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| 1 | SUSTAINABLE EXTRACTION OF PROTEINS AND BIOACTIVE |
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| 2 | SUBSTANCES FROM POMEGRANATE PEEL (PUNICA GRANATUM L.) |
| 3 | USING PRESSURIZED LIQUIDS AND DEEP EUTECTIC SOLVENTS |
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20 Pomegranate peel is a source of proteins, bioactive peptides, and phenolic 21 compounds. The simultaneous extraction of these compounds required the use of 22 polluting solvents and reagents that are non-suitable. This work targets the development 23 of green methodologies based on pressurized liquids (PLE) or deep eutectic solvents 24 (DES) for the extraction of these compounds. Extracts were digested with different 25 proteolytic enzymes and different functionalities (antioxidant, hypocholesterolemia, and 26 capacities) evaluated. antioxidant antihypertensive were Highly and 27 hypocholesterolemic extracts and hydrolysates were obtained using PLE while high 28 antihypertensive capacity was observed in the hydrolysates from proteins extracted 29 using DES. Peptides and polyphenols were identified by HPLC-ESI-Q-TOF/MS. 30 Higher amounts of peptides were shown in hydrolysates from DES extracts while 31 hydrolysates from PLE extracts presented higher amounts of phenolic compounds. 32 Some peptides were assigned to proteins from *Punica granatum*. Both green methods 33 improved the extraction of bioactive compounds from pomegranate peel compared to 34 the non-sustainable method.

35

36 Keywords: pomegranate peel; pressurized liquid extraction (PLE); deep eutectic
37 solvents (DES); peptide; polyphenol; mass spectrometry.

38

40 **1.** Introduction

41 Wastes and coproducts are generated within different phases of the food cycle 42 (Kumar et al., 2017). They are currently discarded in landfills or incinerated. Alternative 43 destinations are their processing into biogas, their composting into biofertilizers (Banerjee et al., 2017) or their use in animal feeding (Kumar et al., 2017). Nevertheless, 44 45 these food wastes can contain valued substances such as proteins and bioactive 46 compounds. The growing consumption of proteins urge for new proteins sources and 47 food coproducts could help to support this demand (Aiking, 2011). Proteins can also be 48 sources of bioactive peptides. Bioactive peptides and, in general, bioactive compounds 49 can provide beneficial effects in terms of health promotion and reduction of the 50 incidence of disorders (Nazir et al., 2019; Santos et al., 2019). Bioactive compounds can 51 be employed for the development of nutraceuticals and functional foods or in medicinal 52 and pharmaceutical preparations (Kumar et al., 2017; Ran et al., 2019).

53 Methods currently employed to extract proteins and bioactive compounds are 54 associated to a high solvent and energy consumption, risk of thermal degradation of 55 heat-labile components, and long extraction times (Kumar et al., 2017; Banerjee et al., 56 2017). Thus, there is a great interest in the development of efficient extraction methods 57 that are more environmentally friendly and that can minimize the degradation of target 58 compounds (Duarte et al., 2014; Sumere et al., 2018). No much progress has taken place 59 in relation with the extraction of proteins that require volatile organic solvents and 60 polluting reagents.

61 Pressurized Liquid Extraction (PLE) uses high pressures and temperatures 62 enabling a reduced extraction time, less solvent consumption, high extraction yields, 63 and use of completely safe solvents (Žlabur et al., 2018). PLE has been used to extract

64 phenolic compounds but it has been scarcely employed for the extraction of proteins65 (Herrero et al., 2015; Ameer et al., 2017).

66 Deep eutectic solvents (DES) are environmentally friendly, easily synthesized, 67 biodegradable, non-volatile, non-toxic, highly stable, and have low cost (Benvenutti et 68 al., 2019). They are constituted by two or more compounds acting as either hydrogen 69 bond donors (HBD) or hydrogen bond acceptors (HBA) (Bai et al., 2017; Jiang et al., 70 2018). The HBA is often a quaternary ammonium salts like choline chloride, whereas 71 the HBD comprises amines, carboxylic acids, alcohols, polyols, acid amides or 72 carbohydrates (Benvenutti et al., 2019). The mixture of these compounds, at a suitable 73 ratio, results in hydrogen bond interactions and the formation of a solvent with lower 74 melting point than those corresponding to its individual components (Ozturk et al., 75 2018; Rajha et al., 2019a). Different DES have been previously employed in the 76 extraction of proteins (Bai et al., 2017; Grudniewska et al., 2018; Wahlström et al., 77 2017) or phenolic compounds (Rajha et al., 2019a; Rajha et al., 2019b; Gullón et al., 78 2019; Pal & Jadeja, 2019; Ozturk et al., 2018; Djaoudene & Louaileche, 2018).

79 Pomegranate peel represents about 40-50% of the whole fruit weight 80 (Kharchoufi et al., 2018; Kaderides et al., 2019; Rajha et al., 2019a). Phenolic fraction 81 has been the most studied within the pomegranate peel (Sumere et al., 2018; Kaderides 82 et al., 2019; Smaoui et al., 2019) while the protein fraction has been hardly considered. 83 Pomegranate peel proteins were extracted by our research group using conventional 84 solvents and high intensity focused ultrasounds (HIFU) (Hernández-Corroto et al., 2019). Results demonstrated that both proteins and phenolic compounds were 85 86 responsible for bioactive properties observing potential synergic effects among them.

87 This work proposes the development of green analytical methods for the 88 extraction of high-added-value compounds from pomegranate peel using sustainable techniques based on pressurized liquids and deep eutectic solvents. Extracted
compounds were identified using HPLC-ESI-Q-TOF/MS and their functionality and
potential synergies were also investigated.

