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(R)-(+)-NORANICANINE
A NEW TYPE OF TRIOXYGENATED
BENZYLISOQUINOLINE
ISOLATION AND SYNTHESIS

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Abstract - (R)-(+)-Noranicanine (1a), a new type of benzylisoquinoline precursor was isolated from *Aniba canelilla* (Lauraceae) and its structure was determined by means of spectroscopic data as well as by synthesis.

1-Benzylisoquinolines and their derivatives represent one of the preeminent group of alkaloids in plants.[1] The benzylisoquinoline skeleton is known to be produced, *in vivo*, *via* cyclization of Schiff bases formed from condensation of dopamine with either 4-hydroxyphenylacetaldehyde or 3,4-dihydroxyphenylacetaldehyde.[2] Natural known 1-benzyltetrahydroisoquinolines with a monosubstituted benzyl group therefore always exhibit a *para* functionality in ring C. We now have isolated from a Bolivian Lauraceae, *Aniba canelilla*,[3] (+)-noranicanine (1), a trioxygenated benzyltetrahydroisoquinoline with a different substitution pattern in ring C, the oxygenated substituent being in the *meta* position. The structure of this new alkaloid has been firmly established by synthesis.

The EI mass spectrum of 1 , $C_{18}H_{21}NO_3$, exhibited a typical benzyltetrahydroisoquinoline type fragmentation pattern[4] with a very weak molecular ion at m/z 299 and a base peak at m/z 192 representing the isoquinoline moiety. The nmr spectrum of (+)-noranicanine (CDCl3, 200 MHz, TMS=0) has been summarized around structure 1a. Noteworthy were the absence of a *N*-methyl singlet and the presence of two methoxy singlets at δ 3.79 and 3.85 ppm and two aromatic singlets at δ 6.57 and 6.50 ppm for the isoquinoline moiety. Unlike coclaurine or other compounds monosubstituted in ring C, H-13 was there in evidence as a deshielded doublet of doublets at δ 7.09 ppm ($J_{12,13}=J_{13,14}=7.5\ Hz$). However, the unresolved multiplet centered at δ 6.67 ppm corresponding to the three remaining aromatic protons did not allow us to unambiguously establish the substitution pattern of the benzene ring.
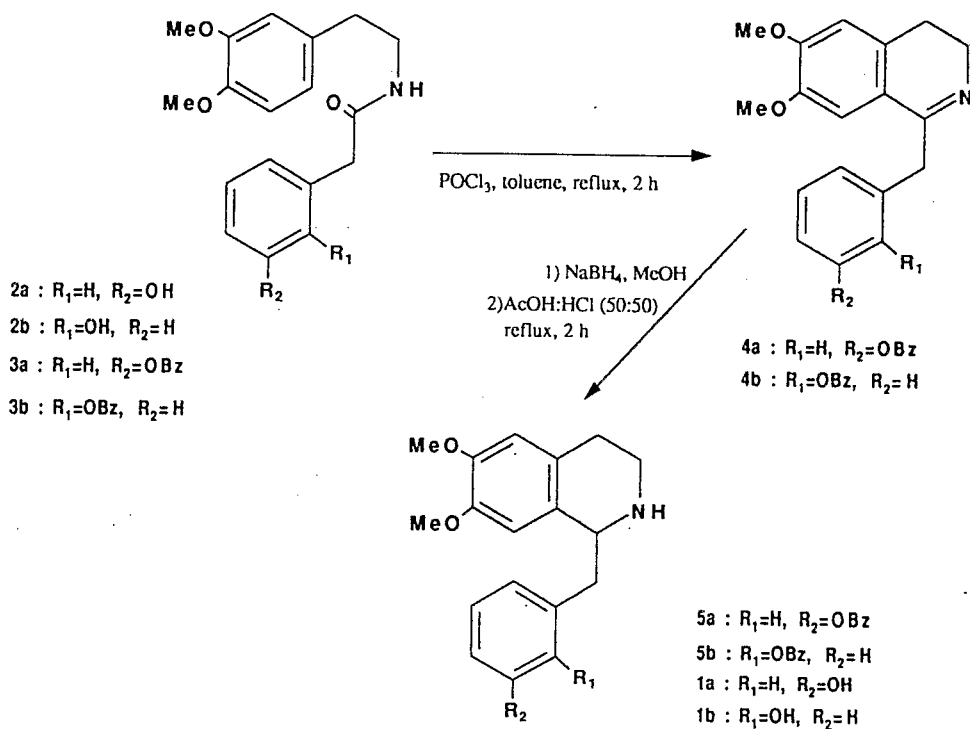
1a

