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Simultaneous nitrogen and dissolved methane removal from a upflow anaerobic sludge blanket reactor effluent using an integrated fixed film activated sludge system

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### Highlights

- Up to 32.5 mg TN  $L^{-1}$  and 93% of the dissolved CH<sub>4</sub> were removed in the IFAS system.
- Suspended biomass was completely washed-out from the IFAS system.
- Biofilm carriers played a key role in nitrogen and CH<sub>4</sub> removals.
- Almost half of the CH<sub>4</sub> and nitrogen removals were observed in the aerobic stage.
- Aerobic methanotrophs and anammox were detected in the anoxic and aerobic compartments.

#### 1 Keywords

- 2 Upflow anaerobic sludge blanket; Methane oxidation; Denitrification; Integrated fixed-
- 3 film activated sludge; Biofilm carriers; Greenhouse gas.

#### 4

#### 5 Abbreviations

- 6 AMO-D: aerobic methane oxidation coupled to denitrification
- 7 Anammox: anaerobic ammonium oxidation
- 8 COD<sub>S:</sub> soluble chemical oxygen demand
- 9 COD<sub>T</sub>: total chemical oxygen demand
- 10 DO: dissolved oxygen
- 11 FA: Free ammonia
- 12 GHG: greenhouse gas
- 13 HYME-D: hypoxic methane oxidation coupled to denitrification
- 14 IFAS: Integrated fixed-film activated sludge
- 15 MBR: membrane bioreactor
- 16 MLTSS: mixed liquor total suspended solids
- 17 MLVSS: mixed liquor volatile suspended solids concentration
- 18 MRR: methane removal rate

- 19 N-damo archaea: nitrate-dependent anaerobic methane oxidation archaea
- 20 N-damo bacteria: nitrite-dependent anaerobic methane oxidation bacteria
- 21 NOB: nitrite-oxidizing bacteria
- 22 NRR: nitrogen removal rate
- 23 OLR: organic loading rate

24 R: recycling ratio

- 25 TN: total nitrogen
- 26 UASB: upflow anaerobic sludge blanket

27

#### 28 **1. Introduction**

Anaerobic wastewater treatment technologies are widely used in warm climate countries 29 for treating low-strength streams at room temperature. Indeed, there are hundreds of 30 upflow anaerobic sludge blanket (UASB) reactors operating in different parts of 31 (semi)tropical regions due to low or even zero energy demand (Chernicharo et al., 32 33 2015). One of the main constraints of the UASB technology is related to its insufficient removal of organic matter and nitrogen. Given the low C/N ratios commonly observed 34 in anaerobic effluents, the addition of expensive external carbon sources (e.g., ethanol 35 and methanol) is frequently necessary to improve nitrogen removal through 36 37 heterotrophic denitrification processes. In addition, Noyola et al. (2006) estimated that up to 50% of the methane produced in methanogenic reactors that treat domestic sewage 38 39 at 20 °C could remain dissolved in their effluents. Methane is a strong greenhouse gas

(GHG) with a global warming potential 28 times higher than  $CO_2$  for a hundred year 40 41 time horizon (Myhre et al., 2013), which contributes to aggravate the climate change problem. However, this dissolved methane present in UASB effluents can be 42 43 biologically used as an inexpensive electron donor to denitrify to reduce both GHG emissions and nitrogen content of the treated wastewater. So far, different simultaneous 44 45 nitrogen and methane removal bioprocesses are distinguished. First, in aerobic methane 46 oxidation coupled to denitrification (AMO-D), strict aerobic methanotrophs oxidize methane into methane oxidation products (Modin et al., 2007). Afterward, conventional 47 heterotrophic denitrifiers can use these oxidation products as an electron donor to 48 49 reduce nitrate and/or nitrite to dinitrogen gas (Equations 1 and 2). Second, in the anaerobic pathway two main microorganisms are involved: nitrate-dependent anaerobic 50 methane oxidizers (n-damo archaea), such as "Candidatus Methanoperedens 51 52 nitroreducens" (Haroon et al., 2013), which are able to reduce nitrate to nitrite using methane as electron donor (Equation 3), and nitrite-dependent anaerobic methane 53 54 oxidizers (n-damo bacteria), such as "Candidatus Methylomirabilis oxyfera" (Ettwig et al., 2010), which are able to reduce nitrite to dinitrogen gas using also methane as 55 electron donor (Equation 4). Third, Cao et al. (2019) also demonstrated an innovative 56 57 pathway for removing methane and nitrogen in extreme oxygen-limited conditions, which consist of hypoxic methane oxidation coupled to denitrification (HYME-D), 58 where, instead of using oxygen as the final electron acceptor, aerobic methanotrophs 59 directly utilize nitrate as electron acceptor to oxidize methane and produce organic 60 compounds that could be subsequently utilized by conventional heterotrophic 61 denitrifiers to denitrify. 62

63 
$$5CH_4 + 5O_2 + 4NO_3^- + 4H^+ \rightarrow 2N_2 + 12H_2O + 5CO_2$$
 (1)

$$64 \qquad 3CH_4 + 3O_2 + 4NO_2^- + 4H^+ \rightarrow 2N_2 + 8H_2O + 3CO_2 \tag{2}$$

$$65 \quad 2CH_4 + 8NO_3^{-} \rightarrow 2CO_2 + 8NO_2^{-} + 4H_2O \tag{3}$$

$$66 \quad 3CH_4 + 8NO_2^- + 8H^+ \to 3CO_2 + 4N_2 + 10H_2O \tag{4}$$

67 Therefore, to minimize the environmental impact of effluents from UASB reactors that 68 treat low-strength sewage at room temperature, and to comply with the increasingly 69 severe effluent discharge regulations, further low-footprint post-treatment systems are 70 required. On this ground, authors such as Kampman et al. (2012), Cai et al. (2015), Pelaz et al. (2017), and van Kessel et al. (2018) proposed the use of dissolved methane 71 72 as an inexpensive electron donor to denitrify. In previous studies carried out in our 73 laboratories, a hybrid MBR system was proposed as an alternative post-treatment 74 system to minimize the impact of effluents from UASB reactors that treat domestic 75 sewage at room temperature (Silva-Teira et al., 2017; Alvarino et al., 2019). In the 76 present research, the same idea of using dissolved methane as an electron donor to denitrify is considered. However, the utilization of an integrated fixed-film activated 77 sludge (IFAS) with a secondary settler instead of a membrane bioreactor (MBR) system 78 79 was evaluated. Unlike the MBR technology, which requires a high energy demand associated with membrane operation, IFAS systems can result in a less-energy 80 consuming alternative, but probably with lower effluent quality. The use of biofilm 81 technologies such as the IFAS system, which combines the use of both attached and 82 suspended biomass, has arisen as an alternative for the treatment of domestic or 83 84 industrial sewage, leading to more compact systems than the conventional activated sludge ones. The IFAS technology enables to increase the effective biomass 85 concentration, improving biomass diversity without the need to enlarge the size of the 86 different compartments of treatment plants. Furthermore, biofilm carriers enhance the 87 88 retention of slow-growth microorganisms, such as nitrifiers (Randall and Sen, 1996) or aerobic/anaerobic methane oxidizers, preventing them from being washed out through 89

90 the IFAS final effluent. Besides, biofilms are spatially heterogeneous, which makes
91 possible the coexistence of anaerobic, anoxic, and aerobic bioprocesses in them (Leyva92 Díaz et al., 2016).

