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EXTRACTION AND IDENTIFICATION OF MAJOR POLYPHENOL CONSTITUENTS OF NORTHERN HUNGARIAN HORSEMINT (MENTHA LONGIFOLIA L. (L.))

Katalin Patonay¹*, Orsolya Szabó-Hudák¹, Helga Szalontai¹, Mihály Jánószky¹, Erika Pénzesné Kónya² & Éva Zámboriné Németh³

¹Food and Wine Research Institute, Eszterházy Károly University, 6G Leányka u., H-3300, Eger, Hungary; ²Department of Botany and Plant Physiology, Eszterházy Károly University, 6D Leányka u., H-3300, Eger, Hungary; ³Department of Medicinal and Aromatic Plants, Szent István University, 29-43 Villányi u., H-1118, Budapest, Hungary; *E-mail: patonay.katalin@uni-eszterhazy.hu

Abstract: (*Mentha longifolia* (L.) L.) is a less studied species from the viewpoint of its nonvolatile constituents and this work is the first phytochemical screening of it in Hungary. The aim of the experiments is to elaborate an efficient extraction method of its polyphenolic antioxidants. Thirty-six accessions were sampled. Soxhlet and an ultrasonic extraction were applied, both with methanol (MeOH) and ethanol:water 7:3 (WA). The analysis of the extracts was carried out by HPLC-DAD method. Rosmarinic acid, hesperidin, diosmin and cynaroside were observed to be common constituents of all samples. Other major constituents were observed in all of the chromatograms at ca. t_R=13.5 min. and t_R=19.7 min (unknown A and B). Their identification needs further examination. Regarding antioxidant activity, WA extraction was more efficient than MeOH Soxhlet extraction.

Keywords: *Mentha longifolia*, horsemint, polyphenolic antioxidants, HPLC, antioxidant activity

INTRODUCTION

Horsemint, *Mentha longifolia* (L.) L., *Lamiaceae* (further: ML) is a wild-growing perennial species. It has a large distribution area in Eurasia and non-tropical Africa (Tucker and Naczi 2007), grows in multiple types of habitats such as wet meadows, roadsides and urban areas (Başer *et al.* 1999). Active ingredients of ML originate from the mevalonate and shikimate pathways. The former produces the volatile terpenes of ML, most frequently ketones as carvone, piperitone, piperitenone and corresponding epoxides

(Başer *et al.* 1999; Sharopov *et al.* 2012). Triterpenes in ML (ursolic, oleanolic acids etc.) also originate from this pathway. The shikimate pathway is the origin of phenolics of ML. Caffeic acid derivatives (predominantly the known antiradical molecule rosmarinic acid) are present. But flavonoids also deserve attention: the presence of 55 different flavonoids was reported (Bourwieg and Pohl 1973; Ghoulami *et al.* 2001; Pereira and Cardoso 2013 Hawrył *et al.* 2016, Patonay *et al.* 2017; Park *et al.* 2019), six of them *de novo* (Sharaf *et al.* 1999; Jahan *et al.* 2001; Ghoulami *et al.* 2001; Ali *et al.* 2002). Regarding flavonoid subclasses, flavones seem to be present in the largest variability.

There are several studies dealing with the phytochemistry of horsemint, although the majority of them deals with volatile terpenoids. Studies focusing to polyphenolic constituents most frequently evaluate a single sample of the species (e. g. Bourwieg and Pohl 1973; Sharaf et al. 1999; Jahan et al. 2001; Ghoulami et al. 2001; Ali et al. 2002; Fialová et al. 2008; Krzyzanowska et al. 2011; Shen *et al.* 2011: Stanislavliević *et al.* 2012: Benedec *et al.* 2013: Bahadori et al. 2018; Park et al. 2019). Relevant studies on a larger representative sample pool dealing with flavonoid composition of horsemint are not available, in contrary to e.g. peppermint (Guédon and Pasquier 1994). For the major nonflavonoid phenolics of horsemint, one single large-sample study has been found (Dudai et al. 2006) providing the rosmarinic and caffeic acid contents of 40 different clones. Therefore, it can be established, that conclusions the distribution pattern of polyphenols in horsemint on populations and on the factors influencing their accumulation would be difficult. Additionally, a part of the former studies does not provide a detailed description of the extraction method applied, making the comparison of data difficult (e. g. Ghoulami *et al.* 2001; Jahan et al. 2001; Benedec et al. 2013). Thus, contrary to the abundance of this species, knowledge on its phytochemistry, especially in case of the polyphenolic compounds, still seems to be incomplete. The present study on the extractability and polyphenol composition of the Northern Hungarian ML populations has been carried out in the framework of a wider project which intends to utilize this species as a cultivated source of antioxidant and/or preservative preparations in the food industry. Thus, one step is to define an effective extraction method of horsemint resulting in the highest antioxidant capacity and polyphenol content of the

samples. In parallel, screening the polyphenol composition supplements our knowledge on this species and may help in elucidating the background of the antioxidant properties.

