

REVIEW ARTICLE

DNA Related Enzymes as Molecular Targets for Antiviral and Antitumoral Chemotherapy. A Natural Overview of the Current Perspectives

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Abstract: Background: The discovery of new chemotherapeutic agents still remains a continuous goal to achieve. DNA polymerases and topoisomerases act in nucleic acids metabolism modulating different processes like replication, mitosis, damage repair, DNA topology and transcription. It has been widely documented that Polymerases serve as molecular targets for antiviral and antitumoral chemotherapy. Furthermore, telomerase is a ribonucleoprotein with exacerbated activity in most of the tumor cell lines, becoming as an emergent target in Cancer treatment.

Methods: We undertook an exhaustive search of bibliographic databases for peer-reviewed research literature related to the last decade. The characteristics of screened bibliography describe structure activity relationships and show the principal moieties involved. This work tries to summarize the investigation about natural and semi-synthetic products with natural origin with the faculty to inhibit key enzymes that play a crucial role in DNA metabolism.

Results: Eighty-five data references were included in this review, showing natural products widely distributed throughout the plant kingdom and their bioactive properties such as tumor growing inhibitory effects, and anti-AIDS activity.

Conclusion: The findings of this review confirm the importance to find new drugs and biologically active natural products, and their potential medicinally useful benefits.

ARTICLE HISTORY

Received: February 05, 2018
Revised: April 17, 2018
Accepted: April 19, 2018

DOI:
10.2174/1389450119666180426103558

Keywords: DNA polymerases, telomerases, topoisomerases, inhibition, molecular target, chemotherapy.

1. INTRODUCTION

Natural products and their derivatives have been used for a long time in traditional medicine. In addition to their low cost and wide practice, natural products and their derivatives attract growing interests and are valuable resources for drug design and pharmaceutical development. Plants have a long history of use in the treatment of cancer. Scientists have identified many molecules with anticancer activities from nature, including taxol, vinblastine, camptothecin, paclitaxel, cephalotaxine, matrine, and colchicine [1-4].

Chemotherapy remains to be the standard treatment for initial or advanced stages of cancer. However, commonly used chemotherapeutic agents may induce damage in healthy cells or tissues. Thus, in recent years, there has been an increased effort in the development of new, efficient anticancer drugs exhibiting low toxicity, that are not affected by mechanisms of chemo-resistance [5-7].

Complexity and molecular diversity, high selectivity and specific biological activities are characteristics of natural

products, and many of these metabolites have some receptor-binding activity in a specific manner [8]. Consequently, some proteins can become true molecular targets and then develop certain pharmacological action. One kind of targets could be DNA-related enzymes. These receptors play a crucial role in DNA metabolism, including processes such as recombination, replication, transcription and chromosome segregation during cell division. The inhibition of these **al-**
low the development of new leads, like antitumoral and antiviral agents. Considering the enormous complexity and structural diversity of natural products, we can find alkaloids, chalcones and flavonoids, iridoids, polyalcohols, lipids and terpenoids, macrocycles, quinones, and polyphenols capable of showing this activity, but with the accuracy of maintaining a defined configuration of its chiral centers. This work tries to summarize the investigation about some remarkable natural and semi-synthetic products that share a natural origin, with the ability to inhibit key DNA-related enzymes [9, 10].

2. DNA POLYMERASES

DNA polymerases are ubiquitous enzymes with the ability to create new polymers of deoxyribonucleic acid from four different kinds of substrates, known as nitrogenous DNA bases. These proteins, together with different topoisomerases

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and helicases, play a crucial role in DNA metabolism, such as recombination, replication, transcription, and chromosome segregation during meiosis and mitosis. Hyperproliferative diseases such as cancer, autoimmune conditions, and viral/bacterial infections are associated with uncontrollable DNA replication. For this reason, it has long been accepted that these enzymes are important molecular targets for the development of cancer and antiviral chemotherapeutic agents [11-17].

In the subsequent paragraphs, some natural products and derivatives with DNA polymerase inhibitory activity are categorized according to their chemical structures.

2.1. Flavonoids and Chalcones

The flavonoid group comprises a large family of plant-derived polyphenolic compounds of medium to low molecular weight, which exhibit diverse biological activities. The naturally occurring chalcones, or benzopyrone derivatives, and the phenylchromones are widely distributed throughout the plant kingdom as components in fruits, vegetables, tea, bark, roots, flowers, grains, *etc.* [18, 19].

Myristinin is a bioflavonoid isolated from *Myristica cinnamomea* [20]. This occurs as different stereoisomers known as myristinin A and myristinin B/C. In 2004, Maloney and co-authors prepared (+)-myristinin stereoselectively with absolute stereochemistry. In addition to DNA-damaging ability, (+)-myristinin was also shown to inhibit DNA polymerase β ($IC_{50} = 2.8 \mu M$), a DNA-repairing enzyme (Fig. 1) [21].

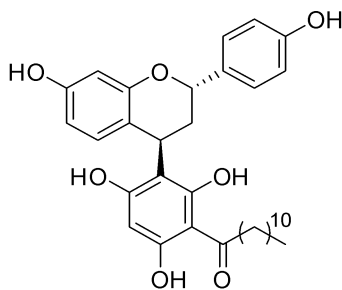


Fig. (1). Chemical structure of (+)-myristinin.

