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Jennifer Y. King University of California, Santa Barbara

Leslie A. Brandt USDA Forest Service

E. Carol Adair National Center for Ecological Analysis and Synthesis

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SYNTHESIS AND EMERGING IDEAS

Shedding light on plant litter decomposition: advances, implications and new directions in understanding the role of photodegradation

Jennifer Y. King · Leslie A. Brandt · E. Carol Adair

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Abstract Litter decomposition contributes to one of the largest fluxes of carbon (C) in the terrestrial biosphere and is a primary control on nutrient cycling. The inability of models using climate and litter chemistry to predict decomposition in dry environments has stimulated investigation of non-traditional drivers of decomposition, including photodegradation, the abiotic decomposition of organic matter via exposure to solar radiation. Recent work in this

Author contributions All authors contributed equally to the manuscript and revisions.

J. Y. King (⊠) Department of Geography, University of California, Santa Barbara, CA 93106-4060, USA e-mail: jyking@geog.ucsb.edu

L. A. Brandt

Northern Institute of Applied Climate Science, USDA Forest Service, Saint Paul, MN 55108, USA e-mail: lbrandt@fs.fed.us

E. C. Adair

National Center for Ecological Analysis and Synthesis (NCEAS), University of California, Santa Barbara, CA 93101, USA e-mail: Carol.Adair@uvm.edu

Present Address:

E. C. Adair Rubenstein School of Environment and Natural Resources, University of Vermont, Burlington, VT 05405, USA developing field shows that photodegradation may substantially influence terrestrial C fluxes, including abiotic production of carbon dioxide, carbon monoxide and methane, especially in arid and semi-arid regions. Research has also produced contradictory results regarding controls on photodegradation. Here we summarize the state of knowledge about the role of photodegradation in litter decomposition and C cycling and investigate drivers of photodegradation across experiments using a meta-analysis. Overall, increasing litter exposure to solar radiation increased mass loss by 23% with large variation in photodegradation rates among and within ecosystems. This variation was tied to both litter and environmental characteristics. Photodegradation increased with litter C to nitrogen (N) ratio, but not with lignin content, suggesting that we do not yet fully understand the underlying mechanisms. Photodegradation also increased with factors that increased solar radiation exposure (latitude and litter area to mass ratio) and decreased with mean annual precipitation. The impact of photodegradation on C (and potentially N) cycling fundamentally reshapes our thinking of decomposition as a solely biological process and requires that we define the mechanisms driving photodegradation before we can accurately represent photodegradation in global C and N models.

Keywords UV-B · Solar radiation · Arid ecosystems · Grasslands · Carbon · Nitrogen · Lignin

Introduction

Ecosystem carbon (C) and nutrient cycling is driven by the two fundamental processes of production and decomposition. In terrestrial ecosystems, the physical and biological controls on ecosystem productivity are relatively well known and can be accurately modeled (Cramer et al. 1999), but researchers have had less success modeling decomposition rates across space and time, especially of leaf litter on the soil surface (Whitford et al. 1981; Moorhead et al. 1999; Gholz et al. 2000; Adair et al. 2008). Work over the last 5 years suggests that an abiotic process, photodegradation, may help to explain some of the problems in modeling decomposition to date (e.g. Parton et al. 2007).

Photodegradation, the breakdown of organic matter via solar radiation, can increase decomposition rates and lead to changes in the way C and nutrients are cycled among plants, soil and atmosphere. There are several paths by which solar radiation has been observed to influence pools and fluxes associated with plant litter decomposition (Fig. 1). Solar radiation may increase decomposition fluxes from soil organic matter (SOM) and litter via abiotic photochemical reactions or facilitation of microbial decomposition through the production of labile photodegraded material (e.g. Austin and Vivanco 2006; Brandt et al. 2009; Foereid et al. 2010). However solar radiation can also have negative effects on fluxes through direct negative impacts on microbial and plant growth (e.g. Johanson et al. 1995; Duguay and Klironomos 2000; Johnson 2003; Belnap et al. 2008). Thus, solar radiation may decrease pool sizes by suppressing plant growth and associated inputs or by increasing outputs from litter, soil and dissolved organic matter (DOM; via photodegradation/facilitation of decomposition). Solar radiation may increase or decrease the recalcitrance and microbial uptake of DOM depending on the source of DOM (Moran and Zepp 1997). CO₂ fluxes from microbial respiration can be positive or negative depending on whether solar radiation has a net positive effect by increasing labile carbon or a net negative effect by reducing microbial growth (e.g. Foereid et al. 2010; Johnson 2003).

Given the complexity of solar radiation effects on decomposition, an increased understanding of the underlying mechanisms is critically needed. Here, we summarize recent advances in understanding the role of photodegradation in plant litter decomposition in three main areas: (1) the mechanisms by which litter is photodegraded; (2) the role photodegradation plays in

Fig. 1 Observed effects of solar radiation on pools and fluxes of decomposition. A negative effect on pool size indicates that outputs exceed inputs. For litter and microbial biomass, this is a consequence of negative effects of UV radiation on growth. For soil and dissolved organic matter, this is a consequence of increased fluxes from the pool without an increase in inputs. Extracellular enzymes are included in the microbial biomass pool. Please refer to text for further discussion



the global carbon cycle; and (3) the factors that influence the role of photodegradation in litter mass loss across all field experiments to date using a metaanalysis. We then explore implications for modeling biogeochemical processes. Finally, we provide recommendations for future research needed to fill critical gaps in the understanding of this important process.

Historical context

Research on decomposition has shown that litter in arid ecosystems decomposes faster than predicted by microbial drivers, namely climate and litter chemistry (Whitford et al. 1981; Adair et al. 2008; Austin 2011). Several hypotheses for this phenomenon have been proposed, such as litter consumption by termites (Johnson and Whitford 1975; Whitford et al. 1982). Pauli (1964) first proposed that solar radiation may play a role in driving decomposition in arid ecosystems, and although the idea was reiterated two decades later (Moorhead and Reynolds 1989), the hypothesis went largely untested with a few exceptions (Zlotin 1979; Mackay et al. 1994).

Recognition of the impacts of chlorofluorocarbons on stratospheric ozone in the 1980s led to research on the impacts of increased ultraviolet-B radiation (UV-BR) on decomposition and nutrient cycling (related to the most recent United Nations Environment Programme (UNEP) Environmental Effects Assessment Panel (EEAP) Report, see Ballaré et al. 2011 and Zepp et al. 2011 and references therein). These studies focused primarily on decomposition in high latitude systems where ozone depletion is greatest (Gehrke et al. 1995; Johanson et al. 1995; Paul et al. 1999). Many of these studies indicated that the primary mechanisms of UV-BR impacts on decomposition were indirect, mediated through changes in litter chemistry or changes in soil biota, rather than the direct result of incident UV-BR inducing litter mass loss (reviewed in Paul et al. 1999). These studies provided important advances in methodology for manipulating solar radiation at different wavelengths and introduced the concept of photodegradation to a larger audience in the terrestrial ecology field.

Research on the contribution of plants and litter to atmospheric trace gas concentrations became a topic of interest in the 1990s and remains so today. Work in the 1990s on carbon monoxide (CO) emissions showed that solar radiation plays a large role in CO emissions from plant litter and SOM (Tarr et al. 1995; Schade et al. 1999; Kisselle et al. 2002). However, the role of solar radiation in the production of other trace gases from plants, litter and soil remained largely ignored until the past 5 years. One exception was work by Anesio et al. (1999), showing that carbon dioxide (CO₂) could also be produced by photodegradation.

In contrast to terrestrial systems, work in the 1990s on biogeochemical cycling in aquatic systems significantly advanced understanding of photodegradation's role in decomposing DOM. The large body of work in marine and freshwater systems indicates that photochemical reactions with DOM produce an additional 1 Gt C year⁻¹ and 15 Mt nitrogen (N) year⁻¹ for heterotrophic utilization and convert 12-16 Gt C year⁻¹ to CO₂ through direct photochemical reactions (Moran and Zepp 1997). Research on DOM continues to explore variation among and within systems in photodegradation rates (e.g., with changes in salinity, DOM source and temperature) and interactions with microbial utilization (Obernosterer and Benner 2004; Anesio et al. 2005; Amado et al. 2007; Nelson et al. 2010). Recent work has focused on identifying underlying mechanisms of DOM photochemical reactions (Boreen et al. 2008; Cory et al. 2010; Vione et al. 2010). Other studies have examined photodegradation of submerged or emergent aquatic vegetation, providing a link between terrestrial litter decomposition and aquatic systems (Denward and Tranvik 1998; Vahatalo et al. 1998; Anesio et al. 1999; Denward et al. 2001: Hernes and Benner 2003).

