango banana pine-apple grape and long an and their total antioxidant activities were studied. A fter ultrasound-assisted extraction with 70% aqueous acetone at room temperature, polyphenols and tann ins in the plant cells were separated and their contents were determined by Prussian blue assay. Antioxidant activities of the aqueous acetone extracts were evaluated by Ferric Reducing/Antioxidant Power (FRAP) assay. A positive linear correlation between antioxidant activities and phenolic contents was observed. The results showed that the fruit peel and seeds of pomegranate, olive, mango, grape and longan could be considered as potential sources of antioxidants rather than just discarded as waste

Key words polypheno; tann ir, antioxidant, fru it peel pulp and seed

Introduction

The research interest in polyphenolic antioxidants has increased remarkably in the last decade because of their free radical scavenging activities associated with various diseases ^[1]. Synthetic antioxidants require high manufacturing costs but show bwer activities than natural antioxidants, and some of them may be toxic to human ^[2]. Therefore, a need is stimulated to identify natural and possibly more economic and effective antioxidants with potential to be used for foods industry. Antioxidant compounds have been identified in the apple pomace ^[3], banana ^[4], and in the seeds of grape ^[5],

mango $^{[6]}$ and olive $^{[7]}$. Besides, the pamegranate peel and brigan seed also showed high antioxidant aetivity. However to our knowledge, there are few studies relating to the antioxidant activity of fruit peel Given the considerable amount of by-products arising from fruits-processing plants improving the utilization of the fruit peel and seeds is a very important issue. One possible solution could be turning the fruit peel and seeds into a source of natural food additives and ingredients Fruits have excellent antioxidant properties and these effects are mainly attributed to their phenolic constituents [1,2]. To find out the distribution of phenolic substance in different kinds of plant fruits and determine their antioxidant activities apple pomegranate olive mango banana, pineapple grape and bugan were selected for this study. The objectives of the present study

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polyphenolic contents of peel pulp and seed extracts of selected fruits and to further investigate the relationships between antioxidant activities and polyphenolic contents in different fruit parts

Materials and Methods

M aterials

Fru its

"Youtanben" longan was purchased from a local farm in Zhangzhou Fujian Province China Apple (Fushi), pom eg ranate, o live, m ango (Zhua), ban an a, pine app le, and grape were purchased from WaHMart supermarket in X iam en, Fujian Province China

Chen icals

Polyviny bolypy rrolidone (PVPP), L-ascorbic acid and 2 4 6-tripyridyl-s-triazine (TPTZ) were purchased from Sigma-Aldrich Chemicals All other chemicals were commonly available reagents made in China

M ethods

Sample preparation

Organic solvent extraction is often used for isolation of antiox idents Both extraction yield and antiox ident activity of extracts are strongly dependent on the solvent species [2]. Aqueous-acetone system is widely employed because of their high efficiency of polyphenol extraction [13]. Therefore, acetonew ater system was selected for the extraction The extraction process was followed by Makkar's [9]. 0 5-3 g of each part of the fruit (fresh tissue) was accurately weighed, and extracted with 50 mL aqueous acetone (70%, v/v) under ultrasonic treatment for 20 m in $(2 \times 10 \text{ m in w ith } 5 \text{ m in break in between})$ at noom temperature. The solution was then transferred to centrifuge tubes and subjected to centrifugation for 10 m in at 3000 g. The supernatant was collected and kept cooled on ice The precipitate from centrifugation was then re-extracted with 50 mL agueous acetone (70%, v/v) and the above procedures were repeated to max ÷ m ize the extract All extracts were concentrated on a rotary evaporator below 40 °C under reduced pressure. The agueous residue (about 30 mL) was transferred into 50 mL volumetric flask

Determination of total polyphenol contents

tracts were evaluated by the Prussian blue assay $^{\left[10\right] }.$ Diluted extracts (3 0 mL) were mixed with 1 mL of $0.016 \text{ mol/L K}_3 \text{ Fe}(\text{CN})_6$ in a tube, then 1 mL of 0.02mol/LFeCl in 0 1 mol/LHCl was added The contents were m ixed well and kept at (24 ± 1) °C for 15 m in Then 3 m L of 6. 03 m o l/L H₃ PO₄ was added and the reagents in the tube were mixed well After 2 min 2 mL of 1% gum acacia was added and the color density was measured at 700 nm against a reagent blank. The amount of total polyphenols was calculated as a tannic acid equivalent from the calibration curve of tannic acid standard solutions (covering the concentration range between 3 and 15 mg/L), and expressed as mg tannic acid equivalent (TAE)/g fresh plant material All measurements were performed in triplicate

