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Efficient cartridge purification for producing high molar activity ^{18}F -glycoconjugates via oxime formation

Abbreviated title (max 45 letters and spaces): High molar activity ^{18}F -glycoconjugates with cartridge purification

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Abstract: Introduction: ^{18}F -fluoroglycosylation via oxime formation is a chemoselective and mild radiolabeling method for sensitive molecules. Glycosylation can also improve the bioavailability, in vivo kinetics, and stability of the compound in blood, as well as accelerate clearance of biomolecules. A typical synthesis procedure for ^{18}F -fluoroglycosylation with [^{18}F]FDG (2-deoxy-2- ^{18}F fluoro-D-glucose) and [^{18}F]FDR (5-deoxy-5- ^{18}F fluoro-D-ribose) involves two HPLC (high performance liquid chromatography) purifications: one after ^{18}F -fluorination of the carbohydrate to remove its labeling precursor, and a second one after the oxime formation step to remove the aminoxy precursor. The two HPLC purifications can be time consuming and complicate the adaptation of the synthetic strategy in nuclear medicine applications and automated synthesis. We have developed a procedure in which SPE (solid phase extraction) and resin purification methods replace both of the needed HPLC purification steps.

Methods: We used [^{18}F]FDR and [^{18}F]FDG as prosthetic groups to radiolabel two aminoxy-modified model molecules, a tetrazine and a PSMA (prostate specific membrane antigen) inhibitor. After fluorination, the excess carbohydrate precursor was removed by derivatizing it with 4,4'-dimethoxytrityl chloride (DMT-Cl). The DMT moiety increases the hydrophobicity of the unreacted precursor making the separation from the fluorinated precursor possible with simple C18 Sep-Pak cartridge. For removal of the aminoxy precursor, we used a commercially available aldehyde resin (AminoLink, Thermo Fisher Scientific). C18 Sep-Pak SPE cartridge was used to separate [^{18}F]FDR and [^{18}F]FDG from the ^{18}F -fluoroglycoconjugate end product.

Results: [^{18}F]FDR and [^{18}F]FDG were efficiently purified from their precursors, free fluorine-18, and other impurities. The aldehyde resin quantitatively removed the unreacted aminoxy precursors after the oxime formation. The fluorine-18 labeled oxime end products were obtained with high radiochemical purity (> 99%) and molar activity (> 600 GBq/ μmol).

Conclusions: We have developed an efficient cartridge purification method for producing high molar activity ^{18}F -glycoconjugates synthesized via oxime formation.

Introduction

^{18}F -fluoroglycosylation has become a useful tool in fluorine-18 labeling. It has been used for the production of various ^{18}F -glycoconjugates, of for example of peptides, proteins, and other compounds, which are vulnerable to harsh reaction conditions [1-26]. Glycosylation also has a favorable impact on the biodistribution and cellular uptake properties of the radiotracer. The introduction of a carbohydrate moiety can improve the bioavailability, in vivo kinetics, stability in blood, and accelerate clearance of biomolecules conjugated with sugars [13, 27-34]. ^{18}F -fluoroglycosylation can be achieved with several methods but it is most frequently carried out via oxime formation, or by using other click reactions such as copper-catalyzed azide-alkyne cycloaddition (CuAAC). Advantages of the click glycosylation methods are their high chemoselectivity and the fact that they can be performed in aqueous solutions with high radiochemical yields. The downside, however, is that they require a preparative HPLC purification before the oxime formation step in order to remove the excess carbohydrate precursor. Due to the high concentration of glucose in the commercially available [^{18}F]FDG (2-deoxy-2-[^{18}F]fluoro-D-glucose) formulations, clinical grade [^{18}F]FDG is also usually purified with preparative HPLC before use in ^{18}F -fluoroglycosylation reactions. In order to produce high apparent molar activity end products, another HPLC purification is required after the ^{18}F -fluoroglycosylation to remove the excess aminoxy functionalized precursor of the second reaction step. The two HPLC purifications can be time consuming and will complicate the automation of the radiosynthesis for clinical production.

Oxime formation is a click reaction between aminoxy and carbonyl groups. [^{18}F]FDG and [^{18}F]FDR (5-deoxy-5-[^{18}F]fluoro-D-ribose) can be used as prosthetic groups to radiolabel wide variety of molecules by attaching aminoxy functional group to the desired molecule. The reaction with [^{18}F]FDG often requires harsher reaction conditions (high temperature, low pH) than reaction with [^{18}F]FDR. Both [^{18}F]FDG and [^{18}F]FDR isomerize between α - and β -anomer through intermediate

acyclic aldehyde (**Scheme 1**, [^{18}F]**3** and [^{18}F]**10**). This mutarotation has a dynamic equilibrium that is favored at high temperatures and for [^{18}F]FDG it is more efficient at acidic conditions. In [^{18}F]FDR, the fluorine in C-5 of the 5-membered ring prompts the formation of the acyclic form, making it more exposed for subsequent reaction to yield the oxime ether. For this reason, reactions with [^{18}F]FDR can be carried out at room temperature even at pH 6 with high yields [23].

Recently, we published a radiosynthesis where [^{18}F]FDR is used to radiolabel an aminoxy-functionalized tetrazine [35]. Tetrazines are base-labile molecules that do not tolerate the typical conditions for nucleophilic ^{18}F -fluorination well. The developed synthesis involved two preparative HPLC purifications. The first one after the ^{18}F -fluorination step, and a second one after the oxime formation step. Here, our aim is to develop an HPLC-free purification method, which would be applicable for [^{18}F]fluoroglycosylation reactions with both [^{18}F]FDR and [^{18}F]FDG. This was achieved by developing efficient cartridge-based procedures for both purification steps (**Figure 1**). The first HPLC purification was replaced by a DMT-Cl (4,4'-dimethoxytrityl chloride) derivatization, followed by subsequent C18 Sep-Pak cartridge separation. DMT-Cl is commonly used in nucleoside and nucleotide chemistry to protect alcohol groups [36, 37]. Here we used the same method for conversion of the unreacted carbohydrate precursor to a more hydrophobic derivative, resulting in increased retention of the unreacted precursor complex in a regular C18 Sep-Pak cartridge and allowing its efficient separation from the ^{18}F -fluorinated carbohydrate. For the second purification step, the removal of the aminoxy precursor, we used a commercially available aldehyde resin (AminoLink, Thermo Fisher Scientific). The AminoLink resin is commonly used to purify biomolecules containing amine groups with reductive amination with cyanoborohydride (NaCNBH_3) in order to react the amine groups with the aldehyde groups on the resin [38]. In our protocol, the precursor contains an aminoxy group that rapidly reacts with the aldehyde groups without the need of any additional reagents. The AminoLink resin quantitatively trapped the unreacted aminoxy precursor, resulting in end products with high apparent molar activity. This is a significant

improvement as due to their similar retention properties in cartridge separation systems based on hydrophobic interaction or size exclusion between the stationary phase and the analyte, it has not been previously possible to separate the aminoxy precursor and the oxime end product using such methods.

Recently, radiosynthesis of [^{18}F]FDG conjugated PSMA (prostate specific membrane antigen) inhibitor was published by Wuest and his group. Since high apparent molar activity was required, HPLC purification was needed for removing the non-radiolabeled glucose-conjugates from the final product [3]. We tested our methodology by radiolabeling two different aminoxy functionalized compounds, the PSMA inhibitor with both [^{18}F]FDR and [^{18}F]FDG and a tetrazine for bioorthogonal radiolabeling using the inverse electron-demand Diels-Alder (IEDDA) reaction. The final radiolabeled products were characterized for radiochemical purity (%), molar activity ($\text{GBq } \mu\text{mol}^{-1}$) and amount of residual organic impurities. The ^{18}F -fluoroglycosylated end products were efficiently separated from free [^{18}F]fluoride, the radiolabeling precursors [^{18}F]FDR and [^{18}F]FDG, and non-radiolabeled organic impurities yielding the end products with > 99% radiochemical purity (RCP) and high molar activity (> 600 $\text{GBq } \mu\text{mol}^{-1}$) and without any detectable amounts of the contaminating non-radioactive carbohydrate conjugates.

Materials and methods

Unless otherwise noted, all reagents were purchased from commercial suppliers and used without further purification. All water used was ultrapure (> 18.2 $\text{M}\Omega \text{ cm}^{-1}$). Reference compound **6**, and precursors **1** and **4** were synthesized according to previously described procedures [24, 35].

Chemical and radiochemical purities were analyzed with analytical HPLC on a Shimadzu Prominence UFLC liquid chromatography system (Shimadzu Corporation, Kyoto, Japan) consisting of two LC-20AD pumps, SIL-20AHT autosampler, CTO-20AC column oven, SPD-20A UV/VIS detector (dual

mode with wavelengths 209 nm and 254 nm) and an external NaI scintillation crystal radiodetector (Ortec, Oak Ridge, TN, USA) operated at +0.90 kV. Separation was carried out on a Grace Alltima C18 column (5 μ m, 250 mm \times 4.6 mm) with a flow rate of 1 ml min⁻¹. The used HPLC eluents were A: 0.1% TFA (trifluoroacetic acid) in water and B: 0.1% TFA in acetonitrile. For compounds **3-7** and **10-12**, the gradient was 5 to 50% B (0-15 min). Retention times: 3.5 min ([¹⁸F]**3**), 10.9 min (**4**), 5.9 min (**5**), 11.8 min (**6 E**), 12.1 min (**6 Z**), 6.6 min (**7 E**), 7.0 min (**7 Z**), 3.1 min ([¹⁸F]**10**), 11.2 min (**11 cyclic**), 11.4 min (**11 E**), 11.5 min (**11 Z**), 6.2 min (**12 cyclic**), 6.5 min (**12 E**), 6.7 min (**12 Z**), 4.2 (pyridine), and 6.0 min (aniline). For compounds **1**, [¹⁸F]**2**, **8**, and [¹⁸F]**9**, the gradient was 40 to 80% B (0-15 min). Retention times: 14.0 min (**1**), 9.0 min ([¹⁸F]**2**), 11.9 min (**8**), 8.7 min ([¹⁸F]**9**), and 3.0 min (pyridine).

Thin layer chromatography was performed using acetonitrile-water (95:5) eluent and TLC Merck Silica gel 60 F₂₅₄ plates. The TLC plates were imaged with digital autoradiography on a FLA-5000 scanner (Fujifilm, Japan) at a nominal resolution of 50 μ m. Autoradiographic images were analyzed using AIDA 2.0 software (Raytest Isotopenmessgeräte GmbH, Straubenhardt, Germany). R_f values: 0.94 ([¹⁸F]**2**), 0.45 ([¹⁸F]**3**), 0.64 ([¹⁸F]**6**), 0.0 ([¹⁸F]**7**), 0.98 ([¹⁸F]**9**), 0.41 ([¹⁸F]**10**), 0.63 ([¹⁸F]**11**), and 0.0 ([¹⁸F]**12**).

Radiosyntheses were carried out in a semi-automated synthesis unit (DM Automation, Stockholm, Sweden). ¹⁸F-Fluoride was produced on an IBA Cyclone 10/5 cyclotron in a ¹⁸O(p,n)¹⁸F reaction.

¹H- and ¹³C-NMR spectra were recorded on a Varian Mercury 300 spectrometer at 27 °C. Chemical shifts are reported in ppm (δ) relative to tetramethylsilane (TMS) and calibrated using solvent residual peaks or TMS as internal standard. Data are shown as follows: chemical shift, multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, quin = quintet, dd = doublet of doublets, dt = doublet of triplets, m = multiplet), integration, and coupling constant (J, Hz).

Bruker Daltonics micrOTOF (TOF-ESI-MS) was used for exact mass measurements.

AminoLink coupling resin was purchased from Thermo Fisher Scientific. The AminoLink column for radiosynthesis was prepared on the day of use as follows. Polyethylene frit was placed on a 2 ml syringe without a plunger. The frit was wetted with acetonitrile, and 0.5 ml of AminoLink resin (1 ml slurry) was added. The resin was washed with 6 ml of water, and capped until use.

Synthesis of aminooxy precursor 5

(Boc-aminooxy)acetic acid (0.1187 g, 620.9 μmol , 1.5 eq) was dissolved in 0.5 ml of dimethylformamide (DMF) under argon atmosphere. 1-[Bis(dimethylamino)methylene]-1*H*-1,2,3-triazolo[4,5-*b*]pyridinium 3-oxid hexafluorophosphate (HATU) (0.2418 g, 635.9 μmol , 1.6 eq) was added to the reaction vial. The mixture was let to stir for 10 minutes before adding di-*tert*-butyl ((6-amino-1-(*tert*-butoxy)-1-oxohexan-2-yl)carbamoyl)glutamate (0.1980 g, 406.0 μmol , 1 eq) with 0.8 ml of DMF. The mixture was let to stir for 20 minutes before adding *N,N*-diisopropylethylamine (DIEA) (180 μl , 1033.4 μmol , 2.5 eq). After 20 hours 15 ml of ethyl acetate was added to the mixture. Extractions were done to remove DMF: 3 x 15 ml 5% LiCl, and 20 ml brine. The organic layer was dried with MgSO_4 and evaporated to dryness. The crude product was dissolved in acetonitrile and purified with HPLC. Separation was carried out on a Grace Alltima C18 column (5 μm , 250 mm \times 10 mm) with a flow rate of 3 ml min^{-1} . The used HPLC eluents were A: water and B: acetonitrile. The gradient program was 0-3 min: 40% B, 3-15 min: 40 to 80% B, 15-25 min: 80% B. Retention time was 19.2 min. After solvent removal by lyophilization the Boc-aminooxy intermediate was isolated as white powder (160.1 mg, 242.3 μmol , 59.7%). TOF-ESI-MS $[\text{M}+\text{H}]^+$ m/z calcd. 661.4018 for $\text{C}_{31}\text{H}_{56}\text{N}_4\text{O}_{11}$, found 661.3756. ^1H NMR (300 MHz, CDCl_3) δ : 8.11 (s, 1H), 7.91 (bs, 1H), 5.32-5.37 (m, 2H), 4.34 (s, 2H), 4.24-4.40 (m, 2H) 3.20-3.43 (m, 2H), 2.23-2.40 (m, 2H), 2.01-2.12 (m, 1H), 1.72-1.90 (m, 2H), 1.49 (s, 9H), 1.46 (s, 9H), 1.45 (s, 9H), 1.44 (s, 9H), 1.22-1.68 (m, 5H). ^{13}C -NMR (300 MHz, CDCl_3) δ : 172.81, 172.57, 169.32, 157.89, 157.23, 83.07, 82.20, 81.63, 80.66, 75.96, 53.50, 53.10, 38.34, 32.31, 31.74, 28.83, 28.50, 28.28, 28.21, 28.16, 22.23.