93 The present study aims to evaluate the feasibility of using an IFAS system as an alternative post-treatment technology to mitigate the environmental impact of two of the 94 95 main pollutants present in UASB effluents, such as dissolved methane and nitrogen. To this end, a low-footprint bench-scale plant (186 L), made up of a UASB (120 L) 96 followed by an IFAS (66 L) system, was operated. In a previous study, within the same 97 integrated system operation, Arias et al. (2018) analyzed the removal efficiencies 98 99 achieved for a wide range of organic micropollutants. However, in the present study, special attention was given to the suspended biomass retention capacity and the 100 101 potential for removal of dissolved methane and nitrogen removal of the IFAS post-102 treatment system. Furthermore, the role of carriers in the processes of denitrification and 103 nitrification and the microbial communities present in the biofilm were also analyzed. 104 Finally, the results achieved in this research were compared with previous studies, 105 where similar operating conditions were applied (Silva-Teira et al., 2017; Alvarino et al., 2019). Nevertheless, instead of using an IFAS, a hybrid MBR was proposed as a 106 107 post-treatment alternative to treat effluents from UASB reactors.

108

#### 109 2. Materials and methods

#### 110 **2.1 Reactor configuration and operation**

A 186 L pilot plant (Figures 1 and S1) was used for 407 days in order to treat domestic sewage at room temperature. The pilot plant was first composed of a 120 L methanogenic reactor, using the UASB configuration with granular sludge. The obtained UASB effluent was subsequently led to a pre-anoxic IFAS system composed

of three different compartments: a first anoxic compartment (36 L) with mechanical 115 116 mixing, in which simultaneous nitrogen and methane removal bioprocesses were intended, using suspended biomass and also biomass attached to polyurethane sponge 117 118 biofilm carriers (Levapor); a second aerobic compartment (20 L), where ammonium oxidation processes were carried out, with the presence of suspended biomass and 119 biofilm adhered to rigid polyethylene supports (Biochip). Both biofilm carriers were 120 moved freely in the liquid volume of the anoxic and aerobic compartments, 121 122 respectively; and finally, a settler (10 L) in which the physical biomass/effluent separation was carried out. Moreover, the system operation was controlled by a PLC 123 124 Micrologix (Allen-Bradley) connected to a computer.

125 Three different types of biomass conformation (granular, attached to carriers, and suspended) and three redox conditions (anaerobic, anoxic, and aerobic) were present in 126 the suggested integrated system. The two different biofilm carriers used, Levapor and 127 Biochip (Figure S2), presented an external specific surface of 486 and 2,174 m<sup>2</sup> m<sup>-3</sup>, and 128 a volume of  $2.8 \times 10^{-3}$  and  $4.15 \times 10^{-4}$  L, respectively. The amount of carrier used for 129 130 both varied throughout the experimentation period. The initial apparent volume percentage for Levapor was of 20%. However, this value was raised up to 23% on day 131 132 250. As a consequence of the higher difficulty in keeping Biochip carriers homogenously mixed, the initial apparent volume percentage was only 7%. 133 Nevertheless, this value was stepwise increased to 12, 16, and 19% on days 135, 247, 134 135 and 309, respectively. Biochip was utilized in the aerobic compartment, instead of Levapor, due to its 4.47 times higher external specific surface, which favors the 136 nitrifying capacity per unit of carrier volume, where oxygen transfer usually limits 137 biological conversions (Henze et al., 2002). Conversely, Levapor carriers were used in 138

the anoxic compartment because of their higher void volume, which could favor thedevelopment of anaerobic environments.

The UASB was continuously fed with synthetic wastewater, mimicking a typical 141 composition for domestic sewage: skimmed milk (diluted with tap water); NaHCO<sub>3</sub> 142 (960 mg L<sup>-1</sup>); NH<sub>4</sub>Cl (18 mg N L<sup>-1</sup>) and trace elements with the same composition as 143 proposed by Silva-Teira et al. (2017). The UASB-IFAS integrated system was 144 continuously fed with a fixed sewage flow of 140 L d<sup>-1</sup>, which resulted in hydraulic 145 retention times (HRT) of 21 h and 10 h for the UASB and the IFAS system, 146 respectively. Two different recirculation streams were applied in the post-treatment 147 148 system: an internal recycling ratio (R) of 1.5 was applied from the aerobic to the anoxic 149 compartment to provide an electron acceptor, and an external recirculation from the 150 settler to the anoxic chamber to return the activated sludge to the post-treatment system (R = 1.5). Therefore, a total recirculation value of 3 was fixed throughout the operation. 151 152 This crucial parameter also governs the contact time in each post-treatment compartment and the quantity of  $O_2$  that is recycled to the anoxic compartment. The 153 154 UASB reactor was inoculated with active biomass from the brewery industry, while the post-treatment system was seeded using stored flocculent biomass obtained from the 155 156 MBR post-treatment system operated by Silva-Teira et al. (2017).

Two main periods were distinguished during this study. In Period I (days 0–167), suspended and attached-to-carriers biomass were observed in the IFAS post-treatment system. Period II (days 168-407) was characterized by the presence of only attached-tocarriers biomass because all the suspended biomass was unintentionally washed out from the IFAS system.

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#### 163 **2.2 Batch tests experimental procedure**

In order to better understand the processes of denitrification and oxidation of methane in 164 the proposed IFAS system, batch assays were performed in 500 mL flasks on day 407, 165 using only biofilm adhered to Biochip and Levapor biofilm carriers. Thus, biomass 166 conditions were specific to that given period. Processes such as anaerobic ammonium 167 oxidation (anammox;  $NH_4^+ + NO_2^-$ ), conventional heterotrophic denitrification (Acetate 168  $+ NO_{3}$ , and anaerobic methane oxidation coupled to denitrification processes, using 169 nitrate and nitrite  $(CH_4 + NO_3/NO_2)$  as electron acceptors, were evaluated. Levapor 170 and Biochip biofilm carriers were extracted from the anoxic and aerobic compartments 171 172 of the IFAS system, respectively. In the Levapor assays, the flasks were filled with 324 mL of a phosphate buffer solution (143 mg KH<sub>2</sub>PO<sub>4</sub>  $L^{-1}$  and 740 mg K<sub>2</sub>HPO<sub>4</sub>  $L^{-1}$ ) and 173 27 foam biofilm carriers (76 mL of volume, 19% apparent volume). In the Biochip 174 assays, the bottles were filled with 221 biofilm carriers (92 mL of volume, 23% 175 176 apparent volume) and 302 mL of the same buffer solution. The exact volume capacity of the flask bottles was 539 mL. The biofilm did not detach from the carriers and was 177 178 washed three times with the phosphate buffer solution to ensure the absence of organic matter or nitrogen species. After adding the carriers into the flask bottles, they were 179 180 flushed for 5 min with nitrogen or methane, according to the conditions of the tests, to assure anoxic conditions and provide electron donors, respectively. For the  $CH_4 + NO_2^{-1}$ 181 /NO<sub>3</sub><sup>-</sup> tests, the headspace of the flasks was saturated in CH<sub>4</sub>. Stock solutions with 182 183 NH<sub>4</sub>Cl, NaNO<sub>2</sub>, and CH<sub>3</sub>COONa·3H<sub>2</sub>O were prepared. Then, 1 mL of each stock solution was provided to the corresponding bottle at the beginning of the experiment. 184 Ammonium and nitrite concentrations of 20 and 30 mg N  $L^{-1}$ , respectively, and COD<sub>T</sub> 185 concentrations of 200 mg L<sup>-1</sup>, were initially added. The flasks were incubated at 25 °C 186 187 and shaken at 150 rpm for 8 h. Gas and liquid samples were collected every 2 h through the bottle's septum. The final sample was taken after 23 h. Once liquid samples were taken using a syringe, they were immediately filtered through a 0.45  $\mu$ m nitrocellulose membrane filter (HA, Millipore). Gas samples were also taken for analysis in the CH<sub>4</sub> + NO<sub>3</sub><sup>-</sup>/NO<sub>2</sub><sup>-</sup> tests.