Data with * are interchangeable
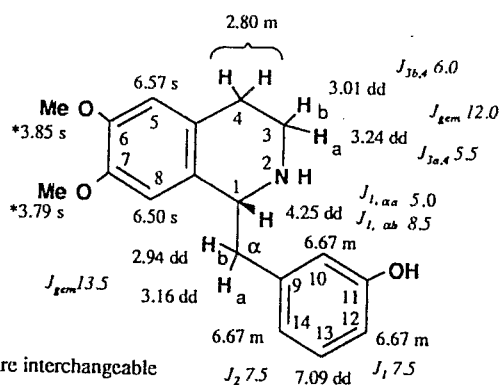
Different assays of solvent effects failed in offering a better resolution than CDCl₃, however the hydroxyl substituent of ring C should be undoubtedly located either in the *meta* (**1a**) or in the *ortho* position (**1b**). We thus undertook the synthesis of both possible compounds *via* a classical route [5,6] in order to confirm the assigned structure **1** (Scheme 1).

2a : R₁=H, R₂=OH

2b : R₁=OH, R₂=H

3a : R₁=H, R₂=OBz

3b : R₁=OBz, R₂=H

POCl₃, toluene, reflux, 2 h

1) NaBH₄, MeOH
2) AcOH:HCl (50:50) reflux, 2 h

4a : R₁=H, R₂=OBz

4b : R₁=OBz, R₂=H

5a : R₁=H, R₂=OBz

5b : R₁=OBz, R₂=H

1a : R₁=H, R₂=OH

1b : R₁=OH, R₂=H

Scheme 1

Condensation of 3,4-dimethoxyphenethylamine with m-hydroxyphenylacetic acid or o-hydroxyphenylacetic acid respectively led to amides (2a) (89%,mp 149-150 °C) and 2b (97%, mp 108-109°C) which were benzylated to give 3a (72%, mp 89-90 °C) and 3b (85%, mp 112-113°C). A Bischler-Napieralski cyclization with 3a and 3b then provided 3,4-dihydroisoquinolines (4a) (71%, mp 79-80 °C) and (4b) (73%, mp 184-185°C) which were further reduced with sodium borohydride into 5a (86%, mp 160-161°C) and 5b (82%, mp 165-166°C), respectively..Debenzylation of 5a and 5b gave the corresponding phenol (1a) (61%, mp 155-156 °C) and (1b) (76%, mp 147-148 °C) as their hydrochlorides.[7] Compound (1a), in its base form, finally was spectroscopically identical with natural (+)-noranicanine.[8]

The 1-R configuration of the natural (+)-noranicanine is suggested by the positive optical rotation [9] and confirmed by the cd curve which presents a negative Cotton effect at 284 nm and a negative tail at 240 nm .[10]

To our knowledge, this is the first report of a natural benzyltetrahydroisoquinoline monosubstituted in ring C with the hydroxyl group being in the *meta* position. Due to the number of alkaloids derived from this basic skeleton, (+)-noranicanine may be therefore regarded as a new type of benzylisoquinoline precursor.

REFERENCES AND NOTES

1   D. W. Hughes, and D. B. MacLean, "The alkaloids : The ^{13}C-nmr spectra of isoquinoline alkaloids", Vol. 18, ed. by R. G. A. Rodrigo , Academic Press, Inc., New York, 1981, pp. 217-262.