Boc-aminooxy intermediate (128.4 mg, 194.3 μmol) was dissolved in 30 ml of 2 M HCl in Et₂O. The mixture was stirred at r.t. for 20 hours. The resulting white precipitate was isolated, dissolved in water, and purified with HPLC. Separation was carried out on a Grace Alltima C18 column (5 μm , 250 mm \times 4.6 mm) with a flow rate of 1 ml min⁻¹. The used HPLC eluents were A: 0.1% TFA in water and B: 0.1% TFA in acetonitrile. The gradient was 5 to 50% B (0-15 min). Retention time: 5.9 min. After solvent removal by lyophilization the aminoxy precursor **5** was isolated as white powder (70.6 mg, 179.9 μmol , 92.6%). TOF-ESI-MS [M+H]⁺ *m/z* calcd. 393.1616 for C₁₄H₂₅N₄O₉, found 393.1640. ¹H-NMR (300 MHz, D₂O) δ : 4.65 (s, 2H), 4.29-4.33 (dd, 1H, *J* = 5.2 Hz, *J* = 9.2 Hz), 4.22-4.26 (dd, 1H, *J* = 5.1 Hz, *J* = 8.9 Hz), 3.29 (t, 2H, *J* = 7.0), 2.54 (t, 2H, *J* = 7.1 Hz), 2.17-2.29 (m, 1H), 1.98-2.08 (m, 1H), 1.85-1.95 (m, 1H) 1.70-1.81 (m, 1H), 1.57-1.66 (m, 2H), 1.41-1.51 (m, 2H).

Synthesis of reference compound 7

Compound **5** (26.4 mg, 67.2 μmol , 1 eq) and 5-deoxy-5-fluoro-D-ribose (**3**) (102.2 mg, 671.8 μmol , 10 eq) were dissolved in 1 ml of water. The mixture was kept at r.t. (room temperature) for 1 hour. The product was purified with HPLC. Separation was carried out on a Grace Alltima C18 column (5 μm , 250 mm \times 4.6 mm) with a flow rate of 1 ml min⁻¹. The used HPLC eluents were A: 0.1% TFA in water and B: 0.1% TFA in acetonitrile. The gradient was 5 to 50% B (0-15 min). Retention time: 6.6 min (*E*) and 7.0 min (*Z*). After solvent removal by lyophilization compound **7** was isolated as white powder (28.2 mg, 53.6 μmol , yield 79.9%, purity 99% (HPLC 209 nm)). Compound **7** has an imine double bond and exists as *E/Z* isomers. The two product peaks were observed with HPLC. The peaks were collected and subjected to TOF-ESI-MS. MS analysis confirmed the same mass for both peaks. Cyclic form was not detected which was in accordance with previous studies with FDR [20, 23, 35].

TOF-ESI-MS [M+Na]⁺ *m/z* calcd. 549.1815 for C₁₉H₃₁FN₄O₁₂Na, found 549.1816. ¹H-NMR (300 MHz, D₂O) δ : 7.74 (d, 0.85H (*E*), *J* = 5.6 Hz), 7.06 (d, 0.15H (*Z*), *J* = 6.2 Hz), 5.19-5.14 (m, 0.15H (*Z*)), 4.61 (s, 0.3H (*Z*)), 4.60 (s, 1.7H (*E*)), 4.65-4.57 (m, 1.85H), 4.57-4.51 (m, 1H), 4.35-4.27 (m,

2H), 4.27-4.19 (m, 2H), 4.01-3.81 (m, 2H), 3.29 (t, 2H, J = 6.8 Hz), 2.56 (t, 2H, J = 7.3 Hz), 2.30-2.14 (m, 1H), 2.09-1.94 (m, 1H), 1.94-1.81 (m, 1H), 1.81-1.67 (m, 1H), 1.67-1.51 (m, 2H), 1.51-1.36 (m, 2H). ¹³C-NMR (300 MHz, D₂O) δ: 177.67, 177.54, 176.67, 172.31, 159.74, 154.09 (Z), 153.24 (E), 86.11, 83.92, 72.61 (E), 72.52 (Z), 72.37 (E), 70.48 (Z), 70.24 (Z), 69.73 (E), 53.58, 52.98, 39.10, 30.94, 30.47, 28.23, 26.65, 22.63.

Synthesis of reference compound 11

Compound **4** (12.3 mg, 47.3 μmol, 1 eq) and 2-deoxy-2-fluoro-D-glucose (**10**) (77.6 mg, 426.0 μmol, 9 eq) were dissolved in 1.5 ml of water. The mixture was kept at r.t. for 3 hours. The mixture was passed through the AminoLink (AL) resin (1 ml of slurry, 0.5 ml of resin, preconditioned with 6 ml of water. Additional 2 ml of water was passed through the AL resin after the application of the reaction mixture. The mixture was passed through C18 Sep-Pak cartridge (Waters, preconditioned with 5 ml of ethanol and 10 ml of water), and the cartridge was washed with 10 ml of water. The product was eluted out with 5 ml of acetonitrile and evaporated to dryness to yield the product as pink solid (14.6 mg, 34.4 μmol, yield 72.8%, purity 99% (HPLC 209 nm)). Three product peaks were observed with HPLC. The peaks were collected and subjected to TOF-ESI-MS. MS analysis confirmed identical mass for all peaks. The peaks belong to the acyclic *E*- and *Z*-isomers, and the cyclic pyranose isomer. Previous studies have also reported the formation of different isomers with hexose conjugates [14, 17, 18, 39].

TOF-ESI-MS [M+H]⁺ *m/z* calcd. 425.1579 for C₁₇H₂₂FN₆O₆, found 425.1586. ¹H-NMR (300 MHz, DMSO) δ: 10.57 (s, 1H), 8.54 (t, 0.14H (Z), J = 6.2 Hz), 8.45 (d, 2H, J = 8.3 Hz), 8.45 (dd, 0.58H (E), J = 9.0, 3.6 Hz), 7.76 (t, 0.58H (E), J = 7.1 Hz), 7.55 (m, 2H), 7.16 (dd, 0.14H (Z), J = 9.1, 6.6 Hz), 5.75 (dt, 0.14H (Z), J = 49.1, 6.4 Hz), 5.04 (dt, 0.58H (E), J = 48.2, 6.9 Hz), 4.57 (s, 2H), 4.47 (t, 2H, J = 5.3 Hz), 4.43-4.38 (m, 0.28H (cyclic)), 4.14-4.18 (m, 0.28H (cyclic)), 4.04-3.92 (m, 1H), 3.63-3.56 (m, 1H), 3.54-3.45 (m, 1H), 3.45-3.31 (m, 2H). ¹³C-NMR (300 MHz, DMSO) δ: 168.64, 165.40, 158.08, 148.93 (E), 148.64 (Z), 144.63, 130.32, 127.96, 127.73, 91.83, 89.57, 78.21 (cyclic),

72.65, 70.98, 70.19 (cyclic), 69.96 (*E* & *Z*), 69.67 (cyclic), 69.63 (*E*), 69.58 (*Z*), 63.20 (*E* & *Z*), 60.84 (cyclic), 41.53.

Synthesis of reference compound **12**

Compound **5** (16 mg, 40.8 μmol , 1 eq) and 2-deoxy-2-fluoro-D-glucose (**10**) (74.3 mg, 407.9 μmol , 10 eq) were dissolved in 1 ml of water. The mixture was kept at r.t. for 3 hours. The product was purified with HPLC. Separation was carried out on a Grace Alltima C18 column (5 μm , 250 mm \times 4.6 mm) with a flow rate of 1 ml min^{-1} . The used HPLC eluents were A: 0.1% TFA in water and B: 0.1% TFA in acetonitrile. The gradient was 5 to 50% B (0-15 min). Retention time was 6.2 min (cyclic), 6.5 min (*E*), and 6.7 min (*Z*). After solvent removal by lyophilization the end product **12** was isolated as white powder (16.5 mg, 29.6 μmol , yield 72.7%, purity 99% (HPLC 209 nm)). Three product peaks were observed with HPLC. The peaks were collected and subjected to TOF-ESI-MS. MS analysis confirmed identical mass for all peaks. The peaks belong to the acyclic *E*- and *Z*-isomers, and the cyclic pyranose isomer. Previous studies have also reported the formation of different isomers with hexose conjugates [14, 17, 18, 39].

TOF-ESI-MS $[\text{M}+\text{Na}]^+$ m/z calcd. 579.1920 for $\text{C}_{20}\text{H}_{33}\text{FN}_4\text{O}_{13}\text{Na}$, found 579.1921. ^1H -NMR (300 MHz, D_2O) δ : 7.80 (dd, 0.73H (*E*), $J = 9.1, 6.2$ Hz), 7.21 (dd, 0.15H (*Z*), $J = 10.4, 5.9$ Hz), 5.87 (dt, 0.15H (*Z*), $J = 48.0, 5.6$ Hz), 5.26 (dt, 0.73H (*E*), $J = 47.4, 6.5$ Hz), 4.66 (s, 0.3H (*Z*)), 4.65 (s, 1.46H (*E*)), 4.36-4.28 (m, 2H), 4.28-4.16 (m, 2H), 4.00-3.76 (m, 2H), 3.74-3.63 (m, 2H), 3.59-3.43 (m, 1H), 3.30 (t, 2H, $J = 6.7$ Hz), 2.56 (t, 2H, $J = 7.3$ Hz), 2.29-2.14 (m, 1H), 2.09-1.94 (m, 1H), 1.94-1.81 (m, 1H), 1.81-1.68 (m, 1H), 1.68-1.52 (m, 2H), 1.52-1.36 (m, 2H). ^{13}C -NMR (300 MHz, D_2O) δ : 177.66, 177.53, 176.66, 172.11, 159.72, 150.03 (*E*), 149.74 (*Z*), 91.96, 89.68, 77.87 (cyclic), 72.63 (*E*), 72.32 (*Z*), 71.65 (cyclic), 71.06, 70.42 (cyclic), 70.10 (*E* & *Z*), 69.72 (cyclic), 69.60 (*E*), 69.48 (*Z*), 63.20 (*E* & *Z*), 60.84 (cyclic), 53.58, 52.97, 39.09, 30.94, 30.45, 28.19, 26.63, 22.62.

^{18}F FDR synthesis (^{18}F **3**)

[¹⁸F]F⁻ was trapped on a Sep-Pak QMA Light anion-exchange cartridge (Waters, preconditioned with 10 ml of 0.5 M K₂CO₃ and 15 ml of water) and eluted as a ¹⁸F⁻/Kryptofix2.2.2/K⁺ complex to a conical reaction vial (Grace Davison Discovery Sciences, Deerfield, IL, USA). The ¹⁸F⁻/Kryptofix2.2.2/K⁺ complex was dried with azeotropic distillation at 120 °C under an Ar flow (40 ml min⁻¹) and cooled to r.t. Precursor **1** (5 mg, 14.0 μmol) in 500 μl of DNA-synthesis grade anhydrous acetonitrile (Merck) was added in to the reaction vial. The reaction mixture was heated to 108 °C for 12 min, and then cooled to r.t. DMT-Cl (24 mg, 70.8 μmol, 5 eq) in pyridine (0.5 ml) was added and the reaction was let to stand for 5 min. The mixture was transferred with gas flow to a vial with 30 ml of water, and 0.5 ml of 6 M HCl to neutralize the pyridine. The mixture was passed through two C18 Sep-Pak cartridges (preconditioned with 5 ml of ethanol and 10 ml of water). The C18 cartridges were rinsed with 15 ml of water. [¹⁸F]**2** was eluted out with 4 ml of 50% acetonitrile into a 5 ml septum-sealed reaction vial, which was loaded with 400 μl of 6 M HCl. The reaction vial was heated to 110 °C for 10 min and cooled down to r.t. For neutralization 390 μl of 6 M NaOH was added. The solution was concentrated with Ar gas flow (90 ml min⁻¹) under heating at 110 °C to 1 ml.