MATERIALS AND METHODS

Sampling: Thirty-six shoot samples of ML were collected in the flowering stage, from the end of June to the beginning of October, 2016. Populations in the Bükk Mountains, Mátra Mountains, Heves-Borsodi Hills and Eger were surveyed. Identification: one item (ID: NOSZ) by the collector Jana Táborská (Eszterházy Károly University, Department of Botany and Plant Physiology) ten items (JOF, HV1, HV2, HV3, HOV1, HOV2, HOV3, KÜH, JÁV and EGR4) by the collector Helga Szalontai and other samples: by Erika Pénzesné Kónya. The identification was performed on the basis of macromorphology, using the manual of the vascular flora of Hungary (Simon 1994). Voucher specimen deposition: at the Eszterházy Károly University in the chemotaxonomical collection of the herbarium (EGR-CH). Samples were hung in a shady place for 21 days. After drying, leaves and inflorescences were separated from stems, collected and deposited in the freezer.

Extraction. Extraction in a cooled ultrasonic bath (45 kHz, 3 stages, 45 min/stage, at maximum 30°C) was applied. Sonication may be expected to be milder toward many plant polyphenols of heat sensitivity as it can be performed at room temperature and normal pressure. Another advantageous traits of this method are the much lesser time demand and moderate solvent demand in comparison with Soxhlet extraction or the conventional room temperature maceration of plant material (Medina-Torres et al. 2017). Furthermore, for rosmarinic and carnosic acids it was observed that the sonication gives better yield of them in comparison with conventional room-temperature maceration of rosemary leaves, both in case of methanolic and ethanolic extractions (Paniwnyk et al. 2009) and the antioxidant activity of extracts gained by ultrasonic treatments is higher than the values of extracts prepared by maceration. A comparative study on extraction of polyphenols of *Tilia cordata* inflorescences established that the sonication was the most efficient method to extract phenolic acids of *Tiliae flos* and the second efficient method to gain flavonol constituents of it when sonication, microwave

extraction, accelerated solvent extraction, heat reflux extraction, and Soxhlet were compared (Oniszczuk *et al.* 2015)

In parallel, the Soxhlet method was applied as a reference method, as nowadays it is still frequently used to extract different plant materials (e. g. Ahmed et al. 2015) including horsemint (e. g. Güllüce et al. 2007; Iqbal et al. 2013) and in comparative investigations on extractability of some plants (Karabegović et al. 2011). The solvents were methanol (MeOH) and 7:3 ethanol: water mixture (WA). Methanol (MeOH) was taken as a reference, as for this species a significant part of investigators use it when polar extracts are prepared (e.g.: Hajlaoui et al. 2009; Güllüce et al. 2007; Iqbal et al. 2013; Murad et al. 2016). The aqueous ethanol with 7:3 ethanol: water ratio (WA) is not frequently mentioned in the references on this species (Mimica-Dukić et al. 1999; Stanislavljević et al. 2012; Benedec et al. 2013) however, the solvent is applied in pharmaceutical practice to prepare polyphenol rich tinctures i. e. from common tormentil (Potentilla erecta) rhizome (Kőszeginé Szalai *et al.* 2003). In this manner, four different extracts of each of the thirty-six horsemint samples were obtained.

