Kyoto University Research Info	mation Repository KYOTO UNIVERSITY
Title	Selective isolation of -glucan from corn pericarp hemicelluloses by affinity chromatography on cellulose column.
Author(s)	Yoshida, Tomoki; Honda, Yoichi; Tsujimoto, Takashi; Uyama, Hiroshi; Azuma, Jun-ichi
Citation	Carbohydrate polymers (2014), 111: 538-542
Issue Date	2014-10-13
URL	http://hdl.handle.net/2433/197293
Right	© 2014 Elsevier Ltd. NOTICE: this is the author's version of a work that was accepted for publication in Carbohydrate polymers. Changes resulting from the publishing process, such as peer review, editing, corrections, structural formatting, and other quality control mechanisms may not be reflected in this document. Changes may have been made to this work since it was submitted for publication. A definitive version was subsequently published in Carbohydrate polymers, 113 538-542 (2014), doi:10.1016/j.carbpol.2014.04.050
Туре	Journal Article
Textversion	author

1	Selective Isolation of β-Glucan from Corn Pericarp Hemicelluloses
2	by Affinity Chromatography on Cellulose Column
3	
4	Tomoki Yoshida ^{a*} , Yoichi Honda ^a , Takashi Tsujimoto ^b , Hiroshi Uyama ^b and
5	Jun-ichi Azuma ^b
6	
7	^a Division of Environmental Science and Technology, Graduate School of Agriculture,
8	Kyoto University, Kitashirakawa Oiwake-cho, Sakyo-ku, Kyoto 606-8502, Japan
9	
10	^b Division of Applied Chemistry, Graduate School of Engineering, Osaka University,
11	Yamadaoka, Suita 565-0871, Osaka, Japan
12	
13	*Corresponding author (T. Yoshida: telephone +81-75-753-6465; fax +81-75-753-6471;
14	e-mail Address; tonki@kais.kyoto-u.ac.jp)
15	

16 Abstract

A combination of an ion-exchange chromatography and affinity chromatography on a 17 cellulose column was found to be effective for the isolation of β -(1,3;1,4)-glucan (BG) 18 from corn pericarp hemicelluloses (CPHs). CPHs containing 6.6% BG were extracted 19 from corn pericarp with 6 M urea-2 wt% NaOH solution and initially fractionated into 20 neutral and acidic parts by anion exchange chromatography to remove acidic 21 22 arabinoxylan consisting of arabinose (35.6%) and xylose (50.9%). The neutral fraction (yield; 10.1% on the basis of CPHs) consisting of 1.0% arabinose, 10.1% xylose and 23 80.3% glucose containing 28.4% BG was then applied to a cellulose column of Watman 24 CF-11. BG could be recovered from the adsorbed fraction on the cellulose column by 25 elution with 2% NaOH in a yield of 2.6% on the basis of CPHs with a purity of 84.7%. 26 The chemical structure of the isolated corn pericarp BG was confirmed by ¹³C-NMR 27 spectroscopic, methylation and lichenase treatment analyses. The results indicate that 28 29 the ratios of (1,4)/(1,3) linkage and cellotriosyl/cellotetraosyl segments of the BG were 30 2.60 and 2.5, respectively.

Keywords: β-Glucan; Corn pericarp hemicelluloses; Cellulose column; Affinity
chromatography

34 **1. Introduction**

Corn pericarp is a by-product of industrial corn starch production and its annual 35 global generation is estimated to be over 4 million tons (Yoshida, Dwianto, Honda, 36 Uvama, & Azuma, 2014). Although it is frequently used as an ingredient in animal feed 37 38 with the addition of corn protein (Shukla, & Cheryan, 2001), finding out more valuable applications is expected. Because hemicelluloses are its major constituents amounting to 39 40 about 75%, its functional use largely depends on their extended characterization. Use of corn pericarp hemicelluloses (CPHs) as an emulsifier is a candidate for this line of 41 42 investigation (Yadav, Johnston, & Hicks, 2007; Yadav, Parris, Johnston, & Hicks, 2008). 43

