University of Texas Rio Grande Valley

ScholarWorks @ UTRGV

Physics and Astronomy Faculty Publications and Presentations

College of Sciences

2-2-2021

Scalable, cost-efficient synthesis and properties optimization of magnetoelectric cobalt ferrite/barium titanate composites

Farnaz Safi Samghabadi

Long Chang

Mohammad Khodadadi

Karen S. Martirosyan The University of Texas Rio Grande Valley, karen.martirosyan@utrgv.edu

Dmitri Litvinov University of Houston

Follow this and additional works at: https://scholarworks.utrgv.edu/pa_fac

Part of the Astrophysics and Astronomy Commons, and the Physics Commons

Recommended Citation

Safi Samghabadi, Farnaz, Long Chang, Mohammad Khodadadi, Karen S Martirosyan, and Dmitri Litvinov. 2021. "Scalable, Cost-Efficient Synthesis and Properties Optimization of Magnetoelectric Cobalt Ferrite/ Barium Titanate Composites." APL Materials 9 (2): 021104. https://doi.org/10.1063/5.0036518.

This Article is brought to you for free and open access by the College of Sciences at ScholarWorks @ UTRGV. It has been accepted for inclusion in Physics and Astronomy Faculty Publications and Presentations by an authorized administrator of ScholarWorks @ UTRGV. For more information, please contact justin.white@utrgv.edu, william.flores01@utrgv.edu.

Scalable, cost-efficient synthesis and properties optimization of magnetoelectric cobalt ferrite/barium titanate composites

Cite as: APL Mater. **9**, 021104 (2021); https://doi.org/10.1063/5.0036518 Submitted: 05 November 2020 . Accepted: 02 January 2021 . Published Online: 02 February 2021

Farnaz Safi Samghabadi, ¹⁰ Long Chang, ¹⁰ Mohammad Khodadadi, ¹⁰ Karen S Martirosyan, and ¹⁰ Dmitri Litvinov

COLLECTIONS

Paper published as part of the special topic on Magnetoelectric Materials, Phenomena, and Devices





ARTICLES YOU MAY BE INTERESTED IN

Strain-gradient effects in nanoscale-engineered magnetoelectric materials APL Materials 9, 020903 (2021); https://doi.org/10.1063/5.0037421

Examination of possible high-pressure candidates of SnTiO₃: The search for novel ferroelectric materials APL Materials **9**, 021103 (2021); https://doi.org/10.1063/5.0029968

Ferroelectric field effect transistors: Progress and perspective APL Materials 9, 021102 (2021); https://doi.org/10.1063/5.0035515





APL Mater. 9, 021104 (2021); https://doi.org/10.1063/5.0036518 © 2021 Author(s).

ſŢ

Export Citation

Scalable, cost-efficient synthesis and properties optimization of magnetoelectric cobalt ferrite/barium titanate composites

Cite as: APL Mater. 9, 021104 (2021); doi: 10.1063/5.0036518 Submitted: 5 November 2020 • Accepted: 2 January 2021 • Published Online: 2 February 2021

Farnaz Safi Samghabadi,¹ D Long Chang,^{2,3} Mohammad Khodadadi,^{1,2} Karen S Martirosyan,⁴ Rand Dmitri Litvinov^{1,2,3,3}

AFFILIATIONS

¹Materials Science & Engineering, University of Houston, Houston, Texas 77204, USA

²Department of Electrical & Computer Engineering, University of Houston, Houston, Texas 77204, USA

³Center for Integrated Bio & Nano Systems, University of Houston, Houston, Texas 77204, USA

⁴Department of Physics and Astronomy, University of Texas Rio Grande Valley, Brownsville, Texas 78520, USA

Note: This paper is part of the Special Topic on Magnetoelectric Materials, Phenomena, and Devices. ^{a)}Author to whom correspondence should be addressed: litvinov@uh.edu

ABSTRACT

Cobalt ferrite (CoFe₂O₄)/barium titanate (BaTiO₃) particulate composites exhibiting high magnetoelectric coefficients were synthesized from low-cost commercial precursors using mechanical ball milling followed by high-temperature annealing. CoFe₂O₄ (20 nm–50 nm) and either cubic or tetragonal BaTiO₃ nanoparticle powders were used for the synthesis. It was found that utilizing a 50 nm cubic BaTiO₃ powder as a precursor results in a composite with a magnetoelectric coupling coefficient value as high as 4.3 mV/Oe cm, which is comparable to those of chemically synthesized core–shell CoFe₂O₄–BaTiO₃ nanoparticles. The microstructure of these composites is dramatically different from the composite synthesized using 200 nm tetragonal BaTiO₃ powder. CoFe₂O₄ grains in the composite prepared using cubic BaTiO₃ powder are larger (by at least an order of magnitude) and significantly better electrically insulated from each other by the surrounding BaTiO₃ matrix, which results in a high electrical resistivity material. It is hypothesized that mechanical coupling between larger CoFe₂O₄ grains well embedded in a BaTiO₃ matrix in combination with high electrical resistivity of the material enhances the observed magnetoelectric effect.