In vitro antioxidant parameters Antiradical activity was investigated by DPPH assay (0.1 mM DPPH in ethanol/30 min/room temperature/normal light). Results are given in EC₅₀ mg/l extract. This means the concentration of the extract consumes half of the initial concentration of DPPH, thus a lower value means stronger activity. Reducing ability was investigated on FRAP assay. Briefly, a modified method of Benzie and Strain (1996) was used at the assay. Fifty ml of buffer (46.3 ml 0.2M acetic acid plus 3.7 ml of 0.2M Na-acetate filled to 100 ml) was measured in with 5 ml 10mM 2,4,6-s-tripyridyl-triazine (TPTZ) acidified with 84 µl cc HCl/25 ml solution. In the last step 5 ml 20mM ferric chloride was added to the mixture. 50 μ l of sample with 1450 μ l of this mixed reagent was reacted for 5 min (under normal light and room temperature). Spectra were collected and at 596 \pm 1 nm evaluated. The results were determined as mg ascorbic acid equivalents (AAE) per kg dry weight (DW). Total polyphenol content was investigated by Folin-Ciocalteu assay, with modifications of the protocol of Waterhouse (Waterhouse 2002) and given in gallic acid eq. (GAE) mg/kg DW. Suppliers of chemicals used for the antioxidant assays are as follows: DPPH, a. r., Sigma-Aldrich. Folin and Ciocalteu's reagent, 2N, VWR. TPTZ, 98%, Alfa Aesar.

HPLC investigations Apparatus: Agilent 1200 Series with diode array detector G1315D Column: ACE C18250 mm*4.6 mm*5 μ m. Eluent A: aqueous acetic acid (5 V/V%) and B: acetonitrile. Injection: 10 μ l. The applied gradient is shown in *Table 1*.

Table 1. Timetable of the gradient program applied to the horsemint extracts.

| t min | ACN % | |
|-------|-------|--|
| 0 | 5 | |
| 5 | 5 | |
| 8 | 20 | |
| 11 | 20 | |
| 13 | 25 | |
| 16 | 30 | |
| 22.5 | 25 | |
| 24.5 | 25 | |
| 30 | 50 | |
| 32 | 95 | |
| 34 | 5 | |
| 37 | 5 | |

Identification of components was based on retention times of analytical standards. Standards were Caffeic acid, Sigma-Aldrich, min. 99%, predominantly trans; rosmarinic acid, Aldrich, min. 99%; Hesperidin, Sigma-Aldrich, min. 97%, analytical standard; Cynaroside, Sigma-Aldrich, min. 98%, analytical standard; Diosmin, Sigma-Aldrich, min. 90%, analytical standard. Solvents: Acetonitrile, VWR Chromanorm, super gradient grade Reag. Ph Eur, USP, ACS. Acetic acid, glacial: VWR AnalAR Normapur.

Quantitation of the identified compounds were based on calibration with their standards. Results here are expressed as the maximal extractable concentration, in mg component/kg dry plant material weight (DW). For each sample, four concentration values were gained for the main phenolics according the different extraction methods described above. These were compared and the maxima were chosen. This approach may better approximate the real concentrations of the ingredients in the plant thus the variability of the concentrations among the 36 populations may be more recognizable.

RESULTS AND DISCUSSION

Extractability and antioxidant activity

The polyphenol contents calculated from the four different extract types has shown a strong solvent dependence. Extractions made with WA were more efficient. *Table 2* depicts the basic statistical data for the TPC of four extract types, which are calculated from the detailed results provided in our previous study (Patonay *et al.* 2019). The similarity of TPC values of the two WA extractions are observable, together with the weak efficiency of MeOH Soxhlet extraction. In Welch tests, significant difference of the obtained TPC values were observable between the two ultrasound extractions (p<0.025), the two Soxhlets (p<0.01), and between the two methanolic extract types but as it was expected from their similar values, no significant difference was found between the two hydroalcoholic treatments.

Table 2. Comparison of descriptive statistics of TPC of ML samples, calculated from the different extracts. All data is given in mg GAE/kg DW. US stands for 'ultrasonic'.

| | MeOH Soxhlet | MeOH US | WA Soxhlet | WA US |
|--------|-----------------|------------|---------------|----------|
| Q1 | 25836 | 39143 | 41875 | 46148 |
| Median | 35373 | 47520 | 53721 | 51684 |
| Q3 | 41295 | 55901 | 61351 | 58333 |
| Min | 6295 | 7105 | 18381 | 23390 |
| Max | 49033 | 67952 | 87024 | 67910 |

The solvent dependence appears in the results of the antioxidant assays, too. WA extractions provide extracts with higher antioxidant activity, than methanolic ones. But additionally, in these assays, dependence on the extraction method is also noticeable. WA sonication was observed to be more efficient than WA Soxhlet, contrary to their similarity in total polyphenol contents. Methanolic Soxhlet extraction resulted in the least efficiency, contrary to its usage appearing in the literature. Basic descriptive statistical data of DPPH EC_{50} measurements on each extract series are shown in *Table 3*. The antiradical activity of the four extract types has shown significant differences on the Welch

test (from p<0.025 to p<0.001) in all arrangements of comparisons (two hydroalcoholic extractions, two methanolic extractions, two Soxhlets and two ultrasonic extractions).