Recently we found that corn pericarp contains 3.2% of β -(1,3;1,4)-glucan (BG) 44 (Yoshida, Sakamoto, & Azuma, 2012). BG is commonly present in cereal grains and is 45 included in many kinds of commercially available cereal based foods as a nutritionally 46 important ingredient, because it improves food qualities such as mouthfeel and texture 47 48 (Lazaridou, & Biliaderis, 2007). In addition, BG provides some specific health benefits, 49 such as attenuating blood postprandial glycemic and insulinemic responses, lowering 50 blood total cholesterol and low-density lipoprotein (LDL) cholesterol, and improving high-density lipoprotein (HDL) cholesterol and blood lipid profiles (Braaten et al., 1994; 51 Daou, & Zhang, 2012; Brennan, & Cleary, 2005). 52

BG is a linear homopolysaccharide comprised of two types of D-glucopyranosyl residues linked by a mixture of β -(1-3) and β -(1-4) linkages, with blocks of (1-4)-linked residues (oligometric cellulose-like segments) separated by (1-3)-linkages. Its structural

features, such as linkage ratio, number of units of cellulose-like segments and distribution of the cellulose-like segments, are known to be important determinants for its physical properties and functionalities, including its use as a food additive (Lazaridou, Biliaderis, Micha-Screttas, & Steele, 2004).

60 Previously we demonstrated the effectiveness of a NaOH-urea solvent system for extraction of hemicelluloses from corn pericarp, including BG (Yoshida, Sakamoto, & 61 Azuma, 2012). Although biorefinement of corn pericarp targeted to produce BG is 62 desirable as it is an innovative utilization of corn starch residues, rather tedious steps for 63 64 the removal of large amounts of other polysaccharides are usually required for the isolation of BG from monocotyledonous crops (Ahmad, Anjum, Zahoor, Nawaz, & 65 Ahmed, 2009; Ahmad, Anjum, Zahoor, Nawaz, & Din, 2007; Beer, Arrigoin, & Amadò, 66 1996; Bhatty, 1993; Burkus, & Temelli, 1998; Lazaridou, Biliaderis, Micha-Screttas, & 67 Steele, 2004; Wood, Weisz, Fedec, & Burrows, 1989). Two important steps so far 68 noticed were removal of contaminating starch and arabinoxylan. Repeated treatments 69 70 with thermo-stable starch-degrading enzymes were usually necessary to enrich BG for 71 removal of starch. Solubility difference in aqueous media was frequently used to 72 remove arabinoxylan (Izydorczyk, & Biliaderis, 1995; Izydorczyk, Biliaderis, Macri, & MacGregor, 1997; Izydorczyk, & MacGregor, 2000). However, in the case of CPHs, 73 74 removal of starch was not prerequisite because of the low content (about 1%) and our trials of fractional precipitation of BG from a mixture with arabinoylan by using ethanol 75 76 and ammonium sulfate were unsuccessful.

77 In this study, we developed a convenient isolation method specific for corn pericarp

BG by using affinity chromatography on a cellulose column, and present the chemical
properties of the isolated BG were also investigated.

80

81 **2. Materials and Methods**

82 *2. 1. Materials*

Kernels of sweet corn cultivated and steamed for food in Hokkaido, Japan, were 83 84 purchased from Kewpie Co., Japan. Corn pericarp was manually peeled from the upper portion of each kernel $(9.4 \pm 0.7\%)$ on basis of dried kernel, n = 10) and treated with hot 85 86 water (121°C) for 1 h. Corn pericarp hemicelluloses (CPHs) consisting of 40.5% of xylose, 29.2% of arabinose, 26.2% of glucose and 4.1% of galactose were prepared by 87 extraction of corn pericarp with 2 wt% NaOH-6 M urea in a yield of 74.8% on the basis 88 of dried corn pericarp as described previously (Yoshida, Sakamoto, & Azuma, 2012). 89 β-Glucan (BG) from barley (>95%) was purchased from Sigma (St. Louis, Missouri, 90 91 USA). Amounts of BG and starch were determined by using the mixed-linkage β -glucan 92 and total starch content assay kits (Megazyme International Ireland Ltd., Wicklow, Ireland), respectively. Whatman CF-11 cellulose powder (WhatmanTM, a part of GE 93 94 Healthcare Life Science, Ltd., Buckinghamshire, UK) was used for affinity chromatography after pre-washing with 5% NaOH and neutralization with acetic acid. 95