As questions remained about the relatively high rates of litter decomposition in arid ecosystems, terrestrial ecologists, informed by the findings and methods from aquatic systems, began to investigate the potential importance of photodegradation.

Mechanisms

How does exposure to solar radiation degrade terrestrial litter and soil? The mechanisms involved, and whether they are similar to those discovered in the fields of aquatic biochemistry, materials science and atmospheric chemistry, remain relatively unknown. Here we discuss our current understanding of the mechanisms underlying litter mass loss through photodegradation, as well as possible mechanisms gleaned from work in other fields.

Wavelength

It was originally hypothesized that wavelengths in the UV-B range (280-320 nm) were responsible for photodegradation, but sufficient evidence suggests that other wavelengths, particularly those in the UV-A range (320-400 nm) and the short-wave visible range (400–500 nm), play an equal if not greater role (Austin and Vivanco 2006; Brandt et al. 2009; Austin and Ballaré 2010). Many plant compounds have a maximum absorbance in the UV-B range, but they also absorb lesser amounts of radiation at other wavelengths. Since the atmosphere absorbs much of the UV-BR before it hits Earth's surface, visible and, to a lesser extent, UV-A radiation make up a much larger proportion of the photon flux to the litter layer. Therefore, litter photodegradation rates may be higher on a *per photon* basis in the UV-B range, but lower *in* total compared to longer wavelengths. Such hypotheses have yet to be tested (Fig. 2, "Mechanistic research"), but recent work has pushed forward our understanding of terrestrial photodegradation by expanding the definition of photo-active radiation to include shortwave visible wavelengths.

> interactions. Measure impacts on N

cycling.

Photoreactive compounds

It has long been assumed that lignin is the primary compound in litter susceptible to photodegradation, but there has been relatively little evidence to support this assumption. Studies that have measured changes in litter chemistry over time under different radiation treatments have typically found only small treatment effects on either lignin or other compounds such as cellulose (Rozema et al. 1997; Brandt et al. 2007, 2010; Day et al. 2007). However, a recent study by Austin and Ballaré (2010) showed that photodegradation did not occur when lignin-free, pure cellulose substrates were exposed to radiation but did occur when a lignin solution was added to the substrates. Furthermore, rates of photodegradation increased with solution lignin concentration, and mass loss patterns were consistent with what would be predicted if only lignin were lost in this process. This study shows that lignin is at least one of the photoreactive compounds and that cellulose (in a highly purified form) does not appear capable of absorbing radiation and photodegrading in isolation. This study only tested two highly purified forms of lignin and cellulose. Results could vary with plant species differences in lignin structure and chemistry and the extent of decomposition that has already occurred. Lignin in the lignocellulose matrix

Fig. 2 Recommended future research directions to advance understanding of the role of photodegradation

Future research in three major areas will enhance our understanding of the role photodegradation plays, from ecosystem to global scales: **Field Experiments** Mechanistic Research Modeling Identify which litter Conduct long-term Incorporate photodegradation into experiments (> 3 years). components are photosensitive. biogeochemical Determine the relative models. Characterize roles of litter chemistry and leaf morphology in absorbance and AQY Validate models with (for mass loss and field measurements. influencing photodegradation rates. photoproduct Investigate interactions formation) of different Understand patterns of photodegradation litter types and/or with C and N cycling across seasons and litter C components. ecosystems. within ecosystems. Partition direct versus Examine Estimate impact of indirect photolysis in photodegradation on global CO₂ fluxes. photodegradation of the decomposition of standing dead litter. litter compounds. coarse woody debris > Quantify temperature soil organic matter, and black carbon. and moisture effects. Delineate microbial decompositionphotodegradation

of plant tissue may have a different susceptibility to photodegradation than purified lignin. It is also possible that compounds not examined in this study could be photoreactive, including proteins and other phenolics besides lignin (see Fig. 3).

Direct and indirect photolysis

The chemical mechanisms behind litter photodegradation remain unknown, but general principles of photochemistry along with evidence from other fields can provide a good starting point. When a molecule absorbs radiation and it results in a permanent change to the molecule, such as fragmentation, intramolecular rearrangement or electron transfer from or to the molecule, it is generally referred to as "direct photolysis" (Fig. 3). For example, a lignin molecule could undergo fragmentation into smaller organic components, which then could be consumed by microbes or leached out of the litter layer. This fragmentation could also result in the formation of inorganic C compounds such as CO or CO₂. Research on photodegradation of DOM in aquatic systems has shown that direct photolysis does occur to some extent (Kieber et al. 1999). This mechanism may be a factor in studies that have shown a decrease in litter lignin content when exposed to solar radiation (Rozema et al. 1997; Day et al. 2007; Henry et al. 2008; Austin and Ballaré 2010).

In addition to direct photolysis, "indirect photolysis" may also play a role in the decomposition of organic substrates (Fig. 3; see also review by Lanzalunga and Bietti 2000). During indirect photolysis, photosensitizers absorb radiation and transfer energy to other molecules (often triplet oxygen), creating reactive intermediates such as singlet oxygen, hydroxyl radical or hydrogen peroxide. Other important reactive intermediates can include reduced iron, copper or manganese. Reactive intermediates then change the chemistry of another non-light-absorbing molecule or part of the same molecule where the photosensitizer resided. In the case of litter photodegradation, light-absorbing (chromophoric) structures in lignin or molecules in litter may act as photosensitizers, facilitating chemical transformations of compounds that do not absorb much radiation, such as nonchromophoric lignin components, cellulose or hemicellulose. Research in aquatic systems and on wood and paper materials has shown that indirect photochemical processes are the primary mechanisms by which many organic materials are photodegraded (reviewed in George et al. 2005; Cory et al. 2010). In addition, recent studies have shown that photosensitized production of reactive oxygen species is the



(reactive intermediates)

Deringer

primary mechanism for photochemical production of CH_4 from plant pectins (McLeod et al. 2008, Messenger et al. 2009). This mechanism could explain why some studies have seen decreases in cellulose but not lignin when litter is exposed solar radiation (Brandt et al. 2007, 2010). Indirect photolysis could be an alternative mechanism for observed lignin loss (Rozema et al. 1997; Day et al. 2007; Henry et al. 2008; Austin and Ballaré 2010). In addition, direct photolysis may also result in the production of reactive intermediates that can facilitate indirect photolysis (Lanzalunga and Bietti 2000).

Interactions with other decomposition processes

Modifications to litter by direct or indirect photochemical processes can lead to modifications in other decomposition processes, including the biodegradability of litter. A recent study showed that preexposure of litter to solar radiation can make it more digestible when subsequently decomposed under conditions favoring microbial decomposition (Foereid et al. 2010). Another recent study suggests that the radiation exposure conditions for plant litter may affect the size of respiration pulses of CO₂ emissions during subsequent precipitation events (Ma et al. 2012). The facilitation of microbial decomposition by photodegradation appears to depend heavily on length of exposure, and studies exposing litter for short periods of time (e.g. a season) do not show the same effects as those conducted over longer time periods (Brandt et al. 2009; Foereid et al. 2010; Kirschbaum et al. 2011). However, studies in the aquatic literature have shown that photodegradation effects on subsequent microbial decomposition can also be negative, depending on the chemistry of organic materials being bio-and photo-degraded (Tranvik and Bertilsson 2001). Since relatively few studies have examined this interaction, it is too early to generalize about the magnitude or direction of this interaction.

Photodegradation can also interact with other physical decomposition processes. It can increase the solubility of litter, leading to increased leaching (Gallo et al. 2006; Feng et al. 2011). Research by Feng et al. (2011) suggests that this effect may be dependent on litter type, as increases in water-extractable organic C from photodegradation were seen for pine but not maize litter. This may explain why a study examining grass and oak litters found no effect of UV-radiation exposure on DOC leached from litter (Brandt et al. 2009), while another study examining conifer litter reported a significant increase (Gallo et al. 2006). In addition to leaching, physical processes common in arid, light-exposed systems such as wind or soil abrasion could interact with photodegradation, but this interaction remains relatively unstudied (Throop and Archer 2007; Austin 2011). One recent study suggests that freeze-thaw cycles in colder climates may contribute to accelerated mass loss by photodegradation (Uselman et al. 2011).

