University of Mississippi

eGrove

Faculty and Student Publications

Pharmacy, School of

4-11-2019

Flavonoid glycosides from endemic bulgarian astragalus aitosensis (Ivanisch.)

Hristo Vasilev Medical University of Sofia

Samir Ross University of Mississippi School of Pharmacy

Karel Šmejkal Veterinární a farmaceutická univerzita Brno

Petr Maršík Czech University of Life Sciences Prague

Dagmar Jankovská Veterinární a farmaceutická univerzita Brno

See next page for additional authors

Follow this and additional works at: https://egrove.olemiss.edu/pharmacy_facpubs

Recommended Citation

Vasilev, H., Ross, S., Šmejkal, K., Maršík, P., Jankovská, D., Havlík, J., & Veselý, O. (2019). Flavonoid Glycosides from Endemic Bulgarian Astragalus aitosensis (Ivanisch.). Molecules, 24(7), 1419. https://doi.org/10.3390/molecules24071419

This Article is brought to you for free and open access by the Pharmacy, School of at eGrove. It has been accepted for inclusion in Faculty and Student Publications by an authorized administrator of eGrove. For more information, please contact egrove@olemiss.edu.

Authors

Hristo Vasilev, Samir Ross, Karel Šmejkal, Petr Maršík, Dagmar Jankovská, Jaroslav Havlík, and Ondřej Veselý





Article Flavonoid Glycosides from Endemic Bulgarian Astragalus aitosensis (Ivanisch.)

Hristo Vasilev ¹, Samir Ross ², Karel Šmejkal ³,*, Petr Maršík ^{4,5}, Dagmar Jankovská ³,*, Jaroslav Havlík ⁴ and Ondřej Veselý ⁴

- ¹ Department of Pharmacognosy, Faculty of Pharmacy, Medical University–Sofia, 2 Dunav str., 1000 Sofia, Bulgaria; christo.vasilev@yahoo.com
- ² National Center for Natural Products Research, School of Pharmacy, University of Mississippi, Oxford, MS 38677, USA; sross@olemiss.edu
- ³ Department of Natural Drugs, Faculty of Pharmacy, University of Veterinary and Pharmaceutical Sciences Brno, 61242 Brno, Czech Republic
- ⁴ Department of Food Science, Czech University of Life Sciences Prague, Kamycka 129, 165 00 Prague 6–Suchdol, Czech Republic; marsik@af.czu.cz (P.M.); havlik@af.czu.cz (J.H.); veselyo@af.czu.cz (O.V.)
- ⁵ Institute of Experimental Botany, Czech Academy of Sciences, Rozvojová 263, 16502 Prague, Czech Republic
- * Correspondence: karel.mejkal@post.cz (K.Š.); jankovskad@vfu.cz (D.J.); Tel.: +42-072-424-3643 (K.Š.); +42-054-156-2832 (D.J.)

Academic Editor: Francesco Epifano Received: 28 March 2019; Accepted: 6 April 2019; Published: 11 April 2019



Abstract: *Astragalus* is a very interesting plant genus, well-known for its content of flavonoids, triterpenes and polysaccharides. Its secondary metabolites are described as biologically active compounds showing several activities, e.g., immunomodulating, antibacterial, antiviral and hepatoprotective. This inspired us to analyze the Bulgarian endemic *A. aitosensis* (Ivanisch.) to obtain deeper information about its phenolic components. We used extensive chromatographic separation of *A. aitosensis* extract to obtain seven phenolic compounds (1–7), which were identified using combined LC-MS and NMR spectral studies. The 1D and 2D NMR analyses and HR-MS allowed us to resolve the structures of known compounds **5–**7 as isorhamnetin-3-*O*-robinobioside, isorhamnetin-3-*O*-(2,6-di-*O*- α -rhamno-pyranosyl- β -galactopyranoside), and alangiflavoside, respectively, and further comparison of these spectral data with available literature helped us with structural analysis of newly described flavonoid glycosides **1–4**. These were described in plant source for the first time.

Keywords: Astragalus aitosensis; flavonoid; glycoside

1. Introduction

Genus *Astragalus* comprises from more than 2500 species, which makes it the largest genus in the family Fabaceae. *Astragalus* species are cosmopolitans, widely distributed in dry and semi-dry regions, mainly in the temperate regions of the Northern hemisphere [1]. Certain sources assign genus *Astragalus* as the largest genus of flowering plants [2]. About 133 species are distributed in Europe [3], and 29 have been identified in Bulgaria [4,5]. Fourteen *Astragalus* species from the Bulgarian flora are protected by the Bulgarian Biodiversity Act and they are included in the Red List of Bulgaria, as is the species *A. aitosensis*, the focus in this investigation [4].

The object of our study—*A. aitosensis* (Ivanisch.) (syn. *Astracantha aitosensis* (Ivanisch.) Podlech, *Astracantha arnacantha* (M. Bieb.) Podlech subsp. *aitosensis* (Ivanisch.) Réer & Podlech)—is a Bulgarian endemic plant and yet it is very scarcely studied for its phytochemical composition [6,7]. *A. aitosensis*

is a low, spiny, tussock-forming shrub with strongly branched stems (30–50 cm in height) [8]. The plant grows in dry stony places (90–550 m alt) with neutral to alkaline soil. It is distributed only in the suburbs of the small Bulgarian town Aytos, which gives rise to its name.

From pharmacological point of view, *Astragalus* species are well-known and widely used as remedies in the traditional folk medicine of different countries, but only about 100 species from the genus are researched for their phytochemical composition and properties. Their activity described in the literature is a result of the presence of flavonoids, saponins, and polysaccharides [9,10], which endue these plants with immunomodulating, antibacterial, antiviral, hepatoprotective and other protective pharmacological effects [1,9,10].

Astagalus species show diverse flavonoid composition: flavons, flavonols, flavanons, flavanonols, chalcones, isoflavons, isoflavans and pterocarpans [1]. Flavonols—in their free and glycosidic forms—are the most common aglycons, particularly kaempferol, quercetin and methylquercetin (syn. isorhamnetin). Astragalin, rutoside, hyperoside and isoquercitrin are the most frequently found glycosidic forms [1,10]. Interesting recent research by Bulgarian scientists reports the presence of a rarely-met group of flavo-alkaloids [11] and heavily glycosylated tri- and tetra-flavonols [10].

The aim of this study was to isolate content compounds of the methanolic extract from aerial parts of *A. aitosensis* and preform their structural elucidation using ¹H, ¹³C, COSY, HSQC, HMBC, NOESY, and TOCSY NMR experiments. HR-ESI-MS was used for additional confirmation of the structures revealed by NMR. We report here the isolation and structural elucidation of six isorhamnetin and one kaempferol glycosides 1-7, four of which are new natural glycosides: three tetra- (1–3) and one triglycoside (4). The other three already known structures were determined for the first time in *A. aitosensis*: with two (5), three (6) and four (7) sugar units, respectively.