2   R. Stadler and M. H. Zenk, Liebig Ann. Chem., 1990, 555.

3   Plant collection, extraction and alkaloid isolation— The stem barks of *Aniba canelilla* (1.1 Kg) were collected in 1988 at Fatima de Chimanes, Bolivia. A voucher specimen (ref AF855) was deposited at the herbarium of ORSTOM-IBBA center of La Paz. The crude alkaloids (1.04%) were obtained from the powdered defatted stem barks following a classical procedure. Specific extraction was followed by MPLC ($CHCl_3$ gradually enriched with MeOH) and preparative TLC ($CHCl_3$-MeOH 13%) on silica gel. This way, 16 mg of amorphous (+)-noranicanine were isolated.

4   M. Tomita, H.Furukawa, T. Kikuchi, A. Kato, and T. Ibuka, Chem. Pharm. Bull., 1966, 14, 232.

5   S. Kubota, T. Matsui, E. Fujita, and S. M. Kupchan, J. Org. Chem., 1966, 31, 516.

6   S. Teitel and A. Brossi, J. Heterocycl. Chem.,1968, 5, 825.

7   Eims data (70 eV) for compounds 2a—>5a and 2b—>5b : 2a m/z : 315 (M$^+$, $C_{18}H_{21}NO_4$, 8.7%), 165 (10.4%), 164 (100%), 152 (4.8%), 151 (20%), 149 (8.8%), 108 (3.1%), 107 (14.8%), 91 (3.1%), 78 (4.8 %), 77 (8.4%), 44 (3.4%) ; 3a m/z : 405 (M$^+$, $C_{25}H_{27}NO_4$, 7.6%), 165 (12.3%), 164 (100%), 151 (12.5%), 149 (4.5%), 134 (3.9%), 107 (3%), 92 (3.8%), 91 (42.3%), 65 (5%) ; 4a m/z : 387 (M$^+$,

$C_{25}H_{25}NO_3$, 5.7%), 386 (6.9%), 297 (22.3%), 296 (100%), 280 (4.6%), 268 (11.5%), 252 (3.5%), 192 (2.1%), 152 (1.7%), 92 (5.7%), 91 (53.1%), 77 (3.9%), 65 (7.4%), 38 (7%), 36 (18.9%), 35 (3.2%) ; **5a** m/z : 389 (M+, $C_{25}H_{27}NO_3$, 0.06%), 280 (1.3%), 193 (12.7%), 192 (100%), 176 (5.8%), 148 (3.5%), 147 (1.6%), 91 (9.5%) ; **2b** m/z : 315 (M+, $C_{18}H_{21}NO_4$, 9.6%), 165 (12.6%), 164 (100%), 152 (4.4%), 151 (15.6%), 107 (12.8%), 106 (3.2%), 91 (3.1%), 79 (3.9%), 78 (4.8%), 77 (8.6%) ; **3b** m/z : 405 (M+, $C_{25}H_{27}NO_4$, 0.8%), 165 (13.5%), 164 (100%), 151 (13.5%), 149 (5.4%), 107 (12.8%), 106 (4.4%), 92 (5.5%), 91 (62.0%), 79 (3.4%), 78 (8.0%), 77 (6.9%), 65 (13.0%) ; **4b** m/z : 387 (M+, $C_{25}H_{25}NO_3$, 0.4%), 377 (3.4%), 296 (6.7%), 281 (13.7%), 280 (71.9%), 264 (5.1%), 128 (5.6%), 126 (15.0%), 92 (8.1%), 91 (100%), 89 (7.5%), 65 (12.5%), 63 (7.1%), 52 (3.1%), 51 (4.7%), 50 (6.2%), 38 (3.2%), 36 (4.5%) ; **5b** m/z : 389 (M+, $C_{25}H_{27}NO_3$, 0.03%), 280 (3.0%), 193 (13.0%), 192 (100%), 176 (7.0%), 148 (3.0%), 91 (11.3%).