96

97 2. 2. Isolation of β -(1,3;1,4)-glucan

Anion exchange chromatography was first applied for partial purification of BG. Hot water-soluble CPHs (94.2 \pm 0.5% on the basis of CPHs, *n* = 4) obtained by extraction of

corn pericarp with a 100 fold excess of water at 80°C for 2-3 h were applied to a column 100 101 (15 × 150 mm) of TOYOPEARL DEAE-650M (Tosoh Co., Tokyo, Japan) equilibrated 102 with 5 mM sodium phosphate buffer (pH 6.8) and eluted with the same solution to recover neutral BG. Acidic arabinoxylan was next recovered by elution with the same 103 buffer containing 1.2 M NaCl. Elution was monitored by the phenol-sulfuric acid 104 105 method. Both polysaccharide fractions were separately pooled, dialyzed against water 106 and freezed-dried. The BG-rich fraction (Neutral fraction; $8.7 \pm 2.4\%$ on the basis of 107 CPHs, n = 3, Table 1) was dissolved in 5 mM sodium acetate buffer (SAB), pH 5.0 and 108 applied to a cellulose column (15×150 mm). After equilibrating for 30 min at room 109 temperature, the column was washed with the same buffer to remove unadsorbed 110 material, eluted with distilled water, and finally adsorbed BG was recovered by elution with 2% NaOH (2% NaOH fraction; yield $3.3 \pm 1.3\%$ on the basis of CPHs, n = 3, 111 Table 1). All carbohydrate containing fractions were pooled, neutralized, dialyzed 112 113 against water and freeze-dried.

114

115 *2. 3. Chemical analysis*

116 CPHs and all of the materials recovered by anion exchange and affinity 117 chromatographic techniques were hydrolyzed according to the Saeman method (Saeman, 118 Bubl, & Harris, 1945), and their monosaccharide compositions were determined by 119 high-performance anion exchange chromatography (HPAEC) on a Dionex DX-500 120 system (Sunnyvale, CA, USA) equipped with a pulsed amperometric detector (ED-40) 121 as described in our previous report (Yoshida, Tusbaki, Teramoto, & Azuma, 2010). The liquid state ¹³C-NMR spectrum of the isolated BG was recorded in D_2O on a Bruker DPX-400 instrument (Billerica, Bruker, MA, USA) operating at 400 MHz and the chemical shifts in ppm were normalized as downfield values from that of internal standard, TSP (sodium 2,2,3,3-tetradeuterio-3-(trimethylsilyl)-propionate).

Permethylation of polysaccharides was carried out according to the Hakomori 126 method (Hakamori, 1964). The permethylated polysaccharides were subjected to two-127 128 step hydrolysis with 90% formic acid for 2 h at 100°C and 0.5 N sulfuric acid for 12 h at 100°C. After neutralization with barium carbonate, the hydrolyzate was reduced with 129 130 sodium borohydride and acetylated with a mixture of acetic anhydride and pyridine (1:1. v/v) for 1 h at 100°C. The resulting mixture of partially methylated additol acetates was 131 analyzed by GC/MS with a Shimadzu Parvum 2 (70 eV) using a column of Shimadzu 132 133 CBP-1 (0.25 μ m, 0.25 mm \times 25 m) and a linear temperature gradient from 140°C to 220°C at 2°C/min. 134

