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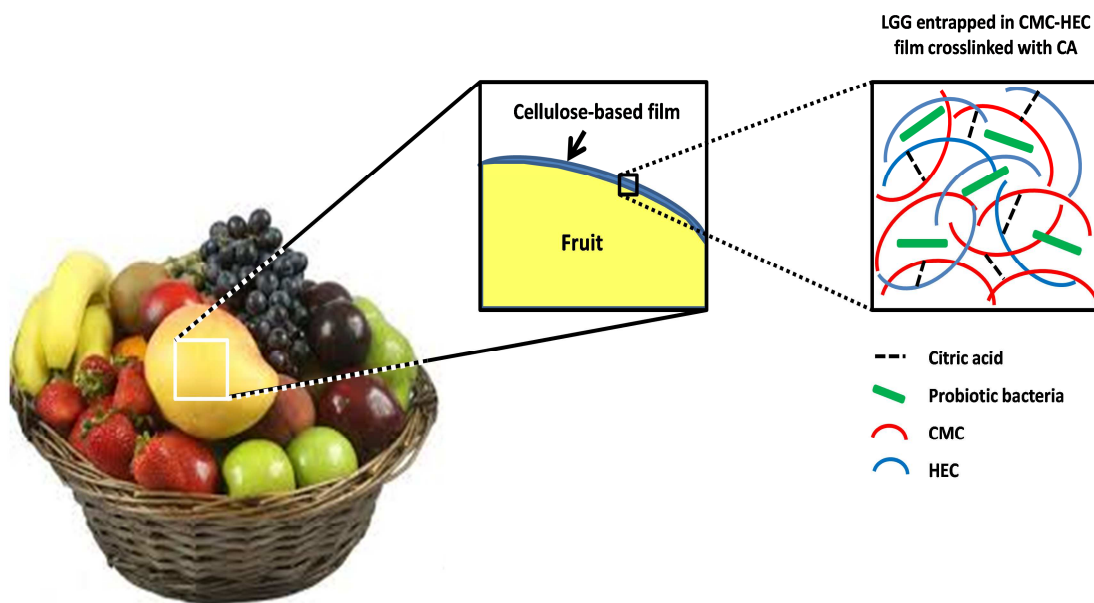
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**Graphical abstract****“Cellulose-Based Edible Films for Probiotic Entrapment”**

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# Cellulose-Based Edible Films for Probiotic Entrapment

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## Abstract:

Encapsulation with edible films is a promising approach that may solve the disadvantages associated with the use of bioactive compounds as food additives. This is particularly relevant in the case of probiotics, since their stability in food matrices and in the gastrointestinal tract may be rather poor. Therefore, new cellulose-based edible films have been successfully developed and characterized. Sodium carboxymethyl cellulose (CMC) and hydroxyethyl cellulose (HEC) were used for the film preparation and cross-linked with citric acid (CA) under reasonably mild conditions. Model probiotic bacteria (*Lactobacillus rhamnosus* GG) were incorporated in the films either during the film formation and casting or after the film synthesis, via bacteria diffusion and adsorption. The later approach could efficiently entrap and preserve viable bacteria. The mechanical properties and swelling ability could be tuned by varying the HEC/CMC ratio and the amount of CA. Moreover, the surface area and total pore volume of the films considerably decreased after cross-linking. Overall, these novel films are regarded as promising inexpensive and friendly matrices for food protection and packaging applications.

