# Silicon oxide-niobium oxide mixture films and nanolaminates grown by atomic layer deposition from niobium pentaethoxide and hexakis(ethylamino) disilane

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# Abstract

Amorphous SiO<sub>2</sub>-Nb<sub>2</sub>O<sub>5</sub> nanolaminates and mixture films were grown by atomic layer deposition. The films were grown at 300 °C from Nb(OC<sub>2</sub>H<sub>5</sub>)<sub>5</sub>, Si<sub>2</sub>(NHC<sub>2</sub>H<sub>5</sub>)<sub>6</sub>, and O<sub>3</sub> to thicknesses ranging from 13 to 130 nm. The niobium to silicon atomic ratio was varied in the range of 0.11-7.20. After optimizing the composition, resistive switching properties could be observed in the form of characteristic current-voltage behavior. Switching parameters in the conventional regime were well defined only in a SiO<sub>2</sub>:Nb<sub>2</sub>O<sub>5</sub> mixture at certain, optimized, composition with Nb:Si atomic ratio of 0.13, whereas low-reading voltage measurements allowed recording memory effects in a wider composition range.

**Keywords**: atomic layer deposition, niobium oxide, silicon oxide, nanolaminates, multilayers, resistive switching, memory

## Introduction

Artificially mixed or layered nanomaterials, particularly those consisting of different oxides, have been of continuously growing interest due to the possibility to tailor functional properties of the constituent compounds, thus extending their application areas. Regarding one of the most basic materials, silicon oxide, one can note that thin solid films containing SiO<sub>2</sub> have been grown and investigated for several purposes, including porous membranes [1], catalysts [2,3], and optical filters [4,5]. SiO<sub>2</sub> films, mixtures of SiO<sub>2</sub> with metal oxides, and other SiO<sub>2</sub>-based nanostructures have been synthesized by different methods such as

galvanostatic etching of SiO<sub>2</sub>[6], electron beam evaporation [4,7, 8], sputtering [5,9-17], solgel technique [2], chemical vapor deposition [17], and atomic layer deposition (ALD). In the latter case, growth of silicon oxide could effectively be realised using, e.g, tris(dimethylamino)silane [1] or hexakis(ethylamino)disilane [18-19] as silicon precursors and ozone.

SiO<sub>2</sub>-Nb<sub>2</sub>O<sub>5</sub> multilayers have been investigated as optical coatings [8-11]. Multilayering Nb<sub>2</sub>O<sub>5</sub> with SiO<sub>2</sub> films has allowed one to reduce internal mechanical stress in the optical coating [10] and, of course, tailor the refractive index and reflecting properties of the coatings [8-9]. Nb<sub>2</sub>O<sub>5</sub> itself as a dielectric high permittivity metal oxide has been considered for the application in dynamic random access memories [20,21], also when tailored with Ta<sub>2</sub>O<sub>5</sub> or  $Al_2O_3$  [22-24].

In certain aspects, nanolaminates as switching media might demonstrate improved performance due to their stacked or periodical structure which can, at first, limit excessive grain growth. Secondly, the internal interfaces in multilayers can confine rupture points of conduction paths or filaments and, thirdly, internal barriers built in the laminates can suppress oxygen ion transport.

A polycrystalline material layer with random distribution of crystallite boundaries may give rise to nonuniformity or instability of resistive switching parameters such as ratios between the high and low resistance states as well as variability in switching (programming) voltages. These instabilities would thereby be consequences of the variations in the size, direction, and length of the conduction paths, especially in the case of filamentary conduction. In order to improve the reliability, one can deposit alternate layers of lower and higher electrical resistance materials. Such kind of structural tuning has been applied, e.g., in Al<sub>2</sub>O<sub>3</sub>/HfO<sub>2</sub>/Al<sub>2</sub>O<sub>3</sub> stacks where, as proposed, the tips of localized conduction paths led to the formation or rupture of the conducting filaments mainly at the interfaces between HfO<sub>2</sub> and Al<sub>2</sub>O<sub>3</sub> layers during the SET and RESET events [25]. The same concept was proposed to suppress the endurance fluctuation by controlling the filament rupture points to internal interfaces in double-layered media consisting of HfO<sub>2</sub> and ZrO<sub>2</sub> layers [26]. Yet in another study, the suppression of data retention failures, i.e. reliability increment, was directly attributed to the blocking of oxygen ion transport by an Al<sub>2</sub>O<sub>3</sub> layer in HfO<sub>2</sub>/Al<sub>2</sub>O<sub>3</sub>

multilayers [27]. Increased control over lengths and depths of the switchable conduction paths enabled stabilization of the switching parameters and increase in the reliability of the memristive characteristics.