Influence on the global carbon cycle

The effects of ozone depletion and climate change on ultraviolet radiation are expected to have measurable impacts on the global C cycle (Zepp et al. 2011), but the specific contributions of photodegradation are not well known. The factors that affect solar radiation exposure and, therefore, photodegradation, are highly complex, varying considerably with ecosystem type, as well as through space and time (Table 1). Foereid et al. (2011) estimated of the importance of photodegradation to overall litter decomposition on a global scale by comparing modeled global photodegradation fluxes, based on experimental data from three sites, against ecosystem NPP. Their modeled estimates suggest that only 0.5-1.6 % of global NPP is photodegraded. However, in dry, light-exposed ecosystems, up to 14 % of NPP can be lost via photodegradation (Foereid et al. 2011). It appears that the primary influence of photodegradation on the C cycle is to accelerate return of C to the atmosphere. Welldocumented pathways of C return to the atmosphere are the direct abiotic losses of CO₂ (Anesio et al. 1999; Brandt et al. 2009; Rutledge et al. 2010), CO (Tarr et al. 1995; Schade et al. 1999; Kisselle et al. 2002; Derendorp et al. 2011b), methane (CH_4 , McLeod et al. 2008; Vigano et al. 2008; Bruhn et al. 2009; Messenger et al. 2009) and trace amounts of ethane and ethylene and other hydrocarbons (McLeod et al. 2008; Derendorp et al. 2011a; Table 2).

The first estimates of photochemical CO_2 emissions were based on exposure of dry aquatic macrophyte litter to only UV radiation in the laboratory (0.0108 g C m⁻² day⁻¹; Anesio et al. 1999). In comparison, measurements made in microcosms under natural solar radiation on clear sunny days in summer resulted

Table 1 Factors affecting solar radiation exposure and subsequent photodegradation from global to plot-level scales

Ozone Possible slight increase with stratospheric ozone thinning.	Smith et al. (2010)
Potential decreases with high tropospheric ozone.	
Latitude Generally negative relationship.	Moody et al. (2001),
High latitudes susceptible during summer months due to ozone thinning.	Pancotto et al. (2003), Brandt et al. (2010)
Season In grasslands, highest rates during summer in grasslands if seasonally dry, but rates may be higher in spring in areas with summer monsoons.	Henry et al. (2008), Brandt et al. (2010), Rutledge et al. (2010)
In temperate deciduous forests, highest directly before leafout in spring or after senescence in autumn.	
In tropical deciduous forests, highest during dry season.	
Elevation Most likely positive relationship due to higher proportion of short-wave radiation and higher total irradiance at high elevations.	Blumthaler et al. (1997)
May be negative relationship in areas where cloud, canopy, or snow cover increases with elevation to the point where litter is shaded.	
Cloud cover Most likely negative relationship.	Madronich et al. (1998)
Modest cloud cover can increase diffuse radiation and potentially increase rates on mostly sunny days	
Leaf area index Generally negative relationship, but especially so with broadleaf architecture.	Rozema et al. (1999)
Canopy architecture Higher rates with vertically-distributed structure (e.g. grasslands) than horizontally-distributed structure (e.g. broadleaf forests).	Rozema et al. (1999)
Landscape patchiness/ Higher rates in open areas versus under shrubs or trees.	Köchy and Wilson (1997):
evenness Rates per unit mass potentially greater with increased evenness.	Throop and Archer (2007), Mlambo and Mwenje (2010)
Soil reflectivity Sandy soils may increase albedo and lead to increased rates in adjacent litter.	Rozema et al. (1999)
Snow No photodegradation when buried.	N/A ^a
Potential increase in photodegradation in standing dead if surrounded by snow due to albedo.	
Litter position/location Standing dead litter may be more susceptible than litter positioned flat on the ground.	N/A ^a
Litter layer thickness Rates per unit mass of litter higher with lower litter layer thickness.	Henry et al. (2008), Brandt et al. (2009)
Soil cover/burial Decreased rates with increasing soil burial.	Throop and Archer (2007), Brandt et al. (2010)
	Barnes et al. (2012)

^a N/A no current publications on this topic

in slightly higher estimates (0.016 g C m⁻² day⁻¹) and showed that other wavelengths play a role in litter photodegradation (Brandt et al. 2009). Rutledge et al. (2010) estimated photodegradation CO_2 emissions based on eddy covariance and clear chamber measurements during daylight and non-daylight hours.

Their estimates $(0.186 \text{ g C m}^{-2} \text{ day}^{-1} \text{ in annual}$ grasslands during the dry season and $0.093-0.18 \text{ g C m}^{-2} \text{ day}^{-1}$ in harvested peatlands) are much higher than the laboratory and microcosm measurements described above. Such contrasting estimates make it clear that a better understanding of how and

Site	Substrate	Radiation source	Measured flux rate	Reference
Carbon dioxide (CO ₂)				
Laboratory	Senesced and fresh aquatic grasses	UV lamps	$0.0108 \text{ g C m}^{-2} \text{ day}^{-1}$	Anesio et al. (1999)
Laboratory	Senesced grasses	UV lamps	$0.004 \text{ g C m}^{-2} \text{ day}^{-1}$	Brandt et al. (2009)
Minnesota grassland	Senesced grasses	Natural solar radiation	$0.016 \text{ g C m}^{-2} \text{ day}^{-1}$	Brandt et al. (2009)
California grassland	Senesced grasses and bare soil	Natural solar radiation	$0.186 \text{ g C m}^{-2} \text{ day}^{-1 \text{ a}}$	Rutledge et al. (2010)
Peatland	Bare peat	Natural solar radiation	0.093–0.18 g C m $^{-2}$ day $^{-1}$ $^{\rm b}$	Rutledge et al. (2010)
Laboratory	Dried grasses	Solar simulator	0.003–0.012 g C ${\rm m^{-2}~day^{-1}~^c}$	Lee et al. (2012)
Laboratory	Pure citrus pectin	UV lamps	118.2 μ g C g dw ⁻¹ h ^{-1 d}	McLeod et al. (2008)
Carbon monoxide (CO)				
Laboratory	South Africa: senescedsavanna grasses	Solar simulator	$5.51 \text{ mg C m}^{-2} \text{ day}^{-1}$	Tarr et al. (1995)
Laboratory	South American savanna grass	Solar simulator	$3.36 \text{ mg C m}^{-2} \text{ day}^{-1 \text{ e}}$	Schade et al. (1999)
Brazilian savanna	Unburned vegetation	Natural solar radiation	$3.34 \text{ mg C m}^{-2} \text{ day}^{-1 \text{ a}}$	Kisselle et al. (2002)
	and soil		$1.98 \text{ mg C m}^{-2} \text{ day}^{-1 \text{ f}}$	
Brazilian savanna	Burned	Natural solar radiation	$3.32 \text{ mg C m}^{-2} \text{ day}^{-1 \text{ a}}$	Kisselle et al. (2002)
			$3.24 \text{ mg C m}^{-2} \text{ day}^{-1 \text{ f}}$	
Brazilian	Unburned vegetation	Natural solar radiation	$3.12 \text{ mg C m}^{-2} \text{ day}^{-1 \text{ a}}$	Kisselle et al. (2002)
shrubland	and soil		$2.39 \text{ mg C m}^{-2} \text{ day}^{-1 \text{ f}}$	
Brazilian	Burned	Natural solar radiation	$3.08 \text{ mg C m}^{-2} \text{ day}^{-1 \text{ a}}$	Kisselle et al. (2002)
shrubland			$3.13 \text{ mg C m}^{-2} \text{ day}^{-1 \text{ f}}$	
Laboratory	Sequoia leaf litter	UV lamps	$5.76 \text{ mg C m}^{-2} \text{ day}^{-1 \text{ g}}$	Derendorp et al. (2011b)
Laboratory	Dried grasses	Solar simulator	$0.88-3.50 \text{ mg C m}^{-2} \text{ day}^{-1 \text{ c}}$	Lee et al. (2012)
Methane (CH ₄)				
Laboratory	Dried grass, milled	UV lamps	$150 \text{ ng C g } dw^{-1} h^{-1}$	Vigano et al. (2008)
Laboratory	Pure cellulose	UV lamps	$6 \text{ ng C g } dw^{-1} h^{-1}$	Vigano et al. (2008)
Laboratory	Pure lignin	UV lamps	24 ng C g $dw^{-1} h^{-1}$	Vigano et al. (2008)
Laboratory	Pure apple pectin	UV lamps	63.75 ng C g $dw^{-1} h^{-1}$	Vigano et al. (2008)
Laboratory	Wide variety of fresh and dry plant leaves	UV lamps	$0-3225 \text{ ng C g } dw^{-1} h^{-1}$	Vigano et al. (2008)
Laboratory	Pure citrus pectin	UV lamps	495.75 ng C g $dw^{-1} h^{-1}$	McLeod et al.((2008)
Laboratory	Twigs	UV lamps	1.125 ng C g dw ⁻¹ h ⁻¹	Bruhn et al. (2009)
Laboratory	Citrus fruit peels	UV lamps	3.75–12.75 ng C g dw ⁻¹ h ^{-1 h}	Bruhn et al. (2009)
Laboratory	Dried grasses	Solar simulator	1.28–4.39 ng C g dw $^{-1}$ h $^{-1}$ c	Lee et al. (2012)
Laboratory	Cellulosic filter paper	Solar simulator	2.90–7.00 ng C g dw $^{-1}$ h^{-1} $^{\rm c}$	Lee et al. (2012)
Laboratory	Basswood sheet	Solar simulator	3.51–6.47 ng C g dw $^{-1}$ h $^{-1}$ c	Lee et al. (2012)
Ethylene (C ₂ H ₄)				
Laboratory	Pure citrus pectin	UV lamps	232.3 ng C g $dw^{-1} h^{-1}$	McLeod et al. (2008)
Ethane (C ₂ H ₆)				
Laboratory	Pure citrus pectin	UV lamps	100.2 ng C g dw ⁻¹ h ⁻¹	McLeod et al. (2008)