**Keywords:** Edible films; citric acid; carboxymethyl cellulose; hydroxyethyl cellulose; probiotic bacteria

## 34 1. Introduction

35 Nowadays, the use of edible coatings to improve the quality of food products is an  
36 interesting approach employed routinely. Typically, these systems work as physical  
37 barriers for gases, moisture and other compounds such as aromas and lipids (Biquet &  
38 Labuza, 1988; Cuq, Gontard, & Guilbert, 1995; Kester & Fennema, 1986; Quiros-  
39 Saucedo, Ayala-Zavala, Olivas, & Gonzalez-Aguilar, 2014). Apart from the barrier  
40 functionality, some innovative applications consider their use to entrap bioactive  
41 compounds of interest combined with later controlled release of the cargo at a specific  
42 target. These functionalized systems are interesting not only to extend shelf life and  
43 reduce the risk of pathogen growth on food products, but also to provide a functional  
44 product with health benefits to the consumer (Dhall, 2013; Pothakamury &  
45 Barbosa-Canovas, 1995). Antioxidants, nutraceuticals, antimicrobials, flavors and  
46 probiotics are among the most used bioactive agents (Espitia, Batista, Azeredo, & Otoni,  
47 2016; Muranyi, 2013). Among them, probiotic bacteria are particularly relevant since  
48 these living microorganisms are believed to provide beneficial health effects to  
49 the host by replenishing the natural gastrointestinal microbiota (Espitia, et al.,  
50 2016; Holzapfel, Haberer, Snel, Schillinger, & Huis in't Veld, 1998; Pamer, 2016;  
51 Salminen, Ouwehand, Benno, & Lee, 1999). Entrapment of probiotics into  
52 polymeric films can protect them from premature degradation and enhance their  
53 controlled release (Corona-Hernandez, et al., 2013; Haffner, Diab, & Pasc, 2016; P.  
54 Singh, et al., 2017). The practicability of coatings based on edible polysaccharides (e.g.  
55 alginate or gellan films) to carry and support viable bifidobacteria in food products was  
56 firstly demonstrated by Tapia et al. in fresh-cut fruits (Tapia, et al., 2007). Later, in a  
57 related study, *L. acidophilus* probiotics have been immobilized in an alginate-coated  
58 surface for strawberries (Moayednia, et al., 2010). Such immobilization of *L.*  
59 *acidophilus* in the alginate-based film effectively protected the bacteria against the low  
60 temperature storage, with no significant change in the viability count (Moayednia, et al.,  
61 2010).

62 Among the systems developed so far, a high focus has been given to biopolymers such  
63 as chitosan, gums, cellulose, pectins and seaweed extracts (Baldwin, Hagenmaier, &  
64 Bai, 2012; Bourtoom, 2008; Espitia, Du, Avena-Bustillos, Soares, & McHugh, 2014).  
65 Cellulose is particularly special since it is a highly available, renewable and  
66 biocompatible raw material and therefore it emerges as a future key resource for a  
67 sustainable planet (Klemm, Heublein, Fink, & Bohn, 2005; P.; Singh, et al., 2015).

68 The film production often requires the use of harsh chemicals and cross-linking agents,  
69 such as formaldehyde-based compounds, that may limit their potential use in food  
70 applications. An interesting alternative considers the use of citric acid, CA, as a natural  
71 cross-linker. Thus, the resulting materials are acceptable in the food and medical fields,  
72 due to their excellent biocompatibility and hydrophilicity (Christian Demitri, et al.,  
73 2008; Raucci, et al., 2015). Even if not well documented in the available literature,  
74 cellulose systems chemically cross-linked by CA are expected to form three-  
75 dimensional networks poorly soluble in water or in biological fluids (Coma, Sebti,  
76 Pardon, Pichavant, & Deschamps, 2003; Glusker, 1980; Wang & Chen, 2005; Xie, Liu,  
77 & Cui, 2006; Yang & Wang, 1998). These hydrogels have also been suggested as new  
78 superabsorbent systems potentially interesting in personal care (i.e. diapers and napkins)  
79 or in agriculture and horticulture for strategical water management (C. Demitri, Scalera,  
80 Madaghiele, Sannino, & Maffezzoli, 2013; Sannino, Demitri, & Madaghiele, 2009). To  
81 the best of our knowledge, the potential use of such systems to entrap probiotic bacteria  
82 has never been explored before. Therefore, in the present work, two cellulose  
83 derivatives, sodium carboxymethylcellulose (CMC) and hydroxyethylcellulose (HEC)  
84 were used for film preparation, cross-linked with CA and loaded with probiotic bacteria.  
85 The film formation, structure and mechanical properties were characterized, as well as  
86 the viability of the entrapped bacteria under different conditions.