192

#### 193 **2.3 Liquid and gas samples analysis**

Nitrogen species (nitrite, nitrate, and ammonium), Total (COD<sub>T</sub>) and soluble chemical 194 oxygen demand (COD<sub>s</sub>), mixed-liquor total and volatile suspended solids (MLTSS and 195 196 MLVSS), pH, temperature, and oxidation-reduction potential (ORP) were measured according to Standard Methods (Rice et al., 2012). The dissolved oxygen (DO) 197 198 concentration in both compartments was also periodically measured by a multiparameter meter (Hach HQ40d), with a luminescent optical probe (IntelliCAL 199 200 LDO101). Oxygen concentration in the aerobic chamber was not controlled throughout 201 the operation. The dissolved methane concentration in the liquid phase of the anoxic and 202 the aerobic compartments was measured according to the procedure described by 203 Sánchez et al. (2016). A MilliGascounter type MGC-10 (Ritter) was used to measure 204 the biogas production, while the biogas composition, including CO<sub>2</sub>, N<sub>2</sub>, and CH<sub>4</sub>, was 205 analyzed using a gas chromatograph 5890 series II (Hewlett-Packard) with a thermal conductivity detector and a column of Porapack Q 80/100 2 m x 1/8" (SUPELCO), with 206 column and detector temperatures of 34 °C and 110 °C, respectively, and helium as 207 208 carrier gas. All the aforementioned parameters were measured at least twice a week.

209

#### 210 2.4 DNA extraction and 16S rRNA gene amplicon sequencing

Biomass samples were taken from Levapor packing materials on days 175, 265, and 211 212 368. Furthermore, the attached biomass composition was also analyzed in Biochip 213 carriers, on day 368. Unfortunately, suspended biomass communities were not 214 evaluated due to complete washing of biomass detected from day 167 onwards. Microbial communities of the two carriers were monitored by sequencing the V3V4 and 215 V2V3 regions of the 16S rRNA gene for bacteria and archaea, respectively. Amplicon 216 libraries, sequencing, and data analysis were performed as described in the 217 Supplementary Material. 218

219

#### 220 **3 Results and discussion**

#### 221 **3.1 General results**

The pilot plant was continuously operated over a period of 407 days, with an average 222 temperature value of 21.1  $\pm$  1.4 °C. A daily feed flow of 138.3  $\pm$  21.9 L d<sup>-1</sup> was treated, 223 with COD<sub>T</sub> and COD<sub>S</sub> concentrations of 891  $\pm$  214 and 782  $\pm$  204 mg COD L<sup>-1</sup>, which 224 resulted in an average organic loading rate (OLR) of around 1 kg COD  $m^{-3} d^{-1}$  (referred 225 to the UASB system). Average HRT values of  $21.1 \pm 3.1$  and  $9.8 \pm 1.5$  h were applied 226 227 for the UASB and IFAS post-treatment systems, respectively. Considering both the 228 external and internal recirculation in the IFAS system, a constant R of  $3 \pm 0.6$  was applied throughout the operation. A pH value of  $7.26 \pm 0.26$  was observed in the UASB 229 throughout the operation, while values of  $7.6 \pm 0.1$  and  $7.9 \pm 0.2$  were determined in the 230 231 anoxic and aerobic compartments of the IFAS system, respectively. During the entire 232 experimentation period, a high average  $COD_T$  removal efficiency of 93  $\pm$  2% was 233 detected in the first methanogenic stage. From the beginning of the operation, high COD removals were observed in the UASB reactor due to the large amount of anaerobic 234

granules used as inoculum. Approximately 75% of the overall COD inlet was 235 methanized in the UASB reactor, producing a biogas flow of  $37.3 \pm 8.5 \text{ L d}^{-1}$ , with the 236 following composition:  $75 \pm 2.5\%$  CH<sub>4</sub>,  $15 \pm 2.8\%$  CO<sub>2</sub>, and  $9 \pm 3\%$  N<sub>2</sub>. Nevertheless, 237 not all methane produced was present in the gas phase, but  $17.6 \pm 2.9 \text{ mg CH}_4 \text{ L}^{-1}$ 238 dissolved in the anaerobic effluent were detected, representing the  $14 \pm 6\%$  of the total 239 methane produced. Low COD<sub>T</sub> and COD<sub>S</sub> concentrations were detected in the anaerobic 240 effluent, with values of only 57  $\pm$  15 and 43  $\pm$  12 mg COD L<sup>-1</sup>, respectively, excluding 241 242 in both measurements the portion related to the dissolved methane COD. This organic matter could be used as an electron donor in the pre-anoxic compartment of the IFAS 243 system, promoting heterotrophic denitrification processes. Nevertheless, COD<sub>T</sub> was also 244 detected in the final effluent of the system, at values of  $77 \pm 21.4$  and  $53.8 \pm 12$  mg 245  $COD L^{-1}$ , during Periods I and II, respectively. During Period I, due to the continuous 246 247 VSS washout in the IFAS system, COD concentrations in the final effluent were even higher than those detected in the UASB effluent. Regarding Period II, COD 248 249 concentrations in the UASB outlet and in the final effluent were similar. This fact 250 indicates that the biodegradable COD fraction of the UASB effluent is very low. Therefore, conventional heterotrophic denitrifiers, using the remaining COD of the 251 UASB effluent (excluding methane), should have played a minor role in the nitrogen 252 253 removal processes detected in the IFAS post-treatment system.

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#### **3.2 Impact of biofilm carriers' addition on the nitrification potential**

Most of the total nitrogen (TN) of the UASB effluent was present in the form of ammonium, with an average concentration value of  $54 \pm 8 \text{ mg NH}_4^+$ -N L<sup>-1</sup> (Figure S3). Oxygen was continuously provided in the aerobic compartment to allow nitrification processes, and average DO concentration values of  $3.2 \pm 0.7$  and  $4.4 \pm 0.6$  mg O<sub>2</sub> L<sup>-1</sup> were measured during Periods I and II, respectively.