| | MeOH | MeOH | WA | WA |
|--------|---------|-------|---------|-------|
| | Soxhlet | US | Soxhlet | US |
| Q1 | 466.7 | 403.9 | 303.2 | 323.1 |
| Q3 | 727.8 | 608.3 | 462.6 | 383.3 |
| Median | 583.1 | 501.6 | 362.1 | 355.8 |
| Min | 353.3 | 291.7 | 168.5 | 226.9 |
| Max | 3183 | 1736 | 983.4 | 806.4 |

Table 3. Comparison of descriptive statistics of EC_{50} -mg/L values of the different ML extracts. US stands for 'ultrasonic'.

The ferrous reducing activities follow a similar trend as the DPPH data. Methanolic extraction is less efficient, than the WA extractions – partially, methanolic Soxhlet which is found to be the least effective. Sonication with WA is noticed to be more efficient, than WA Soxhlet extraction. The statistical characterization of FRAP values are presented in *Table 4*.

| | MeOH | MeOH | WA | WA |
|--------|---------|------|---------|-------|
| | Soxhlet | US | Soxhlet | US |
| Q1 | 5236 | 5498 | 6587 | 7173 |
| Q3 | 6704 | 7301 | 9975 | 10447 |
| Median | 6118 | 6275 | 8017 | 9102 |
| Min | 3264 | 2875 | 3441 | 5671 |
| Max | 7463 | 9101 | 11969 | 12543 |

Table 4. Comparison of descriptive statistics of FRAP values calculated from the different extracts. All data is given in mg AAE/kg DW. US stands for 'ultrasonic'.

The constituents detected in all the thirty-six samples of the horsemint

Six major peaks were observed to appear in all of the extracts and until now, four of them have been identified and quantified. These are, in order of their elutions, the following: i) cynaroside, the 7-Oglucoside of luteolin (5,7,3',4'-tetrahydroxyflavone) ii) diosmin, the 7-O-rutinoside of diosmetin (5,7,3'-trihydroxy-4'-methoxyflavone) iii) hesperidin, 7-O-rutinoside of hesperetin (5,7,3'-trihydroxy-4'- methoxyflavanone) and iv) rosmarinic acid, the caffeoyl ester of 3,4-dihydroxyphenyllactic acid. *Figure 2* is the chromatogram of a WA Soxhlet extract, also showing the structure of the identified four ingredients and designation of their peaks. Their concentrations in the studied extracts are different, depending on the plant material and, as it is expected, on solvent type. For example, hesperidin is concentrated in methanolic extracts up to 29815 mg/kg calculated to DW, while it is present in significantly lower concentrations in hydroalcoholic extracts. Identification of the two other major peaks Unknown A and B requires further investigations.

As it was expected, rosmarinic acid is a dominant constituent in the horsemint samples while other hydroxycinnamic acids as major components were not found. Another peculiar trait of the samples is the high concentration of hesperidin. This compound is frequently detected in *Mentha* plants (Guédon and Pasquier 1994) (Hanafy *et al.* 2017) but in varying concentrations, not often as a dominant constituent. Diosmin is also frequent in mints, e. g. in peppermint (Guédon and Pasquier 1994). In the case of the horsemint, presence of this compound was only reported once, from a single sample (Bourwieg and Pohl 1973).

Figure 2 shows the distribution of the identified four polyphenols in the thirty-six plant samples and *Table 2* summarizes the concentration intervals of them.

As it was expected, a high variability is observable in the concentration of each polyphenol amongst the plant samples.

| Component | Highest extracted concentration, mg/kg DW |
|-----------------|---|
| Rosmarinic acid | 7043-38667 |
| Cynaroside | <200-2944 |
| Hesperidin | 1985-29815 |
| Diosmin | 398-7987 |

Table 5. Intervals of the highest extracted concentrations of the identified polyphenols in the thirty-six horsemint samples.



Figure 1. Chromatogram of mint MBA1, WA Soxhlet extract. Table included to the left upper corner contains the retention times of the six common constituents of the mint samples, together with their most intensive detection wavelengths. Structural formulae of the identified four polyphenols are inserted above their peaks.



Figure 2. Highest extracted concentrations of the identified polypenols in the horsemint samples, given in mg/kg DW. On the x-axis of the chart the sample ID-s are provided.