© 2021 Author(s). All article content, except where otherwise noted, is licensed under a Creative Commons Attribution (CC BY) license (http://creativecommons.org/licenses/by/4.0/). https://doi.org/10.1063/5.0036518

INTRODUCTION

Magnetoelectric (ME) effect^{1–5} has a number of promising applications in sensors, energy harvesting, magnetoelectric random access memory, antennas, drug delivery, etc.^{6,7} In ME materials, the electric polarization can be controlled by varying the material's magnetization state, and conversely, varying the electric polarization affects the material's magnetization state. The ME effect has been observed in a few single-phase *multiferroic* materials; however, the effect is relatively weak at room temperature, which hampers useful applications. More robust ME behavior has been achieved in composites that combine mechanically coupled magnetostrictive and piezoelectric materials arranged in a matrix.^{6,8} The ME effect in such composites is due to the induced stresses within the tion of external magnetic or electric fields, which transfer through the interface to the piezoelectric or magnetostrictive phases, respectively.⁹ These composites can be produced in versatile connectivity configurations/matrices with a wide choice of materials, volume fractions, and microstructures^{8,10} and can exhibit several orders of magnitude stronger ME effect than single-phase ME materials. Composite ME materials have been synthesized using a variety of techniques including sol–gel electrospinning of nanofibers,⁹ polyol mediated process of composite ceramics,¹⁰ molten-salt synthesis route for bulk composites,¹¹ core–shell structures,^{12,13} wet ballmilling,¹⁴ one-pot process,¹⁵ carbon combustion synthesis,¹⁶ and feather-like nanostructures.¹⁷ The ME coefficient for these composites ranges from a few μ V/Oe cm to several mV/Oe cm.⁹ Core–shell

magnetostrictive or piezoelectric phases controlled by the applica-

nanostructures, where the magnetostrictive core is fully enclosed by a piezoelectric shell, typically exhibit a higher value of ME coefficients. However, relatively complex chemical synthesis techniques along with the relatively low material yields limit their applications. Other synthesis approaches often suffer from poor interfaces between the two phases and/or low resistivity of magnetostrictive components, which hinders effective electrical poling.

In this study, the synthesis and characterization of CoFe₂O₄-BaTiO₃ particulate composites using dry ball-milling followed by annealing as a simple, low-cost, and highly reproducible powder processing method are presented. Cobalt ferrite, CoFe₂O₄, is a magnetostrictive material with magnetostriction coefficient values as high as 100 ppm-200 ppm at saturation,18-20 and the tetragonal phase of barium titanate, BaTiO₃, is piezoelectric with reported d₃₃ piezoelectric coefficient values in the 190 pC/N-260 pC/N range at room temperature.²¹⁻²⁵ The overarching goal of this work is to develop a scalable synthesis of an ME composite utilizing readily available commercial CoFe₂O₄ (CFO) and BaTiO₃ (BTO) precursor nanopowders. While there are several published reports on CoFe₂O₄-BaTiO₃ composites prepared by wet mechanical ball-milling, these reports have only partially explored the relationships between the process parameters and the properties of the composite.^{14,26,27} The structural, magnetic, and ME properties of CoFe₂O₄-BaTiO₃ composites were investigated as functions of preparation conditions and the types of BaTiO₃ precursor nanoparticles used, namely, 50 nm cubic BaTiO₃ powder and 200 nm tetragonal BaTiO3 powder. The ME coefficient in the optimized CoFe₂O₄-BaTiO₃ composite, where individual CoFe₂O₄ grains are fully enclosed by the BaTiO3 matrix, was found to be comparable with the numbers reported for the composites based on core-shell structures.28,29

$\mbox{CoFe}_2\mbox{O}_4\mbox{-BaTiO}_3$ composite synthesis and sample conditioning