 Table 2
 Trace gas emissions from photodegrading plant materials

^a Dry season measurement

^b Value depends on equation used to model dark fluxes

^c Value depends on temperature (15–55°C)

^d dw denotes dry weight

^e Value depends on radiation intensity (taken from Schade et al. 1999, Fig. 7)

f Wet season measurement

g Value extrapolated using radiation intensity

^h Value depends on radiation wavelengths and intensity. (Color figure online)

why rates of photochemical CO_2 production vary across ecosystems (e.g., variation in radiation exposure; Table 1) is needed before we can make accurate estimates of its contribution to global CO_2 emissions.

Measured rates of photochemical CO production from litter under natural or simulated solar radiation conditions generally range from approximately $2-5.5 \text{ mg C m}^{-2} \text{ day}^{-1}$ (Tarr et al. 1995; Schade et al. 1999; Kisselle et al. 2002; Table 2). Using response factors developed for CO emissions together with global land area and irradiance datasets, Schade and Crutzen (1999) estimated the global source strength of CO emissions from photodegradation to be on the order of 100 Tg CO year⁻¹. Schade et al. (1999) also measured thermal emissions of CO, but those emissions were lower, by as much as an order of magnitude, than photochemical emissions. In a recent study, Lee et al. (2012) documented the production of CO, as well as CO_2 and CH_4 , through photodegradation and thermal processes. Their controlled laboratory study of six different plant materials showed that higher production of trace gases occurred during photodegradation compared to thermal degradation and that photodegradation production rates varied depending on temperature. The molar ratio of thermal degradation to photodegradation for CO ranged from 0.07 to 0.28 (for temperatures from 25 to 55 °C; Lee et al. 2012). This result is similar to the findings of Schade et al. (1999). Consistent with our summary of trace gas production in separate studies (Table 2), Lee et al. found, using simultaneous measurements of these trace gases, that the primary trace gas produced during both photodegradation and thermal degradation is CO_2 (Lee et al. 2012).

The magnitude of photochemical CH₄ fluxes varies depending on experimental conditions and on materials exposed (Table 2; values in Table 2 converted to ng C g dry weight⁻¹ h⁻¹ for comparison). Most studies to date have focused on a limited set of plant materials (tobacco leaves, citrus pectin) or plant structural compounds (lignin, pectin, but see Lee et al. 2012). McLeod et al. (2008) reported CH₄ emissions of up to 661 ng CH₄ g dry weight⁻¹ h⁻¹ for citrus pectin, while Bruhn et al. (2009) reported emissions that ranged from 5 to 17 ng CH₄ g dry weight⁻¹ h⁻¹ for similar material. The widest range of materials studied is reported by Vigano et al. (2008) in which the range of CH₄ emissions was also wide (0–4300 ng CH₄ g dry weight⁻¹ h⁻¹). Bloom et al. (2010) used an approach similar to Schade and Crutzen (1999) to estimate global CH₄ emissions derived from photodegradation of foliar pectin. Their estimate, 0.2–1.0 Tg CH₄ year⁻¹, is small compared to other estimates of foliar CH₄ emissions (1–7 Tg CH₄ yesr⁻¹, Keppler et al. 2006) and points to the need for further mechanistic studies (Fig. 2).

Little is known about the influence of photodegradation on other biogeochemical cycles, such as nitrogen (N). Because photodegradation is an abiotic process, litter that is low in N and is primarily decomposed by photodegradation should not have the same immobilization and mineralization patterns that are often observed in systems controlled by microbial decomposition (Parton et al. 2007). Two studies have found partial support for this hypothesis (Brandt et al. 2007; Smith et al. 2010). In addition, a study by McCalley and Sparks (2009) showed that soil exposed to solar radiation releases trace N gases at higher rates than when kept under dark conditions, and Mayer et al. (2012) recently reported N release with photodissolution of SOM. This suggests that, as with C, photodegradation speeds up N losses to the atmosphere and reduces N storage in litter and SOM.

Effects on litter mass loss: a meta-analysis

What factors across ecosystems influence the role photodegradation plays in litter decomposition? Field researchers have attempted to quantify the role of photodegradation in litter decomposition by manipulating the amount of solar radiation reaching the litter layer and measuring litter mass loss over time. Most field studies examine the effect of photodegradation in the presence of microbial decomposition (but see Austin and Vivanco 2006). The photodegradation effect is thus the difference in mass loss or decay rates between a control (e.g., that allows all radiation to pass through) and a treatment that either reduces solar radiation by filtering photodegradative wavelengths (UV-B, UV-A, and/or other wavelengths), blocking radiation completely, or increasing UV-BR using lamps. To examine the magnitude of solar radiation effects on mass loss (ML), we built a data set consisting of 50 field experiments drawn from 16 published, peer-reviewed sources (through July of 2011; see Appendix). We tested the influence of different treatment combinations, experimental

approaches, initial lignin content, litter C to N ratio (C/N) and litter area to mass ratio (area/mass) for the log response ratio for final mass loss (ML) of the higher solar radiation treatment compared to the lower solar radiation treatment for all studies (LRR_{enh}, see Appendix for more details).

Effects of radiation treatment and experimental methods

Higher exposure of litter to solar radiation increased ML by 23 % on average across all studies (Fig. 4a); however, results varied in relation to the type of experimental treatment. The experiments that compared enhanced UV-BR to reduced or ambient UV-BR found that increasing exposure had no effect on ML (95 % CI included zero), although these results may change with a larger sample size. Often, such experiments were designed to investigate the role of ozone depletion and the resulting increases in UV-BR on plant growth and litter chemistry, and thus only manipulated UV-B wavelengths by dosage changes projected to occur with ozone thinning. In these experiments, UV-BR-induced changes in litter quality also failed to impact photodegradation (95 % CI encompassed zero for litter produced under ambient, N = 10, and enhanced, N = 4, radiation; data not shown). In contrast, mass loss consistently increased in experiments that compared litter exposed to ambient versus reduced UV-BR, UV-A+BR, or total solar radiation (32 % on average; $LRR_{enh} > 1$; Fig. 4a). These experiments were more often designed specifically for testing the hypothesis that photodeg-radation played a role in litter mass loss in systems with high levels of radiation exposure and low moisture availability, and thus compared ambient controls to treatments that blocked a certain wavelength entirely.

Methods used to impose radiation treatments also affected the ML response, but it is difficult to tease apart the effect of method from the effect of experimental treatment (above). Enhancing litter exposure to UV-BR via supplementary lamps had no effect on ML (N = 14, 95 % CI included zero; Fig. 4b). To date, no other methods have been used to increase the exposure of litter to UV-BR to examine effects on ML. Using shade cloths or filter treatments to reduce total solar radiation or UV radiation (UVR) reduced ML by 45 and 25 %, respectively (n = 11 and 25, respectively; Fig. 4b).

