Note

Synthesis of some new spirothiazolidinone and spiroazetidinone derivatives incorporated with quinazoline

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3-Amino-2-methyl-3*H*-quinazolin-4-one 1 reacts with cyclic ketones 2a-d to afford the corresponding cycloalkylidene-3-aminoquinazolinone derivatives 3a-d. which on treatment reacted with mercaptoacetic acid to give the spirothiazolidinones 4a-d. Also the reaction of compounds 3a-d with chloroacetyl chloride in the presence of triethylamine as a catalyst yield the respective spiroazetidinones 5a-d. All the synthesized spiroheterocycle derivatives have been identified by conventional methods IR and ¹H NMR spectroscopy and elemental analyses.

Certain spiro compounds show anticancer¹, central nervous system activity², antiinflammatory activity³ and antibiotic aranorosin⁴. Recent literature reports revealed the synthesis of spiroheterocycles which were used as intermediates for aldose reductase inhibitors⁵. Some new spiroheterocycles are found to have activity as herbicides and pesticides⁶. Also, a facile access to aphidicolane and stemodane B/C/D-ring systems have been reported^{7,8}, and the syntheses of some new heterobicyclic compounds containing the spiro-1,2,4triazine moiety as potential anti HIV and anticancer agents were investigated9 by us. The preparation of fluoran compounds for use as recording materials has also been carried out¹⁰⁻¹². Also, the preparation of spiroazafuranone derivatives to be used for the treatment of neurodegenerative disorders and as anxiolytics were achieved 13,14. Quinazoline derivatives showed divers biological activities¹⁵⁻¹⁹. Thiazolidinone derivatives have considerable commercial importance as drugs e.g. bactericidal, pesticidal, fungicidal, insecticidal, anticonvulsant, tuberculostatic, antiinflammatory, antithyroidal and potentiation of pentobarbital induced sleeping time^{20,21}. Spirocycloalkylsubstituted azetidinones were used as hypocholesterolemic agents²². The syntheses of β-lactams using different techniques and catalysts have been accomplished^{21,23,24}. The syntheses of spiro compounds containing nitrogen have gained importance because of their biological activity, but, in some cases, the preparation of these compounds required seven steps²⁵. Also, the preparation of the wellknown spiro derivatives, fredericamycin A, required eight steps²⁶. From all of the forgoing facts it was very interesting to synthesize a new series of spirothiazolidinones and spiroazetidinones incorporated with quinazoline. We report herein a facile synthesis of some spiroquinazolineheterocycles derivatives analogous to spiro[indan-1,1'[1H]-3benzenepine] derivatives²⁵ and fredericamycin A²⁶. The advantages of our syntheses were the use of inexpensive precursors such as anthranilic acid, hydrazine hydrate, cyclic ketones and mercaptoacetic acid, facile reactions, readily available reagents and simple techniques. Our syntheses were initiated with the reaction of 3-amino-2-methyl-3H-quinazolin-4-one 1 with 1-indanone 2a, 1tetralone 2b, fluorenone 2c and anthrone 2d to afcorresponding cycloalkylidene-3ford aminoquinazolinone derivatives 3a-d in good yields (Scheme 1). The structure of compounds 3ad were established from their elemental analyses and spectroscopic data (Table 1). The IR spectrum of 3a showed characteristic strong absorption bands at 1685 cm⁻¹ corresponding to the stretching vibrations of the carbonyl group of the quinazoline ring and 1600 cm⁻¹ for the C=N stretching. The ¹H NMR spectrum of 3a (DMSO- d_6) showed the following signals: δ 2.59 (2H, t) for the methylene protons of the indan moiety, 3.05 (2H, t) for the benzylic methylene protons of indan residue, 3.39 (3H, s) for the methyl protons at C₂ of the quinazoline ring and 7.00-8.14 ppm (8H, m) for the aromatic protons of both quinazoline and indan rings. The 13 C NMR spectrum of 3a (DMSO- d_6) showed the following signals: δ 21.88, 25.40, 35.83, 119.76, 122.84, 125.84, 126.60, 126.92, 127.06, 127.16, 133.82, 134.59, 136.64, 146.63, 155.20, 155.46, 160.08, 206.24 ppm.