Whenever the Biochips' apparent volume was higher, the ammonium oxidizing rates, 261 262 referred to the aerobic compartment, accordingly increased. Average values of  $157.1 \pm$ 61.5, 165.4  $\pm$  34.7, 191.5  $\pm$  47.9, and 240.6  $\pm$  55.5 mg NH<sub>4</sub><sup>+</sup>-N L<sup>-1</sup> d<sup>-1</sup> were attained for 263 the aforementioned periods, respectively (Figure S4). Nevertheless, despite the supports 264 265 addition, full ammonium oxidation was not accomplished during the study and concentrations of 25.3  $\pm$  12.3, 23.2  $\pm$  5.2, 23.2  $\pm$  6.7, and 19.3  $\pm$  6.5 mg NH<sub>4</sub><sup>+</sup>-N L<sup>-1</sup> 266 were measured (Figure S5). During Period II, the ammonia consumption rate referred 267 per unit of suspended carrier surface was of  $0.59 \pm 0.15$  g NH<sub>4</sub><sup>+</sup>-N m<sup>-2</sup> d<sup>-1</sup>. 268

Although the suspended biomass was totally washed-out from the IFAS system on day 269 167, significant ammonium oxidation rates were achieved during Period II (without the 270 presence of suspended biomass in the IFAS system), indicating that biofilms were responsible 271 for the removal of ammonium observed. This highlights the essential role of biofilm 272 carriers in nitrification processes. In order to accomplish better nitrification rates, a 273 higher surface area of biofilm carriers is required by increasing either the aerobic 274 275 compartment volume or the carrier added volume. A second strategy is to generate higher DO concentrations in the aerobic compartment, which is usually the limiting 276 substrate in biofilm processes. 277

Partial nitritation processes were observed in the aerobic compartment during most of Period I. In fact, throughout the first 119 days of operation, nitrite was the main available electron acceptor, with an average concentration of  $3.5 \pm 1.5 \text{ mg NO}_2^{-}\text{N L}^{-1}$ (Figure S5).

From that day onwards, the nitrite concentration considerably declined due to the nitrite-oxidizing bacteria (NOB) activation. As a result, nitrate became the most

abundant oxidized nitrogen species in the aerobic stage. During most of Period I, the 284 estimated average free ammonia (FA) concentration of 0.81 mg NH<sub>3</sub>-N L<sup>-1</sup> may have 285 partially inhibited the NOB activity. It has been reported in the literature that FA values 286 between 0.1 and 1.0 mg NH<sub>3</sub>-N L<sup>-1</sup> I initiated the inhibition of NOB activity 287 (Anthonisen et al., 1976). Nevertheless, over time, the biomass colonization of the 288 packing media could have provided a more suitable environment for NOB activity to 289 tolerate the inhibitory effect of FA (Villaverde et al., 2000). During Period II, a nitrite 290 concentration of only  $0.25 \pm 0.17$  mg NO<sub>2</sub><sup>-</sup>N L<sup>-1</sup> was detected. 291

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#### **3.3 Suspended biomass washout in the IFAS system**

One of the main concerns of the present study was related to the suspended biomass retention capacity of the IFAS system when treating low-C/N ratio UASB effluents, mainly due to the reports regarding the poor settling properties of suspended biomass when treating effluents with these characteristics (Metcalf & Eddy et al., 2013). At the beginning of the operation, the MLTSS and MLVSS concentrations measured in the pre-anoxic compartment were of 3.5 and 3.1 g L<sup>-1</sup> (Figure 2), respectively.

300 Nevertheless, from day 47 to 89, as a consequence of the low mixing in the anoxic compartment, large amounts of biomass gradually accumulated in the bottom part, 301 302 resulting in a dramatic drop in the suspended solids concentration. Once the mechanical mixing was increased on day 83, the mixed liquor was immediately resuspended, 303 obtaining a homogeneous mixture, which increased the MLVSS concentration. Between 304 days 109 and 167, the MLVSS concentration declined in the IFAs system from 2.62 to 305 only 0.05 g L<sup>-1</sup>. During that period, sludge washout was observed in the effluent of the 306 307 IFAS system, with MLTSS and MLVSS concentrations of approximately  $61 \pm 11$  and  $45 \pm 12 \text{ mg L}^{-1}$ , respectively. The average liquid up-flow velocity in the settler 308

compartment was 0.31 m<sup>3</sup> m<sup>-2</sup> h<sup>-1</sup> throughout the entire operation. Curiously, the biomass washout was concomitant with the period in which nitrite began to oxidize in the aerobic compartment, making nitrate the main electron acceptor. From day 167 onwards, the MLTSS and MLVSS observed in the final effluent were around 39.9  $\pm$ 11.4 and 33.7  $\pm$  9 mg L<sup>-1</sup>, respectively. Once all suspended biomass was totally removed from the IFAS post-treatment system (Period II), the role of biofilm carriers in nitrogen and methane removal-processes was thoroughly studied.

The low average OLR applied to the IFAS system throughout the entire operation, 0.14 kg COD m<sup>-3</sup> d<sup>-1</sup>, and the poor organic removal rate observed in the IFAS system due to the low biodegradable COD content of the UASB effluent could be the main causes responsible for the biomass washout. Consequently, the only way of treating these kinds of anaerobic effluents with low C/N ratios is by promoting biofilm growth adhered to supports. Therefore, if the complete retention of suspended biomass is intended, a preanoxic MBR system, such as that done by Alvarino et al. (2019), must be used.

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#### 324 **3.4 Methane removal in the IFAS post-treatment system**

Methane removals in the anoxic compartment of the IFAS system were calculated using 325 mass balances (Supplementary Material). During Period I, an upward trend in methane 326 327 removal was clearly observed, attaining a maximum volumetric methane removal rate (MRR) and methane removal efficiency of 52.5 mg CH<sub>4</sub>  $L^{-1} d^{-1}$  and 72.7% on day 126, 328 respectively (Figure 3), both values referred only to the anoxic compartment. 329 330 Nevertheless, once the suspended biomass from the IFAS system was lost, and nitrate became the main electron acceptor detected, methane removals significantly dropped 331 into the anoxic compartment. During Period II, values of  $23.9 \pm 10.4 \text{ mg CH4 } \text{L}^{-1} \text{ d}^{-1}$ 332 and  $36.6 \pm 10.5\%$  were attained, respectively. Thereby, the complete loss of suspended 333

biomass in the IFAS system led to a reduction of around 50% in both parameters. In
spite of the negative effect of the suspended biomass washout, the methane removals
detected during Period II, highlight the important role of the adhered biomass in the
methane oxidation bioprocesses.