In studies that reduced solar radiation, reducing only UV-BR had the smallest effect on ML (95 % CI included zero; Fig. 4c). In contrast, blocking all solar radiation (via shade), radiation with wavelengths <450 nm, or UVA + BR all increased ML (Fig. 4c). Evidence to date suggests that short wavelength radiation in the visible range (400–500 nm) may substantially increase photodegradation (Fig. 4c;



Fig. 4 Log response ratio for increasing the exposure of litter to solar radiation (LRR_{enh}) in (**a**) all experiments and in experiments that increased the level of solar radiation (either UV-B, UV-A+B, or total) exposure: (i) from reduced solar radiation (via filters or shade) to ambient solar radiation, (ii) from ambient solar radiation to enhanced radiation (via lamps), and (iii) from reduced solar radiation to enhanced radiation, **b** by treatment type (lamp for enhanced radiation,

shade or filter for reduced radiation), and c by the amount and/or wavelengths of radiation blocked (total for shade, wavelengths of less than 450 nm, UVA + UVB, or only UVB) in reduced experiments. *Positive values* indicate an increase in mass loss in response to increasing radiation. *Negative values* indicate reduced mass loss in response to increasing radiation. Sample sizes shown in *parentheses*

Brandt et al. 2009). However, this study has yet to be replicated, so the magnitude of this effect is uncertain.

Influence of radiation exposure

The amount of radiation to which litter is exposed can be influenced by a variety of environmental factors including latitude, elevation, canopy cover and cloud cover, leading to increases or decreases in photodegradation rates (Table 1). It is difficult to tease apart the relative influence of these factors on photodegradation rates because the number of studies is limited, studies often use slightly different experimental approaches (see above), and many of these factors tend to co-vary. In general, our meta-analysis suggests that studies conducted at mid-latitude arid sites with low canopy cover demonstrate larger photodegradation effects compared to higher latitude sites, which could be due in part to greater radiation dosage at mid-latitude arid sites (LRR_{enh} decreased with increasing latitude in field experiments that compared ambient to reduced solar radiation; Fig 5a). This effect was independent of method (shade or filter) or wavelengths excluded (e.g., UV-B or UV-A + B; data not shown).

Not examined in the meta-analysis (due to lack of sufficient data) are the effects of other factors on radiation exposure (Table 1). For example, Brandt et al. (2010) conducted a field study comparing photodegradation rates in three contrasting grassland systems, hypothesizing that photodegradation rates

should increase with decreasing latitude and increasing aridity. However, they found that factors such as seasonal monsoons and soil burial decreased the influence of photodegradation in the most arid site. Other studies have shown that burial by soil or additional litter can reduce exposure to radiation and thus reduce photodegradation rates (Henry et al. 2008; Throop and Archer 2009; Barnes et al. 2012).

Influence of precipitation

Our meta-analysis suggests that photodegradation plays the largest role in dry environments. With one exception, the LRR_{enh} decreased with mean annual precipitation (MAP; Fig. 5b). Precipitation data were only available for experiments that compared ambient to reduced solar radiation, so it is not known if this relationship would be significant for experiments that enhanced UV-BR. The exception to this trend was a single photodegradation study conducted in a Mediterranean climate (Henry et al. 2008). This site had both high MAP and photodegradation rates (Fig. 5b; Appendix). The large effect of increasing litter exposure to solar radiation was likely due to the highly seasonal nature of rainfall, which began several months after litter was placed in the field. This result suggests that MAP may not be a good predictor of photodegradation in climates with highly seasonal rainfall patterns.



Fig. 5 a LRR_{enh} by the absolute value of site latitude (in degrees). Linear regressions are significant for all data (n = 50, LRR_{enh} = $0.805 - 0.014 \times \text{abs}(\text{Latitude})$, P = 0.0002, $r^2 = 0.261$; regression not shown) and for experiments (exp) that reduced solar radiation (Reduced exp; n = 36, LRR_{enh} = $1.155 - 0.023 \times \text{abs}(\text{Latitude})$, P = 0.0012, $r^2 = 0.268$). **b** LRR_{enh} by mean annual precipitation (MAP). Note that only reduced

solar radiation experiments had MAP data available and so are the only experiments shown in the figure. A linear regression of all data versus MAP was not significant; removal of one outlier point (shown as a *black circle*; Henry et al. 2008) resulted in a significant relationship between LRR_{enh} and MAP (n = 36, LRR_{enh} = 0.431 - 0.0008 × MAP, P = 0.0041, $r^2 = 0.224$)

A few studies have directly examined the influence of precipitation or moisture on the relative importance of photodegradation (Brandt et al. 2007; Gallo et al. 2009; Smith et al. 2010). These studies tested the hypothesis that under dry conditions, photodegradation should be more easily detectable, while under wet conditions microbial decomposition should dominate, not because photodegradation is reduced under wet conditions, but rather because microbial decomposition is reduced by dry conditions, making the photodegradation signal clearer. In addition, these studies hypothesized that under wet conditions, overall decomposition rates would decrease when litter is exposed to radiation because of negative impacts of UV-BR on the microbial community (Moody et al. 1999; Pancotto et al. 2003). In general, findings support these hypotheses (Brandt et al. 2007; Smith et al. 2010). However, one study conducted in a riparian forest in New Mexico did not find a precipitation effect, which the authors attributed to a low (15 %) difference between the precipitation treatment and control (Gallo et al. 2009). It is important to note that photochemical reactions could potentially be moisture-dependent (Schade et al. 1999) or could lead to increased litter solubility (Gallo et al. 2006), so actual photodegradation rates may be lower under dry conditions despite the fact that it accounts for a larger proportion of litter mass loss in dry versus mesic environments.

Influence of litter characteristics

Because lignin has been identified as a photosensitive compound (George et al. 2005), photodegradation is generally hypothesized to increase with lignin content (e.g. Moorhead and Callaghan 1994; Austin and Ballaré 2010). Instead, we found no relationship between LRR_{enh} and initial lignin content for experiments that compared ambient to reduced solar radiation, and LRR_{enh} actually decreased with increased initial lignin content in experiments that compared enhanced UV to ambient or reduced UV treatments (Fig. 6a). The lack of relationship between LRR_{enh} and lignin content may be accurate, but it could also be a consequence of the narrow range of lignin contents used in reduced solar radiation experiments thus far (Fig. 6a; the one exception is Day et al. 2007 with a lignin content of ~ 32 %). However, LRR_{enh} did increase with initial litter C/N (Fig. 6b). An increase in litter C/N can reduce microbial decomposition rates because of low N availability, thus increasing the proportion of decomposition that is due to photodegradation. LRR_{enh} was also positively related to litter area/mass (Fig. 6c), a relationship that held across all reduced radiation experiments regardless of method or wavelengths excluded (data not shown; Appendix). This suggests an influence of litter density, presumably due to the shading of bottom litter layers from radiation by top layers in low area/mass experiments. Brandt et al. (2009) found support for this hypothesis by demonstrating that increasing litter density decreased photochemically-derived CO_2 on a per unit mass basis, while CO_2 released per unit area remained constant.

In addition to lignin content and litter density, other chemical and physical properties of litter could potentially influence photodegradation rates, but have not yet been examined. Studies in aquatic systems have shown that iron (Fe^{3+}) can play a large role in photodegradation of DOM (Gao and Zepp 1998); thus litter iron content may be an important factor. Other studies of DOM have shown that the degree to which DOM has been biodegraded can affect subsequent photodegradation rates and vice versa (Obernosterer and Benner 2004), so this may also hold true for litter. Other radiation-absorbing compounds such as flavonoids, tannins and anthocyanins may also influence photodegradation but have remained unexamined. Physical characteristics such as the presence of trichomes or cutin on the leaf surface may affect photodegradation rates by preventing radiation transmission. Many plants in arid environments have these characteristics to (at least in part) protect themselves from solar radiation, so these physical characteristics should be addressed when attempting to estimate photodegradation rates in arid systems.