87

## 88 **2. Experimental**

### 89 **2.1. Materials and Methods**

90 The cellulose derivatives, CMC (Mw of ca. 250 kDa with a degree of substitution  
91 ca. 0.80-0.85) and HEC (Mw of ca. 720 kDa with a molecular substitution of 2.5 mol  
92 per mol of cellulose), were purchased from VWR international (Belgium) and Sigma  
93 Aldrich (USA), respectively, and used as received. The CA and HCl were obtained from  
94 Sigma Aldrich (USA). The chemical structures of CA, CMC and HEC can be found in  
95 supporting material section, Figure S1. *Lactobacillus rhamnosus* GG LMG 18243  
96 (LGG) was bought from the Belgian Coordinated Collection of Microorganisms. The  
97 MRS broth pH 6.4 and MRS agar pH 5.7 were obtained from VWR International. The  
98 PBS buffer was prepared in the laboratory using disodium hydrogen phosphate, sodium  
99 chloride, potassium chloride and potassium dihydrogen phosphate, all with analytical  
100 grade from Sigma Aldrich (USA). The Live/Dead® BacLight™ Bacterial Viability Kit

101 L7012 was purchased from Thermofisher Scientific, USA. Milli-Q water (18.2 MΩ.cm-  
102 1 at 25 °C, MQ) was used for the preparation of all samples.

103

## 104 **2.2. Film formation**

105 The films were obtained by mixing different amounts of CMC, HEC and CA in  
106 Milli-Q water until full dissolution was achieved. The total polymer concentration was  
107 set to 2 wt% while the ratio between the cellulose derivatives was systematically varied.  
108 The CA (cross-linking agent) amount ranged from 5 to 10 wt% based on the total  
109 polymer concentration. Lower amounts of CA were also tested but the obtained films  
110 were too fragile for handling. After obtaining a clear solution, 15 g of it were poured  
111 into Petri dishes and allowed to cure in an oven at 50 °C for 15 h. in such conditions, the  
112 cross-linking reaction between the cellulose derivatives and CA occurs. The film  
113 thickness depends on the amount of CA and presents an average value of ca. 0.12 mm.  
114 It should be noted that this is a fairly mild procedure in comparison with the standard  
115 cross-linking approach reported elsewhere (i.e. 80 °C for 24 h) (Christian Demitri, et al.,  
116 2008).

117

## 118 **2.3. *Lactobacillus rhamnosus* GG culture and entrapment in the films**

119 All culture media and buffers were autoclaved for sterilization at 121 °C for 15  
120 min. Freeze-dried cells of *Lactobacillus rhamnosus* GG were rehydrated in 5 mL MRS  
121 broth and incubated at 37 °C for 40 h in a CO<sub>2</sub> incubator. Cell growth was followed for  
122 two days and, in order to evaluate the purity of the bacteria, one inoculation loop with  
123 bacteria was streaked on a MRS agar plate while a plate with pure sterilized culture was  
124 treated as control. The optical density was measured with a UV/vis spectrophotometer.

125 The incorporation of bacteria into the films was performed in two ways: 1)  
126 soaking the cross-linked films in a bacteria medium for 30 min at room temperature and  
127 removing the excess of medium to allow a moderate drying at 37 °C for 30 min. 2)  
128 mixing the bacteria inoculum in the CMC/HEC/CA mixture before casting and curing.  
129 While in the former approach the bacteria are expected to diffuse and adsorb into the  
130 film during its swelling and solvent uptake, in the latter the bacteria are expected to be  
131 entrapped in the cellulosic polymer matrix already before casting and curing. The  
132 results from both methodologies will be discussed later.

133

## 134 **2.4. Evaluation of the Viability of *Lactobacillus rhamnosus* GG.**

135 To estimate the viable counts and matrix effect, the encapsulated bacteria were  
136 released by re-suspending ca. 0.01 g of film either in a PBS buffer solution (pH 7.4) or  
137 in HCl aqueous solution (pH 2.4) at 37 °C for 30 min, under stirring. Sequential  
138 dilutions were performed following the Miles and Misra approach to count the number  
139 of viable bacteria (Hedges, 2002; Miles, Misra, & Irwin, 1938). All different dilutions  
140 were plated in triplicates and kept in a CO<sub>2</sub> incubator for 48 h before the bacterial  
141 colony was counted using a colony reader. The experiments were performed in Faster  
142 BH-EN and BHG Class II Microbiological Safety Cabinets.

143

#### 144 **2.5. Rheology**

145 The rheological measurements on the gel-like films were carried out on a HAAKE  
146 MARS III rheometer (Thermo Fisher Scientific, Germany) set with a plate-plate  
147 geometry (35 mm, 0.2 mm gap). A Peltier unit was used to ensure strict temperature  
148 control, which was set at 20.0 ± 0.1 °C. The storage modulus, G', was accessed by  
149 performing dynamic oscillatory experiments from 10 to 0.01 Hz, at a constant stress of  
150 5 Pa.