Another kind of non-glycosylated flavonoids able to inhibit DNA polymerase β was 5-methoxyflavone. The three-ringed molecular scaffold of this flavone is easy to synthesize in the development of nature-inspired compounds (Fig. 2) [22].

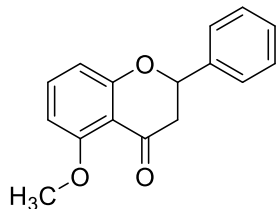


Fig. (2). Chemical structure of 5-methoxyflavone.

Drugs with low molecular weight have usually shown a good ability to cross cell membranes by passive diffusion, such as the nuclear membrane. Furthermore, simple alkylat-

ed flavonoids have shown selective pharmacological properties, such as xanthohumol. This prenylated chalcone exhibited remarkable inhibitory activities against human DNA polymerase α (Fig. 3) [23].

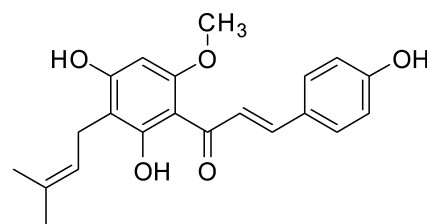


Fig. (3). Chemical structure of xanthohumol.

It should be noted that the use of more complex structures increases the probability of improving the physicochemical interactions between molecules and receptors. In this case, biflavonoids from *Dacrydium balansae* have shown a potent inhibition of the Dengue virus NS5 RNA-dependent RNA polymerase (Fig. 4) [24].

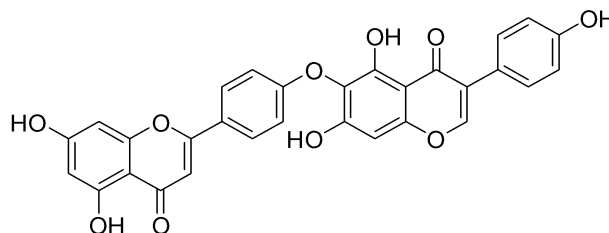


Fig. (4). Chemical scaffold of a kind of active biflavonoid.

Although the receptor used is not specifically a DNA-dependent polymerase, it would be interesting to conduct a study on this kind of enzyme using flavonoid dimers. The use of duplicate molecular scaffolds, in addition to favoring greater structural complexity, generally leads to a better cellular permeability through passive diffusion mechanisms, which results from a greater hydrophobicity [25].

2.2. Lipids and Terpenoids

Lipids represent a class of metabolites with interesting values of cLogP mainly due to their hydrocarbon constitution. These can be made as different forms: from molecules with very low molecular weight to complex structures with more than 100 carbon atoms like dolichol. Terpenes are secondary metabolites derived from a unit of five-carbon atoms called isoprene, thus sharing different building blocks of this [26-29].

The first attempts to discover β -polymerase inhibitors conducted by Mizushima *et al.* revealed that long chain fatty acids suppressed polymerase activity [30]. They also found that terpeno-benzoic acids, from the plant *Myrsine seguinii*, showed an inhibitory effect against enzymes involved in DNA synthesis, such as calf DNA polymerase α (pol. α), rat DNA polymerase β (pol. β) and one of the β family polymerases, calf thymus terminal deoxynucleotidyl transferase (TdT). Its IC_{50} values were $22 \mu M$ for pol. α , $11 \mu M$ for pol. β , and $46 \mu M$ for TdT, respectively (Fig. 5) [31]. These authors suggested that the presence of carboxylic acid is an

important structural factor for the DNA polymerase inhibitory activity.

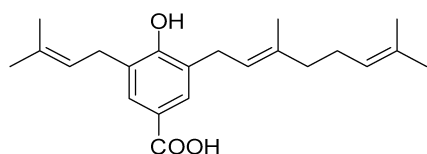


Fig. (5). Chemical structure of a terpeno-benzoic acid from *Myrsine seguinii*.

Another class of attractive lipid carboxylic acid which has shown this activity is 7,8-euphadien, a type of triterpenoid from *Brackenridgea nitida* and *Bleasdalea bleasdalei*. This product inhibited rat DNA polymerase α with an IC_{50} value of $23 \mu M$ in the presence of bovine serum albumin (BSA) and $9.7 \mu M$ in the absence of BSA (Fig. 6) [32].

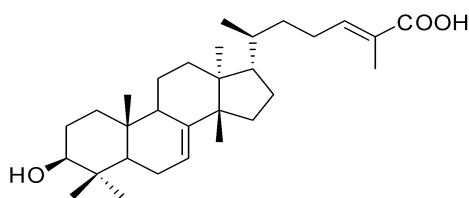


Fig. (6). Chemical structure of 7,8-euphadien.

Furthermore, bioassay-guided fractionation of extracts prepared from *Couepia polyandra* and *Edgeworthia gardneri* resulted in the isolation of the DNA polymerase β inhibitors, oleanolic acid ($IC_{50} = 24.98 \pm 3.3 \mu M$) and betulinic acid ($IC_{50} = 46.25 \pm 3.1 \mu M$), (Fig. 7) [33].

