

of 2-methylthionicotino-nitrile, Pyrazolopyridine and Pyridopyrazolotriazine Derivatives

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Síntesis, reacciones y caracterización de derivados de 2-metilthionicotinonitrilo, pirazolopiridina y piridopirazolotriazina

Síntesi, reaccions i caracterització de derivats de 2-metilthionicotinonitril, pirazolopiridina i piridopirazolotriazina

Recibido: 30 de agosto de 2007; aceptado: 4 de octubre de 2007

RESUMEN

Se obtienen las 4,6-diaril-1H-pirazolo[3,4-b]piridin-3-aminas 4a-c en estado muy puro y se usan como buenos productos de partida para el presente estudio. El compuesto 4a se diazotiza para dar la correspondiente sal de diazonio 11 y, además, se hace reaccionar con 2-bromo-1-feniletanona para dar el correspondiente derivado de pirazolo[3,4-b]piridin-2-il)-1-feniletanona 7 que se usa para preparar los derivados de hidrazona 8 y de formamida 10 mediante reacción con hidrato de hidrazina y ácido fórmico, respectivamente. El compuesto 11 se utiliza para preparar derivados de piridopirazolotriazina vía acoplamiento con diversos compuestos que contienen -CH₂- activos. Se determina la estructura química de los nuevos compuestos heterocíclicos sintetizados a partir de los datos de IR, 1H-RMN, espectros de masas y análisis elementales.

Palabras clave: Cianoetanoamida. 2-metilthionicotinonitrilo. Pirazolopiridina. Hidrazona. Formamida. Pirazolo-piridinilhidrazono. Piridopirazolotriazina.

SUMMARY

4,6-Diaryl-1H-pyrazolo[3,4-b]pyridin-3-amines 4a-c were obtained in very pure state and used as the good starting materials for the present study. Compound 4a diazotized to give the corresponding diazonium salt 11 and also, reacted with 2-bromo-1-phenylethanone to give the corresponding pyrazolo[3,4-b]pyridin-2-yl)-1-phenylethanone derivative 7 which in turn, used for the preparation of the hydrazone and formamide derivatives 8 and 10 respectively through its reaction with hydrazine hydrate and formic acid respectively. Compound 11 was used for the preparation of pyridopyrazolotriazine derivatives

via its coupling with several active -CH₂- containing compounds. Considering the data from IR, ¹H NMR, the mass spectra and elemental analyses the chemical structures of the newly synthesized heterocyclic compounds were elucidated.

Key words: Cyanoethanethioamide. 2-methylthionicotinonitrile. Pyrazolopyridine. Hydrazone. Formamide. Pyrazolopyridinylhydrazone and Pyridopyrazolotriazine.

RESUM

S'obtenen les 4,6-diaril-1H-pirazolo[3,4-b]piridin-3-aminas 4a-c en estat molt pur i s'empren com a bons productes de partida per al present estudi. El compost 4a es diazotitza per donar la corresponent sal de diazoni 11 i a més, es fa reaccionar amb 2-bromo-1-feniletanona per donar el corresponent derivat de pirazolo[3,4-b]piridin-2-il)-1-feniletanona 7, que s'empra per preparar els derivats d'hidrazona 8 i de formamida 10 mitjançant reacció amb hidrat d'hidrazina i àcid fòrmic, respectivament. El compost 11 s'utilitza per preparar derivats de piridopirazolotriazina via acoblament amb diversos compostos que contenen -CH₂- actius. Es determina l'estructura química dels nous compostos heterocíclics sintetitzats a partir de les dades de IR, 1H-RMN, espectres de masses i anàlisis elementals.

Mots clau: Cianoetanoamida. 2-metilthionicotinonitrilo. Pirazolopiridina. Hidrazona. Formamida. Pirazolo-piridinilhidrazono. Piridopirazolotriazina.

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INTRODUCTION

In previous papers⁽¹⁻²²⁾ the synthesis of pyrazolopyridine and pyridopyrazolotriazine derivatives were described. As a continuation of our research on the synthesis of these ring systems and owing to the reported biological activities of pyrazolopyridine⁽²³⁻²⁵⁾ and as well as that of triazines,⁽²⁶⁻³⁰⁾ we were interested to synthesize several derivatives of these ring systems that are required for several chemical transformation as well as for our medicinal chemistry program.

RESULTS AND DISCUSSION

The synthetic potentiality of 1a-c was used to build up new ring containing active functional groups, this goal achieved through their reaction with hydrazine hydrate. Thus, it has been found that 1a reacted with methyl iodide in methanolic sodium methoxide to afford the corresponding 4-(4-bromophenyl)-2-(methylthio)-6-phenylnicotinonitrile 2a which in turn, reacted with hydrazine hydrate in ethanol to give the sulfur free product that formulated as 4-(4-bromophenyl)-6-phenyl-1H-pyrazolo[3,4-b]pyridin-3-amine 4a through the non-isolable product 3a. Chemical structure of 4a was elucidated based on the spectroscopic and elemental analyses (cf. Exp. Part). A further elucidation of structure 4a arose from its preparation authentically, thus 1a reacted with hydrazine hydrate to afford directly 4a through the hydrogen sulfide molecule removal. It is remarkable to report here that the products of the two pathways were identical in all physical and chemical properties (cf. Exp. Part). Similarly, 1H-pyrazolo[3,4-b]pyridin-3-amine derivatives 4b,c were synthesized by the reaction of either 1b,c or 2b,c with hydrazine hydrate and their structures were elucidated by considering the data of elemental and spectroscopic analyses (cf. Exp. Part and Chart 1).