As was stated earlier [28], interfaces between the component layers in laminates allow one to control the electronic band structure of the system leading to the barriers for charge transport, introduce local mechanical strains influencing structure and crystallization at the interface, and thus provide additional degrees of freedom for device implementation. In a comparative study on Al<sub>2</sub>O<sub>3</sub> and HfO<sub>2</sub> single layers, and Al<sub>2</sub>O<sub>3</sub>/HfO<sub>2</sub> nanolaminates [29], the superiority of the nanolaminates over single layers in terms of uniformity and self-compliance was demonstrated.

Stacks and periodical nanolaminates for ReRAM cells can be fabricated by physical vapor deposition, using, e.g., sputtering technique [26,30]. Hereby atomic layer deposition method has become the emerging method for the fabrication of switching media consisting of, e.g., Al<sub>2</sub>O<sub>3</sub> and HfO<sub>2</sub> [25,27,28], or Al<sub>2</sub>O<sub>3</sub> and TiO<sub>2</sub> [28].

Ta<sub>2</sub>O<sub>5</sub> has probably been the most extensively considered and prospective memristor oxide [**31**], whereas Nb<sub>2</sub>O<sub>5</sub> films have been studied less frequently for this purpose, although Nb<sub>2</sub>O<sub>5</sub> is related to Ta<sub>2</sub>O<sub>5</sub> in terms of abundance, crystallography, electronic and optical properties. Nb<sub>2</sub>O<sub>5</sub> may be regarded as somewhat more conductive (electrically leaky) oxide compared to Ta<sub>2</sub>O<sub>5</sub>, due to its smaller band gap [**23**, **32**], and possibly larger stoichiometry deviations accompanied by oxygen vacancies. Despite the electrically relatively leaky character of niobium oxide, memristive structures [**33**] based on Nb<sub>2</sub>O<sub>5</sub> have been proposed, either combined with Ta<sub>2</sub>O<sub>5</sub> [**34**], deposited as single oxide layers [**33-38**], or layered alternately with other oxides such as TiO<sub>2</sub> [**39**]. The effective band gap of Nb<sub>2</sub>O<sub>5</sub> can be widened and, concurrently, insulating properties improved in Nb<sub>2</sub>O<sub>5</sub> films upon mixing them with widerband-gap oxides such as Al<sub>2</sub>O<sub>3</sub> [**40**] or SiO<sub>2</sub>. Electronic properties of wider band-gap Al<sub>2</sub>O<sub>3</sub> doped with Nb<sub>2</sub>O<sub>5</sub> by ALD using Nb(OC<sub>2</sub>H<sub>5</sub>)<sub>5</sub> and water as precursors have been investigated [**41**]. Regarding the ALD method, Nb<sub>2</sub>O<sub>5</sub> films have mainly been grown from Nb(OC<sub>2</sub>H<sub>5</sub>)<sub>5</sub> and water [**42**, **43**]. ALD-Nb<sub>2</sub>O<sub>5</sub> has been studied as a component of negative electrode in energy storage microdevices [**44**], or as a resistively switching medium [**45**].

SiO<sub>2</sub> has also been considered as a prospective memristive material exhibiting resistive switching behavior either deposited alone [12,46,47] or combined with metals or metal oxides [7,13-17]. Many metal oxides have, to date, been explored and listed as candidates for memristive dielectrics [48] but SiO<sub>2</sub> has been described as a compound which after combination with some other oxides or metals may actually promote the resistive switching behavior in a hosting metal oxide [49].

In the present paper, ALD of SiO<sub>2</sub>-Nb<sub>2</sub>O<sub>5</sub> mixture films and nanolaminates was carried out using a novel combination of cation precursors: niobium pentaethoxide, Nb(OC<sub>2</sub>H<sub>5</sub>)<sub>5</sub>, and hexakis(ethylamino) disilane, Si<sub>2</sub>(NHC<sub>2</sub>H<sub>5</sub>)<sub>6</sub>. Ozone, O<sub>3</sub>, was applied as an oxidizer. The goal of the study was to obtain primary knowledge about the correlation between elemental composition, cation ratio, and growth cycle ratio of the constituent oxides. In terms of functionality, the Nb<sub>2</sub>O<sub>5</sub>-SiO<sub>2</sub> films were characterized by electrical measurements conducted in the resistive switching regime. The goal of the electrical evaluation was to clarify whether the multilayers grown at a rather low temperature (300 °C) and stacked between electrodes of moderate work function could perform as switching media. It was examined whether it is possible to empirically find a balance between defect densities and composition of the mixtures in order to improve their electrical performance. Further, small signal current and admittance measurement procedures were exploited to help distinction of the conductance states also in the case of films which are so leaky that the low and high resistivity states become difficult to distinguish in the common d.c. measurements.

