Note

Nucleophilic substitutions on 3-chloro-4- fluoronitrobenzene

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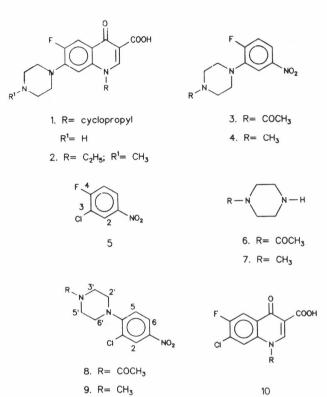
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Received 4 November 1996; accepted 23 December 1996

Nucleophilic substitutions on 3-chloro-4- fluoronitrobenzene 5 with piperazines occur on the fluorine rather than on the chlorine atom as reported, to yield 3-chloro-4-piperazinylni-trobenzenes 8 and 9.

Our interest in antibacterial 6-fluoro-7-piperazinylquinolones like ciprofloxacin 1 and pefloxacin 2¹ led us to consider 4-fluoro-3-(substituted)-piperazinvlnitrobenzenes 3 and 4 as starting materials in one route. A recently published Indian Patent² reported the formation of 3 in 90% yield from 3-chloro-4fluoronitrobenzene 5 by reaction with N-acetylpiperazine 6 in pyridine at 120-125° for 4 hr. The patent further describes the conversion of 3 into ciprofloxacin 1 via several obvious steps. A priori the substitution of chlorine meta to nitro group in 5 in preference to the intrinsically more reactive fluorine at position - 4 which is further activated by a paraplaced nitro group seemed unlikely, but the patent claim had to be tested since 1 had been reportedly obtained from 3.

In the event, the reaction of 5 with 6 in pyridine at 120-125° for 4 hr gave a yellow product. m.p. 91-92° (from MeOH) in poor yield (lit². m.p. claimed for 3, 48-51°). The reaction of 5 and 6 went better when heated neat at 60° for 2 hr to afford the same product in 53% yield. Physical data however established the identity of the product beyond doubt as 8, arising from 5 by displacement of fluorine as expected and not chlorine. Analysis. Found: C, 50.72; H, 4.94; N,



14.83; M⁺ 283, 285. Calcd for $C_{12}H_{14}ClN_{3}O_{3}$: C,50.81; H, 4.97; N, 14.82%; M⁺, 283.72; ¹H NMR (60 MHz, CDCl₃): δ 8.27 (C₂-H, d, *J* =2.5Hz), 8.13 (C₆-H, dd, *J* =8.2, 2.5 Hz), 7.03 (C₅- H, d, *J* =8Hz), 3.73 (4H at C- 3' and C- 5', unresolved m), 3.23 (4H at C- 2' and C-6', unresolved m), 2.15 (COCH₃, s).

Condensation of 5 with N-methylpiperazine 7 should have given 4 according to the patent which would lead to pefloxacin 2 by the reaction sequence outlined therein. However, the reaction of 5 and 7 in pyridine at 120-125° for 4 hr gave only 9 in 59% yield [m.p. 94-95° (from MeOH)] and not 4. Analysis. Found : C, 51.82; H, 5.65; N, 16.32; M+255, 257. Calcd for C₁₁H₁₄ClN₃O₂: C, 51.67; H, 5.52; N, 16.47%; M+255.71; ¹H NMR (60 MHz, CDCl₃: δ 8.18 (C₂-H, d, *J* =2.5 Hz), 8.03 (C₆-H, dd, *J* =8, 2.5 Hz), 7.03 (C₅-H, d, *J* =8 Hz), 3.20 (4H at C- 2' and C-6', m), 2.57 (4H at C- 3' and C-5', m), 2.32 (CH₃, s).

The regrettable conclusion that 3 and 4 which could well serve as starting materials for 1 and 2 respectively cannot be obtained from 5 by nucleophilic substitution thus becomes inevitable. We wish to point out that the conventional synthesis of 1, 2 and similar compounds depends upon a preferential nucleophilic substitution of chlorine in 10 with piperazines in the final step but now the chlorine is activated by a *para*-placed carbonyl group while the fluorine does not enjoy such activation.

Acknowledgement

We thank Dr Paul Baynes and Mr Ashok Konaji for mass and ¹H NMR spectra and Profs S. Chandrashekaran and G.S.R. Subba Rao of the Indian Institute of Science, Bangalore for microanalysis.

References

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