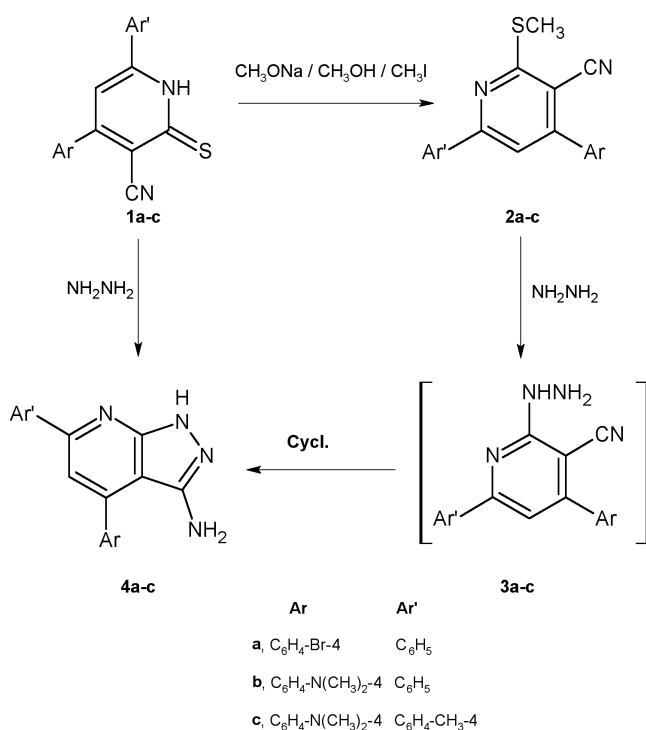


Chart 1

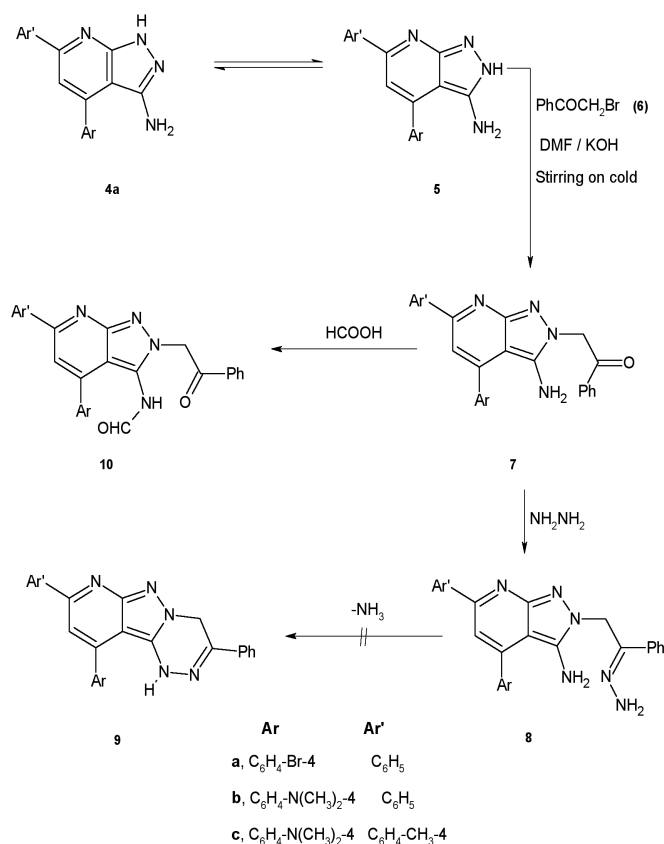


Chart 2

The chemical reactivity and synthetic potentiality of 4a was investigated through its reaction with 2-bromo-1-phenylethanone (6) in stirred DMF/KOH on cold to give the corresponding 2-(3-amino-4-[4-bromophenyl]-6-phenyl-2H-pyrazolo[3,4-b]pyridin-2-yl)-1-phenylethanone (7). The reaction seemed to be proceeded through the dehydrobromination between the tautomer 5 and 6.

The IR, ¹H NMR, and elemental analyses of this reaction product found to be in a good agreement with the assigned structure (cf. Exp. Part and Chart 2). Furthermore, the structure of compound 7 was confirmed chemically through its reaction with both hydrazine hydrate and formic acid. Thus, it has been found that compound 7 reacted with hydrazine hydrate under reflux 3-5 hours to afford the reaction product which assumed to be formed via the condensation of 7 with hydrazine hydrate to give the final isolable hydrazone derivative 8 which confirm the presence and chemical reactivity of CO group. The IR (cm⁻¹) of this reaction product showed no bands of C=O and instead the newly formed =N-NH₂ group was detected.

On the other hand, the ¹H NMR spectrum of this reaction product revealed the signals of NH₂ at pyrazole ring and =N-NH₂ protons. Moreover, its mass spectrum gave m/z = 497 (M⁺, 14.9%) which corresponding to the molecular weight of the molecular formula C₂₈H₂₁BrN₆ of the assigned structure. Several peaks such as 364 [M-CH₂-C(Ph) = N-NH₂, 100%], 335 (364-N₂, 20%) and 270 (335-C₃H₅N₂, 13.1%) were detected and their detec-

tion elucidates structure 8 (cf. Exp. Part). Considering the above-mentioned data in addition to that of elemental analyses it is concluded that this reaction product formulated as 2-(3-amino-4-(4-bromophenyl)-6-phenyl-2H-pyrazolo[3,4-b]pyridin-2-yl)-1-phenyl-ethanone hydrazone 8 not triazine derivative 9 (cf. Chart 2 and Exp. Part).

The position and chemical reactivity of NH₂ in 7 could be established through its reaction with formic acid. It has been found unexpectedly that the IR (cm⁻¹) of this reaction product showed the bands of C=O, NH function and the ¹H NMR (δppm) showed the signals of aldehydic, NH and -CH₂COPh protons. Moreover, the mass spectrum of this reaction product gave m/z = 511 (M⁺, 11.1%) which corresponding to the molecular weight of the molecular formula C₂₇H₁₉BrN₄O₂ of the assigned structure, in addition to the peak at m/z = 405 (M-COPh, H, 71.5%) and m/z = 77 (for the fragment Ph, 100%). Considering the above mentioned data we can conclude that N-[4-(4-bromophenyl)-2-(2-oxo-2-phenylethyl)-6-phenyl-2H-pyrazolo[3,4-b]pyridin-3-yl]formamide 10 represents the reaction product (cf. Chart 2 and Exp. Part).

