



David Pinto Brandão **Resposta citotóxica de células de melanoma a diferentes nanopartículas de conversão ascendente e a hipertermia**

Melanoma cells cytotoxic response to different upconversion nanoparticles and hyperthermia



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Resposta citotóxica de células de melanoma a diferentes nanopartículas de conversão ascendente e a hipertermia

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Dissertação apresentada à Universidade de Aveiro para cumprimento dos requisitos necessários à obtenção do grau de Mestre em Biologia Aplicada, realizada sob a orientação científica da Doutora Helena Cristina Correia de Oliveira, Investigadora Auxiliar do Departamento de Biologia da Universidade de Aveiro, e coorientação da Doutora Verónica Isabel Correia Bastos, Investigadora do Departamento de Biologia e CESAM da Universidade de Aveiro.

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Ao Gigi por me mostrar a beleza da biologia.

o júri

presidente

Professora Doutora Etelvina Maria de Almeida Paula Figueira,
Professora Auxiliar, Universidade de Aveiro

arguente principal

Doutora Fernanda de Oliveira Esteves Rosário,
Investigadora Contratada, Universidade do Porto

orientador

Doutora Helena Cristina Correia de Oliveira,
Equiparada a Investigadora Auxiliar, Universidade de Aveiro

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palavras-chave

Hipertermia, Melanoma, NaYF₄:Yb,Er (20/2%)@mSiO₂, Gd₂O₃:Yb,Er, Citotoxicidade, Nanopartículas de conversão ascendente

Resumo

O melanoma é um dos tipos de cancro da pele mais agressivos e com uma alta taxa de mortalidade. Portanto, tem havido uma demanda crescente de novas abordagens terapêuticas para neutralizar taxas tão elevadas. A hipertermia é uma abordagem terapêutica que atua ao aumentar a temperatura dentro do tumor, desde 41 a 45 °C. O aumento da temperatura pode interromper os processos bioquímicos das células tumorais, que por sua vez pode se traduzir em morte celular por apoptose ou necrose. No entanto, apesar de promissora, a hipertermia tem alguns obstáculos, especialmente manter uma distribuição homogênea de calor por todo o tumor. Devido às limitações desta técnica, há necessidade de criar novas formas de aplicá-la com alta eficiência.

As nanopartículas podem ser usadas para induzir hipertermia, e têm a vantagem de poderem ser ajustadas especificamente para o tumor, e dessa forma aplicar calor de dentro para fora do tumor. Vários tipos de nanopartículas têm sido usados para gerar hipertermia, mas, mais recentemente, as nanopartículas de conversão ascendente (UCNPs) têm atraído muito interesse por causa das suas características únicas.

O objetivo deste trabalho foi desenvolver as bases necessárias para aplicar a hipertermia usando UCNPs. Para atingir este objetivo, quatro linhas de células de melanoma diferentes foram utilizadas, MNT-1, B16-F10, A375 e SK-MEL-28, juntamente com 2 tipos diferentes de UCNPs, NaYF₄:Yb,Er (20/2%)@mSiO₂ (NaYF₄UCNPs@mSiO₂) e Gd₂O₃:Yb,Er (Gd₂O₃UCNPs).

Antes de cada ensaio, ambos os tipos de UCNPs foram dispersas num banho de ultrasons durante 20 minutos. A caracterização físico-química das nanopartículas foi realizada por espalhamento dinâmico de luz (DLS) para NaYF₄UCNPs@mSiO₂ tamanho das nanopartículas e morfologia também foi avaliada por microscopia eletrônica de transmissão de varrimento (STEM). Os resultados de DLS de Gd₂O₃UCNPs mostraram altos valores de diâmetro hidrodinâmico e índice de polidispersidade, indicando aglomeração das nanopartículas. Além disso, o potencial zeta apresentou baixo valor, indicando a instabilidade das nanopartículas e tendência de agregação. Os resultados de DLS de NaYF₄UCNPs@mSiO₂ mostraram valores aceitáveis de diâmetro hidrodinâmico, com baixos valores de índice de polidispersidade indicando um tamanho mais uniforme de nanopartículas. O potencial zeta de NaYF₄UCNPs@mSiO₂ indica que eles têm estabilidade incipiente a 25 µg/mL e estabilidade inferior a 100 µg/mL. A imagem e análise STEM indicaram tamanho de 77.78 ± 3.53 nm. A citotoxicidade de ambos os UCNPs foi testada pelo protocolo WST-8, com Gd₂O₃UCNPs sendo testados nas linhas celulares MNT-1 e A375, e NaYF₄UCNPs@mSiO₂ sendo testados nas quatro linhas celulares mencionadas acima. Em ambos os casos, as células foram expostas a 12,5, 25, 50, 100 e 200 µg/mL de UCNPs. Gd₂O₃UCNPs causou uma diminuição na viabilidade de A375 nas maiores concentrações após 48 horas de exposição, em comparação com o grupo de controle. NaYF₄UCNPs@mSiO₂ causou uma diminuição na viabilidade celular para todas as linhas celulares para 100 e 200 µg/mL após 24 e 48 horas, com as células MNT-1 também tendo uma diminuição da viabilidade em 25 e 50 µg/mL por 48 horas após a exposição. As células A375 têm uma diminuição da viabilidade para 50 µg/mL em 48 horas após a exposição. A internalização das NaYF₄UCNPs@mSiO₂ só aconteceu nas linhas celulares MNT-1 e SK-MEL-28. O perfil de sensibilidade à hipertermia das linhagens celulares MNT-1 e A375 foi realizado pela exposição a 43 e 45 °C por 30, 60 e 120 minutos, e a viabilidade celular medida 24, 48 e 72 horas após a exposição por meio do ensaio MTT. Em quase todos os casos a viabilidade celular MNT-1 diminuiu com o aumento do tempo de exposição onde, após 120 minutos de exposição, a viabilidade celular ficou abaixo de 60% para todos os tempos de exposição em ambas as temperaturas testadas. Comparativamente, as células A375 expostas a 43 °C não tiveram viabilidade inferior a 60% em todos os casos. Por fim, as células MNT-1 expostas a 45 °C por 120 minutos apresentaram valores de viabilidade abaixo de 20% após 48 e 72 horas, enquanto, por outro lado, a viabilidade das células A375 variou de 40 a 60%, dependendo do tempo de exposição.

Este trabalho permitiu definir um intervalo de concentrações de UCNPs que podem ser usados sem comprometer a viabilidade celular, sendo bons candidatos para hipertermia induzida por radiação próxima do infravermelho para células de melanoma. Este trabalho também permitiu concluir quais temperaturas e tempos de exposição aplicar para potencializar o efeito da hipertermia em células de melanoma. As condições definidas no trabalho atual (concentrações e temperaturas de UCNPs) podem ser replicadas para gerar hipertermia desencadeada por radiação no infravermelho próximo em células de melanoma usando UCNPs.

keywords

Hyperthermia, Melanoma, NaYF₄:Yb,Er (20/2%)@mSiO₂, Gd₂O₃:Yb,Er, Cytotoxicity, Upconversion Nanoparticles

abstract

Melanoma is one of the most aggressive types of skin cancer with a high mortality rate. Therefore, there has been an increasing demand for new therapeutic approaches to counteract such elevated rates. Hyperthermia is a therapeutic approach that works by raising the temperature inside of the tumour, ranging between 41 and 45 °C. The temperature increase may disrupt the biochemical processes of the tumour cells, which in turn can translate into cellular death either by apoptosis or necrosis. However, this promising therapeutic therapy has some hurdles, especially regarding the homogeneous distribution of heat throughout the tumour. Because of the limitations of this technique, there is a need to create new ways of applying it with higher efficiency.

Nanoparticles can be used to induce hyperthermia, and they have the upside of being able to be fine-tuned in order to specifically target the tumour and apply heat from inside-out. Various types of nanoparticles have been used to generate hyperthermia, but more recently upconversion nanoparticles (UCNPs) have garnered a lot of interest due to their unique characteristics.

The objective of this work was to develop the groundwork needed to apply hyperthermia by using UCNPs. To achieve this objective four different melanoma cells lines were used, MNT-1, B16-F10, A375 and SK-MEL-28, along with 2 different types of UCNPs, NaYF₄:Yb,Er(20/2%)@mSiO₂ (NaYF₄UCNPs@mSiO₂) and Gd₂O₃:Yb,Er (Gd₂O₃UCNPs).

Prior to every assay, both types of UCNPs were dispersed in ultrasound bath for 20 minutes. Physicochemical characterization of nanoparticles was performed by dynamic light scattering (DLS) for NaYF₄UCNPs@mSiO₂ nanoparticles size and morphology was also assessed by

Scanning transmission electron microscopy (STEM). DLS results of Gd₂O₃UCNPs showed high values of hydrodynamic diameter and polydispersity index, indicating agglomeration of the nanoparticles. Furthermore, zeta potential showed a low value, indicating the instability of the nanoparticles and tendency to aggregate. DLS results of NaYF₄UCNPs@mSiO₂ showed acceptable values of hydrodynamic diameter, with low values of polydispersity index indicating a more uniform size of nanoparticles. Zeta potential of NaYF₄UCNPs@mSiO₂ indicate that they have incipient stability at 25 µg/mL, and lower stability at 100 µg/mL. STEM imaging and analysis indicated size of 77.78 ± 3.53 nm. Cytotoxicity of both UCNPs were tested by WST-8 protocol, with Gd₂O₃UCNPs being tested on MNT-1 and A375 cell lines, and NaYF₄UCNPs@mSiO₂ being tested on the four cell lines mentioned above. In both cases, cells were exposed to 12,5, 25, 50, 100 and 200 µg/mL of UCNPs. Gd₂O₃UCNPs caused a decrease in the viability of A375 at the highest concentrations after 48 hours of exposure, compared to the control group. NaYF₄UCNPs@mSiO₂ caused a decrease in cell viability for all cell lines for 100 and 200 µg/mL after 24 and 48 hours, with MNT-1 cells also having a decrease of viability at 25 and 50 µg/mL for 48 hours after the exposure. A375 cells have a decrease of viability for 50 µg/mL at 48 hours after the exposure. Cellular uptake of NaYF₄UCNPs@mSiO₂ only happened on MNT-1 and SK-MEL-28 cell lines. Hyperthermia sensitivity profile of MNT-1 and A375 cell lines was performed by exposure to 43 and 45 °C during 30, 60 and 120 minutes, and cell viability measured 24, 48 and 72 hours after exposure through the MTT assay. In almost every case MNT-1 cell viability decreased with the increase of exposure time where, after 120 minutes of exposure, cell viability was below 60% for all the exposure times in both of the tested temperatures. Comparatively, A375 cells exposed to 43 °C did not have viability lower than 60% in all cases. Finally, MNT-1 cells exposed to 45 °C for 120 minutes showed viability values below 20% after 48 and 72 hours, while on the other hand, A375 cells viability ranged from 40 to 60%, depending on the exposure time.

This work allowed to set a range of concentrations of UCNPs that can be used without compromising cell viability, being good candidates for near-infrared induced hyperthermia to melanoma cells. This work also allowed to conclude which temperatures and exposure times to apply in order to potentiate the effect of hyperthermia in melanoma cells. The conditions defined in the current work (UCNPs concentrations and temperatures) can be replicated to generate near-infrared light-triggered hyperthermia in melanoma cells using UCNPs.

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Table 1: Cell seeding densities used on WST-8 assay

List of abbreviations

Abs – Absorbance

dH₂O – MiliQ water

DLS – Dynamic light scattering

DMEM – Dulbecco's modified Eagle's medium

DMSO – Dimethyl sulfoxide

DNA – Deoxyribonucleic acid

FBS – Fetal bovine serum

FCM – Flow cytometry

FS – Forward scattering

Hsp – Heat shock proteins

MTT – Colorimetric 3-(4,5-dimethyl-2-thiazolyl)-2,5-diphenyl tetrazolium bromide

NIR – Near-infrared

PBS – Phosphate-buffered saline

PDI – Polydispersity index

PDT – Photodynamic therapy

PTT – Photothermal therapy

RNA – Ribonucleic acid

ROS – Reactive oxygen species

SD – Standard deviation

SS – Side scatter

STEM – Scanning transmission electron microscopy

TEM – Transmission electron microscopy

UCNPs – Upconversion nanoparticles

UV – Ultraviolet

UVA – Ultraviolet A

UVB – Ultraviolet B

UVC – Ultraviolet C

WHO – World health organization

1. General introduction and objectives

1.1 Cancer

In the current society, cancer is a serious health problem, being the second leading cause of death across the world, following cardiovascular diseases ¹. The World Health Organization (WHO) reported that in 2018 there were 18,078,957 new cases of cancer with around 9,555,027 deaths globally, with Asia and Europe, being the most affected continents (Figure 1) ². Cancer is a group of diseases that can be defined as a multi-cellular disease characterized by uncontrolled growth and proliferation of abnormal cells, that can present different aetiology and occurrence on various types of tissues and organs ³. Different factors may increase the risk of developing cancer, such as genetic factors, ultraviolet rays' exposure, lifestyle among others ³.