2. Results

2.1. Isolation of Compounds

The aerial parts of *A. aitosensis* were extracted with 80% methanol under reflux. The preliminary chromatographic analysis showed a bunch of signals of flavonoid compounds, with retention times predicting a high degree of glycosylation (Figure 1).

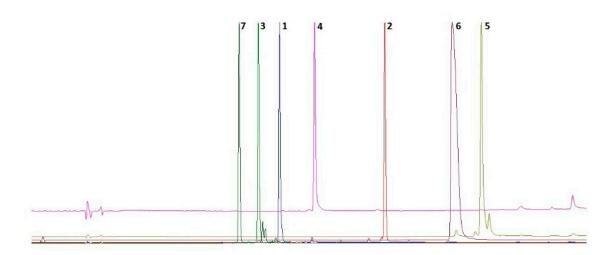


Figure 1. The exemplified overlay of chromatograms of compounds 1–7 at λ 254 nm.

The crude extract was therefore defatted by liquid–liquid partitioning with chloroform and further fractionated via successive column chromatography with final step of semi-preparative HPLC purification of the isolated compounds (Figure 2). Their nature (UV spectral properties) and the behavior of these compounds during separation on reversed phase—the polar character of the compounds—gave us the idea of flavonoid glycosides with a high glycosylation pattern.

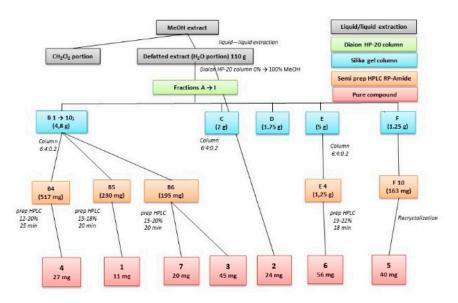


Figure 2. Simplified scheme of separation.

Our separation procedures resulted in isolation of seven pure compounds (1–7) (Figure 3).

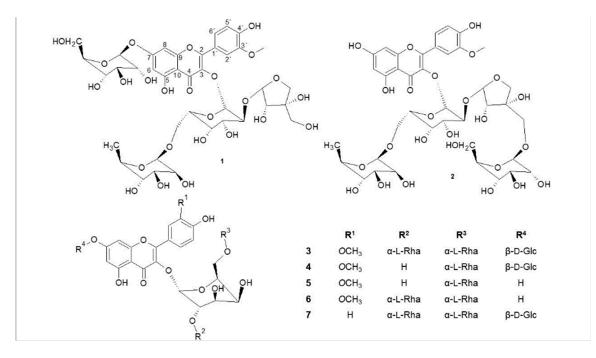


Figure 3. Structures of isolated compounds 1–7.

2.2. Structural Analysis

To the best of our knowledge, spectral data of four of the isolated compounds (1–4) did not correspond to the data of compounds previously published in the literature. The 1D and 2D-NMR analysis, HR-MS and comparison with literature allowed us to resolve the structures of compounds 5–7 as isorhamnetin-3-*O*-robinobioside [12], isorhamnetin-3-*O*-(2,6-di-*O*- α -rhamno-pyranosyl- β -galactopyranoside) [13,14], and alangiflavoside [15], respectively. The further comparison of these spectral data with available literature helped us with structural analysis of newly described flavonoid glycosides 1–4. For a detailed description, please see Supplementary Materials, Figures S1–S77.

The aglycones for compounds **1–6** were determined based on HR-ESI-MS and NMR (¹H, ¹³C, COSY, HSQC and HMBC) spectral analysis. ¹H and ¹³C spectra are shown in Table **1**. HMBC spectra showed the following significant correlations: proton at C-2' was a doublet with *meta* coupling and, in the HMBC, it showed strong correlation to C-4' and weak to C-3'. The proton at C-6' was observed as doublet of doublet (*ortho* and *meta* coupling) and showed strong correlation in the HMBC to C-4'. The proton at C-5' was a doublet with *ortho* coupling, displaying in the HMBC the strong correlation to C-3' and weak to C-3'. The proton of the methoxy is at C-3' and the aglycone of compounds **1–6** was finally identified as 3'-O-methylquercetin, syn. isorhamnetin [16].

Table 1. ¹³C-NMR δ_C (ppm) (100 MHz); ¹H-NMR (600 MHz for 1 and 500 MHz for 2). δ_H (ppm), multiplicity (*J* in Hz).

	1		2		
	δ _C , Type	δ _H (J in Hz)	δ _C , Type	δ _H (J in Hz)	
2	157.8, C		157.2, C		
3	133.6, C		133.3, C		
4	178.0, C		177.8, C		
5	161.3, C		161.6, C		
6	99.4, CH	6.45, d (2.02)	98.5, CH	6.19, d (2.00)	
7	163.0, C		164.4, C		
8	94.4, CH	6.75, d (2.02)	93.4, CH	6.38, d (2.00)	
9	156.5, C		157.0, C		
10	106.3, C		104.5, C		
1′	121.8, C		121.9, C		
2'	113.3, CH	8.05, m (1.94)	113.7, CH	8.04, d (1.97)	
3'	147.0, C	-OCH ₃	147.0, C	-OCH ₃	
4'	149.4, C	-OH	149.2, C	-OH	
5'	114.5, CH	6.91, m (8.39)	114.5, CH	6.90, d (8.40)	
6'	122.3, CH	7.62, m (1.92; 8.45)	122.1, CH	7.57, dd (1.90; 8.43)	
-OCH3	55.7, CH	4.00 s	55.7, CH ₃	3.98, s	
3-O-gal	,		,	- · · · , -	
1	99.8, CH	5.57, d (7.84)	99.9, CH	5.58, d (7.76)	
2	75.1, CH	3.96, dd	74.8, CH	3.97, dd	
3	73.9, CH	3.73 dd	73.9, CH	3.74 dd	
4	69.0, CH	3.77 dd	69.0, CH	3.77 dd	
5	74.1, CH	3.67 dt	74.0, CH	3.64 dt	
6	66.0, CH ₂	3.47/3.68 dd	65.8, CH ₂	3.47/3.71 dd	
	0000, 0112	0117 / 0100 du		0.11 / 0.1 1 du	
1	109.2, CH	5.43, d (1.55)	109.1, CH	5.44, d (1.73)	
2	76.6, CH	4.01, d	76.9, CH	4.07, d	
3	79.5, C	-OH	78.8, C	-OH	
4	74.2, CH ₂	3.64, d (9.58)	74.3, CH ₂	3.66, d (n/a)	
4	74.2, CH ₂	3.75, d (9.59)	74.3, CH ₂	4.07, d (9.70)	
5	65.1, CH ₂	3.64, d (11.45)	61.3, CH ₂	3.67, d (11.95)	
5	05.1, CH ₂	3.75 d (11.45)	$01.3, C11_2$	3.85 d (11.90)	
Rha (1→6)		5.75 d (11.50)		5.65 ú (11.90)	
				4 50 1 (1 50)	
1	100.5, CH	4.50, d (1.56)	100.5, CH	4.52, d (1.53)	
2	70.7, CH	3.51, dd	70.7, CH	3.57, dd	
3	70.9, CH	3.46, dd	70.9, CH	3.49, dd	
4	72.4, CH	3.25, pt	72.4, CH	3.26, pt	
5	68.3, CH	3.49, dq	68.3, CH	3.51, dq	
6	16.5, CH ₃	1.15, d (6.17)	16.5, CH ₃	1.16, d (6.23)	
Glc	7-O-Glc		Api (5→1) Glc		
1	100.1, CH	5.06, d (7.53)	103.4, CH	4.25, d (7.54)	
2	73.3, CH	3.49, dd	73.6, CH	3.17, dd	
3	77.0, CH	3.55, dd	76.3, CH	3.21, dd	
4	69.9, CH	3.38, dt	70.4, CH	3.27, dd	
5	76.4, CH	3.49, dq	76.3, CH	3.27, dt	
6	61.1, CH ₂	3.69/3.94 dd	72.9, CH ₂	3.68/4.15, dd	