Determination of tannins

Tannin content was determined by Prussian blue assay as above, after removal of tann in s by their adsorption on in solub le matrix (po lyviny lpo lypyrro lidone PVPP) [11]. Insoluble cross-linked PVPP (approximately 250 mg) was weighed and mixed with 10 mL diluted extracts in test tubes After 15 m in at 4 °C, tubes were vortexed and centrifuged for 10 m in at 4350 g. A liquots of supernatantwere transferred into test tubes and nonabsorbed polyphenols determined as described above Calculated values were subtracted from total polyphenol contents and total tannin contents expressed as mg tannic acid equivalent (TAE)/g fresh plantmaterial All measurements were performed in triplicate.

Determination of total antioxidant activity

Total antioxidant activity of investigated aqueous acetone extracts was evaluated by the modified FRAP (Ferric Reducing/Antioxidant Power) assay [12]. Dilıted extracts (0 1 mL) were transferred into test tubes and 3. 0 mL of freshly prepared FRAP reagent (25 mL acetate buffer 300 mm ol/L, pH 3 6 + 2 5 mL 10 mmol/L TPTZ in 40 mmol/L HCl + 2 5 mL 20 mmol/L FeCh • 6H2O) were added After 5 min, the absorbance was measured at 593 nm against a blank containing 0 1 mL of solvent Relative activities were calculated from the calibration curve of L-ascorbic acid standard solutions (0 1-1 mmol/L) under the same

Total polyphenol contents of the aqueous acetone exexperimental conditions and expressed as µmol ascorbic acid equivalents (AAE) /g of fresh plantmaterial All measurements were performed in triplicate

Results and Discussion

Phenolic contents

Total polyphenol and tann in contents of peel, pulp and seed fractions of eight selected varieties of fruits were measured by the use of Prussian blue method As shown in Table 1, total polyphenol and tann in contents varied considerably from one kind of variety to another. In addition, they were found to be different in different parts of the fruits Mean total polyphenol and tann in contents varied from 0 28 to 71. 93 mg/g and from 0 05 to 50 94 mg/g of fresh matter, respectively. A good linear relationship between total polyphenol and tann in contents was found ($r^2 = 0.9901$, P < 0.001, n = 22) (Fig. 1). The sbpe of curve in Fig. 1 indicated that the mean tann in content value corresponded to 71% of the mean total polyphenol content value

Table 1 Total antioxidant activities and contents of polyphenol and tannin of peel pulp and seed extracts

Sample		$Total\ polypheno \mathbf{k}^a$	Tann ins ^a	FRAP values ^b
Apple	Peel	1 52 ±0. 03	1. 27±0 05	10 07±0.33
	Pulp	0 47 ±0.06	0. 40±0 02	301 ± 017
	Seed	1 41 ±0. 10	0.86±0 01	3 55±0 07
Pom egranate	Peel	71 93 ±7. 84	50. 94±0 33	571 92±11. 14
	Pulp	0 72 ±0.06	0. 53±0 02	3 22±0 18
	Seed	2 27 ±0. 42	1. 08±0 29	18 52±3.44
O live	Peel	14 47 ±0. 28	10. 61±0 08	75 21±2 23
	Pulp	14 39 ±0. 64	10.84±0 57	73 15±5.91
	Seed	12 80 ±0. 60	10. 67±0 41	69 54±1.92
M ango	Peel	16 08 ±0. 40	7. 83±0 14	52 22±4.50
	Pulp	1 17 ±0.06	0. 92±0 05	4 79±0 26
	Seed	27. 44 ±0. 76	17. 82±0 40	87. 42±6. 56
Banana	Peel	1 05 ±0.11	0. 58±0 08	7. 63 ± 0.53
	Pulp	0 47 ±0.04	0. 31±0 08	3 53±0 27
P ineapple	Peel	0 95 ±0.01	0.63 ± 0.02	6 00±0 30
	Pulp	0 42 ±0.00	0. 24±0 00	2 46±0 14
Grape	Peel	4 78 ±0. 13	3. 50±0 08	29 05±0.85
	Pulp	0 28 ±0.02	0. 05±0 01	4 05±0 12
	Seed	15 22 ±0. 42	11. 38±0 44	91 80±4.43
Longan	Peel	21 11 ±0. 99	16. 13±0 58	118 34±5.49
	Pulp	0 90 ±0.06	0.56±0 06	5 15±0 16
	S eed	26 72 ±1. 95	21. 84±1 32	155 32±4.78