94 2. Materials and methods

95 2.1 Chemicals and samples

96 All reagents were of analytical grade and water was purified in a Milli-Q system 97 from Millipore (Bedford, MA, USA). Hydrochloric acid (HCl), acetonitrile (ACN), 98 acetic acid (AA), and urea were from Scharlau (Barcelona, Spain). Sodium chloride 99 (NaCl) and phosphate buffer (PB) were from Merck (Darmstadt, Germany). DL-100 dithiothreitol (DTT), albumin from bovine serum (BSA), thermolysin, sodium 101 tetraborate, β-mercaptoethanol, o-phthaldialdehyde (OPA), L-glutathione (GSH), 2,2'-102 azino-bis(3-ethylbenzothiazoline-6-sulfonic acid) diammonium (ABTS), potassium 103 persulphate, 1,10-phenantroline, ferrous sulphate, hydrogen peroxide (H₂O₂), bovine 104 pancreatic cholesterol esterase (CEase), p-nitrophenylbutylrate (p-NPB), taurocholic 105 acid, oleic acid, phosphatidylcholine, sodium taurocholate hydrate, angiotensin 106 converting enzyme (ACE), N-(2-hydroxyethyl)piperazine-N'-(2-ethanesulfonic) acid 107 (HEPES), and tripeptide hippuryl-histidyl-leucine (HHL) were obtained from Sigma-108 Aldrich (Saint Louis, MO, USA). Bradford reagent (Coomassie Blue G-250), Laemmli 109 buffer, Tris/glycine/SDS running buffer, Mini-Protean precast gels, Bio-Safe Coomassie 110 G-250 stain, and Precision Plus Protein All Blue standards were acquired at Bio-Rad 111 (Hercules, CA, USA). Ethanol (EtOH) was from Thermo Fisher Scientific (Waltham, 112 MA, USA). Cholesterol assay kit, which contained the assay buffer, the cholesterol 113 reagent, the enzyme mix, and the dye reagent, was obtained from BioAssay Systems 114 (Hayward, CA, USA). Alcalase 2.4 L FG was donated by Novozymes Spain S.A. 115 (Madrid, Spain). Pomegranates were purchased in a local market.

Polyphenols standards (punicalagin, gallic acid, ellagic acid, and punicalin) wereacquired in Sigma-Aldrich.

119 2.2 Preparation of deep eutectic solvents (DES)

120 Different DES were prepared according to Rajha et al. (2019a). HBA and HBD 121 components of DES are described in Table 1. All DES contained water (3th component) 122 to control solvent viscosity. The three components were mixed at a 1:1:3 molar ratio. Mixtures were heated in a water bath at 80 °C with agitation until a clear liquid was 123 124 obtained.

125

126 2.3 *Extraction of proteins and bioactive compounds from pomegranate peels*

127 Pomegranates peels were dried to 50 °C for 48 h. Dried peels were ground in a 128 mortar and, next, in a domestic mill. Finally, they were stored at -20 °C until use. After 129 extraction, proteins in extracts were evaluated by Bradford assay and separated by SDS-130 PAGE (polyacrylamide gel electrophoresis) following a procedure previously described 131 (Hernández-Corroto et al., 2019).

- 132
- 133 2.3.1

Pressurized Liquid Extraction (PLE)

134 Extraction of proteins and bioactive substances was performed using an 135 accelerated solvent extractor system (ASE 150, Dionex, Sunnyvale, CA, USA). 136 Solvents were degassed in an ultrasound bath for 30 min. In every extraction, 2 g of 137 dried pomegranate peels were mixed with 8 g sand and put into a 10 mL stainless steel 138 extraction cell. A circular cellulose filter (2.5 cm, Whatman) was placed at the bottom 139 of the extraction cell to prevent suspended particles from entering the collection bottles. 140 Before extraction, the oven was preheated for 6 min. Optimal extraction conditions 141 were: extraction pressure, 1500 psi; extraction solvent, 70% (v/v) EtOH; extraction 142 temperature, 120 °C; static extraction time, 3 min; extraction time, 12 min; and one static cycle. The extracts were evaporated in a centrifugal concentrator (Eppendorf AG,
Hamburg, Germany) and pellets were stored at -20°C.

145

146 2.3.2 Extraction with DES

147 Extractions were firstly carried out by mixing 150 mg of dried pomegranate 148 peels with 5 mL of the DES grouped in Table 1 using a High Intensity Focused 149 Ultrasounds (HIFU) probe (model VCX130, Sonics Vibra-Cell, Hartford, CT, USA) for 150 1 min at 30% amplitude (Hernández-Corroto et al., 2019). After DES selection, an 151 incomplete factorial experimental design of second order, based on three levels, was 152 employed to optimize HIFU and other extraction conditions. Four different factors such 153 as molar ratio between DES components (1:1, 1:2, and 1:3), molar ratio of water (10, 154 15, and 20), HIFU amplitude (30%, 45%, and 60%), and extraction time (1, 8, and 15 155 min) were used. The response was the protein content (mg protein/g peel), which was 156 determined by the Bradford method. Twenty nine experiments were conducted in a 157 randomized order, corresponding to twenty four points of the factorial design and five 158 additional center points to consider the experimental errors. The experimental design 159 and data analysis were carried out by Box-Behnken design with Statgraphics Centurion 160 XVII software (Statpoint Technologies, Inc., Warranton, VA, USA). Experimental data 161 were fitted to a quadratic model using a second-order polynomial model equation:

162

Protein content (mg protein/g peel) = $\beta_0 + \sum \beta_i X_i + \sum \beta_{ii} X_i^2 + \sum \beta_{ij} X_i X_j$

where β_0 is the constant, β_i is the linear regression coefficient, β_{ii} is the quadratic regression coefficient, and β_{ij} is the interaction regression coefficient, while X_i and X_j are the independent variables. The determination coefficient (\mathbb{R}^2) and the analysis of variance (ANOVA) at a confidence level of 95% were employed to evaluate the fittingof data to the polynomial model equation.

Extracts were next centrifuged (10 min at 4000 xg) and supernatants were collected. Proteins in supernatants were precipitated with cold EtOH (15 mL, 4 °C, 24 h) and centrifuged (10 min, 4000 xg). The pellet was purified again using the same procedure. The resulting pellet was dried at room temperature and stored at -20 °C.

172

173 2.4 Enzymatic digestion of proteins

Protein hydrolysis was carried out using two different enzymes (alcalase and
thermolysin) under optimal conditions (Hernández-Corroto et al., 2019).

176 Evaporated extracts obtained by PLE were dissolved in a 5 mM borate buffer 177 (pH 9.0) at a concentration of 5 mg/mL, when digesting with the alcalase enzyme, and 178 in a 5 mM phosphate buffer (pH 7.5) at the same concentration, when digesting with the 179 thermolysin enzyme. For these purposes, the HIFU probe was employed for 5 min at 180 30% amplitude. The alcalase/substrate ratio was 0.3 AU/g protein while the 181 thermolysin/substrate ratio was 0.1 g enzyme/g substrate. Solutions were incubated in a 182 Thermomixer Compact (Eppendorf AG, Hamburg, Germany) at 50 °C for 2 h, for the 183 digestion with alcalase, and at 70 °C for 1 h, for the digestion with thermolysin. After 184 hydrolysis, the temperature of both solutions was raised to 100 °C and kept at this 185 temperature for 10 min to stop the reaction. Resulting solutions were centrifuged (10 186 min, 6000 rpm) and supernatants, containing peptides, were stored at -20 °C.

187 Proteins extracted using DES were dissolved in a 100 mM borate buffer (pH
188 9.0), for the digestion with both enzymes, at a concentration of 5 mg/mL. For this

purpose, the HIFU probe was employed for 15 min at 30% amplitude. Digestions werenext carried at the conditions described before.