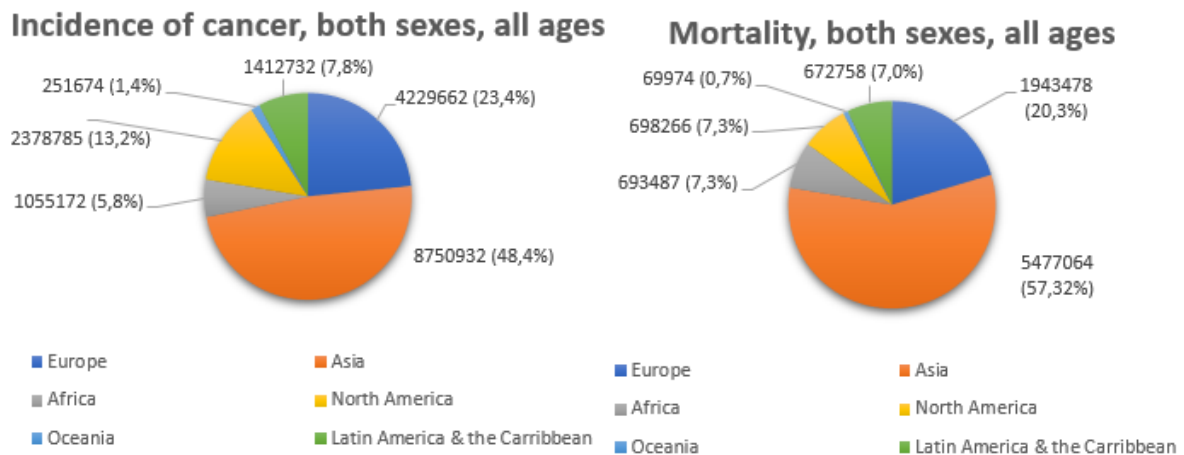


Figure 1: Global cancer incidence and mortality, in both sexes and ages. Adapted from [2].

Tumour cells vary from normal cells by having different structural and functional characteristics, which leads to genetic changes that may bolster tumorigenesis ⁴. However, there are several steps required to develop malignant cells. Accordingly, in 2011, Douglas Hanahan and Robert Weinberg ⁵ proposed 6 necessary hallmarks (Figure 2) for such event:

1. Sustaining proliferative signalling – the most fundamental trait of cancer cells, which is its ability to maintain a chronic proliferation ⁵.
2. Dodging growth suppressors – Tumour suppressors normally regulate cell division, DNA repair and promote apoptosis, but mutations on the genes responsible for these suppressors are normally associated with loss of function, resulting in the inactivation of the proteins produced by these genes.
3. Resisting cell death – Apoptosis is one of the main barriers preventing cancer development. Tumour cells have different strategies to prevent circumvent apoptosis, being the loss of the TP53 tumour suppressor the most common.
4. Enabling replicative immortality – Tumour cells show an unlimited replicative potential making them immortal. In these cells, telomerases are heavily expressed, which may explain why they are immortal.
5. Inducing Angiogenesis – Tumours have the ability to create new vessels from endothelial cells and form the sprouting of existent vessels. These new vessels provide the sustenance required for tumours.
6. Activating invasion and metastasis – The ability to create secondary tumours, by deploying tumour cells from the primary tumour to invade other tissues. This is one of the main characteristics of cancer and one of its major complications.

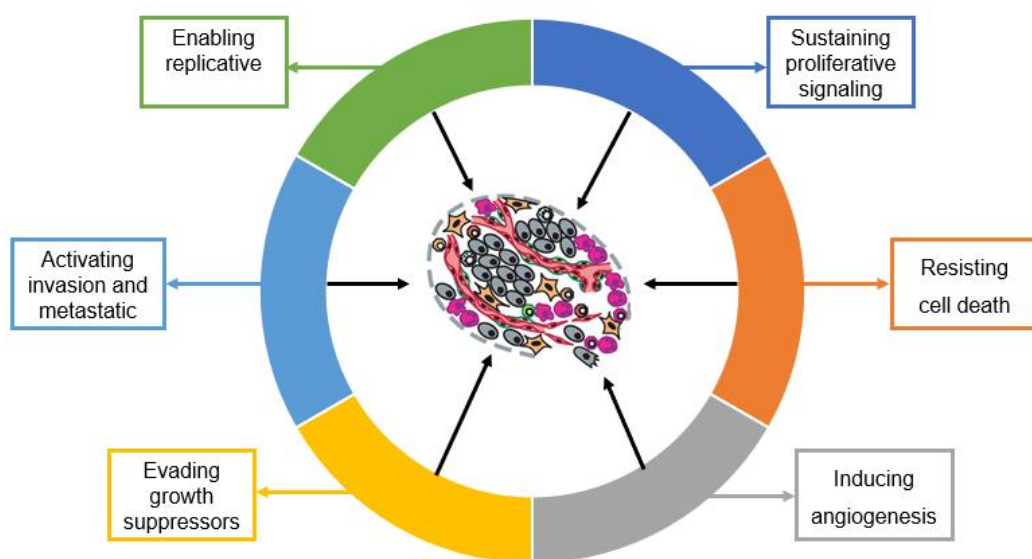


Figure 2: Hallmarks of cancer. Adapted from [5].

These hallmarks represent characteristics that normal cells go through to become tumour cells ⁵, and it is necessary to understand them in order to better treat cancer.

Despite its high incidence, there is not yet a cure or an effective treatment for cancer. There are some therapies that show some success, such as chemotherapy and radiotherapy, but they also affect normal cells and leads to side effects that can condition the life quality of the patient ⁶.

1.1.2 Melanoma

Melanoma is a type of skin cancer that develops from melanocytes, the cells responsible for producing the pigment of the skin - melanin. It is one of the most frequent types of cancer, being the most aggressive form of skin cancer with over 70% mortality rate among skin cancers ⁷. Even though the mortality rate is stable or even decreasing, its incidence rate is increasing, being primarily diagnosed in young patients ^{8,9}. The survival rate for this type of cancer depends mainly in which stage it is diagnosed, the earlier the best, with a high survival rate of 98% with treatment by local surgery ⁶. Survival rate decreases drastically according to which of the later stages it is diagnosed, especially when it reaches metastatic melanoma.

Melanoma can be divided into 2 classes based on histopathological classification: cutaneous and noncutaneous. The cutaneous melanoma can further be divided into 4 different types in relation to clinical and histological features: superficial spreading melanoma, nodular melanoma, acral lentiginous melanoma and desmoplastic melanoma ¹⁰. Non-cutaneous melanoma can appear all over the body, in sites where melanocytes normally occur (ocular, nasopharyngeal, gastrointestinal and genitourinary) ¹⁰.

The incidence of melanoma is also influenced by factors such as genetic component, physiology, and behaviour. Since the decade of 1960, there has been an increasing incidence in Caucasian populations ¹¹, which have a ten times higher risk of developing melanoma skin cancer than those of dark skin ¹². This type of cancer affects mostly young and middle-aged people, with the median age of

diagnosis standing at 57 years ¹². In relation to sex, there is no consistent information, with distinct studies giving different information ¹².

Melanoma is normally correlated to exposure to Ultraviolet (UV) radiation. Ultraviolet radiation is classified as shortwave Ultraviolet C (UVC, 200-290 nm), midwave Ultraviolet B (UVB, 290-320 nm) and longwave Ultraviolet A (UVA, 320-400 nm) ¹³⁻¹⁵. These radiations are differently absorbed by the ozone layer and as such, solar radiation that reaches earth surface is composed of 5 to 10% of UVB and \pm 95% of UVA.

Ultraviolet A radiation and ultraviolet B radiation are able to penetrate the skin (Figure 3), with the former causing the generation of reactive oxygen species (ROS) by endogenous non-DNA chromophores, which cause cell cycle arrest and pyrimidine dimer-type DNA damage ^{16,17}. Reactive Oxygen Species are responsible for various cellular effects by the redox-signalling pathway and oxidative DNA damage ¹⁸. Ultraviolet B is partially absorbed by DNA and induces the formation of DNA photoproducts. These types of damage cause a disruption of the DNA responsible for inhibiting the activity of polymerase during the replication and/or transcription of DNA ^{13,19}.

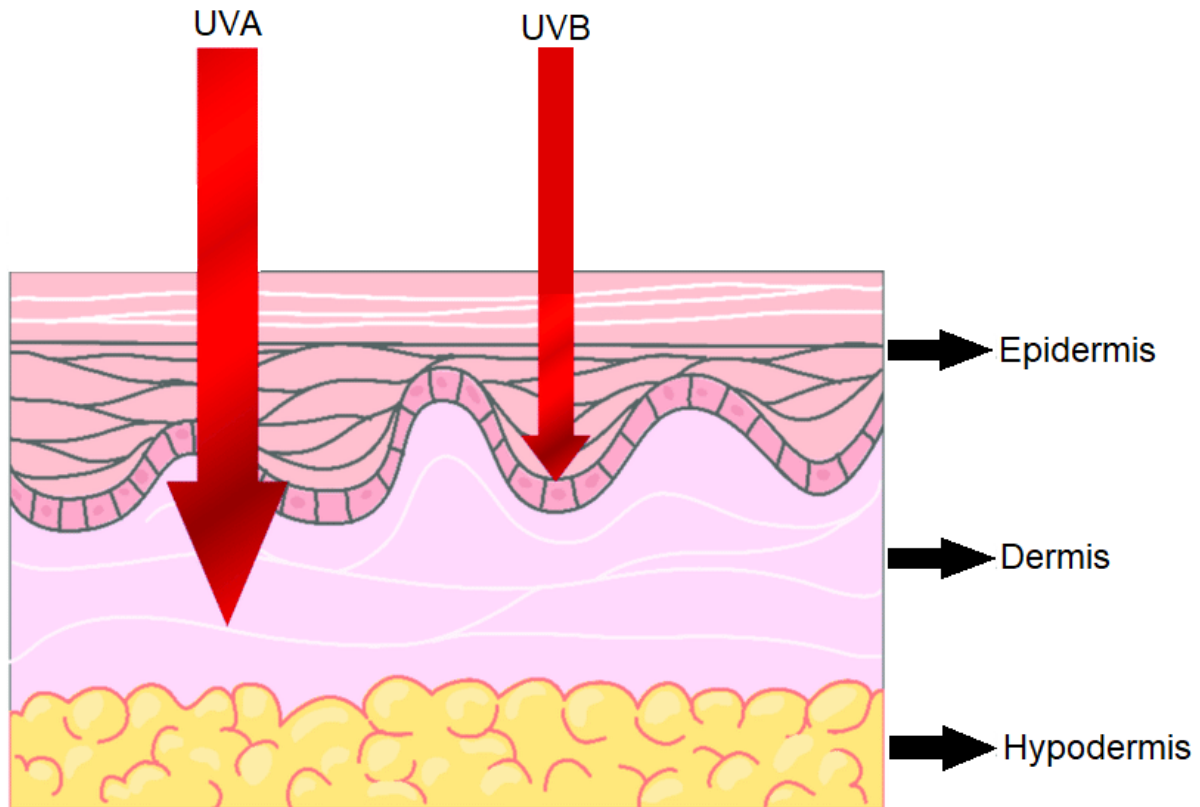


Figure 3: UVA and UVB penetration on skin. Adapted from [144].

Ultraviolet radiation has different effects on melanoma, such as causing a high mutation rate of the genetic material, mainly the conversion of C nucleotide to a T nucleotide at the TpC nucleotides²⁰, photolesions on DNA that codify cellular regulators (p53, EGFr, p16INK4A) which are responsible for cell growth, DNA damage repair, cell-cycle dysregulation and immunosuppression, which increases the probability of skin cancer, by upregulating the immunosuppressive cytokine IL-10^{14,18,21,22}.

Familiar and individual history of melanoma are factors that increase the risk of developing this disease^{23,24}. This happens through the transmission of unreliable genes regarding the repair mechanisms and cell damage-induced signalling pathways^{6,25}. There are numerous genes essential for melanoma development such as cell proliferation regulation (BRAF, NRAS and NF1), DNA repair (PARP1 and ATM) and some genes in which the function is not yet determined (ARNT-SETDB1, CASP8, FTO and MX2)^{24,26}.

The main step to prevent melanoma is protection from UV radiation ¹⁴, with the incidence of melanoma decreasing mainly on increased awareness and regular check-ups of skin lesions such as moles ⁶. Various social campaigns advocate for lower sun exposure during the most UV intense part of the day and the use of sunscreen. The incidence of melanoma is lower when the population is more informed on the effects of UV exposure ¹⁴.

Treatment for melanoma is normally conducted by surgical excision of the entire tumour. The mortality of melanoma increases with the age of the tumour, being early detection and treatment fundamental for a high survival rate.

Over the past decade, there has been an impressive clinical advance on melanoma therapy. One decade ago metastatic melanoma was considered to be a terminal disease with a survival rate lower than 5% ²⁷, however now-a-day there has been a big increase in the survival rate due to new mechanism-based therapeutic strategies ²⁸. These new therapies can be grouped into 3 groups: BRAF inhibitors, MEK inhibitors and immunotherapy.

BRAF mutations are frequent in melanoma, with different incidences depending on the type of melanoma, with nodular melanoma and superficial spreading melanoma having over 50% of BRAF mutations ²⁹⁻³¹. For this reason, BRAF inhibitors show promise against melanoma as well as other types of cancer, by deregulating the signalling cascade RAS-RAF-MEK-ERK. Studies showed a 50% response rate to a single BRAF inhibitor and a 60-70% to a combination of 2 BRAF inhibitors ^{21,32,33}. MEK inhibitors, which also act by downregulating the MAPK pathway signalling also have potential to treat melanoma. Studies on MEK inhibitors showed a survival rate of 81% with an average duration of the response of 5 and a half months ³⁴.

Immunotherapy focus on the blockage of one or two immune checkpoints, which increases the immune response of the cells. This therapy has high durability of response, but the downside is its unpredictability of response. There are currently two available therapies which target PD-1 and CTLA-4, both of which are co-inhibitory protein receptors that are expressed on the lymphocytes cell surface. Lymphocytes principal function is to preserve self-tolerance and reduce inflammatory responses in normal tissues ^{28,35}. This therapy has a lower survival

rate than the previous therapies mention and can also have side effects due to the disruption of normal immunological responses that can affect healthy tissues ^{28,36}.