135 The distribution of (1,4)-linked-glucopyranosyl segments in BG was determined by lichenase treatment and high-performance liquid chromatography (HPLC). BG samples 136 137 were dissolved in sodium phosphate buffer (20 mM, pH6.5) and incubated with 138 lichenase [(1,3;1,4)-beta-glucan-4-glucanohydrolase, 1000 U/mL, included in the 139 Megazyme kit for measurement of BG content] for 2 h at 50°C. After centrifugation, the 140 supernatant was purified by passage through a joint column of cation (Dowex 50x, 8 H⁺ form) and anion (Dowex 1x8, acetate form) exchange resins. The passed solution and 141 washed solution with pure water were freeze-dried (recovery, 62.7%). The distribution 142 of segments was analyzed by HPLC on a column of MCI GEL CK04SS (7.5×200 mm, 143

- 144 Mitsubishi Chemical Industry Co., Tokyo, Japan) at 80°C with refractive index detector
- 145 (RI-8, Tosoh Co., Tokyo, Japan). The eluent was deionized water and flow rate was 0.3
- 146 mL/min. Elution was monitored using Chrom NAV Station, Jasco, Co., Tokyo, Japan).
- 147

148 **3. Results and discussion**

149 CPHs (BG content 6.6%) were separated into neutral and acidic fractions by anion exchange chromatography on a DEAE-column as shown in Fig. 1 (A). The neutral 150 fraction was further separated into three fractions by affinity chromatography on a 151 cellulose column as shown Fig. 1 (B). Yields of the separated fractions are listed in 152 Table 1. The relative monosaccharide compositions of CPHs and the fractions separated 153 154 by anion exchange and affinity chromatography are listed in Table 1. The results indicate that the neutral fraction contained BG (28.4%) together with a small amount of 155 156 xylan (10.1%). On the other hand, the acidic fraction was predominantly arabinoxylan consisting of xylose (50.9%) and arabinose (35.6%), with glucose as a minor constituent 157 (3.5%). These results indicate that anion exchange chromatography was effective for 158 partial purification of BG in corn pericarp. Previously, Gruppen et al. (1992) have 159 reported that anion exchange chromatography is an efficient tool for the fractionation of 160 161 arabinoxylans present in wheat flour. On the other hand, our results indicate its 162 suitability for the removal of the acidic arabinoxylan present in abundance in CPHs.

In the present study, affinity chromatography on a cellulose column was found to be more effective for selective purification of BG. The glucose contents of the fractions eluted with SAB, distilled water and 2% NaOH were 29.5, 50.6 and 91.5%, respectively (Table 1). The starch content of the SAB, water and 2% NaOH fractions were 4.3, 5.7 and 3.5%, respectively. BG was recovered in 84.7% purity from the column by elution with 2% NaOH (Table 1) on the basis of hot water (100°C)-soluble materials. When the BG content in the NaOH fraction was initially examined by using its whole amount, BG content was calculated as $67.0 \pm 3.6\%$. This value, however, seemed to be invalid, because this fraction was mainly consisted of glucose (91.5%) with minor contaminants of starch (3.5%) and contained a large amount of insolubilized materials (20.8%) mainly composed of glucose (78.6%). Insolubilization of BG after purification was pointed out by Lazaridou and Biliaderis (2007). Therefore, in the present study, the purity of the NaOH fraction was estimated by calculation on the basis of soluble materials.

When the water-soluble portion of CPHs was directly applied to the cellulose column, the adsorbed fraction was found to be contaminated with a larger amount of xylose (23.2%). This result shows the necessity of anion-exchange chromatography prior to cellulose affinity chromatography.