The chemical reactivity of 4a was further investigated via its reaction with nitrous acid to give the corresponding 4-(4-

bromophenyl)-6-phenyl-1H-pyrazolo[3,4-b]pyridine-3-diazonium chloride (11). Compound 11 used as a good synthon for the preparation of several pyrido[2',3':3,4]pyrazolo[5,1-c]triazine derivatives 15a-f. Thus, it has been found that compound 11 coupled with each of malononitrile, ethyl cyanoacetate and 2-cyanoethanethioamide (12a-c) in stirred ethanol-sodium acetate on cold. The IR (cm⁻¹) of the reaction product in case of 12a showed the bands of CN, NH₂ groups. The reaction product in case of 12b showed the bands of NH₂, CO (ester) in its IR spectrum and its ¹H NMR spectrum revealed the signals of COOCH₂CH₃ protons. On the other hand, the reaction product in case of 12c showed the bands of NH₂, C=S in IR spectrum. Moreover, the mass spectra of these reaction products gave m/z = 442, 489 and 476 and these corresponding to the molecular weights of the molecular formulas C₂₁H₁₂BrN₇, C₂₃H₁₇BrN₆O₂ and C₂₁H₁₄BrN₅S respectively in addition to several fragments confirm the structures 15a-c (cf. Chart 3 and Exp. Part). Considering the above-mentioned data (cf. Exp. Part) we concluded that the reaction seemed to be proceeded via dehydrochlorination between each of 12a-c and 11 to afford the corresponding non-isolable derivatives 13a-c, 14a-c respectively which spontaneously cyclized to afford the corresponding triazines 15a-c.

Similarly, synthon 11 reacted with each of diethylmalonate, acetylacetone and ethyl acetoacetate 12d-f under the same above-mentioned experimental conditions to afford the corresponding pyrido[2',3':3,4]pyrazolo[5,1-c][1,2,4]triazines 15d-f via the non-isolable intermediates 13d-f respectively. By considering the data of IR, ¹H NMR and elemental analyses (cf. Exp. Part) the chemical structures of each of 15d-f were elucidated. Moreover, the mass spectra of 15d-f gave m/z = 490, 458 and 488 which corresponding to the molecular weights of the molecular formulas C₂₃H₁₆BrN₅O₃, C₂₃H₁₆BrN₅O and C₂₄H₁₈BrN₅O₂ of the assigned structures. The presence of broad band in IR spectrum of 15d confirm the presence of more stable enol-form and this confirmed further by the detection of peak at 489 (M⁻¹) at mass spectrum (cf. Chart 4 and Fig. 1). The presence of a peak at m/z = 488 in mass spectrum for the reaction product of 11 with 12f confirm the structure 15f and reject that of 15g and this also, further confirmed by the presence of signal at δ(ppm) 1.4 (t, 3H, COOCH₂CH₃) and 4.5 (q, 2H, COOCH₂CH₃) and absence of m/z = 460 for the molecular formula C₂₂H₁₄BrN₅O₂ of the structure 15g (cf. Charts 4, Fig. 2, and Exp. Part). An authentic samples of pyrido[2',3':3,4]pyrazolo[5,1-c][1,2,4]triazines 15a-f obtained through another pathway via the reaction of each of 12a-f with 11 in ethanol containing the catalytic amounts of triethylamine under reflux 3-5hrs. It is important to report here that compounds obtained through the two pathways are identical in all physical and chemical properties.