## **Experimental details**

The films were grown in a commercial flow-type hot-wall reactor F120 [**50**] (ASM Microchemistry, Ltd.). Growth temperature was held at 300 °C. The silicon precursor used was hexakis(ethylamino)disilane [**18,19**], i.e. Si<sub>2</sub>(NHEt)<sub>6</sub>, AHEAD (Air Liquide, Inc.), evaporated at 65-67 °C. The niobium precursor, Nb(OC<sub>2</sub>H<sub>5</sub>)<sub>5</sub>, was evaporated at 90-93 °C. Both precursors were evaporated from open boats inside the reactor, and transported to the substrates by the N<sub>2</sub> flow. The pressure in the reactor chamber during the deposition process was below 10 mbar. Ozone as the oxygen precursor was produced with a Wedeco Ozomatic

Modular 4 HC ozone generator from oxygen (99.999%, Linde Gas). The estimated ozone concentration output of the generator was about  $100 \text{ g/m}^3$ .

The cycle times used were 0.5-0.5-2.0-0.5 s, denoting the sequence Si or Nb precursor pulse– purge–O<sub>3</sub> pulse–purge. The substrates were pieces of wafers, cut out of undoped Si(100) covered with less than 2.0 nm thick wet-chemically-grown SiO<sub>2</sub>. Electrically conducting substrates were also used, based on (100) silicon with a resistivity of 0.014–0.020  $\Omega \cdot$  cm, i.e., Si boron-doped to the concentrations up to  $5 \times 10^{18} - 1 \times 10^{19}$ /cm<sup>3</sup>, and coated with 10 nm thick titanium nitride layer. Such structures enable convenient through-wafer measurements during electrical evaluation of the oxide layers. TiN was chemical vapor deposited using TiCl<sub>4</sub>/NH<sub>3</sub> process in an ASM A412 Large Batch 300 mm reactor at Fraunhofer IPMS-CNT. The SiO<sub>2</sub>-Nb<sub>2</sub>O<sub>5</sub> films were grown to thicknesses ranging from ca. 10 to 150 nm. The thickest films were grown for the conveniency of composition analysis.

X-ray diffraction (XRD) measurements were performed with PANalytical X'Pert Pro MPD diffractometer using CuK $\alpha$  radiation (1.5406 Å), 1/4° divergence slit and parallel beam optics. The same setup with a smaller  $1/16^{\circ}$  divergence slit, Cu attenuator and parallel plate collimator slit was used for X-ray reflectivity (XRR) measurements. The density, thickness and roughness of the layers was acquired by fitting the theoretical model to the measured data using either Reflectivity 1.2 (PANalytical) or Reflex35 [51] softwares. Energy dispersive Xray spectrometry (EDX) was used for the measurements of the relative niobium and silicon contents in the films by a Hitachi S-4800 scanning electron microscope (SEM) equipped with an Oxford INCA 350 EDX spectrometer. In order to avoid interference from the substrate while measuring the silicon content, these measurements were done on samples grown on aluminum foil placed into the reactor chamber simultaneously with the samples grown on silicon. The EDX spectra were measured at 10 keV. The beam current and spectrometer gain were determined from a calibration measurement performed under the same beam conditions. The calculations were done on the basis of k ratios measured for Nb La and Si Ka X-ray lines using a GMRFILM program [52]. Oxygen was calculated from the stoichiometry. The error of the EDX measurements was ~10 %, and could increase to 15 % in the case of the thinnest films, e.g. the 13 nm thick reference Nb<sub>2</sub>O<sub>5</sub>. Contents of residual light elements in selected representative samples were determined by time-of-flight elastic recoil detection

analysis (TOF-ERDA) [53]. The TOF-ERDA was performed with 5 MV tandem accelerator using 50 MeV <sup>127+</sup>I ion beam.

In order to carry out the electrical measurements, metal-insulator-metal (MIM) structures were prepared on the conducting substrates where the TiN served as the bottom electrode. Top electrodes with an area of either 0.052 or 0.204 mm<sup>2</sup> were formed by electron beam evaporation of 120 nm thick Al/Ti dot electrodes through a shadow mask, with the Ti layer contacting the SiO<sub>2</sub> –Nb<sub>2</sub>O<sub>5</sub> dielectric layers. Admittance-voltage and current-voltage measurements were carried out in a light-proof and electrically shielded box. Samples were electrically characterized in both d.c. and a.c. regimes using a Keithley 4200SCS semiconductor analyzer. The bias voltage was applied to the top electrode, while the bottom electrode was grounded. To record the admittance parameters, a small signal of 30 mV r.m.s. was superimposed with the d.c. bias voltage. The measurement frequency did not affect the resistive switching behavior in the range of 20 kHz - 1 MHz.