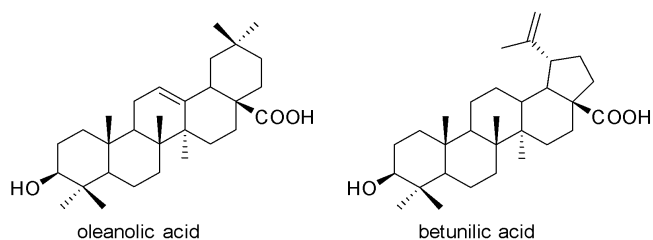


Fig. (7). Chemical structure of oleanolic and betulinic acids.

2.3. Iridoids and Other Polyols

Iridoids, glycosides and other polyols display an elevated content of oxygen, giving these molecules high water solubility. Their poor lipophilicity might account for the lack of biological activity against human solid tumor cell lines. This may appear, in principle, to be a disadvantage when crossing lipid membranes; but the presence of hydroxyl groups increases the probability of direct enzyme recognition through hydrogen bonds [34-36].

Iridoids may even emulate the nitrogenous substrates of DNA-related enzymes. The bicyclic aglycone of DNA polymerase inhibitor catalpol could mimic the purine scaffold present in natural nucleosides because they share similar electronic surface distribution (Fig. 8) [37].

In 2010, our research group synthesized a novel catalpol derivative able to inhibit the KlenTaq fragment. By conduct-

ing molecular dynamic studies, we proposed a recognition inside the catalytic site by hydrogen bonds between glycosidic hydroxyls and polar chain residues (Fig. 9) [36].

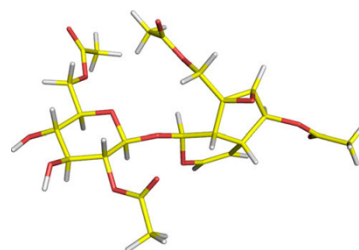


Fig. (8). Conformational structure of a catalpol derivative.

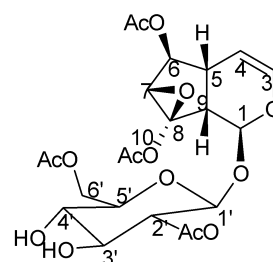


Fig. (9). Chemical structure of 6,10,2',6'-tetraacetyl-*O*-catalpol.

Furthermore, Pungitore *et al.* prepared a series of novel lipophilic catalpol analogs by the regioselective addition of silylether moieties. Antimetabolites cannot enter cells easily by passive diffusion due to their low lipophilicity. However, these products did show antitumoral activity, possibly due to their higher cLogP values (Fig. 10) [38].

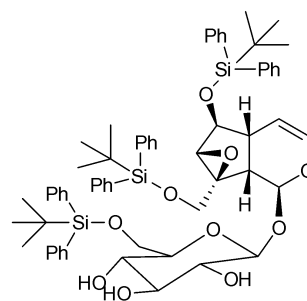


Fig. (10). Chemical structure of a lipophilic catalpol derivative.

Recently, Garro *et al.* isolated a new epimer iridoid of catalpol from *Buddleja cordobensis* [39]. This metabolite was even more active than catalpol. Similarly, two silylether derivatives with higher lipophilicity also showed inhibition. Considering a different configuration of carbon six, the activity may improve; and hydrophobic derivatives could act against human solid tumors (Fig. 11).

From *Buddleja cordobensis*, we also could isolate verbascoside as a *per-O*-acetyl derivative. This disaccharide has shown to be a potent inhibitor with $IC_{50} = 1.21 \mu M$, possibly due to its complex structure and the large number of chiral centers (Fig. 12).

Finally, the natural polyalcohol petasiphenol, extracted from the Japanese plant *Petasites japonicus*, was shown to selectively inhibit pol λ activity, but it resulted to be inactive

towards the structurally-related polymerase β , as well as towards replicative polymerases (Fig. 13). Petasiphenol inhibited the pol λ activity indirectly by acting at the BRCT (C-terminal domain of a breast cancer susceptibility protein), and could not recognize the BRCT domain structure of other proteins, suggesting that the three-dimensional structure of the BRCT domain of pol λ differs of pol β [40].

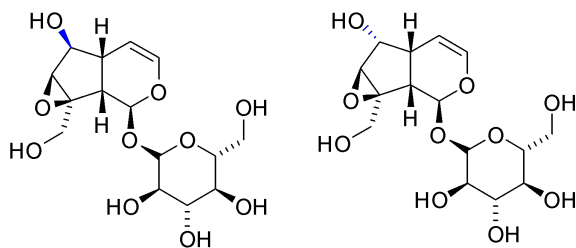


Fig. (11). Chemical structures of catalpol (left) and 6-*epi*-catalpol (right).

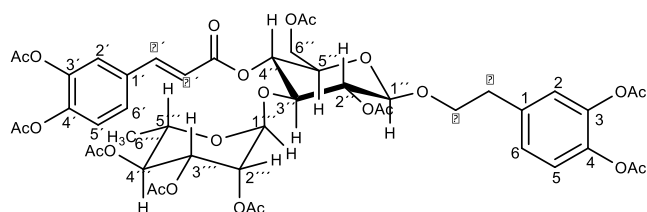


Fig. (12). Chemical structure of *per-O*-acetyl-verbascoside.