Even though there have been significant advances in the treatment of melanoma, there is still a need to further develop additional approaches for patients that are resistant to the currently available treatments and create new strategies to better apply the already existent therapies ²⁸. Other than improving the therapies already available to us, there is also a demand for new strategies to have a better range of alternatives in order to have a more fitted treatment of each patient.

1.2 Hyperthermia

Hyperthermia is a type of cancer therapy where heat is applied to a tumour mass by an external physical mean, in which the temperature inside the tumour mass increases between 41 and 48 °C ³⁷. This type of therapy has long been described in the literature, but its use on medicine is somewhat new.

The first correlation between hyperthermia and its effects on cancer was noted in 1779 by Kizowitz, where he stated that high fever in patients infected purposely with malaria showed a retarded tumour growth. The first published paper on the subject would only come 90 years later, when Busch described a complete remission of face sarcoma when the patients had two erysipelas infections caused by bacteria, following a two-year disease-free survival ³⁸. A year later, in 1887, Bruns described complete remission in a patient with recurrent melanoma after infection of erysipelas with high fever (over 40 °C) for several days, with an eight-year disease-free survival ³⁹. William B. Coley caused fever on patients by injecting them with streptococcus and noticed a correlation with the severity of the infection and the degree of regression on the tumour ⁴⁰.

Hyperthermia can be divided into three different types according to the area it is applied to, with local or interstitial corresponding to the appliance of hyperthermia to the area where the tumour is located, regional hyperthermia when it is applied to a part of the body, limb, organ or a hollow cavity, and whole-body hyperthermia where body temperature is elevated to mimic a fever, with the use of warm-water immersion, heating blankets or thermal chambers³⁷.

The effect of hyperthermia on tumours vary according to the distinct characteristics of each tumour and its environment ³⁷. One way of fine-tuning the effect of heat on tumours is to modify the microenvironment where it is located since these changes can increase the response of tumours to heat ⁴¹. Most cancer cells die when exposed to temperatures between 40 and 43 °C, while healthy cells more likely survive ⁴². Cancer cells exposed to those temperatures endure irreversible damage, which can lead to cellular death, dependent on the time and dose applied ^{37,43}.

Pietrangeli and Mondovi ⁴² outlined the biochemical process of tumour cells that are affected by heat, which are:

- Inhibition of RNA synthesis, DNA, DNA repair mechanism and cell respiration,
- Strong inhibition of DNA polymerases- β crucial enzymes in multistep repair system,
- Cell membranes of tumour cells become permeable and fluid in the presence of heat,
- Heighten production of heat shock proteins (hsp) which affects thermo-tolerance and tumour immunogenicity,
- Alterations of cristae in mitochondria,
- Increased influx of reactive oxygen radicals (ROS),
- Promotion of apoptosis. All these effects, plus the microenvironment of the tumour, are responsible for the sensitivity of the tumour to heat ^{41,44}.

Tumour growth requires new vasculature to provide nutrients ³⁷. This new vasculature normally is insufficient and disorganized, which results in rudimentary oxygenation and the presence of areas of hypoxia and necrosis, randomly distributed across the tumour ³⁷. This usually results in a hostile environment, which has negative effects on chemotherapy and radiotherapy, but it can be a boon to hyperthermia. This environment doesn't affect the action of hyperthermia on the tumour, as shown by Koutcher and Gerweck on glioblastomas ^{45,46}, but hypoxic cells, especially the ones in acute hypoxia, are more sensitive to heat ⁴⁵⁻⁴⁷, and Hahn *et al.* showed that the metabolic state and the energy privation of these cells increases their heat sensitivity ⁴⁸.

The effectiveness of hyperthermia depends on the type of tumour it is applied to since different type of cells show different sensitivity to heat treatment. Another factor that can affect cell sensitivity to heat is the phase of their cell cycle ⁴⁹. In most cases, the phase most sensitive is the mitotic phase, where heat damages the mitotic apparatus, which leads to inefficient mitosis and by consequence polyploidy. Cells in S-phase are also sensitive to hyperthermia, and in this phase, the damage is made to the chromosomes. Cells in these two phases suffer a slow mode of cell death after hyperthermia, while cells arrested in G1-phase have comparatively more heat resistant, with no damage observed microscopically, and suffer a rapid mode of death after hyperthermia ⁴⁹. The response varies according to which phase of the cell cycle the cell is in, which indicates the possibility of different cellular death mechanisms after heat exposure ⁵⁰⁻⁵².

Tumours have reduced apoptosis because of an imbalance of pro-apoptotic (Bcl-2 family proteins, p53 expression) and anti-apoptotic proteins (reduced caspase activity, anomalies in the death receptors, increased endogenous inhibitors of apoptosis proteins). On the other hand, hyperthermia was reported by Ahmed and Zaidi to enhance apoptosis ⁵³, by biochemical process's described previously, such as increasing membrane permeability of tumours, increased production of ROS, alterations of the cellular cytoskeleton ^{42,53,54}. Other authors have described that hyperthermia enhances apoptosis by adjusting the expression of apoptotic genes such as p52, Bcl-2 and Bax ⁵⁵, and other have shown that in melanoma, hyperthermia activates a non-conventional apoptotic pathway caspases 3/7 ⁵⁶.

Other than affecting the balance of apoptosis, hyperthermia also has the ability to increase the antigenic presentation of effectors cells and the production of heat shock proteins ⁵⁷⁻⁶⁰, as well as recruiting neutrophils, natural killer cells, regulatory T cells, myeloid suppressor cells and macrophages into the area of the tumour ^{57,58}. Hyperthermia also is able to inhibit angiogenesis, by controlling extracellular matrix degradation by inducing Plasminogen activator inhibitor-1, as demonstrated by Roca *et al.* ⁶¹.

Even though hyperthermia shows a lot of benefits, there is one major hurdle that is the homogenous heat distribution in treated tissues ⁶². Current modalities of hyperthermia do not have the capability of tumour selection, and as such, they can

damage healthy tissue ⁶³. One way of tackling this problem is by using nanoparticles.

Different types of nanoparticles can be used in hyperthermia, such as magnetic nanoparticles, gold nanoparticles or upconversion nanoparticles. All these types of nanoparticles can be fine-tuned to target the tumour, thus circumscribe the hyperthermia to mostly the tumour alone. This ability is normally done by decorating the nanoparticles with tumour-specific molecules, such as folic acid, and as such increase the accumulation of nanoparticles inside the tumour.

Another interesting characteristic of nanoparticle use in hyperthermia is the ability to heat the tumour from the inside (inside-out hyperthermia). This ability tackles another problem of hyperthermia, that is the dissipation of heat. Normally hyperthermia is applied with an external heat source, and in this case, the blood flow inside the tumour allows for quick dissipation of the heat inside the parenchyma of the tumour ⁶⁴, but when nanoparticles are used as the source of heat, they are normally placed inside the parenchyma of the tumour, overcoming this problem. The only possible problem with this type of hyperthermia is that nanoparticles might not be distributed evenly across the tumour, especially the parts further away from the vasculature where the most sensitive cells (hypoxic cells) are located ⁶⁴.

Nanoparticles might be the future of hyperthermia since they can provide a more efficient and localized hyperthermia, and they also have the capability of both treating the disease as well as diagnosing it, if they are constructed in a way that allows them to be visualized by an imaging technique ⁶⁴.

Even though hyperthermia is a promising field and with new discoveries and techniques still being developed, at present time the techniques available do not have a high efficiency when used alone. This inability makes it impossible to replace any of the commonly used therapies to treat cancer, but they have the ability to improve other therapies, such as radiotherapy and chemotherapy, by enhancing the cell-killing effect of cytotoxic drugs and/or radiation ⁴⁹. Because of this, there is a need to create new techniques to apply hyperthermia in a more controlled way and with high efficiency. One possibility is the use of nanoparticles, especially upconversion nanoparticles since they have special characteristics that will be discussed in the next topic.

1.3 Lanthanide nanomaterials

Nanoparticles are solid colloidal particles with size ranging from 1 to 100 nm⁶⁵. They are composed of 3 layers: (i) surface layer, that can be functionalized, (ii) shell layer, which is chemically unlike the core of the nanoparticles and (iii) the core, which is the central part of the nanoparticle⁶⁶. Nanoparticles can be made from a wide variety of materials, such as transitional metals, carbon, metal oxides and inorganic materials. They can be synthesized by various chemical and physical methods⁶⁷, and they are classified according to their dimensionality, morphology, composition, uniformity and agglomeration^{67,68}.

Lanthanides are a group of rare earth chemical elements where the atomic numbers range between 57 and 71 and are located in the sixth period and the IIIB group of the periodic table. They range from lanthanum and lutetium. These elements plus scandium and yttrium are also called rare earth elements. This term was first coined by the chemist Johan Gadolin in 1794, because of the low concentration on rare minerals⁶⁹. The lanthanides were first discovered in 1787 when a mineral was identified in a Swedish town named Ytterby⁷⁰. This mineral was then separated into the various types of lanthanide elements, and then yttria, an impure form of yttrium oxide, was obtained by Professor Gadolin in 1794⁶⁹. Over the past two centuries, the further development of lanthanide chemistry has led to the application of these elements in different fields such as agriculture, chemical industry and biomedicine⁷¹⁻⁷³.

Lanthanides have inherent luminescence which derives from $f-f$ electron transitions in the $4f$ sub-shell, which provide exclusive properties for optical imaging. In recent years, lanthanide luminescence has been gaining a lot of consideration, in particular when doped in inorganic hosts with the size on the nanometre region⁷⁴. The luminescence typically occurs from direct excitation that provokes an excited state, which is followed by emission and return to the ground state. It is necessary, though, a suitable excitation source, in which the wavelength is resonant with the energy gap between the ground state and the excited states⁷⁴.

Luminescence emission of lanthanides can be achieved by a process known as upconversion. This is a process where the excitation light, a low energy near-

infrared (NIR), is converted into higher energy wavelengths shorter than the excitation source, that covers a broad region of the ultraviolet, visible and NIR ⁷⁵. In this case, the absorption of the photons is sequential and not simultaneous ⁷⁶, and it requires that the excitation is provided by ultrafast (femtosecond) pulse lasers. The sequential and step by step absorption of NIR photons is possible by the intermediate excited states, which have lifetimes in the μs to ms range, and therefore allow the excitation to the final energy state of the photons and the generation of a higher energy photon ⁷⁴.

There are three main mechanisms in which upconversion occur:

- Excited-state absorption – Involves a single ion and is the successive absorption of two photons by a single Ln^{3+} ion ⁷⁴ (Figure 4);

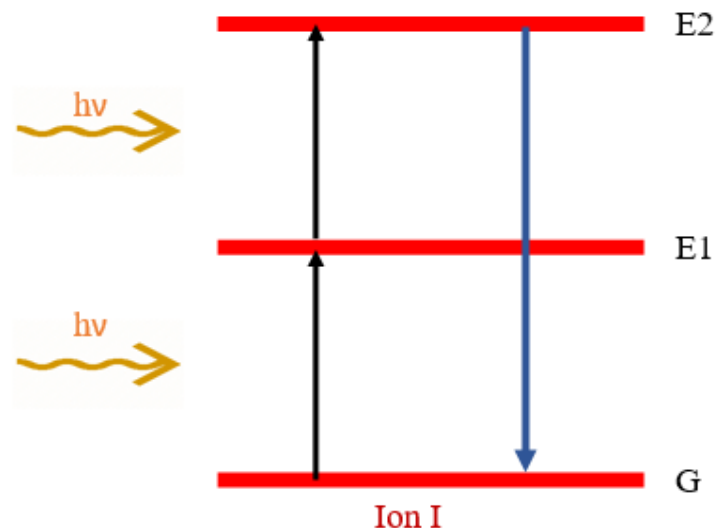


Figure 4: Schematic representation of the excited-state absorption process. Adapted from [74].

- Cross-relaxation upconversion – Two identical neighbouring ions are excited from the ground state to intermediate excited state by ground-state absorption, and then one ion returns to the ground state while the other is promoted to upper emitting level (Figure 5).

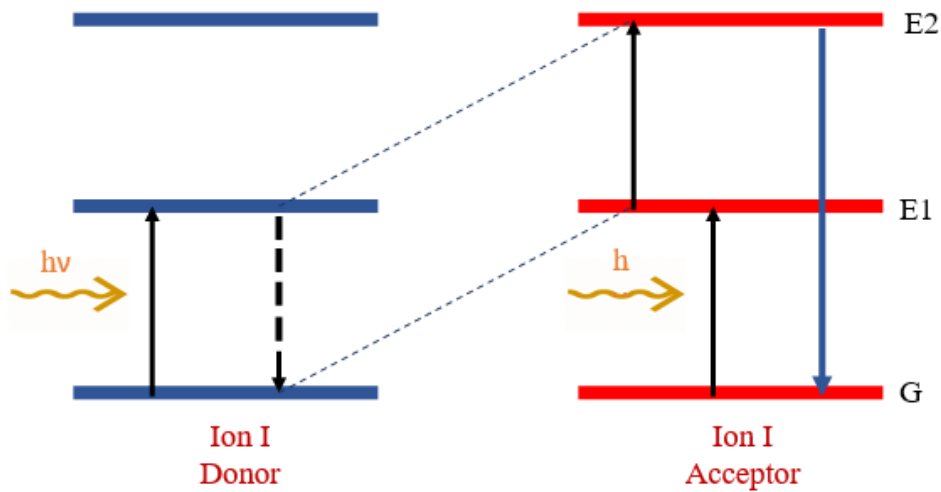


Figure 5: Schematic representation of the cross-relaxation upconversion process. Adapted from [74]

- Energy transfer upconversion – Occurs in co-doped materials, in which successive energy transfer from a donor ion to an acceptor ion happens. The donor ions are excited to intermediate states by ground-state absorption, and then energy transfer from the donor ions to the acceptor ions result in the promotion to its intermediate state. Afterwards, a second energy transfer occurs and promotes the acceptor ion to its emitting state (Figure 6).