Antioxidant properties of the identified polyphenols and their connection with the antioxidant capacities of the extracts

Table 6 contains the DPPH and FRAP data measured on the standards of the four identified polyphenols, compared with the values of a widely used control antioxidant, di(*tert*-)butyl-hydroxytoluene (*BHT* or *E321*).

| Standard | DPPH EC50 mg/l | | DPPH EC ₅₀ mg/l FRAP mg AAE/kg | | AE/kg |
|-----------------|----------------|-------|---|-------|-------|
| | Mean | S. D. | Mean | S. D. | |
| Rosmarinic acid | 16 | 0.75 | 172499 | 1044 | |
| Cynaroside | 33 | 0.03 | 68637 | 2910 | |
| Hesperidin | 6352 | 84.00 | 13333 | 873 | |
| Diosmin | >25000 | NA | 3098 | 42 | |
| BHT | 87.35 | 4.51 | 12169 | 34 | |

Table 6. DPPH and FRAP values of the four identified polyphenol compounds in comparison with BHT as positive control.

Rosmarinic acid and cynaroside were observed to be strong antiradical and reducing agents, being more active than BHT. Rosmarinic acid shows outstandingly strong activity in both antioxidant activity assays, especially in FRAP. This is in accordance with the available literature data each in ferrous reducing activity (Hadjimehdipoor *et al.* 2014) and in DPPH testing (Damien-Dorman *et al.* 2003; Koşar *et al.* 2004). Hesperidin was observed to be a weak antiradical agent and diosmin is practically inactive in both assays.

The difference in the antiradical activity of cynaroside and the other two flavonoids may be justified with structural background: cynaroside has free C3'-OH and C4'-OH (free catechol moiety) which is highly important in effective radical scavenging (Bors *et al.* 1990). Contrary to this, in the case of hesperidin and diosmin the C4' site is blocked by a methoxy group. The correlation between the antioxidant properties of the extracts and the concentrations of the four phenolics in them was investigated. Rosmarinic acid concentration was found to be in significant correlation both with the antiradical and the ferrous reducing activity. Strength of correlations of DPPH EC₅₀ values and rosmarinic acid content of the extracts was medium (R between 0.4990 and 0.5555 depending on extract type) with p<0.01. To ferrous reducing activity, R was calculated between 0.3551 and 0.5530 (also in dependence with the extract type) with p<0.01. To cynaroside concentrations, contrary to the high antioxidant activity of this substance, no significant correlation was found with the antioxidant properties of the extracts. No correlation between the two other identified flavonoids and the antioxidant capacities was detected, which is not unexpected regarding the weaker activity of them.

Concluding remarks

Based on our results, WA may be the appropriate extract type for assuring a high *in vitro* antiradical activity. Besides, it can be used in food industry without restrictions, contrary to methanol to which strict limitations have been declared (The European Parliament and Council 2009, 2010). The better efficacy of WA than methanol in our study is especially interesting as in numerous former references methanol was frequently used as a polar medium to gain polyphenolic antioxidants of the plant.

The investigated horsemint samples contained high levels of rosmarinic acid. Concentration of this constituent may reach high values also in other mints: 0.97-3.86 m/m% DW, ca. 9700-38600 mg/kg were observed in a survey dealing with 40 different samples of different peppermint cultivars (Guédon and Pasquier 1994). Our findings show, that horsemint may be a useful source of polyphenolic antioxidants, too. The high amount of hesperidin and the lack of caffeic acid in higher ratios was unexpected in comparison with literature of ML (Dudai *et al.* 2006). The presence of diosmin and its concentration intervals are also similar to the

values of peppermint detected in the aforementioned study (0.03-0.47 m/m %, ca. 300-4700 mg/kg DW). A medium, significant correlation between rosmarinic acid content and antioxidant properties of samples was determined while the connection of antioxidant power and content of cynaroside was not significant. These results indicate that identification and quantification of Unknown A and B in the extracts would be necessary to decide how flavonoids and other polyphenols beyond rosmarinic acid contribute antioxidant properties to the of horsemint. The investigation discussed above is part of the first phytochemical survey of horsemint populations in Hungary. The high heterogeneity of concentration of the phenolics in the horsemint samples supports the assumption that the wild-collected material is not suitable for gaining extracts of standard quality. Selection, propagation and cultivation of ML chemotypes bearing the required polyphenol composition may provide a sustainable alternative.

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