Fig. 2 shows the ¹³C-NMR spectrum of the materials separated into the 2% NaOH 181 fraction. Each signal was assigned according to the previous report (Bock, Duus, 182 183 Norman, & Pedersen, 1991; Cui, Wood, Blackwell, & Nikiforuk, 2000; Roubroeks, 184 Andersson, & Aman, 2000) and the peak assignments are listed in Table 2. The 185 spectrum was identical to that of pure β-glucan (Cui, Wood, Blackwell, & Nikiforuk, 186 2000; Roubroeks, Andersson, & Aman, 2000). The spectrum of materials isolated in the 2% NaOH fraction showed characteristic intense peaks at 81.3 and 86.6 ppm, assigned 187 188 to the signals of C-3 of (1,4)-linked- and (1,3)-linked-D-glucopyranosyl residues of β -glucan, respectively. Although very weak signals assignable to arabinoxylan 189 (Roubroeks, Andersson, & Aman, 2000) were detected, no intense peaks other than BG 190 191 could be detected. These results also indicate that the material in the 2% NaOH fraction

192 was high purity BG. The present study shows for the first time that BG can be 193 effectively isolated from other hemicellulosic polysaccharides present in corn pericarp 194 by using a combination of anion exchange and cellulose affinity column 195 chromatography.

The glucosidic linkage analysis of the purified BG was investigated by methylation 196 analysis. The corn pericarp BG consisted of 2,3,6-Me-Glcp, 2,4,6-Me-Glcp and 197 198 2,3,4,6-Me-Glcp in a molar ratio of 58.1 (1,4-linked Glcp), 22.3 (1,3-linked Glcp) and 199 0.5% (terminal Glcp), respectively (Table 3). The ratio of (1,4)-glucose linkages to 200 (1,3)-linkages for the corn pericarp β -glucan was calculated as 2.60, which was slightly higher than that for barley (2.37) but within the ranges previously reported; 2.3-2.8 for 201 oat, 1.9-2.8 for barley and 2.3 for rye (Lazaridou, & Biliaderis, 2007). The high affinity 202 203 of the corn pericarp BG to cellulose suggests the existence of strong interactions between BG and cellulose in corn pericarp. 204

205 The distribution of (1,4)-linked glucopyranosyl segments in corn pericarp BG was 206 examined by fragmentation analyses with lichenase which splits (1,3)- β -D-glycosidic 207 linkages in BG (Table 4). After lichenase treatment, cello-octomer and shorter 208 oligomers were detected in the corn pericarp BG. The ratio of cellotriosyl/cellotetraosyl 209 units for corn pericarp BG was 2.5, which was slightly higher than that for barley (2.2). 210 Previously, the same ratios in the native cereal β -glucan structures were reported to be within the range of 1.5-2.3 for oat, 1.8-3.5 for barley, 1.9-3.8 for rye and 3.0-4.5 for 211 wheat (Lazaridou, & Biliaderis, 2007). Ebringerová et al. (2005) described in their 212 review as 'In comparison to the water-extractable β -glucan-rich fractions, the 213

alkali-extractable ones were characterized by high ratios of cellotriosyl/cellotetraosyl units and large amounts of long, contiguously linked $(1\rightarrow 4)$ -linkage segments. Such polymers exhibit a tendency for interchain aggregation through strong hydrogen bonding along the cellulose-like regions and hence lower solubility'. The present results might fit their descriptions and be suffice to show that the affinity of BG for cellulose was the basis for its isolation.

220

221 **3. Conclusion**

The effectiveness of the combination of anion-exchange and cellulose affinity chromatographic techniques for the isolation of BG from CPHs was established for the first time. By using the present system, the BG present at 6.6% in CPHs was purified to 84.7%. The methylation and fragmentation analyses showed that the BG isolated from the CPHs has (1,4)/(1,3) linkage and cellotriosyl/cellotetraosyl segment ratios of 2.60 and 2.5, respectively. The chemical structure of the BG was also confirmed by ¹³C-NMR spectroscopic analysis.