191 Peptide content in all hydrolysates was determined following the OPA method192 described in Hernández-Corroto et al. (2018).

193

194 2.5 Evaluation of the functionality of extracts and hydrolysates

Antioxidant capacity was determined by the evaluation of the capacity of samples to inhibit the formation of hydroxyl radicals and to scavenge free radicals (Hernández-Corroto et al., 2018). Extracts and hydrolysates of two independent experiments were analyzed by triplicate.

Hypocholesterolemic capacity was determined by the evaluation of the capacity of samples to inhibit the cholesterol esterase (CEase) enzyme and the cholesterol micellar solubility. Both procedures were previously described in Hernández-Corroto et al. (2019). Antihypertensive capacity was evaluated using a methodology previously described in the same work. Extracts and hydrolysates of two independent experiments were analyzed by triplicate.

205

206 2.6 Identification of peptides and phenolic compounds by RP- and HILIC-HPLC207 ESI-Q-TOF

208 Peptides and phenol compounds were identified using a 6530 series high 209 sensitivity mass spectrometry Quadrupole-Time-of-Flight (Q-TOF) coupled to a High-210 Performance Liquid Chromatograph (HPLC), model 1100, both from Agilent 211 Technologies. Extracts obtained by PLE were filtered using regenerated cellulose 212 syringe filters from Sartorious (Barcelona, Spain) (for the analysis of peptides) or 213 nonsterile hydrophobic PTFE syringe filters from Labbox (Barcelona, Spain) (for the 214 analysis of polyphenols). Both filters had a pore size of 0.45 µm. Extracts obtained 215 using DES passed through a solid phase extraction C18 columns from Isolute (Uppsala, 216 Sweden) to remove salts. RP-HPLC separation was carried out in an Ascentis Express 217 Peptide ES-C18 column (100 mm x 2.1 mm, 2.7 µm particle size) with a guard column 218 (5 mm x 2.1 mm, 2.7 µm particle size), both from Supelco (Bellefonte, PA, USA). 219 HILIC separation was carried out in an Ascentis Express column (100 mm x 2.1 mm, 220 2.7 µm particle size) with a guard column (5 mm x 2.1 mm, 2.7 µm particle size), also 221 from Supelco. Chromatographic conditions for the separation of peptides by RP-HPLC 222 and their detection by MS were the described in Hernández-Corroto et al. (2019). 223 Chromatographic conditions for the separation of peptides by HILIC were: mobile 224 phase A, 65 mM ammonium acetate in water; mobile phase B, ACN; injection volume, 225 15 µL; flow rate, 0.3 mL/min; column temperature, 25 °C. The optimized elution 226 gradient was: 95-78% B in 25 min, 78-60% B in 5 min, 60% B for 5 min, and a 227 reversed gradient from 60 to 95% B in 5 min to recover initial eluting conditions.

228 MS/MS spectra of peptides were analyzed using the *de novo* tool of PEAKS 229 Studio Version 7 software from Bioinformatics Solutions Inc. (Waterloo, Canada). 230 Peptide sequences were accepted if the average local confidence (ALC, expected 231 percentage of correct amino acids in the peptide sequence) was equal or higher to 90%. 232 Since the *de novo* tool cannot differentiate between I and L amino acids, only isoforms 233 with L are shown although both isoforms are equally possible. Peptide sequences were 234 also analyzed by PEAKS DB (database search tool) using FASTA database that 235 included protein sequences from Punica granatum organism extracted from UNIPROT 236 database. Peptides sequences were associated to a protein if the error tolerance was less 237 than 10 ppm and the mass tolerance was 0.5 Da for the fragments. Peptides and proteins with a -10lgP equal or higher to 15 and peptides with ALC equal or higher to 90%confirmed the confidence between them.

240 Phenolic compounds were identified by comparison of their retention times and 241 MS spectra with those corresponding to standards. Polyphenol standards were dissolved 242 in water or MeOH at a concentration of 100 μ M. Rest of polyphenols were assigned 243 according to their fragmentation pattern. All samples were injected, at least, by 244 triplicate.

245

246 2.7 Statistical analysis

247 Statistical analysis was performed using Statgraphics Centurion XVII software 248 (Statpoint Technologies, Inc., Warranton, VA, USA). Values were expressed as mean \pm 249 standard deviation. The analysis of variance (ANOVA) was performed using a 250 significant level of 0.05.

251

252 **3.** Results and discussion

3.1. Optimization of the extraction of proteins from pomegranate peels using greenmethods

Two green methodologies were developed to extract proteins and bioactive compounds from pomegranate peel. First methodology employed PLE and second methodology used DES.

258

259 *3.1.1. Pressurized Liquid Extraction (PLE)*

Different parameters were optimized for the extraction of proteins from pomegranate peel: concentration of extracting solvent, temperature, static cycles, static time, and presence of additives in the solvent.

EtOH was employed as extracting solvent. Protein content in extracts is 263 264 displayed in Table 2. Extraction yield increased at higher percentages of EtOH up to 265 70% (v/v), which was selected for further optimizations. Different temperatures were 266 next tried observing that the protein extraction yield improved at higher temperatures. A 267 temperature of 120 °C was selected as optimum. Afterwards, different static cycles and 268 times were employed, although no significant differences were observed in the 269 extraction yield. Moreover, two additives (DTT and urea) were added to the optimum 270 extracting solvent to denature proteins and promote their extractability although no 271 significant affect was observed.

The extract obtained under optimal conditions was analyzed by SDS-PAGE (see Fig.S1A). Different bands corresponding to proteins from 25 and 150 kDa were observed. Under these conditions, it was possible the extraction of 9 ± 1 mg proteins/g pomegranate peel which is lower than the amount of proteins extracted by the nonsustainable methodology (15 ± 2 mg/g) (Hernández-Corroto et al., 2019).

277

278 *3.1.2. Extraction using DES*

Eight different DES were firstly employed (Table 1). The protein content in extracts is shown in Table 3. The highest protein extraction yield was obtained with ChCl:urea, ChCl:EG, ChCl:AA, and NaOAc:urea DES. Since urea is a reagent that usually interferes in the assay employed for the estimation of proteins (Bradford assay), the protein content determined in extracts obtained with ChCl:urea and NaOAc:urea DES are likely overestimated. Extracts were analyzed by SDS-PAGE (Fig.S1B) observing main bands from 100 to 250 kDa and, in some cases, additional bands at lower molecular weights. From these results, the ChCl:AA DES was selected.