[¹⁸F]FDG synthesis ([¹⁸F]**10**)

[¹⁸F]F⁻ was trapped on a Sep-Pak QMA Light anion-exchange cartridge (Waters, preconditioned with 10 ml of 0.5 M K₂CO₃ and 15 ml of water) and eluted as a ¹⁸F⁻/Kryptofix2.2.2/K⁺ complex to a conical reaction vial (Grace Davison Discovery Sciences, Deerfield, IL, USA). The ¹⁸F⁻/Kryptofix2.2.2/K⁺ complex was dried with azeotropic distillation at 120 °C under an Ar flow (40 ml min⁻¹) and cooled to r.t. Precursor **8** (25 mg, 52.0 μmol) in 500 μl of DNA-synthesis grade anhydrous acetonitrile (Merck) was added in to the reaction vial. The reaction mixture was heated to 85 °C for 5 min, and then cooled to r.t. DMT-Cl (88 mg, 259.7 μmol, 5 eq) in pyridine (0.5 ml) was added and the reaction was let to stand for 5 min. The mixture was transferred with gas flow to a vial with 30 ml of water, and 0.5 ml of 6 M HCl to neutralize the pyridine. The mixture was passed through

C18 Light Sep-Pak cartridge (preconditioned with 5 ml of ethanol and 10 ml of water). The C18 Light Sep-Pak cartridge was rinsed with 15 ml of water. The removal of protective groups was done on cartridge by eluting the cartridge first with 2 x 0.5 ml of 1 M NaOH and then with 1 ml of water into a vial with 166 μ l of 6 M HCl. For final purification, the solution was passed through a C18 Light Sep-Pak cartridge (preconditioned with 5 ml of ethanol and 10 ml of water) and an Alumina N Sep-Pak cartridge (preconditioned with 50 ml water). The cartridges were rinsed with 1 ml of water.

Oxime formation between [18 F]**3** ([18 F]FDR) and precursor **4** to yield [18 F]**6**

The aminoxy precursor **4** (0.33 mg, 1.27 μ mol), 1 ml of [18 F]**3**, and anilinium acetate buffer (330 μ l, 1.2M, pH 4.6) were mixed. The reaction was kept at r.t. for 1 min. The reaction mixture was passed through AminoLink resin (1 ml slurry (0.5 ml of resin), preconditioned with 6 ml of water), and the resin was rinsed with 1 ml of acetonitrile. The mixture was diluted with 20 ml of 10 mM HCl and passed through a C18 Sep-Pak cartridge (preconditioned with 5 ml of ethanol and 10 ml of water). The cartridge was washed with 20 ml of 10 mM HCl and 2 ml of water. The product was eluted with 1.5 ml of 50% ethanol.

Oxime formation between [18 F]**3** ([18 F]FDR) and precursor **5** to yield [18 F]**7**

The aminoxy precursor **5** (0.5 mg, 1.27 μ mol), 1 ml of [18 F]**3**, and anilinium acetate buffer (330 μ l, 1.2M, pH 4.6) were mixed. The reaction was kept at r.t. for 1 min. The mixture was diluted with 10 ml of 10 mM HCl and passed through a C18 Sep-Pak cartridge (preconditioned with 5 ml of ethanol and 10 ml of water). The cartridge was washed with 20 ml of 10 mM HCl and 2 ml of water. The product was eluted with 1.5 ml of 50% ethanol. The reaction mixture was passed through AminoLink resin (1 ml slurry (0.5 ml of resin), preconditioned with 6 ml water), and the resin was rinsed with 1 ml of water.

Oxime formation between [18 F]**10** ([18 F]FDG) and precursor **4** to yield [18 F]**11**

The aminoxy tetrazine precursor **4** (0.33 mg, 1.27 μmol), 1 ml of [^{18}F]**10**, and anilinium acetate buffer (330 μl , 1.2M, pH 4.6) were mixed. The reaction mixture was kept at 60 $^{\circ}\text{C}$ for 5 min. After the oxime formation, the reaction mixture was passed through AminoLink resin (1 ml slurry (0.5 ml of resin), preconditioned with 6 ml water), and the resin was rinsed with 1 ml of acetonitrile. The mixture was diluted with 20 ml of 10 mM HCl and passed through a C18 Sep-Pak cartridge (preconditioned with 5 ml of ethanol and 10 ml of water). The cartridge was washed with 20 ml of 10 mM HCl and 2 ml of water. The product was eluted with 1.5 ml of 50% ethanol.

Oxime formation between [^{18}F]**10** ([^{18}F]FDG) and precursor **5** to yield [^{18}F]**12**

The aminoxy precursor **5** (0.5 mg, 1.27 μmol), 1 ml of [^{18}F]**10**, and anilinium acetate buffer (330 μl , 1.2M, pH 4.6) were mixed. The reaction mixture was kept at 85 $^{\circ}\text{C}$ for 30 min. After the oxime formation, the mixture was diluted with 10 ml of 10 mM HCl and passed through a C18 Sep-Pak cartridge (preconditioned with 5 ml of ethanol and 10 ml of water). The cartridge was washed with 20 ml of 10 mM HCl and 2 ml of water. The product was eluted with 1.5 ml of 50% ethanol. The reaction mixture was passed through AminoLink resin (1 ml slurry (0.5 ml of resin), preconditioned with 6 ml water), and the resin was rinsed with 1 ml of water.

Results

We used ^{18}F -fluoroglycosylation via oxime formation to radiolabel an aminoxy functionalized tetrazine derivative and a PSMA inhibitor with [^{18}F]FDR and [^{18}F]FDG (**Scheme 1**). The focus of the work was to develop HPLC-free purification methods to decrease the synthesis time and facilitate future automation of ^{18}F -fluoroglycosylation radiosynthesis. Excess unreacted carbohydrate precursor was derivatized with 4,4'-dimethoxytrityl chloride (DMT-Cl), thus increasing the lipophilicity of the unreacted precursor and enabling its separation from the fluorinated precursor with reverse-phase C18 Sep-Pak cartridge. The aminoxy precursor was trapped on an aldehyde resin

(AminoLink), and [^{18}F]FDR or [^{18}F]FDG was separated from the ^{18}F -fluoroglycoconjugate end product with C18 Sep-Pak cartridge.

[^{18}F]FDR synthesis ([^{18}F]3)

Fig. 2 represents the HPLC chromatograms after each step in the synthesis of the ribose precursor [^{18}F]FDR ([^{18}F]3). After the reaction with DMT-Cl, we did not observe the carbohydrate precursor **1** with HPLC. After the C18 cartridge purification and deprotection, the RCP (radiochemical purity) and decay corrected yield of [^{18}F]3, when starting from $^{18}\text{F}^-$, 500-2000 MBq at EOS (end of synthesis), were > 99% and $80 \pm 5\%$, respectively ($n = 12$). The total synthesis time was typically 85 min starting from $^{18}\text{F}^-$.

[^{18}F]FDG synthesis ([^{18}F]10)

Fig. 3 represents the HPLC chromatograms after each step in the synthesis of the [^{18}F]FDG ([^{18}F]10) radiolabeling precursor. After the reaction with DMT-Cl, the carbohydrate precursor **8** was not detected with HPLC. After the C18 cartridge purification and deprotection, the RCP and decay corrected yield of [^{18}F]FDG, starting from $^{18}\text{F}^-$, 500-1500 MBq at EOS, were > 99% and $65 \pm 7\%$, respectively ($n = 12$). The total synthesis time was typically 45 min starting from $^{18}\text{F}^-$.

[^{18}F]FDR-tetrazine ([^{18}F]6)

Samples were taken at 1 min, 5 min, 10 min, 20 min, and 30 min to follow the reaction between [^{18}F]3 and precursor **4** with radio-TLC. The oxime formation proceeded to near completion in one minute at r.t. and the end product remained stable over the whole observation period of 30 minutes (**Fig. 4 A**). Based on the results, reaction time of 1 min was chosen to be used in further studies. After the AminoLink and C18 cartridge purification, the RCP, decay-corrected yield (starting from $^{18}\text{F}^-$, 300-700 MBq at EOS), and molar activity for [^{18}F]6 were > 99%, $73 \pm 7\%$ and $> 800 \text{ GBq } \mu\text{mol}^{-1}$, respectively ($n = 3$). **Fig. 5** represents the HPLC chromatograms of the [^{18}F]FDR-tetrazine conjugate [^{18}F]6 before and after purification. The amount of the residual aminoxy precursor was below LOD

(limit of detection, < 0.1 nmol). The total synthesis time was typically 95 min (starting from $^{18}\text{F}^-$).

Table 1 represents comparison of these results to previously used HPLC-purification method [35].

^{18}F]FDR-PSMA inhibitor (^{18}F]**7**)

Samples were taken at 1 min, 5 min, 10 min, 20 min, and 30 min to follow the reaction between ^{18}F]**3** and **5** with radio-TLC. The oxime reaction proceeded to nearly completion in one minute at r.t. and the end product remained stable over the whole observation period of 30 minutes (**Fig. 4 B**). Reaction time of 1 min was chosen to be used in further studies. After the AminoLink and C18 cartridge purifications, the RCP, decay-corrected yield (starting from $^{18}\text{F}^-$, 250-700 MBq at EOS), and molar activity for ^{18}F]**7** were > 99%, $75 \pm 6\%$ and > 600 GBq μmol^{-1} , respectively (n = 3). The amount of the residual aminoxy precursor was below LOD (< 0.1 nmol). The total synthesis time was typically 95 min (starting from $^{18}\text{F}^-$).

^{18}F]FDG-tetrazine (^{18}F]**11**)

In order to find the optimal reaction conditions for the conjugation between ^{18}F]**10** and **4**, the reaction mixture was kept at r.t. (n = 2), or heated to 60 °C (n = 3) or 85 °C (n = 3). Samples were taken at 1 min, 5 min, 10 min, 20 min, and 30 min to follow the reaction progress with radio-TLC. The highest conjugation yields were observed 5 minutes after reaction start when heating was applied: $61.3 \pm 3.5\%$ (60 °C, n = 3), and $31.6 \pm 1.6\%$ (85 °C, n = 3) (**Fig. 4 A**). Reaction time of 5 min and temperature of 60 °C were chosen to be used in further studies. After the AminoLink and C18 cartridge purifications, the RCP, decay-corrected yield (starting from $^{18}\text{F}^-$, 200-500 MBq at EOS), and molar activity for ^{18}F]**11** were > 99%, $35 \pm 4\%$ and > 800 GBq μmol^{-1} , respectively (n = 4). The amount of the residual aminoxy precursor was below LOD (< 0.1 nmol). The total synthesis time was typically 60 min (starting from $^{18}\text{F}^-$, oxime formation: 5 min at 60 °C).

^{18}F]FDG-PSMA inhibitor, (^{18}F]**12**)

In order to determine the optimal reaction conditions for the conjugation between [^{18}F]**10** and **5**, the reaction mixture was kept at r.t. ($n = 2$), or heated to 60 °C ($n = 3$) or 85 °C ($n = 3$). Samples were taken at 1 min, 5 min, 10 min, 20 min, and 30 min to follow the reaction progress with radio-TLC. The highest conjugation yields were observed 30 minutes after the start of the reaction. The reaction was facilitated by increased temperature: $12.9 \pm 1.2\%$ (r.t., $n = 2$), $36.3 \pm 0.5\%$ (60 °C, $n = 3$), and $90.4 \pm 1.6\%$ (85 °C, $n = 3$) (**Fig. 4 B**). Reaction time of 30 min and temperature of 85 °C were chosen to be used in further studies. After the AminoLink and C18 cartridge purifications, the RCP, decay-corrected yield (starting from $^{18}\text{F}^-$, 300-650 MBq at EOS), and molar activity for [^{18}F]**12** were $> 99\%$, $54 \pm 6\%$ and $> 600 \text{ GBq } \mu\text{mol}^{-1}$, respectively ($n = 4$). **Fig. 5** represents the HPLC chromatograms of [^{18}F]**12** before and after purification. The amount of the residual aminoxy precursor was below LOD ($< 0.1 \text{ nmol}$). The total synthesis time was typically 85 min (starting from $^{18}\text{F}^-$, oxime formation: 30 min at 85 °C).

Residuals of aniline and pyridine

Extra precautions should be taken to ensure that there are not residues of either aniline and pyridine present in the end products, since both compounds are toxic and carcinogenic. Before the last C18 Sep-Pak cartridge purification, the reaction mixture was diluted with 10 mM HCl (pH 2) and the cartridge was washed with excess of 10 mM HCl. In these conditions both compounds, aniline (pK_a of conjugate acid: 4.6) and pyridine (pK_a of conjugate acid: 5.25), are ionized and not retained on C18 cartridge. We did not detect any residual peaks of aniline and pyridine when the non-diluted end products (100 μl) were injected to the HPLC. The limit of detection (LOD) with our HPLC method was 54 pmol and 253 pmol for aniline and pyridine, respectively. Based on these LODs, the end products would contain $< 0.8 \text{ nmol}$ and $< 3.8 \text{ nmol}$ of aniline and pyridine, respectively.

Discussion

A typical synthesis procedure for ^{18}F -fluoroglycosylation with $[^{18}\text{F}]\text{FDR}$ and $[^{18}\text{F}]\text{FDG}$ involves two HPLC purifications. In this study, we have developed an efficient cartridge purification method to replace both of the HPLC purification steps. If the carbohydrate precursor is not removed after the fluorination step during the radiosynthesis of $[^{18}\text{F}]\text{FDR}$ and $[^{18}\text{F}]\text{FDG}$, it will get deprotected and hydrolyzed in the next steps, leading to an end product that contains ribose (FDR synthesis) or glucose (FDG synthesis) that will compete in the subsequent oxime formation reaction. Therefore, the $[^{18}\text{F}]\text{FDR}$ and $[^{18}\text{F}]\text{FDG}$ are usually purified with HPLC before the oxime formation, or the end product is separated from glucose/ribose conjugate with HPLC after the oxime formation. Here, we present a way to circumvent the HPLC purification by adding one step, the derivatization of the carbohydrate precursor with DMT-Cl, to the radiosynthesis of $[^{18}\text{F}]\text{FDR}$ and $[^{18}\text{F}]\text{FDG}$. Conjugation of the DMT moiety increased the lipophilicity of the carbohydrate precursor and enables its separation from the fluorinated carbohydrate precursor with C18 Sep-Pak cartridge before the deprotection step. The addition of the DMT-Cl to the reaction vial after the fluorination reaction proved to be very efficient, since no carbohydrate precursor could be detected with HPLC. A facile C18 Sep-Pak SPE purification retained the DMT-carbohydrate precursor complex entirely and the following oxime formation reaction proceeded in high yields with a low concentration of aminoxy precursor (~ 1 mM).

