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### Note

# Synthesis and fungicidal activity of 3-aryl-2-(4'-aryl thiazol-2'-ylaminomethyl) quinazol-4 (3*H*)-ones

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A series of 3-aryl-2-(4'-aryl thiazol-2'-ylaminomethyl) quinazol-4(3*H*)-ones **3a-p** have been prepared by condensing 3-aryl-2-chloromethylquinazol-4(3*H*)-ones with 2-amino-4-substituted phenylthiazoles. Another group of 3-aryl-6, 8-dibromo-2-(4'-arylthiazol-2'-ylaminomethyl) quinazol-4 (3*H*)-ones **3q-gg** have also been synthesized from 3-aryl-6, 8-dibromo-2-chloromethylquinazol-4(3*H*)-ones and 2-amino-4-substituted phenylthiazoles in the same manner. The prepared compounds have been characterised by spectral data. Their antifungal activity has also been determined.

Quinazolones possess antiinflammatory, analgesic and antitumor activities<sup>1,2</sup>. Some 2-alkyl- and 3aryl quinazol-4(3*H*)-ones also show antibacterial as well as tuberculostatic activities <sup>3-5</sup>. Inspired by these observations some new quinazolone derivatives having halogen in the quinazolone moiety were prepared with a view to study their fungicidal activities.

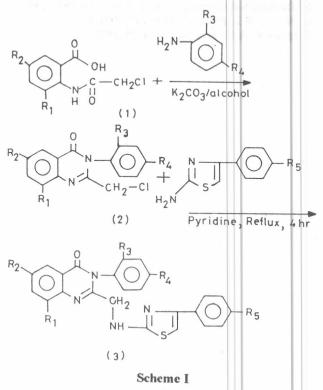
Quinazolone derivatives 3 showed a strong band at  $\approx 1668$  cm<sup>-1</sup> indicating the presence of an

-N-C=O group. Bands at around 1532, 1488,

435 cm<sup>-1</sup> were observed which were attributed to thiazole and phenyl ring vibration<sup>6</sup>. <sup>1</sup>H NMR spectra of compounds **3** in DMSO- $d_6$  exhibited a singlet at  $\delta$  2.5 integrating for 2 protons and was assigned to a methylene substituted in position-2 of the quinazolone nucleus. Other signals appeared in the aromatic region at  $\delta$  6.5-8.2 integrating for 10 to 14 protons depending on the substitution<sup>7</sup>. The reaction path is shown in **Scheme I**. Analytical data of quinazolones are given in **Table I**.

#### **Experimental Section**

General. The purity of the compounds was checked by TLC. Melting points were determined



using a Melting Point apparatus and are uncorrected. IR spectra of the compounds were recorded on a Perkin-Elmer infrared-283 spectrophotometer in KBr discs.

Preparation of *N*-chloroacetyl-3, 5-dibromoanthranilic acid  $1^{8, 9}$ . 3,5-Dibromoanthranilic acid (0.01 mole) was treated with chloroacetyl chloride (0.015 mole) in benzene with 4 to 5 drops of pyridine. The mixture was refluxed for 6 hr and the solid thus obtained was recrystallised from acetone-ethyl alcohol mixture (1:1) and to give 1 which was used in subsequent steps, yield 59%, mp 198°C.

Following the same procedure Nchloroacetylanthranilic acid (1;  $R_1=R_2=H$ ) as prepared, yield 72%, mp 169°C.