A 99.9% purity cobalt ferrite powder with 20 nm–50 nm particle sizes and 99.9% purity cubic and tetragonal barium titanate powders with 50 nm and 200 nm average particle sizes, respectively, were used in this study (see Fig. 1).³⁰ The composite mixture [x-CoFe₂O₄–(1 – x)-BaTiO₃, where x is the weight fraction] was prepared using mechanical ball milling. The mixture of CoFe₂O₄ and BaTiO₃ powders was ball-milled using a high-speed vibrating milling machine (MTI Corporation MSK-SFM-3) at 288 rpm for 5 h in a dry medium. Ball milling was performed using a nylon jar filled with zirconia balls with a ball to powder mass ratio of 10:1. The resulting mixture was pressed into pellets with a diameter of ~12 mm and a thickness of ~1 mm using a hydraulic press at a pressure of 120 bars. Next, the pellets were sintered in air for 4 h at 1200 °C in a tube furnace (MTI Corporation OTF-1200X-S).

Tetragonal BaTiO₃ is piezoelectric, but cubic BaTiO₃ does not exhibit ferroelectric behavior due to its centrosymmetric crystal structure. However, cubic BaTiO₃ undergoes a structural transformation as it reaches certain critical temperatures. The most typical conversion from paraelectric cubic to ferroelectric tetragonal crystal structure occurs through cooling through its Curie point at 120 °C. This structural transformation is due to a slight displacement of Ti⁴⁺ cations with respect to the anion center along the crystallographic c axis.^{21–23}

The annealed pellets were electrically poled in the direction perpendicular to the pellet surface in a heated silicone oil bath at 150 $^{\circ}$ C, which is above the BaTiO₃ Curie temperature for 15 min. The poling voltage was maintained after the pellet was removed from the bath for an additional 15 min as the pellet cools down to room temperature. As discussed later, the magnitude of the poling voltage affects the measured ME properties.

Materials characterization

The magnetoelectric behavior was evaluated using a custombuilt ME characterization system similar to the one described in the literature.^{31–33} The opposing surfaces of the samples were coated with a thin layer of conductive silver paste (MG chemicals 842-AR silver print) as the electrical contacts for magnetoelectric measurement. The ac ME voltage across the pellet induced by the ac magnetic field (1 Oe at 1 KHz) superimposed over a dc magnetic field is recorded as a function of dc magnetic field, which can be swept between -7 kOe and 7 kOe. Stimulating the sample with an ac magnetic field superimposed to the dc magnetic field eliminates the contribution of charges accumulated in the grain boundaries and defects in the material during the poling process into the ME signal.^{33–35} Piezo-force microscopy (PFM) measurements were performed utilizing an MFP-3D Origin+ (Asylum Research-Oxford Instruments) atomic force microscope in the Dual AC Resonance Tracking (DART) mode using a silicon tip coated with Ti/Ir (5/20). The microstructure was examined utilizing an FEI Dual Beam 235 Focused Ion Beam instrument. The magnetostriction



FIG. 1. SEM images of (a) 20 nm–50 nm CoFe₂O₄, (b) 50 nm cubic BaTiO₃, and (c) 200 nm tetragonal BaTiO₃.

measurement was performed using a linear strain gauge (Micro-Measurements C2A-06-062LW-350) mounted on the surface of the pellets using a commercial bonding kit (Micro-Measurements MMF006678/M-BOND-200), and the data were acquired through a DAQ device (Micro-Measurements MM01-350) with a built-in Wheatstone quarter-bridge circuit. The magnetic properties of the samples were characterized using a LakeShore (model 735) vibrating sample magnetometer (VSM). All the magnetic measurements were conducted in a transversal (field lines perpendicular to the plane of the pellet) configuration at ambient temperature. The crystal structure of the composites was studied using a Rigaku Smartlab x-ray diffractometer with Cu–k α radiation ($\lambda = 1.540$ 60 Å). The data were collected in the range of 20 < 2 θ < 80 with a step size of 0.01° and a scan step time of 1 s and analyzed using X'Pert HighScore software.

RESULTS AND DISCUSSION

Two factors play a crucial role in the effectiveness of the ferroic phase conjugation: resistivity of the ceramic, which directly governs the electrical poling effectiveness, and the interface of the two materials, which influences the strain transfer.³⁶ As discussed above, the $CoFe_2O_4$ pellets were conditioned prior to the measurements through electric poling to optimize ME properties. Due to the finite resistivity of the $CoFe_2O_4$ phase in the composite and the leakage currents through a network of electrically interconnected