Compounds 3a-d reacted with mercaptoacetic

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Scheme I

Scheme II

acid in dry benzene to give the target products $\bf 4a-d$ in good yields (Table 1, Scheme 1). The structures of compounds $\bf 4a-d$ were confirmed on the basis of their elemental analyses and spectroscopic data (Table I). The ¹H NMR spectrum of $\bf 4a$ (DMSO- d_6) showed the following signals: δ 2.43

(2H, t) for the methylene protons of the indan ring, 2.59 (2H, t) for the benzylic methylene protons of the indan residue, 2.92 (2H, s) for the methylene protons of thiazolidine ring, 3.40 (3H, s) for methyl protons at C₂ of quinazoline ring, 7.39-8.34 ppm (8H, m) for the aromatic protons of

(Contd)

Compd Yi'eld mp Mol. formula Compd Yi'eld mp Mol. formula Compd Yi'eld (5) (C) (Solvent of Compd (%) (C) (C) (C) (C) (C) (C) (C) (C) (C) (C			Table	Table I—Physical Data of -	-3-aminoqu	inazolino	nes (3a-d),	spirothiazo	olidinone	-3-aminoquinazolinones (3a-d), spirothiazolidinones (4a-d) and spiroazetidinones (5a-d)—Contd	(Sa-d)—Contd
61 64-66 C ₂ H ₁ N ₁ O ₂ S 70.00 4.00 10.10 7.70 - 2.90 (2H.s.), 3.40 (3H.s.), (cthanol) 770.077 (4.13) (10.21) 7.78 - 2.90 (2H.s.), 3.40 (3H.s.), (cthanol) 770.28) (4.47) (9.88) 7.29 - 2.85 (2H.s.), 2.90 (2H.s.), (cthanol) 770.28) (4.47) (9.88) 7.29 - 2.85 (2H.s.), 2.90 (2H.s.), (cthanol) 770.28) (4.47) (9.88) 7.29 - 2.85 (2H.s.), 2.90 (2H.s.), (cthanol) 770.28) (4.47) (9.88) 7.29 - 2.85 (2H.s.), 2.90 (2H.s.), (cthanol) 770.28) (4.47) (9.88) 7.29 - 2.85 (2H.s.), 2.90 (2H.s.), (cthanol) 770.28) (4.47) (9.88) 7.29 - 2.85 (2H.s.), 2.90 (2H.s.),	Compd	Yield (%)		Mol. formula (Solvent of crystallization)			Found (%	6) (Calc.)		'H NMR (DMSO-d ₆), (6, TMS) ppm	¹³ C NMR (DMSO- <i>d</i> ₆), (δ, TMS) ppm
61 64-66 C ₃ H ₁ N ₁ O ₂ S 70.00 4.00 10.10 7.70 - 2.90 (2H, s), 3.40 (3H, s), (ethanol) (70.07) (4.13) (10.21) (7.78) - 2.85 (2H, s), 2.90 (2H, s), 3.40 (3H, s), (ethanol) (70.58) (4.47) (9.88) (7.52) - 2.85 (2H, s), 2.90 (2H					၁	Н	z	S	C	·1	
58 190-192 C ₂₃ H ₁₈ N ₃ O ₂ S 70.50 4.40 9.80 7.49 – 2.85 (2H, s), 2.90 (2H, s), (12H, m) (12H, m) (12H, m) 58 220-222 C ₂₉ H ₁₆ N ₃ O ₂ Cl 65.70 4.35 11.40 – 9.50 2.43 (2H, t), 2.59 (2H, t), (benzene) (65.75) (4.38) (11.50) – 9.50 2.43 (2H, t), 2.59 (2H, t), (benzene) (66.40) (4.74) (11.00) – 9.10 2.03 (2H, m), 2.43 (2H, t), (benzene) (69.73) (3.87) (11.08) – 9.10 2.03 (2H, m), 2.43 (2H, t), (benzene) (69.73) (3.87) (10.16) – 8.40 3.40 (3H, s), 5.52 (1H, s), (benzene) (69.73) (3.87) (10.16) – 8.40 3.40 (3H, s), 5.52 (1H, s), (benzene) (69.73) (70.20 4.15 9.80 – 8.10 2.90 (2H, s), 7.00-8.29 (12H, m) (12H, m) (12H, m) (12H, m) (12H, m)	40	61	64-66	C ₂₄ H ₁₇ N ₃ O ₂ S (ethanol)	70.00	4.00 (4.13)	10.10 (10.21)	7.70 (7.78)		2.90 (2H, s), 3.40 (3H, s), 7.00-8.35 (12H, m)	27.70, 28.85, 119.67, 120.67, 120.49, 125.83, 126.16, 126.42, 126.56, 126.92, 127.40, 133.35, 133.82, 142.25, 142.50, 144.60, 146.66, 146.54, 152.85, 155.46, 155.79, 159.99, 174.67, 197.45