**Preparation of 2-chloromethyl-3-substitutedphenylquinazol-4** (3H) **ones**<sup>10,</sup> <sup>11</sup> 2. N-Chloroacetyl-3, 5-dibromoanthranilic acid 1 was refluxed for 6 hr with equimolecular amount of an aromatic amine in the presence of anhydrous  $K_2CO_3$  and ethanol (20 mL). The reaction mixture

1	2	n	5
1	2	υ	2

	in the second		R <sub>5</sub>	mgicidal data of 3-aryl-2 Mol. formula	rmp* (°C)	thiazol-2'-ylamino methyl) quinazol-4 (3H) one					
Compd	R <sub>3</sub>	R <sub>4</sub>				N ( Found	%) Calcd	S (% Found	%) Calcd	% of fungal inhibi- tion	Yield (%)
3a	Н	н	CH3	C <sub>25</sub> H <sub>20</sub> ON₄S	193	12.87	13.25	7.88	7.56	65.3	52.0
3b	H	Н	C1	C24H17ON4SCI	172	12.99	12.67	6.98	7.21	68.8	55.0
3c	CH <sub>3</sub>	Н	CH <sub>3</sub>	C <sub>26</sub> H <sub>22</sub> ON <sub>4</sub> S	189	12.83	12.81	7.01	7.32	64.9	56.2
3d	CH <sub>3</sub>	Н	C1	C25H19ON4SCI	153	12.15	12.24	6.81	6.99	66.5	49.1
3e	OCH <sub>3</sub>	Η	OCH <sub>3</sub>	C26H22O3N4S	208	11.52	11.94	6.52	6.82	71.2	51.7
3f	OCH3	Н	C1	C25H19ON4SCI	154	11.32	11.82	6.48	6.75	73.6	60.0
3g	Cl	Н	CH <sub>3</sub>	C25H19ON4SCI	188	12.01	12.24	6.53	6.99	71.8	55.5
3h	C1	Н	Cl	C24H16ON4SCl2	138	11.63	11.71	6.48	6.69	76.5	57.1
31	Н	CH <sub>3</sub>	CH <sub>3</sub>	C <sub>26</sub> H <sub>22</sub> ON <sub>4</sub> S	158	12.65	12.81	7.06	7.32	65.5	58.0
3j	Н	CH <sub>3</sub>	OCH <sub>3</sub>	C <sub>26</sub> H <sub>22</sub> O <sub>2</sub> N <sub>4</sub> S	152	12.29	12.36	6.90	7.06	66.9	52.0
3k	Н	CH <sub>3</sub>	C1	C <sub>25</sub> H <sub>19</sub> ON <sub>4</sub> SCl	137	12.05	12.24	6.80	6.99	67.2	48.0
31	Н	C1	CH <sub>3</sub>	C <sub>25</sub> H <sub>19</sub> ON <sub>4</sub> SCl	163	11.97	12.24	6.72	6.99	66.2	53.0
3m	Н	C1	OCH <sub>3</sub>	C25H19O2N4SC1	135	11.53	11.82	6.57	6.75	72.9	54.0
3n	Н	Cl	C1	C24H16ON4SCl2	142	11.29	11.71	6.32	6.69	76.4	52.0
30	Н	OCH <sub>3</sub>	CH <sub>3</sub>	C26H22O2N4S	181	12.12	12.36	6.81	7.06	67.1	51.0
3p	Н	OCH <sub>3</sub>	Cl	C <sub>25</sub> H <sub>19</sub> O <sub>2</sub> N <sub>4</sub> SCl	167	11.21	11.82	6.52	6.75	73.8	61.5
Bq	Н	Н	CH <sub>3</sub>	C <sub>25</sub> H <sub>18</sub> N <sub>4</sub> OSBr <sub>2</sub>	151	9.45	9.98	5.43	5.70	71.5	52
Br	Н	Н	C1	C24H15ON4SBr2Cl	154	9.07	9.31	5.02	5.32	73.5	53.
Bs	CH <sub>3</sub>	Н	CH <sub>3</sub>	$C_{26}H_{20}ON_4SBr_2$	178	9.20	9.41	5.08	5.37	75.9	54.
3t	CH <sub>3</sub>	Н	OCH <sub>3</sub>	$C_{26}H_{20}O_2N_4SBr_2$	172	8.88	9.09	4.95	5.19	75.8	53.9
3u	OCH <sub>3</sub>	Н	OCH <sub>3</sub>	C <sub>26</sub> H <sub>20</sub> O <sub>3</sub> N <sub>4</sub> SBr <sub>2</sub>	191	8.36	8.96	4.78	5.12	81.4	55.0
3v	OCH <sub>3</sub>	Н	C1	C <sub>25</sub> H <sub>17</sub> O <sub>2</sub> N <sub>4</sub> SBr <sub>2</sub> Cl	207	8.58	8.86	4.81	5.06	82.5	57.2
3x	Cl	Н	OCH <sub>3</sub>	C <sub>25</sub> H <sub>17</sub> O <sub>2</sub> N <sub>4</sub> SBr <sub>2</sub> Cl	166	8.83	9.09	4.93	5.19	76.2	49.0
3y	C1	Н	C1	C24H14ON4SBr2Cl2	152	8.63	8.80	4.13	5.06	79.9	51.1
3z	Н	CH <sub>3</sub>	CH <sub>3</sub>	C <sub>26</sub> H <sub>20</sub> ON <sub>4</sub> SBr <sub>2</sub>	151	9.26	9.41	5.02	5.37	72.9	59.
aa	Н	CH <sub>3</sub>	OCH <sub>3</sub>	C <sub>26</sub> H <sub>20</sub> O <sub>2</sub> N <sub>4</sub> SBr <sub>2</sub>	149	8.57	8.73	4.73	4.99	74.5	50.
bb	Н	CH <sub>3</sub>	C1	C <sub>25</sub> H <sub>17</sub> ON <sub>4</sub> SBr <sub>2</sub> Cl	127	8.87	9.10	4.94	5.19	76.5	60.
lee	Н	C1	CH <sub>3</sub>	C25H17ON4SBr2Cl	173	8.92	9.10	4.88	5.19	73.8	48.
dd	Н	C1	OCH <sub>3</sub>	C <sub>25</sub> H <sub>17</sub> O <sub>2</sub> N <sub>4</sub> SBr <sub>2</sub> Cl	148	8.32	8.87	4.78	5.66	82.3	56.
lee	Н	C1	C1	C24H14ON4SBr2Cl2	136	8.39	8.80	4.77	5.03	85.9	59.
Bff	Н	OCH <sub>3</sub>	OCH <sub>3</sub>	C <sub>25</sub> H <sub>20</sub> O <sub>3</sub> N <sub>4</sub> SBr <sub>2</sub>	206	8.71	9.13	5.02	5.22	83.4	54.
3gg	Н	OCH <sub>3</sub>	C1	C25H17O2N4SBr2Cl	162	8.68	8.86	4.85	5.06	82.4	52