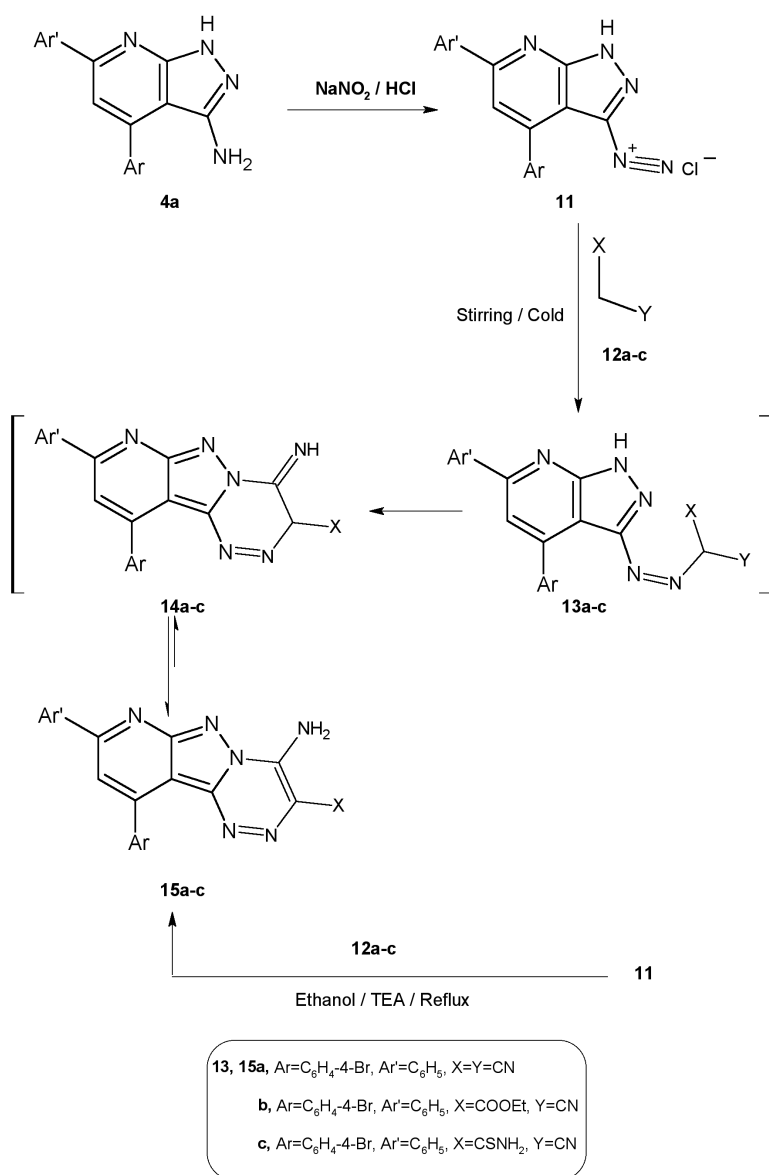
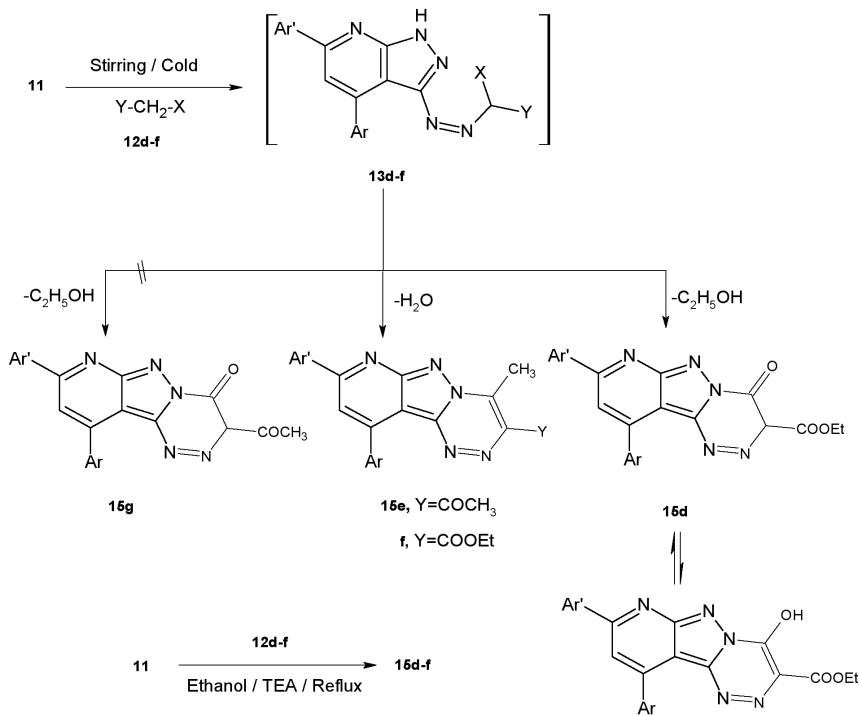
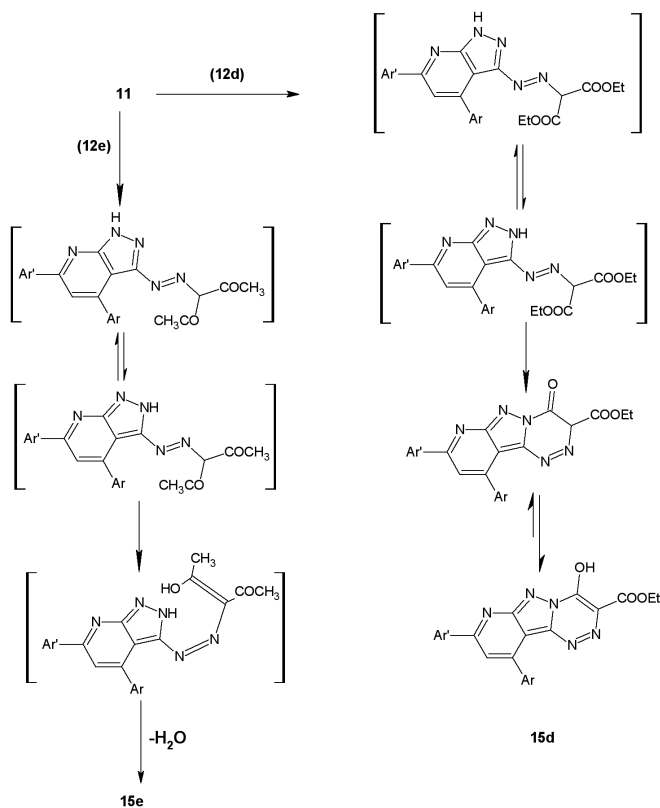


Chart 3

Figure 1



Ar=C₆H₄-4-Br, Ar'=C₆H₅

12d, X=Y=COOEt
 e, X=Y=COCH₃
 f, X=COCH₃, Y=COOEt

Chart 4

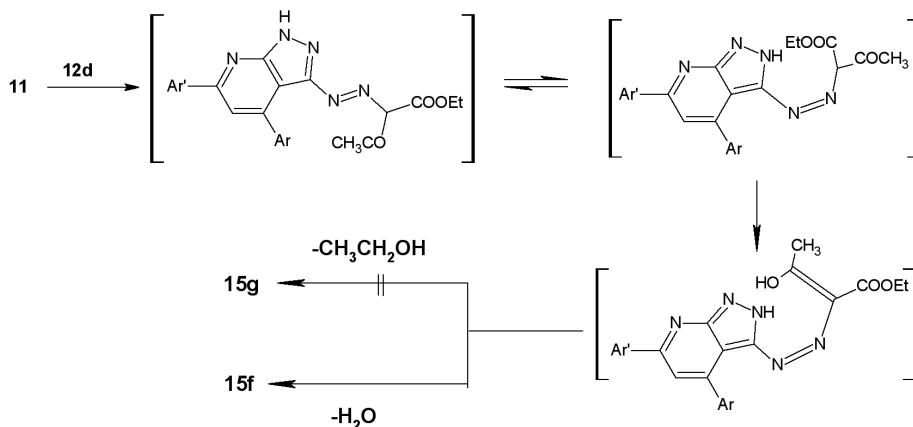


Figure 2

EXPERIMENTAL

All melting points were uncorrected. I.R. (KBr discs) spectra were recorded on a Shimadzu FTIR-8201PC Spectrophotometer. $^1\text{H-NMR}$ spectra were recorded on a Varian Mercury 300 MHz., and a Varian Gemini 200 MHz. spectrometers using TMS as an internal standard and CDCl_3 , DMSO-d_6 , and $(\text{CD}_3)_2\text{CO}$ as solvents. Chemical shifts were expressed as δ (ppm) units. Mass spectra were recorded on Shimadzu GCMS-QP1000EX using an inlet type at 70 eV. The Microanalytical Center of Cairo University performed the microanalyses.

Synthesis of 2a-c: (General Procedure): Mix each of 2-thioxopyridine derivatives 1a-c (0.1 mole) and iodomethane (0.15 mole) in methanolic sodium methoxide (0.23 g of sodium with about 30-50 mL methanol). Stir on cold for 1-1.5 hours then pour onto ice-cold water and acidified with concentrated HCl. Collect the formed solids by filtration and wash with water then crystallize from the proper solvent to afford 2a-c respectively.