CoFe₂O₄ grains, there are limitations on the magnitude of the electric field that can be applied across the pellet. The dependence of the maximum achievable electric field that was applied during the poling (before samples get damaged by local Joule heating) and the corresponding resistivity values of the samples at the poling temperature (150 °C, see above) on CoFe₂O₄ content and the type of $BaTiO_3$ precursor (cubic or tetragonal) are shown in Fig. 2(a). Since BaTiO₃ is an insulator, higher BaTiO₃ content leads to better electrical isolation of CoFe₂O₄ grains and the increase in pellet resistance, which, in turn, enables higher poling voltages. To measure the maximum achievable ME coefficient value for a given composition of the composite, the maximum achievable electric field was applied to each sample during electric poling. Furthermore, the values of the ME coefficient resulting from poling under identical conditions were compared. For these measurements, an electric field of 0.8 kV/cm was used, which is the maximum achievable electric field for the composite with the lowest resistivity (using the cubic BaTiO₃ precursor). The ME response as a function of dc magnetic field for x·CoFe₂O₄-(1 - x)·BaTiO₃ composites prepared with two different types of BaTiO₃ precursors and poled with the highest dc voltage applicable depending on their resistivity is shown in Figs. 2(b) and 2(c). All the curves exhibit hysteretic behavior^{10,28} that originates from the hysteretic nature of the magnetization reversal in CoFe2O4. A comparison of the maximum ME coefficient of the pellets prepared and conditioned differently is shown in Fig. 2(d). The cubic samples produce higher ME coefficients than the



FIG. 2. (a) Correlation of maximum poling electric field and resistivity with CoFe₂O₄ content in CoFe₂O₄ composites, ME coefficient as a function of dc magnetic field for different compositions prepared with (b) cubic and (c) tetragonal BaTiO₃ precursors, (d) maximum ME coefficient vs CoFe₂O₄ content in different composites depending on the BaTiO₃ precursor type (top) and magnitude of the poling electric field applied to cubic samples (bottom).

tetragonal samples when the concentration of $CoFe_2O_4$ is below 50%. The highest ME coefficient is achieved at a $CoFe_2O_4$:BaTiO₃ concentration of 20%–80% for cubic samples (4.3 mV/Oe cm) and 25%–75% for tetragonal samples (1.83 mV/Oe cm). When the samples based on cubic BaTiO₃ precursor were poled at the same electric fields (0.8 kV/cm), the ME coefficient of each composite is reduced, but their relative performance remains approximately the same. The samples were prepared multiple times to verify these trends.

Piezo-Force Microscopy (PFM) images of 0.3CoFe₂O₄-0.7BaTiO₃ composites are shown in Fig. 3. These samples were polished to a mirror finish (rms roughness below 10 nm) using progressively higher grit sandpaper followed by polishing paste. The PFM probe is placed in contact with the sample surface. Upon applying a 5V ac driving voltage to the probe, the piezoelectric domains in the sample respond by straining, which leads to the deflection of the cantilever. This deflection is interpreted as piezo-amplitude and piezo-phase images that exhibit a contrast between piezoelectric BaTiO3 and non-piezoelectric CoFe2O4 materials. The images show CoFe₂O₄ grains (darker regions on PFM amplitude and phase scans) within a BaTiO₃ matrix (lighter regions), as the brighter domains have higher piezoresponse and a different phase compared to dark regions. The variations in the PFM amplitude observed for BaTiO3 grains result from the variations of crystallographic orientation of individual grains with respect to the excitation electric field, properties of surrounding grains, individual grain sizes, residual stresses, etc. The characteristic grain sizes observed in the sample prepared using cubic BaTiO₃ powder are significantly larger than the grain sizes in the sample prepared using tetragonal BaTiO₃ powder [see Figs. 3(b) and 3(e)], which is the result of significant differences in the morphologies and

compositions of $BaTiO_3/CoFe_2O_4$ precursor mixtures that affect the grain growth dynamics.^{37–41}

Due to the better isolation of $CoFe_2O_4$ grains in cubic samples compared to the networked $CoFe_2O_4$ grains in tetragonal ones, they can be poled more efficiently. Moreover, the formation of a higher interfacial area or core-shell-like structure between piezomagnetic and piezoelectric phases in samples prepared with cubic BaTiO₃ precursor facilitates the strain transfer between two phases.