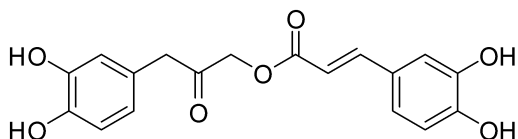


Fig. (13). Chemical structure of petasiphenol.

3. TELOMERASE

Telomeres are repetitive DNA elements at the end of chromosomes that protect against chromosomal erosion and rearrangement inside the nucleus. This unique structural formation of DNA is different from the well-known helical structure that DNA adopts. Furthermore, it has recently been proposed as a molecular target for new kinds of anticancer agents, in particular, molecules able to stabilize the intramolecular G-quadruplex. Telomerase is part of an enzyme complex that counteracts telomeres shortening. This protein is constituted by the catalytic subunit telomerase reverse transcriptase (TERT) and its RNA component sequence [41, 42].

Human telomeres are simple (TTAGGG) n islands of 5–15 kb repeated sequences located at the ends of chromosomes. This structure capping the chromosome termini prevents aberrant recombination. The mutagenesis events caused by these aberrant changes could eventually result in a malignant cell differentiation. Highly proliferative cells require elevated levels of telomerase activity to offset such loss of telomeric DNA. This enzyme has become an obligatory objective for anti-cancer research, as it promised a unique target for chemotherapy without the related side effects of conventional chemotherapeutics [43, 44]. As an anticancer

strategy, efforts have been invested in targeting and lowering telomerase activity, which is frequently found to be overexpressed in cancerous cells [45].

3.1. Alkaloids

Quinoline compounds are characterized by a double-ring structure that contains a benzene ring fused to a heterocyclic aromatic nitrogen pyridine. These products have shown strong inhibition with sub-toxic concentrations, such as 11-substituted cryptolepine analogs (Fig. 14).

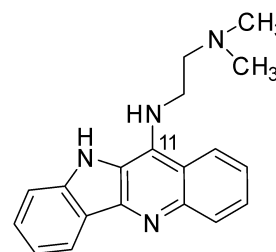


Fig. (14). Chemical structure of 11 substituted quindolines.

In 2005, Zhou *et al.* demonstrated its capacity to stabilize a G-quadruplex, and that an electron-donating substituent at the 11-position can enhance the basicity of the nitrogen atom, thus increasing the electrostatic interaction between the derivatives and the negative electrostatic center of the G-quadruplex [46, 47].

In the last decade, Neidle's and Huang's groups reported that 13 and 9-substituted berberine derivatives had the ability to selectively bind to G-quadruplex over double-stranded DNA, including those present in the human oncogene *c-myc* sequence (Fig. 15) [48, 49].

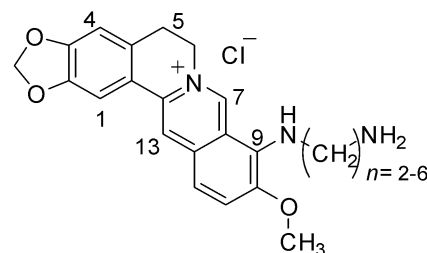


Fig. (15). Chemical structure of 9 substituted berberine analogs.

Overexpression of this oncogene is related to the increase of cellular and malignant proliferation. Guanine rich sequences present in its promoter region controls 80-90% of the transcription activity of *c-myc*. Several small molecules, like chelerythrine, which can stabilize this quadruplex structure, are putative agents to downregulate *c-myc* expression (Fig. 16). Interestingly, the protonated iminium form of chelerythrine is able to intercalate into double stranded DNA [50].

Electrospray ionization mass spectrometry (ESI-MS) and Circular Dichroism (CD) can be used to determine the formation and recognition of human telomere G-quadruplex and selectivity *versus* duplex DNA. For example, the binding of homobarringtonie, a natural alkaloid with a seven-membered heterocycle, was established using these methods (Fig. 17) [51].

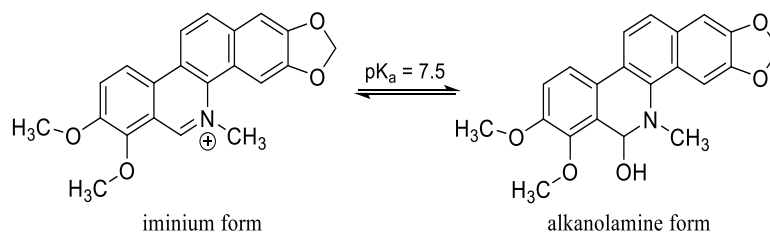


Fig. (16). Chelerythrine forms.

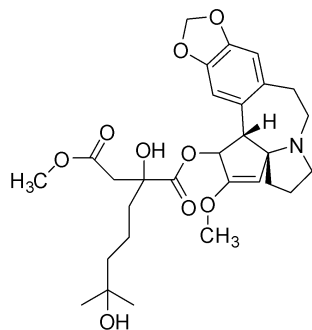


Fig. (17). Chemical structure of homobarringtonic.

3.2. Porphyrins and Perylene Compounds

Polycyclic aromatic hydrocarbons (PAHs) are a large group of hydrophobic chemicals, characterized by presenting conjugate systems within a planar geometry [52]. Some porphyrin derivatives bearing *N*-methylpyridinium cationic side arms are able to bind with parallel four-stranded (TG₄T)₄ G-quadruplex DNA in buffers containing Na⁺ cation (Fig. 18) [53].