Because of the identification procedure, we describe the elucidation of structures of known compounds prior to the new compounds. Detailed ¹H and ¹³C chemical shifts for compound **5** are listed in Tables 2 and 3, respectively. The LC-MS analysis showed single peak (t_R 7.95 min) chromatogram with a signal of a deprotonated molecule at m/z [M – H]⁻ 623.16205, which showed a good correlation with the calculated value of m/z 623.16176 for C₂₈H₃₁O₁₆⁻ (Δ = 0.00029). The (+)HRMS supported the idea of this molecular formula; the glycosylation pattern was predicted to be the rhamnose and a hexose from MS/MS analysis of m/z [M + H]⁺ 625.1175 and observed fragments m/z [M + H – rhamnose]⁺ 479.1184, m/z [M + H – rhamnose-oxygen]⁺ 463.1213, and finally m/z 317.0645 showing the aglycone. The molecular formula C₂₈H₃₂O₁₆ (calcd 624.54408) accorded with the NMR and further HR-ESI-MS spectral data, and the data of compound **5** corresponded to the reported data for isorhamnetin-3-*O*-robinobioside [12].

Compound/Position -	3	4	5	6	7
Compound/rosition	δ _C , Type	δ _C , Type	δ _C , Type	δ _C , Type	δ _C , Type
2	157.7, C	157.4, C	156.8, C	157.0, C	158.0, C
3	133.3, C	133.8, C	133.5, C	133.0, C	133.4, C
4	177.9, C	178.0, C	177.8, C	177.8, C	177.1, C
5	161.3, C	161.3, C	161.6, C	161.7, C	161.4, C
6	99.3 <i>,</i> CH	99.8, CH	99.2, CH	98.4, CH	99.4, CH
7	163.0, C	163.4, C	164.7, C	164.4, C	163.0, C
8	94.4, CH	95.1, CH	94.2, CH	93.3, CH	94.3, CH
9	156.5, C	156.4, C	156.8, C	157.0, C	156.6, C
10	106.3, C	106.1, C	104.4, C	104.5, C	106.2, C
1′	121.8, C	121.3, C	121.5, C	122.0, C	121.5, C
2′	113.3, CH	113.9, CH	113.9, CH	113.3, CH	131.0, CH
3'	147.1, CH	147.5, CH	147.4, C	147.0, C	114.8, CH
4′	149.3, C	150.1, C	149.9, C	149.1, C	160.1, C
5'	114.6, CH	115.6, CH	114.6, CH	114.5, CH	114.8, CH
6′	122.1, CH	122.6, CH	122.4, CH	121.8, CH	131.0, CH
OCH ₃	56.37, CH ₃	56.39, CH	56.37, CH	55.79, CH ₃	
3-O-Gal					
1	99.3, CH	102.1, CH	102.3, CH	99.4, CH	99.4, CH
2	76.4, CH	74.1, CH	74.0, CH	76.4, CH	76.1, CH
3	74.2, CH	71.6, CH	71.5, CH	74.2, CH	74.3, CH
4	69.2, CH	68.4, CH	68.4, CH	69.1, CH	69.4, CH
5	74.1, CH	73.4, CH	73.4, CH	73.9, CH	74.1, CH
6	65.9, CH ₂	65.6, CH ₂	65.6, CH ₂	65.7, CH ₂	66.0, CH ₂
Rha (1→2)					
1	101.4, CH			101.3, CH	101.2, CH
2	71.0, CH			71.0, CH	71.0, CH
3	71.0, CH			71.0, CH	70.9, CH
4	72.5, CH			72.5, CH	72.7, CH
5	68.4, CH			68.4, CH	68.4, CH
6	16.0, CH ₃			16.0, CH ₃	16.1, CH ₃
Rha (1→6)					
1	100.6, CH	100.5, CH	100.5, CH	100.5, CH	100.5, CH
2	70.7, CH	70.9, CH	70.9, CH	70.7, CH	70.7, CH
3	70.9, CH	71.1, CH	71.1, CH	70.9, CH	70.9, CH
4	72.4, CH	72.2, CH	72.3, CH	72.6, CH	72.5, CH
5	68.3, CH	68.7, CH	68.7, CH	68.3, CH	68.3, CH
6	16.6, CH ₃	18.3, CH ₃	18.3, CH ₃	16.6, CH ₃	16.6, CH ₃
7-0-Glc					
1	100.1, CH	100.3, CH			100.1, CH
2	73.3, CH	73.6, CH			73.3, CH
3	77.0, CH	76.9, CH			77.0, CH
4	69.9, CH	70.0, CH			69.9, CH
5	76.4, CH	77.7, CH			76.4, CH
6	61.1, CH ₂	61.2, CH ₂			61.1, CH ₂

Table 2. ¹³C-NMR assignments (100 MHz) for compounds 3–7, δ (ppm).