A Box-Behnken experimental design was employed to optimize HIFU and other extraction conditions using the protein extraction yield as response variable. Factors employed in the 29 experiments and protein content in extracts are grouped in Table 4. The second-order polynomial model best fitting collected variables to predict the protein content is,

292 Protein content (mg protein/g peel) = $25.4 - 1.57 X_1 - 0.157 X_2 - 0.225 X_3 - 0.278 X_4 + 0.221 X_1^2 + 0.0410 X_1 X_2 - 0.0123 X_1 X_3 + 0.0425 X_1 X_4 + 0.00248 X_2^2 - 0.00443 X_2 X_3 + 0.0198 X_2 X_4 + 0.00332 X_3^2 + 0.00583 X_3 X_4 - 0.0149 X_4^2$

295 where X_1 is the molar ratio of acetic acid, X_2 is the molar ratio of water, X_3 is the 296 amplitude of the HIFU probe, and X₄ is the extraction time. The mathematical model 297 enabled to predict the 92% of the response variability. An ANOVA determined the 298 suitable fitting of data to the model (p-value of the lack-of-fit > 0.05). The effect of 299 explanatory variables on the protein extraction yield is displayed in a response surface 300 3-D contour plot at different acetic acid concentrations (Fig.1). The higher was the ratio 301 of water and acetic acid in the solvent, the lower was the protein extraction yield while 302 the probe amplitude and the extraction time were positively correlated with the 303 extraction yield. Optimal conditions for the extraction of proteins were: a 304 ChCl:AA:H₂O DES at 1:1:10 molar ratio using an HIFU amplitude of 60% for 11 min. 305 Under these conditions, it was possible the extraction of 20 ± 1 mg protein/g peel. This 306 is more than twice the amount of proteins extracted by PLE $(9 \pm 1 \text{ mg/g})$ and higher 307 than those extracted by the non-sustainable methodology $(15 \pm 2 \text{ mg/g})$ (Hernández-308 Corroto et al., 2019).

309

310 3.2. Fitting of protein extracts to release peptides by enzymatic hydrolysis

Extracted proteins were next hydrolyzed with alcalase and thermolysin, under previously optimized conditions, in order to obtain peptides (Hernández-Corroto et al., 2019). For that purpose, the extract obtained by PLE was evaporated and next dissolved in a suitable digestion buffer.

Since the extract obtained using ChCl:AA DES showed a pH < 3 and alcalase and thermolysin enzymes activity at this pH was very low, it was necessary the precipitation of proteins and their solubilization in a more suitable buffer. Proteins were precipitated with EtOH (Bai et al., 2017) and the resulting pellet was dissolved in 100 mM borate buffer (pH 9), since the pellet could not be dissolved in a buffer with a lower concentration.

321 Peptide content in the hydrolysates of the extract obtained by PLE was 0.3 ± 0.1 322 mg/mL, when using alcalase enzyme, and 0.58 ± 0.02 mg/mL, when using thermolysin, 323 while the peptide content in the hydrolysates obtained from proteins extracted using the 324 DES was 0.9 ± 0.1 mg/mL, in the case of alcalase enzyme, and 0.8 ± 0.1 mg/mL, in the 325 case of thermolysin. These peptide concentrations are lower than the observed in the 326 non-sustainable method (2.9 \pm 0.1 mg/mL and 2.2 \pm 0.1 mg/mL in the hydrolysates 327 obtained with alcalase and thermolysin, respectively). In the case of the extract obtained 328 by PLE, the lower peptide concentration can be attributed to the lower protein 329 concentration of the PLE extract, while in DES hydrolysates, the lower peptide 330 concentration can be because only part of the extracted proteins precipitated with EtOH.

Indeed, extracted proteins decreased from 20 to 5.2 mg protein/g peel after precipitationof proteins with EtOH.

333

334 3.3. Identification of peptides in the hydrolysates by HPLC-ESI-Q-TOF

335 Peptides present in hydrolysates were analyzed by RP-HPLC-ESI-Q-TOF. 336 Fig.2A shows the total ion chromatogram (TIC) of the hydrolysates obtained by PLE (a) 337 or with the DES (b) using alcalase enzyme and the mass spectra corresponding to three 338 peptides. Higher amounts of peptides and intensity were observed in the hydrolysates 339 from proteins extracted using the DES which can be explained taking into account the 340 higher peptide content in hydrolysate obtained from the DES extract. Moreover, 23 341 different peptides were identified in this extract, while only 14 peptides were found in 342 the hydrolysate from proteins obtained by PLE (see Table 5). Fig.2B compares the TICs 343 of hydrolysates obtained using thermolysin. Again, more intense signals were observed 344 for the hydrolysate obtained from the DES extract. In this case, the number of identified 345 peptides was 4, in the hydrolysate obtained from proteins extracted by PLE, and 20, in 346 the hydrolysate obtained from proteins extracted with the DES. A similar situation was 347 observed when the protein isolate obtained using the non-sustainable method was 348 analyzed. In this case, 26 peptides were identified in the hydrolysates obtained with 349 alcalase and 16 peptides in the case of the hydrolysate obtained with thermolysin.

Peptides contained between 4 and 9 amino acids, in hydrolysates from proteins obtained by PLE, and between 4 and 13 amino acids, in hydrolysates obtained using DES (using alcalase enzyme in both cases). Some peptides in the hydrolysate obtained from the DES extract were found in BIOPEP database (Minkiewicz et al., 2008) such as peptides KVLL, responsible for antioxidant activity, KVLI and KVIL, with ACE and dipeptidyl peptidase III (DPP-III) inhibitor activities, respectively, and FEEI, with antithrombotic activity. Peptides in the hydrolysate obtained from the PLE extract using thermolysin contained only 4 amino acids while peptides in hydrolysates obtained from the DES extract, using the same enzyme, presented between 4 and 11 amino acids. Some of these peptides were described in BIOPEP database like ILSS and IISS, with antioxidant and antibacterial activities, respectively, LLEK, with calpain inhibitor and antioxidant activities, and ILEK, with antibacterial activities. Most peptides observed in hydrolysates had a molecular weight below 1 kDa.