#### **Results and discussion**

The Nb to Si ratio was varied by changing the ratio of the SiO<sub>2</sub> and Nb<sub>2</sub>O<sub>5</sub> deposition cycles. The numbers of these deposition cycles were varied separately in order to change the thicknesses of the constituent metal oxide layers. In this way, films of different artificial structures were deposited, ranging from SiO<sub>2</sub> doped or mixed with low amounts of Nb<sub>2</sub>O<sub>5</sub> to SiO<sub>2</sub>-Nb<sub>2</sub>O<sub>5</sub> nanolaminates. The growth cycle sequences applied for the different samples are presented in Table I.

ToF-ERDA results of selected films are depicted in Figures 1 and 2 in the form of elemental depth profiles. ToF-ERDA was conducted in order to examine the chemical purity of the films especially in terms of the light residual elements, hydrogen and carbon, originating primarily from the precursor ligands. Elemental contents were calculated omitting the very surface as well as the substrate-film interface.

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|   |  |           | •        |
|---|--|-----------|----------|
| Cycle sequence  | notes  | Nb:Si     | EDX      |
|   |  | EDX       | thick-   |
|   |  | at. ratio | ness, nm |
|   |  |           |          |
| $2000 \times SiO_2$   | thick reference SiO <sub>2</sub>                       | 0         | 122      |
| $250 \times SiO_2$  | reference SiO <sub>2</sub>                             | 0         | 18       |
| $20 \times [10 \times SiO_2 + 2 \times Nb_2O_5] + 10 \times SiO_2$      | Nb <sub>2</sub> O <sub>5</sub> -doped SiO <sub>2</sub> | 0.11      | 19       |
| $50 \times [5 \times SiO_2 + 1 \times Nb_2O_5] + 5 \times SiO_2$        | Nb <sub>2</sub> O <sub>5</sub> -doped SiO <sub>2</sub> | 0.13      | 22       |
| $10 \times [50 \times Nb_2O_5 + 150 \times SiO_2] + 50 \times Nb_2O_5$  | thick laminate with thicker SiO <sub>x</sub>           | 0.18      | ca. 110  |
| $350 \times [3 \times Nb_2O_5 + 3 \times SiO_2] + 3 \times Nb_2O_5$     | thick mixture  | 0.40      | 133      |
| $150 \times [1 \times Nb_2O_5 + 1 \times SiO_2] + 1 \times Nb_2O_5$     | thin "homogeneous" mixture                             | 0.46      | 19       |
| $1000 \times [1 \times Nb_2O_5 + 1 \times SiO_2] + 1 \times Nb_2O_5$    | thick "homogeneous" mixture                            | 0.48      | 120      |
| $30 \times [12 \times Nb_2O_5 + 3 \times SiO_2] + 12 \times Nb_2O_5$    | SiO <sub>2</sub> -doped Nb <sub>2</sub> O <sub>5</sub> | 1.2       | 19       |
| $5 \times [55 \times Nb_2O_5 + 15 \times SiO_2] + 55 \times Nb_2O_5$    | thin "defined" laminate                                | 1.6       | 18       |
| $15 \times [30 \times Nb_2O_5 + 2 \times SiO_2] + 30 \times Nb_2O_5$    | SiO <sub>2</sub> -doped Nb <sub>2</sub> O <sub>5</sub> | 2.9       | 21       |
| $10 \times [170 \times Nb_2O_5 + 20 \times SiO_2] + 170 \times Nb_2O_5$ | thick laminate, "thick" NbOx                           | 3.5       | 78       |
| $30 \times [15 \times Nb_2O_5 + 1 \times SiO_2] + 15 \times Nb_2O_5$    | SiO <sub>2</sub> -doped Nb <sub>2</sub> O <sub>5</sub> | 7.2       | 19       |
| $350 \times Nb_2O_5$  | reference Nb <sub>2</sub> O <sub>5</sub>               | -         | 13       |
|   |  |           |          |

Table I. ALD cycle sequences for SiO<sub>2</sub>-Nb<sub>2</sub>O<sub>5</sub> films with the thicknesses and Nb:Si cation ratios measured by EDX in the order of ascending Nb/Si ratio.