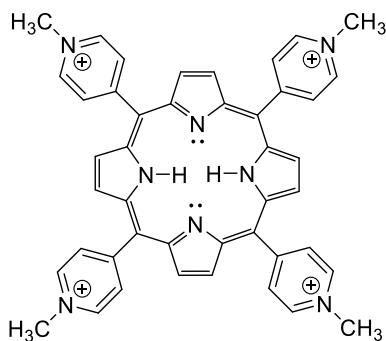


Fig. (18). Chemical structure of porphyrin derivatives bearing *N*-methylpyridinium cations.

Perylenes and coronenes are hydrosoluble PAHs able to recognize and specific binding of telomeric G-quadruplexes. These compounds contain a large planar aromatic system moiety, which can bind to the terminal G-quartet of a G-quadruplex *via* $\pi-\pi^*$ stacking in a threading intercalation mode. For example, Franceschin *et al.* prepared four “not self-aggregated” perylene derivatives with activities lower than 20 μ M of telomerase inhibition (Fig. 19).

The most interesting point was the strong inhibition and the water solubility of these, which suggest a potential low toxicity [54]. The same group of researchers also prepared coronene derivatives capable of specifically recognizing secondary structures of G-quadruplex (Fig. 20) [55].

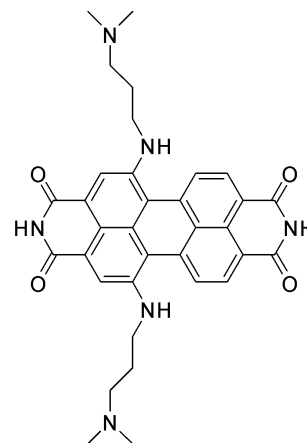


Fig. (19). Chemical structure of a perylene derivative.

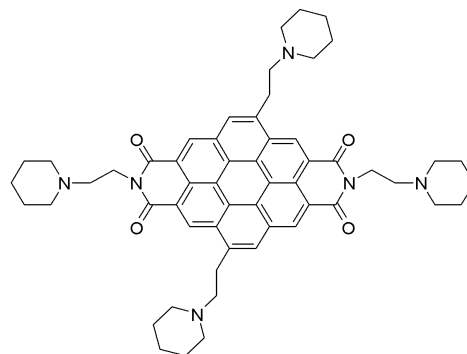


Fig. (20). Chemical structure of a coronene derivative.

3.3. Quinones

Quinones containing compounds have been widely used for their antitumour activity. Nature is an important source of novel quinone analogues, such as *para* and *ortho*-benzoquinones, naphthoquinones and anthraquinones. Almost 350 natural naphthoquinones have been discovered. Now, the co-existence of naphthoquinones and anthraquinones in plants is not atypical, and they have a common precursor. New types include isofuranonaphthaquinones and naphthoquinones linked to coumarin, which form substantial subgroups with interesting activities [56]. For example, celastrol a quinone methide triterpenoid from the Chinese medicinal root bark of *Tripterygium wilfordii*, possesses beneficial therapeutic properties. A series of celastrol derivatives were prepared as potential telomerase inhibitors, showing a direct correlation between telomerase inhibition and anti-proliferative inhibition of SMMC-7721 in human hepatoma cells. This compound with hydrophilic moieties show potential non-toxic side effects (Fig. 21) [57, 58].

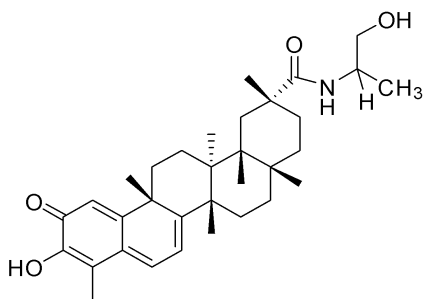


Fig. (21). Chemical structure of a hydrophilic celastrol derivative.

The tanshinone natural products have a variety of pharmacological properties including, anti-neoplastic activity. *Ortho*-quinone containing tanshinones were telomerase inhibitors through an oxidative mechanism mediated by production of hydrogen peroxide. To examine the importance of the *ortho*-quinone moiety, Soares *et al.* examined the telomerase inhibition by telomerase inhibitors and the simple *ortho*-quinone 9,10-phenanthrenequinone (PHQ). They found that the *ortho*-quinone tanshinone Tan II-A and PHQ were equipotent inhibitors with $IC_{50} = 5.0 \mu M$ (Fig. 22) [59].

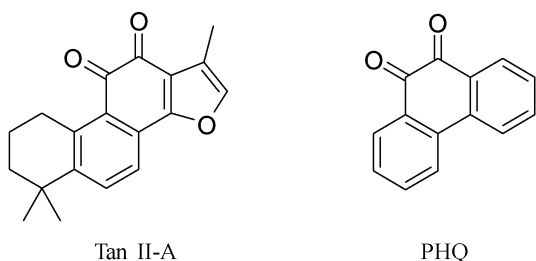


Fig. (22). Chemical structures of an *ortho*-quinone tanshinone (left) and 9,10-phenanthrenequinone (right).