363 Peptides in hydrolysates obtained with alcalase presented a higher amount of 364 hydrophobic amino acids than peptides in hydrolysates obtained with thermolysin. 365 Amino acids leucine/isoleucine (L/I) and valine (V) highlighted within peptides released from proteins obtained by PLE and DES. The presence of aromatic amino acids 366 367 was higher in hydrolysates obtained with alcalase than in hydrolysates obtained with 368 thermolysin. Hydrophobic and aromatic amino acids could contribute to antioxidant 369 capacity (Erdmann et al., 2008; Hernández-Corroto et al., 2018). Furthermore, acidic 370 amino acids were identified in hydrolysates from proteins obtained by PLE and using 371 the DES. The presence of these amino acids has been related to hypocholesterolemic 372 peptides (Hernández-Corroto et al., 2019; Prados et al., 2018). This fact could also 373 explain the acidic isoelectric points (pI) observed for identified peptides. Most 374 hydrolysates obtained with alcalase showed poor water solubility, which is related to 375 that fact that these peptides presented high content in hydrophobic and aromatic amino 376 acids. The presence of these amino acids have been linked to a high antioxidant 377 capacity. Unlike them, hydrolysates obtained with thermolysin showed a lower content 378 in hydrophobic amino acids and, thus, they presented a good water solubility (calculated 379 by Peptide2.0). This feature has been observed within hypocholesterolemic peptides 380 (Zanoni et al., 2017). Peptides in hydrolysates obtained with thermolysin also showed a

high amount in phenylalanine (F), tyrosine (Y), and lysine (K) as C-terminus amino acids which seems to be a common characteristic in ACE inhibitory peptides (Erdmann et al., 2008). These solubilities and isoelectric points were obtained using Innovagen's peptide property calculator. Surprisingly, the highest percentage of peptides with F, Y, or K as C-terminus amino acids was found in the hydrolysate obtained using thermolysin from the DES extract.

Hydrolysates from proteins extracted using the DES, which contained higher amounts of peptides, were also analyzed by HILIC. Only one peptide (HPVLV) was identified in the hydrolysate obtained with alcalase while four additional peptides (VTYDYYEL, LSGGPMVVAHE, MPVVAEH, and ARAR) were observed in the hydrolysate obtained with thermolysin. Mass spectra are displayed in Fig.S2.

392 Some identified peptides in Table 5 were also found in hydrolysates obtained 393 from proteins extracted using the non-sustainable method (Hernández-Corroto et al., 394 2019). Additional peptides, shown in Table S1, were also in common among peptides 395 when ALC was reduced at 80% or when these peptides appeared, at least, in one 396 replicate of the hydrolysates obtained by green methodologies. Higher amount of 397 common peptides was found in hydrolysates with alcalase.

Furthermore, some of these peptides could be assigned to proteins from *Punica granatum*. Table 6 grouped the name of three proteins and the number of peptides from Table 5 that were within their sequences. A more detailed description of these peptides is in Table S2.

402

403 *3.4.* Evaluation of the antioxidant capacity

404 Protection capacity against oxidation damage of hydrolysates and extract was 405 evaluated. Fig.4 shows that all hydrolysates and extracts presented a high antioxidant 406 capacity. Extract and hydrolysates obtained by PLE presented higher antioxidant 407 capacity than the extract and hydrolysates obtained using the DES. Indeed, it was 408 necessary a 1:12 dilution in the extract obtained by PLE and in hydrolysates to avoid 409 signal saturation. No significant differences were observed between the antioxidant capacity of hydrolysates obtained with alcalase and thermolysin enzymes and that of the 410 411 extract obtained by PLE. Probably, polyphenols were coextracted along with proteins 412 contributing to the antioxidant capacity while contribution of peptides released from 413 extracted proteins was minimal.

Hydrolysate obtained using alcalase from the DES extract showed a slightly more antioxidant capacity than the hydrolysate obtained with thermolysin. This fact could be justified taking into account the higher amount of hydrophobic amino acids in peptides released with alcalasa. In addition, the assay evaluating the capacity to scavenge ABTS radicals showed a higher antioxidant capacity for the hydrolysate obtained with alcalase than for the non-hydrolyzed extract.

420 Antioxidant capacity for the protein isolate and its hydrolysates, obtained by the 421 non-sustainable method (77-83% inhibition of hydroxyl radical formation and 68-74% 422 scavenging of ABTS radicals, previous three times dilution of extracts) (Hernández-423 Corroto et al., 2019), were similar to the observed when using the DES but much lower 424 than the observed when using PLE. The PLE method probably co-extracted phenolic 425 compounds that highly contribute to the observed antioxidant capacity while the extract 426 obtained using the DES and the extract from the non-sustainable method showed a 427 higher contribution of proteins to the antioxidant capacity.

428 3.5. Evaluation of hypocholesterolemic capacity

429 Hypocholesterolemic capacity of the extracts and hydrolysates were grouped in 430 the Fig.5. The hypocholesterolemic capacity of both extracts was always higher or 431 similar to that of hydrolysates. Hydrolysates obtained with thermolysin presented higher 432 capacity to inhibit cholesterol esterase enzyme and to reduce micellar cholesterol 433 solubility than the hydrolysates obtained with alcalase with the exception of the 434 hydrolysate obtained with thermolysin from the DES extract. This fact can be explained 435 taking into account the lower amount of hydrophobic amino acids observed in the 436 hydrolysates obtained with thermolysin. Furthermore, the extract obtained by PLE 437 showed the highest inhibition of cholesterol esterase. This fact could be attributed to the 438 intact proteins or to polyphenols co-extracted with proteins by PLE. The hydrolysis of 439 this extract resulted in a reduction of the capacity to inhibit cholesterol esterase that 440 could be due to two reasons. If capacity of the extract was due to the extracted proteins, 441 probably peptides released during the hydrolysis have no capacity to inhibit the 442 cholesterol esterase enzyme. If capacity of the extract was due to the presence of 443 phenolic compounds, probably they have been degraded under hydrolysis conditions. 444 Neither proteins nor phenolic compounds seem to contribute to the capacity to reduce 445 micellar cholesterol solubility that can be attributed just to the released peptides.

The capacity to reduce cholesterol esterase of the DES extract and its hydrolysates was significantly lower than the observed for the PLE extract and hydrolysates. This fact could be related to the lower amount of phenolic compounds in the extract obtained by the DES. Different results were observed for the protein isolate obtained by the non-sustainable method (Hernández-Corroto et al., 2019). Probably proteins extracted by this method are different from those extracted using DES or PLE.

452

453 *3.6. Evaluation of antihypertensive capacity*

454 ACE inhibition capacity is shown in Fig.6. Hydrolysates obtained from proteins 455 extracted with the DES presented higher percentage of ACE inhibition than the 456 precursor extract. IC₅₀ values of these hydrolysates were $28 \pm 1 \mu g/mL$, when using 457 alcalase, and $23 \pm 5 \,\mu\text{g/mL}$, when using thermolysin. This fact demonstrated that main 458 contributors to antihypertensive capacity in hydrolysates were released peptides. 459 Peptides released from proteins extracted using DES showed a higher antihypertensive 460 capacity than those obtained from proteins extracted by the non-sustainable method (75 461 \pm 8 µg/mL, when using alcalase, and 49 \pm 3 µg/mL, when using thermolysin) 462 (Hernández-Corroto et al., 2019). Taking into account the higher peptide concentration 463 in hydrolysates from the non-sustainable method (2.9 \pm 0.1 mg/mL and 2.2 \pm 0.1 464 mg/mL in the hydrolysate obtained with alcalase and thermolysin, respectively), it is 465 possible to confirm that peptides released from the DES extract are much more 466 antihypertensive than those obtained by the non-sustainable method. This supports that 467 proteins extracted by each method were different.