			•		
	3	4	5	6	7
	δ _H (J in Hz)	δ _H (J in Hz)	$\delta_{\rm H}$ (J in Hz)	δ _H (J in Hz)	δ _H (J in Hz)
2					
3					
4					
5					
6	6.44, d (2.07)	6.44, d (1.91)	6.19, d (1.97)	6.15, d (1.97)	6.46, d (2.15)
7					
8	6.75, d (2.09)	6.77, d (1.97)	6.42, d (1.97)	6.37, d (1.97)	6.75, d (2.14)
9					
10					
1′					
2'	8.09, m (1.93)	8.00, d (1.84)	7.98, d (1.97)	8.07, d (1.94)	8.09, m (8.95)
3'					6.90, m (8.90)
4′					
5'	6.91, m (8.44)	6.90, d (8.39)	6.88, d (8.44)	6.90, d (8.42)	6.90, m (8.90)
6′	7.57, m (1.91;	7.54, dd (1.82;	7.49, dd (2.03;	7.52, dd (1.99;	8.09, m (8.95)
OCH ₃	8.42) 3.83, s	8.48) 3.84, s	8.39) 3.83, s	8.43) 4.00, s	_
3-O-Gal	,-	,-			
	E 70 J (7 94)	E 47 J (772)		E E() J (7.92)	E EQ 1 (7.74)
1	5.79, d (7.84)	5.47, d (7.73)	5.45, d (7.67)	5.59, d (7.83)	5.59, d (7.74)
2 3	3.97, dd	3.60, dd	3.57, dd	3.96, dd	3.95, dd
4	3.78, dd 3.80, dd	3.57, dd 3.62, dd	3.57, dd 3.63, dd	3.76, dd 3.81, dd	3.70, dd 3.75, dd
5	3.74, dt	3.42, dt	3.41, dt	3.71, dt	3.64, dt
6	3.54/3.73dd	3.30/3.60 dd	3.31/3.61 dd	3.52/3.72 dd	3.46/3.69 dd
0 Rha (1→2)	5.54/ 5.75dd	5.507 5.00 dd	5.51/ 5.01 dd	5.527 5.72 dd	3.40/ 3.07 du
1	5.16, d (1.52)			5.16, d (1.53)	5.21, d (1.32)
2	4.00, dd			4.00, dd	4.00, dd
3	3.75, dd			3.76, dd	3.78, dd
4	3.33, pt			3.32, pt	3.33, pt
5	4.04, dq			4.03, dq	4.06, dq
6	0.89, d (6.25)			0.89, d (6.23)	0.98, d (6.23)
Rha (1→6)					
1	4.53, d (1.53)	4.41, d (1.53)	4.41, d (1.53)	4.55, d (1.55)	4.49, d (1.24)
2	3.54, dd	3.37, dd	3.38, dd	3.59, dd	3.48, dd
3	3.49, dd	3.28, dd	3.28, dd	3.51, dd	3.46, dd
4	3.25, pt	3.06, pt	3.07, pt	3.26, pt	3.25, pt
5	3.52, dq	3.36, dq	3.34 <i>,</i> dq	3.54, dq	3.49, dq
6	1.16, d (6.08)	1.04, d (6.18)	1.03, d (6.25)	1.17, d (6.22)	1.16, d (6.21)
7-0-Glc					
1	5.06, d (7.24)	5.05, d (7.89)			5.07, d (7.45)
2	3.48, dd	3.24, dd			3.48, dd
3	3.53, dd	3.27, dd			3.53, dd
4	3.39, dd	3.15, dd			3.38, dd
5	3.50, dt	3.43, dt			3.49, dt
6	3.69/3.92, dd	3.44/3.68, dd			3.69/3.91, dd

Table 3. ¹H-NMR assignments (600 MHz) of compounds 3–7 δ H (ppm), multiplicity (*J* in Hz).

ESI⁻ HRAM spectrum of compound **6** showed presence of signal of deprotonated molecule m/z [M – H]⁻ 769.21948 (calcd for C₃₄H₄₁O₂₀⁻ 769.219667 m/z; Δ = 0.00019). The MS/MS analysis (a minimal fragmentation using 20 eV) in a positive mode showed the presence of a protonated parental ion m/z [M + H]⁺ 787.2284 and fragments corresponding to loss of two rhamnose units and a hexose m/z 641.1708, 479.1180 and 317.0653. The molecular formula C₃₄H₄₂O₂₀ (Mr 770.68528) is in accordance with the NMR and HR-ESI-MS spectral data and ¹H and ¹³C chemical shifts for compound **6** are listed in Tables 2 and 3, respectively. Compound **6** was therefore identified as isorhamnetin-3-*O*-(2,6-di-*O*-α-rhamno-pyranosyl-β-galactopyranoside) [13,14].

The molecular formula $C_{39}H_{50}O_{24}$ (Mr 902.7999) is in accordance with the NMR and HR-ESI-MS spectral data, where the mass of deprotonated molecule was found at m/z [M – H][–] 901.2624 (calcd 901.2619; $\Delta = 0.0005$). The detailed MS analysis of fragments observed in spectrum showed signal

7 of 13

m/z [M + H]⁺ 903.2759, m/z [M + H – rhamnose]⁺ 757.2178, m/z [M + H – rhamnose – rhamnose]⁺ 611.1601, m/z [M + H – rhamnose – rhamnose – hexose]⁺ 449.1078, and aglycone m/z 287.0549, with pairs of fragments showing loos of water 18 m/z [M + H – rhamnose – H₂O]⁺ 741.2245, m/z [M + H – rhamnose – rhamnose – H₂O]⁺ 595.1660. The spectrum also showed fragments of sugars m/z 309.1186 (hexose-rhamnose). ¹H and ¹³C chemical shifts for compound 7 are listed in Tables 2 and 3 respectively. NMR data of compound 7 and mass spectral analysis results show a good accordance with the data already reported for alangiflavoside [15].

Based on the previously described analysis of known compounds, we tried to identify the other isolated compounds, which showed differences from those previously described in the literature. Some of the fragmentation MS/MS results used for the identification of compounds **1–4** are depicted in Figure 4.