Rubromycins are a family of structurally related compounds originally isolated from *Actinomycetes* that exhibit an attractive array for pharmacology (Fig. 23). Telomerase is a dimeric protein consisting of two reverse transcriptase motifs (TERT) with two RNA components (TR). β -rubromycin has revealed a competitive interaction with the telomerase substrate primer interacting both with the human telomerase RNA (TR) and the catalytic subunit (TERT), ($IC_{50} = 3.06 \mu M$) [60].

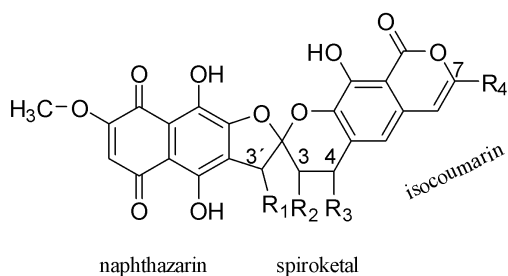


Fig. (23). General scaffold of rubromycins.

It was established that the high biological activity of rubromycin compounds significantly depends on the presence of the [5,6]-bisbenzannulated spiroketal moiety as a

central structural motif. An elegant synthesis of (\pm) γ -rubromycin was made by Michael Wilsdorf and Hans-Ulrich Reissig, opening novel and challenging methods (Fig. 24) [61, 62].

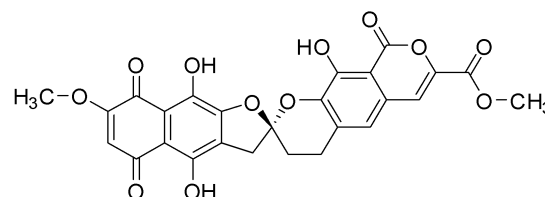


Fig. (24). Chemical structure of (\pm)- γ -rubromycin.

4. TOPOISOMERASE

Another vital enzymes implicated in the replication, transcription, recombination or DNA repair processes are the topoisomerases [63, 64]. These proteins direct and supervise the topological arrangement and orientation of DNA by relaxing torsion stress in supercoiled DNA sectors [65].

There are two major classes of topoisomerases, that share common characteristic, considering the nicking mechanism. Enzymes that cleave one strand of the DNA double strand are defined as topo I class; isoforms that cleave both strands are catalogued as topo II proteins [66, 67]. Topo I produces the relaxation of DNA chain during replication by a single strand break in DNA. The single strand break is then resealed, thus restoring the DNA double strands, without any energy cofactor [68].

Today, in clinical and medical practice, mammalian DNA topoisomerases are considered prominent cellular targets of several anti-cancer drugs [69]. Perhaps, camptothecin analogues are the most topoisomerase inhibitors used; however, another kind of natural products are promising candidates.

4.1. Lipids and Terpenes

Mizushima *et al.* reported that some triterpenes are able to inhibit human topoisomerase II and some DNA polymerases. Remarkably, topo II and polymerase β have a similar three-dimensional triterpene-binding region, which is a hydrophobic pocket, able to specifically bind terpenoid compounds (Fig. 25).

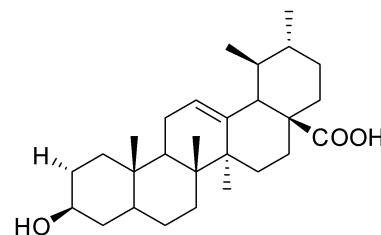


Fig. (25). Chemical triterpene scaffold from plant callus tissues of some Chinese medicinal plants.

Moreover, the hydroxyl moiety at “A ring” might generate a weaker inhibitory activity, but not the carboxyl group. This acidic group could have the required electronic density surface with the rest of terpene scaffold necessary for such inhibition.

Moreover, Guo-Wei Wang *et al.* have isolated new Lanostane-type triterpenoids from *Abies faxoniana* (Fig. 26) [70].

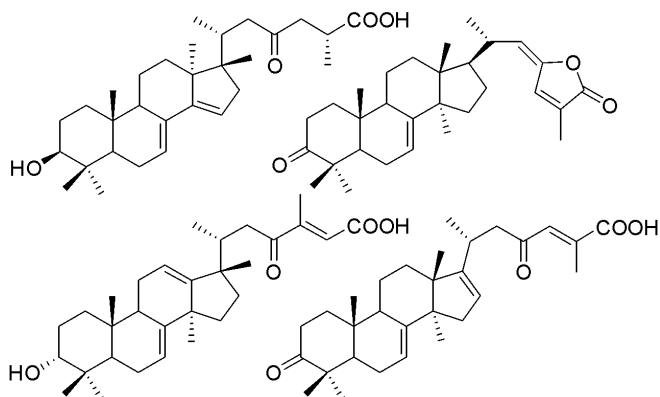


Fig. (26). Chemical structures of some lanostane-triterpenoids from *Abies faxoniana*.

Three of them have shown activity, and they share the carboxylic moiety at the end of the macrocyclic terpene structure. Interestingly, the fourth active compound shows a lactone moiety instead of the carboxylic residue, showing the best IC_{50} values. These chemically related groups share approximately the same van der Waals surface for each arrangement. However, the lactone group show a higher probability to insert inside of terpene topo hydrophobic pocket [71].