151

#### 152 **2.6. Fourier transform infrared (FTIR) spectroscopy**

153 The cross-linking between the cellulose derivatives and CA was investigated by  
154 FTIR at 25 °C with an ATR-FTIR spectrophotometer Thermo Nicolet IR300 (USA),  
155 using a universal ATR sampling accessory. The FTIR spectral analysis was performed  
156 between 400–4000 cm<sup>-1</sup>. A total of 256 scans were performed in the transmission mode  
157 for each spectrum, with a resolution of 1 cm<sup>-1</sup>.

158

#### 159 **2.7. Thermal gravimetric analysis (TGA)**

160 TGA was performed on a TG 209 F3Tarsus thermogravimetric analyzer (Netzsch  
161 Instruments). The films (ca. 8-10 mg) were weighed in alumina pans and heated from  
162 30 to 650 °C with a constant heating rate of 10 °C min<sup>-1</sup> under N<sub>2</sub> atmosphere.

163

#### 164 **2.8. Water uptake**

165 The equilibrium swelling measurements for all chemical films were carried out in  
166 Milli-Q water or in aqueous media at different pHs. The percentage of swelling (S) was  
167 estimated by weighing a ca. 1 g square-shaped specimen before and after its immersion

168 in the aqueous media for about 24 h. The films were observed to reach their swelling  
169 equilibrium within this period.  $S$  is defined as follows:

170

$$S = \frac{(W_s - W_d)}{W_d} \times 100$$

171

172 where  $W_s$  is the weight of the swollen film and  $W_d$  is the weight of the dried sample.

173

## 174 **2.9. Scanning Electron Microscopy (SEM)**

175 A VEGA3 SBH from TESCAN scanning electron microscope, equipped with a  
176 selected energy dispersive X-ray microanalyser, was used to observe the superficial  
177 morphology of the cross-linked dried films. Briefly, the freeze-dried samples were  
178 deposited directly over the carbon tape on the support and sputtered with an  
179 approximately 6 nm thin Au/Pd film by cathodic pulverization using a SPI Module  
180 Sputter Coater before SEM analysis. The accelerating voltage ranged from 5 to 15 kV.

181

## 182 **2.10. Mechanical analysis**

183 A Texture Analyzer TA.XT Plus (Stable Micro Systems Ltd., Surrey, UK) was  
184 used to access the mechanical properties of the films following the ASTM D882-12  
185 standard. Tensile grips with 35 mm were used to hold the specimens which consisted of  
186 40 mm x 40 mm films. Up to four repetitions were made for each film formulation  
187 using a grip speed of 1.0 mm/s.

188

## 189 **2.11. Surface area and porosimetry**

190 The surface area and the total pore volume were determined by  $N_2$  gas adsorption  
191 isotherms using an ASAP 2000, from Micromeritics and considering the BET  
192 (Brunauer, Emmett and Teller) model for evaluation (Brunauer, Emmett, & Teller, 1938;  
193 Sing, 2001).

194

## 195 **2.12. Statistical analysis**

196 Most of the experiments were performed in triplicate and data was subjected to  
197 one-way analysis of variance (ANOVA). Multiple comparisons were performed by LSD  
198 test. Statistical significance was set at  $p < 0.05$  using SPSS (SPSS Inc, USA).

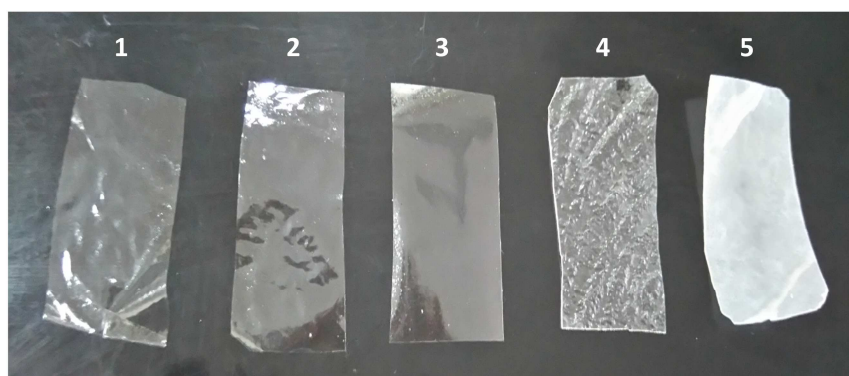
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### 3. Results and discussion