The SiO<sub>2</sub> film grown on Si(100) substrate using 2000 ALD cycles (Fig. 1) contained 28.2  $\pm$  0.3 at.% silicon, 64.0  $\pm$  0.5 at.% oxygen, 7.7  $\pm$  0.6 at.% hydrogen, less than 0.01 at.% carbon, and less than 0.03 at.% nitrogen. The SiO<sub>2</sub>–Nb<sub>2</sub>O<sub>5</sub> film grown using the cycle sequence 1000  $\times$  [ 1  $\times$  Nb<sub>2</sub>O<sub>5</sub> + 1  $\times$  SiO<sub>2</sub>] + 1  $\times$  Nb<sub>2</sub>O<sub>5</sub> (Fig. 2) contained 21.0  $\pm$  0.3 at.% silicon, 8.9  $\pm$  0.1 at.% niobium, 68.3  $\pm$  0.5 at.% oxygen, 1.5  $\pm$  0.3 at.% hydrogen, 0.23  $\pm$  0.03 at.% carbon, and 0.15  $\pm$  0.02 at.% nitrogen. The Nb:Si cation ratio measured for the latter film by EDX on Al foil was 0.48, compared to the ratio of 0.42 obtained by ToF-ERDA. Thus the EDX and ToF-ERDA results agree reasonably well in terms of film composition.

It can be seen, that the hydrogen content is markedly high in the SiO<sub>2</sub> films (Figure 1), exceeding 7 at.%. At the same time, in the films mixed with Nb<sub>2</sub>O<sub>5</sub> (Figure 2) the hydrogen content was decreased down to 1-2 at.%. In addition, elemental composition of a 15 nm thick reference Nb<sub>2</sub>O<sub>5</sub> film was measured. The elements were detected in the relative contents of 27  $\pm$  0.5, 71.2  $\pm$  1.4 and 0.91  $\pm$  0.13 at.% for niobium, oxygen and hydrogen, respectively. The

content of carbon remained below the reliable measurement level in this  $Nb_2O_5$  sample, i.e. below 0.2 at.%. No other elements were detected.



Figure 1. Elemental depth profile from ToF-ERDA for a  $SiO_2$  film grown on Si(100) substrate using 2000 ALD cycles. The thickness measured by EDX was 122 nm.



Figure 2. Elemental depth profile from ToF-ERDA for a Nb<sub>2</sub>O<sub>5</sub>-SiO<sub>2</sub> film grown on Si(100) substrate using cycle sequence of  $1000 \times [1 \times Nb_2O_5 + 1 \times SiO_2] + 1 \times Nb_2O_5$ . The thickness measured by EDX was 120 nm.

All the films - single oxides, mixed oxides and nanolaminates - were amorphous as examined by X-ray diffraction measurements. X-ray reflectivity (XRR) measurements were performed to examine the distinction between the SiO<sub>2</sub> and Nb<sub>2</sub>O<sub>5</sub> layers constituting the nanolaminate. In this way, one could prove the nanolaminate structure of the samples with relatively thicker constituent metal oxide layers. XRR measurements of selected samples showed a clear evidence of layered structure consisting of the niobium and silicon oxides (Fig. 3). The highest sharp maxima that are the most prominent, are characteristic of the superlattice structures [54-57] and indicate that the bilayers are uniform with appreciably sharp interfaces. The period of the superlattice maxima becomes gradually longer or shorter as the bilayer thickness decreases or increases, respectively [54-57]. The shorter oscillation period is inversely proportional to the total nanolaminate thickness. These shorter oscillations are in a good agreement with the simulations, giving further evidence of an appreciable layer thickness uniformity (Fig. 3). Anyhow, some variations in the intermediate layer thicknesses and densities may be caused by thickness profiles formed across the substrate due to the deviations in adsorption rates and nucleation densities along the gas flow directions in the cross-flow type reactor used. The niobium oxide layers deposited with smaller numbers of cycles have lower densities, which is quite ntural because these films represent the earliest stages of the nucleation and growth (Fig. 3). Such factors may also somewhat complicate the modelling of relatively complex reflection patterns obtained from the periodical multilayers.

Further proof of the nanolaminate quality was obtained by fitting the measured data, as the stack could be modelled adequatelyby repeating equal bilayers of Nb<sub>2</sub>O<sub>5</sub> and SiO<sub>2</sub> and a Nb<sub>2</sub>O<sub>5</sub> capping layer on top. In other words, all the bilayers in the nanolaminate were considered equal. The differences observed in the Figure 3, the top and second panels in the left column, are due to small differences between the adjacent bilayers. Thus, the nanolaminates, which were deposited using the different cycle sequences indicated in the Table I and in Fig. 3 allowed one to achieve satisfactory fit between the measured data and modelled curves. Native SiO<sub>2</sub> layer of ~1.2-1.4 nm thickness had to be included between the film and silicon substrates in the models in order to achieve the best fits.