Another kind of derivatives bearing α,β -unsaturated lactone are gibberellins. These potent topoisomerases I inhibitors belong to the tetracyclic diterpenes and show attractive $cLogP$ values (Fig. 27) [72].

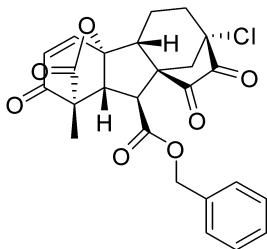


Fig. (27). Chemical structure of a gibberellin hydrophobic analogue.

4.2. Alkaloids

Fagaronine has been shown to interact inside of DNA, as well as with double stranded regions of ribonucleic acids, like tRNA (Fig. 28). As a cytotoxic agent, its activity seems to depend on the presence of the quaternary nitrogen because the desmethyl congener was inactive [73, 74].

Berberine salts analogues show a slightly buckled structure due to the partial saturation of the central ring. Berberine has been previously characterized as a DNA intercalating agent and as a cationic ligand, and electrostatic forces could play an important role in its interaction with DNA phosphates. Furthermore, electric linear dichroism experiments have shown that burasaine also interacts with DNA minor groove in an intercalation mode of binding (Fig. 29) [75].

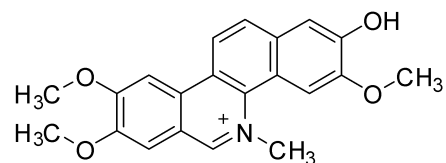


Fig. (28). Chemical structure of fagaronine.

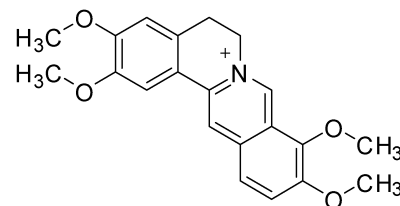


Fig. (29). Chemical structure of burasaine.

Theoretical simulations using molecular docking studies have shown that these planar compounds bound with topoisomerase/DNA complex, mainly through π - π stacking interactions [76].

Last year, Ashish Kumar *et al.* isolated a protoberberine group of isoquinoline alkaloids from *Thalictrum foliolosum*, including thalifendine and berberine. Surprisingly, they could find two new bisbenzylisoquinoline alkaloids. One of them displayed potent inhibitory activity against topoisomerase I in a concentration dependent manner (Fig. 30). Apparently, it did not stabilize enzyme/DNA covalent complex, acting as a class II inhibitor, which is a catalytic inhibitor of the enzyme [77].

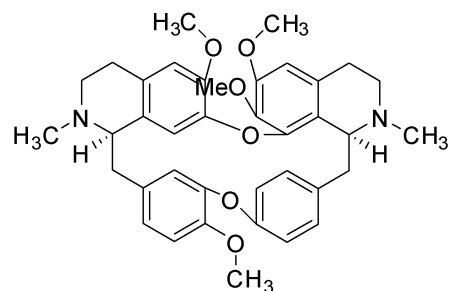


Fig. (30). Chemical structure of a bisbenzylisoquinoline alkaloid isolated from *Thalictrum foliolosum*.

Finally, β -carboline alkaloids represent a promising class of antitumoral drugs. *Galianthe thalictroides* (Rubiaceae), is a widespread shrub in Argentina, Brazil, Paraguay, and Uruguay, where it is popularly used to treat and prevent Cancer. A new cytotoxic β -carboline alkaloid from its roots, 1-methyl-3-(2-hydroxypropan-2-yl)-2-(5-methoxy-9H- β -carbolin-1-yl) cyclopentanol, inhibited both topoisomerase I and topoisomerase II α better than etoposide, in a tested concentration of 1 μ M (Fig. 31) [78, 79].

4.3. Flavonoids and Chalcones

Flavonoids are widely spread in the plant kingdom and are common constituents of vegetables, fruits, and certain beverages. In 2010 López-Lázaro *et al.* evaluated the effects of four common dietary flavonoids (quercetin, apigenin,

fisetin and myricetin) on topoisomerases at several concentrations using a cell-based assay (Fig. 32) [80, 81].

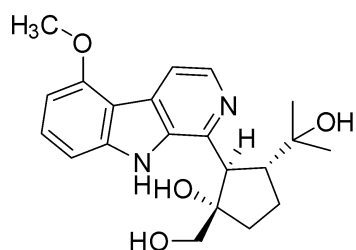


Fig. (31). Chemical structure of a β -carboline alkaloid isolated from *Galianthe thalictroides*.

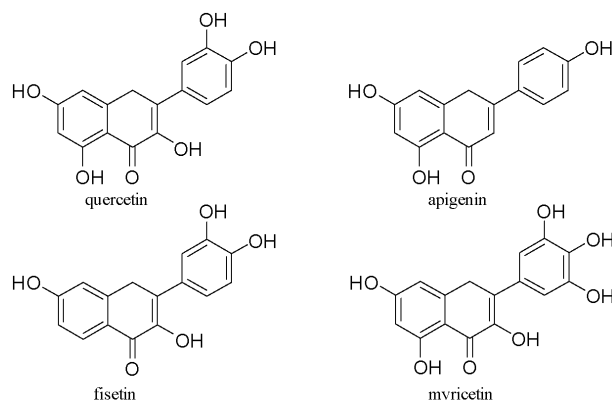


Fig. (32). Chemical structures of some common dietary flavonoids.