468 On the other hand, hydrolysates from the extract obtained by PLE showed a 469 lower ACE inhibition percentage, which is likely related to the smaller peptide 470 concentration of hydrolysates. This inhibition percentage was even lower than the 471 observed for the extract obtained by the non-sustainable method.

- 472
- 473

3.7. Identification of polyphenols in the hydrolysates by RP-HPLC-ESI-Q-TOF

474 Since results have suggested that polyphenols could be coextracted in the 475 extraction of proteins using DES or PLE and contribute to the antioxidant and 476 hypocholesterolemic capacity, a study of its presence was next carried out. For that 477 purpose, hydrolysates were analyzed by RP-HPLC-ESI-Q-TOF. Fig.3 shows the 478 chromatogram obtained by RP-HPLC at 260 nm. Twenty different peaks were assigned 479 to polyphenols in hydrolysates from the extract obtained by PLE, confirming previous 480 suspicions. Furthermore, these polyphenols were previously identified and described in 481 the hydrolysate obtained by the non-sustainable method (Hernández-Corroto et al., 482 2019). Mass spectra of identified polyphenols are in the Fig.S3. Especially interesting 483 was the presence of punicalin (peak 5) and punicalagin (peaks 11 and 13) due to its high 484 bioactivity (ref). Unlike these results, extraction with DES was more selective for 485 proteins and it hardly extracted polyphenols. Only three peaks, corresponding to 486 galloyl-HHDP-hexoside, ellagic acid-hexoside, and ellagic acid, with extremely reduced 487 intensities were observed in the Fig.3b. A similar situation was observed when 488 analyzing the hydrolysates obtained with thermolysin. These results demonstrated that 489 the PLE was less selective for pomegranate peel proteins and coextracted phenolic 490 compounds while the selected DES mainly extracted proteins.

491

493 Conclusions

Two different green methods, based on pressurized liquid extraction (PLE) and on a deep eutectic solvent (DES), have been developed for the extraction of proteins and bioactive peptides from pomegranate peel. Antioxidant and hypocholesterolemic capacities observed in the PLE extract and its hydrolysates were attributed to the coextraction of phenolic compounds and to the intact proteins. A high antihypertensive capacity was observed in hydrolysates from proteins extracted by DES, which was attributed to released peptides.

501 A higher number of peptides were identified in hydrolysates obtained from the 502 DES extract than in hydrolysates obtained from the PLE extract. Some of these peptides 503 were assigned to proteins from Punica granatum. Identified peptides showed common 504 features within antioxidant, hypocholesterolemic, and antihypertensive peptides. A 505 higher number and amount of phenolic compounds were observed in the hydrolysates 506 obtained from the PLE extract. Developed green methodologies enabled to obtain 507 extracts and hydrolysates from pomegranate peels with higher bioactivity than the observed when using a previous non-sustainable method. Selection of the most suitable 508 509 extracting method will depend, in each case, on the desired bioactivity.

510

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654 Figure captions

Figure 1. 3-D contour plot showing the effect of the extraction time (min), the HIFU probe amplitude (%), and the molar ratio of water in the DES at different acetic acid molar ratios (AA = 1, 2, and 3) on the protein content (mg protein/g peel).

658

Figure 2. TIC corresponding to the pomegranate peel hydrolysates obtained from the
PLE extract (a) and from the DES extract (b) using alcalase (A) and thermolysin (B)
and mass spectra of three common peptides.

662

663 Figure 3. Chromatograms monitorized at 260 nm corresponding to the hydrolysates 664 obtained with alcalase and thermolysin enzymes from PLE extract and from the DES 665 extract. Peaks identification: 1, HHDP-hexoside; 2, galloyl-hexoside; 3, gallloyl-666 HHDP-gluconate; 4, gallic acid; 5, punicalin; 6, pedunculagin I; 7, pedunculagin III; 8, 667 digalloyl-hexoside; 9, gallocatechin; 10, valoneic acid dilactone; 11, punicalagin α ; 12, 668 punicalagin isomer; 13, punicalagin β ; 14, pedunculagin II; 15, galloyl-HHDP-669 hexoside; 16, digalloyl-gallagyl-hexoside; 17, ellagic acid-hexoside; 18, ellagic acid-670 pentoside; 19, ellagic acid; 20, ellagic acid-deoxyhexoside.

671

Figure 4. Capacity to inhibit the formation of hydroxyl radicals and to scavenge ABTS
free radicals of pomegranate peel extracts obtained by PLE or using a DES and their
hydrolysates obtained with alcalase or thermolysin enzymes. Significant differences are
indicated by a letter (a-c).

676

Figure 5. Capacity to inhibit cholesterol esterase enzyme and to decrease cholesterol
micellar solubility of pomegranate peel extracts obtained by PLE or using a DES and

- 679 their hydrolysates obtained with alcalase or thermolysin enzymes. Significant680 differences are indicated by a letter (a-c).
- 681
- 682 Figure 6. Antihypertensive activity of pomegranate peel extracts obtained by PLE or
- 683 using the DES and their hydrolysates obtained with alcalase or thermolysin enzymes.
- 684 Significant differences are indicated by a letter (a-b).
- 685
- 686

Table 1. Hydrogen bond acceptor (HBA) and hydrogen bond donor (HBD) used in thesynthesis of deep eutectic solvent (DES).