As mentioned above, the period of the maxima caused by the multilayer structure considerably increased due to the decrease in the bilayer thickness. Obviously, in order to visualise the ordering in the multilayer with the thinnest interlayers in this study, one had to measure the reflectivity patterns in markedly extended ranges of angles (Fig. 3). It became evident, that two ALD cycles of SiO<sub>2</sub> deposited alternately with the hosting Nb<sub>2</sub>O<sub>5</sub> layers was sufficient to create multilayered structures with distinctive periodicity. Thus, the film deposited using the cycle sequence  $15 \times [30 \times Nb_2O_5 + 2 \times SiO_2] + 30 \times Nb_2O_5$  (Fig, 3, the 3rd panel), was modelled to consist of 1.05 nm thick Nb<sub>2</sub>O<sub>5</sub> and, nominally, 0.07 nm thick SiO<sub>2</sub> layers, providing a satisfactory fit with the measured curve. The thickness obtained for SiO<sub>2</sub>, 0.07 nm, remains below a thickness of a monolayer, meaning that the SiO<sub>2</sub> layer was not continuous, and the result rather reflects a periodical two-dimensional confinement of the elemental constituents with specific electron density, within the multilayer structure.



Figure 3. Representative X-ray reflectivity patterns measured from Nb<sub>2</sub>O<sub>5</sub>-SiO<sub>2</sub> nanolaminates deposited using ALD cycle sequences as given by labels, and thickness, d, density,  $\rho$ , and roughness,  $\sigma$  (right column), as results obtained from fitting the XRR patterns. Note the differences in the ranges of the measurement angles. In the two upper panels of the left column, the measured and fitted curves are shifted vertically in relation to each other for the sake of clarity. The schematic layer thicknesses in the right column are not to scale.

# **Electrical performance**

Leakage currents in Nb<sub>2</sub>O<sub>5</sub> and SiO<sub>2</sub>-Nb<sub>2</sub>O<sub>5</sub> nanolaminates occurred too high for the reliable recognition and stabilization of low resistivity state when measured in the RRAM regime. In such samples, no current-voltage hysteresis and memory windows between low resistance states (LRS) and high resistance states (HRS) developed. Besides the highly defective nature of amorphous Nb<sub>2</sub>O<sub>5</sub> films and their likely oxygen deficiency, high leakage current could also arise due to the application of low work function Ti and TiN electrodes. Earlier, resistive switching behavior of Nb<sub>2</sub>O<sub>5</sub> films were stacked between higher work function and noble platinum bottom and top Ti electrodes [45]. In the present study, the emphasis was put on the behavior of the dielectric material and its internal ability or disability to host the switching current. As a result and in a clear contrast with the nanolaminates, behavior characteristic of RRAM material was achieved and recorded with the SiO<sub>2</sub>-Nb<sub>2</sub>O<sub>5</sub> mixture films grown using SiO<sub>2</sub>:Nb<sub>2</sub>O<sub>5</sub> ALD cycle ratio of 5:1 (Figure 4, top panel).

Figure 4, middle panel, depicts in a generalized form the voltage and corresponding current signals applied and measured, respectively, on resistively switching stacks consisting of an oxide or a combination of different oxides, mounted between metallic electrodes. During the conventional resistive switching measurement, rectangular voltage pulses are applied on the electrodes, with the amplitude increasing within the sequence of pulses. At the increasing voltage pulses, values of current are recorded, which, in the sequence, form the current-voltage curve until the transition (switching) to the low resistance state. Upon back-forth sweeping the sequence of voltage pulses with gradually and alternately increasing and decreasing amplitudes, the envelope curves of the current-voltage dependences were recorded (Fig. 4, top panel) with hysteresis characteristic of resistive switching behavior in a memristive material. Fig. 4 thus depicts defined memristive behavior characterizing Nb<sub>2</sub>O<sub>5</sub>:SiO<sub>2</sub> films with Nb:Si atomic ratio of 0.13 grown using SiO<sub>2</sub>:Nb<sub>2</sub>O<sub>5</sub> cycle ratio of 5:1.



Figure 4 Sequential current-voltage loops measured in resistive switching regime from 22 nm thick SiO<sub>2</sub>-Nb<sub>2</sub>O<sub>5</sub> film deposited with SiO<sub>2</sub>:Nb<sub>2</sub>O<sub>5</sub> cycle ratio of 5:1 (upper panel); sampling voltage pulses applied on a film to record switching currents and small signal current values detected at 0.1 V as memory maps (middle panel); and current-voltage loop measured at low-reading voltage (0.1 V) (bottom panel).