They found that, in addition to the cancer preventive effects showed by some flavonoids, at higher concentrations (micromolar), these products may induce topoisomerase-mediated DNA alterations that might have carcinogenic effects. This denotes the importance of finding more selective and active inhibitors at low concentrations.

Furthermore, quercetin diacylglycoside analogues have shown interesting activities against bacterial DNA gyrase and topoisomerase IV (Fig. 33). These compounds could be interesting for the study against human topoisomerases using a more rational design [82].

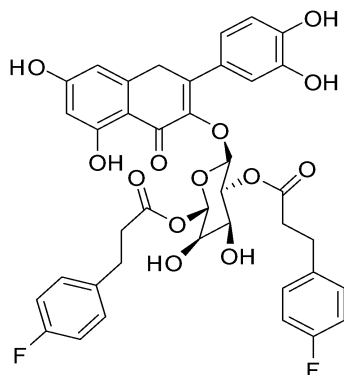


Fig. (33). Chemical scaffold of a quercetin diacylglycoside analogue.

The acyl glycoside present in flavonoids products seems to be an attractive moiety for activity. For example, a

kaempferol glycoside derived from parts of two unique varieties of the Leguminosae, *Vicia faba* and *Lotus edulis* from Crete acted as catalytic inhibitor of wheat germ topoisomerase I and human topoisomerases I and II (Fig. 34) [83].

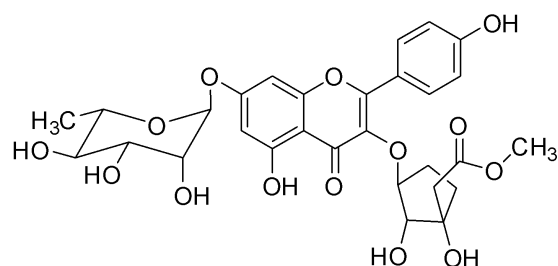


Fig. (34). Chemical structure of 2, kaempferol 3-O-(5-O-acetyl- α -D-apiofuranosyl)-7-O- α -L-rhamnopyranoside.

Finally, using the chalcone scaffold as starting material, interesting families of inhibitors can be achieved by simple preparation methods; for example, by incorporating heteroaromatic cyclopropane rings (Fig. 35) [84, 85].

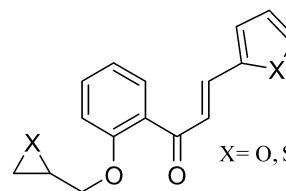


Fig. (35). Chemical scaffold of chalcone derivatives.

CONCLUSION

Several of the essential molecular mechanisms underlying malignant tumor biology remain still elusive and, thus, developing specific anticancer therapies remains a challenge. Medicinal plants and their natural products possess numerous bioactive principles which attack multiple targets, showing minor side effects, low potential to cause resistance, and remarkably low costs of production. The idea of DNA replication or DNA-repair interference as a potential tool approach to overcome intrinsic or acquired tumor resistance is gaining substantial attention. With respect to the inhibition of DNA repair pathways, it is desirable to avoid hurting normal cells.

Polymerases have a highly preserved structure, which means that their overall catalytic subunits motifs fluctuate very little from species to species. A preserved structure usually indicates that the protein has a crucial, irreplaceable function in the cell, the maintenance of which provides evolutionary advantages.

Telomerases are distinguished by having a particular catalytic activity from which RNA sequences can be employed as a template in a retro-transcription process. This feature could use reverse transcriptase inhibitors as potential telomerase inhibitors, thus increasing selectivity and specificity.

Topoisomerases are a special class of enzymes that do not require enzymatic cofactors or external energy to carry out their action. However, since its main function focuses on relieving DNA topology, and certain secondary structures,

topoisomerases tend to have an affinity for the recognition of conjugated planar molecules, which can be found in nature.

Various products of natural origin and derivatives are useful pharmacological agents, but the search for new antitumoral and antiviral drugs remains still necessary.

LIST OF ABBREVIATIONS

BSA	=	Bovine serum albumin
CD	=	Circular dichroism
DNA	=	Deoxyribonucleic acid
DNA Pol	=	DNA polymerase
ESI-MS	=	Electro spray ionization mass spectrometry
hTERT	=	Human telomerase reverse transcriptase
PCR	=	Polymerase chain reaction
PHQ	=	Phenanthrenequinone
Pol	=	Polymerase
RNA	=	Ribonucleic acid
SAR	=	Structure-activity relationship
Taq	=	Thermus aquaticus
tRNA	=	Transcriptional RNA

CONSENT FOR PUBLICATION

Not applicable.

CONFLICT OF INTEREST

The authors declare no conflict of interest, financial or otherwise.

ACKNOWLEDGEMENTS

All Financial support from CONICET, UNSL (PROICO 02-2516 UNSL 2016) and ANPCyT is gratefully acknowledged. We thank Organic Area staff of the National University of San Luis for their help and also we appreciate revision of the manuscript by staff from the "Instituto de Lenguas, Universidad Nacional de San Luis".

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