Compound 1 was isolated as a yellow amorphous powder. The UV spectra analysis showed a typical flavonoid course, with the retention time typical for glycosylated substances. The HR-ESI-MS in a negative mode displayed a molecular ion m/z [M – H]⁻ 917.2572 (calc. for [M – H]⁻ 917.256841), confirming the molecular formula C₃₉H₅₀O₂₅ with calculated molecular mass 918.7993 Da. HRESIMS showed an adduct formation m/z [M + HCOO]⁻ 963.26269 (confirming the analyte), and a hexose (m/z162.0235) loss leading to *m/z* 755.21208 (possible representing cleavage of glucose from 7-O position or galactose from 3-O- position of expected flavonol skeleton). Furthermore, the MS in a positive mode confirmed this loss showing precursor ion m/z [M + H]⁺ 919.2705 with product ions formed either by subsequent loses of a pentose (m/z 787.2264), a rhamnose (m/z 625.1750), and a hexose (479.1170) unit, or loss of rhamnose (m/z 773.2136) followed by hexose (m/z 611.1613) moieties (see Figure 4). After lining up interpretation of NMR spectra (Table 1) with MS analysis, the structure of compound 1 was elucidated to be the isorhamnetin substituted by four sugars. One of them is a 6-deoxyhexose (a methyl group as doublet), two are hexoses (a glucose, a galactose), and one is a pentose. DEPT spectrum of compound 1 showed four methylene groups. Altogether, two of them belong to each hexose (the glucose and galactose), respectively, and since a 6-deoxyhesose (expected rhamnose) does not have a methylene moiety, the two left CH_2 moieties must belong to a pentose. Four anomeric protons are found in HMBC spectrum: δ_H 5.57 ppm, (1H, d, J = 7.84 Hz), δ_H 5.43 ppm (1H, d, J = 1.55 Hz), δ_H 4.50 ppm (1H, d, J = 1.56 Hz) and δ_H 5.06 ppm (1H, d, J = 7.53 Hz), corresponding to carbon atoms with δ_C 99.8 ppm, δ_C 109.2 ppm, δ_C 100.5 ppm and δ_C 100.1 ppm from the HSQC spectrum. After complete resonance assignments, analyses of coupling constants, intensities, interpretation of cross-peaks in the COSY spectrum, and ¹³C-NMR chemical shift values, one hexose moiety was identified as a β -glucosyl unit, the second as a β -galactosyl moiety, the 6-deoxysugar was identified to be the α -rhamnose, and the pentose was recognized as the β -apiose, which contains two of the above-mentioned methylene groups. ¹³C values of C-6 indicate that β -glucose residue is not connected to other sugar unit (δ_C 61.1 ppm), while the β -galactose moiety is connected at C-6 position (δ_C 66.0 ppm—shifted to a higher field). HMBC correlations allowed us to elucidate the precise structure of the sugar chains and the positions of their attachment to the aglycone. The anomeric proton ($\delta_{\rm H}$ 5.06 ppm) of the β -glucose moiety showed a three-bond correlation to C-7 ($\delta_{\rm C}$ 163.0 ppm) of the aglycone, while a HMBC correlation between the anomeric proton of the galactose ($\delta_{\rm H}$ 5.57 ppm) and the carbon at $\delta_{\rm C}$ 133.6 ppm indicated that the galactosyl unit is connected at C-3 towards the aglycone. The anomeric proton of α -rhamnose correlated to C-6 in the β -galactose (δ_C 66.0 ppm). The anomeric atom of the last pentose sugar (apiose) $\delta_{\rm H}$ 5.43 ppm bonded to $\delta_{\rm C}$ 109.2 ppm showed a HMBC correlation with C-2 of the β -galactose molecule (δ_C 75.1 ppm). The remaining two methylene groups (δ_C 74.2 ppm and δ_C 65.1 ppm) were recognized as carbons C-4 and C-5 in the apiosyl moiety. Hence, according to all interpreted spectra, we identified compound 1 as a new natural product named isorhamnetin-3-O-[β -D-apiofuranosyl-(1 \rightarrow 2)-[α -L-rhamnopyranosyl-(1 \rightarrow 6)]- β -D-galactopyranosyl]-7-O- β -D-glucopyranoside.

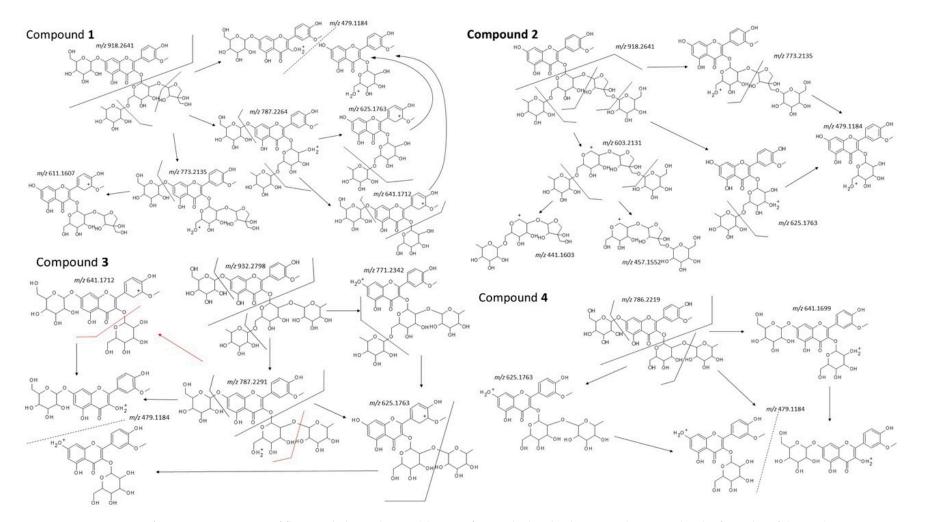


Figure 4. A fragmentation pattern of flavonoid glycosides 1–4 (shown *m/z* are calculated values according to molecular formulas of the ions).

Compound 2 was obtained as a yellow amorphous powder. The spectral analysis showed data very similar to those observed for compound 1 (Table 1), and the only difference found was the presence of a β -glucosyl residue connected to the sugar chain, attached to the 3-O position (particularly C-5 of apiosyl residue); a connection to the 7-O position of the aglycone was not found. Further interpretation of HRESIMS supported this suggestion by observing a lacking of the fragment of a hexose loss (m/z 162.0235), which can be observed in spectra of all compounds possessing glucose attached at 7-O position (compounds 1, 3, 4 and 7). The HR-ESI-MS in negative mode displayed a molecular ion m/z [M - H]⁻ 917.25696 (calculated for m/z [M - H]⁻ 917.256841), confirming molecular formula C₃₉H₅₀O₂₅ with calculated molecular mass 918.7993 Da. Adduct formation [M + HCOO]⁻ (m/z 953.2336) and [M + Cl]⁻ (m/z 963.2621) was observed in a negative ESI mode. Furthermore, MS in a positive mode confirmed this by $[M + H]^+$ with m/z 919.2709, and a series of corresponding losses of four sugar units at a minimal fragmentation of 20 eV: m/z 773.2136 as loss of rhamnosyl unit, m/z 625.1759 loss of rhamnose and hexose, and 479.1185 corresponding to isorhamnetin hexoside after loss of three sugar units of rhamnose, hexose and pentose. In addition, the aglycone signal m/z 317.0655 and the fragment ion m/z 603.2144, interpreted as the chain of all sugar moieties with corresponding fragments m/z 441.1610 and m/z 457.1552 after losses of terminal hexose and rhamnose, respectively, were detected. According to differences in chemical shifts in NMR spectral data (compared with 1), the glucosyl moiety changed its bonding position from 7-O position of the aglycone to C-5 position of the β -apiosyl residue (see Table 3). The compound was therefore identified as isorhamnetin-3-O-[β -D-glucopyranosyl-(1 \rightarrow 5)- β -D-apiofuranosyl]-(1 \rightarrow 2)-robinobioside or $[\alpha$ -L-rhamnopyranosyl- $(1 \rightarrow 6)]$ - β -D-galactopyranosyl], a newly described flavonoid glycoside.