| Component 1 (HBA) | Component 2 (HBD) | Abbreviation | Reference |
|----------------------|----------------------|--------------|---|
| Choline Chloride | Urea | ChCl:urea | Moore et al., 2016; Bai et al., 2017; Sanchez-Fernandez et al., 2017; Jiang et al., 2018; Pal & Jadeja, 2019 |
| Choline Chloride | Ethylene glycol | ChCl:EG | Xu et al., 2015; Bai, et al., 2017; Ozturk et al., 2018; Gullón et al., 2019 |
| Choline Chloride | Glycerol | ChCl:gly | Xu et al., 2015; Bai et al., 2017; Sanchez-Fernandez et al., 2017; Grudniewska et al., 2018; Ozturk et al., 2018 |
| Choline Chloride | Acetic acid | ChCl:AA | Bai et al., 2017 |
| Choline Chloride | Glucose | ChCl:gluc | Xu et al., 2015; Gullón et al., 2019 |
| Choline Chloride | Sorbitol | ChCl:sorb | Xu et al., 2015 |
| Choline Chloride | Citric acid | ChCl:CA | Lores et al., 2017 |
| Sodium acetate | Urea | NaOAc:urea | Wahlström et al., 2017 |

| Concentration of EtOH (%) | Temperature (°C) | Cycles | Time (min) | Additives | Protein content (mg prot/g peel) |
|------------------------------|---------------------|--------|---------------|----------------------|-------------------------------------|
| 100 | 120 | 1 | 3 | - | 4.5 ± 0.4 |
| 80 | 120 | 1 | 3 | - | 7.9 ± 0.8 |
| 70 | 120 | 1 | 3 | - | 9 ± 1 |
| 60 | 120 | 1 | 3 | - | 8 ± 1 |
| 50 | 120 | 1 | 3 | - | 5.4 ± 0.5 |
| 40 | 120 | 1 | 3 | - | 5.2 ± 0.4 |
| 30 | 120 | 1 | 3 | - | 5.3 ± 0.5 |
| 0 | 120 | 1 | 3 | - | 4.5 ± 0.6 |
| 70 | 21 | 1 | 3 | - | 6.2 ± 0.8 |
| 70 | 50 | 1 | 3 | - | 6.6 ± 0.6 |
| 70 | 100 | 1 | 3 | - | 8.3 ± 0.9 |
| 70 | 120 | 1 | 3 | - | 9 ± 1 |
| 70 | 150 | 1 | 3 | - | 8 ± 1 |
| 70 | 120 | 1 | 3 | - | 9 ± 1 |
| 70 | 120 | 3 | 3 | - | 9 ± 1 |
| 70 | 120 | 5 | 3 | - | 9 ± 1 |
| 70 | 120 | 1 | 3 | - | 9 ± 1 |
| 70 | 120 | 1 | 15 | - | 7.9 ± 0.9 |
| 70 | 120 | 1 | 3 | - | 9 ± 1 |
| 70 | 120 | 1 | 3 | 0.25% DTT | 8.7 ± 0.6 |
| 70 | 120 | 1 | 3 | 0.25% DTT + 3 M urea | 9 ± 1 |

691 Table 2. Optimized parameters for the extraction of proteins from pomegranate peel by 692 PLE.

693 694 Bold conditions were selected for next experiment

| DES | Protein content (mg prot/g peel) | | | | |
|------------|-------------------------------------|--|--|--|--|
| ChCl:urea | 14 ± 1 | | | | |
| ChCl:EG | 13 ± 1 | | | | |
| ChCl:gly | 7.1 ± 0.2 | | | | |
| ChCl:AA | 15 ± 1 | | | | |
| ChCl:gluc | 5 ± 3 | | | | |
| ChCl:sorb | 4 ± 1 | | | | |
| ChCl:CA | 6 ± 2 | | | | |
| NaOAc:urea | 11 ± 5 | | | | |

Table 3. Protein content of extracts obtained using different DES.

Bold conditions were selected for next experiments.

| Europinont | | Response Variable | | | | |
|------------|-----------------------|----------------------|----------------------|------------------|---------------|--|
| number | ChCl (molar ratio) | AA (molar ratio) | H2O (molar ratio) | Amplitude (%) | Time (min) | Protein content (mg prot/g peel) |
| 1 | 1 | 3 | 20 | 45 | 8 | 16.6 ± 0.5 |
| 2 | 1 | 2 | 15 | 60 | 1 | 15 ± 1 |
| 3 | 1 | 2 | 10 | 45 | 15 | 16 ± 1 |
| 4 | 1 | 3 | 15 | 45 | 1 | 15 ± 2 |
| 5 | 1 | 2 | 15 | 45 | 8 | 17 ± 1 |
| 6 | 1 | 2 | 10 | 30 | 8 | 17.5 ± 0.4 |
| 7 | 1 | 2 | 15 | 45 | 8 | 17 ± 1 |
| 8 | 1 | 1 | 20 | 45 | 8 | 17.0 ± 0.9 |
| 9 | 1 | 2 | 15 | 45 | 8 | 17.0 ± 0.2 |
| 10 | 1 | 2 | 10 | 60 | 8 | 19 ± 1 |
| 11 | 1 | 2 | 15 | 45 | 8 | 17 ± 1 |
| 12 | 1 | 2 | 20 | 30 | 8 | 17.3 ± 0.5 |
| 13 | 1 | 3 | 15 | 45 | 15 | 17.4 ± 0.4 |
| 14 | 1 | 2 | 15 | 30 | 15 | 17.1 ± 0.8 |
| 15 | 1 | 2 | 15 | 45 | 8 | 17 ± 1 |
| 16 | 1 | 2 | 20 | 45 | 1 | 15 ± 1 |
| 17 | 1 | 1 | 15 | 60 | 8 | 19 ± 1 |
| 18 | 1 | 3 | 15 | 30 | 8 | 17.7 ± 0.8 |
| 19 | 1 | 3 | 15 | 60 | 8 | 18.0 ± 0.8 |
| 20 | 1 | 2 | 15 | 60 | 15 | 19.2 ± 0.3 |
| 21 | 1 | 2 | 10 | 45 | 1 | 16.4 ± 0.3 |
| 22 | 1 | 2 | 15 | 30 | 1 | 16 ± 1 |
| 23 | 1 | 1 | 15 | 30 | 8 | 17.7 ± 0.4 |
| 24 | 1 | 1 | 15 | 45 | 1 | 17 ± 1 |
| 25 | 1 | 1 | 10 | 45 | 8 | 18.0 ± 0.3 |
| 26 | 1 | 2 | 20 | 60 | 8 | 18 ± 1 |
| 27 | 1 | 2 | 20 | 45 | 15 | 17.8 ± 0.4 |
| 28 | 1 | 3 | 10 | 45 | 8 | 16.8 ± 0.9 |
| 29 | 1 | 1 | 15 | 45 | 15 | 17 ± 1 |

Table 4. Optimization of different parameters for the extraction of proteins frompomegranate peel using ChCl:AA DES.