SEM images of 0.3CoFe₂O₄-0.7BaTiO₃ samples for the cases of cubic and tetragonal BaTiO₃ precursors are shown in Figs. 4(a)and 4(b), respectively. To distinguish between CoFe₂O₄ and BaTiO₃ phases, the samples were etched in hydrochloric acid (HCl 50% v/v aqueous solution) for 3 h to selectively remove CoFe₂O₄. The resulting SEM images of HCl etched 0.3CoFe2O4-0.7BaTiO3 samples for the cases of cubic and tetragonal BaTiO₃ precursors are shown in Figs. 4(b) and 4(e), respectively. Focused ion-beam (FIB) cross-sectioning was used to reveal the depth profile of the composite microstructure. It should be noted that material damage due to Ga ion implantation during typical FIB cross-sectioning used in this work is limited to ~20 nm penetration depth, which is significantly smaller than the characteristic length scales probed here.⁴² As shown in Fig. 4(f), the voids (dissolved $CoFe_2O_4$) in the samples based on tetragonal BaTiO₃ precursors extend up to 8 μ m below the surface of the sample. Since the pores (etched away CoFe₂O₄) are observed all the way down to 8 µm below the surface, CoFe₂O₄ grains in the tetragonal BaTiO₃ precursor based samples are fairly well interconnected. On the other hand, the FIB cross section of the sample based on cubic BaTiO₃ precursor reveals that individual CoFe₂O₄ grains are physically isolated from each other (by BaTiO₃) since no voids are observed below the sample surface [see Fig. 4(c)]. This is



FIG. 3. PFM imaging of 0.3CoFe₂O₄-0.7BaTiO₃ composites prepared with cubic (top row) and tetragonal (bottom row) BaTiO₃ precursors: [(a) and (d)] height (surface topography), [(b) and (e)] PFM amplitude, and [(c) and (f)] PFM phase.



FIG. 4. SEM images of polished 0.3CoFe₂O₄-0.7BaTiO₃ composites (cubic: top and tetragonal BaTiO₃: bottom row): [(a) and (d)] surface, [(b) and (e)] etched surface, and [(c) and (f)] FIB-etched cross section.

in agreement with lower resistance observed in tetragonal precursor samples as compared to cubic ones.

The dependence of magnetostriction, λ , and piezomagnetic coefficient, $d\lambda/dH$, on the magnitude of applied dc magnetic field

for $0.3\text{CoFe}_2\text{O}_4$ -0.7BaTiO₃ composite prepared with two types of BaTiO₃ is shown in Figs. 5(a) and 5(b), respectively. The composite synthesized with the cubic BaTiO₃ precursor exhibits lower magnetostriction values compared to the tetragonal BaTiO₃ precursor



FIG. 5. (a) Comparison of magnetostriction, λ , as a function of DC magnetic field in 0.3CoFe₂O₄–0.7BaTiO₃ composites, (b) comparison of the derivative of magnetostriction, and [(c) and (d)] comparison of the derivative of magnetostriction and ME coefficient for cubic and tetragonal composites, respectively.





based samples. This is attributed to the finer microstructure and smaller $CoFe_2O_4$ grain sizes in tetragonal $BaTiO_3$ samples that help to more effectively distribute the strain throughout the composite.¹⁸ It should be noted that the coercivity values are the same in the magnetic field dependencies of the ME coefficient, magnetostriction, and magnetization (see Fig. 6).^{43,44}

The dependence of the magnetoelectric coefficient on several materials properties can be approximately described using the following equation:⁴⁵

$$\alpha_{ME} = x(1-x) \cdot \frac{d\lambda}{dH} \cdot \frac{d_{33}}{\epsilon_{r33}} \cdot Y_{33}, \qquad (1)$$

where *x* is the volume fraction of the magnetostrictive component, $d\lambda/dH$ is the piezomagnetic coefficient, d_{33} is the piezoelectric

coefficient, ε_{r33} is the relative permittivity of the composite, and Y_{33} is Young's modulus. The above equation is a relatively crude approximation; however, it aids in the interpretation of the trends observed in experimental data. According to Eq. (1), the magnetoelectric coefficient, α_{ME} , is proportional to the value of piezomagnetic coefficient, $d\lambda/dH$, at a given dc magnetic field, which is consistent with the respective data shown in Figs. 5(c) and 5(d).^{32,34} Although the piezomagnetic coefficient is higher for CoFe₂O₄–BaTiO₃ composites based on the tetragonal BaTiO₃ precursor, it does not result in higher magnetoelectric coefficient values, as shown in Fig. 2. The value of the magnetoelectric coefficient is affected by a higher dielectric constant due to the lower resistivity of tetragonal BaTiO₃ based composites [see Eq. (1)], limitations to effectively pole the composites (again due to low resistivity),



FIG. 7. Comparison of PFM amplitude scans of CoFe₂O₄–BaTiO₃ composites prepared with cubic (top row) and tetragonal (bottom row) BaTiO₃ precursors for CoFe₂O₄–BaTiO₃ composites: [(a) and (b)] 10%–90%, [(c) and (d)] 30%–70%, and [(e) and (f)] 70%–30%.