Reference

230	Ahmad, A., Anjum, F. M., Zahoor, T., Nawaz, H., & Ahmed, Z. (2009). Extraction and
231	characterization of beta-D-glucan from oat for industrial utilization.
232	International Journal of Biological Macromolecules, 46(3), 304-309.
233	Ahmad, A., Anjum, F. M., Zahoor, T., Nawaz, H., & Din, A. (2007). Physicochemical
234	and functional properties of barley beta-glucan as affected by different
235	extraction procedures. International Journal of Food Science and Technology,
236	44(1), 181-187.
237	Beer, M. U., Arrigoin, E., & Amadò, R. (1996). Extraction of oat gum from oat bran:
238	Effects of process on yield, molecular weight distribution, vscosity and
239	$(1\rightarrow 3)(1\rightarrow 4)$ - β -D-glucan content of the gum. <i>Cereal Chemistry</i> , 73(1), 58-62.
240	Bhatty, R. S. (1993). Extraction and enrichment of $(1\rightarrow 3), (1\rightarrow 4)-\beta$ -D-glucan from
241	barley and oat brans. Cereal Chemistry, 70(1), 73-77.
242	Bock, K., Duus, J. O., Norman, B., & Pedersen, S. (1991). Assignment of structures to
243	oligosaccharides produced by enzymic degradation of a beta-D-glucan from
244	barley by ¹ H- and ¹³ C-n.m.r. spectroscopy. Carbohydrate Research, 211(2),
245	219-233.
246	Braaten, J. T., Wood, P. J., Scott, F. W., Wolynetz, M. S., Lowe, M. K., Bradleywhite,
247	P., & Collins, M. W. (1994). Oat beta-glucan reduces blood cholesterol
248	concentration in hypercholesterolemic subjects. European Journal of Clinical
249	Nutrition, 48(7), 465-474.
250	Brennan, C. S., & Cleary, L. J. (2005). The potential use of cereal

- 251 $(1\rightarrow 3, 1\rightarrow 4)$ -β-D-glucans as functional food ingredients. *Journal of Cereal* 252 *Science*, 42(1), 1-13.
- Burkus, Z., & Temelli, F. (1998). Effect of extraction conditions on yield, composition,
 and viscosity stability of barley β-glucan gum. *Cereal Chemistry*, 75(6),
 805-809.
- Cui, W., Wood, P. J., Blackwell, B., & Nikiforuk, J. (2000). Physicochemical properties
 and structural characterization by two-dimensional NMR spectroscopy of
 wheat beta-D-glucan comparison with other cereal beta-D-glucans.
 Carbohydrate Polymers, 41(3), 249-258.
- Daou, C., & Zhang, H., (2012). Oat beta-glucan: its role in health promotion and
 prevention of diseases. *Comprehensive Reviews in Food Science and Food Safety*, 11(4), 355-365.
- Ebringerová, A., Hromádková, Z., & Heinze, T. (2005). Hemicellulose. In T. Heinze
 (Ed.). *Polysaccharides 1: Structure, Characterization and Use* (Vol. 186, pp.
 1-67). Berlin: Springer-Verlag Berlin.
- Gruppen, H., Hamer, R. J., & Voragen, A. G. J. (1992). Water-unextractable cell wall
 material from wheat flour. 2. Fractionation of alkali-extracted polymers and
 comparison with water-extractable arabinoxylans. *Journal of Cereal Science 16*(1), 53-67.
- Hakamori, S. (1964). A rapid permethylation of glycolipid and polysaccharide catalyzed
 by methylsulfinyl carbanion in dimethyl sulfoxyde. *The Journal of Biochemistry*, 55(2), 205-208.

