Rapid Communication

A synthetic approach towards Pinakene, a C₂₈H₁₄ fragment of [70]- fullerene

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During a projected synthesis of 4,7,11,14- tetramethyldibenzo [fg,op] naphthacene 3 en route to Pinakene 2, a [70]-fullerene fragment, an unprecedented photochemical reaction leads to dimethylnaphthacene derivative 7. Flash vacuum pyrolysis on 7 leads to 8 via a transannular bridging.

While the 'gold-rush' towards the exploration of the chemistry of fullerenes continues unabated¹, there are spinoffs in the form of emerging synthetic interest² in accessing curved, non-planar aromatic fragments and siblings of these new carbon allotropes, which might possess novel physico-chemistry characteristics. Noteworthy progress towards the synthesis of several C₆₀ fragments has already been achieved during the past few years.² However, synthetic efforts towards the fragments of [C₇₀]- fullerene 1 have yet to mark much headway.³ Recently, we have recognized 'Pinakene' 2, a C₂₈H₁₄ bowl-shaped hydrocarbon as a dominant sub-unit of 1, and delineated its molecular structure based on MNDO



and MM2 calculations⁴. In this effort, a possible synthetic approach to 2 was also considered and 4,7,11,14- tetramethyldibenzo[fg,op]naphthacene 3 was identified as a key precursor, in which thermally induced four-fold transannular bridging could be induced under flash vacuum pyrolysis conditions. A literature search revealed that synthetic routes to the dibenzonaphthacene system were relatively few and neither the strained 3 (E_s 83.1 kcal/mol) nor any derivative bearing an alkyl substituent in *peri*-position were known. Thus, devising a synthesis of 3 and evaluating the feasibility of the contemplated bridging process ($3\rightarrow 2$) became our initial concern and some interesting observations in this quest form the subject matter of this communication.

Hart reaction⁵ of 2,5-dimethylphenylmagnesium bromide with 2,6- dichloroiodobenzene 4 and quenching the reaction with iodine furnished a mixture of configurationally stable iodoterphenyl derivatives, *syn*-5a and *anti*-5b, from which the latter readily- crystallized and was identified through Xray crystal structure determination (Figure 1)⁶.

Palladium catalyzed Suzuki cross-coupling reaction between 5a,b and cyclicphenylboronate ester furnished a highly crowded *m*- terphenyl derivative **6** (Scheme I) as a mixture of rotamers. Irradiation of **6** under conditions of oxidative photoelectrocyclization, to our surprise, furnished 4,7- dimethyl-





Reagents & yields: (i) 2,5-Dimethylphenylmagnesiumbromide, THF, Δ , I₂, 80%; (ii) Phenyl boronate ester, Et₃N, PPh₃, Pd(OAc)₂, DMF, 140°C, 40% (iii) 450W Hg Lamp, Vycor, C₆H₆: Acetone, (80:20), Cat. I₂, -20%; (iv) FVP, 950°C, 0.1 torr (quant. based on recovery of 7)

Scheme I

dibenzo[fg,op]naphthacene 7, instead of the expected 3. The structure of 77 was secured through its spectral characteristics, particularly through 2D NMR (¹H-¹H COSY) experiments. Formation of 7 could involve an unprecedented photodearylationrearylation process in which the *p*-xylene moiety in 6 is exchanged with a phenyl moiety from the solvent benzene prior to photocyclization. The driving force for the photodearylation of 6 is the mitigation of steric strain and is probably manifested through the cleavage of one of the ArC-CAr (CH₃)₂ bonds and rearylation through solvent capture. We are not aware of any example of such photodearylation procss in the literature. Alternately, it is possible that 3 is indeed formed during the photo-irradiation, but suffers unprecedented loss of methyl groups to fur-

nish 7. However, we have been unable to detect 3 in photolystate from 6.

The feasibility of the proposed transannular bridging was explored by subjecting 4,7-dimethyldibenzonaphthacene 7 to flash vacuum pyrolysis. The only product isolated in this reaction was the monobridged compound 8^7 (-15% conversion), whose structure was revealed through the spectral data. Further efforts towards incorporating additional bridges on to dibenzonaphthacene framework, via the appropriately placed halogen substituents are in progress.

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References

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 - Data was collected on a Siemens R3m/V diffractometer and refined using Shelxtl Plus. Crystal data for the compound C₂₂H₂₁I:Transparent plate crystals, Triclinic, Space group P-1, a = 6.979(1), b = 10.175(1) and c =13.711(1) Å, $\alpha = 92.0(1)$, $\beta = 97.2(1)$, $\gamma = 98.2(1)^\circ$, V =954.69Å³, Z = 2, T = 293K, D_c = 1.434 mg-m⁻³, crystal dimensions 0.15 × 0.17 × 0.13mm³, 2762 reflections measured, $2\theta_{max} = 45^\circ$, $0 \le h \le 7$, $-10 \le k \le 10$, $-14 \le I \le$ 14, 2516 unique reflections (R_{int} =0.021 averaging double measured) and 2304 observed with $I \ge 3\sigma(I)$, $\mu =$ 1.68mm⁻¹. Final R =0.032 and R_w0.046 {208 parameters and w =1/ σ^2 {(Fo) +0.000878Fo²)} maximum shift/error =0.001, $\Delta\rho_{max} = 0.56eÅ^3$, $\Delta\rho_{min} = -0.92eÅ^3$.
 - Selected spectral data 7: ¹H NMR (200 MHz, CDCl₃): δ 8.92 (d, 2H, J =10Hz), 8.87 (d, 2H, J =10Hz), 8.72 (d, 2H, J=9Hz), 8.63 (s, 2H), 8.01-8.09 (m, 2H), 7.57 (d, 2H, J=8Hz), 2.68 (s, 6H); MS: 330 [M⁺]. 8: ¹H NMR (400 MHz, CDCl₃): δ 0.84 (d, 1H, J=8Hz), 8.82 (d, 1H, J = 8Hz), 8.80 (d, 1H, J =8Hz), 8.70 (d, 1H, J =8Hz), 8.66 (d, 1H, J =8Hz), 8.42 (d, 1H, J =8Hz), 8.10 (d, 1H, J = 8Hz), 8.04 (d, 1H, J =8Hz), 7.70 (m, 2H), 7.51 (d, 1H, J =8Hz), 4.53 (s, 2H), 2.68 (s, 3H); MS: 328 [M⁺].