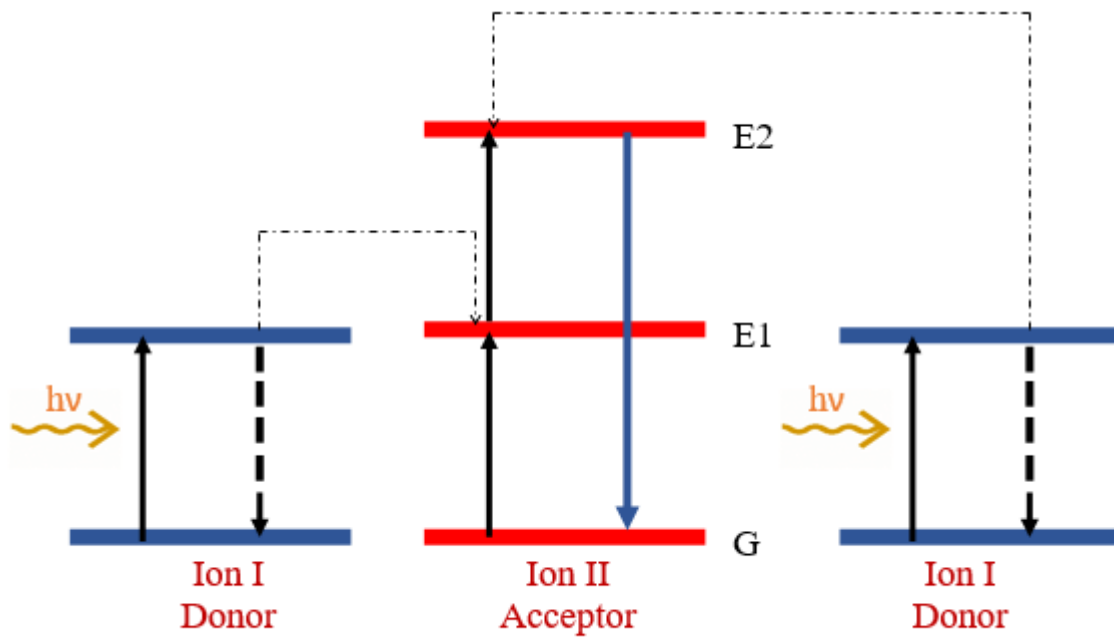


Figure 6: Schematic representation of the energy transfer upconversion process. Adapted from [74].

1.3.1 Composition of UCNPs

Upconversion nanoparticles are composed of a host material, sensitizers and activators. Normally UCNPs are composed of an inorganic host doped with Ln^{3+} ions. These ions should be less than 10 \AA apart in order to have the most efficient method of upconversion ^{74,77}.

1.3.2 Host material for UCNPs

Host material absorption should be as low as possible, so they should be transparent in the NIR and visible wavelengths region, as well as low energy phonons. With these conditions luminescence of the nanoparticles should be maximized.

To this date, the best host materials belong to the fluorides family, since they normally have phonon energy levels of below 500 cm^{-1} and are highly chemically stable ⁷⁸. For the mentioned reasons, the most popular host has been NaREF_4 , such as NaYF_4 ⁷⁹⁻⁸², NaGdF_4 ^{83,84}, NaLuF_4 ⁸⁵.

There are two different structural phases of NaREF₄ UCNPs, the cubic phase (α -) and the hexagonal phase (β -), with the latter being the most efficient phase due to its crystal structure ⁸⁶.

1.3.3 Sensitizers for UCNPs

Sensitizers should have a high absorption cross-section which matches the energy levels and excitation wavelength of the activator. Ytterbium (Yb³⁺) fulfills all the previous conditions, with his energy level only having one excited state (²F_{5/2}) that resonates with 980 nm wavelength light, making him the best choice for a sensitizer ⁷⁴.

With Yb³⁺ a 980 nm laser is applied as the excitation source in order to match the ²F_{7/2} → ²F_{5/2} transition. The gap between ²F_{7/2} and ²F_{5/2} resonates with energy gaps of different excited states of ions normally used as activators, such as Er³⁺ ⁷⁴ (Figure 7).

1.3.4 Activators for UCNPs

Activators are characterized by having homogeneously distributed energy levels that have an energy separation equal to the sensitizer emission. When sensitizers are excited, activators obtain energy from the sensitizers close by in order to promote transitions to higher energy levels. Based on this Er³⁺, Tm³⁺ and Ho³⁺ are excellent activators, especially Er³⁺ and Tm³⁺, due to the cascading arrangement of their energy levels ⁸⁷.

Erbium ion (Er³⁺) has a higher upconversion efficiency due to the energy gaps of the first transition (⁴I_{15/2} to ⁴I_{11/2}) and the second transition (⁴I_{11/2} to ⁴F_{7/2})).

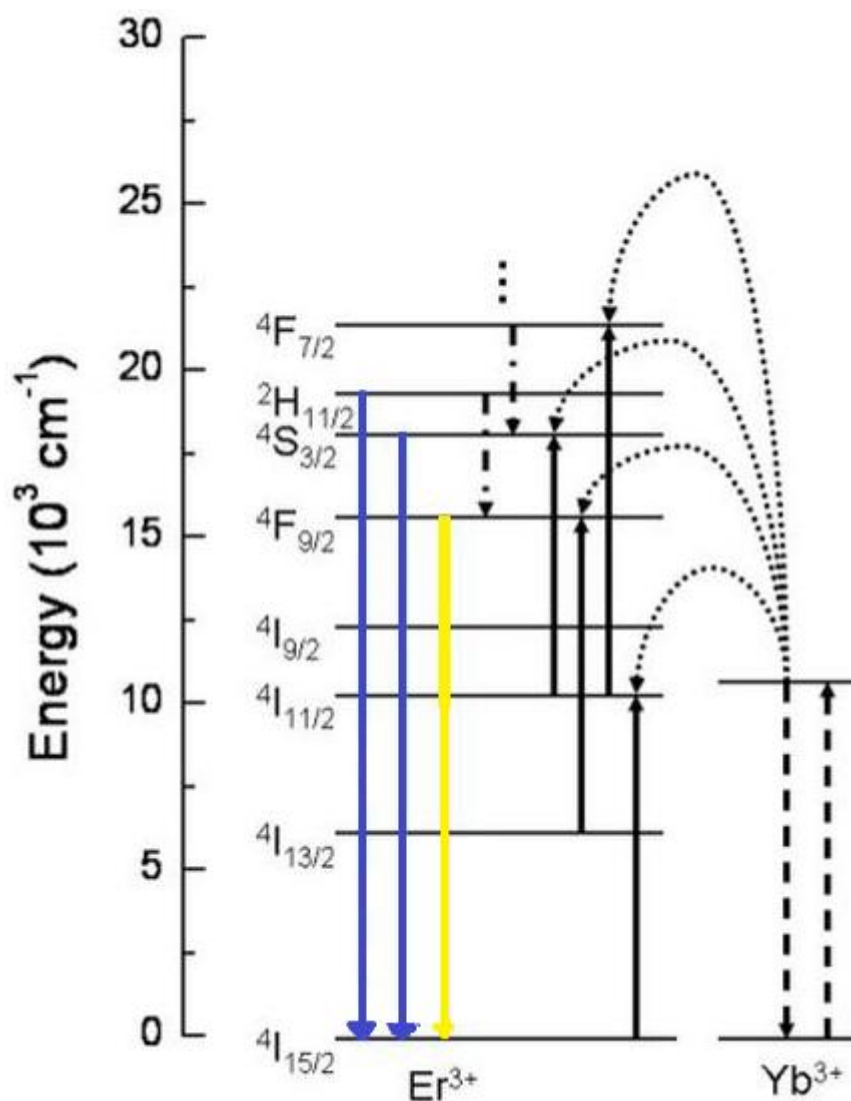


Figure 7: Energy level diagram for UCNPs in Yb³⁺/Er³⁺. Adapted from [145].

1.3.5 Synthesis of UCNPs

In order to use UCNPs for bioimaging and therapeutic applications, there is a need to have a controlled synthesis of monodispersed particles. There are several methods to achieve monodispersed UCNPs such as microwave^{88,89}, hydrothermal/solvothermal^{90,91}, thermal decomposition (thermolysis)^{92,93}, between others.

Of the three mentioned methods, thermal decomposition with metal trifluoroacetate as precursors is the most common route to synthesize monodispersed UCNPs of high quality. This method is based on the chemical

decomposition caused by heat and was first used to the synthesis of highly dispersed LaF₃ in 2005⁹⁴. In the following year, this approach was later modified and extended to the synthesis of NaYF₄ nanoparticles⁹².

Over the last few years, a new type of UCNPs has been developed, hollow UCNPs. This type of UCNPs can be used effectively has storage tools or delivery of therapeutic agents.

1.3.6 Bioapplications of UCNPs

Various optical imaging techniques make use of particles, such as organic dyes or semiconductor quantum dots, for biological applications, however, these particles require emission at longer wavelengths and excitation by high energy radiation in the UV or visible region^{95,96}. This type of wavelengths come with their own flaws when applied in biological tissues, because they have a lower penetration as a result of scattering processes and diverse absorption, low contrast because of background fluorescence, tissue damage due to phototoxicity of UV light and photobleaching restricting their use (in organic dyes)⁷⁴. Due to these limitations, there is a need to produce other optical imaging techniques that are able to overcome these limitations, granting a reliable and efficient optical bioimaging. Hence, UCNPs can overcome these obstacles, making them a very promising luminescent probe for biological applications⁹⁷⁻⁹⁹.

Upconversion nanoparticles main advantage lies in their capability of being excited by NIR photons and an emission range from UV light to NIR. Other than this advantage they also have an exceptional penetration depth *in vivo*, low cyto- and phototoxicity and a reduced background autofluorescence from the biological structures. Due to these reasons, UCNPs have shown a compelling potential for bioimaging and cancer therapy applications. In bioimaging, they have been used for *in vitro* imaging¹⁰⁰, *in vivo* imaging^{101,102}, and have been combined with other bioimaging techniques, such as positron emission tomography¹⁰³ and diffuse optical tomography¹⁰⁴. As for cancer therapy, the luminescence of these particles can be exploited for other applications, such as photodynamic therapy^{105,106}, drug delivery¹⁰⁷⁻¹⁰⁹ and photothermal therapy^{110,111}.

1.3.7 *In vivo* and *in vitro* bioimaging with UCNPs

Since the first synthesized, UCNPs have been applied in *in vitro* (cells) and *in vivo* (small animals) imaging. In *in vitro* imaging, cytotoxicity and water dispersion pose as the main obstacles to the effective use of UCNPs. Vetrone *et al.* used polyethyleneimine capped NaYF₄:Er³⁺, Yb³⁺ UCNPs to study intracellular imaging of HeLa cells, with the results showing a repositioning of UCNPs inside the cell with an increase in incubation time, showing a promising application for real-time imaging of cellular dynamics ¹¹². Cellular cytotoxicity and cellular imaging were tested by Shan *et al.*, by preparing hydrophobically bonded NaYF₄ UCNPs with amino and carboxyl groups, which showed low cytotoxicity and uptake confirmed by TEM image analysis on HOS cells ¹¹³.

Regarding *in vivo* imaging, Lim *et al.* first reported that *Caenorhabditis elegans* (roundworm) were injected with UCNPs. In this experiment, luminescence from the intestines of the worms was reported, with no cytotoxicity reported in a 24 hours period ¹¹⁴. In 2008, Zhang *et al.* used rats as models and subcutaneously injected UCNPs. Compared to the control group, which was injected subcutaneously with inorganic green-emitting quantum dots, the rats injected with UCNPs showed a stronger luminescence ¹¹⁵.

It is also possible to study multimodal imaging by combining UCNPs with other imaging techniques, such as ultrasound imaging, magnetic resonance imaging, positron emission tomography, computed tomography ^{116–118}.

1.3.8 Photodynamic therapy with UCNPs

Photodynamic therapy (PDT) is a clinical treatment where a light-sensitive drug, a photosensitizer, causes controlled cell death in the presence of light in a specific wavelength. In this technique, the photosensitizer is introduced in the cells, then it is excited by a light at a specific wavelength, which depends on the type of photosensitizer. After the excitation, energy is transferred to molecular oxygen in its surroundings, which generates reactive oxygen species (ROS) that can trigger cell death ^{119,120}.

Photodynamic therapy has some limitations, such as the need for UV or visible light to excite the photosensitizer. These types of light have low penetration on biological tissues and cause cellular damage, so in order to overcome this obstacle, UCNPs can be used.

Upconversion nanoparticles can be excited by near-infrared light and emit in the UV or visible photons, which can be used to activate photosensitizers and subsequently produce ROS leading to cell death. For this reason, the combination of UCNPs and PDT has gained attention.

Several studies of the combination of UCNPs and PDT have been done, most using NaYF₄:Er³⁺, Yb³⁺, with one photosensitizer, for example, Rose Bengal hexanoic acid ¹²¹, or more, when the absorption bands match different upconversion emissions of the UCNPs ¹⁰⁶. Zhang *et al.* ¹¹⁹ published the first work that combined UCNPs and PDT, that proved that this concept works. X. M. Liu *et al.* also worked with this combination, having a NaYF₄:Tm³⁺, Yb³⁺/Er³⁺, Yb³⁺ (core/shell) nanoparticle, which was used as a donor to monomalononic fullerene. Their results showed higher efficiency of this treatment ¹²².

