Note

Synthesis and antibacterial activity of nitrofurylvinylquinazolinones

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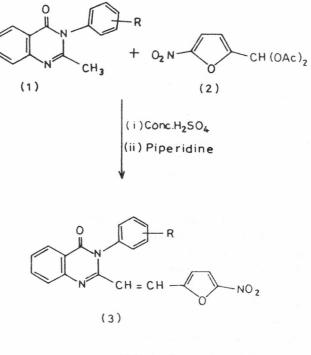
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The reaction of 3-aryl-2-methyl-quinazolin-4-ones 1 with acidified nitrofurfuraldehyde diacetate 2 in the presence of piperidine affords 3-aryl-2-[(5-nitro-2-furyl)vinyl]quinazolin-4-ones 3 in 50-80% yield. These compounds have been characterized by nitrogen analysis, and IR, ¹H NMR and mass spectral data. The newly synthesized compounds have been screened for their antibacterial activities.

Quinazolinone derivatives exhibit a wide range of antiparkinson¹, antibacterial¹, anthelmintic², CNS depressant³, antitubercular⁴, analgesic⁵, antiinflammatory⁶, hypoglycemic⁷, non-steroidal⁸ and fungicidal activities. The 4(3H)-quinazolinone⁹ moiety constitutes an important structural feature in several useful sedatives, hypnotic and anticonvulsant drugs such as Methaqualone and Mecloqualone. Further, nitrofuran derivatives are being used as potential chemotherapeutic agents¹⁰⁻¹¹. In view of these observations and in continuation of our studies on chemotherapeutically important nitrofurans¹²⁻¹⁴, it was thought of interest to synthesize novel 3-aryl-2-[(5-nitro-2-furyl)vinyl]-quinazolin-4-ones **3**.

For the present work, various anilines were condensed with 2-methyl-3,1-benzoxazin-4-one in the presence of phosphorus oxychloride employing toluene as solvent. Such condensations afforded 3aryl-2-methyl-quinazolin-4-ones 1 in 60-70% yield. The required 2-methyl-3,1-benzoxazin-4-one was prepared by the cyclization of anthranilic acid with acetic anhydride. Quinazolinones 1 were characterized by their melting points and reference to literature¹⁵. 3-Aryl-2-methylquinazolin-4-ones 1 were then condensed with acidified nitrofurfuraldehyde diacetate 2 in the presence of a base such as



Scheme I

piperidine to yield **3** (Scheme I). Nitrofurfuraldehyde diacetate **2**, required for this reaction, was obtained by the literature method¹⁶. Characterization and spectral data of compounds **3** are given in Table I.

Antibacterial Activity. All the newly synthesized compounds were screened for their antibacterial activity against *Bacillus subtilis*, *Staphylococcus aureus*, *Escherichia coli* and *Pseudomonas aeruginosa*. All of them showed antibacterial activity at 5 μ g/mL to a lesser degree than that of Furacin, a standard drug used for comparison.

Experimental Section

General. Melting points were determined in open capillaries and are uncorrected. IR spectra were recorded on a Shimadzu FT-IR 157 infrared spectrophotometer, ¹H NMR spectra in DMSO- d_6 on a 90 MHz Perkin-Elmer R-32 instrument using TMS as internal reference (chemical shifts in δ , ppm) and mass spectra on a Jeol JMS-D 300 mass spectrometer.

Compd	R	Yield (%)	m.p. °C	Nature of the compound		Mol. Formula
3a 3b 3c 3d 3e 3f 3g	H p-Bromo [†] p-Methyl m-Chloro 3-Chloro-4-fluoro [‡] o-Methoxy m-Methyl	80 81 78 57 81 51 68	254 229 213 231 196 191 197 242	Orange yellow crystals Yellow crystals Orange crystals Yellowish shining needles Yellow crystals Light yellow crystals Yellow crystals	5 N) 5 7	$\begin{array}{c} C_{20}H_{13}N_3O_3\\ C_{20}H_{12}BrN_3O_3\\ C_{21}H_{15}N_3O_3\\ C_{20}H_{12}ClN_3O_3\\ C_{20}H_{12}ClN_3O_3\\ C_{20}H_{11}ClFN_3O_3\\ C_{21}H_{15}N_3O_4\\ C_{21}H_{15}N_3O_3\\ C_{11}H_{15}N_3O_3\\ C_{11}H_{1$
3h 3I	<i>p</i> -Nitro 2,5-Dichloro	50 53	242 226	Brownish orange crystals Light yellow crystals		$\begin{array}{c} C_{20}H_{12}N_4O_5\\ C_{20}H_{11}Cl_2N_3O_3 \end{array}$

Table I-Characterization data of 3-aryl-2-[(5-nitro-2-furyl)vinyl]quinozolin-4-ones 3*

*Satisfactory elemental analyses (±0.3%) were obtained.

[†]IR: 1675 (C=O), 1600 (C=C), 1550 and 1340 cm⁻¹ (NO₂ asym and sym)

[‡]IR: 1681 (C=O), 1600 (C=C), 1545 and 1340 cm⁻¹ (NO₂ asym and sym)

3-Aryl-2-[(5-nitro-2-furyl)vinyl] quinazolin-4ones 3: General procedure. An equimolecular mixture of suitably substituted quinazolinone 1 (0.01 mole) and 5-nitro-2-furfuraldehyde diacetate (2.43 g, 0.01 mole) in absolute ethanol (20 mL) was treated first with conc. sulphuric acid (0.5 mL) followed by the *addition of piperidine* (10 mL). The mixture was heated under reflux for 1-2 hr. On cooling the reaction mixture, the resultant product was filtered, washed and recrystallized from dimethylformamide. The characterization data of compounds **3** are given in Table I.

As a representative case the spectral data of 3c are given below: IR: 1680 (C=O), 1600 (C=C), 1540 and 1340 cm⁻¹ (NO₂ *asym* and *sym*). ¹H NMR (DMSO-*d*₆): 3c δ 2.5 (s, 3H, *p*-tolyl protons), 6.5-6.7 (d, 2H, vinyl protons), 7.1-7.4 (m, 8H, aromatic protons), 7.7-7.9 (d, 2H, furan protons). MS: m/z 359 (M⁺), 313 (M-NO₂).

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