To replace the second HPLC purification for the removal of the unreacted aminoxy precursor, we used an aldehyde resin (AminoLink) which quantitatively trapped the precursor. In addition to $[^{18}\text{F}]\text{FDR}$ and $[^{18}\text{F}]\text{FDG}$ other ^{18}F -labeled aldehydes, such as 4- $[^{18}\text{F}]\text{fluorobenzaldehyde}$, have been used to radiolabel aminoxy-functionalized molecules [40-46]. The AminoLink purification method developed in the present work could be a valuable tool for these applications as well. The developed method enables the HPLC-free separation of the oxime end product from the aminoxy precursor, and the production of high apparent molar activity oxime end products in cases where it has been

impossible to separate the aminoxy precursor from the oxime end product with liquid chromatography due to the very similar retention properties of the two compounds.

Earlier, we have reported the synthesis of the [^{18}F]FDR-conjugated tetrazine [^{18}F]6 using a method requiring two HPLC purifications [35, 47, 48]. With the HPLC purifications, the radiochemical purity was > 99% and the molar activity was up to 809 GBq μmol^{-1} . With the HPLC-free SPE and resin method reported here, we achieved the same radiochemical purity and molar activity levels (> 800 GBq μmol^{-1}). The total radiosynthesis time, however, was decreased by 25 minutes (95 min versus 120 min), and the simplified purification procedures are extremely useful towards the automation of ^{18}F -fluoroglycosylation synthesis via oxime formation.

In the present study, both tetrazine and PSMA inhibitor aminoxy precursors conjugated rapidly with [^{18}F]FDR at room temperature with aniline as a catalyst. With [^{18}F]FDG heating of the reaction was required as expected. [^{18}F]FDG has been used in the ^{18}F -fluoroglycosylation of a variety of molecules, always with heating [1-3, 7, 9, 14-16, 18]. The [^{18}F]FDG conjugation also required longer reaction times than the [^{18}F]FDR conjugation: for the PSMA inhibitor, for example, 30 min with [^{18}F]FDG was required compared to the 1 min with [^{18}F]FDR.

Our results corroborate that the milder conditions of the [^{18}F]FDR fluoroglycosylation are more suitable for heat-sensitive molecules, such as tetrazines. The aminoxy tetrazine **4** could be conjugated with [^{18}F]FDR at radiochemical yields exceeding 95%, but with [^{18}F]FDG the highest yields achieved were around 60%. The yields of the [^{18}F]FDG-tetrazine ([^{18}F]11) conjugation reaction started dropping after the first 10 minutes when heating was applied, and a temperature of 85 °C resulted in lower yields than 60 °C. This might be due to the degradation of the tetrazine moiety upon heating. Tetrazines are known to decompose at high temperatures and at basic conditions. For this reason their direct fluorination usually fails or has poor yields. The tetrazine structure could be stabilized by using electron donating groups as substituents [49]. For example Rashidian et al. have successfully radiolabeled a methyl tetrazine derivative with [^{18}F]FDG with high yield, when using *p*-

phenylenediamine as catalyst [15]. The reaction at 70-75 °C for 5-10 min provided the [¹⁸F]FDG conjugate with > 90% RCY. Concentration of the aminoxy precursor has also a significant role in the yield of the oxime formation. The radiolabeling of the methyl tetrazine in the work of Rashidian et al. was carried out in ~ 0.2-0.3 M range of aminoxy precursor, whereas our reaction was in ~ 1 mM of the aminoxy precursor.

The oxime end products **6**, **7**, **11**, and **12** have an imine double bond and occur in *E* and *Z* isomers in solution. The *E/Z* isomeric ratio depends on the size of the substitutes on the double bond. With the FDR-conjugated compounds no cyclic form was detected. In contrast, with the FDG conjugated compounds the formation of the cyclic isomer was observed with both HPLC and NMR. These findings are in accordance with previous studies with FDR and FDG glycoconjugates [14, 17, 18, 20, 23, 35, 39]. Because the different isomers equilibrate quickly in solution, they are not usually isolated from each other and are considered as one compound. Reduction of the imine double bond would lead to single isomer. However, the reduction might take too long with respect to the physical half-life of fluorine-18 (109.8 min), and at least in the case of FDG, the yield would not be quantitative due to the presence of the cyclic form, the removal of which would require an additional purification step [39].

The hydroxyl groups in both [¹⁸F]FDR and [¹⁸F]FDG lower the lipophilicity of the oxime end product. This is often desired since low lipophilicity usually favors rapid urinary elimination of the radiolabeled tracer. Glycosylation of small radiolabeled peptides has in fact been shown to reduce their hepatobiliary excretion [29-31, 50]. The aminoxy functional group can be attached to almost any molecule and the radiolabeling can be carried out under mild reaction conditions using a carbohydrate prosthetic group. Consequently ¹⁸F-fluoroglycosylation is a handy tool for fluorine-18 radiolabeling of various molecules that cannot tolerate the harsh reaction conditions of direct nucleophilic fluorination. The versatility, mild conditions, and chemoselectivity of glycoconjugation are the most appealing features prompting the use of carbohydrate prosthetic groups for radiolabeling.

With the present work, the feasibility of applying this highly versatile radiosynthetic strategy for the automated production of ^{18}F -radiolabeled compounds is greatly advanced driving clinical translation of ^{18}F -fluoroglycosylated tracers.

Conclusions

Herein, we report efficient cartridge purification method for ^{18}F -fluoroglycosylation via oxime formation. The carbohydrate precursors of [^{18}F]FDR and [^{18}F]FDG were efficiently removed before oxime formation, and the aminoxy precursors of the two model compounds were quantitatively trapped with an aldehyde resin. The oxime end products were obtained with high yield, radiochemical purity, and apparent molar activity. The developed purification method enables simpler and faster automated radiosynthesis of ^{18}F -glycoconjugates for clinical translation.

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Table 1. Comparison between HPLC-purified and cartridge purified [¹⁸F]**6**.

	HPLC-purified [35]	Cartridge purified
Yield ^a (%)	50.5 ± 1.7	73 ± 7
Synthesis time (min)	120	95
Molar activity (GBq/μmol)	800	800
Radiochemical purity (%)	> 99	> 99

^aDecay-corrected to the start of synthesis

Scheme 1. Oxime formation using [^{18}F]FDR ([^{18}F]3) and [^{18}F]FDG ([^{18}F]10) as prosthetic groups to radiolabel two aminoxy functionalized compounds, tetrazine derivative (4) and PSMA (prostate specific membrane antigen) inhibitor (5). Yield refers to decay-corrected yield. RCP = radiochemical purity.

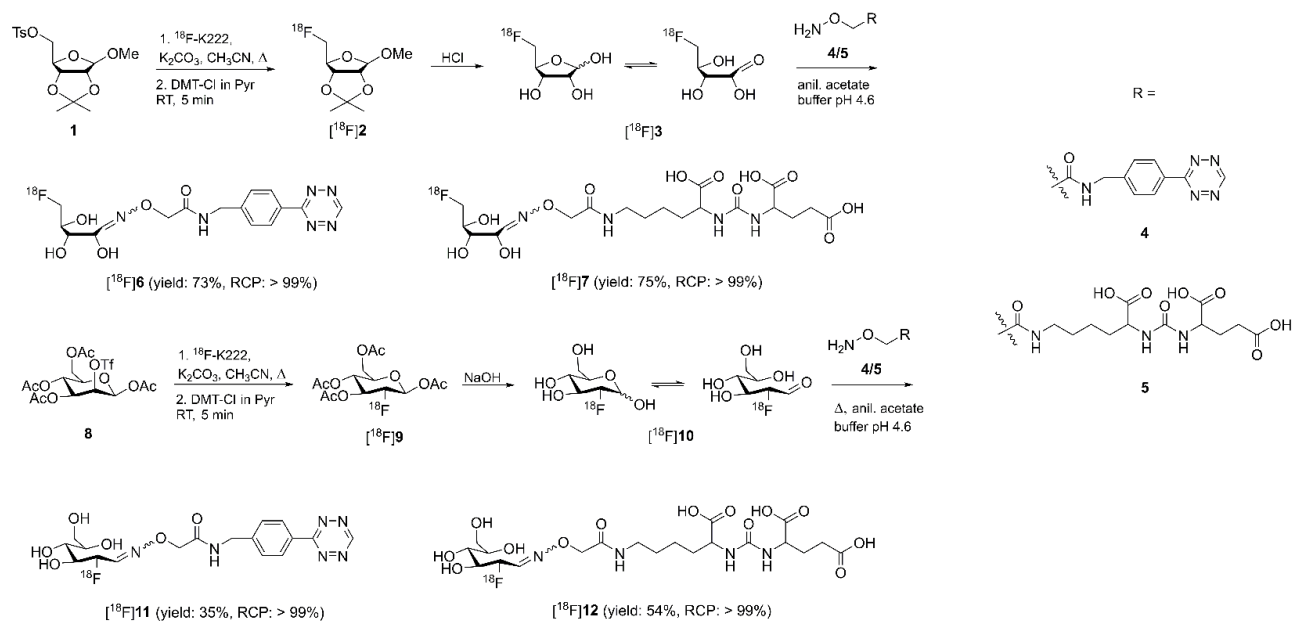
Figure 1. Schematic presentation of the synthetic and purification steps during [^{18}F]FDR and [^{18}F]FDG radiosynthesis and ^{18}F -fluoroglycosylation via oxime formation.

Fig. 2. HPLC chromatograms of the reaction mixture and the intermediates during synthesis of [^{18}F]FDR ([^{18}F]3) after A) fluorination, B) reaction with DMT-Cl, C) C18 purification, and D) deprotection. Top row: ultraviolet (UV) absorbance at 254 nm. Bottom row: radioactivity detector.

Fig. 3. HPLC chromatograms of the reaction mixture and the intermediates during synthesis of [^{18}F]FDG ([^{18}F]10) after A) fluorination, B) reaction with DMT-Cl, C) C18 purification and deprotection. Top row: ultraviolet (UV) absorbance at 254 nm. Bottom row: radioactivity detector.

Fig. 4. Oxime formation between [^{18}F]FDR ([^{18}F]3) or [^{18}F]FDG ([^{18}F]10) with aminoxy functionalized tetrazine (A) and aminoxy functionalized PSMA inhibitor (B). The yields (mean \pm s.d.) refer to the amount of [^{18}F]3 or [^{18}F]10 converted into the desired ^{18}F -labeled conjugate evaluated with radio-TLC. Dark gray square: conjugation with [^{18}F]3 at room temperature (r.t.), black circle: conjugation with [^{18}F]10 at r.t., medium gray triangle up: conjugation with [^{18}F]10 at 60 °C, and light gray triangle down: conjugation with [^{18}F]10 at 85 °C.

Fig. 5. HPLC chromatograms before and after purification with the AminoLink resin and C18 Sep-Pak cartridge for [^{18}F]FDR conjugated tetrazine ([^{18}F]6) and [^{18}F]FDG conjugated PSMA inhibitor ([^{18}F]12). Similar results were obtained with the corresponding [^{18}F]FDR conjugate [^{18}F]7 (PSMA inhibitor) and [^{18}F]FDG conjugate [^{18}F]11 (tetrazine) (see Supplemental Information **Fig. S9**). Top row: ultraviolet (UV) absorbance at 209 nm. Bottom row: radioactivity detector.



Scheme 1. Oxime formation using [¹⁸F]FDR ([¹⁸F]3) and [¹⁸F]FDG ([¹⁸F]10) as prosthetic groups to radiolabel two aminoxy functionalized compounds, tetrazine derivative (**4**) and PSMA (prostate specific membrane antigen) inhibitor (**5**). Yield refers to decay-corrected yield. RCP = radiochemical purity.