1.3.9 UCNPs for drug delivery

Upconversion nanoparticles have also been applied to drug delivery therapy. At present time there are different nanostructures for drug delivery, such as nanoparticles, micelles, liposomes, *etc.* ¹²³. All the different structures have a size distribution between 10 and 100 nm, in order to prevent discharge by the reticuloendothelial system ¹²⁴, allowing them to be in the bloodstream for longer periods ¹²⁵.

The present time nanostructures can be improved by adding optical tagging, in order to see their distribution *in vivo* and have a better monitoring service of the treatment. Upconversion nanoparticles can achieve this, making them of great interest in drug delivery systems.

There are 3 main techniques of drug-delivering with the use of UCNPs:

- Porous shell to load the drug and slowly release it by diffusion or erosion of the shell. When UCNPs are used like this, they normally are

encapsulated in a mesoporous silica shell, where the drug is loaded by capillarity, and are slowly released.

- Hollow nanoparticles with a mesoporous silica shell, where the drug is loaded inside the nanoparticle. Hollow UCNPs release the drug as they are excited, normally having more cytotoxicity than drug delivery without nanoparticles.
- Upconversion nanoparticles can be coated with an amphiphilic polymer, which transfers hydrophobic UCNPs to their biological environment. In this case, drug delivery is normally pH-sensitive, making them more specific.

1.3.10 Photothermal therapy with UCNPs

Photothermal therapy (PTT) treatment works by inducing high temperature to kill cells ¹²⁶. This type of therapy has the advantage of selectivity towards cancer cells since tumour cells are more sensitive to heat than normal cells ¹²⁷.

In order to achieve high temperatures inside the cells, there has to be an efficient mechanism to transform the energy from a light source (normally a laser because of its monochromaticity and high power) by optical absorption.

This therapy normally uses noble metals, such as gold or silver, since they have a strong surface plasmon resonance absorption and light emission that releases heat. Upconversion nanoparticles are normally used with other nanoparticles for PTT, since they can be excited by NIR light and emit radiate in the wavelength that excites other nanoparticles, such as gold nanoparticles, to induce heat. This option has been explored by other groups, such as the work developed by Qian *et al.* that used gold decorated UCNPs ($\text{NaYF}_4:\text{Er}^{3+}, \text{Yb}^{3+}/\text{NaYF}_4/\text{silica}$ (core/shell/shell)), which showed that BE(2)-C cancer cells were killed after irradiation with a laser at 980 nm¹²⁸.

Besides combining UCNPs with other nanoparticles, several studies showed that UCNPs with a silica shell ($\text{NaYF}_4@\text{SiO}_2$) can also be used the source of heat. Carrasco *et al.* showed that neodymium ions (Nd^{3+}) coated UCNPs were able to irradiate a tumour from a mouse by increasing the temperature inside the tumour to around 47 °C ¹²⁹. The work by Zhang *et al.* ¹³⁰also showed an increase in

temperature of NaYF₄:Er³⁺/Yb³⁺ when irradiated with a 980 nm laser, revealing good potential for PTT.

With the possibility of using UCNPs alone to cause hyperthermia, and their ability to act both as the source of heat and as a sensor for imaging techniques, it makes them a prime candidate to further develop hyperthermia therapy, until it can be regularly applied in medicine.

1.4 Objectives

In this work, the main objective was to develop the groundwork to apply hyperthermia by UCNPs. To achieve this goal the specific aims were:

- Assess and analyse the physicochemical properties of two different types of upconversion nanoparticles;
- Verify the biocompatibility of upconversion nanoparticles in four different melanoma cell lines;
- Evaluate the sensitivity of melanoma cells to hyperthermia by studying the effects of different temperatures and exposure times.

2. Materials and methods

2.1 Upconversion Nanoparticles

In this work two different nanoparticles were used: NaYF₄:Yb/Er(20/2%)@mSiO₂ (NaYF₄UCNPs@mSiO₂), provided by Dr. Maysoon Saleh from Dr. Ute Resch-Genger group, from Federal Institute for Material Research testing (BAM), Berlin, Germany, and Gd₂O₃:Yb/Er (Gd₂O₃UCNPs) provided by Dr. Mengistie Debasu and Professor Luís Carlos from CICECO – Aveiro Institute of Materials, Department of Physics, Universidade de Aveiro.

2.2 Physicochemical characterization of UCNPs

Morphology and size were analysed by scanning transmission electron microscopy (STEM) using a 200 kV Hitachi HD-2700 (Hitachi High-Technologies Europe GmbH, Germany) STEM microscope equipped with energy-dispersive X-ray spectroscopy and secondary electron detectors. Samples for STEM analysis were prepared by evaporating 10 µL of nanoparticle solution on a grid.

The hydrodynamic diameter and polydispersity index (PDI) of the nanoparticles were measured by dynamic light scattering (DLS) and the zeta potential was assessed by electrophoretic mobility. Both measurements were performed using a Zetasizer Nano ZS (Malvern Instruments, UK), which uses multi-scatter laser diffraction to determine the size of the nanoparticles and its agglomeration.

Prior to any protocol, nanoparticles were dispersed by ultrasonic bath for 20 minutes.

For all the experiments UCNPs were dispersed in Mili-Q water and DMEM medium at 25 and 100 µg/mL, with 3 replicas each.

2.3 Cell culture

The MNT-1 cell line (a tumorigenic immortalized highly pigmented human melanoma cell line) was provided by Dr Manuela Gaspar (iMed.Ulisboa, Portugal),

while B16-F10 (a tumorigenic immortalized *Mus musculus* skin cell line) A375 (amelanotic human epithelial skin cell line) and SK-MEL-28 (a lightly pigmented human melanoma cell line) were purchased from ATCC. Cells were cultured in DMEM (Dulbecco's modified Eagle's medium) supplemented with 10% fetal bovine serum (FBS), 2mM-glutamine, 100 U/mL penicillin, 100 µg/mL streptomycin and 2.5 µg/mL fungizone, at 37 °C in 5% CO₂ humidified atmosphere.

Cells were observed daily to verify confluence, morphology and the presence of contaminants by an inverted phase-contrast microscope Nikon Eclipse TS100 (Tokyo, Japan). When the confluence of the culture reached 80%, the subculture was performed. Briefly, the old medium was removed, the cells were washed with PBS to remove any impurities. Afterwards, trypsin was added to detach the cells from the flask and was incubated for 5 minutes at 37 °C, 5% CO₂. After the incubation, medium was added to the flask, twice the volume of trypsin, in order to inactivate trypsin. Subsequently, the cells were counted in a haemocytometer and the desired number of cells were seeded into a new flask.

2.4 Viability assay

Cell viability of all the aforementioned cell lines was determined by the colourimetric WST-8, which is based on the bioreduction of tetrazolium salt WST-8 in the presence of an electron carrier, producing a water-soluble formazan. Cells were seeded in 96-well plates, at densities described in table 1, and cultured as previously described. Cell viability was assessed in cells exposed to UCNP dissolved in medium at 200, 100, 50, 25 and 12,5 µg/mL, previously dispersed in ultrasonic bath for 20 minutes. Cell viability was measured after 24 h and 48 h. The medium was removed and substituted by 110 microliters of WST-8 at 10% in medium, and incubated for 2 h at 37 °C, 5% CO₂.

Table 1 Cell seeding densities used on WST-8 assay

	24 HOURS	48 HOURS	72 HOURS
MNT-1	35000 cells/mL	25000 cells/mL	15000 cells/mL
B16-F10	25000 cells/mL	10000 cells/mL	5000 cells/mL
A375	35000 cells/mL	25000 cells/mL	20000 cells/mL
SK-MEL-28	50000 cells/mL	20000 cells/mL	10000 cells/mL

The optical density of WST-8 was measured at 450 nm in a microtiter plate reader (Synergy HT Multi-Mode, BioTek, Winooski, VT), and the cell viability was calculated as:

$$Cell\ viability = \left(\frac{(Sample\ Abs - Blank\ Abs)}{(Control\ Abs - Blank\ Abs)} \right) * 100$$

Two independent assays were performed with 4 replicates each, comparing the results with the control group (no exposure).

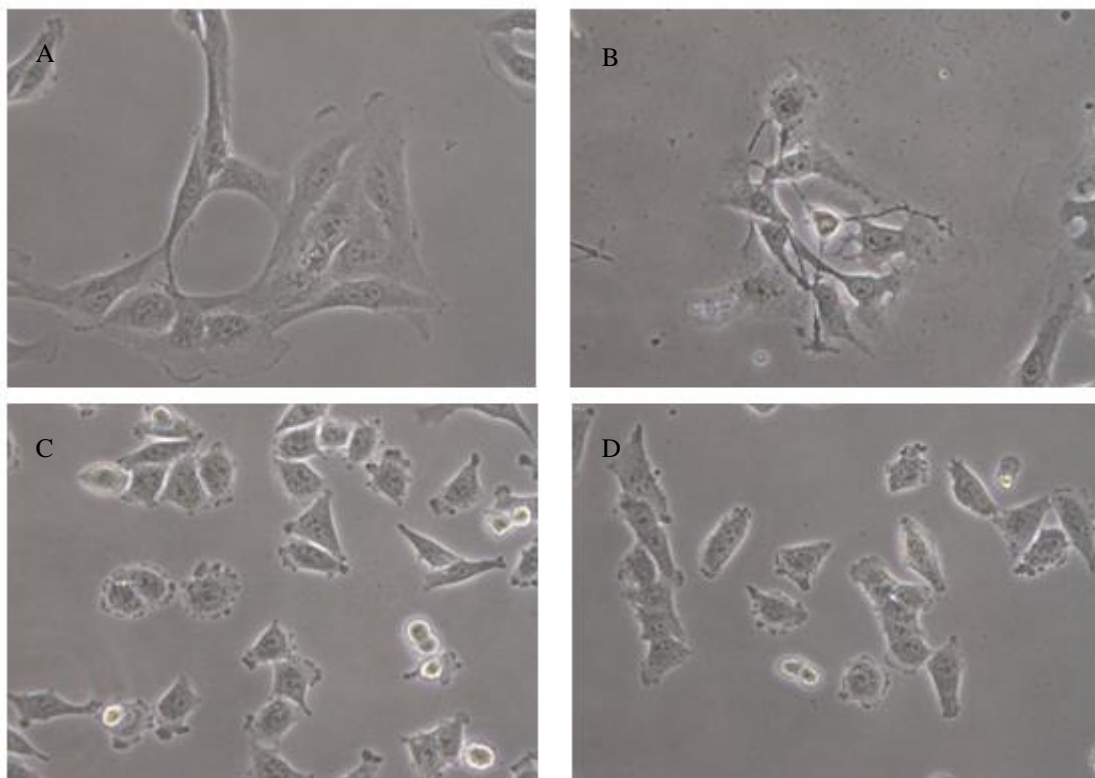


Figure 8: Light microscopy images of (A) MNT-1, (B) B16-F10, (C) A375 and (D) SK-MEL-28. Photos were taken with magnification of 20x and 10x combined.

2.5 Uptake potential by flow cytometry

Uptake potential of MNT-1, B16-F10, A375 and SK-MEL-28 cells, at the same concentration as the WST-8 assay, was determined by flow cytometry, as described by Suzuki *et al.* (2007). Briefly, cells were seeded in 24-well plates and after UCNP exposure they were trypsinized, collected to Eppendorf tubes and analysed by flow cytometry (FCM). Both parameters, forward scatter (FS), providing information on the particle's size, and side scatter (SS), providing information on the complexity of particles, were measured in an Attune[®] Acoustic Focusing Cytometer (TermoFisher Scientific) equipped with a 488 nm laser. For each sample, at least 100000 cells were analysed.

2.6 Effect of hyperthermia on cell viability

To evaluate the effect of temperature, cell viability was assessed by the colourimetric 3-(4,5-dimethyl-2-thiazolyl)-2,5-diphenyl tetrazolium bromide (MTT) assay, which is based on the bioreduction of the tetrazolium dye MTT by oxireductases enzymes, producing insoluble formazan. Briefly, cells were seeded in 96-well plates and cultured as described previously in WST-8 viability assay, where instead of adding UCNPs, the temperature effects were assessed in cells exposed to 43 and 45 °C for 30, 60 and 120 minutes. Cell viability was measured after 24 h, 48 h and 72 h. The medium was removed and substituted by 50 microliters of MTT and incubated for 4 hours at 37 °C, 5% CO₂. Then, MTT was removed and 150 microliters of dimethyl sulfoxide (DMSO) were placed in each well and were agitated in the dark for 2 hours.

The optical density of MTT was measured at 570 nm in a microtiter plate reader (Synergy HT Multi-Mode, BioTek, Vinooski, VT), and the cell viability was calculated as:

$$Cell\ viability = \left(\frac{(Sample\ Abs - Blank\ Abs)}{(Control\ Abs - Blank\ Abs)} \right) * 100$$

Results were compared with the control group (without exposure).

2.7 Statistical analysis

The results are reported as mean \pm standard deviation (SD) of 4 technical replicates in each of the 2 independent experiments. For all assays one-way ANOVA was performed, followed by Dunnet and Dunn's tests (as a parametric and non-parametric test, respectively), using Sigma Plot 11.0 software (Systat Software Inc.). The differences were considered statistically significant for $p < 0,05$.

3. Results

3.1 Physicochemical characterisation

3.1.1 Dynamic light scattering of Gd₂O₃UCNPS

Physicochemical characterization by DLS of Gd₂O₃UCNPs, after dispersion in ultrasonic bath for 20 minutes showed that in water the average hydrodynamic diameter (Dh) was near 1840 nm, while in DMEM medium the Dh values were considerably lower, respectively 306 nm and 560 nm for the suspensions of 25 µg/ml and 100 µg/ml (Figure 9).