Compound 3 was isolated as a yellow amorphous powder. Rt of compound 3 was 4.61 min, again slightly different from other isolates. The HR-ESI-MS in a negative mode displayed a molecular ion m/z [M – H]⁻ 931.27352 (calcd [M – H]⁻ 931.272491), confirming molecular formula C₄₀H₅₂O₂₅ with calculated molecular mass 932.82588 Da. HRESIMS showed additional adduct formation at m/z $[M + HCOO]^{-}$ 977.27862, and a hexose loss (*m*/*z* 162.05387) leading to *m*/*z* 769.21965 (representing supposed cleavage of glucose from 7-O position). In the HRAM ESI positive mode, $[M + H]^+$ ion with m/z 933.2810 (calcd [M + H]⁺ 933.2876) was obtained. Fragmentation with collision energy of 20 eV gave ions with m/z 787.2229 and m/z 771.2324 formed by loss of hexose and rhamnose, respectively. Further subsequent loses of the hexosyl and rhamnosyl moieties resulted fragments with m/z 625.1679 (aglycone-rhamnosyl-hexoside) and m/z 641.1652 (aglycone-hexosyl-hexoside). Finally, ions of m/z479.1082 (hexosylated aglycone) and aglycone m/z 317.0526 (supposed isorhamnetin) were found. Together with an interpretation of NMR spectra (Tables 2 and 3), the structure was predicted to be composed of isorhamentin and four sugar moieties. Two of them were recognized as 6-deoxyhexoses (two methyl groups in the form of the doublet), and the other two were hexoses (possibly a glucose, or a galactose). DEPT spectrum showed their two methylene groups (δ_C 61.1 ppm and 65.9 ppm) that belonged to the galactose and the glucose, respectively. We observed four anomeric protons in HSQC spectrum: $\delta_{\rm H}$ 5.79 ppm, (1H, d, J = 7.84 Hz), $\delta_{\rm H}$ 5.16 ppm (1H, d, J = 1.52 Hz), $\delta_{\rm H}$ 4.53 ppm (1H, d, J = 1.53 Hz) and $\delta_{\rm H}$ 5.06 ppm (1H, d, J = 7.24 Hz), corresponding to carbon atoms $\delta_{\rm C}$ 99.3 ppm, 101.4 ppm, 100.6 ppm and 100.1 ppm, respectively. After complete resonance assignments and analyses of coupling constants, intensities of cross-peaks in the COSY spectrum, and ¹³C-NMR chemical shift values, one hexose moiety was identified as a β -glucosyl unit, the other as a β -galactosyl moiety, and the 6-deoxy sugars were found to be α -rhamnosyl moieties. ¹³C values indicated that β -glucosyl residue is free at C-6 (δ_C 61.1 ppm), while the β -galactose is bonded to C-6 (δ_C 65.9 ppm—shifted to a higher field). According to HMBC correlations, structure of the side chains and their attachment to the aglycone were established. Anomeric proton (δ_H 5.06 ppm) of the glucose moiety showed a three-bond correlation to C-7 (δ_C 163.0 ppm) of the aglycone, while an HMBC correlation between the anomeric proton of the β -galactosyl moiety (δ_H 5.79 ppm) and the carbon at δ_C 133.3 ppm indicated that the β -galactosyl unit is bonded at C-3 toward the aglycone. The anomeric proton of one of the α -rhamnose residues (δ_H 5.16 ppm) was correlated to position C-2 of the β -galactose (δ_C 76.4 ppm),

while the anomeric proton of the other α -rhamnose showed correlation to C-6 of the β -galactose ($\delta_{\rm C}$ 65.9 ppm). Methyl residues of α -rhamnopyranosyl residues were located in the low-field region of ¹H spectrum at $\delta_{\rm H}$ 0.89 ppm, (3H, d, J = 6.25 Hz) for the α -rhamnosyl residue attached to C-2 of the β -galactose and at $\delta_{\rm H}$ 1.16 ppm (3H, d, J = 6.08 Hz) for the α -rhamnosyl residue attached to C-6 of the β -galactose, respectively. Hence, we identified compound **3** as the new natural product isorhamnetin-3-O-(2,6-di-O- α -rhamnopyranosyl- β -D-galactopyranoside)-7-O- β -D-glucopyranosyl. The compound is similar to compound **7**; the difference lies in the absence of β -D-glucopyranosyl moiety at 7-O position in compound **7**.

Compound 4 was isolated as a yellow amorphous powder. Chromatographic analysis showed Rt 5.40 min. The HR-ESI-MS in negative mode displayed a molecular ion m/z [M – H]⁻ 785.21509 (calcd for [M – H]⁻ 785.21403), confirming molecular formula C₃₄H₄₂O₂₁ with calculated molecular mass 786.68468 Da. HRESIMS in negative mode showed additional adduct formation m/z [M + HCOO]⁻ 831.22089, and a hexose (m/z 162.05404) loss leading to m/z 623.16174 (representing cleavage of glucose from 7-*O* position or a galactose from 3-*O*). The moderate fragmentation in positive ESI (20 eV) showed the presence of m/z 787.2284 for parental ion, and then the corresponding fragment with a cleavage of deoxyhexose m/z 641.1714, hexose m/z 625.1753, two sugar (rhamnosyl and hexosyl) units m/z 479.1178 and an aglycone at m/z 317.0658, possible isorhamnetin. After the interpretation of NMR spectra (Tables 2 and 3), we confirmed the presence of isorhamnetin and three sugar moieties. Compound 4 possesses similar structure as compound 3, with absence of the α -rhamnopyranosyl moiety connected to C-2 of β -galactosyl residue, and we identified compound 4 as isorhamnetin-3-*O*-robinobioside-7-*O*-glucoside, a new flavonoid glycoside.

3. Discussion

As described above, we isolated 7 flavonoid compounds from *A. aitosensis* extract. Their structures were elucidated by $1D(^{1}H, ^{13}C)$ and 2D NMR experiments (COSY, HSQC, HMBC, NOESY and TOCSY) and confirmed by HR-ESI-MS. We report the structures of six isorhamnetin and one kaempferol glycosides (1–7), including three new tetra- (1–3), one new tri-glycoside (4) and three already known compounds with two (5), three (6) and four (7) sugar units, respectively.

Genus *Astragalus* is one of the largest genera of Fabaceae family. As mentioned, bioactivity of *Astragalus* plants are connected with a presence of flavonoids, saponins and polysaccharides. The use of *Astragalus* spp. is mainly connected with immunomodulation, antibacterial and antiviral activity, and hepatoprotection [9,10]. The reviews of Gorai et al. [16], Bratkov et al. [10], and Li et al. [17] show an overview of *Astragalus* genera and flavonoids isolated, showing the presence of flavones, flavonols, flavanoes, flavan-4-ols, isoflavones, isoflavans, pterocarpans and others in 60 different *Astragalus* species. Their reviews also include isorhamnetin and kaempferol glycosides. Another comprehensive review of Bulgarian *Astragalus* species, published in 2016 [18], similarly describes the presence of several kaempferol and isorhamnetin glycosides, including alangiflavoside from *A. monspessulanus* sp. *monspessulanus* [18,19].

In recent years, Bulgarian researchers isolated and reported new tri- and tetraglycosides of flavonols, including some new compounds from the rarely-met group of flavo-alkaloids [10,11]. Many species of *Astragalus* possess in nature the widely-distributed aglycones—kaempferol, quercetin and methylquercetines—in their free and glycosidic forms [10]. *A. aitosensis* has previously shown presence of rutin, quercetin-3-*O*-β-D-glucoside and astragalin [7]. As visible from comparison of glycosides isolated from *A. aitosensis* with the literature, similar compounds—glycosides—were obtained for example from A. *monspessulanus* ssp. *monspessulanus*, *A. cicer* and A. centralpinus [18–20]. This may, from chemotaxonomic point of view, confirm their close relationships.