The recirculations applied in the post-treatment system will determine the amount of 338 339 DO provided to the anoxic compartment, which could cause the proliferation of aerobic methanotrophs. If it is assumed that all oxygen reduced in the anoxic stage during 340 Period I and II,  $32.5 \pm 9.3$  and  $41.8 \pm 5.9$  mg O<sub>2</sub> L<sup>-1</sup> d<sup>-1</sup>, respectively, was consumed 341 through aerobic methanotrophs, and considering its stoichiometry (2 g O<sub>2</sub> g<sup>-1</sup> CH<sub>4</sub>; 342 Equations 1 and 2), these microorganisms could explain up to 48% and 87% of the 343 average MRRs observed in the anoxic compartment during Period I and II, respectively. 344 Besides aerobic methane oxidation processes, some other metabolic pathways could be 345 346 involved, especially during Period I, such as anaerobic methane oxidation and even the 347 possibility of other lesser-studied metabolic pathways, such as the HYME-D suggested 348 by Cao et al. (2019).

349 In previous studies, Silva-Teira et al. (2017) treating the same type of domestic sewage, under similar operating conditions, but using an MBR instead of an IFAS system, 350 achieved a maximum MRR of  $195 \pm 17$  mg CH<sub>4</sub> L<sup>-1</sup> d<sup>-1</sup> in the anoxic compartment. This 351 352 value is almost four times higher than the maximum attained in the current research (52.5 mg CH<sub>4</sub>  $L^{-1}$  d<sup>-1</sup>). This fact could be attributed to the presence of an ultrafiltration 353 membrane, which ensured the complete retention of suspended biomass in the MBR. 354 355 Furthermore, they observed that MRRs depended on the applied feed flow, which was approximately three times higher in the previous study (355  $L_{feed} d^{-1}$ ) than in the current 356 one (138  $L_{feed} d^{-1}$ ). However, the highest methane removal efficiencies observed in the 357 anoxic compartment during this research, with the presence of nitrite and suspended 358

biomass (78%), were similar to the maximum attained by Silva-Teira et al. (2017). This suggested that the MRR observed during Period I, with the presence of microorganisms growing in biofilms and in suspension, was limited by the hydraulic load applied to the system. During Period II, in which only biofilms were responsible for the methane oxidation, MRRs could be limited by the transfer of either methane or oxygen to the biofilms.

365 The dissolved methane that was not oxidized in the first anoxic compartment was led to 366 the aerobic one. This methane could be biologically oxidized or stripped-off into the atmosphere. Throughout the entire operation, dissolved methane was detected in the 367 aerobic compartment, with values ranging from 2.10 to 0.01 mg CH<sub>4</sub> L<sup>-1</sup>. Average 368 concentrations of 0.25  $\pm$  0.2 and 0.32  $\pm$  0.4 mg CH<sub>4</sub> L<sup>-1</sup> were measured during Periods I 369 370 and II, respectively. The continuous detection of methane allowed to estimate the 371 amount of this compound that was either desorbed or biologically oxidized in the 372 aerobic compartment (Supplementary Material).

Average MRRs of 53.1  $\pm$  33.4 and 43.5  $\pm$  20.3 mg CH<sub>4</sub> L<sup>-1</sup> d<sup>-1</sup> were estimated in the 373 374 aerobic compartment during Period I and II, respectively. This indicates the crucial role of the aerobic compartment to minimize GHGs emissions from the process. In fact, the 375 376 average MRRs observed in the aerobic compartment for Period I are similar to the 377 maximum achieved in the anoxic compartment, and almost double the average values 378 achieved during Period II. Considering the methane removals observed in the anoxic compartment and those estimated in the aerobic compartment, high overall removal 379 380 efficiencies of 89.1  $\pm$  10% and 77  $\pm$  12% were achieved in the whole IFAS posttreatment system during Periods I and II, respectively. The remaining fraction of 381 382 methane that was not biologically oxidized was considered to be stripped off into the atmosphere. Due to the high oxygen concentration measured in the aerobic 383

compartment, a high fraction of the methane was surely completely oxidized to carbon
dioxide. However, the development of AMO-D processes should not be discarded due
to the presence of anoxic environments inside the Biochip.

387

#### 388 **3.5** Nitrogen removal performance in the IFAs post-treatment system

Similarly, methane mass balances in the anoxic and aerobic compartments, and in the 389 whole IFAS post-treatment system were carried out. An average value of  $54 \pm 8$  mg 390 NH<sub>4</sub><sup>+</sup>-N L<sub>feed</sub><sup>-1</sup> was observed in the UASB throughout the research. In order to promote 391 denitrification processes in the anoxic compartment, previous ammonium oxidation into 392 oxidized nitrogen compounds  $(NO_2^- \text{ and } NO_3^-)$  is required. Due to the low organic 393 394 matter content observed in anaerobic effluents, the use of a coexisting alternative 395 electron donor to denitrify, such as ammonium or dissolved methane, is required to 396 accomplish the denitrification processes. From the beginning of the study, the overall 397 TN removal progressively increased throughout the whole IFAS system, reaching a maximum removal of 32.5 mg TN  $L_{feed}^{-1}$  on day 109 (Figure 4). Nevertheless, nitrogen 398 399 removals dropped between days 126 and 167. This decrease was concomitant to the suspended biomass loss and also with the lower presence of nitrite in the aerobic 400 compartment. However, although only attached biomass was available throughout 401 Period II, an important average nitrogen removal of  $18 \pm 3.9$  mg TN  $L_{feed}^{-1}$  was attained. 402 Regarding the nitrogen removal rate (NRR), a maximum value of 119.6 mg TN  $L^{-1} d^{-1}$ 403 and an average value of  $29.9 \pm 17.1 \text{ mg TN } \text{L}^{-1} \text{ d}^{-1}$  for Periods and II, respectively, were 404 observed in the first anoxic compartment of the IFAS system. Curiously, denitrification 405 406 processes were not only restricted to this anoxic compartment but were also 407 continuously detected in the aerobic compartment from the beginning of the study. This 408 fact could be demonstrated because the average TN concentration in the final effluent

was 2.9  $\pm$  1.7 mg N L<sub>feed</sub><sup>-1</sup> lower than the measured in the anoxic compartment. 409 Differences in the TN concentrations between these two streams were permanently 410 411 detected throughout the study (Figure S3). Using mass balances, it was calculated that  $45.5 \pm 24.4$  and  $54.9 \pm 24.2\%$  of the TN removal achieved was surprisingly carried out 412 in the aerobic compartment in the IFAS system during Period I and II, respectively. The 413 existence of different DO concentration gradients, which is characteristic of biofilm 414 415 systems, may promote anoxic environments in the inner layers of the Biochip carriers, explaining the observed nitrogen removals in the aerobic compartment. This evidences 416 that simultaneous nitrification-denitrification processes were conducted in the Biochip 417 carriers. 418

The COD concentration observed in the final effluent of the post-treatment system was 419 higher or similar than in the UASB effluent. This could indicate a minor role of 420 421 conventional heterotrophic denitrifiers in the nitrogen removal achieved in the IFAS 422 system. Therefore, other compounds have been alternatively used as an electron donor 423 to denitrify. Part of the nitrogen removal could be explained by means of conventional 424 heterotrophic denitrifiers, using the methane oxidation products produced by aerobic methanotrophs (AMO-D) in the anoxic and aerobic compartments, or by other methane 425 426 oxidation pathways coupled to denitrification processes, such as anaerobic methane 427 oxidation, or HYME-D. Furthermore, the coexistence of nitrite and ammonium under 428 anoxic conditions could induce the development of anammox microorganisms in the 429 anoxic and aerobic compartments, contributing to improving the nitrogen removals.