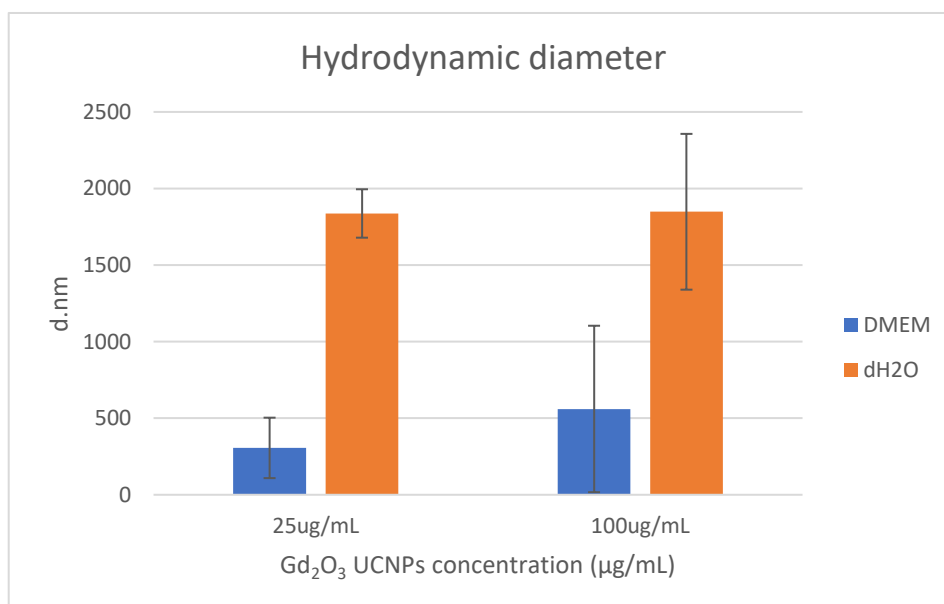


Figure 9: Hydrodynamic diameter results of Gd₂O₃UCNPs. Results are shown as average ± standard deviation (SD).

Hydrodynamic diameter results (Figure 9) showed higher values than expected, being the dispersion in dH₂O the highest. These results might be explained by agglomeration of the nanoparticles.

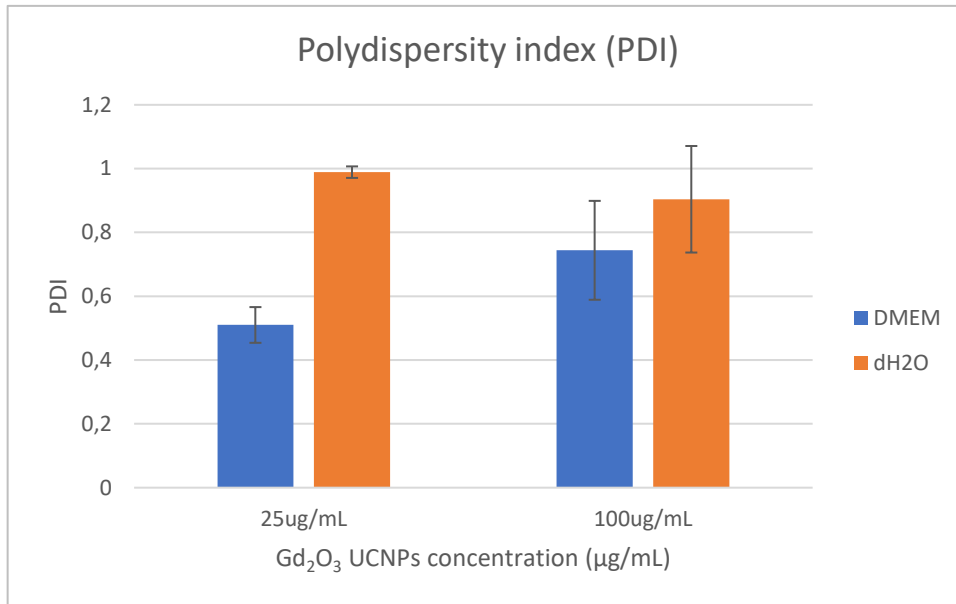


Figure 10: Polydispersity index results of Gd₂O₃UCNPs. Results are shown as average ± standard deviation (SD).

The values of polydispersity index (PDI) for Gd₂O₃UCNPs (Figure 10) in water were 0.99 and 0.90, respectively for 25 and 100 µg/ml suspensions. In DMEM medium PDI values were lower than in water, being respectively 0.51 and 0.74 for the 25 and 100 µg/ml suspensions .

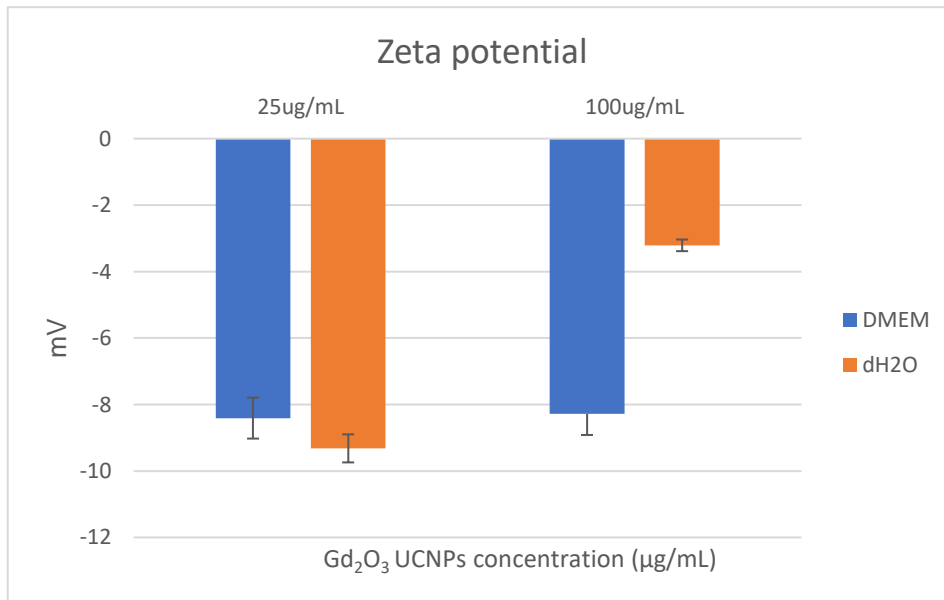


Figure 11: Zeta potential results of Gd₂O₃UCNPs. Results are shown as average ± standard deviation (SD).

Zeta potential values of Gd₂O₃UCNPs nanoparticles (Figure 11) in dH₂O were -9.32 and -3.21 for 25 and 100 µg/mL, respectively. In DMEM medium zeta potential was more uniform, with -8.41 for 25 µg/mL and -8.28 µg/mL.

3.1.2 Dynamic light scattering of NaYF₄UCNPs@mSiO₂

Physicochemical characterization by DLS of NaYF₄UCNPs@mSiO₂, after dispersion in ultrasonic bath for 20 minutes.

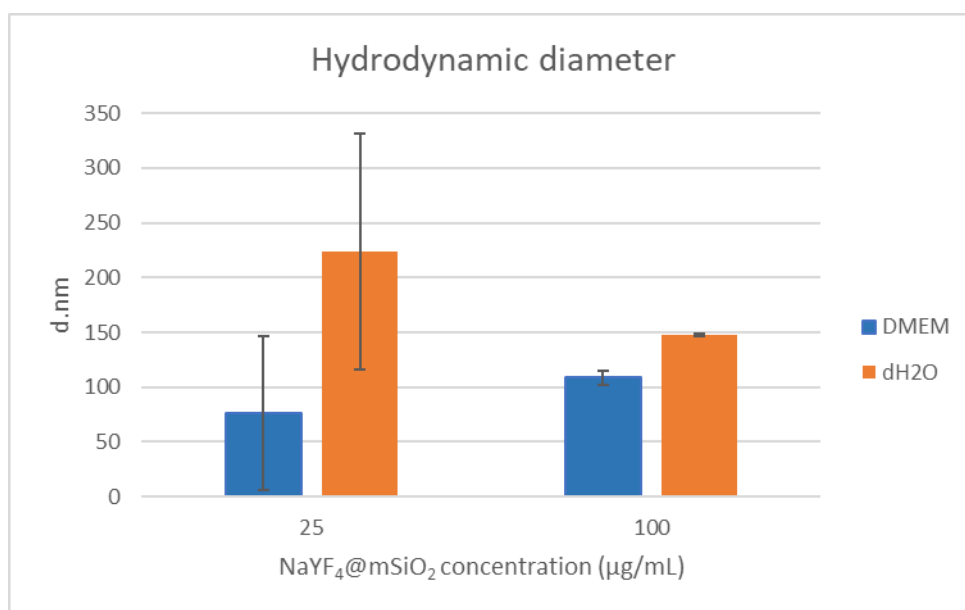


Figure 12: Hydrodynamic diameter results of NaYF₄UCNPs@mSiO₂. Results are shown as average ± standard deviation (SD).

The average hydrodynamic diameter (Dh) for NaYF₄UCNPs@mSiO₂ was 76.68 and 108.7 nm in DMEM medium, for 25 and 100 µg/mL respectively (Figure 12).

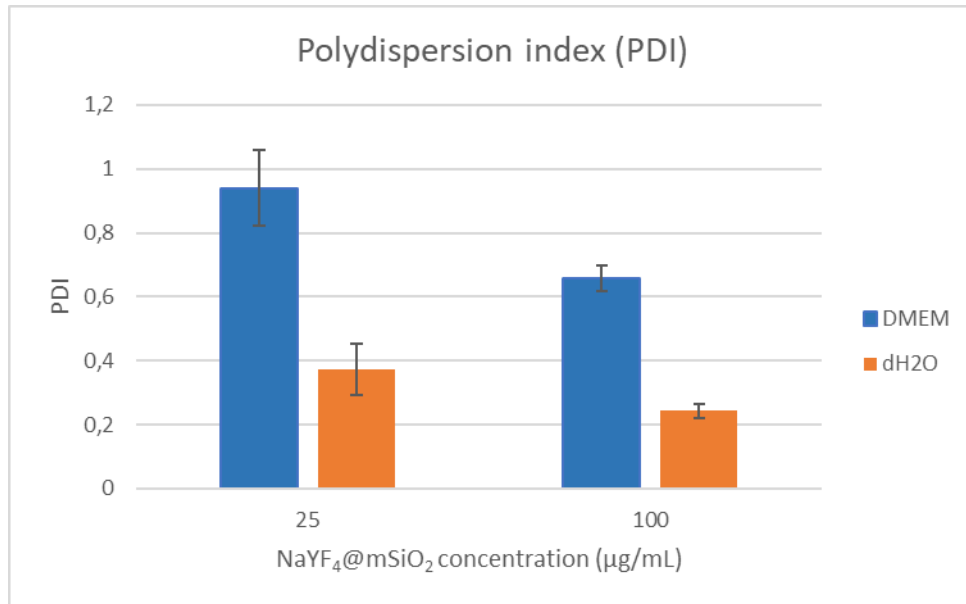


Figure 13: Polydispersity index results of NaYF₄UCNPs@mSiO₂. Results are shown as average ± standard deviation (SD).

Polidispersity index values for NaYF₄UCNPs@mSiO₂ (Figure 13) in water were 0.373 and 0.244 for 25 and 100 µg/mL, respectively. In DMEM medium PDI values were 0.94 for 25 µg/mL and 0.658 for 100 µg/mL.

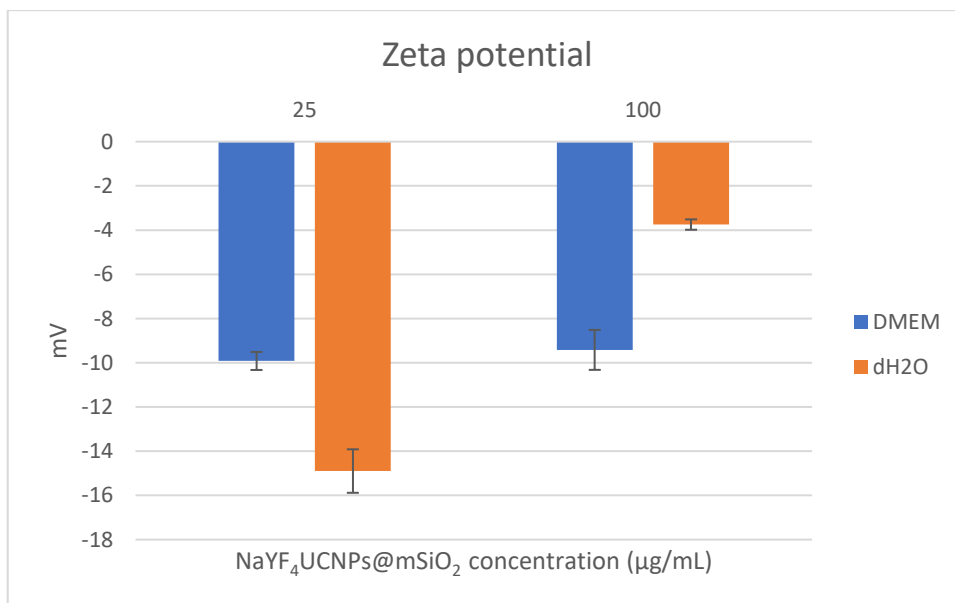


Figure 14: Zeta potential results of NaYF₄UCNPs@mSiO₂. Results are shown as average ± standard deviation (SD).

Zeta potential for NaYF₄UCNPs@mSiO₂ nanoparticles (Figure 14) in dH₂O were -14.9 and -3.75 for 25 and 100 µg/mL, respectively. In DMEM medium zeta potential was more uniform, with -9.92 for 25 µg/mL and -9.42 for µg/mL.