4-(4-bromophenyl)-2-(methylthio)-6-phenylnicotinonitrile 2a: crystallized from acetic acid as yellowish-white crystals (64%); m.p 260 °C; IR (ν cm^{-1}): 2218 (CN) and 1605 (C=C); *Anal. for* $\text{C}_{19}\text{H}_{13}\text{BrN}_2\text{S}$ (381.28): Calcd./Found (%): C,59.85/60.10, H,3.44/3.50, Br,20.96/21.0, N,7.35/7.30, S,8.41/8.50.

4-[4-(dimethylamino)phenyl]-2-(methylthio)-6-phenylnicotinonitrile 2b: crystallized from ethanol as pale yellow crystals (64%); m.p 220 °C; IR (ν cm^{-1}): 2218 (CN) and 1605 (C=C); *Anal. for* $\text{C}_{21}\text{H}_{19}\text{N}_3\text{S}$ (345.46): Calcd./Found (%): C,73.01/73.10, H,5.54/5.50, N,12.16/12.20, S,9.28/9.30.

4-[4-(dimethylamino)phenyl]-6-(4-methylphenyl)-2-(methylthio)-nicotinonitrile 2c: crystallized from ethanol as orange crystals (66%); m.p 222 °C; IR (ν cm^{-1}): 2222 (CN) and 1602 (C=C); *Anal. for* $\text{C}_{22}\text{H}_{21}\text{N}_3\text{S}$ (359.48): Calcd./Found (%): C,73.50/73.50, H,5.89/5.90, N,11.69/11.50, S,8.92/9.0.

Synthesis of 4a-c: A mixture of 1a-c or 2a-c (0.01 mole of each) and hydrazine hydrate (20-25 mL) and pyridine (20 mL) was heated under reflux for 20 hours till the odour of H_2S or CH_3SH ceased. The products so formed after cooling were filtered off, washed with cold ethanol, and crystallized from the proper solvent to give 4a-c respectively.

4-(4-Bromophenyl)-6-phenyl-1H-pyrazolo[3,4-b]pyridin-3-amine 4a: crystallized from ethanol as yellow crystals (58 %); mp.232-4 °C; IR (cm^{-1}): 3466, 3302, 3194 (NH_2 , NH), 2858 (sat. CH) and 1642 (C=N); *Anal. for* $\text{C}_{18}\text{H}_{13}\text{BrN}_4$ (365.22): Calcd./Found (%): C,59.19/59.30, H,3.59/3.60, Br,21.88/21.70, N,15.34/15.40.

4-[4-(Dimethylamino)phenyl]-6-phenyl-1H-pyrazolo[3,4-b]pyridin-3-amine 4b: crystallized from ethanol as yellow crystals (65 %); mp.256-8 °C; IR (cm^{-1}): 3446, 3287, 3185 (NH_2 , NH) and 1638 (C=N); *Anal. for* $\text{C}_{20}\text{H}_{19}\text{N}_5$ (329.39): Calcd./Found (%): C,72.93/73.0, H,5.81/5.9, N,21.26/21.30.

4-[4-(Dimethylamino)phenyl]-6-(4-methylphenyl)-1H-pyrazolo[3,4-b]pyridin-3-amine 4c: crystallized from ethanol as yellow crystals (65%); mp.256-8 °C; IR (cm^{-1}): 3462, 3309, 3194 (NH_2 , NH) and 1641 (C=N); *Anal. for* $\text{C}_{21}\text{H}_{21}\text{N}_5$ (343.42): Calcd./Found (%): C,73.44/73.3, H,6.16/6.20, N,20.39/20.40.

Synthesis of 7: A solution of 4a and potassium hydroxide (0.01 mole of each) in DMF (30 mL) stirred with 2-bromo-1-phenylethanone (6) (0.01 mole) for 30 minutes on cold. Pour onto ice-cold water, the solid so formed was filtered off, washed with water, and crystallized from ethanol to give 7.

2-(3-Amino-4-(4-bromophenyl)-6-phenyl-2H-pyrazolo[3,4-b]pyridin-2-yl)-1-phenylethanone 7: as yellow crystals (60 %); mp.194-6 °C; IR (cm^{-1}): 3418, 3294, 3189 (NH_2), 3059 (CH-aromatic) and 1692 (C=O); $^1\text{H-NMR}$ (δ): 2.60 (s, 2H, $-\text{CH}_2-$), 5.67 (s, br., 2H, NH_2) and 7.08-7.89 (m, 15H, aromatic and pyridine-CH protons); MS: 483 (M^+ , 11.1%) which corresponding to the molecular weight of the molecular formula $\text{C}_{26}\text{H}_{19}\text{BrN}_4\text{O}$ of the assigned structure, 484 ($\text{M}^+ + 1$, 10.6 %), 485 ($\text{M}^+ + 2$, 8.5 %), 482 ($\text{M}^+ - \text{H}$, 9.8%), 380 ($\text{M}^+ - \text{PhCN}$, 72.5%), 379 (482-PhCN, 96.6%), 377 (482-PhCO, 100%), 363 ($\text{M}^+ - \text{PhCOCH}_2$, 1.6%), 105 (PhCO, 41.7%), 103 (PhCN, 1.5%) and 77 (Ph, 61.8%); *Anal. for* $\text{C}_{26}\text{H}_{19}\text{BrN}_4\text{O}$ (483.35): Calcd./Found (%): C,64.61/65.0, H,3.96/4.0, Br,16.53/16.40, N,11.59/11.60.

Synthesis of 8: A mixture of 7 (0.01 mole), hydrazine hydrate (20-25 mL) and ethanol (25 mL) was heated under reflux for 5-7 hours. The product so formed after cooling was filtered off, washed with cold ethanol, and crystallized from diluted ethanol to give 8.