| | Α | Thermolysin enzyme | | | | | | |
|-------------------|------------------|--------------------|-----------|-----------------|------------------|----------|-----------|-----------------|
| | Peptide sequence | RT (min) | Mass (Da) | Ip ^b | Peptide sequence | RT (min) | Mass (Da) | Ip ^b |
| | YYGK | 2.08 | 529.2537 | 9.33 | HVNR | 1.09 | 524.2819 | 10.59 |
| | NAGDY | 2.79 | 538.2023 | 0.74 | LVSE | 2.75 | 446.2376 | 1.00 |
| | STYPTN | 3.47 | 681.2969 | 3.43 | LLSS | 3.51 | 418.2427 | 3.72 |
| | NEGTL | 4.92 | 532.2493 | 0.92 | FADY | 19.41 | 514.2063 | 0.74 |
| | FLGGQ | 15.30 | 520.2645 | 3.45 | | | | |
| Peptides obtained | YDTL | 16.88 | 510.2326 | 0.69 | | | | |
| by PLE | ADGAELEVF | 26.22 | 949.4392 | 0.65 | | | | |
| hydrolysates | WNNF | 26.41 | 579.2441 | 3.58 | | | | |
| using RP-HPLC | YVLV | 27.23 | 492.2948 | 3.37 | | | | |
| | VFDNL | 28.96 | 606.3013 | 0.69 | | | | |
| | AYVLV | 29.43 | 563.3318 | 3.65 | | | | |
| | FYDTL | 30.12 | 657.3010 | 0.69 | | | | |
| | TFYDTL | 32.57 | 758.3486 | 0.69 | | | | |
| | VAFDNV | 33.81 | 663.3228 | 0.72 | | | | |
| | APPPGPH | 2.12 | 671.3391 | 7.88 | LYSK | 1.87 | 509.2849 | 9.74 |
| Peptides obtained | DFGGH | 2.83 | 531.2078 | 4.87 | VVAAE | 2.35 | 487.2642 | 1.00 |
| by HIFU | SLGGASGSTAFQQ | 3.47 | 1209.5625 | 3.43 | LLEK | 3.09 | 501.3162 | 6.85 |
| DES using RP- | KGTTF | 3.65 | 552.2908 | 9.91 | LLSS | 3.72 | 418.2427 | 3.72 |
| HPLC | KDLDLK | 5.24 | 730.4225 | 6.57 | WTSSTTAGK | 4.38 | 937.4505 | 10.01 |
| | HVGEL | 7.65 | 553.2860 | 5.10 | VTDYT | 4.59 | 597.2646 | 0.75 |

Table 5. Sequences and characteristics of peptides identified in hydrolysates from extracts obtained by PLE or using a DES with alcalase and
 thermolysin enzymes using RP-HPLC- and HILIC-QTOF^a.

| | A | lcalase enzy | me | Thermolysin enzyme | | | | |
|--------------------------------|--------------------|--------------|-----------|--------------------|-----------------------------|----------|-----------|-----------------|
| | Peptide sequence | RT (min) | Mass (Da) | Ip ^b | Peptide sequence | RT (min) | Mass (Da) | Ip ^b |
| | ALYE | 13.06 | 494.2376 | 1.00 | LVSEADANN | 6.05 | 931.4247 | 0.71 |
| | FYDTT | 17.54 | 645.2646 | 0.75 | LENY | 8.29 | 537.2435 | 0.95 |
| | SKFYDTT | 18.45 | 860.3916 | 6.36 | LLYK | 9.89 | 535.3370 | 9.74 |
| | KVLL | 18.79 | 471.3420 | 9.91 | LTEDVDAH | 9.91 | 898.4032 | 3.54 |
| | VVDL | 21.32 | 444.2584 | 0.69 | LSYE | 10.38 | 510.2325 | 1.00 |
| | DLPGLK | 24.32 | 641.3748 | 6.66 | LVNYD | 12.26 | 622.2962 | 0.88 |
| | FEEL | 25.37 | 536.2482 | 0.76 | LLNEPT | 15.38 | 685.3646 | 0.97 |
| | NNFL | 25.77 | 506.2489 | 3.21 | YDTTY | 18.02 | 661.2595 | 0.74 |
| | WNNF | 26.47 | 579.2441 | 3.58 | LEGDL | 19.39 | 545.2697 | 0.63 |
| | EHPVLL | 26.80 | 706.4014 | 5.10 | VGAGGF | 19.55 | 506.2489 | 3.67 |
| | AVPLLAK | 30.19 | 710.4691 | 10.19 | FADY | 19.97 | 514.2063 | 0.74 |
| | FYDTL | 30.26 | 657.3010 | 0.69 | LYSKF | 22.76 | 656.3533 | 9.74 |
| | KDFPFPN | 30.76 | 863.4177 | 6.43 | HTMEEYSTT | 26.06 | 1097.4336 | 4.14 |
| | LLTF | 32.78 | 492.2948 | 3.70 | FADENF | 27.19 | 741.2969 | 0.70 |
| | TFYDTL | 32.83 | 758.3486 | 0.69 | | | | |
| | AVFDNV | 34.13 | 663.3228 | 0.72 | | | | |
| | NFADYL | 36.53 | 741.3333 | 0.69 | L | | | |
| Peptides obtained | HPVLV ^c | 10.10 | 563.3431 | 7.56 | VTYDYYEL° | 31.55 | 1064.4702 | 0.63 |
| by HIFU | | | | | LSGGPMVVAHE ^c | 31.69 | 1095.5383 | 5.10 |
| hydrolysates with DFS using | | | | | MPVVAEH ^c | 32.11 | 781.3793 | 5.10 |
| HILIC-HPLC | | | | | ARAR ^c | 34.83 | 472.2870 | 12.10 |

^a All identified peptides showed ALC > 90%, at least in three replicates.

- ^b Isoelectric point (Ip) was determined by Innovagen's peptides property calculator.
- ^c Peptides obtained by HILIC column.

| | Gene name | Protein name | Length | Mass (Da) | Number of peptides found in the protein sequence | | | | | |
|----------|--------------|---|--------|-----------|--|---|---------------|----|------------------|---|
| Accesion | | | | | PLE | | DES (RP-HPLC) | | DES (HILIC-HPLC) | |
| | | | | | Α | Т | Α | Т | Α | Т |
| G1UH28 | PSC | Acidic endochitinase Pun g 14, amyloplastic | 299 | 31,747 | 36 | 7 | 50 | 28 | 0 | 2 |
| Q84VT2 | FAD12 | Delta(12)-acyl-lipid-desaturase | 387 | 44,280 | 0 | 0 | 1 | 0 | 0 | 1 |
| Q84UB8 | FADX | Bifunctional fatty acid conjugase/Delta(12)-oleate desaturase | 395 | 45,828 | 0 | 0 | 1 | 0 | 0 | 0 |

707 Table 6. Peptides in hydrolysates obtained from PLE and DES extracts assigned to proteins from *Punica granatum*.

708 A: Hydrolysates obtained with alcalase enzyme

709 T: Hydrolysates obtained with thermolysin enzyme













➢ Hydrolysate obtained with alcalase enzyme

≅ Hydrolysate obtained with thermolysin enzyme



721 Fig.4





723 Fig.5