FIG. 8. X-ray diffraction patterns of (a) cubic BaTiO₃ before and after annealing at 1200 °C and (b) annealed 0.3CoFe₂O₄-0.7BaTiO₃ composites. The inset shows the BaTiO₃ (200) reflection.

and a variety of other factors including significant microstructural differences.

The M–H loops for $0.3\text{CoFe}_2\text{O}_4$ – 0.7BaTiO_3 composites and the dependence of their coercivity on CoFe_2O_4 content are shown in Figs. 6(a) and 6(b), respectively. The magnetic field was applied perpendicular to the pellet surface. The observation of the lower coercivity values in samples prepared using the cubic BaTiO₃ precursor is consistent with the above observation that CoFe_2O_4 grain size in these samples is significantly larger than CoFe_2O_4 grain size in samples prepared using the tetragonal BaTiO₃ precursor. The large CoFe_2O_4 grain size (>1 μ m) facilitates magnetization reversal processes via domain wall motion resulting in lower coercivity values.

PFM amplitude scans shown in Fig. 7 for samples prepared cubic and tetragonal $BaTiO_3$ precursors illustrate the increase in the characteristic $CoFe_2O_4$ grain sizes [dark regions with zero (or near zero) PFM amplitude correspond to $CoFe_2O_4$ grains] with the increase in $CoFe_2O_4$ content in the $CoFe_2O_4$ -BaTiO₃ composite. Similarly to the above observations of coercivity value differences between samples prepared cubic and tetragonal $BaTiO_3$ precursors, the increase in $CoFe_2O_4$ grain size for higher $CoFe_2O_4$ content samples leads to the reduction of the composite coercivity. This is also consistent with previously published reports.⁴⁶⁻⁴⁸

The XRD spectra of cubic BaTiO₃ before and after annealing at 1200 °C for 4 h are shown in Fig. 8(a). Non-annealed cubic BaTiO₃ holds Pm-3m space group (JCPDS data No.: 01-074-1961) with the calculated lattice parameters of a = b = c = 3.9966 Å.^{49–51} While characteristic peaks of tetragonal BaTiO₃ are observed for the annealed powder and indexed with P4mm space group (JCPDS No.: 01-081-2201) with lattice parameters of a = b = 3.9961 Å and c = 4.03 219 Å. The estimated tetragonality factor (c/a) is 1.009, which is close to the bulk value of 1.011. The XRD spectra of $0.3CoFe_2O_4-0.7BaTiO_3$ samples after annealing are shown in Fig. 8(b). The two patterns do not exhibit peak splitting at 45° corresponding to the Miller index of (200) and (002) as expected for the tetragonal crystal structure. This can be attributed to the peak broadening of nano-crystalline particles or the cubic-dominant structure, which leads to the lower tetragonality factor.^{52,53} However, as demonstrated in this work, the

composite samples based on both cubic and tetragonal precursors do exhibit magnetoelectric properties, which confirms the presence of tetragonal $BaTiO_3$ in the composite.

CONCLUSIONS

In summary, magnetoelectric composites of CoFe₂O₄-BaTiO₃ were synthesized using commercial CoFe₂O₄ and BaTiO₃ precursors and an inexpensive and scalable process based on ball-milling and high-temperature sintering/annealing. It was observed that magnetoelectric properties are strongly affected by the CoFe₂O₄-BaTiO₃ composition, where the highest magnetoelectric coefficient is exhibited in samples prepared from a 20% CoFe₂O₄/80% BaTiO₃ precursor mixture. The effect of size and crystal structure of the BaTiO₃ nanoparticle precursor on magnetoelectric and related magnetic properties of the CoFe₂O₄-BaTiO₃ composite was investigated. Higher magnetoelectric coupling coefficients comparable to those of core-shell nanostructures were observed for composites prepared with 50 nm cubic BaTiO₃. Microstructural characterization using SEM and PFM revealed that CoFe₂O₄ grains in a composite based on cubic BaTiO₃ precursor nanoparticles are larger and better isolated from each other by the surrounding BaTiO₃ matrix than CoFe₂O₄ grains in a composite based on tetragonal BaTiO₃ precursor nanoparticles. Larger, higher crystallinity magnetostrictive (CoFe₂O₄) and piezoelectric (BaTiO₃) phases likely result in enhanced magnetostrictive and piezoelectric coefficients; the enhanced interfaces between the phases improve the coupling between the phases; and higher electrical resistivity enables more effective poling and leads to the effective reduction of the relative permittivity. Combining these effects results in higher ME coefficient values that are observed in the optimized composite based on cubic BaTiO₃ powders.