2-(3-amino-4-(4-bromophenyl)-6-phenyl-2H-pyrazolo[3,4-b]pyridin-2-yl)-1-phenylethanone hydrazone 8: as yellow crystals (57%); mp.114-6 °C; IR (cm^{-1}), 3311 (NH_2), 3055 (CH-aromatic) 2966(aliphatic CH) and 1578 (C=N); $^1\text{H-NMR}$

(δ): 1.55 (s, 2H, $-\text{CH}_2-$), 5.41 (s, br., 2H, NH_2 at pyrazole), 6.75 (s, br., 2H, NH_2 , $\text{N}=\text{NH}_2$) and 7.13-8.02 (m, 15H, aromatic and pyridine-CH protons); **MS**: 497 (M^+ , 14.9%) which corresponding to the molecular weight of the molecular formula $\text{C}_{26}\text{H}_{21}\text{BrN}_6$ of the assigned structure, 496 (M^+-H , 23.4%), 480 (M^+-NH_2 , 46.9%), 377 ($\text{M}^+-\text{PhC}=\text{N}-\text{NH}_2$, 73.1%), 364 (M^+-133 , $-\text{CH}_2-\text{C}(\text{Ph})-\text{N}=\text{NH}_2$, 100%), 335 (364- N_2 , H, 20%); *Anal. for* $\text{C}_{26}\text{H}_{21}\text{BrN}_6$ (497.38): Calcd./Found (%): C,62.78/62.80, H,4.26/4.30, Br,16.06/16.10, N,16.90/17.10.

Synthesis of 10: A mixture of 7 (0.01 mole) and formic acid (20-25 mL) was heated under reflux for 5-7 hours. The product so formed after cooling was filtered off, washed with cold ethanol, and crystallized from dioxane to give 10.

N-[4-(4-bromophenyl)-2-(2-oxo-2-phenylethyl)-6-phenyl-2H-pyrazolo-[3,4-b]pyridin-3-yl]formamide 10: as pale yellow crystals (67%); mp. 262-4 °C; **IR** (cm^{-1}): 3178 (NH), 3058 (CH-aromatic), 2965 (aliphatic CH), 1699 (CO) and 1580 (C=N); **¹H-NMR** (δ): 2.11 (s, 2H, $-\text{CH}_2-\text{COPh}$), 6.26 (s, br., 1H, NH), 7.51-8.23 (m, 15H, aromatic and pyridine-CH protons) and 10.26 (s, 1H, $-\text{CHO}$); **MS**: 511 (M^+ , 11.1%) which corresponding to the molecular weight of the molecular formula $\text{C}_{27}\text{H}_{19}\text{BrN}_6\text{O}_2$ of the assigned structure, 512 (M+1, 15%), 513 (M+2, 11.3%), 510 (M^+-H , 13.7%), 405 (M+ $-\text{COPh}$, 71.5%), 392 (M+ $-\text{CH}_2-\text{COPh}$, 1.4%), 377 ($\text{M}^+-\text{NCH}_2\text{COPh}$, 54.4%), 363 (377-N, 3.5%), 334 (363-CHO, 1.6%), 77 (Ph, 100%); *Anal. for* $\text{C}_{27}\text{H}_{19}\text{BrN}_6\text{O}_2$ (511.36): Calcd./Found (%): C,63.42/63.50, H,3.75/3.70, Br,15.63/15.70, N,10.96/11.10.

Synthesis of 11: Mix 4a (0.01 mole), HCl (1.5 mL) and acetic acid (5 mL). Dissolve the formed salt in about 10 ml cold water. Add a mixture sodium nitrite (0.01 mole) and cold water drop wise. Stir for 30 min. then filter the formed solid, wash with water and crystallize the formed solid from ethanol to give 11.

4-(4-bromophenyl)-6-phenyl-1H-pyrazolo[3,4-b]pyridine-3-diazonium chloride (11): as yellow crystals (62%); mp.180-2 °C; **IR** (cm^{-1}): 3454 (NH), 3048 (CH-aromatic) and 2232 (N=N); *Anal. for* $\text{C}_{18}\text{H}_{11}\text{BrClN}_5$ (412.67): Calcd./Found (%): C,52.39/52.40, H,2.69/2.70, Br,19.36/19.30, Cl,8.59/8.60, N,16.97/17.0.

Synthesis of 15a-f: (General Procedure):

Method A: A solution of 12a-f (0.01 mole) in ethanol (50 mL) containing fused sodium acetate (2g) was stirred for 30 min. in ice-cold mixture. Add a mixture of each of 11 in ethanol (50 mL) drop wise, stir for 1hr after complete addition. The solids so obtained were filtered off, washed with water and crystallized from the proper solvent to give 15a-f respectively.

Method B: A solution of 11 (0.01 mole) in ethanol (50 mL) and each of 12a-f (0.01 mole) in the presence of triethylamine (0.5 mL) was heated under reflux for 3 hrs. The solids so obtained were filtered off and crystallized from the proper solvent to give 15a-f respectively.

4-Amino-10-(4-bromophenyl)-8-phenylpyrido[2',3':3,4]pyrazolo[5,1-c][1,2,4]triazine-3-carbonitrile 15a: crystallize from ethanol-DMF mixture as orange crystals (60%); mp. >320 °C; **IR** (cm^{-1}): 3413, 3323, 3159 (NH_2) and 2227 (CN); **MS**: 442 (M^+ , 56.7%) which corresponding to the molecular weight of the molecular formula $\text{C}_{21}\text{H}_{12}\text{BrN}_7$ of the assigned structure, 443 (M+1, 100%), 444 (M+2, 25.22%), 441 (M^+-H , 100%), 425 (M^+-NH_2 , 32.38%), 268 ($\text{M}^+-\text{C}_6\text{H}_5\text{N}_7$, 25.29%); *Anal. for* $\text{C}_{21}\text{H}_{12}\text{BrN}_7$ (442.27): Calcd./Found (%): C,57.03/57.0, H,2.73/2.80, Br,18.07/18.10, N,22.17/22.20.