273	Izydorczyk, M. S., & Biliaderis, G. C. (1995). Arabinoxylans: advances in structure and
274	physicochemical properties. Carbohydrate Polymers, 28(1), 33-48.
275	Izydorczyk, M. S., Biliaderis, G. C., Macri, L. J., & MacGregor, A. W. (1997).
276	Fractionation of oat $(1\rightarrow 3),(1\rightarrow 4)$ - β -D-glucans and characterization of the
277	fractions. Journal of Cereal Science, 27(3), 321-325.
278	Izydorczyk, M. S., & MacGregor, A. W. (2000). Evidence of intermolecular
279	interactions of beta-glucans and arabinoxylans. Carbohydrate Polymers, 41(4),
280	417-420.
281	Lazaridou, A., & Biliaderis, C. G. (2007). Molecular aspects of cereal beta-glucan
282	functionality: Physical properties, technological applications and physiological
283	effects. Journal of Cereal Science, 46(2), 101-118.
284	Lazaridou, A., Biliaderis, C. G., Micha-Screttas, M., & Steele, B. R. (2004). A
285	comparative study on structure-function relations of mixed-linkage $(1\rightarrow 3)$,
286	$(1\rightarrow 4)$ linear beta-D-glucans. Food Hydrocolloids, 18(5), 837-855.
287	Roubroeks, J. P., Andersson, R., & Aman, P. (2000). Structural features of $(1\rightarrow 3)$,
288	$(1\rightarrow 4)$ - β -D-glucan and arabinoxylan fractions isolated from rye bran.
289	Carbohydrate Polymer, 42(1), 3-11.
290	Wood, P. J., Weisz, J., Fedec, P., & Burrows, V. D. (1989). Large-scale preparation and
291	properties of oat fractions enriched in $(1\rightarrow 3)(1\rightarrow 4)$ -beta-D-glucan. Cereal
292	Chemistry, 66(2), 97-103.
293	Saeman, J. F., Bubl, J. L., & Harris, E. E. (1945). Quantitative saccharification of wood
294	and cellulose. Industrial and Engineering Chemistry, Analytical Edition, 17(1),

5 35-37.

- Shukla, R., & Cheryan, M. (2001). Zein: the industrial protein from corn. *Industrial Crops and Products*, *13*(3), 171-192.
- Yadav, M. P., Johnston, D. B., & Hicks, K. B. (2007). Structural characterization of
 corn fiber gums from coarse and fine fiber and a study of their emulsifying
 properties. *Journal of Agricultural and Food Chemistry*, 55(15), 6366-6371.
- Yadav, M. P., Parris, N., Johnston, D. B., & Hicks, K. B. (2008). Fractionation,
 characterization, and study of the emulsifying properties of corn fiber gum.
 Journal of Agricultural and Food Chemistry, 56(11), 4181-4187.
- Yoshida, T., Tsubaki, S., Teramoto, Y., & Azuma, J. (2010). Optimization of
 microwave-assisted extraction of carbohydrates from industrial waste of corn
 starch production using response surface methodology. *Bioresource Technology*, 101(20), 7820-7826.
- Yoshida, T., Sakamoto M., & Azuma, J. (2012). Extraction of Hemicelluloses from
 Corn Pericarp by the NaOH-Urea Solvent System. *Procedia Chemistry*, 4(1),
 294-300.
- Yoshida, T., Dwianto, W., Honda, Y., Uyama, H., & Azuma, J. (2014). Removal of
 arabinose substituents from corn pericarp arabinoxylan. *Wood Research Journal.* 4(1), 46-50.
- 314

Figure captions 316

- 317 Fig. 1. Anion exchange chromatographic (A) and affinity chromatographic (B) profiles. (A) The neutral and acidic fractions shown as separate bars at the top of the figure were pooled. (B) 318 319 Neutral fraction was applied to the cellulose column. The three fractions eluted with SAB, 320 distilled water and 2% NaOH were recovered.
- Fig. 2. ¹³C-NMR spectrum of materials isolated in 2% NaOH fraction using a cellulose column 321
- 322
- 323

324 Table 1. Yields of the separated fractions and sugar compositions of corn pericarp

Coursel.	TT: 118 (0/)	Relative	monosaco	charide con	mposition (%, w/w)	β-Glucan	Starch
Sample	Yield (%)	Ara	Gal	Glc	Xyl	Man	content ^b (%)	content ^b (%)
CPHs	-	29.2	4.1	26.2	40.5	tr ^c	6.6	1.4
		Anior	ı exchang	e chroma	tography			
Neutral fraction	10.1 ± 0.4	1.0	2.0	80.3	10.1	6.7	28.4	5.0
Acidic fraction	42.8 ± 4.8	35.6	8.2	3.5	50.9	1.9	0.6	0.6
	Ą	ffinity chr	omatogra	phy on ce	ellulose co	lumn		
SAB fraction	7.1 ± 3.2	23.4	7.6	29.5	35.8	3.7	1.3	4.3
Water fraction	1.0 ± 0.7	14.3	7.1	50.6	23.4	4.6	1.4	5.7
2% NaOH fraction	2.6 ± 0.5	1.3	0.9	91.5	6.3	-	84.7^{d}	3.5

325 hemicelluloses (CPHs) and separated fractions.