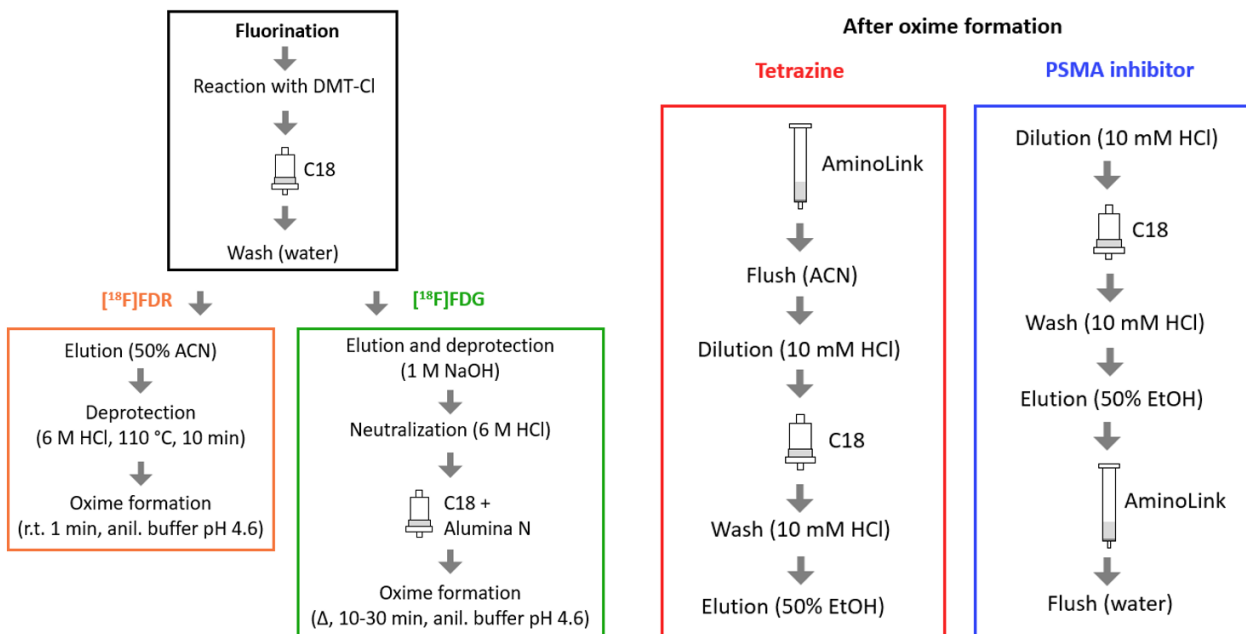


Figure 1. Schematic presentation of the synthetic and purification steps during $[^{18}\text{F}]$ FDR and $[^{18}\text{F}]$ FDG radiosynthesis and ^{18}F -fluoroglycosylation via oxime formation.

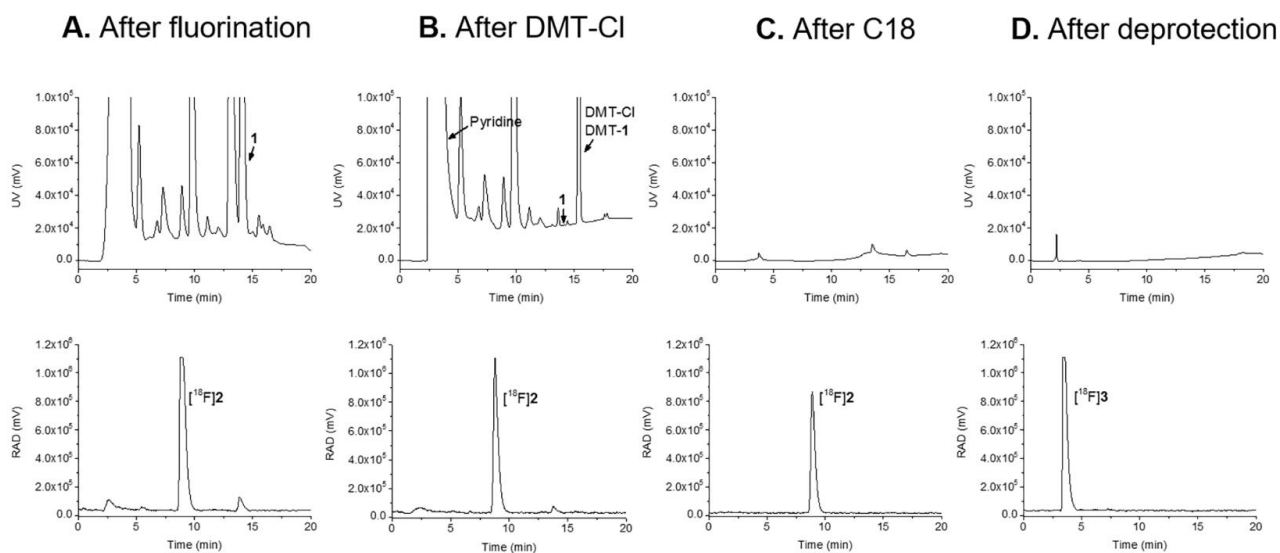


Fig. 2. HPLC chromatograms of the reaction mixture and the intermediates during synthesis of $[^{18}\text{F}]\text{FDR}$ ($[^{18}\text{F}]\mathbf{3}$) after A) fluorination, B) reaction with DMT-Cl, C) C18 purification, and D) deprotection. Top row: ultraviolet (UV) absorbance at 254 nm. Bottom row: radioactivity detector.

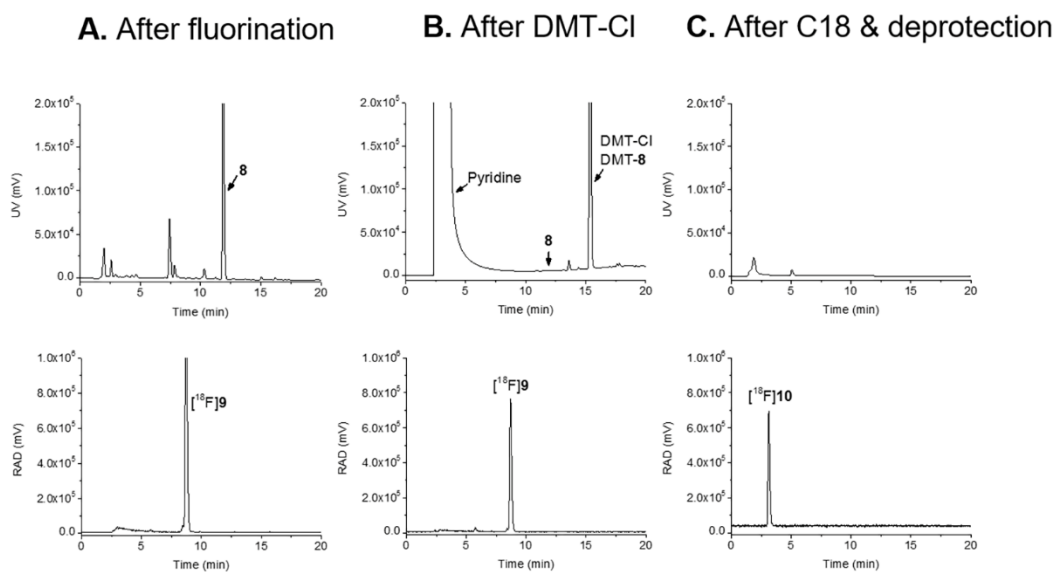


Fig. 3. HPLC chromatograms of the reaction mixture and the intermediates during synthesis of $[^{18}\text{F}]$ FDG ($[^{18}\text{F}]10$) after A) fluorination, B) reaction with DMT-Cl, C) C18 purification and deprotection. Top row: ultraviolet (UV) absorbance at 254 nm. Bottom row: radioactivity detector.

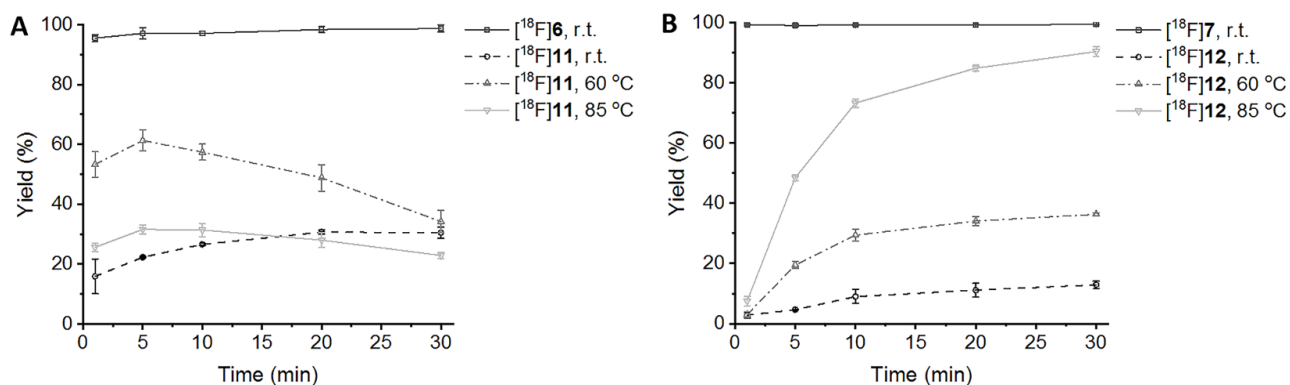


Fig. 4. Oxime formation between ^{18}F FDR (^{18}F 3) or ^{18}F FDG (^{18}F 10) with aminoxy functionalized tetrazine (A) and aminoxy functionalized PSMA inhibitor (B). The yields (mean \pm s.d.) refer to the amount of ^{18}F 3 or ^{18}F 10 converted into the desired ^{18}F -labeled conjugate evaluated with radio-TLC. Dark gray square: conjugation with ^{18}F 3 at room temperature (r.t.), black circle: conjugation with ^{18}F 10 at r.t., medium gray triangle up: conjugation with ^{18}F 10 at 60 °C, and light gray triangle down: conjugation with ^{18}F 10 at 85 °C.

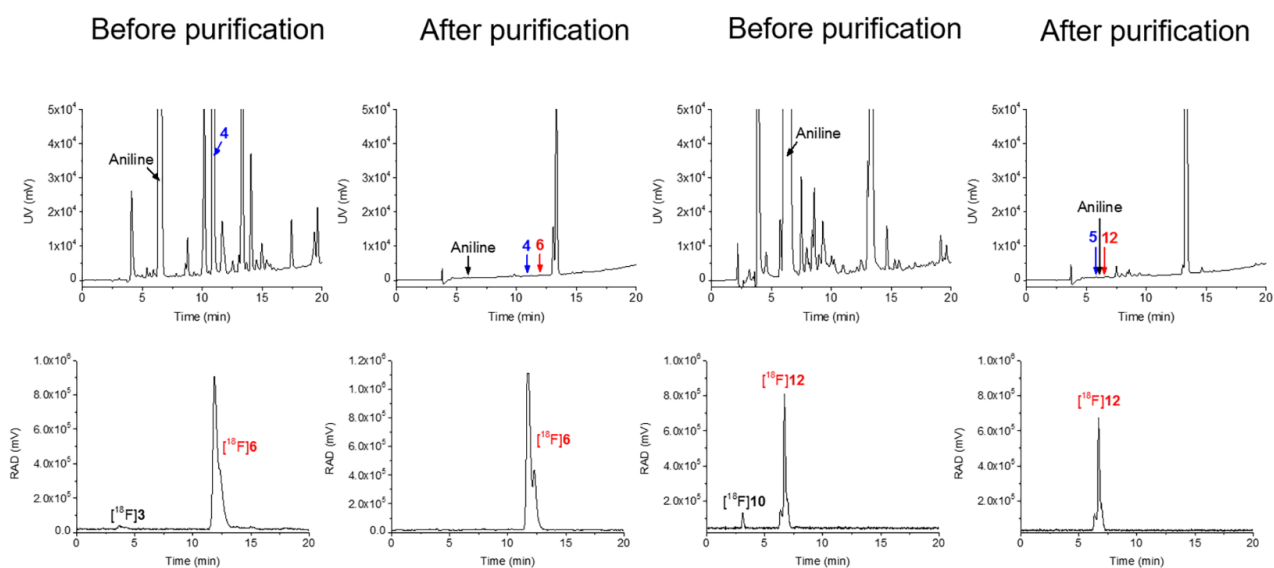


Fig. 5. HPLC chromatograms before and after purification with the AminoLink resin and C18 Sep-Pak cartridge for [¹⁸F]FDR conjugated tetrazine ([¹⁸F]6) and [¹⁸F]FDG conjugated PSMA inhibitor ([¹⁸F]12). Similar results were obtained with the corresponding [¹⁸F]FDR conjugate [¹⁸F]7 (PSMA inhibitor) and [¹⁸F]FDG conjugate [¹⁸F]11 (tetrazine) (see Supplemental Information **Fig. S9**). Top row: ultraviolet (UV) absorbance at 209 nm. Bottom row: radioactivity detector.