3.1.3 STEM analysis of NaYF₄UCNPs@mSiO₂

Size and morphology of NaYF₄UCNPs@mSiO₂ was analysed by scanning transmission electron microscopy. As observed in Figure 15 NaYF₄UCNPs@mSiO₂ showed monodispersed and uniform spherical morphologies, being each core well encapsulated with a silica shell (Figure 15 B and C). Their average size was 77.88 ± 3.53 nm.

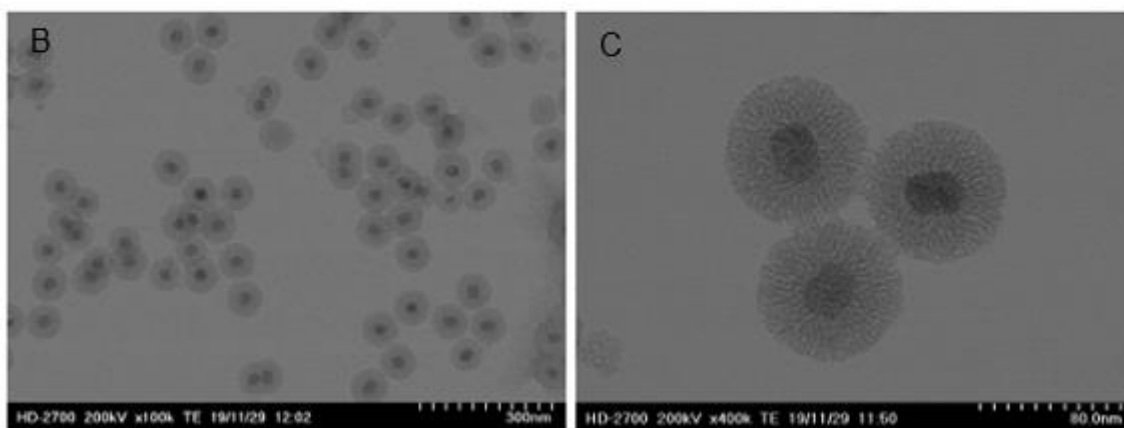
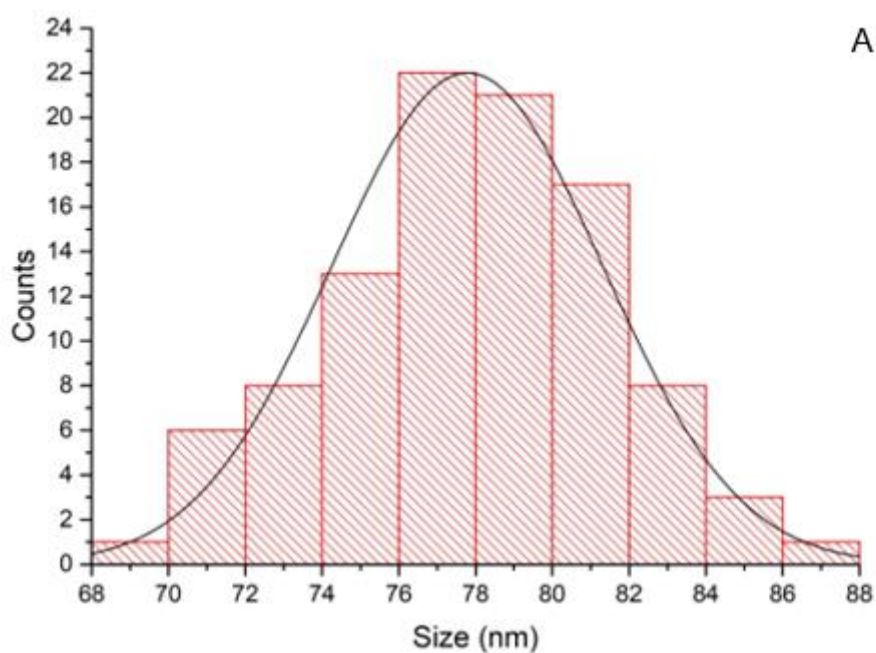


Figure 15: (A) Particle size distribution of NaYF₄UCNPs@mSiO₂ nanoparticles determined via image analysis with ImageJ, with an average nanoparticle size of 77.78 ± 3.53 nm. STEM images of NaYF₄@mSiO₂ nanoparticles at two different ampliations: (B) 100,000x and (C) 400,000x.

3.2 Cell viability

Cell viability was performed by WST-8 assay on 2 different nanoparticles: Gd₂O₃UCNPS, NaYF₄UCNPs@mSiO₂.

3.2.1 Gd₂O₃UCNPS

Gadolinium oxide UCNPs were tested on MNT-1 and A375 cell lines, for 24 and 48 hours (Figure 16).

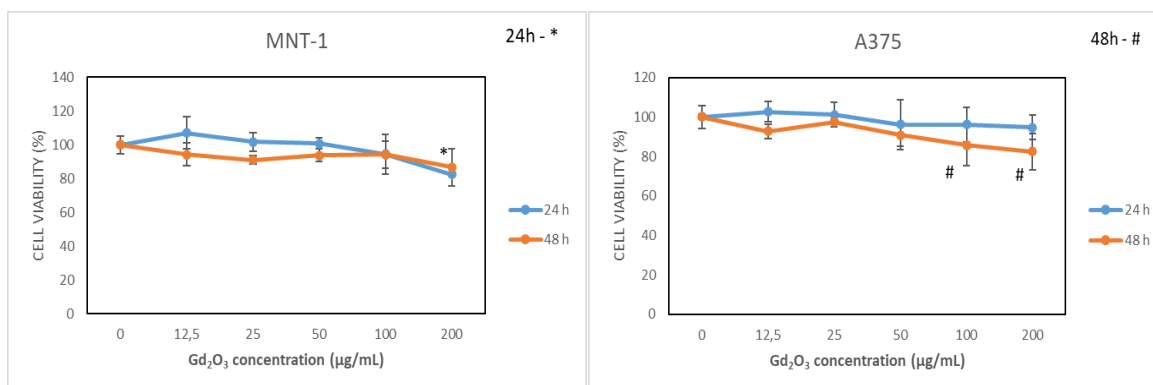


Figure 16: Cell viability by WST-8 protocol on MNT-1 (left) and A375 (right) cells, when exposed to Gd₂O₃ for 24 and 48 hours. Results are shown as average \pm standard deviation (SD). * or # indicates significant statistical difference in relation to the control condition ($p < 0.005$).

For all tested concentrations of Gd₂O₃UCNPs MNT-1 cell viability was above 90% with no statistical difference, showing that this nanoparticles do not affect the cell's viability. For A375 cells, viability 24 hours after exposure showed no statistical differences, but after 48 hours there was a significant decrease at 100 and 200 µg/mL, showing that these concentrations of nanoparticles result in a significant viability decrease.

3.2.2 NaYF₄UCNPs@mSiO₂

The nanoparticle NaYF₄UCNPs@mSiO₂ effect on viability was tested on 4 different cell lines: MNT-1, B16-F10, A375 and SK-MEL-28 (Figure 17).

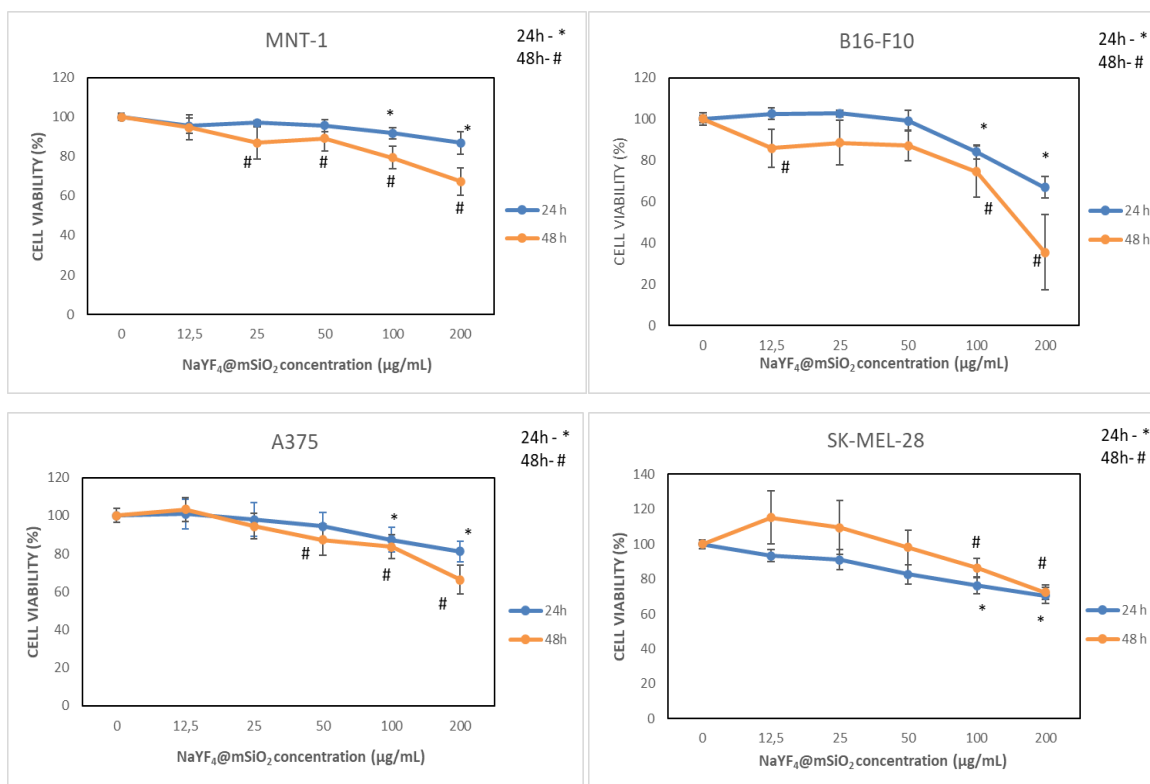


Figure 17: Cell viability by WST-8 protocol of MNT-1 (top left), B16-F10 (top right), A375 (bottom left) and SK-MEL-28 (bottom right) cells, when exposed to NaYF₄UCNPs@mSiO₂ for 24 and 48 hours. Results are shown as average \pm standard deviation (SD). * or # indicates significant statistical difference in relation to the control condition ($p < 0.005$).

On all cases there is a significant decrease in viability at 100 and 200 µg/mL of NaYF₄UCNPs@mSiO₂, implying that these concentrations are too high for further tests with these nanoparticles.

MNT-1 cells also showed that even a lower concentration of 25 µg/mL for 48 hours has a negative impact in cellular viability, suggesting either the use of low concentrations of nanoparticle for 48 hours exposure or a smaller exposure time. A375 cells show a similar profile as the MNT-1 cells, except that at 48 hours of exposure there was an only significant decrease in viability at 50 µg/mL and above.

3.3 Uptake

Uptake was measured by flow cytometry of the 4 cell lines used in this work, with NaYF₄UCNPs@mSiO₂ nanoparticles at 2 different concentrations, 25 and 100 µg/mL, and a control group that had no nanoparticles in the medium (Figure 18).

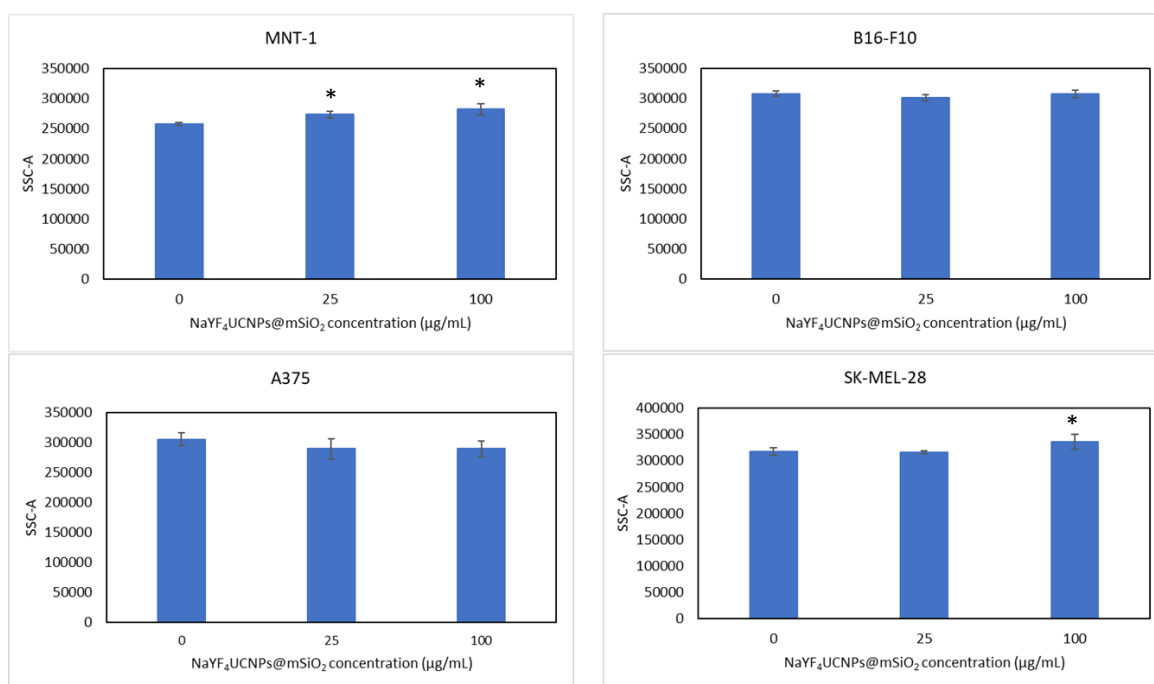


Figure 18: Cell uptake results of MNT-1 (top left), B16-F10 (top right), A375 (bottom left) and SK-MEL-28 (bottom right) cells, when exposed to 25 and 100 µg/mL of NaYF₄UCNPs@mSiO₂ for 24 hours. Results are shown as average ± standard deviation (SD). * indicates significant statistical difference in relation to the control condition (p<0.005).