Methodological considerations

Some of the variation in the meta-analysis results likely stems from the wide range of methods used to quantify litter photodegradation. Only a few studies have attempted to quantify photodegradation in the absence of microbial activity (Austin and Vivanco 2006; Brandt et al. 2009), and it is unclear whether the methods used may affect photodegradation measurements. Sterilization techniques such as autoclaving or gamma irradiation may change litter chemistry as it





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Enhanced exp Reduced exp regression

Enhanced exp regression

Fig. 6 a LRR_{enh} by % initial lignin content of litter. Linear regressions were only significant for enhanced UVR experiments (n = 14, LRR_{enh} = $0.185 - 0.015 \times \%$ lignin, P = 0.0108, $r^2 = 0.494$). Removal of one outlier point (from Day et al. 2007) did not result in significant regressions for either all or reduced solar radiation experiments. **b** LRR_{enh} by initial litter C/N. Linear regressions were significant for all experiments (n = 28, LRR_{enh} = $-0.125 + 0.004 \times C/N$, P = 0.0001, $r^2 = 0.436$; regression not shown) and for reduced solar

does for SOM chemistry (e.g. Kelsey et al. 2010), and chemical biocides may interfere with or promote photochemical reactions in the litter (Katagi 2004). Finally, eliminating the microbial component may under- or overestimate the total contribution of photodegradation to decomposition by eliminating the potential positive or negative effects that photodegradation may have on subsequent microbial decomposition (Henry et al. 2008; Gallo et al. 2009; Foereid et al. 2010). Thus, most field studies have examined the effect of photodegradation in the presence of microbial decomposition by placing unaltered litter in the field.

Another factor that may contribute to variation among studies is the method used to contain litter. Many studies have used the standard litterbag

radiation experiments (n = 24, LRR_{enh} = $-0.0998 + 0.004 \times C/N$, P = 0.0006, $r^2 = 0.420$). Note that there were not enough data points to calculate a regression for enhanced UV-BR experiments. **c** LRR_{enh} by area/mass of litter. Linear regressions were significant for all experiments (n = 48, LRR_{enh} = $-0.031 + 0.004 \times \text{area/mass}$, P < 0.0001, $r^2 =$ 0.309; regression not shown) and for reduced solar radiation experiments (n = 36, LRR_{enh} = $-0.005 + 0.004 \times \text{area/mass}$, P = 0.0002, $r^2 = 0.337$)

technique with either supplementary UV-B lamps (Gehrke et al. 1995; Newsham et al. 1997; Moody et al. 2001), filters (Pancotto et al. 2003, 2005; Brandt et al. 2007, 2010), or shade cloths (Mackay et al. 1994; Köchy and Wilson 1997; Gallo et al. 2009) to manipulate the amount of radiation reaching the litterbag. The use of lamps or filters above traditional litterbags has the advantage of being directly comparable to other litterbag studies. This technique also has relatively minor effects on microclimate because filters and lamps are not in contact with the litter and can have perforations large enough to allow precipitation to reach the litter layer. In addition, filters and lamps can be replaced periodically to ensure consistent UV treatments. The disadvantage is that the mesh used in constructing litterbags covers a substantial proportion of the litter (50 % for fiberglass and 30 % for aluminum), which leads to an underestimate of the effects of solar radiation (either positive or negative).

A few studies have constructed litter "boxes" or "packets" with clear UV-transparent or UV-blocking tops and bottoms made out of mesh or UV transparent material (Austin and Vivanco 2006; Day et al. 2007; Austin and Ballaré 2010), minimizing the problem of obstructing solar radiation. However, the UV-transparent materials themselves can photodegrade, thus altering their radiation transmission properties over time (Spartech Polycast, personal communication). The materials can also have significant effects on microclimate when placed directly above the litter, increasing temperature and decreasing the amount of precipitation reaching the litter. These factors could potentially lead to an overestimate of photodegradation rates if reaction rates are temperature-dependent or an underestimate of microbial decomposition processes due to low levels of water availability and very high temperatures. Since no method is without problems, we recommend that researchers fully disclose the limitations of their treatments to inform comparisons among studies.

Modeling photodegradation

Now that a basic understanding of the role of photodegradation in litter decomposition and C cycling is emerging, how can this information be incorporated into models of decomposition and biogeochemical cycling? Traditional decomposition models based on biotic drivers of decomposition (i.e., climate and litter chemistry) do not accurately predict decomposition in semi-arid and arid ecosystems (Meentemeyer 1978; Parton et al. 2007; see also Throop and Archer 2009), in some cases underpredicting long-term decomposition by as much as 25 % (Adair et al. 2008). A primary candidate for explaining this discrepancy is photodegradation (Austin and Vivanco 2006; Adair et al. 2008; Vanderbilt et al. 2008). Despite this, photodegradation has yet to be adequately incorporated into terrestrial ecosystem models, a failure that is likely due to incomplete understanding of this process.

Large unanswered questions remain regarding the mechanisms driving photodegradation, substrate(s) and product(s) of photodegradation, and interactions of photodegradation with climate, local environmental conditions and biotic decomposition. Thus, the four existing terrestrial photodegradation models consist of exploratory exercises (Moorhead and Callaghan 1994; Rozema et al. 1999) or simple attempts to scale up CO production (Schade and Crutzen 1999) and mass loss (Foereid et al. 2011) rates. However, even these few modeling efforts have increased understanding of how photodegradation may fit into ecosystem functioning and large-scale C cycling. Moorhead and Callaghan (1994) modified the CENTURY ecosystem model to account for litter photodegradation by increasing the C transfer rate from structural litter to recalcitrant SOM and adding a C transfer from structural litter to labile SOM. These modifications increased litter turnover rates, but had little effect on SOM dynamics (Moorhead and Callaghan 1994). Another exploratory model allowed exposure to UV-BR to increase litter mass loss (all fractions) and increase the lignin content of litter inputs (due to exposure during growth; Rozema et al. 1999). While increasing UV-BR exposure did increase mass loss, UV-BR-induced lignin increases had larger negative effects on mass loss, resulting in slower mass loss rates regardless of UV exposure level (Rozema et al. 1999). However, while many studies have demonstrated that UVR exposure during growth affects plant tissue chemistry (see review in Caldwell et al. 2003), empirical studies have shown that the effects of UV-BR exposure during plant growth on decomposition may be minor or non-existent (e.g. Hoorens et al. 2004). Schade and Crutzen (1999) and Foereid et al. (2011) used models to scale up local photoproduction or photodegradation rates to global levels. Both estimated that photodegradation contributions to global fluxes of CO (Schade and Crutzen 1999) and litter mass loss (Foereid et al. 2011) were relatively small. Such models are useful at a coarse scale, but developing a model that fully accounts for the role of photodegradation in ecosystem function or the global C cycle requires greater understanding of the underlying mechanisms of photodegradation, its substrates and products, as well as its interactions with climate and microbial decomposition (Fig. 7).

A substantial challenge for modeling litter photodegradation lies in determining what is being photodegraded. Decomposition models often divide litter into different pools that decompose at different rates (e.g., fast, slow cellulosic and very slow lignin pools; Adair et al. 2008). If only lignin is photodegraded, incorporating photodegradation into traditional pool based models could significantly alter mass loss predictions (as in Moorhead and Callaghan 1994). Currently, there is substantial contradictory evidence as to whether (or how) photodegradation differentially affects various litter compounds or litter types (e.g., Brandt et al. 2009; Austin and Ballaré 2010). It is therefore unclear how to best represent litter mass loss in photodegradation models. While Moorhead and Callaghan (1994) assumed that only lignin was photodegraded, the remaining three studies modeled either litter mass losses without specifying which components were lost (Rozema et al. 1999; Foereid

et al. 2011) or product formation without accounting for mass loss (Schade and Crutzen 1999).

If photodegradation affects all litter types similarly (if, e.g., mass loss is a function of surface area or there is a universal action spectrum for litter photodegradation), then differences in litter quality (e.g., lignin content) will not affect photodegradation rates. However, this seems unlikely given findings that DOM photodegradation rates and photoreactivity change with source and composition (e.g., Stubbins et al. 2011). Because the exact chemical structure of DOM is often unknown, DOM loss (or product formation) is often modeled via the development of "apparent" quantum yields (AQY; Miller et al. 2002). The AQY



Fig. 7 Conceptual model of the differences in carbon cycling between \mathbf{a} open canopy, arid systems and \mathbf{b} relatively closed canopy, mesic systems. Both biotically-driven (*gray*) and photodegradation-driven (*black*) fluxes are shown. (Color figure online)

describes the quantitative relationship between DOM breakdown (or photoproduct formation) and the amount of radiation absorbed by DOM (i.e., photoproduct amount per photon light absorbed). The AQY and DOM absorption are defined for individual wavelengths, and the product of AQY, DOM absorption and light availability is integrated across the wavelengths of interest to predict DOM disappearance (or photoproduct formation; e.g., Miller et al. 2002; Swan et al. 2009). Similar to aquatic systems, where the AQY and DOM absorption change with DOM source and composition (Stubbins et al. 2011), predicting photoproduct formation during terrestrial photodegradation may depend on plant litter photoreactivity changes with litter chemistry.