Ethyl 4-amino-10-(4-bromophenyl)-8-phenylpyrido[2',3':3,4]pyrazolo-[5,1-c][1,2,4]triazine-3-carboxylate 15b: crystallize from ethanol-DMF mixture as orange crystals (58%); mp. >300 °C; **IR** (cm^{-1}), 3398, 3220, 3124 (NH_2), 2927 (CH-aliphatic) and 1693 (C=O); **¹H-NMR** (δ): 1.25 (t, 3H, $\text{CH}_3\text{CH}_2\text{COO}$), 4.18 (q, 2H, $\text{CH}_2\text{CH}_2\text{COO}$), 7.50-8.41 (m, 10H, aromatic and pyridine-CH protons) and 8.78 (s, br. 2H, NH_2); **MS**: 489 (M^+ , 16.36%) which corresponding to the molecular weight of the molecular formula $\text{C}_{23}\text{H}_{17}\text{BrN}_5\text{O}_2$ of the assigned structure, 490 (M+1, 54.75%), 491 (M+2, 12.97%), 488 (M^+-H , 53.34%), 459 (488- C_2H_5 , 14.86%), 416 ($\text{M}^+-\text{COOC}_2\text{H}_5$, 51.89%), 388 (416- N_2 , 19.22%), 364 ($\text{M}^+-\text{COOC}_2\text{H}_5-\text{N}_2-\text{C}_2$, 100%), 334 (364- NNH_2 , 8.88%), 103 (PhCN, 7.30%); *Anal. for* $\text{C}_{23}\text{H}_{17}\text{BrN}_5\text{O}_2$ (489.32): Calcd./ Found (%): C,56.45/56.50, H,3.50/3.60, Br,16.33/16.20, N,17.17/17.20.

3-[Aminosulfanylidene]methyl]-10-(4-bromophenyl)-8-phenylpyrido-[2',3':3,4]pyrazolo[5,1-c][1,2,4]triazin-4-amine 15c: crystallize from ethanol-DMF mixture as yellow crystals (60%); mp. >320 °C; **IR** (cm^{-1}), 3457, 3318, 3162 (two NH_2) and 2925 (CH-aliphatic); **MS**: 477 (M+1, 16%), 475 (M^+-H , 16.62%), 459 (475- NH_2 , 37.08%), 443 (475-S, 100%), 416 (M^+-CSNH_2 , 27.54%), 388 (416- N_2 , 21.54%), 103 (PhCN, 21.69%); *Anal. for* $\text{C}_{21}\text{H}_{14}\text{BrN}_5\text{S}$ (476.35): Calcd./Found (%): C,52.95/53.0, H,2.96/3.10, Br,16.77/16.80, N,20.58/20.60, S,6.73/6.70.

Ethyl 10-(4-bromophenyl)-4-oxo-8-phenyl-3,4-dihydropyrido[2',3':3,4]pyrazolo[5,1-c][1,2,4]triazine-3-carboxylate 15d: crystallize from ethanol-DMF mixture as orange crystals (56%); mp. >300 °C; **IR** (cm^{-1}): 3400, 3200 (H-bonded OH), 2981 (CH-aliphatic) and 1714 (CO ester); **MS**: 489 (M^+-H , 0.59%), 460 (489- CH_3CH_2 , 1.23%), 458 (M^+-OH , CH_3 , 1.58%), 428 (458- OCH_2 , 1.86%), 400 (428- N_2 , 1.62%), 388 (400-C, 1.58%), 285 (388-PhCN, 5.38%), 73 (COOEt , 81.16%), 68 ($-\text{N}=\text{N}-\text{C}-\text{CO}-$, 24.46%), 103 (PhCN, 5.46%), 73 (COOC_2H_5 , 81.16%); *Anal. for* Molecular Formula $\text{C}_{23}\text{H}_{16}\text{BrN}_5\text{O}_3$ (490.30): Calcd./Found (%): C,56.34/56.50, H,3.29/3.30, Br,16.30/16.40, N,14.28/14.30.

1-[10-(4-Bromophenyl)-4-methyl-8-phenylpyrido[2',3':3,4]pyrazolo-[5,1-c][1,2,4]triazin-3-yl]ethanone 15e: crystallize from ethanol-DMF mixture as orange crystals (52%); mp. 320 °C; **IR** (cm^{-1}): 3073 (CH aromatic), 2923 (CH-aliphatic) and 1697 (CO acetyl); **¹H-NMR** (δ): 1.27 (s, 3H, CH_3), 2.11 (s, 3H, COCH_3), 7.63-8.48 (m, 10H, aromatic and pyridine-CH protons); **MS**: 458 (M^+ , 30.56%) which corresponding to the molecular weight of the molecular formula $\text{C}_{23}\text{H}_{16}\text{BrN}_5\text{O}$ of the assigned structure, 459 (M+1, 100%), 460 (M+2, 26.49%), 415 (M^+-COCH_3 , 8.71%), 416 (459- COCH_3 , 29.76%), 387 (415- N_2 , 8.21%), 388 (416- N_2 , 9.72%), 103 (PhCN, 3.02%); *Anal. for* Molecular Formula $\text{C}_{23}\text{H}_{16}\text{BrN}_5\text{O}$ (458.31): Calcd./Found (%): C,60.27/60.30, H,3.52/3.60, Br,17.43/17.40, N,15.28/15.30.

Ethyl 10-(4-bromophenyl)-4-methyl-8-phenylpyrido[2',3':3,4]pyrazolo-[5,1-c][1,2,4]triazine-3-carboxylate 15f: crystallize from ethanol-DMF mixture as orange crystals (58%); mp. 258-60 °C; **IR** (cm^{-1}): 3049 (CH aromatic), 2923 (CH-aliphatic) and 1719 (CO ester); **¹H-NMR** (δ): 1.49 (s, 3H, CH_3), 1.57 (t, 3H, $\text{CH}_3\text{CH}_2\text{COO}$), 4.63 (q, 2H, COCH_2CH_3) and 7.26-8.37 (m, 10H, aromatic and pyridine-CH protons); **MS**: 488 (M^+ , 15.98%) which corresponding to the molecular weight of the molecular formula $\text{C}_{24}\text{H}_{18}\text{BrN}_5\text{O}_2$ of the assigned structure, 489 (M+1, 54.94%), 460 (M+1- C_2H_5 , 12.79%), 459 ($\text{M}^+-\text{C}_2\text{H}_5$, 38.04%), 415 ($\text{M}^+-\text{COOC}_2\text{H}_5$, 98.03%), 416 (M+1- COOC_2H_5 , 63.17%), 387 (415- N_2 , 12.60%), 388 (416- N_2 , 50.95%), 361 (388- CCH_3 , 7.37%), 348 ($[\text{M}^+-\text{N}_2\text{C}(\text{COOC}_2\text{H}_5)=\text{C}(\text{CH}_3)]$, 10.82%); *Anal. for* Molecular Formula $\text{C}_{24}\text{H}_{18}\text{BrN}_5\text{O}_2$ (488.33): Calcd./Found (%): C,59.03/59.10, H,3.72/3.80, Br,16.36/16.40, N,14.34/14.20.