Besides the envelope curves consisting of current values measured at the variable bias voltage amplitudes, low reading voltage hysteron curves were recorded as well. In the latter case, currents were read at voltage values of 0.1 V in between the sequential sampling bias voltage pulses. The value of the measured current was, dominantly, determined by the two resistivity states achieved alternately, and two clearly defined plateaus were reached and passed through before and after the sequential SET and RESET transitions (Fig. 4, bottom panel). Current-voltage loop with remarkable squareness, also defined as a current memory map, was thus formed with the memory window between the high and low current states reaching 0.023 A/cm<sup>2</sup>.

Admittance parameters were also recorded while sweeping the bias voltage. In order to measure differential capacitance, 30 mV rms ac signal at a frequency of 500 kHz was superimposed to the voltage bias. Parallel admittance model [58] was exploited to determine conductance, G, and capacitance, C, values against lowreading voltage sampling. In order to obtain the memory maps for G and C return-to-zero voltage pulse sequences were used. That is, on the sample in high resistance state, positive voltage pulses of 1 ms were applied. After each pulse, the voltage was returned to 0 V and the admittance-derived values (G, C) were recorded immediately. The amplitude of the voltage pulse (V<sub>P</sub>) was increased linearly until the transition from the high resistance state to the low resistance state occurred. Once the sample had reached the low resistance state, V<sub>P</sub> was linearly decreased in the opposite direction, i.e. towards negative values. When the V<sub>P</sub> reached values negative enough to enable switching the device back to the high resistance state (i.e. provide RESET transition), it was linearly increased to the zero voltage. The plots of C (Fig. 5) and G (Fig. 6) as functions of the bias voltage, V<sub>P</sub>, thus constituted the corresponding memory maps. The bias sampling voltage herewith serves also as the programming voltage.



Figure 5. Capacitance-voltage hysteron loop measured at programming voltage values during sampling voltage pulse swept on 22 nm thick SiO<sub>2</sub>-Nb<sub>2</sub>O<sub>5</sub> film deposited with SiO<sub>2</sub>:Nb<sub>2</sub>O<sub>5</sub> cycle ratio of 5:1.



Figure 6. Conductance-voltage hysteron loop measured at at programming voltage values during sampling voltage pulse swept on 22 nm thick SiO<sub>2</sub>-Nb<sub>2</sub>O<sub>5</sub> film deposited with SiO<sub>2</sub>:Nb<sub>2</sub>O<sub>5</sub> cycle ratio of 5:1.

The parameters of the memory maps, i.e., the switching voltages (fields), and windows between conductance and capacitance extrema, were rather insensitive to the measurement frequency in the whole range examined, 20 kHz - 1 MHz. Regardless of frequency,  $G_0$  and  $C_0$ remained very stable in a wide  $V_P$  range, being indicative of stable conduction paths formed during switching. The width of the conductance hysteron extended to 700  $\mu\Omega^{-1}$ , with 110 and  $800 \ \mu\Omega^{-1}$  in high and low resistivity states, respectively, promising lowered power consumption while storing information if read at the high resistivity state. RESET and SET transitions were well-defined one-step processes (Figs. 5, and 6), although not as abrupt as RESET in the case of some other materials, such as Ta<sub>2</sub>O<sub>5</sub>-TiO<sub>2</sub>-Ta<sub>2</sub>O<sub>5</sub> trilayer structures also grown by ALD [58]. In the latter case, the SET process tended to be gradual, passing certain steps between the high and low resistance states, which is not observed in the present case. It is possible, that a few-layer stacks may become more asymmetric being RESET more steeply than SET with an implication that switching on, as well as restoring highly conductive state can be an instantaneous process, whereas its breaking and disappearance is more gradual. For instance, in the present case, in a 22 nm thick film grown using cycle sequence of  $50 \times [5 \times$  $SiO_2 + 1 \times Nb_2O_5 + 5 \times SiO_2$ , both SET and RESET transitions could become relatively slow without marked dependence on polarity. Earlier, certain differences in conduction current densities in materials between Ti and TiN contact electrodes were measured [60] and, assumptionally, asymmetric behavior of oxide stacks between Ti and TiN could be assumed. Nonetheless, the resistive switching performance in the case of SiO<sub>2</sub>:Nb<sub>2</sub>O<sub>5</sub> film with thin SiO<sub>2</sub>:Nb<sub>2</sub>O<sub>5</sub> intermediates grown using small numbers of the growth cycles for both constituents in periodical stacks remained symmetric. The latter was valid at least when the film had appreciably high thickness. Further, a more important issue related to the possibility of application of such material layers, was generally connected to the highly leaky nature of SiO<sub>2</sub>:Nb<sub>2</sub>O<sub>5</sub> films, accompanied by difficulties in keeping the material initially in a low resistivity state, and also difficulties to find the approprite cycle and cation ratios between the constituent oxides to provide switching resistance regime.