8   Spectral data for compounds **1a** and **1b** : <u>**1a**</u> : $C_{18}H_{21}NO_3$, uv λ max (MeOH) nm (log ε) : 207 (4.55), 229 sh (4.03), 282 (3.71), λ max (MeOH-KOH) nm (log ε) : 212 (4.66), 235 sh (3.96), 288 (3.61) ; ir ν max (KBr) : 3450, 3260, 1610, 1585 cm^{-1} ; eims (70 eV) m/z : 299 (M+, 0.2%), 193 (12.5%), 192 (100%), 177 (3.7%), 148 (4.0%), cims (NH3) m/z : 300 ([MH]+, 100%), 193 (3.1%), 192 (24.2%) ; ^{1}H-nmr (200 MHz, CDCl3, δ ppm) : see **1a**, ^{13}C-nmr (68 MHz, CDCl3, δ ppm) : 56.3 (C-1), 40.3 (C-3), 28.5 (C-4), 126.5 (C-4a), 111.7 (C-5), 147.1 and 147.7 (C-6 and C-7), 109.3 (C-8), 128.9 (C-8a), 41.9 (C-α), 139.4 (C-9), 116.2 (C-10), 157.6 (C-11), 114.7 (C-12), 130.0 (C-13), 120.4 (C-14), 55.8 and 55.9 (2 OCH3) ; spectral data for natural (+)-noranicanine , $[\alpha]_D+36°$(c=1, CHCl3), cd (MeOH) $\Delta\varepsilon$ (nm) 0 (299), -0.15 (284), -0.04 (255), negative tail at 240 nm. <u>**1b**</u> : $C_{18}H_{21}NO_3$, uv λ max (MeOH) nm (log ε) : 206 (4.63), 230 sh (4.06), 281 (3.79), λ max (MeOH-KOH) nm (log ε) : 210 (4.76), 233 sh (4.11), 286 (3.80) ; ir ν max (KBr) : 3440, 3290, 1610, 1585 cm^{-1} ; eims (70 eV) m/z : 299 (M+, 0.2%), 193 (15.7%), 192 (100%), 178 (8.7%), 177 (4.0%), 176 (10.9%), 148 (5.7%), 131 (4.6%), 77 (7.8%), cims (NH3) m/z : 300 ([MH]+, 100%), 286 (6.7%), 193 (3.2%), 192 (23.9%), 178 (1.7%) ; ^{1}H-nmr (200 MHz, CDCl3, δ ppm) : 2.83 and 3.35 (2m, 6H, 3 CH2), 3.83 and 3.85 (2s, 6H, 2 OCH3), 4.36 (dd, 1H, J=8.0, 1.0 Hz, H-1), 6.54 (s, 1H, H-8), 6.60 (s, 1H, H-5), 6.73 (dd, 1H, J=7.0, 7.0 Hz, H-13), 6.87 (d, J=7.0 Hz, H-11), 7.03 (d, J=7.0 Hz, H-14), 7.10 (dd, J=7.0, 7.0 Hz, H-12) ; ^{13}C-nmr (68 MHz, CDCl3, δ ppm) : 55.9 (C-1), 38.5 (C-3), 27.1 (C-4), 125.6 and 127.1 and 125.0 (C-4a and C-8a and C-9), 111.0 (C-5), 147.6 and 147.9 (C-6 and C-7), 109.9 (C-8), 40.3 (C-α), 156.7 (C-10), 117.4 (C-11), 128.4 (C-12), 119.1 (C-13), 131.6 (C-14), 55.8 and 55.9 (2 OCH3).

9   J. E. Leet, S. F. Hussin, R. D. Minard, and M. Shamma, <u>Heterocycles</u>, 1982, <u>19</u>, 2355.

10   J. C. Craig, M. Martin-Smith, S. K. Roy, and J. B. Stenlake, <u>Tetrahedron,</u> 1966, <u>22</u>, 1335.