4. Materials and Methods

4.1. General Experimental Procedures

Optical rotations were measured on a JASCO P-2000 spectropolarimeter (Easton, MD, USA) at 20 °C in MeOH with Spectramanager software.

HPLC used configuration of analytical system by Agilent 1100 Series (Degasser G1322A, Quaternary Pump G1311A, Autosampler ALS G1313A, Column Compartment G1316, DAD G1315B, Loop 20 μ L, UV spectrum 200–900 nm) with column Kinetex[®] PFP 100 A, 250 mm × 4.6 mm I.D., 5 μ m (Phenomenex, CA, USA), and flow rate of 1 mL/min. Semi-preparative HPLC was carried out using Dionex UltiMate 3000 system (Pump Dionex UltiMate 3000 UPLC+ Focused, Dionex UltiMate 3000 RS Variable Wavelength Detector, fraction collector Dionex UltiMate 3000 with 6 positions, LCO 101 ECOM column oven, constant temperature 40 °C, autosampler Dionex UltiMate 3000, loop 100 μ L), column Ascentis[®] RP-AMIDE, 250 mm × 10 mm, 5 μ m (Supelco, PA, USA), and flow rate of 5 mL/min. TLC was carried out on precoated silica gel plates (Supelco Kieselgel G, F254, 60, Merck, Darmstadt, Germany) with the solvent systems EtOAc:MeOH:H₂O (100:13.5:10, v/v/v). Spots were visualized under UV light (365 nm) after spraying with NTS/PEG reagent. Column chromatography (CC) was performed using Diaion HP-20 (Supelco, PA, USA), \emptyset = 80 mm, height 70 cm ~ 700 g and Silica gel (40–63 μ m, Sigma-Aldrich[®], St. Louis, MO, USA) \emptyset = 35 mm, height 60 cm.

4.2. Plant Material

The aerial parts of *A. aitosensis* (Ivan.) Podl (Fabaceae) (syn. *Astracantha aitosensis* (Ivan.) Podl.) was collected and identified by Hristo Vasilev in June 2015 in the suburbs of town Aytos, Bulgaria (coordinates Google maps: 42.702191 N, 27.266976 E, UTM: NH22), voucher specimen has been deposited in Herbarium of the Institute of Biodiversity and Ecosystem Research, Bulgarian Academy of Sciences with Ref No. SOM001362.

4.3. Extraction and Isolation

Dried aerial parts (3600 g) of A. aitosensis were extracted under reflux with 80% MeOH (20 \times 1.25 L, 40 min each) at 70 °C. The total methanol extract was evaporated to dryness (128 g, 3.5%), and it was re-suspended in H_2O (800 mL) to remove nonpolar compounds by liquid–liquid extraction with chloroform (200 mL \times 5), giving 18 g of CHCl₃ fraction and 110 g of MeOH soluble material. This defatted methanol portion was subjected to Diaion HP-20 column, and eluted with gradient system of H_2O and MeOH (100:0 to 0:100, v/v), giving 10 combined fractions assigned A–J. Fraction B, C, E, and F were further purified via open column chromatography (Silica gel, eluted with CHCl₃:MeOH:H₂O 6:4:0.2, v/v/v). Fraction B was re-chromatographed on silica gel (CHCl₃:MeOH:H₂O 6:4:0.2, v/v/v, as mobile phase,) and resulted in 6 combined fractions (B_1-B_6) . Fraction B_4 was further purified using a semipreparative HPLC (gradient of acetonitrile in 0.2% HCOOH from 12 to 20% in 25th minute), and this purification gave compound 4 (27 mg). Fraction B_5 , after semipreparative HPLC (gradient of acetonitrile in 0.2% HCOOH from 15% to 18% in 20th minute), gave compound 1 (11 mg). Fraction B₆ was purified on semi-preparative HPLC (gradient of acetonitrile in 0.2% HCOOH from 15% to 20% in 20th minute), gave compounds 7 (20 mg) and 3 (45 mg), respectively. Fraction C was re-chromatographed on silica gel (mobile phase 6:4:0.2 CHCl₃:MeOH:H₂O, v/v/v) and this separation resulted in 45 fractions. Pure compound 2 (24 mg) precipitated from fraction C_{34} . Three hundred milligrams of fraction E were purified with semipreparative HPLC (gradient of acetonitrile in 0.2% HCOOH from 19 to 22% in 18th minute), and gave compound 6 (56 mg). Fraction F was re-chromatographed on silica gel (mobile phase 6:4:0.2 CHCl₃:MeOH:H₂O, v/v/v), which resulted in 35 fractions, 100 mL each. Pure compound 5 (40 mg) yielded from fraction F_{10} after re-crystallization in MeOH.

NMR spectra were recorded on a NMR Agilent DD2 600 MHz (compounds 1-7) and a NMR Agilent VNMRS 500 MHz (compound 2)—equipped with four and three channels, respectively, and structure elucidation was carried out using modern 1D and 2D pulse sequences, following 1D and 2D experiments were carried out: ¹H, ¹³C, COSY, HMBC, HSQC, TOCSY, and NOESY. The spectra were processed with MestReNova version 12.0.0 (Mestrelab Research, Santiago de Compostela, Spain). Mass spectra were recorded using a Thermo Scientific Q Exactive Plus quadrupole—Orbitrap mass spectrometer coupled with a UPLC Dionex Ultimate 3000 RSLC system equipped with an RP-18 Kinetex column (2.10 mm \times 100 mm, 2.6 μ m, Phenomenex Corporation, Torrence, CA, USA). Full-scan data were recorded in negative ESI mode from m/z 100 to 1500 at a resolution of 70,000 (at m/z 200). Full-scan dd-MS² (Top 5) was performed at a resolution of 17 500 (at m/z 200), AGC target 1^{e5} with maximum IT 30 ms. For HR-MS in positive mode, Q-TOF mass spectrometer with ultra-high resolution and high mass accuracy (HRAM) Impact II (Bruker Daltonik, Bremen, Germany) were used. UHPLC Dionex UltiMate 3000 (Thermo Fisher Scientific, Waltham, MA, USA) was used for LC, with mobile phases: 0.1% formic acid (A) and MeOH (B), flow rate: 0.2 mL/min, gradient elution: 0 min 5% of B, 3.6 min 10% of B, 10th min 100% of B. Column block temperature was 35 $^{\circ}$ C, and injection volume 5 μ L. Kinetex P5, 100A, 1.7um, 100 \times 2.1 mm (Phenomenex, Torrance, CA, USA) was used as column. Solution of natrium formate clusters was used as calibration mixture for accurate mass calibration, with MS source settings: end plate offset: 500 V, capillary 4500 V, nebulizer pressure 0.3 Bar, dry gas: 4.0 L/min, dry temperature: 250 °C. MS/MS spectra were collected at three collision energy levels of 20, 40, and 60 eV per each peak. Data acquisition were carried out by Control 4.0 and HyStar 3.2 software and the results were processed using Compass DataAnalysis 4.3 (all SW of Bruker Daltonik, Bremen, Germany). Chromatography was controlled by Chromeleon Xpress link (Thermo Fisher Scientific, Waltham, MA, USA). For fragmentation analysis Mass Frontier 7.0.5.9 SR3 (High Chem Ltd., Bratislava, Slovakia) and Thermo Excalibur 3.0.63 (Thermo Fisher Scientific, Waltham, MA, USA) software was used.