Supplemental information:

Efficient cartridge purification for producing high molar activity ^{18}F -glycoconjugates
via oxime formation

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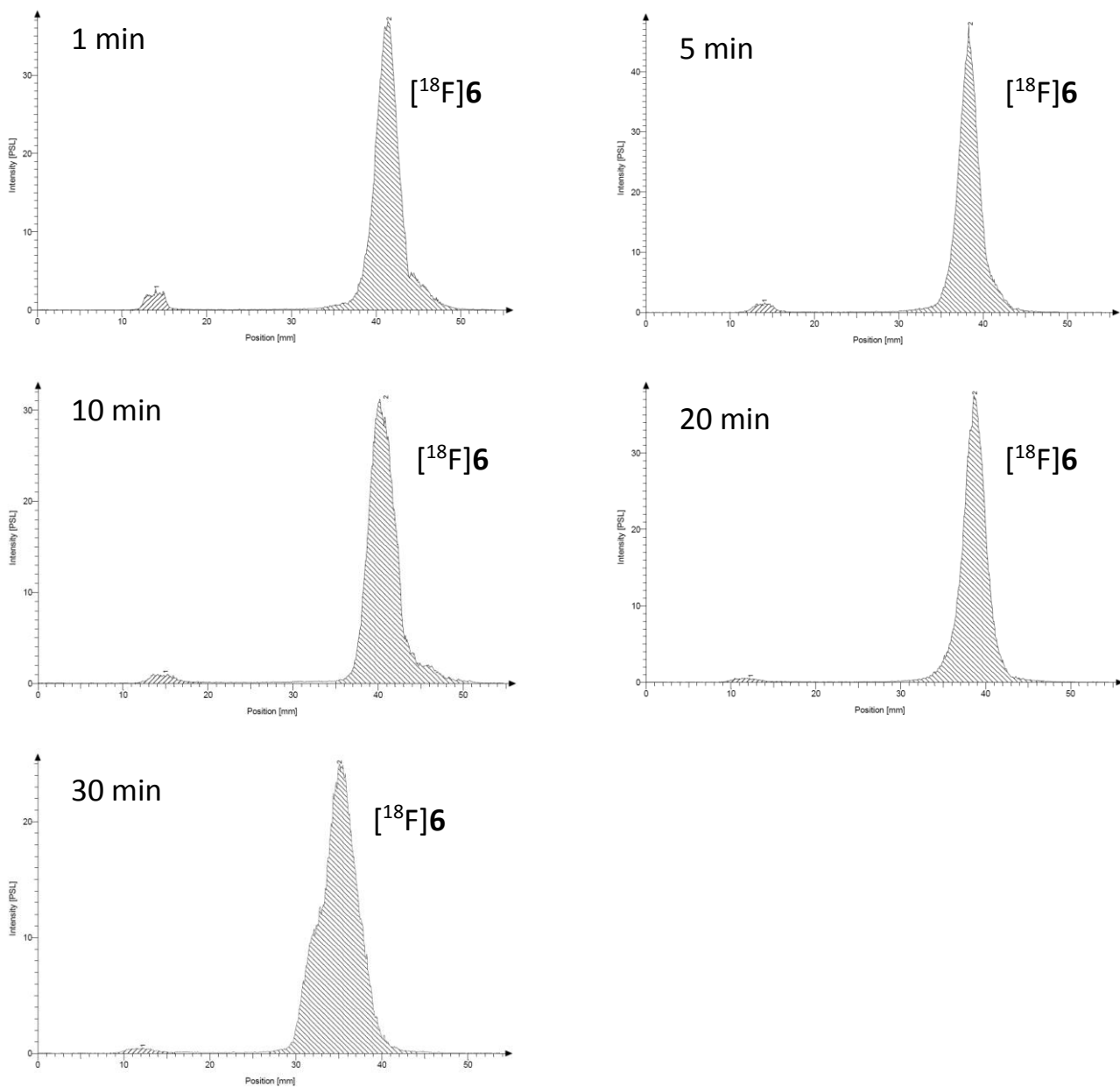


Fig. S1. Radio-TLC chromatograms of $[^{18}\text{F}]\mathbf{6}$ at different time points after reaction start. Reaction temperature: r.t.

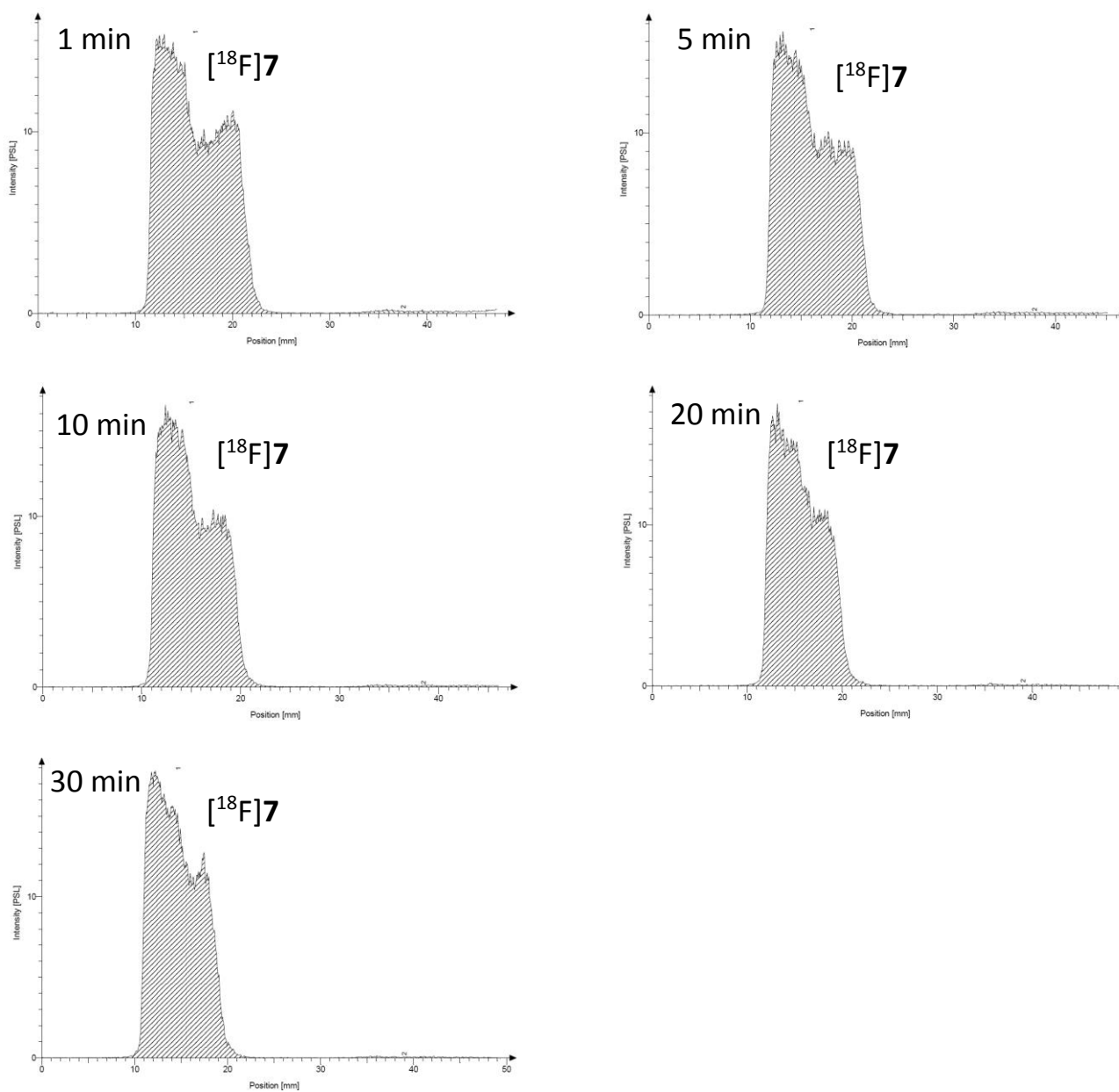


Fig. S2. Radio-TLC chromatograms of $[^{18}\text{F}]\mathbf{7}$ at different time points after reaction start. Reaction temperature: r.t.

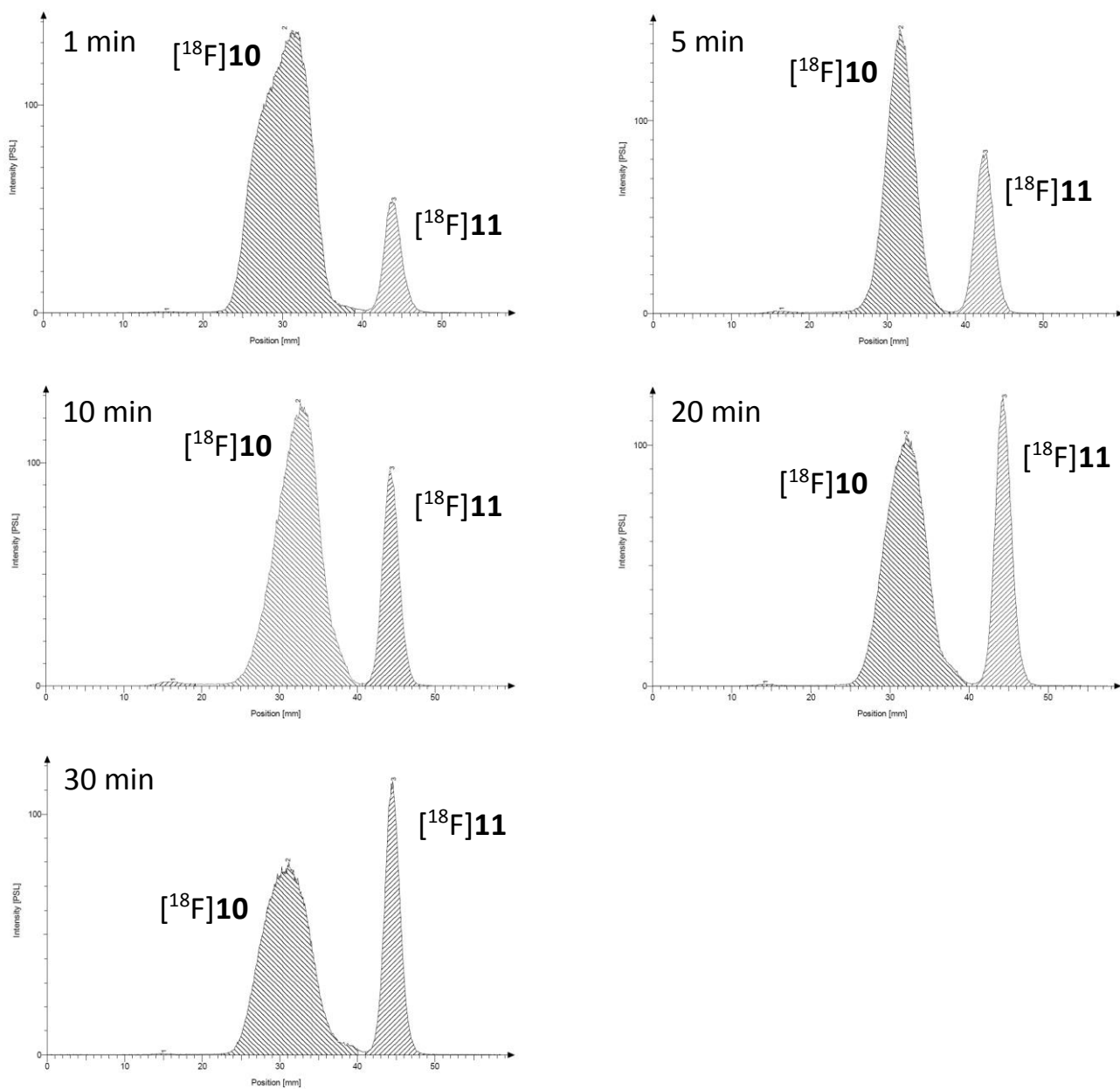


Fig. S3. Radio-TLC chromatograms of $[^{18}\text{F}]11$ at different time points after reaction start. Reaction temperature: r.t.

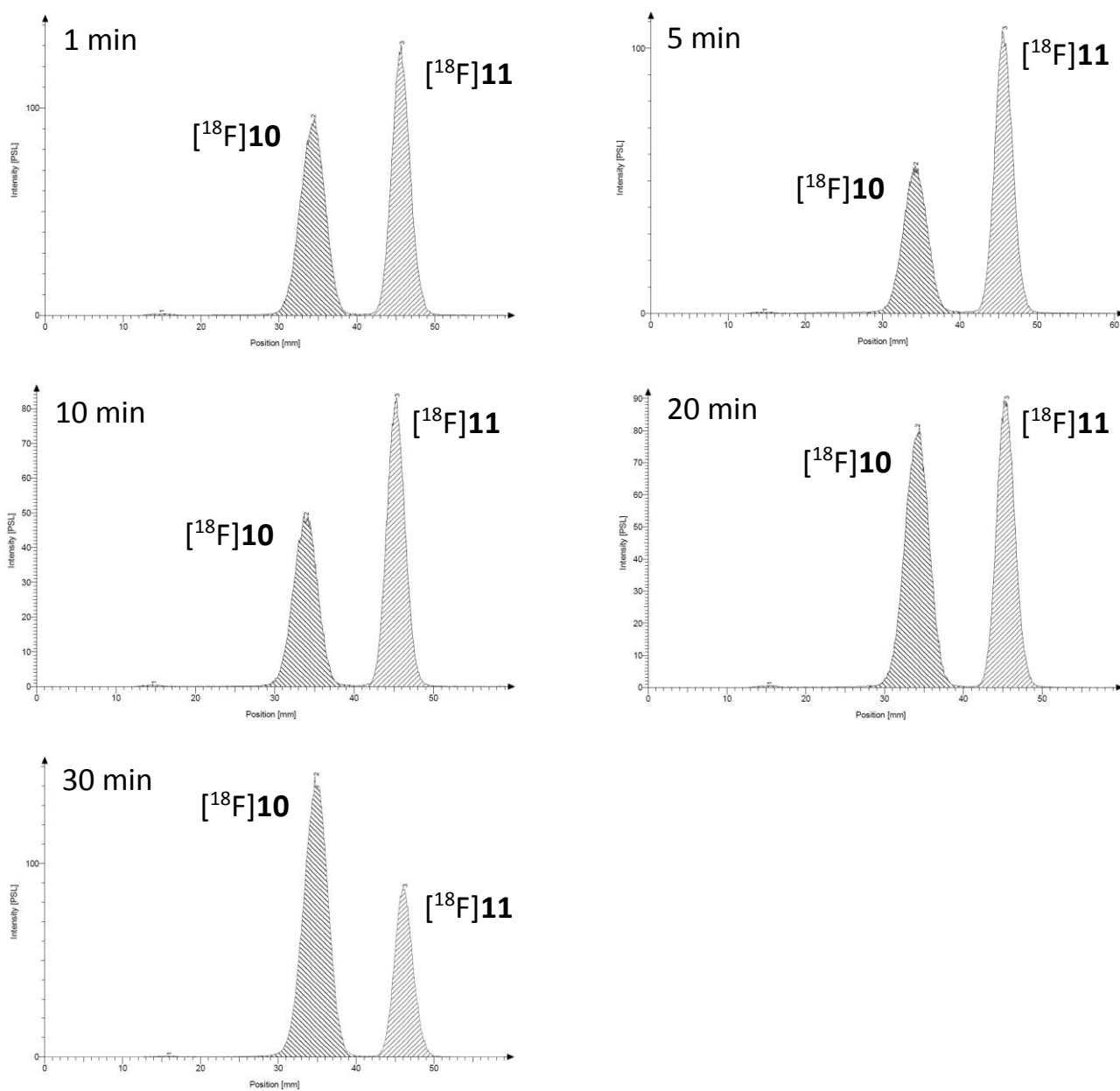


Fig. S4. Radio-TLC chromatograms of $[^{18}\text{F}]11$ at different time points after reaction start. Reaction temperature: 60 °C.