In the rest of the films grown in this study, the leakage currents occurred too high for the clear distinction between the low and high resistance states in the conventional resistive switching regime. The memory window between the currents in LRS and HRS remained inadequately narrow for reliable resistive switching behavior (Fig. 7, topmost panel). That was most likely due to the leaky nature of both Nb<sub>2</sub>O<sub>5</sub> and SiO<sub>2</sub> layers, at first due to their disordered structure

and, secondly, also because of the relatively high contents of residual elements, such as hydrogen. In the nanolaminate structures, the potential barriers likely forming between the constituent oxide layers may, in some extent, assist in suppressing leakage currents. However, in the present case the number of barriers, if existing in such thin films, were clearly not high enough to compensate the detrimental effects of structural and chemical disorder.

It is noteworthy that the small signal current hysteron measurements enabled significantly improved distinction between the two resistivity states in the film grown with a SiO<sub>2</sub>:Nb<sub>2</sub>O<sub>5</sub> cycle ratio of 10:2 (Fig. 7, the 2nd panel from top). The current-voltage loops within the low-reading voltage measurements were, obviously, not as well defined in terms of squareness as those in the adequately performing SiO<sub>2</sub>-Nb<sub>2</sub>O<sub>5</sub> sample grown with a cycle ratio of 5:1 (Fig. 4, bottom panel). However, the measurement mode had a major effect in terms of the distinction between the memory states. Notably, that behavior was recorded for capacitance (Fig. 7, the 3rd panel from top) and conductance (Fig. 7, bottom panel) as well.

Similar results were found for other selected samples as well as for reference  $SiO_2$  film (not shown), which alone could not satisfactorily perform as memristive material, plausibly due to high amount of residues, such as hydrogen, and also carbon. Clearly, to ensure performance of materials as resistively switching media, one has to either adjust the composition of mixtures at high accuracy, or write/read the resistivity states in the admittance regime.



Figure 7. Current-voltage loops measured in conventional resistive switching mode (top panel); low-reading voltage memory map (the 2nd panel from top), small signal capacitance memory map (the 3rd panel from top); and small signal conductance memory map (bottom panel) against sampling voltage pulses applied on 19 nm thick SiO<sub>2</sub>-Nb<sub>2</sub>O<sub>5</sub> film deposited with SiO<sub>2</sub>:Nb<sub>2</sub>O<sub>5</sub> cycle ratio of 10:2.

# Conclusions

Thin mixed films and nanolaminates consisting of Nb<sub>2</sub>O<sub>5</sub> and SiO<sub>2</sub> layers grown alternately by atomic layer deposition were devised at a substrate temperature of 300 °C. The precursor system used was a novel combination of Nb(OC<sub>2</sub>H<sub>5</sub>)<sub>5</sub>, Si<sub>2</sub>(NHC<sub>2</sub>H<sub>5</sub>)<sub>6</sub>, and O<sub>3</sub>. The niobium to silicon atomic ratio in the films grown to thicknesses from 13 to 130 nm was varied in the range of 0.11-7.20. The films were amorphous in the as-deposited state and were also characterized in the as-deposited state.

In the nanolaminate structures consisting of 2-5 nm thick distinct layers, the densities of  $Nb_2O_5$  and  $SiO_2$  layers reached 4.9 and 1.8 g/cm<sup>3</sup>, respectively, approaching bulk values. The content of hydrogen as the main light residue exceeded 7 at. % in  $SiO_2$ , but decreased in mixed  $SiO_2$ -Nb<sub>2</sub>O<sub>5</sub> films down to 1-2 at.%, and in reference Nb<sub>2</sub>O<sub>5</sub> film even below the reliable measurement level.

Electrical measurements on Nb<sub>2</sub>O<sub>5</sub>-SiO<sub>2</sub> nanolaminates implied that the performance of the films was strongly influenced by the relative content of the constituent oxides in the whole structure. Appreciably high current ratios between the high and low resistance states extending over two orders of magnitude could actually be achieved only in a mixture film deposited with a SiO<sub>2</sub>:Nb<sub>2</sub>O<sub>5</sub> cycle ratio of 5:1, possessing Nb:Si cation ratio of 0.13. At the same time, small signal measurements enabled recording, in several SiO<sub>2</sub> and Nb<sub>2</sub>O<sub>5</sub> based films, voltage-dependent admittance maps showing two distinct memory states with low and high values of both capacitance and conductance. Such admittance-voltage loops demonstrated well-defined squareness. One can conclude, that the admittance signal measurements allow definition of memory windows also for such materials which possess too high conductivity to allow switching between the low and high resistance states in the conventional measurements.

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