#### 3.1. Characterization of the cellulose-based films

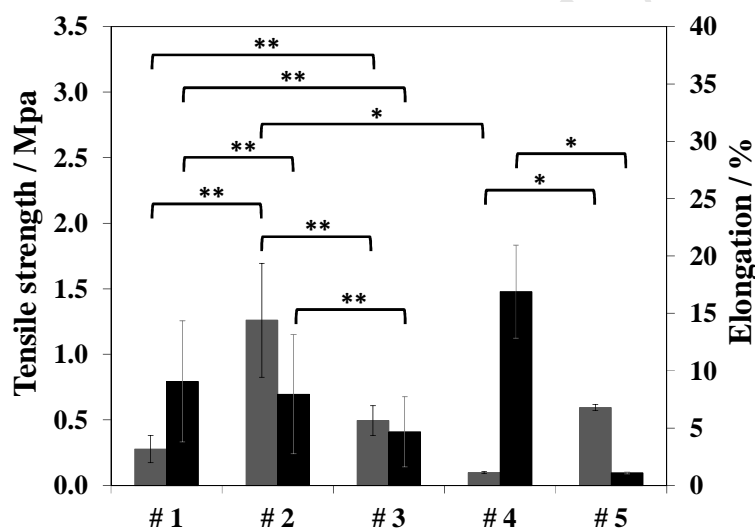
Photographs of the films formed following the procedure described in the experimental section are presented in Figure 1. The films made from HEC (#1) and CMC (#2) alone are reasonably flexible and transparent. The same is observed for the physical mixture of HEC and CMC (#3). On the other hand, the addition of CA followed by cross-linking resulted in less flexible and brittle films. The transparency is also observed to decrease with the amount of CA. While the HEC (#1), CMC (#2) and their physical mixture (#3) dissolve in water, the cross-linked films (#4 and #5) do not.



**Figure 1.** Photograph of cellulose-based film strips after being casted and dried in Petri dishes. 1) 2 wt% HEC; 2) 2 wt% CMC; 3) 1 wt% HEC + 1 wt% CMC; 4) 1 wt% HEC + 1 wt% CMC + 5 wt% CA; 5) 1 wt% HEC + 1 wt% CMC + 10 wt% CA.

The synthesized films were further characterized by FTIR spectroscopy (Figure S2 in supporting material section). The spectra of the films were compared to those of the individual compounds, highlighting the region of interest between  $1100$  and  $1900\text{cm}^{-1}$ . The cross-linking mechanism between polycarboxylic acids and cellulose has been proposed to occur firstly by molecular dehydration of the acid followed by the reaction of esterification (Figure S3 in the supporting material section). Indeed, for the cross-linked films, a new band is observed at  $1722\text{cm}^{-1}$  (dashed line in Figure S2) which is assigned to the stretching mode of the  $\text{C}=\text{O}$  ester bond, expected to appear at higher frequencies than for carboxylic acids. All the remaining bands of CMC and HEC are essentially kept after reaction. Overall, the FTIR data supports the success of the esterification reaction between the cellulose derivatives and CA.

233 The obtained films (cross-linked or not) were subjected to tensile tests in order to  
 234 study their mechanical properties. In Figure 2 it is possible to observe that the CMC  
 235 film presents the higher tensile strength (maximum load at break) while the mixture of  
 236 CMC and HEC cross-linked with 5% CA is the weaker one. On the other hand, a  
 237 significant difference in the tensile strength ( $p < 0.05$ ) is observed when increasing the  
 238 CA content to 10 wt%. These results should not be overrated since the film thickness  
 239 slightly changes with the CA amount and this is expected to influence the mechanical  
 240 properties. It is also interesting to note that the tensile strength and elongation to break  
 241 follow an opposite trend, as reported in related systems (Borges, et al., 2004).  
 242 Increasing the CA content improves the tensile strength but decreases the elongation to  
 243 break (compare films #4 and #5). Overall, the mechanical properties obtained are  
 244 promising even without the use of any plasticizer.