327 ^aValues are expressed as a percentage on the basis of the raw material corn pericarp

328 hemicelluloses. Values are expressed as mean \pm SD (n = 3).

329 ^bValues represent the average of duplication.

330 ^ctr represents trace.

 d Value is expressed as a percentage on the basis of the hot water (100°C)-soluble materials.

332

Linkaga tuna			Chemical	shift (ppm)		
	C1	C2	C3	C4	C5	C6
\rightarrow 4)- β -Glc p (1 \rightarrow 3)	105.41	76.08	81.29	76.92	77.68	62.81
\rightarrow 3)- β -Glc p (1 \rightarrow 4)	105.22	75.84	86.60	70.81	78.45	63.41
\rightarrow 4)- β -Glc p (1 \rightarrow 4)	105.22	76.08	81.29	76.92	77.68	62.81

333 Table 2. Chemical shifts (ppm) of the 13 C responses of the glucose residues of β -glucan isolated

334 in the NaOH fraction in D_2O

Table 3. Methylation analysis of the materials isolated in the 2% NaOH fraction using a

Mathulation position	L intra contras	Molar ratio (%)					
	NaOH fraction B		Barley β-glucan				
2,3,6-Me ₃ -Glc <i>p</i>	1,4-	71.8	69.4				
2,4,6-Me ₃ -Glc <i>p</i>	1,3-	27.6	29.3				
2,3,4,6-Me ₄ -Glc <i>p</i>	Terminal	0.6	1.0				
3,6-Me ₂ -Glc <i>p</i>	1,2,4-	-	0.3				

337 cellulose column and of barley β -glucan.

339 Values are expressed as a relative percentage of the total partially methylated glucose residues.

340

341	Table 4.	Fragmentation	analysis	with	lichenase	of	the	materials	isolated	in	the	2%	NaOH
		<u> </u>	2										

Oligmar	תרו	Molar ratio (%)			
Oliginei	DP	NaOH fraction	Barley		
cellobiose	2	0.3	1.3		
cello-trimer	3	67.5	63.8		
cello-tetramer	4	26.7	28.7		
cello-pentamer	5	3.4	4.4		
cello-hexamer	6	1.7	1.1		
cello-heptamer	7	0.2	0.4		
cello-octamer	8	0.1	0.3		
cello-trimer + cello-tetramer	3 + 4	94.2	92.5		
cello-trimer/cello-tetramer	3/4	2.5	2.2		

342 fraction after affinity chromatography on a cellulose column and of barley β -glucan.

344 DP represents degree of polymerization.



Fig. 1. Anion exchange chromatographic (A) and affinity chromatographic (B) profiles. (A) The 347 348 neutral and acidic fractions shown as separate bars at the top of the figure were pooled. 349 Fractions 5-15 and 25-36 separated by anion exchange chromatography were designated as 350 neutral and acidic fractions, respectively. The eluent for the anion exchange chromatography 351 was changed at fraction 21 from 5 mM sodium phosphate buffer (pH 6.8) to the same buffer 352 containing 1.2 M NaCl. (B) The neutral fraction was applied to the cellulose column. Three 353 fractions eluted with SAB, distilled water and 2% NaOH were recovered. Fractions 4-11, 19-21 354 and 35-44 separated by affinity chromatography on a cellulose column were designated as SAB, 355 water and NaOH fractions, respectively. The eluent for the affinity chromatography was 356 changed at fraction 17 from SAB to water, and at fraction 33 to 2% NaOH. The volume of each 357 fraction was 3 mL.



360 Fig. 2. ¹³C-NMR spectrum of materials isolated in 2% NaOH fraction using a cellulose column