\*All the compounds were recrystallised from ethanol.

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was filtered and ethanol evaporated. The residue was washed thoroughly with boiling water and recrystallised from acetone-ethanol mixture (1:1). Melting point of the synthesized compound was comparable with that reported<sup>11</sup> for **2**, yield 57.5%, mp 177°C.

In a similar manner other members of the series **2** were prepared. The yield varied from 53. 5 to 61.5%.

**Preparation of 3-aryl-2-(4'-arylthiazol-2'ylaminomethyl)quinazol-4(3H)-one-3**. Compound 2 (0.01 mole) and 2-amino-4-substitutedphenylthiazole (0.01 mole) in pyridine were refluxed for 4 hr, and thereafter excess of pyridine was evaporated. The residue was mixed with crushed ice, filtered, washed and recrystallised from ethanol. Analytical and fungicidal activity data of the synthesized compounds are given in **Table I**.

Antifungal activity. For evaluating the antifungal activities of the synthesized compounds, poisoned food technique was adopted<sup>12</sup>. The quinazolone derivatives were tested for their toxicity towards *Helmonthsporium sativum* and the % of inhibition was calculated from the expression given below and the data are recorded in **Table I**.

% of inhibition =  $(1-X/Y) \times 100$ where X = growth with drug, and Y = growth in control.

The data indicate that quinazolones with bromoand methoxy-substituents are more active than the corresponding unsubstituted ones.

**<sup>3</sup>a-p**:  $R_1 = R_2 = H$ **3q-3gg**:  $R_1 = R_2 = Br$ 

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