The uptake results of SK-MEL-28 were transformed by square root to pass the normality test. This assay only showed a significant difference in MNT-1 cells that had more uptake when exposed to the nanoparticle in both concentrations, and SK-MEL-28, which only had a significant difference when exposed to 100 µg/mL.

3.4 Temperature exposure

The effects of heat on cells were measured by MTT cell viability assays after temperature exposure (Figure 19).

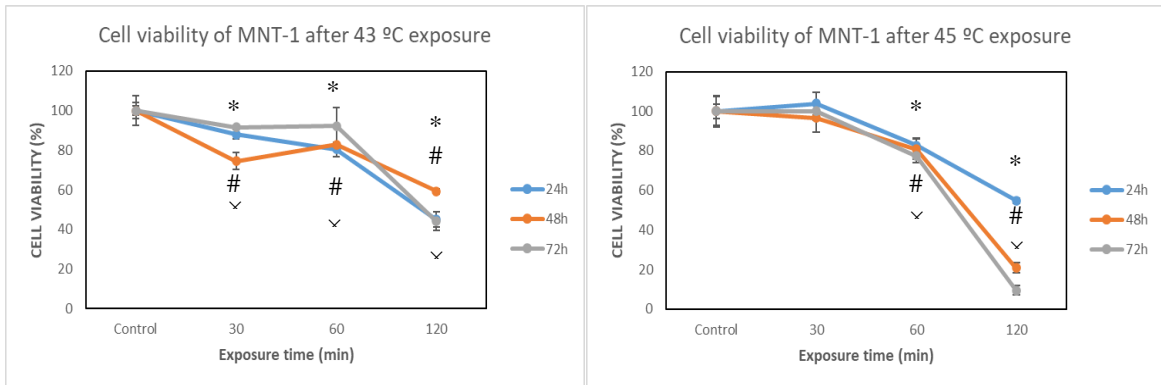


Figure 19: Cell viability by MTT protocol of MNT-1 cells when exposed to 43 °C (left graph) and 45 °C (right graph), for 30, 60, 90 and 120 minutes, 24 hours, 48 hours, and 72 hours after exposure. Results are shown as averages \pm standard deviation (SD). * or # indicates significant statistical difference in relation to the control condition ($p < 0.005$).

In these assays, we can see that a long exposure of temperature to MNT-1 cells causes lower cellular viability independently of how much time it has passed since the exposure, as shown by the low cellular viability of cells exposed for 120 minutes.

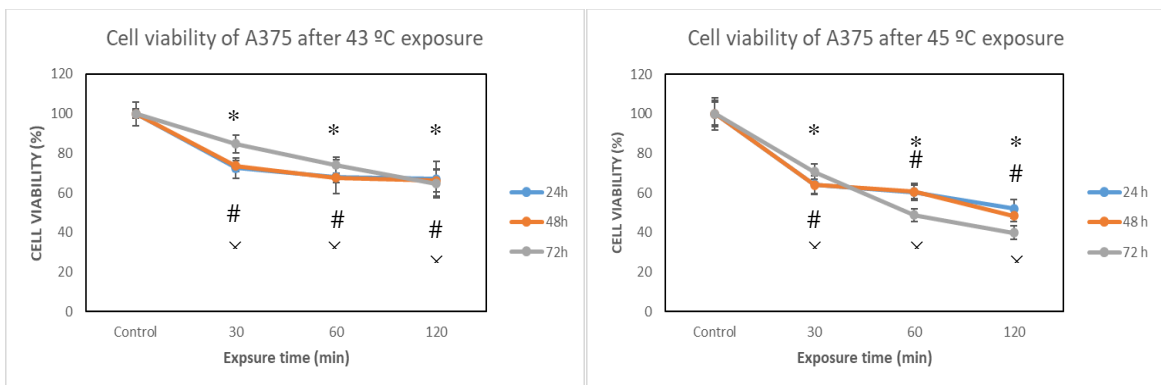


Figure 20: Cell viability by MTT protocol of A375 cells when exposed to 43 °C (left graph) and 45 °C (right graph), for 30, 60, 90 and 120 minutes, 24 hours, 48 hours, and 72 hours after exposure. Results are shown as averages \pm standard deviation (SD). *, # or x indicates significant statistical difference in relation to the control condition ($p < 0.005$).

A375 cells showed a uniform response in each temperature exposure, independently of the time of exposure. Cell viability, when exposed to 43 °C, did not drop below 66%, while cell viability of cells exposed to 45 °C showed lower results, with the lowest value of 39%, 72 hours after exposure for 120 minutes.

4. Discussion

This work assessed physicochemical properties of Gd₂O₃UCNPs and NaYF₄UCNPs@mSiO₂, and their dark cytotoxicity on melanoma cell lines. The effect of high temperature (43 and 45 °C) on melanoma cell lines was also studied.

The physicochemical characterization of Gd₂O₃UCNPs showed overall high values of hydrodynamic diameter in both DMEM and dH₂O solutions, with dH₂O solution being the highest. Such high values can be the result of the agglomeration of these nanoparticles. This hypothesis is corroborated by the high PDI values, which shows high differences in size measurements of the nanoparticles. Furthermore, Zeta potential had values lower than -10 mV, which indicates the instability of the nanoparticles and tendency to aggregate ¹³¹. These results show that the dispersion protocol used, ultrasonic bath for 20 minute, may not be appropriate to disperse correctly these nanoparticles.

NaYF₄UCNPs@mSiO₂ nanoparticles showed good values of hydrodynamic diameter in DMEM medium, while dH₂O medium had higher values in the tested concentrations, 25 and 100 µg/mL. Polydispersity index showed low values in dH₂O medium, which indicates that there was more uniformity in the sizes measured in the sample. On the other hand, DMEM medium values were higher, but this increase can be explained by the fact that this medium has different free-floating constituents, such as proteins, which increase the PDI reading. Zeta potential results showed that nanoparticles have electrostatic repulsion, but they were not fully stable ¹³¹. Similar results concerning hydrodynamic diameter, PDI and Zeta potential, were obtained on similar nanoparticles, NaYF₄ nanoparticles, by Bazyńliska *et al.*, 2017 ¹³².

Scanning transmission electron microscopy showed that NaYF₄UCNPs@mSiO₂ had a size of 77.78 ± 3.53 nm, which are similar to the results reported by Li *et al.*, 2012 ¹³³ in which NaYF₄UCNPs@mSiO₂ nanoparticles had a size around 80 nm. The hydrodynamic diameter of NaYF₄UCNPs@mSiO₂ nanoparticles was similar to the size determined by STEM at 25 µg/mL, with the hydrodynamic diameter at 100 µg/mL being slightly higher.

Cell viability assay of melanoma cells exposed to Gd₂O₃UCNPs showed a significant decrease of MNT-1 cells viability at 200 µg/mL at 24 hours after exposure.

A375 cells viability significantly decreased after exposure to 100 and 200 $\mu\text{g/mL}$ during 48 hours. Even though there was a significant decrease in viability, it was maintained above 80% in all cases indicating a overall low cytotoxicity of $\text{Gd}_2\text{O}_3\text{UCNPs}$. Results obtained by *Li et al.*, 2015 reported a decrease in the viability of HeLa cells when exposed to $\text{Gd}_2\text{O}_3\text{UCNPs}$ starting at a concentration of 281.25 $\mu\text{g/mL}$, but maintaining cell viability at 80%¹³⁴. Even though the work conducted by *Li et al.*, 2015 used higher concentrations of $\text{Gd}_2\text{O}_3\text{UCNPs}$ the same overall profile can be seen, and this difference can be caused by different sensitivity of MNT-1 and A375 cells comparatively to HeLa cells¹³⁴.

Viability assays of $\text{NaYF}_4\text{UCNPs}@m\text{SiO}_2$ nanoparticles on the four melanoma cell lines were used to test the cytotoxicity of the nanoparticles, in order to discover at what concentration they can be used without harming the cells. In all four cell lines, there was no significant effect on the viability at 50 $\mu\text{g/mL}$, 24 hours after exposure. Hence, nanoparticles at this concentration can be used without causing a significantly negative effect on the cells. Results obtained by other studies with UCNPs applied on different cell lines also showed biocompatibility for concentrations at and below 50 $\mu\text{g/mL}$ ^{135,136}. However, MNT-1 and A375 cells at 48 hours after exposure to 50 $\mu\text{g/mL}$ of $\text{NaYF}_4\text{UCNPs}@m\text{SiO}_2$, showed a significant decrease in viability, with MNT-1 even showing a negative effect on viability with 25 $\mu\text{g/mL}$ of $\text{NaYF}_4\text{UCNPs}@m\text{SiO}_2$. The results obtained demonstrate that for these two cell lines, the prolonged exposure to $\text{NaYF}_4\text{UCNPs}@m\text{SiO}_2$ at these concentrations has a negative impact on the viability.

Cell uptake assay was performed by flow cytometry, which compares the side scattering of the cells, indicating their granular complexity¹³⁷. Based on this principle, a higher side scattering indicates the presence of nanoparticles inside the cell¹³⁷. The results showed a significant increase of MNT-1 cells side scatter after exposure to both UCNPs concentrations used (25 and 100 $\mu\text{g/mL}$). SK-MEL-28 cells increased the side scatter only after exposure to the highest concentration, while A375 and B16-F10 cells did not show any differences. Even though uptake by flow cytometry is not as reliable as microscopic techniques, it serves as a preliminary result, indicating that at least MNT-1 and B16-F10 cells are able to uptake $\text{NaYF}_4\text{UCNPs}@m\text{SiO}_2$. The uptake of nanoparticles might be restricted if a negative

charge is present on the surface of the nanoparticles, which can cause an electrostatic repulsion with the cell membrane ¹³⁸⁻¹⁴⁰. Cellular uptake of UCNPs might be increased by modifications of the surface, for example by adding folic acid or poly(ethylene glycol), as reported in several studies ^{100,109}.

Temperature assays showed that hyperthermia causes cellular death in a time and temperature-dependent manner, with cell viability of MNT-1 cells being lower within higher exposure time and temperature (120 min at 45 °C). MNT-1 cells exposed to 43 °C showed similar viability responses, independently of the time after exposure. Such might be explained by the fact that temperature of 43 °C causes reversible cellular damage ¹⁴¹, and that the cells 24 hours after exposure did not have enough time to revert the damages, showing similar cell viability as the cells exposed to 45 °C. MNT-1 cells when exposed to 45 °C showed a similar profile of response, with the only difference being on the exposure time of 120 minutes, where 24 hours after exposure present a higher viability than the other two cases. This might be explained by the fact that cells were still in the process of cell death, and as such 24 hours after exposure there was a higher viability than 48 and 72 hours after exposure. Similar results were obtained in other studies, such as Shellman *et al.* 2008 ⁵⁶ which reported a decrease in cell viability of A375 cells when exposed to 45 °C. A375 cells exposed to 43 °C showed a similar pattern of cell viability, independently of how much time it had passed since the exposure, with cell viability not dropping below 60% in all cases. Similar results were reported by Shellman *et al.* 2008 ⁵⁶, in which viability of A375 cells dropped to around 40% when exposed to 45 °C, but it did not suffer alterations when exposed to 41 °C, and as such, the temperature of 43 °C yielded results in between the two temperatures reported. The same A375 cells when exposed to 45 °C showed lower cell viability, as expected, with cell viability 24 and 48 hours after exposure showing similar results, never dropping below 50%. However, cell viability 72 hours after exposure had lower cell viability in every period tested, with the lowest being 39% at 120 minutes of exposure. Cell viability results 24 hours after exposure, showed lower viability at 30 and 60 minutes exposure than the results reported by Shellman *et al.* 2008 ⁵⁶, but the viability after 90 minutes was around 40%, however in this work showed the same viability at 120 min. Even though there are differences between the results

obtained in this work and Shellman *et al.* 2008⁵⁶, they share a similar profile, and these differences can be explained by the different effect hyperthermia has on the same cells in different phases for the cell cycle⁴⁹.

Even though the mechanism of cell death was not explored in this thesis, several studies have shown that it mostly happens by apoptosis at a temperature of 43 °C^{56,142}, and necrosis with higher temperature, such as 45 °C^{49,143}.

5. Final remarks

Cancer is one of the most common diseases and one with a higher mortality rate, and as such, there is a need to create new approaches to treat it. Hyperthermia has long been studied as a therapeutic treatment for cancer since heat affects the biochemical process of the tumour cells leading to their death. However, this therapeutic technique has its limitations, such as the inability to apply heat only to tumour cells.

Nanoparticles have been revolutionary in the sciences, and their use spreads throughout all branches. Many advances have been made in this area, and in present times upconversion nanoparticles have garnered a lot of interest for their special characteristics, such as their anti-stokes shift. These nanoparticles, have been applied to generate hyperthermia. They offer the ability to provoke hyperthermia from the inside-out and to target cancer cells with the possibility of all this being done noninvasively by lasers.

In this work, we established a range of concentrations of $\text{NaYF}_4@m\text{SiO}_2$ that is possible to use without affecting the viability of the four different cell lines used, and also a range of concentration of $\text{Gd}_2\text{O}_3\text{UCNPS}$ for MNT-1 and A375 cell lines. The aim of this work was also to study the response profile of melanoma cell lines to hyperthermia which was obtained for MNT-1 and A375 cell lines after exposure to 43 °C and 45 °C for different times. In this study it was possible to see that viability drops with time of exposure, and that higher temperatures induced lower viability.

The conditions defined in the current work (UCNPs concentrations and temperatures) can be replicated in the future in order to generate near-infrared light triggered hyperthermia in melanoma cells using UCNPs.

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