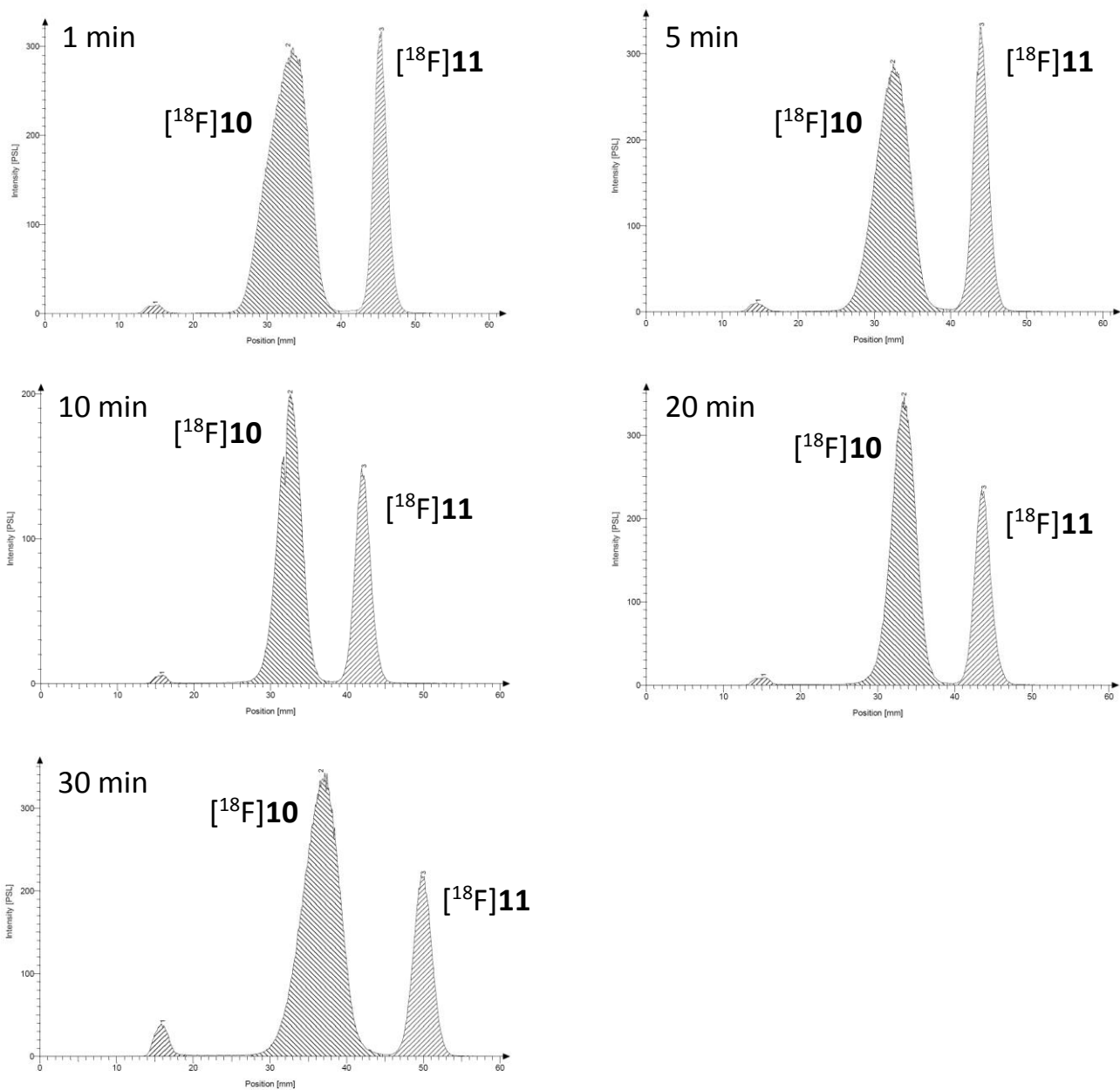


Fig. S5. Radio-TLC chromatograms of $[^{18}\text{F}]11$ at different time points after reaction start. Reaction temperature: 85 °C.

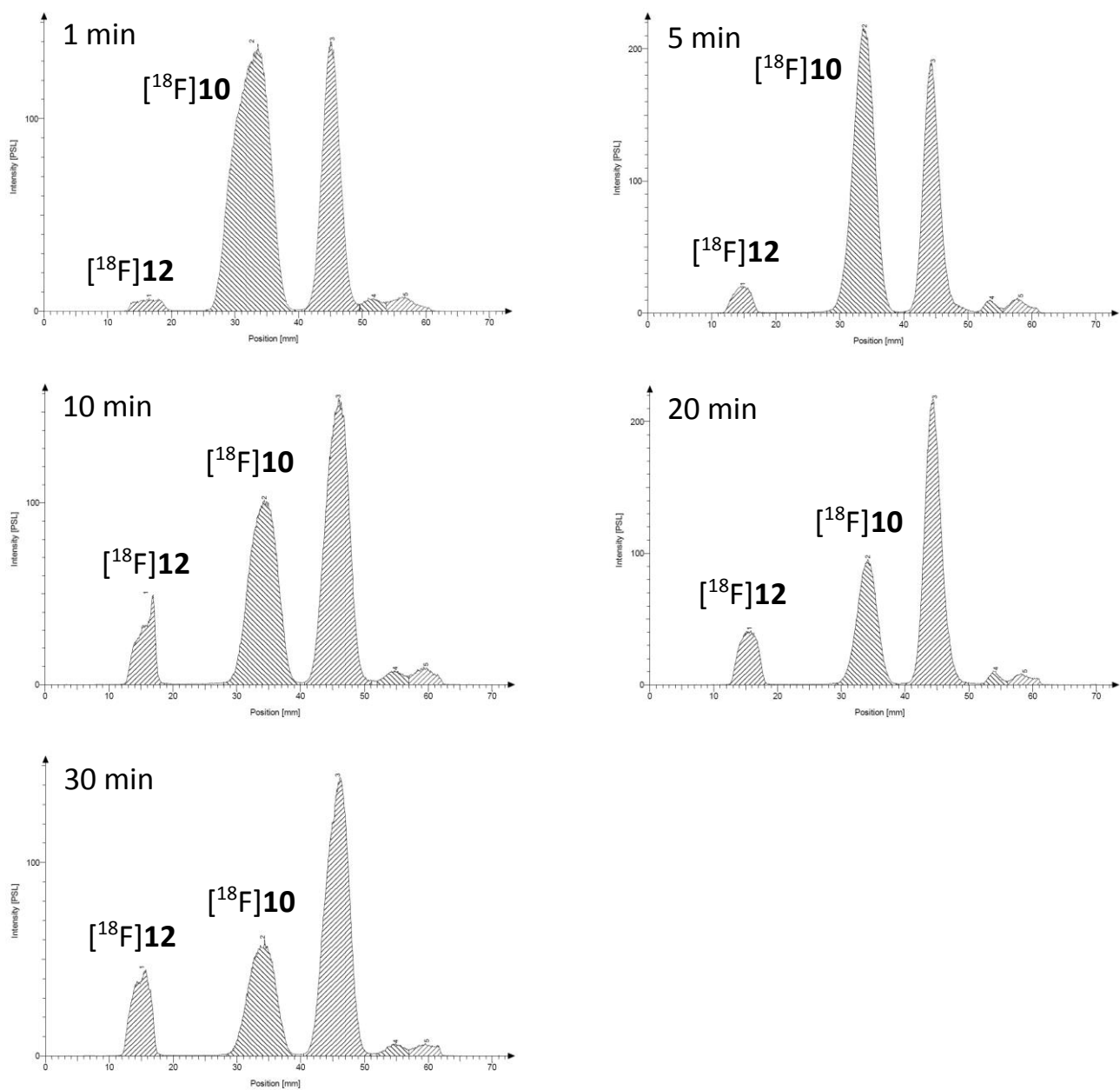


Fig. S6. Radio-TLC chromatograms of $[^{18}\text{F}]\mathbf{12}$ at different time points after reaction start. Reaction temperature: r.t.

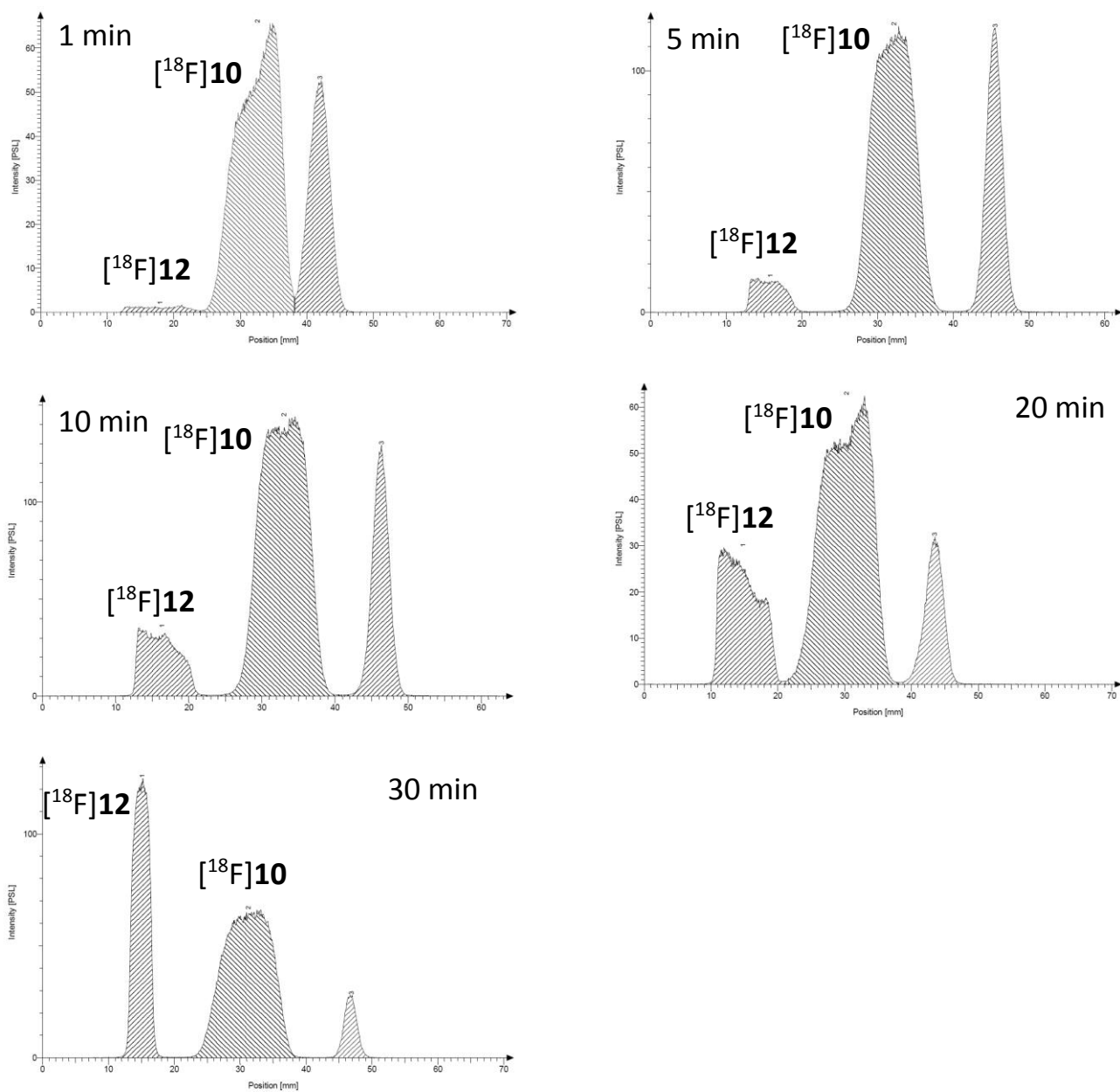


Fig. S7. Radio-TLC chromatograms of $[^{18}\text{F}]12$ at different time points after reaction start. Reaction temperature: 60 °C.