430 The maximum nitrogen removals observed in this research (32.5 mg TN  $L_{feed}^{-1}$ ) were 431 similar to the maximum achieved using a hybrid MBR system (35.1 mg TN  $L_{feed}^{-1}$ ; 432 Alvarino et al., 2019). In both cases, nitrite was present and enhanced nitrogen 433 removals. Furthermore, even without suspended biomass in the IFAs post-treatment

system, high average TN removals were attained (18  $\pm$  3.9 mg TN  $L_{\text{feed}}^{-1}$ )., similar 434 values to those achieved by Silva-Teira et al. (2017; 15 mg TN  $L_{feed}^{-1}$ ) using the MBR 435 system. In both cases, the nitrite concentration was negligible, with nitrate being the 436 main electron acceptor. Therefore, the current outcomes emphasize the key role 437 performed by the aerobic compartment, with the presence of biofilm carriers, in 438 nitrogen removal processes. It should be emphasized that the use of an IFAS system, 439 with lower MRR, but similar dissolved methane and total nitrogen removal than the 440 MBR system, could be simpler and cheaper, to improve those sewage treatment plants 441 that treat municipal wastewater with UASB systems, as occurs in some warm and 442 443 temperate countries of the world.

444

#### 445 **3.6 Microbial community analysis**

446 During Period II, the microbial community was monitored in the two types of free-447 moving biofilm carriers presented in the reactor (Levapor and Biochip), by means of 448 partial amplicon sequencing of the 16S rRNA gene. Three samples of the attached 449 biomass in Levapor (operating days 175, 265, and 368) and a sample of the Biochip (day 368) biomass were analyzed. Due to the observed biomass washout, suspended 450 biomass could not be studied. On days 175, 265, and 368, several aerobic 451 452 methanotrophs (Methylomonas, Methylocaldum, Crenotrix, Methylosiuns, and a putative methanotroph from the Methylococcales order) were detected in Levapor 453 carriers with total relative abundances of 3.20, 3.48, and 2.90%, respectively (Figure 454 455 5a). On day 368, its total relative abundance in Biochip carriers was even higher (7.40%), which could explain the high methane removals observed in the aerobic 456 457 compartment of the IFAS system.

In contrast, n-damo bacteria (Ca. Methylomirabilis) were only detected in Levapor 458 459 carriers (anoxic compartment) in the last days of operation (day 368), with a relative abundance of 0.02%, while n-damo archaea were not detected. Biochips carriers were 460 461 quite enriched in nitrifying organisms during the last days of operation when more than 10% of the bacteria belonged to Nitrospira genus (Figure 5b). Furthermore, ammonium-462 463 oxidizing bacteria of the *Nitrosomonadaceae* family were detected in Biochip carriers 464 with a relative abundance of 3%. As ammonia and low concentrations of nitrite were available, conditions for the growth of anammox bacteria were present in the system. 465 Indeed, Ca. Brocadia was present with a relative abundance of 0.03%, 1.20%, and 466 467 3.18% in Levapor carriers on days 175, 265, and 368, respectively, and with a relative abundance of 6.5% in Biochip carriers on day 368. During Period II, the rest of the 468 469 community was composed mostly of bacterial families of heterotrophs considered 470 saprophytic microorganisms, such as Saprospiraceae or Moraxellaceae (Martins and 471 Carreira, 2014). There also species able to carry out nitrate reduction, such as the 472 Comamonadaceae family, although their relative abundance decreased through the 473 operation of the IFAS system (Figure 6a). Most Archaea present in both, Levapor and Biochip, were well known methanogenic genera, such as Methanosaeta, Candidatus 474 475 Methanoregula, Methanobacterium, and Methanomethylovorans, among others (Figure 476 6b).

These data indicated that during Period II, methane removal was surely achieved in both compartments of the IFAS system through aerobic methanotrophs. Meanwhile, the nitrogen removals observed in Levapor carriers could have been carried out by anammox bacteria or conventional heterotrophic denitrifiers that use methane oxidation products released by aerobic methanotrophs. Moreover, the consortium of nitrifying and

20

482 anammox bacteria probably contributed to the nitrogen removals observed in the483 adhered-to-Biochip biomass.

484

#### 485 **3.7 Batch assays**

Batch experiments were performed under anaerobic conditions at the end of Period II to 486 determine whether anaerobic ammonium oxidation (anammox), conventional 487 heterotrophic denitrification, and anaerobic methane oxidation along with denitrification 488 processes were feasible in both biofilm media carriers (Levapor and Biochip). The batch 489 experiments containing  $NO_2^-$  and  $NH_4^+$  in the liquid, and  $N_2$  in the gas phase, showed 490 high denitrification in both carriers' media. Regarding Levapor, high volumetric NRRs 491 of 80.1 and 52.8 mg N L<sup>-1</sup> d<sup>-1</sup> for nitrite and ammonium were observed, respectively 492 (Figure 7a). The resulting  $NO_2^{-}$ -N/NH<sub>4</sub><sup>+</sup>-N ratio (1.5) is similar to that required for the 493 494 anammox process (1.32). The nitrite and ammonium removal rates per unit of Levapor surface were 0.87 and 0.58 g N  $m^{-2} d^{-1}$ , respectively (Table 1). 495

496

# Table 1. Nitrogen removal rates per unit of Levapor and Biochip surface for the different batch tests.

499

500		Nitrogen removal rate (g N m <sup>-2</sup> d <sup>-1</sup> )		
501	Batch test (type)	Levapor	Biochip	_
502	$NO_{2}^{-} + NH_{4}^{+}$	NO <sub>2</sub> <sup>-</sup> -N: 0.87 NH <sub>4</sub> <sup>+</sup> -N: 0.58	NO <sub>2</sub> <sup>-</sup> -N: 0.49 NH <sub>4</sub> <sup>+</sup> -N: 0.29	
503	Acetate $+ NO_3$	0.83	0.27	
	$CH_4 + NO_3$	0.17	-	
504	$CH_4 + NO_2$	0.29	-	

505

506 In the Biochip carriers, higher volumetric NRRs were observed. In fact, nitrite was totally consumed in less than 4 h. Given the lack of nitrite as electron acceptor between 507 4 and 23 h, the ammonium concentrations remained constant with values of around 10.5 508 mg N L<sup>-1</sup> (Figure 7b). The maximum volumetric removal rates of nitrite and ammonium 509 achieved were 246.2 and 142.4 mg N L<sup>-1</sup> d<sup>-1</sup>, respectively. The nitrite and ammonium 510 surface removal rates for Biochip carriers were 0.49 and 0.29 g N m<sup>-2</sup> d<sup>-1</sup>, respectively. 511 Therefore, the presence of anammox processes in both biofilm carriers' media was 512 clearly confirmed and even with remarkable activities. 513

With respect to the conventional heterotrophic denitrification tests using acetate as 514 electron donor, nitrate removal rates of 75.9 and 137.5 mg NO<sub>3</sub><sup>-</sup>-N  $L^{-1} d^{-1}$  were achieved 515 for Levapor and Biochip, respectively (Figures 7c and 7d). The nitrate removal rates per 516 unit of support surface were 0.83 and 0.27 g NO<sub>3</sub><sup>-</sup>-N m<sup>-2</sup> d<sup>-1</sup>, respectively. These results 517 518 pointed out the presence of conventional heterotrophic denitrification processes in both 519 biofilm carriers. However, the low concentration of organic matter provided to the IFAs 520 post-treatment system possibly limited the denitrification rates achieved during the 521 continuous operation.