256 **Figure 2.** Elongation to break (black bars) and tensile strength (grey bars) of the  
 257 cellulose-based films. 1) 2 wt% HEC; 2) 2 wt% CMC; 3) 1 wt% HEC + 1 wt% CMC;  
 258 4) 1 wt% HEC + 1 wt% CMC + 5 wt% CA; 5) 1 wt% HEC + 1 wt% CMC + 10 wt%  
 259 CA. The data is presented as means  $\pm$  SD. \*Significance at  $p \leq 0.05$ , \*\* not significant.

261 The BET surface area was also assessed and it is summarized in Table 1. The first  
 262 observation is that the non cross-linked films present a substantial higher BET surface  
 263 area when compared with the cross-linked films. Moreover, increasing the CA amount  
 264 decreases the BET surface area. This decrease is certainly related with the denser matrix  
 265 formation due to the cross-linking reaction. The total pore volume decreased ca. 50%  
 266 when the films were cross-linked with 5 wt% CA. For the films with 10 wt% CA, the

267 pore volume drop is even more remarkable, ca. 95% less in comparison to the non  
268 cross-linked film.

269

270 **Table 1.** N<sub>2</sub> surface area and total pore volume of the novel cellulose-based films.

Film composition	BET surface area (m <sup>2</sup> .g <sup>-1</sup> )	Total pore volume (cm <sup>3</sup> /g)
2 wt% HEC	4.079	0.022069
2 wt% CMC	3.142	0.002636
1 wt% HEC + 1 wt% CMC	5.934	0.005838
1 wt% HEC + 1 wt% CMC + 5 wt% CA	0.782	0.003025
1 wt% HEC + 1 wt% CMC + 10 wt% CA	0.196	0.000311

271

272 The amount of water adsorbed by the different films was measured after 24 h of  
273 equilibration in aqueous media at different pHs (Table 2). Note that all the physical  
274 films, i.e. without chemical cross-linking, dissolved in water after a couple of minutes  
275 and thus their average swelling is not possible to report. It was observed that all the  
276 tested systems, except the cross-linked HEC film, present a rather similar trend with a  
277 low swelling capacity at low pH (ca. pH 2) and two to three-fold swelling increment at  
278 high pH, ca. pH 10. The integrity of the cross-linked films was essentially preserved  
279 regardless of the pH used. All results are statistically different ( $p < 0.05$ ) within the same  
280 pH.

281

282 **Table 2.** Average equilibrium swelling of the cross-linked films in water, at 25 °C at  
283 different pHs. Mean values with the same letter are not significantly different ( $p < 0.05$ ).

284

Sample	Average swelling / %		
	pH 2	pH 6	pH 10
CMC + 5% CA	83 ± 6.2 <sup>a</sup>	236 ± 10.8	273 ± 7.5 <sup>c</sup>
HEC + 5% CA	2578 ± 13.7	2630 ± 8.4 <sup>b</sup>	2650 ± 10.0 <sup>b</sup>
CMC + HEC + 5% CA	171 ± 8.5	280 ± 15.0 <sup>c</sup>	312 ± 11.8
CMC + HEC + 10% CA	59 ± 4.0	86 ± 9.0 <sup>a</sup>	93 ± 2.5 <sup>a</sup>

285

286 The cross-linked HEC film presents the highest swellability, ca. 20 times more than  
287 cellophane (Stamm, 1956). The drawback of such a huge water uptake is that the highly

288 swollen film behaves as a weak gel (rheology data in Figure 3) and thus its handling is  
289 rather difficult in contrast to other formulations. Comparing the films containing only  
290 CMC and CMC/HEC, both cross-linked with 5% CA, it is interesting to observe an  
291 increase in swelling when HEC is present. The data suggests that CMC is more reactive  
292 under the present conditions, leading to a better cross-linking when compared with the  
293 HEC film. Previous work reported a different trend where the higher HEC reactivity, in  
294 comparison with CMC, was attributed to lower sterical hindrance (Christian Demitri, et  
295 al., 2008). On the other hand, since the film forming solution has a pH ca. 4, CMC is  
296 charged and swelling may induce some structural changes/rearrangements thus allowing  
297 a more efficient cross-linking. Another interesting observation is related to the lower  
298 swelling of the films when the CA concentration increases. Most likely, the increment  
299 in CA concentration leads to an increase in the cross-linking density and network  
300 rigidity, thus reducing the swelling ability. This observation is also in agreement with  
301 the observed low BET surface area and total pore size of the films cross-linked with 10  
302 wt% CA.

