Claisen rearrangment of 3-(prop-2ynylsulfanyl)-1,2,4-triazinone: A convenient route to 2-methylthiazolo[3,2-*b*][1,2,4]triazinone

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6-Methyl-3- (prop-2-ynylsulfanyl)-1,2,4-triazin-5(2*H*)one 1 on refluxing with hexamethyl phophoric triamide (HMPT), in the presence of catalytic amount of PdCl₂(PhCN)₂ affords 2,6-dimethylthiazolo [3,2b][1,2,4] triazin-7-one 10 via an interesting Claisen rearrangment. This structure has been confirmed by unambiguous synthesis of 10.

Palladium catalyzed S \rightarrow N allylic rearrangment of allylsulfanyl pyrimidinone and allylsulfanyl-1,2,4triazinone have been studied extensively owing to their synthetic utility^{1,2}. However examples of S \rightarrow N propynylic rearrangments are very scarce. The S \rightarrow N propynylic rearrangment of 2-(prop-2 ynylsulfanyl) benzimidazole is typical example^{3,4}.

Recently we have demonstrated that base⁵ and Pd-salt^{6,7} very effectively catalyze the cyclization of prop-2-ynylsulfanyl-1,2,4-triazinone to corresponding 3-substituted thiazolo [1,2,4] triazinone. In this pattern of cyclization and aromatization and other methods^{8,9} to synthesize thiazolotriazine, only 3-substituted thiazolo [1,2,4] triazinone can be obtained. Here we wish to report two independent routes to synthesis 2,6-dimethylthiazolo [3,2-*b*][1,2,4] triazin-7-one **10**.

6-Methyl-3(prop-2-ynylsulfanyl)-1,2,4-triazin-5one 1 was refluxed in HMPT with catalytic amount of Pd Cl₂(PhCN)₂ for 32 hrs. After evaporation of solvent under reduced pressure, the crude material was subjected directly to column chromatorgraphy to obtain a crystalline compound as a major product.

Taking into consideration the problem of regiochemistry, first we thought of alkylation at the N-2 vs. N-4, by direct nucleophilic addition of amide to



acetylenic moiety and subsequent isomerization of either intermediate 2 or 3 to either 4 or 5 respectively. This pattern of cyclization and isomerizaiton by base has been noticed previously⁶. Comparison of spectroscopic data of obtained compound with those known 4^5 and 5^7 showed considerable differences. Therefore we considered thermal and catalyzed [3,3] signatropic rearrangment of 1 to either 6 or 7 with the subsequent cyclization to either 8 or 9 followed by isomerization to either 10 or 11 respectively. (Scheme I).

Note



Scheme I

To prove this pattern of one pot [3,3] sigmatropic shift, cyclization and isomerization we decided to synthesize 2-substituted thiazolotriazinone 10 or 11 unambiguously.

6-Methyl-3-thioxo-1,2,4-triazin-5-one **12** was refluxed with excess of hexamethyl disilazane (HMDS) and a catalytic amount of $(NH_4)_2 \text{ SO}_4^{10}$ and thereafter propynylbromide was added. After usual work-up a single (TLC) compound was isolated in fairly good yield which could be either **6** or **7**.

The corresponding N-propynyl compound 6 or 7 was cyclized and isomerized by acid catalysis¹¹. to afford the identical compound which was obtained by refluxing 1 with HMPT (10 or 11) (Scheme II). The structure 10 or 11 was carefully discriminated by the comparison of its UV spectrum with those of well established 4 (λ max 280 nm)⁵ and 5 ((max 298 nm)⁷. 3,4-Disubstituted-1,2,4-triazin-5 ones are known to show the absorption maxima at the longer wavelenghts compared to 2,3-disubstituted compounds⁹ 5 also showed the absorption maxima at the longer wavelength compared to 4. The UV spectrum of compounds 10 or 11, was quite similar to that of 4. Therefore we concluded that in $S \rightarrow N$ propynylic rearragment in HMPT, the propynylic group migrates to N-2 of 1,2,4-triazine regioselectively to afford 6 which subsequently cyclizes and isomerizes to 10. N-Alkylation of 1,2,4-triazine 12 in the presence of HMDS also occures on N-2 selectively, leading to 2-propynyl derivative 6. Acid catalyzed cyclization and isomerization of the latter led to 10.



Scheme II

Experimental Section

Melting points were obtained on a Buchi 530 and are uncorrected. ¹HNMR spectra were recorded in δ ppm a Brucker Ac 80 spectrometer in dimethyl sulfoxide as solvent using TMS as internal standard. IR spectra were recorded on a Perkin-Elmer model 883 using KBr disk and mass spectra on a Varian CH-7.

2,6-Dimethyl thiazolo [**3,2-***b*][**1,2,4**] triazin-7one **10**—Compound **1** (1g,55 mmoles) and PdCl₂(PhCN)₂ (0.1g) were refluxed in HMPT (20 mL) for 32 hr. The solvent was evaporated off *in vacuo*. The crude mixture was washed with water and extracted with chlorofom. This solution was dried over anhyd. Na₂SO₄ and evaporated off. The residue obtained was subjected to column chromatography using CHCl₃: MeOH, 90:10 to afford the title compound. Yield 0.38g, (38%), m.p.168-69°; ¹HNMR (CDCl₃): δ 2.38(s,6H,2Me), 6.43(s,1H,CH of thiazole ring). IR: (KBr): 3420, 2928, 1640 (amide carbonyl), 1489 and 1372 cm⁻¹; UV (CHCl₃): (λ max 282 nm. M.S: m/z, M⁺, 181(25), 180(55), 156(18), 139(100), 72, 40(95).

6-Methyl-2N-Propynyl-3-thioxo-1,2,4-triazin-5-one 6. 6-Methyl-3-thioxo-1,2,4-triazin-5-one 12-(0.36g, 2.5 mmoles), was refluxed with hexamethyldisilazane (HMDS) (0.7 mL, excess) and a catalytic amount of (NH₄)₂SO₄ (12.5 mg) until a clear solution was obtained. Excess HMDS was removed by disitlation in vacuo. The catalytic amount of I_2 and propynyl bromide (0.18g, 1 mmole) in CH₃CN (5 mL) was added and the reaction mixture was refluxed for further 4 hr. After evaporation of solvent in vacuo, the residue was treated with water to afford the spectroscopic pure product. Yield (0.4 g, 93%), m.p. 98-99°; ¹HNMR 3.1(t,1H,-C-CH, (DMSO-d₆): $\delta 2.1(s, 3H, Me),$ J=2.1Hz), 3.95 (d, 2H, CH₂, J=2.1 Hz); IR: (KBr): 3417, 3258, 2928, 1640, 1590, 1459, 1033 cm⁻¹; M.S: m/z, M⁺, 181(22), 143(18), 140(85), 72(100), 71(75), 80(90), 55(65).

Acid catalyzed cyclization of 6 to 10—Compound 6 (0.25g, 1.38 mmole) was dissolved in conc H_2SO_4 . The reaction mixture was kept at 50 C for 5 hrs. It then poured onto crushed ice and the solution was neutralized by adding NaOH. The precipitated solid was crystallized from EtOH to afford spectroscopic pure 10. Yield 0.175g, (70%), mp 168-69.

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