Means of six determinations $\pm SD$ (standard deviation)

The peel and seed extracts contain more polyphenolic and tann in contents than the pulp extract Some previous studies [2,5,6] focused on phenolic compounds of the seeds of subtropical and tropical fruits, but there are few reports on the phenolic contents of the peel extracts. The results obtained showed that the polyphenolic content of pomegranate peel was the highest, 71, 93 mg/g followed by the peel of bngan (21, 11), mango (16, 08), olive (14, 47), grape (4, 78), apple (1, 52), banana (1, 05), and pineapple (0, 95).

A ntioxidant activities

FRAP assay is widely employed to evaluate the total antioxidant activity of plant materials [14]. As shown in Table 1, FRAP values of different fruits are very different ranging from 2 46 to 571. 92 \(\mu\)molAAE/g of fresh plant materials The FRAP values vary considerably from one kind of variety to another. In addition, they are found to be different for different parts of the fruits On the basis of the wet weight, longan seed had the highest antioxidant activity followed by the seeds of grape mango, olive pom egranate and apple For the peel portion, pomegranate peel showed the highest antiox idant activity followed by longar, olive, mango grape apple banana and pineapple And for the pulp fraction olive showed much higher antioxidant activity than the seven other fruits Overall fruit peel and seed extracts showed a much higher antioxidant capacity and polyphenolic contents than the pulp extracts of the test ed fruits

As illustrated in Fig 2 and Fig 3, good correlations were found between FRAP values and total polyphenol content ($r^2 = 0.9123$, P < 0.001, n = 21) and between FRAP values and tannin content ($r^2 = 0.9678$) P < 0.001, n = 21). This result strongly suggested that polyphenol content should be considered as an important feature of the fruits and their high levels of antioxidant activities were mainly attributed to the presence of phenolic constituents

^a Expressed as mg tannic acid/g fresh plant material

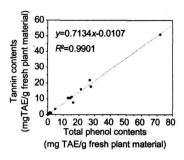


Fig. 1 Correlation of total polyphenol and tamin contents of peel, pulp and seed extracts

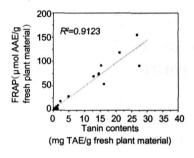


Fig. 2 Correlation of FRAP and total polyphenol contents of peel pulp and seed extracts

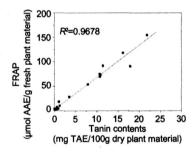


Fig. 3 Correlation of FRAP and tannin contents of peel pulp and seed extracts

Overall, the observed antioxidant activities and polyphenol contents were greatly dependent on different extracts of different fruit varieties. Interestingly, the peel extracts also exhibited high polyphenolic content and antioxidant activity, especially of the subtropical fruit, as well as the seed extracts. As phenolic compounds with different structures are likely to have different chemical and biological properties, it is important to clearly identify the individual phenolic compounds so that their single or synergistic effect can be further studied.