303 If not protected, the growth and survival of probiotics during their transit in the  
304 gastrointestinal tract may be seriously compromised. Moreover, the inherent physical  
305 and chemical properties of food carriers, such as the low pH found in, for instance, fruit  
306 juices, salads and condiments, may represent an additional problem for probiotic  
307 survival (Rodgers, 2007). The data suggests that these cellulose-based films, cross-  
308 linked with CA, may be a viable alternative for entrapping viable probiotics and protect  
309 them from any unfavorable food matrix properties or even from the harsh  
310 gastrointestinal conditions.

311 As previously mentioned, after soaking and swelling the films for 24 h in aqueous  
312 media they behave as jelly-like materials. The rheological mechanical spectra are  
313 represented in Figure 4.

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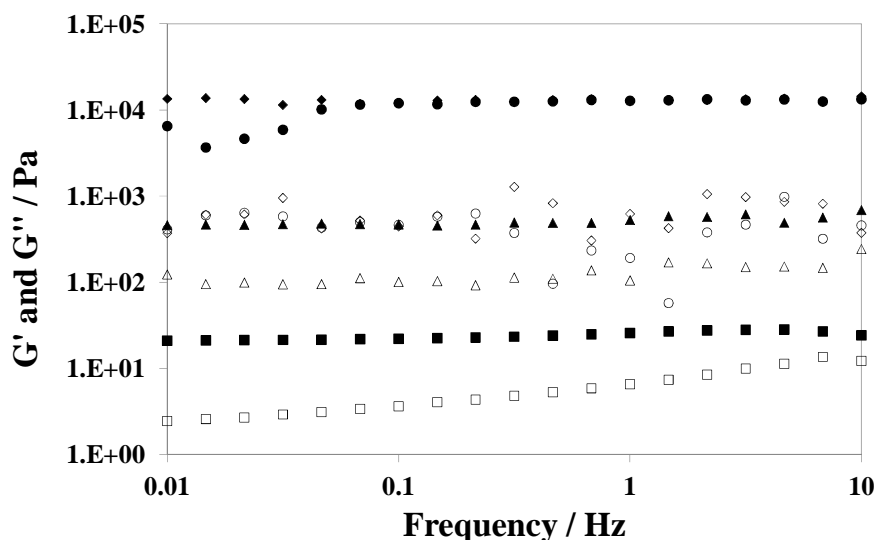
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332

333 **Figure 3.** Mechanical spectra (frequency sweep) of the films after 24h swelling in Milli-  
 334 Q water (neutral pH) at 25 °C.  $G'$  (filled symbols) and  $G''$  (empty symbols) of 2 wt%  
 335 CMC + 5 wt% CA (diamond); 2 wt% HEC + 5 wt% CA (square); 1 wt% HEC + 1 wt%  
 336 CMC + 5 wt% CA (triangle) and 1 wt% HEC + 1 wt% CMC + 10 wt% CA (circles).

337

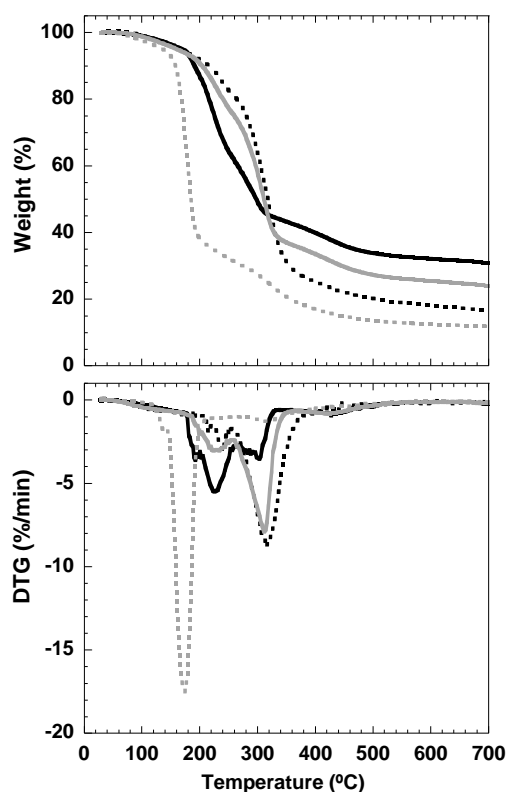
338 The cross-linked CMC films show a reasonably high storage modulus,  $G'$ , almost  
 339 three orders of magnitude higher than the cross-linked HEC films, which behave as a  
 340 weak gel-like material. Nevertheless, as observed in Table 2, such high elasticity  
 341 compromises the extension of swelling. HEC films have a lower cross-linking density  
 342 and thus swell up to one order magnitude more than the CMC films. The combination  
 343 of the two polymers results in films with interesting rheological properties, where  $G'$   
 344 stands in between the values of the films composed by the individual polymers and the  
 345 swelling degree is higher than the film of CMC alone. The role of CA cannot be  
 346 neglected; an increase in CA concentration from 5 to 10 wt% leads to a significant  
 347 increase in rigidity of the polymer network which results in the highest  $G'$  observed and  
 348 very poor swelling ability. Therefore, the mechanical, rheological and swelling  
 349 properties can be optimized by properly tuning the CMC/HEC/CA ratios.