In the anaerobic methane oxidation tests performed only with Levapor carriers and 522 using nitrate as an electron acceptor, a low NRR of 15.7 mg NO<sub>3</sub><sup>-</sup>-N L<sup>-1</sup> d<sup>-1</sup> was 523 524 observed (Figure 7e), which corresponds to a removal rate per unit of Levapor surface of 0.17 g NO<sub>3</sub><sup>-</sup>-N m<sup>-2</sup> d<sup>-1</sup>. At the same time, using nitrite as an electron acceptor instead 525 of nitrate, removal rate values of 22.6 mg NO<sub>2</sub><sup>-</sup>-N  $L^{-1} d^{-1}$  and 0.29 g NO<sub>2</sub><sup>-</sup>-N m<sup>-2</sup>  $d^{-1}$  were 526 527 attained (Figure 7f). Therefore, the NRRs achieved using nitrite as an electron acceptor were 1.44 times higher than for nitrate. For the nitrate test, if the total  $CH_4$  and  $NO_3$ -N 528 consumed are considered, a ratio of 0.61 mol  $CH_4$  mol<sup>-1</sup> NO<sub>3</sub><sup>-</sup>-N was attained. 529 Regarding the nitrite tests, the same ratio was achieved (0.61 mol  $CH_4$  mol<sup>-1</sup> NO<sub>2</sub><sup>-</sup>-N). 530

531 These ratios are similar to those of anaerobic methane oxidation or HYME-D processes532 (Cao et al., 2019).

533

#### 534 4. Conclusions

The feasibility of using an IFAS system as a post-treatment alternative to mitigate the 535 environmental impact of UASB treating sewage, in terms of greenhouse gas emissions 536 537 and nutrients, was evaluated. High overall COD removal efficiencies were achieved 538 during the whole experimentation period, with  $COD_T$  concentrations in the final effluent that never exceeded 77 mg L<sup>-1</sup>. Regarding ammonium removal, an average  $57 \pm 15.7\%$ 539 540 was achieved. The accomplished ammonium oxidation rates were improved by the gradual addition of Biochip carriers in the aerobic compartment. Therefore, a larger 541 volume of the aerobic compartment, a higher biofilm carrier surface, or higher DO 542 543 concentrations would have been required in order to achieve complete nitrification.

544 The maximum nitrogen removal and methane removal efficiencies achieved throughout the IFAS system, 32.5 mg TN L<sub>feed</sub><sup>-1</sup> and around 93%, respectively, were detected 545 546 during the period in which suspended biomass and nitrite were present. Once the suspended biomass was completely washed out, and nitrate became the main electron 547 acceptor, this value progressively decreased. However, despite the absence of 548 549 suspended biomass, average nitrogen removals and methane removal efficiencies of 18  $\pm$  3.9 mg TN L<sub>feed</sub><sup>-1</sup> and 77.2  $\pm$  12.2% were attained, respectively, in the IFAS system. 550 Therefore, the use of an IFAS system could be an alternative to reduce the problems 551 552 related to high nutrient concentrations and greenhouse emissions of UASB systems used in many warm and temperate countries as a secondary sewage treatment method. 553 554 During the study, dissolved methane and nitrogen were not only removed in the anoxic

compartment, but also in the aerobic one, and with significant importance. The results

556 obtained pointed out the importance of both biofilm carriers, Levapor and Biochip, in 557 the methane and nitrogen removals observed in the IFAS system. The significant 558 detection in Biochip carriers of aerobic methane oxidizers and *Ca*. Brocadia confirmed 559 the existence of methane and nitrogen removal processes in the aerobic compartment of 560 the IFAS system. Both groups of microorganisms were also detected in the Levapor 561 carriers present in the anoxic compartment.

562

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#### 692 Supplementary Material

#### 693 **1. Mass balances calculations.**

The volumetric removal rates of oxygen, methane and nitrogen (mg  $L^{-1} d^{-1}$ ) were calculated in the anoxic compartment of the IFAS system by applying mass balances (Equation S1). Similarly, methane and nitrogen removal rates were also calculated in the aerobic compartment by means of Equation S2. It was taken into consideration that the approximation to an ideal continuous stirred tank reactor (CSTR) can be applied to both compartments.

700 Removal rates = 
$$\frac{Q_f (C_x)_{UASB} + (Q_{R1} + Q_{R2}) (C_x)_{aero} - (Q_f + Q_{R1} + Q_{R2}) (C_x)_{anox}}{V_{anox}}$$
(S1)

701 
$$Removal \, rates = \frac{(Q_f + Q_{R1} + Q_{R2}).(C_{x_{anox}} - C_{x_{aero}})}{V_{aero}}$$
(S2)

Where  $Q_f$  is the feeding flow (L d<sup>-1</sup>) applied to the combined UASB-IFAS system; C<sub>x</sub> is the concentration of oxygen, methane and nitrogen in the liquid phase (mg L<sup>-1</sup>) of each compartment: UASB, aerobic (aero) and anoxic (anox); Q<sub>R1</sub> is the recirculation flow from the aerobic to the anoxic compartment of the IFAs system (L d<sup>-1</sup>); Q<sub>R2</sub> is the recirculation flow from the settler to the anoxic compartment of the IFAS system (L d<sup>-1</sup>); Q<sub>R2</sub> is the recirculation flow from the settler to the anoxic compartment of the IFAS system (L d<sup>-1</sup>); i, and V is the compartment volumen (36 and 20 L for anoxic and aerobic, respectively).

However, not all the dissolved methane that was led from the anoxic to the aerobic compartment was biologically oxidized, but a fraction of it could be also stripped-off into the atmosphere. The continuous detection of methane in the aerobic compartment allowed to estimate the amount of methane that was desorbed from the aerobic compartment. To this end, the oxygen volumetric mass transfer coefficient ( $k_La$ ) was calculated (Equation S3).

715 
$$k a_{0_2} = \frac{ORR_{aerobic}}{(DO_{saturation} - DO_{aerobic})}$$
 (S3)

716 Where  $k_L a_{O2}$  represents the volumetric mass transfer coefficient for oxygen (d<sup>-1</sup>); ORR, 717 corresponds to the measured oxygen removal rate in the aerobic compartment (mg O<sub>2</sub> L<sup>-</sup> 718 <sup>1</sup> d<sup>-1</sup>); DO<sub>saturation</sub>, represents the estimated dissolved oxygen saturation concentration at 719 the given temperature (mg O<sub>2</sub> L<sup>-1</sup>); DO<sub>aerobic</sub>, is the dissolved oxygen concentration 720 measured in the aerobic compartment (mg O<sub>2</sub> L<sup>-1</sup>).