56 250-222 C ₃ H ₁₈ N ₃ O ₅ Cl 65.75 (4.38) (11.50)	b	80	190-192	C ₂₅ H ₁₉ N ₃ O ₂ S (ethanol)	70.50	4.40 (4.47)	9.80	7.49	1	2.85 (2H, s), 2.90 (2H, s), 3.40 (3H, s), 7.10-8.40 (12H, m)	26.70, 27.60, 28.85, 119.67, 120.67, 121.40, 125.83, 126.16, 126.42, 126.50, 126.92, 127.40, 133.35, 133.82, 142.25, 142.50, 144.60, 146.66, 146.54, 152.85, 155.46, 155.79, 159.99, 174.67, 197.45
56 250-252 C ₂₁ H ₁₈ N ₅ O ₂ Cl 66.49 (4.74) (11.08) — 9.10 2.03 (2.H, m), 2.43 (2H, t), (benzene) (66.49) (4.74) (11.08) — (9.23) 2.57 (2H, t), 3.40 (3H, s), 5.70 (1H, s), 7.00-8.30 (8H, m) (69.73) (3.87) (10.16) — 8.40 3.40 (3H, s), 5.52 (1H, s), (benzene) (69.73) (3.87) (10.16) — (8.47) 7.00-8.34 (12H, m) (69.73) (70.20 4.15 9.80 — 8.10 2.90 (2H, s), 3.40 (3H, s), 7.00-8.29 (12H, m) (12H, m) (12H, m)	S S	28	220-222	C ₂₀ H ₁₆ N ₃ O ₂ CI (benzene)	65.70 (65.75)	4.35 (4.38)	11.40	.1	9.50	2.43 (2H, t), 2.59 (2H, t), 3.39 (3H, s), 5.82 (1H, s), 7.00-8.14 (8H, m)	22.60, 25.84, 28.81, 119.67, 120.49, 125.83, 126.42, 126.56, 128.29, 132.06, 133.82, 142.25, 144.60, 146.66, 152.85, 155.46, 155.79, 159.99, 174.67, 197.40
55 230-232 C ₂₄ H ₁₆ N ₃ O ₂ Cl 69.70 3.80 10.10 8.40 3.40 (3H, s), 5.52 (1H, s), (benzene) (69.73) (3.87) (10.16) - (8.47) 7.00-8.34 (12H, m) 54 245-247 C ₂₅ H ₁₈ N ₃ O ₂ Cl 70.20 4.15 9.80 8.10 2.90 (2H, s), 3.40 (3H, s), (benzene) (70.25) (4.21) (9.83) - (8.19) 5.52 (1H, s), 7.00-8.29 (12H, m)	Sb	26	250-252	C ₂₁ H ₁₈ N ₃ O ₂ CI (benzene)	(66.49)	4.70 (4.74)	(11.08)	t	9.10 (9.23)	2.03 (2H, m), 2.43 (2H, t), 2.57 (2H, t), 3.40 (3H, s), 5.70 (1H, s), 7.00-8.30 (8H, m)	21.82, 22.79, 27.84, 28.81, 119.76, 120.49, 125.83, 126.42, 126.56, 128.29, 132.06, 133.82, 142.25, 144.60, 146.66, 152.85, 155.46, 155.79, 159.99, 174.67, 197.40
54 245-247 C ₂₅ H ₁₈ N ₃ O ₂ Cl 70.20 4.15 9.80 8.10 2.90 (2H, s), 3.40 (3H, s), (benzene) (70.25) (4.21) (9.83) (8.19) 5.52 (1H, s), 7.00-8.29 (12H, m)	20 20 20 20 20 20 20 20 20 20 20 20 20 2	55	230-232	C ₂₄ H ₁₆ N ₃ O ₂ Cl (benzene)	69.70 (69.73)	3.80	10.10	1	8.40 (8.47)	3.40 (3H, s), 5.52 (1H, s), 7.00-8.34 (12H, m)	28.70, 119.67, 120.49, 121.30, 121.50, 122.64, 122.70, 123.40, 125.89, 126.42, 126.56, 128.29, 132.06, 132.60, 133.82, 142.25, 144.60, 146.66, 152.85, 155.46, 155.79, 159.99, 174.67, 197.40
	2d	54	245-247	C ₂₅ H ₁₈ N ₃ O ₂ Cl (benzene)	70.20 (70.25)	4.15 (4.21)	9.80 (9.83)	1	8.10 (8.19)	2.90 (2H, s), 3.40 (3H, s), 5.52 (1H, s), 7.00-8.29 (12H, m)	27.80, 28.70, 119.67, 120.49, 121.30, 121.50, 122.64, 122.70, 123.40, 125.83, 126.42, 126.56, 128.29, 132.06, 133.82, 142.25, 144.60, 146.66, 152.85, 155.46, 155.79, 159.99, 174.67, 197.40