BIBLIOGRAPHY

- ⁽¹⁾ F.A. Attaby: *Arch. Pharmacol. Res.*, **13**, 342 (1990).
- ⁽²⁾ F.A. Attaby, L. I. Ibrahim, S. M. Eldin and A. K. K El-Louh: *Phosphorous, Sulfur and Silicon*, **73**, 127 (1992).
- ⁽³⁾ F.A. Attaby, S. M. Eldin and M. Abdel Razik: *Phosphorous, Sulfur and Silicon*, **106**, 21 (1994).
- ⁽⁴⁾ F.A. Attaby and A.M. Abdel-Fattah: *Phosphorous, Sulfur and Silicon*, **119**, 257 (1996).
- ⁽⁵⁾ F.A. Attaby: *Phosphorous, Sulfur and Silicon*, **126**, 27 (1997).
- ⁽⁶⁾ F.A. Attaby: *Phosphorous, Sulfur and Silicon*, **139**, 1 (1998).
- ⁽⁷⁾ F.A. Attaby, S.M. Edin and M. A. A. Elneairy: *Heteroatom Chemistry*, **9**, 571 (1998).
- ⁽⁸⁾ F.A. Attaby, S.M. Edin and M.A.A. Elneairy; *J. Chem. Res. (M)*, **10**, 2754, (S) 10 (1998).
- ⁽⁹⁾ F.A. Attaby, M.A.A. Elneairy and M.S. Elsayed: *Phosphorous, Sulfur and Silicon*, **149**, 230 (1999).
- ⁽¹⁰⁾ F.A. Attaby and A.M. Abdel-Fattah: *Phosphorous, Sulfur and Silicon*, **155**, 253 (1999).
- ⁽¹¹⁾ M.A.A. Elneairy, Sanaa M. Eldin, Fawzy A. Attaby and A.K.K. El-Louh: *Phosphorous, Sulfur and Silicon*, **167**, 289 (2000).
- ⁽¹²⁾ M.A.A. Elneairy, F.A. Attaby and M.S. Elsayed: *Phosphorus, Sulfur and Silicon*, **167**, 161 (2000).
- ⁽¹³⁾ F.A. Attaby, M.A.A. Elneairy, S.M. Eldin and A.K.K. El-Louh: *J. C. C. S. (China)*; **48(5)**, 893 (2001).
- ⁽¹⁴⁾ F.A. Attaby, H.M. Mostafa, A.H.H. Elghandour and Y.M. Ibrahim: *Phosphorus, Sulfur and Silicon*; **177**, 2753 (2002).
- ⁽¹⁵⁾ F.A. Attaby, A.H.H. Elghandour, H.M. Mustafa and Y.M. Ibrahim: *J. C. C. S. (China)*; **49(4)**, 561 (2002).
- ⁽¹⁶⁾ F.A. Attaby, S.M. Eldin, M.A.A. Elneairy and A.K.K. Elouh: *Phosphorus, Sulfur and Silicon*; **179**, 2205 (2004).
- ⁽¹⁷⁾ F.A. Attaby, A.H.H. Elghandour, Ali M.A. and Y.M. Ibrahim: *Phosphorus, Sulfur and Silicon*; **181**, 1 (2006).
- ⁽¹⁸⁾ F.A. Attaby, Ali M.A., A.H.H. Elghandour and Y.M. Ibrahim: *Phosphorus, Sulfur and Silicon*; **181**, 1087 (2006).
- ⁽¹⁹⁾ Fawzy A. Attaby, A.H.H. Elghandour, M.A. Ali and Yasser M. Ibrahim: *Afinidad*; **63**, 525 (2006).
- ⁽²⁰⁾ Fawzy A. Attaby, A.H. Elghandour, M.A. Ali and Yasser M. Ibrahim: *Phosphorus, Sulfur and Silicon*; **182**, 133 (2007).
- ⁽²¹⁾ Fawzy A. Attaby, A.H.H. Elghandour, M.A. Ali and Yasser M. Ibrahim: *Phosphorus, Sulfur and Silicon*; **182**, 695 (2007).
- ⁽²²⁾ Fawzy A. Attaby, Mostafa M. Ramla and Eman M. Gouda: *Phosphorus, Sulfur and Silicon*; **182**, 517 (2007).
- ⁽²³⁾ B.R. Tolf, R. Dahlbom, H. Theroell, and A. Akeson: *Acta Chem. Scand., Ser. B.*; **36**, 101 (1982).
- ⁽²⁴⁾ M. Komuro, R. Ishida and H. Uchida: *Arzneim-Forsch*, **42**, 48 (1992); *C. A.*, **116**, 98851q (1992).
- ⁽²⁵⁾ F.E. Goda, A.A.M. Abdel-Aziz and O.A. Attef: *Bioorg. Med. Chem.* **12**, 1845 (2004).
- ⁽²⁶⁾ F.C. Brown and C.K. Bradsher: *Nature*; **168**, 171 (1951).
- ⁽²⁷⁾ D.R. Rao, S.P. Raychaudhuri and V.S. Verma: *International Journal of Tropical Plant Diseases*; **12**, 177 (1994).
- ⁽²⁸⁾ F. Szurdki, L. Jaerger, A. Harris, H. Kido, I. Wengatz and M.H. Goodrow: *Journal of Environmental Science and Health part B*; **31**, 451 (1996).
- ⁽²⁹⁾ F.F. Benzie and J.J. Strain: *Anal. Biochem.*; **15**, 239 (1996).
- ⁽³⁰⁾ P.G. Baraldi, B. Cacciari, A. Dalpiaz and S. Dionisotti: *Arzneim-Forsch*; **46**, 365 (1996).