Supplementary Materials: The following are available online. Figures S1–S77: NMR spectra of compounds 1–7, HRMS of compounds 1–7, and HPLC chromatograms of compounds 1–7.

Author Contributions: Conceptualization, J.H.; Data curation, H.V., S.R., D.J. and O.V.; Formal analysis, D.J. and O.V.; Funding acquisition, K.Š. and P.M.; Investigation, H.V., P.M. and J.H.; Methodology, H.V. and K.Š; and Writing—original draft, H.V., K.Š and P.M.

Funding: The financial support of GACR No. 16-07193S project to K.Š.; MŠMT No. LD14079 to P.M.; CIGA project No. 20,172,031 to P.M., J.H., and O.V; projects of Erasmus Program; and project No. 30016/672 DBU to H.V. is gratefully acknowledged.

Acknowledgments: The authors acknowledge the technical assistance of Ing. Hanka Vítková, Klaus Bergander, and Paraskev Nedialkov.

Conflicts of Interest: The authors declare no conflict of interest. The funders had no role in the design of the study; in the collection, analyses, or interpretation of data; in the writing of the manuscript, or in the decision to publish the results.

References

- 1. Pistelli, L.F. Secondary metabolites of genus *Astragalus*: Structure and biological activity. *Stud. Nat. Prod. Chem.* **2002**, *27*, 443–545. [CrossRef]
- 2. Frodin, D.G. History and concepts of big plant genera. *Taxon* 2004, *53*, 753–776. [CrossRef]
- 3. Tutin, T.G.; Heywood, V.H.; Burges, N.A.; Moore, D.M.; Valentine, D.H.; Walters, S.M. *Flora Europaea*. 2; Cambridge University Press: Cambridge, UK, 1972.
- 4. Assyov, B.; Petrova, A.; Dimitrov, D.; Vassilev, R. *Conspectus of the Bulgarian vascular flora*, 4rd ed.; Assyov, B., Petrova, A., Eds.; Bulgarian Biodiversity Foundation: Sofia, Bulgaria, 2012; pp. 86–89.
- 5. Andreev, N.; Anchev, M.; Kozucharov, S.; Markova, M.; Peev, D.; Petrova, A. *Field Guide of the Bulgarian Vascular Plants*; Nauka i Izkustvo: Sofia, Bulgria, 1992; pp. 392–397.

- Ionkova, I. Isolation and HPLC-TLC Analysis of the Major Flavonoids from *Astragalus aitosensis*. *Planta Med.* 1990, 56, 581. [CrossRef]
- 7. Ionkova, I. Farmacevtichno Znachimi Biologichno Aktivni Veshtestva ot Iztochnici s Optimiziran Fitohimichen Potencial (Pharmaceutically Significant Biologically Active Compounds from Sources with Optimized Phytochemical Potential). DSci Dissertation, Medical University, Sofia, Bulgaria, 2008.
- 8. Apostolova, I. Red Data Book of the Republic of Bulgaria, Digital ed. Astracantha arnacantha. Available online: www.e-ecodb.bas.bg/rdb/en/vol1/Astarnac.html (accessed on 4 March 2019).
- 9. Ionkova, I.; Shkondrov, A.; Krasteva, I.; Ionkov, T. Recent progress in phytochemistry, pharmacology and biotechnology of *Astragalus* saponins. *Phytochem. Rev.* **2014**, *13*, 343–374. [CrossRef]
- 10. Bratkov, V.; Shkondrov, A.; Zdraveva, P.; Krasteva, I. Flavonoids from the genus Astragalus: Phytochemistry and biological activity. *Pharmacogn. Rev.* **2016**, *10*, 11–32. [PubMed]
- 11. Krasteva, I.; Bratkov, V.; Bucar, F.; Kunert, O.; Kollroser, M.; Kondeva-Burdina, M.; Ionkova, I. Flavoalkaloids and flavonoids from *Astragalus monspessulanus*. J. Nat. Prod. **2015**, 78, 2565–2571. [CrossRef] [PubMed]
- 12. Buschi, C.A.; Pomilio, A.B. Isorhamnetin 3-O-robinobioside from *Gomphrena martiana*. J. Nat. Prod. **1982**, 45, 557–559. [CrossRef]
- 13. Yasukawa, K.E.N.; Sekine, H. Two flavonol glycosides from *Lysimachia fortunei*. *Phytochemistry* **1989**, *28*, 2215–2216. [CrossRef]
- 14. Yasukawa, K.; Takido, M. A flavonol glycoside from *Lysimachia mauritiana*. *Phytochemistry* **1987**, *26*, 1224–1226. [CrossRef]
- 15. Kijima, H.; Ide, T.; Otsuka, H.; Takeda, Y. Alangiflavoside, a new flavonol glycoside from the leaves of *Alangium premnifolium. J. Nat. Prod.* **1995**, *58*, 1753–1755. [CrossRef]
- Zhang, Y.; Wang, D.; Yang, L.; Zhou, D.; Zhang, J. Purification and characterization of flavonoids from the leaves of *Zanthoxylum bungeanum* and correlation between their structure and antioxidant activity. *PLoS ONE* 2014, 9, e105725/1–e105725/11. [CrossRef] [PubMed]
- 17. Gorai, D.; Jash, S.K.; Roy, R. Flavonoids from Astragalus genus. Int. J. Pharm. Sci. Res. 2016, 7, 2732–2747.
- 18. Krasteva, I.; Shkondrov, A.; Ionkova, I.; Zdraveva, P. Advances in phytochemistry, pharmacology and biotechnology of Bulgarian *Astragalus* species. *Phytochem. Rev.* **2016**, *15*, 567–590. [CrossRef]
- 19. Li, X.; Qu, L.; Dong, Y.; Han, L.; Liu, E.; Fang, S.; Zhang, Y.; Wang, T. A review of recent research progress on the *Astragalus* genus. *Molecules* **2014**, *19*, 18850–18880. [CrossRef] [PubMed]
- 20. Alaniya, M.D.; Aneli, D.N.; Patudin, A.V.; Komelin, R.V. Flavonoid glycosides of *Astragalus cicer*. *Chem. Nat. Comp.* **1983**, *19*, 500–501. [CrossRef]

Sample Availability: Samples of the compounds are not available from the authors.



© 2019 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (http://creativecommons.org/licenses/by/4.0/).