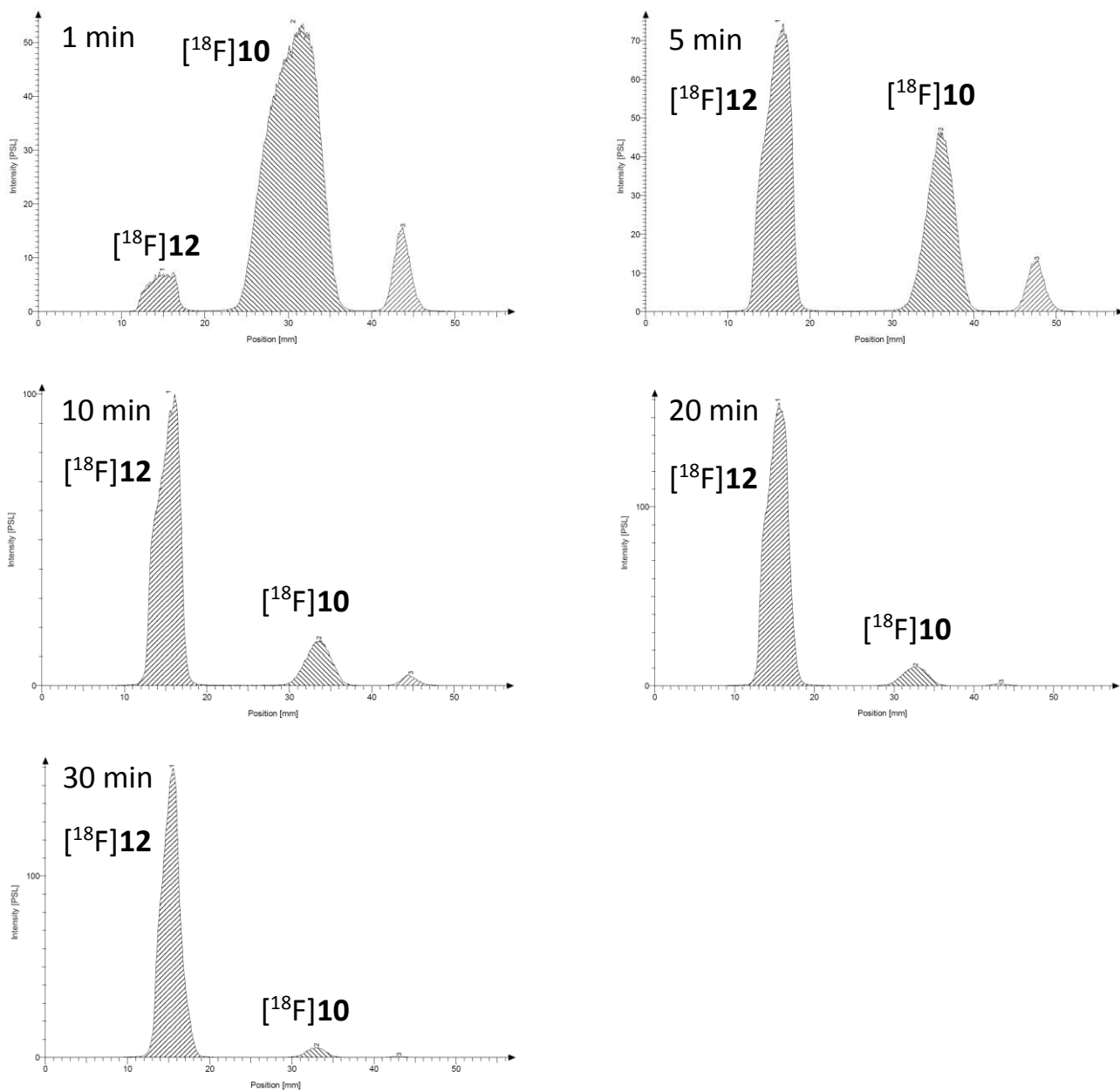


Fig. S8. Radio-TLC chromatograms of $[^{18}\text{F}]\mathbf{12}$ at different time points after reaction start. Reaction temperature: 85 °C.

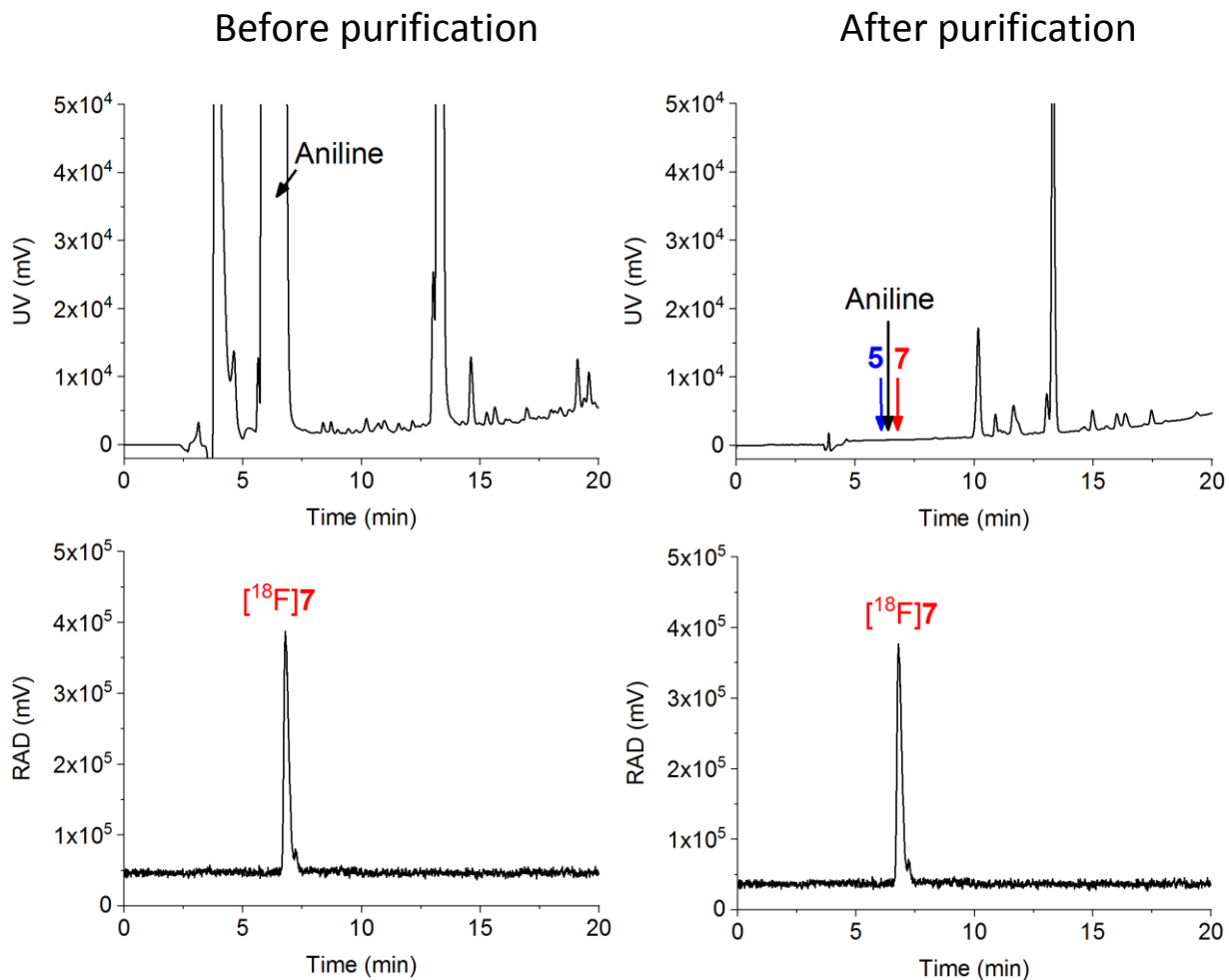


Fig. S9. HPLC chromatograms before and after purification with the AminoLink resin and C18 Sep-Pak cartridge for [¹⁸F]FDR conjugated PSMA inhibitor ([¹⁸F]7). Top row: ultraviolet (UV) absorbance at 209 nm. Bottom row: radioactivity detector.

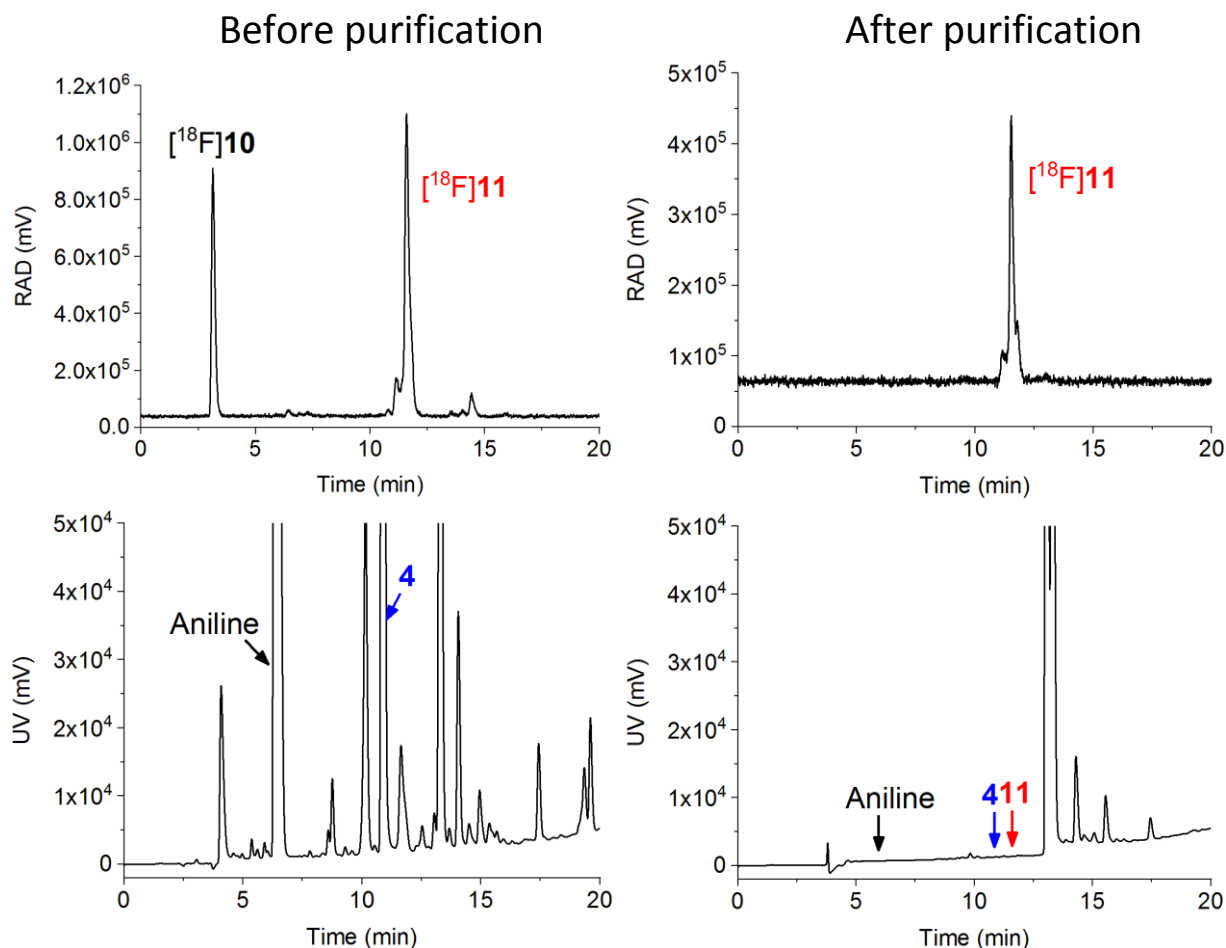


Fig. S10. HPLC chromatograms before and after purification with the AminoLink resin and C18 Sep-Pak cartridge for [^{18}F]FDG conjugated tetrazine ([^{18}F]11). Top row: ultraviolet (UV) absorbance at 209 nm. Bottom row: radioactivity detector.

Table S1. Reactivity of HPLC- and cartridge purified [^{18}F]6 with *trans*-cyclooctene-modified nanoparticles (TCO-NPs). TCO-NPs (0.005-0.2 mg/ml) were incubated with HPLC-purified and cartridge-purified [^{18}F]6 in PBS pH 7.41 (total volume in each experiment was 1 ml). The reaction was let to proceed for 5 minutes at room temperature after which the samples were centrifuged (10 000 g, 5 min) and radioactivity of the supernatant (unreacted [^{18}F]6) and the pellet (reacted [^{18}F]6 with the NPs) were measured in order to determine the yield of the IEDDA reaction.

Amount of TCO-NPs (mg)	IEDDA yield with cartridge-purified [^{18}F]6 (%)	IEDDA yield with HPLC-purified [^{18}F]6 (%) [1]
0.005	25.56 ± 2.27	27.42 ± 2.09
0.010	44.07 ± 2.03	42.62 ± 2.60
0.020	53.34 ± 0.48	56.51 ± 2.75
0.100	91.01 ± 2.05	90.24 ± 0.03
0.200	92.81 ± 0.92	94.95 ± 1.68

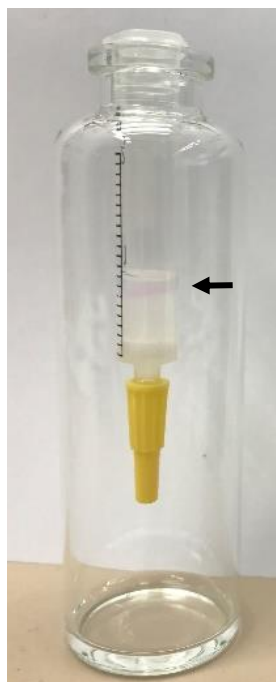


Fig. S12. AminoLink resin after it was used to purify [^{18}F]**6**. Tetrazines have distinct pink color, and a thin light pink line (arrow) was observed on the AminoLink resin after purification. Based on HPLC, the aminoxy functionalized precursor **4** was quantitatively trapped on the AminoLink resin.

[1] Keinänen O, Mäkilä EM, Lindgren R, Virtanen H, Liljenbäck H, Oikonen V, et al. Pretargeted PET Imaging of trans-Cyclooctene-Modified Porous Silicon Nanoparticles. *ACS Omega* 2017;2:62-9.