While some products of photodegradation are well known (e.g., CO_2 , CH_4 , CO), not all of the relevant products have been identified. For example, there is little direct evidence of the conversion of litter to SOM by photodegradation (Mayer et al. 2012; Fig. 7), but it has been long hypothesized and was incorporated into Moorhead and Callaghan's (1994) photodegradation model. Even less is known about direct photodegradation of SOM or the contributions of photodegradation to soil N cycling, and no models currently address these processes.

As previously noted, solar radiation and photodegradation may interact with biotic decomposition by changing the biodegradability of litter or negatively impacting microbes (Fig. 7). Yet, none of the four terrestrial photodegradation models attempt to account for these interactions, likely because so little is known about them. Failing to account for these interactions will likely result in biased predictions of mass loss in systems exposed to high levels of solar radiation.

Developing a global photodegradation model requires understanding how photodegradation interacts with climate. We do not yet have a quantitative understanding of how litter moisture and temperature influence photodegradation rates and photoproduct yields. Results from aquatic systems suggest that photodegradation may be influenced by temperature; Zhang et al. (2006) found that the AQY of CO produced from DOM increased substantially with water temperature (by up to 70 % between 0.5 and 32 °C). Without understanding of how photodegradation processes change with temperature and moisture, attempts to scale up local photodegradation results to the globe will likely yield inaccurate estimates.

In contrast to models of aquatic systems (e.g., Miller et al. 2002), current terrestrial photodegradation models have been inconsistent in parameterizing litter exposure to solar radiation. Thus far, Rozema et al. (1999) took the most thorough approach: mass is lost linearly as a function of radiation exposure while accounting for light extinction through a canopy. Schade and Crutzen's (1999) model predicts global CO evolution from litter as a function of solar radiation, but does not explicitly account for light extinction (an "ecosystem factor" accounts for both light and litter availability). Similarly, Foereid et al. (2011) developed a linear equation that related mass loss to incident radiation, but their model did not account for light extinction. The Moorhead and Callaghan (1994) model did not use solar radiation as a model input. Basing photodegradation or photoproduct yield on the availability of solar radiation is an important first step. It remains to be determined whether it is necessary, as in aquatic models, to account for light extinction, canopy interception, substrate radiation absorption and photoproduct AQYs.

Whether terrestrial photodegradation will be best described by simple pool based models (e.g., Moorhead and Callaghan 1994) and/or linear equations (e.g., Foereid et al. 2011) or by more complex photochemical models (such as those developed for DOM) will only be known when the mechanisms driving terrestrial photodegradation are defined and sufficient data exist to conduct sophisticated model comparisons.

Conclusions

Research on the role of photodegradation in the decomposition of terrestrial plant litter over the past decade has fundamentally altered our perception of terrestrial C and N cycling as primarily biotic processes. Current estimates of C fluxes from photo-degradation vary widely, but evidence suggests that a substantial fraction of net primary productivity can be returned to the atmosphere via this abiotic process, primarily as CO_2 . Research points to photodegradation as a source of other trace gases, including CO,

CH₄, ethylene, and trace N gases. Some evidence also suggests that photodegradation may interact with biotic decomposition, potentially increasing microbial respiration rates by producing labile C, but more studies are needed to better understand this interaction (Fig. 2).

Recent work and our meta-analysis show that photodegradation is most important in low latitude ecosystems and ecosystems characterized by low mean annual precipitation. However, photodegradation may also play an important role in climates with highly seasonal rainfall (e.g., Henry et al. 2008) or in mesic environments during very dry years (e.g., Brandt et al. 2010). Still, it is likely that photodegradation plays the largest role in arid and semi-arid ecosystems, where plant litter is highly exposed to solar radiation and where microbial decomposition may be limited by moisture and radiation conditions.

Rates of photodegradation are likely influenced by several factors, including temperature, moisture, litter chemistry and level of exposure to solar radiation. While research has revealed much about the climates where photodegradation is important, exactly how photodegradation interacts with changes in temperature and moisture remains unknown (Fig. 2). Our meta-analysis results indicate that photodegradation increases with litter C/N and area/mass, but the effects of lignin content and litter chemistry in general are unclear. Much of this confusion likely originates from an incomplete understanding of the mechanisms driving terrestrial photodegradation.

A critical gap still exists in our understanding of the mechanisms by which photodegradation occurs. Thus far, most terrestrial photodegradation research has failed to differentiate between direct and indirect photochemical mechanisms. Research in aquatic systems and on wood and paper materials suggests that terrestrial ecologists should consider how these distinct mechanisms may differentially affect how litter is photodegraded. Working to better understand the mechanisms driving photodegradation will help us identify the important photo-active plant compounds, photodegradation products, and wavelengths, as well as allow us to improve models of C and N cycling in arid and semi-arid ecosystems.

Although many questions about photodegradation in terrestrial ecosystems have yet to be answered (Fig. 2), results to date indicate that, in arid and semiarid ecosystems, photodegradation speeds up C (and likely N) losses to the atmosphere and reduces storage in litter and SOM. The results of research on photodegradation patterns and mechanisms will continue to provide critical information for biogeochemical models and advancing understanding of ecosystem and global C and N cycling.

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Appendix: Meta-analysis methods, data and references

We conducted an extensive keyword and citation search using the ISI Web of Science for the words "photodegradation," "UV," "UV-B," "photolysis," and "solar radiation" in combination with "litter," "decomposition," and "organic matter." Within each reference, we collected information on the field site (latitude and mean annual precipitation, MAP), litter (species, initial lignin content, initial C/N, area/mass), treatment (supplementary lamp, filter, or shade cloth), experiment duration and the final mass loss (ML) in each treatment as a percentage of initial litter mass.

Many of the references we found contained research investigating photodegradation of different litter types, in different locations or by excluding or increasing different amounts/types of radiation, etc. Each of these was incorporated into the database as a single experiment. Thus, the database contained 16 references and 50 experiments (Table 3). For each experiment (N = 50) we calculated a metric that examined the effect of increasing litter UV exposure based on the log response ratio (LRR of enhancing solar radiation or LRR_{enh}): $LRR_{enh} = \ln(ML_{enh}/$ ML_{red}), where ML_{enh} is ML in the treatment with more solar radiation exposure and ML_{red} is ML in the treatment with less UV exposure. ML_{enh} values >0 indicate that ML increases with increasing solar radiation exposure; values <0 indicate that ML decreases with increasing solar radiation exposure.

We examined LRR_{enh} averaged across all experiments (N = 50) and by solar radiation treatment combinations. Across the 50 experiments, we examined three types of treatment combinations:

Initial litter characterist	lics											
Reference	Climate/system	Experiment	Wavelengths manipulated	Light method	Latitude	Length (mo)	MAP (mm)	C/N	Lignin (%)	Area/Mass (cm ² g ⁻¹)	Lignin method	LRR _{enh}
Austin and Ballaré (2010)	Semi-arid grassland	Amb vs red	UV-B	Filter	-31.07	4	0^{a}	ŊŊ	7.3	139	Van Soest 1963	0.2585
Austin and Ballaré (2010)	Semi-arid grassland	Amb vs red	UV-A + B	Filter	-31.07	4	0^{a}	Ŋ	7.3	139	Van Soest 1963	0.6893
Austin and Ballaré (2010)	Semi-arid grassland	Amb vs red	<450 nm	Filter	-31.07	4	0^{a}	Ŋ	7.3	139	Van Soest 1963	0.8887
Austin and Vivanco (2006)	Semi-arid steppe	Amb vs red	UV-B	Filter	-45.68	18	152	QN	ŊŊ	200	NA	0.4555
Austin and Vivanco (2006)	Semi-arid steppe	Amb vs red	Total	Filter	-45.68	18	152	Ŋ	ŊŊ	200	NA	0.7904
Brandt et al. (2007)	Semi-arid grassland	Amb vs red	UV-A + B	Filter	40.82	36	401	48	6.47	33	Van Soest 1967	-0.0532
Brandt et al. (2007)	Semi-arid grassland	Amb vs red	UV-A + B	Filter	40.82	36	226	48	6.47	33	Van Soest 1967	0.0109
Brandt et al. (2007)	Semi-arid grassland	Amb vs red	UV-A + B	Filter	40.82	36	401	61	6.18	33	Van Soest 1967	0.2616
Brandt et al. (2007)	Semi-arid grassland	Amb vs red	UV-A + B	Filter	40.82	36	226	61	6.18	33	Van Soest 1967	0.2332
Brandt et al. (2010)	Mesic grassland	Amb vs red	UV-A + B	Filter	45.40	24	726	154	8.1	22.5	Van Soest 1967	0.1687
Brandt et al. (2010)	Mesic grassland	Amb vs red	UV-A + B	Filter	45.40	24	726	43	6.61	22.5	Van Soest 1967	0.0528
Brandt et al. (2010)	Semi-arid grassland	Amb vs red	UV-A + B	Filter	40.80	24	309	154	8.1	22.5	Van Soest 1967	0.4087
Brandt et al. (2010)	Semi-arid grassland	Amb vs red	UV-A + B	Filter	40.80	24	309	43	6.61	22.5	Van Soest 1967	0.0133
Brandt et al. (2010)	Arid grassland	Amb vs red	UV-A + B	Filter	34.40	24	222	154	8.1	22.5	Van Soest 1967	0.6067
Brandt et al. (2010)	Arid grassland	Amb vs red	UV-A + B	Filter	34.40	24	222	43	6.61	22.5	Van Soest 1967	0.0008
Day et al. (2007)	Desert (arid)	Amb vs red	UV-B	Filter	33.50	4	195	18	32.48	56	Modified Allen 1989	0.2534
Day et al. (2007)	Desert (arid)	Amb vs red	UV-B	Filter	33.50	4	195	ND	ND	56	NA	0.2133
Day et al. (2007)	Desert (arid)	Amb vs red	UV-B	Filter	33.50	4	195	ND	ND	56	NA	0.1335
Gallo et al. (2009)	Semi-arid riparian	Amb vs red	Total	Shade	35.13	16	250	133	Ŋ	40	NA	0.3845
Gallo et al. (2009)	Semi-arid riparian	Amb vs red	Total	Shade	35.13	23	250	LL	Ŋ	40	NA	0.2154
Gallo et al. (2009)	Semi-arid riparian	Amb vs red	Total	Shade	35.13	23	250	36	ŊŊ	40	NA	0.3981