quinazoline and indan rings. The ¹³C NMR spectrum of 4a (DMSO- d_6) showed the following signals: 8 22.79, 27.84, 28.81, 119.67, 120.49,125.83, 126.42, 126.56, 128.29, 132.06, 133.82, 142.25, 144.60, 146.66, 152.85, 155.46, 155.79, 159.99, 174.67, 197.45. For the rigid identification of compounds 4a-d, an unequivocal syntheses for 4a-d were established by the reaction of cyclic ketones namely 1-indanone 2a, 1-tetralone 2b, fluorenone 2c and anthrone 2d with mercaptoacetic acid in the presence of p-toluenesulfonic acid in toluene to afford the spiroheterocyclic derivatives 6a-d in good yields²⁷ (Scheme II), which were subsequently reacted with 3-amino-2-methyl-3Hquinazolin-4-one 1 to afford the target compounds 4a-d. For the synthesis of the new spiroazetidinone derivatives 5a-d, compounds, 3a-d reacted with chloroacetyl chloride in benzene in the presence of triethylamine as a catalyst to give the corresponding spiroazetidinone derivatives 5a-d in good yields (Scheme I). The structures of compounds 5a-d were elaborated on the basis of their elemental analyses and spectroscopic data (Table I). The IR spectrum of compound 5a showed characteristic strong absorption bands at 1730 cm⁻¹, 1700 cm⁻¹ corresponding to the stretching vibrations of the carbonyl group of the azetidinone ring and the quinazoline ring respectively, and at 790 cm⁻¹ for C-Cl stretching vibration. The ¹H NMR spectrum of 5a (DMSO- d_6) showed the following signals: δ 2.43 (2H, t) for the methylene protons of the indan moiety, 2.59 (2H, t) for the benzylic methylene protons of the indan residue, 3.39 (3H, s) for methyl at C₂ of the quinazoline ring, 5.82 (1H, s) the proton at C₃ of the azetidinone ring, 7.00-8.4 ppm (8H, m) for the aromatic protons of the quinazoline and indan rings. The 13C NMR spectrum of 5a (DMSO- d_6) showed the following signals: δ 22.60, 25.84, 28.81, 119.67, 120.49, 125.83, 126.2, 126.56, 128.29, 132.06, 133.82, 142.25, 144.60, 146.66, 152.85, 155.46, 155.79, 159.99, 174.67, 197.40 ppm.

Experimental Section

The time required for completion of each reaction was monitored by TLC. Melting points are uncorrected. ¹H NMR (δ, ppm) spectra were measured on an EM-360 90-MHz spectrophotometer using TMS as internal standard. A Varian FT-80

was used to obtain all 13 C NMR (δ , ppm) spectra. IR (ν , cm $^{-1}$) spectra were recorded on a Pye-Unicam SP 200-G spectrophotometer. Elemental analyses were determined on a Perkin-Elmer 240 C microanalyser.

Synthesis of 3-amino-2-methyl-3H-quina-zolin-4-one 1. This compound was prepared according to the reported method²⁸.

Synthesis of cycloalkylidene-3-aminoquina-zolin-4-one derivatives (3a-d): General Procedure. Each compound 2a-d (0.01 mole) was fused with compound 1 (0.01 mole) for 1/2 hr, then 25 mL of absolute ethanol was added to the reaction mixture. The reaction mixture was refluxed for 6 hr, cooled to room temperature whereby compounds 3a-d were precipitated, filtered off and dried. Yields, melting points and spectral analyses are depicted in Table I.

Synthesis of spirothiazolidinone derivatives 4a-d: General procedure. Each compound 3a-d (0.001 mole) was dissolved in 25 mL of dry benzene and to this solution, mercaptoacetic acid (0.001 mole) was added. Then the reaction mixture was refluxed for 8 hr. At the end of the refluxing time, the solvent was removed by distillation and the residue was poured into 50 mL of 20% sodium carbonate solution, whereby compounds 4a-d were precipitated, filtered off and dried. The results are given in Table I.

Synthesis of spiroazetidinone derivatives 5a-d: General Procedure. Each compound 3a-d (0.001 mole) was dissolved in 25 mL of dry benzene and to this solution, chloroacetyl chloride (0.001 mole) and triethylamine (0.001 mole) were added. Then the reaction mixture was stirred at room temperature for 10 hr. At the end of stirring time, the solvent was removed by distillation and the residue was poured into 25 mL of iced water, whereby compounds 5a-d were precipitated, filtered off and dried. Results are shown in Table I.

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