721 By considering  $k_{L}a_{CH4} = 0.8 \cdot k_{L}a_{O2}$  (Sánchez et al., 2016), the methane desorbed in the 722 aerobic compartment can be calculated as follows:

723 
$$CH_{4 \text{ desorbed}} = C_{CH_4} \cdot 0.8 \cdot k \ a_{O_2} \tag{S4}$$

Where  $CH_{4desorbed}$  represents the amount of methane desorbed per unit of volume of the aerobic compartment and per unit of time (mg  $CH_4$   $L^{-1}$   $d^{-1}$ ) and  $C_{CH4}$  is the concentration of methane measured in the aerobic compartment (mg  $CH_4$   $L^{-1}$ ).

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#### 728 **2. DNA extraction, sequencing and analysis.**

To characterize the microbial community, total genomic DNA was extracted using the 729 730 Biofilm DNA Isolation Kit (Norgen, Thorold, Canada) following the manufacturer 731 instractions. Then, total DNA concentrations were determined using a Qubit fluorometer (Thermo Fisher Scientific, Waltham, MA, USA). DNA size and integrity 732 was tested by standard electrophoresis. The V3V4 region of the bacterial 16S rRNA 733 734 gene was amplified with the primer pair S-D-Bact-0341-b-S-17 and S-D-Bact-0785-a-A 735 (Klindworth et al. 2013). The V2V3 region of the archaeal 16S rRNA gene was amplified with the primer set Arch1F and Arch1R (Cruaud et al. 2014). For that, an 736 737 initial amplification was carried out in 25 µL volumes containing 3 ng of total DNA, 738 100 nM of bacterial primers or 200 nM of archaeal primers, and 1X Q5® High Fidelity Master Mix (New England BioLabs) which contains the DNA polymerase, 2mM MgCl<sub>2</sub> 739 740 and 200 µM dNTPs. PCR conditions were: initial denaturation at 98°C for 30 s 741 followed by 20 (Bacteria) or 22 (Archaea) cycles (denaturation: 98°C for 10 s, annealing: 50°C for Bacteria or 48°C for Archaea during 20 s, extension: 72°C for 20 742 743 s); followed by an extension step of 2 min at 72°C. A second PCR was used to add the 744 Illumina adapters and barcodes to the amplicons. The conditions were similar to the first 745 PCR but in this case, only 15 cycles were applied and the annealing temperature was 746 60°C. Library DNA concentration was then determined in a Bioanalyzer (Bioanalyzer, Agilent Technologies, Santa Clara, CA, USA). Libraires were pooled in equimolar 747 amounts and sequenced in a MiSeq Illumina sequencer (Illumina, Inc.) at Unidad de 748

Genómica, Parque Científico de Madrid. Paired-end reads (2x300) were generated
following manufacturer protocols (Illumina, Inc.).

Obtained sequences were de-multiplexed and trimmed to remove Illumina adapters, 751 752 barcodes. In addition, the last 50 pb of the 5' ends were removed due to the low-quality scores (Q<30). Next, paired-end reads were merged as previously described (Eren et al. 753 754 2013) filtering sequences with quality scores below Q30 and a minimum overlap of 50 755 bp. In addition, all sequences containing indeterminations were removed from further analysis. The obtained high-quality sequences, were analyzed with VSEARCH in de 756 757 novo mode for chimera removal (Rognes et al. 2016) and clustered into Operational 758 Taxonomic Units (OTUs) using the 97% cutoff for sequences similarity using QIIME (Caporaso et al. 2010). OTUs taxonomic affiliation was determined with USEARCH 759 760 (Edgar 2010) with the Greengenes database version 13 8 (DeSantis et al. 2006).

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- Other
- Moraxellaceae unk
- Chloracidobacteria DS-100
- C.BrocadiaNitrosomonadaceae unk
- Nitrospira
- Rubrivivax
- Methylococcales unk
- Novosphingobium
- Gallionella
- Ignavibacteriaceae unk
- Rhodocyclaceae Dok59 unk
- Chlorobi OPB56 unk
- Myxococcales unk
- Rhodobacter
- Saprospirales unk
- Chloracidobacteria RB41 Ellin6075 unk
- Bacteroidales unk
- Sphingobacteriales unk
- Bacteria GN02 3BR-5F unk
- Anaerolineae envOPS12 unk
- MethylomonasOxalobacteraceae unk
- Chitinophagaceae unk
- Saprospiraceae unkComamonadaceae unk

#### Other

- Methanocorpusculum
- Methanomassiliicoccus
- Methanomassillicoccaceae unk
- Methanolinea
- Methanobacteriales WSA unk
- Methanospirillum
- Methanoregulaceae unk
- Methanomethylovorans
- C. Methanoregula
- Methanosaeta
- Methanobacterium













Figure 1. Schematic diagram of the UASB methanogenic reactor coupled to an IFAS post-treatment system.

**Figure 2.** Evolution of the mixed-liquor total (MLTSS) and volatile (MLTVSS) suspended solids concentration in the anoxic compartment.

**Figure 3.** Evolution of methane removal efficiencies and volumetric methane removal rates (MRR) in the pre-anoxic compartment of the IFAS system.

**Figure 4.** Evolution of TN removal efficiencies and TN removal observed in the IFAS post-treatment system throughout the operating period.

**Figure 5.** Relative abundances of aerobic methanotrophs (a) and of the most abundant nitrifying and anammox bacteria (b) in the Levapor (days 175, 265 and 368) and Biochips (day 368) carriers. Only the aerobic methanotrophos with relative abundances over 0.5% in at least one observation are shown.

**Figure 6.** a) Relative abundances of the 25 most dominant bacteria in the Levapor carriers on days 175, 265 and 368; and in the Biochip carriers on day 368 and b) Composition of the archaeal community in the Levapor and Biochip carriers. Numbers indicate the operational day.

**Figure 7.** Batch denitrification tests results for anammox process in Levapor (a) and Biochip (b) carriers media; for conventional heterotrophic denitrification in Levapor (c) and Biochip (d); and anaerobic methane oxidation processes in Levapor using nitrate (e) and nitrite (f) as electron acceptor.

1	Figure S1. Real image of the used UASB-IFAS integrated system (pilot plant).
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3	Figure S2. Sponge Levapor (left) and rigid-plastic Biochip (right) biofilm carriers.
4	
5	Figure S3. Evolution of the ammonium concentration fed to the IFAS system and the
6	measured TN ions concentration in the anoxic compartment and the IFAs effluent.
7	
8	Figure S4. Evolution of the ammonium oxidation rates observed in the aerobic
9	compartment of the IFAS system. Black lines separate periods of time according to the
10	apparent volume of the Biochip carriers (%). The orange line separates Period I from
11	Period II.
12	
10	

Figure S5. Evolution of different nitrogen species in the aerobic compartment of the posttreatment system.

#### **CRediT** author statement

**Tomás Allegue**: investigation, writing original draft, formal analysis, conceptualization; **María Nieves Carballo-Costa**: investigation, methodology; **Nuria Fernández González**: methodology, formal analysis, investigation; **Juan Manuel Garrido Fernández**: conceptualization, funding acquisition, supervision, project administration.