Conclusions

A great variation of polyphenolic contents and FRAP values in different extracts of different fruit varieties was observed Total antioxidant capacities and polyphenolic contents of fruit peel and seed extracts were found to be significantly higher than those of the pulp extracts Approximately 71% of the total polyphenols of eight determined fruits were tannins, which were the major antioxidant components of fruit Due to the elevated values of their antioxidant activity, the peel and seed extracts of pomegranate, olive, mango, grape and longan could be considered as new potential sources of antioxidants. The further study on their chemical properties and processing technologies, the identification of polyphenolic structures and stability of polyphenols, will be of importance

R eferences

- Silva EM, Souza JNS, Rogez H, et al. Antioxidant activities and polyphenolic contents of fifteen selected plant species from the Amazonian region Food Chon, 2007, 101: 1012-1018.
- 2 Soong YY, Barlow PJ Antioxidan tactivity and phenolic content of selected fruit seeds Food Cham, 2004, 88, 411–417.
- 3 Lu YR, Foo LY. Antioxidant and radical scavenging activities of polyphenols from apple pom ace Food Chan, 2000, 68 81-85
- 4 Som eya S, Yoshik i Y, Okubo K. Antioxidant compounds from bananas (Musa cavend ish). Food Cham, 2002, 79, 351–354.
- 5 Jayaprakasha GK, Singh RP, Sakariah KK. Antioxidan tactivity of grape seed(Vitis vinifera) extracts on peroxidation models in vitro Food Chon, 2001, 73 285-290
- 6 Puravankara D, Boghra V, Sharma RS Effect of antioxidant principles isolated from mango (Mang fera indica L.) seed kemels on oxidative stability of buffalo ghee (butter fat). J Sci Food Agric, 2000, 80: 522-526.
- 7 Owen RW, Haubner R, Mier W, et al Isolation, structure electrication and antioxidant potential of the major phenolic and flavonoid compounds in brined olive drupes Food Chem Toxicol 2003, 41: 703-717.

- (Solanaceae). Biochem Syst Ecol, 1993, 21: 629-644
- 10 Oncina R, DelRio JA, Gomez P, et al. Effect of ethylene on diosgen in accumulation in callus cultures of Trigonella foenum-graeaum L. Food Chem., 2002 76, 475-479.
- 11 Quan H.J. Koyanagi J. Ohmori K, et al. Preparations of heterospirostanok and their pharmacobgical activities Eur J. M. et Chem., 2002. 37: 659-669.
- 12 Li JR (李家儒), He J(何骥), Hu JL (胡江丽). Diosgenin assay by ELISA (薯蓣皂素含量的 ELISA 检测方法). China Patent 2004 ZL02147877. 5
- 13 Hu L (胡江丽), Li R (李家儒), He J (何骥). Prepara-

- tion of immunogens of diosgen in and their immuno-effects analysis J Wuhan Univ, Nat Sci(武汉大学学报, 理科版), 2003. 49.783-786.
- 14 Huang YH (黄亚辉), Sheng XB (盛孝邦). Advances on diosgen in research. Chin Wild PlantResour(中国野生植物资源), 2005, 24: 20-23.
- 15 Yang HY (杨红艳), Yuan LH (袁丽红), WuHL (吴红利), et al. Synthesis of diosgen in-BSA and diosgen in-OVA conjugates Nat Prod Res Dev (天然产物研究与开发), 2006, 18 254-256

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- 8 LiYF, Guo CJ, Yang JJ et al. Evaluation of antioxidant properties of pomegranate peel extract in comparison with pomegranate pulp extract Food Chen, 2006, 96 254-260.
- 9 Makkar HPS Quantification of tann ins in tree and shrub foliage a laboratory manual FAO/IAEA, Vienna, 2003 45-46
- 10 G raham H D. Stabilization of the Prussian blue color in the determination of polyphenols J Agric F ood Chom, 1992, 40 801–805.
- 11 Maksimovic Z, Malencic D, Kovacevic N. Polyphenol contents and antioxidant activity of Mayd is stigma extracts Bioresource

Tech, 2005 96: 873-877.

- 12 Benzie IFF, Strain JJ The ferric reducing ability of plasma as a measure of "antioxidant power": the FRAP assay *Anal Biodom*, 1996 239 70-76
- 13 Ri dtjer A, Sk ibsted LH, Andersen ML Antioxidative and prooxidative effects of extracts made from cherry liqueur pon ace Food Chan, 2006 99 6-14.
- 14 Guo CJ Yang JJ Wei JV, et al. Antioxidant activities of peel, pulp and seed fractions of common fruits as determined by FRAP assay Nutr Res 2003, 23 1719-1726