

## Rapid Communication

### A rapid one-pot synthesis of 5-substituted-2-mercapto-1,3,4-thiadiazoles using microwaves

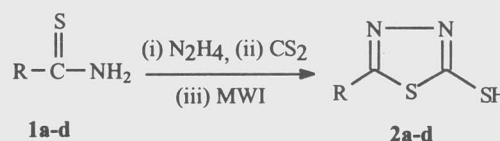
Mazaahir Kidwai\* & Kumar Ranjan Bhushan

Department of Chemistry, University of Delhi, Delhi 110 007, India

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A rapid one-pot synthesis is described for 5-substituted-2-mercapto-1,3,4-thiadiazoles **2a-d** under microwave irradiation starting from thioamides **1a-d**.

In recent times there has been much interest in the development of microwave assisted technique in the synthesis of heterocycles.<sup>1</sup> In view of the utility of microwave irradiation<sup>2</sup> and the biological importance<sup>3</sup> of 5-substituted-2-mercapto-1,3,4-thiadiazoles, it was thought worthwhile to develop a convenient method for the synthesis of the title compounds. One of the title compounds 5-methyl-2-mercapto-1,3,4-thiadiazole is a side-chain at C-3 position of the antibiotic Cefazolin sodium.<sup>4</sup> 5-Substituted-2-mercapto-1,3,4-thiadiazoles have been synthesised earlier from esters,<sup>1f-g</sup> orthoesters, and thioacid esters,<sup>5-8</sup> in low yields via multistep and lengthy procedures which involve difficult work-ups. We report herein one-pot, rapid synthesis of the title compounds **2** from thioamides **1** in high yields using microwave irradiation (cf. Scheme I).



Scheme I

Thioamides **1** were treated with hydrazine hydrate followed by CS<sub>2</sub> solution in the cold. The reaction mixture was then irradiated in a microwave oven to yield the title compounds **2**. This was evidenced by the disappearance of the IR absorption band in the region 3250-3400 due to NH<sub>2</sub> and the appearance of a band at 2550-2650 cm<sup>-1</sup> due to SH group. The <sup>1</sup>H NMR showed signal for SH proton appeared at δ 12.7-13.1 and the peak for the NH<sub>2</sub> protons was absent. This is the first report on one-pot rapid synthesis of the title compounds **2** where microwave technique has been utilized. The analytical and spectral data of the products **2a-d** are given in Table I.

Table I — Analytical and spectral data of compounds **2a-d**

Compd	R	m.p. (°C) (lit)	Reaction period (min)	Yield (%)	M <sup>+</sup> observed (expected)	<sup>1</sup> H NMR (DMSO-d <sub>6</sub> + CDCl <sub>3</sub> ) δ, ppm
<b>2a</b>	CH <sub>3</sub>	185-87 (186-87) <sup>9</sup>	1.0	88	132 (132)	2.7 (s, 3H, CH <sub>3</sub> ), 12.7 (s, 1H, SH)
<b>2b</b>	C <sub>6</sub> H <sub>5</sub>	216-17 (215-17) <sup>10</sup>	1.5	86	194 (194)	7.4-7.7 (m, 5H, Ar-H), 12.9 (s, 1H, SH)
<b>2c</b>	4-C <sub>5</sub> H <sub>4</sub> N <sup>a</sup>	218-20	1.5	79	194 (195)	7.7 (d, 2H, 3'-CH & 5'-CH), 8.75(d, 2H, 2'-CH and 6'-CH), 13.0 (s, 1H, SH)
<b>2d</b>	3-C <sub>5</sub> H <sub>4</sub> N <sup>b</sup>	80-82	2.0	81	195 (195)	7.55 (t, 1H, 5'-CH), 8.20 (d, 1H, 4'-CH), 8.75 (d, 1H, 6'-CH), 9.10 (s, 1H, 2'-CH), 13.1 (s, 1H, SH)

<sup>a</sup> Anal. Calcd for C<sub>7</sub>H<sub>7</sub>N<sub>3</sub>S<sub>2</sub>: C, 43.07; H, 2.56; N, 21.53. Found: C, 43.20; H, 2.48; N, 21.55%.

<sup>b</sup> Anal. Calcd for C<sub>7</sub>H<sub>5</sub>N<sub>3</sub>S<sub>2</sub>: C, 43.07; H, 2.56; N, 21.53. Found: C, 42.95; H, 2.51; N, 21.45%.

### Experimental Section

**General procedure for preparation of 5-substituted-2-mercapto-1,3,4-thiadiazoles 2a-d :** The substituted thioamide 1 (0.02 mole) was dissolved in 10 mL DMF in an Erlenmeyer flask (100 mL) and hydrazine hydrate (0.02 mole) added to it slowly with stirring followed by the dropwise addition of CS<sub>2</sub> (0.021 mole) maintaining the temperature of the reaction mixture at 0-5 °C. Thereafter, the reaction mixture was subjected to microwave irradiation (MWI) at 450 watts. TLC was run after every half a min. to check the progress of the reaction. On completion of the reaction, the reaction mixture was poured over crushed ice and treated with HCl to bring down the pH to 5. The solid obtained was collected and washed with water. It was dried and recrystallised from acetone.

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### References

- 1 (a) Kidwai M & Kumar P, *J Chem Res (S)*, **1996**, 254.
- (b) Kidwai M & Goel Y, *Polyhedron*, **15**, **1996**, 2819.
- (c) Kidwai M, Kumar R & Kumar P, *Indian J Chem*, **35B**, **1996**, 1004.
- (d) Kidwai M, Kumar P & Kohli S, *J Chem Res (S)*, **1997**, 24.
- (e) Kidwai M & Kumar P, *J Chem Res (S)*, **1997**, 178.
- (f) Kidwai M, Kumar R & Goel Y, *Main Gp Met Chem*, **20**, **1997**, 367.
- (g) Kidwai M, Bhushan K R, Kumar P & Kumar R, *Monatsh Chemie*, **128**, **1997**, 1291.
- 2 Caddick S, *Tetrahedron*, **51**, **1995**, 10403.
- 3 Mullican M D, Wilson M W, Conner D T, Kostlan C R, Schrier D J & Dyer R D, *J Med Chem*, **36**, **1993**, 1090.
- 4 Kariyone K, Harada H, Kurita M & Takano T, *J Antibiot*, **23**, **1970**, 131.
- 5 Kariyone K, Harada K & Kurita Y, *Jap Pat*, 7207, 371 (1972); *Chem Abstr*, **76**, **1972**, 153749v.
- 6 Kariyone K, Harada K & Kurita Y, *Japan Pat*, 7207, 370 (1972); *Chem Abstr*, **76**, **1972**, 153750p.
- 7 Kariyone K, Harada H, Kurita M, Ueda Y, Furuhashi T, Nakamura H & Watanabe H, *Ger Offen*, **2**, 162, 324 (1972); *Chem Abstr*, **77**, **1972**, 140085w.
- 8 Kariyone K, Harada H, Kurita M, Ueda Y & Furuhashi T, *Brit Pat*, 1, 383, 292 (1975); *Chem Abstr*, **83**, **1975**, 43340s.
- 9 Sandstrom J & Wennerbeck I, *Acta Chem Scand*, **20**, **1966**, 57.
- 10 Young R W & Wood K H, *J Am Chem Soc*, **77**, **1955**, 400.