ACKNOWLEDGMENTS

This work was supported, in part, by the grant from the National Science Foundation (Grant No. CBET-1928334 and Award/Contract No. DMR-1523577).

DATA AVAILABILITY

The data that support the findings of this study are available from the corresponding author upon reasonable request.

REFERENCES

¹N. A. Spaldin, S.-W. Cheong, and R. Ramesh, Phys. Today **63**(10), 38 (2010).

²J. Ma, J. Hu, Z. Li, and C.-W. Nan, Adv. Mater. 23, 1062 (2011).

³J. S. Andrew, J. D. Starr, and M. A. K. Budi, Scr. Mater. 74, 38 (2014).

⁴D. C. Lupascu, H. Wende, M. Etier, A. Nazrabi, I. Anusca, H. Trivedi, V. V. Shvartsman, J. Landers, S. Salamon, and C. Schmitz-Antoniak, GAMM-Mitteilungen 38, 25 (2015).

⁵J.-M. Hu, L.-Q. Chen, and C.-W. Nan, Adv. Mater. 28, 15 (2016).

⁶Y. Cheng, B. Peng, Z. Hu, Z. Zhou, and M. Liu, Phys. Lett. A 382, 3018 (2018).

⁷H. Palneedi, V. Annapureddy, S. Priya, and J. Ryu, Actuators 5, 9 (2016).

⁸N. Ortega, A. Kumar, J. F. Scott, and R. S. Katiyar, J. Phys.: Condens. Matter 27, 504002 (2015).

⁹A. Baji, Y.-W. Mai, R. Yimnirun, and S. Unruan, RSC Adv. 4, 55217 (2014).

¹⁰T. Walther, U. Straube, R. Köferstein, and S. G. Ebbinghaus, J. Mater. Chem. C 4, 4792 (2016).

¹¹J. Nie, G. Xu, Y. Yang, and C. Cheng, Mater. Chem. Phys. 115, 400 (2009).

¹²M. Etier, Y. Gao, V. V. Shvartsman, D. C. Lupascu, J. Landers, and H. Wende, in Proceedings of the ISAF-ECAPD-PFM 2012, 2012.

¹³G. V. Duong, R. Groessinger, and R. S. Turtelli, IEEE Trans. Magn. 42, 3611 (2006).

¹⁴A. Khamkongkaeo, P. Jantaratana, C. Sirisathitkul, T. Yamwong, and S. Maensiri, Trans. Nonferrous Met. Soc. China 21, 2438 (2011).

¹⁵ S. Q. Ren, L. Q. Weng, S.-H. Song, F. Li, J. G. Wan, and M. Zeng, J. Mater. Sci. 40, 4375 (2005).

¹⁶C. T. D. Leo, C. T. De Leo, G. C. Dannangoda, M. A. Hobosyan, J. T. Held, F. Safi Samghabadi, M. Khodadadi, D. Litvinov, K. A. Mkhoyan, and K. S. Martirosyan, Ceram. Int. 47(4), 5415–5422 (2021).

¹⁷Y. Deng, J. Zhou, D. Wu, Y. Du, M. Zhang, D. Wang, H. Yu, S. Tang, and Y. Du, Chem. Phys. Lett. **496**, 301 (2010).

¹⁸S. D. Bhame and P. A. Joy, Sens. Actuators, A **137**, 256 (2007).

¹⁹S. D. Bhame and P. A. Joy, J. Am. Ceram. Soc. **91**, 1976 (2008).

²⁰ A. Muhammad, R. Sato-Turtelli, M. Kriegisch, R. Grössinger, F. Kubel, and T. Konegger, J. Appl. Phys. **111**, 013918 (2012).

²¹ B. Jiang, J. Iocozzia, L. Zhao, H. Zhang, Y.-W. Harn, Y. Chen, and Z. Lin, Chem. Soc. Rev. 48, 1194 (2019).

²² M. Acosta, N. Novak, V. Rojas, S. Patel, R. Vaish, J. Koruza, G. A. Rossetti, and J. Rödel, Appl. Phys. Rev. 4, 041305 (2017).

²³ A. Filippetti and N. A. Hill, Phys. Rev. B 65, 195120 (2002).

²⁴B. Ertug, Am. J. Eng. Res. 2(8), 1–7 (2013).