Table 3 Data from 16 references used in the meta-analysis of field photodegradation experiments

nitial litter characteristi	ics											
Reference	Climate/system	Experiment	Wavelengths manipulated	Light method	Latitude	Length (mo)	MAP (mm)	C/N	Lignin (%)	Area/Mass $(cm^2 g^{-1})$	Lignin method	LRR _{enh}
Gallo et al. (2009)	Semi-arid riparian	Amb vs red	Total	Shade	35.13	16	286	133	ΟN	40	NA	0.7023
Gallo et al. (2009)	Semi-arid riparian	Amb vs red	Total	Shade	35.13	23	292	LL	Ŋ	40	NA	0.3140
Gallo et al. (2009)	Semi-arid riparian	Amb vs red	Total	Shade	35.13	23	292	36	QN	40	NA	0.5040
Gehrke et al. (1995)	Subarctic heathland	Enh vs red	UV-B	Lamp	68.21	12	Ŋ	QN	20.5	ND	Allen 1989	-0.1196
Gehrke et al. (1995)	Subarctic heathland	Enh vs red	UV-B	Lamp	68.21	12	Ŋ	QN	18	ND	Allen 1989	-0.0961
Henry et al. (2008)	Mediterranean grassland	Amb vs red	Total	Shade	37.67	12	689	ND	ND	196	NA	1.0068
Köchy and Wilson (1997)	Semi-arid forest	Amb vs red	Total	Shade	50.47	S	364	ŊŊ	ND	50	NA	-0.2958
Köchy and Wilson (1997)	Semi-arid forest	Amb vs red	Total	Shade	50.47	5	364	ŊŊ	ND	50	NA	-0.0478
Köchy and Wilson (1997)	Semi-arid grassland	Amb vs red	Total	Shade	50.47	S	364	ŊŊ	ND	50	NA	0.0785
Köchy and Wilson (1997)	Semi-arid grassland	Amb vs red	Total	Shade	50.47	S	364	ŊŊ	ND	50	NA	0.2118
Moody et al. (2001)	Arctic tundra	Enh vs amb	UV-B	Lamp	78.00	4	QN	4	14	43.1	Rowland and Roberts 1994	0.0000
Moody et al. (2001)	Subarctic shrub heath	Enh vs amb	UV-B	Lamp	68.00	14	QN	4	14	43.1	Rowland and Roberts 1994	-0.0502
Moody et al. (2001)	Dune grassland	Enh vs amb	UV-B	Lamp	52.00	14	QN	4	14	43.1	Rowland and Roberts 1994	-0.0374
Moody et al. (2001)	Experimental garden	Enh vs amb	UV-B	Lamp	38.00	14	QN	4	14	43.1	Rowland and Roberts 1994	-0.0843
Vewsham et al. (1997)	Temperate woodland	Enh vs amb	UV-B	Lamp	52.00	16	QN	ND	ND	125 ^b	NA	0.0000
Pancotto et al. (2003)	Sub-Antarctic shrubland	Amb vs red	UV-B	Filter	-55.51	4	629.92	16	6.45	50	Van Soest 1963	-0.1445
Pancotto et al. (2003)	Sub-Antarctic shrubland	Amb vs red	UV-B	Filter	-55.51	4.6	629.92	15	6.75	50	Van Soest 1963	-0.3162
Pancotto et al. (2005)	Sub-Antarctic shrubland	Amb vs red	UV-B	Filter	-54.51	29	629.92	61	4.36	19	Van Soest 1963	0.0017
Pancotto et al. (2005)	Sub-Antarctic shrubland	Amb vs red	UV-B	Filter	-54.51	29	629.92	54	3.94	19	Van Soest 1963	0.0052

Table 3 continued

Initial litter characterist	ics											
Reference	Climate/system	Experiment	Wavelengths manipulated	Light method	Latitude	Length (mo)	MAP (mm)	C/N	Lignin (%)	Area/Mass $(cm^2 g^{-1})$	Lignin method	LRR _{enh}
Rozema et al. (1997)	Dune grassland	Enh vs amb	UV-B	Lamp	52.00	2	ND	QN	5.7	36	Allen 1989	0.0416
Rozema et al. (1997)	Dune grassland	Enh vs amb	UV-B	Lamp	52.00	2	ŊŊ	QN	8.9	36	Allen 1989	0.1269
Rozema et al. (1997)	Dune grassland	Enh vs amb	UV-B	Lamp	52.00	1.5	ŊŊ	QN	5.7	36	Allen 1989	0.0520
Rozema et al. (1997)	Dune grassland	Enh vs amb	UV-B	Lamp	52.00	1.5	ŊŊ	QN	8.9	36	Allen 1989	0.0139
Rozema et al. (1997)	Experimental garden	Enh vs amb	UV-B	Lamp	52.00	7	QN	QN	5.7	36	Allen 1989	0.0549
Rozema et al. (1997)	Experimental garden	Enh vs amb	UV-B	Lamp	52.00	7	QN	ŊŊ	8.9	36	Allen 1989	0.2564
Uselman et al. (2011)	Desert riparian	Amb vs red	UV-B	Filter	39.85	12	173	32	10.57	50	Van Soest 1963	-0.3802
Uselman et al. (2011)	Desert riparian	Amb vs red	UV-B	Filter	39.85	12	173	17	9.57	50	Van Soest 1963	-0.0704
Uselman et al. (2011)	Desert riparian	Amb vs red	UV-B	Filter	39.85	12	173	32	11.29	50	Van Soest 1963	-0.0418
Verhoef et al. (2000)	Experimental garden	Enh vs amb	UV-B	Lamp	52.00	9	Ŋ	Ŋ	Ŋ	43.75	NA	0.0000
ND no data, NA not app	plicable, C/N carbon to	o nitrogen ratio.	Negative latitu	des are in	the Southe	rn Hemis	phere					

2 E Ξ Ę INEGALI Jgen 2 3 υĵ *ND* no data, *NA* not applicat ^a Precipitation excluded ^b Estimated

Table 3 continued

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(1) enhanced versus ambient; (2) enhanced versus reduced; and (3) ambient versus reduced. Enhanced solar radiation treatments were accomplished via supplementary UV lamps, while reduced treatments used filters or shade cloths to block UVR and/or total solar radiation. We examined average LRR_{enh} for each manipulation method (filter, supplementary lamp, or shade). For each mean LRR_{enh}, we calculated a 95 % confidence interval (CI; Student's *t* distribution). Means were considered to be significantly different from zero if the 95 % CIs did not include zero. We also examined linear relationships between experiment LRR_{enh} and latitude, MAP, initial litter lignin content, initial litter C/N and leaf litter area/mass.

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