²⁵L. Simon-Seveyrat, A. Hajjaji, Y. Emziane, B. Guiffard, and D. Guyomar, Ceram. Int. **33**, 35 (2007). ²⁶ R. Kumar, S. Guha, R. K. Singh, and M. Kar, J. Magn. Magn. Mater. 465, 93 (2018).

²⁷ M. Rasly, M. Afifi, A. E. Shalan, and M. M. Rashad, Appl. Phys. A **123**, 331 (2017).

²⁸G. V. Duong and R. Groessinger, J. Magn. Magn. Mater. **316**, e624 (2007).

²⁹G. V. Duong, R. S. Turtelli, and R. Groessinger, J. Magn. Magn. Mater. **322**, 1581 (2010).

³⁰US Research Nanomaterials, Inc., https://www.us-nano.com/. Stock numbers US3843, US3835, and US3828, respectively.

³¹ M. M. Kumar, M. Mahesh Kumar, A. Srinivas, S. V. Suryanarayana, G. S. Kumar, and T. Bhimasankaram, Bull. Mater. Sci. **21**, 251 (1998).

³²G. V. Duong, R. Groessinger, M. Schoenhart, and D. Bueno-Basques, J. Magn. Magn. Mater. **316**, 390 (2007).

³³J. V. Vidal, A. A. Timopheev, A. L. Kholkin, and N. A. Sobolev, *Nanostructures and Thin Films for Multifunctional Applications* (Springer, Cham, 2016).

³⁴J. Paul Praveen, V. R. Monaji, S. Dinesh Kumar, V. Subramanian, and D. Das, Ceram. Int. 44, 4298 (2018).

³⁵D. Damjanovic, *The Science of Hysteresis* (Academic Press, 2006), p. 337.

³⁶M. Fiebig, ChemInform 36, R123 (2005).

³⁷A. Rollett, G. S. Rohrer, and J. Humphreys, *Recrystallization and Related Annealing Phenomena* (Newnes, 2017).

³⁸ M. N. Kelly, W. Rheinheimer, M. J. Hoffmann, and G. S. Rohrer, Acta Mater. 149, 11 (2018).

³⁹S. J. Dillon, M. Tang, W. C. Carter, and M. P. Harmer, Acta Mater. 55, 6208 (2007).

⁴⁰ R. D. MacPherson and D. J. Srolovitz, Nature **446**, 1053 (2007).

⁴¹J.-S. Park, "Microstructural evolution and control in the directional solidification of Fe-C-Si alloys," Ph.D. dissertation (Iowa State University, 1990).
⁴²J. Mayer, L. A. Giannuzzi, T. Kamino, and J. Michael, MRS Bull. **32**, 400 (2007).

⁴³E. Hristoforou and A. Ktena, J. Magn. Magn. Mater. 316, 372 (2007).

⁴⁴ M. M. Vopson, Y. K. Fetisov, G. Caruntu, and G. Srinivasan, Materials 10, 963 (2017).

⁴⁵A. S. Zubkov, Elektrichestvo 10, 77 (1978).

⁴⁶M. M. El-Okr, M. A. Salem, M. S. Salim, R. M. El-Okr, M. Ashoush, and H. M. Talaat, J. Magn. Magn. Mater. **323**, 920 (2011).

⁴⁷R. S. Yadav, I. Kuřitka, J. Vilcakova, J. Havlica, J. Masilko, L. Kalina, J. Tkacz, J. Švec, V. Enev, and M. Hajdúchová, Adv. Nat. Sci.: Nanosci. Nanotechnol. 8, 045002 (2017).

⁴⁸O. Perales-Pérez and Y. Cedeño-Mattei, *Magnetic Spinels: Synthesis, Properties and Applications* (IntechOpen, 2017).

⁴⁹Y. Kobayashi, A. Nishikata, T. Tanase, and M. Konno, J. Sol-Gel Sci. Technol. **29**, 49 (2004).

⁵⁰K. Uchino, E. Sadanaga, and T. Hirose, J. Am. Ceram. Soc. **72**, 1555 (1989).

⁵¹X. Li and W.-H. Shih, J. Am. Ceram. Soc. **80**, 2844 (1997).

⁵²H.-W. Lee, S. Moon, C.-H. Choi, and D. K. Kim, J. Am. Ceram. Soc. **95**, 2429 (2012).

⁵³X. Zhu, Z. Zhang, J. Zhu, S. Zhou, and Z. Liu, J. Cryst. Growth 311, 2437 (2009).