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Figure 4 shows the thermal decomposition of the different films as well as the  
 corresponding derivative thermogravimetric curves (DTG).

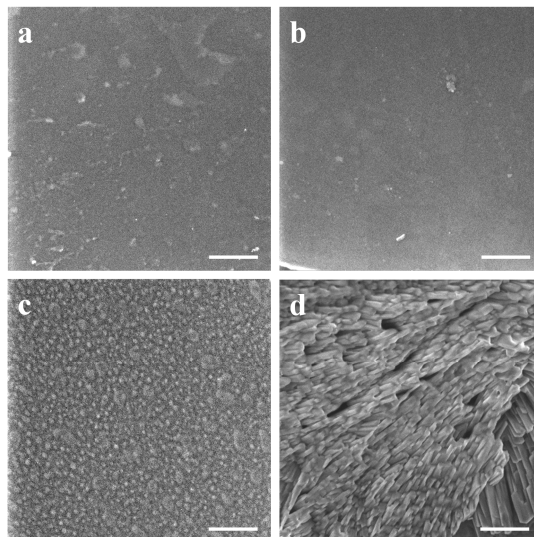


**Figure 4.** TGA (top) and corresponding DTG (bottom) of cellulose-based cross-linked films. 2 wt% CMC + 5 wt% CA (full black line); 2 wt% HEC + 5 wt% CA (dashed black line); 1 wt% HEC + 1 wt% CMC + 5 wt% CA (full grey line); 1 wt% HEC + 1 wt% CMC + 10 wt% CA (dashed grey line).

The CMC-based films generally present a lower thermal stability when compared with the formulations using HEC. The thermal degradation,  $T_{dm}$ , of the native polymers HEC and CMC occurs at ca. 310 °C and 280 °C, respectively (data not shown) but after the cross-linking reaction with 5 wt% CA,  $T_{dm}$  is shifted to ca. 315 °C and 230 °C, respectively. That is, while the HEC films slightly improve their stability, the CMC based films display a substantial reduction. Moreover, it is interesting to note that the film containing HEC, CMC and 5 wt% CA presents a  $T_{dm}$  similar to HEC cross-linked with CA but with an intermediate final mass loss of the individual cross-linked polymers. The formulation containing 10 wt% CA results in films with decreased thermal stability (i.e.  $T_{dm}$  is ca. 175 °C). This reduction in  $T_{dm}$  can be related to a higher disorder in the structure due to the high concentration of CA in the reaction. Such behaviour has been observed in other systems where a too high cross-linking density may lead to a less organized material and, consequently, to a lower thermal stability (Capanema, et al., 2018). Nevertheless, the TGA curves indicate that the obtained

387 chemically cross-linked cellulose-based films are thermally stable in the temperature  
388 range suitable for entrapment and delivery of probiotic bacteria.

389 The surface morphology of the cross-linked films was characterized using SEM  
390 (Figure 5).



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401  
402 **Figure 5.** SEM images of different cross-linked films: a) 2 wt% CMC + 5 wt% CA; b)  
403 2 wt% HEC + 5 wt% CA; c) 1 wt% HEC + 1 wt% CMC + 5 wt% CA; d) 1 wt% HEC +  
404 1 wt% CMC + 10 wt% CA. The scale bar represents 5  $\mu\text{m}$ .

405  
406 The cross-linked films of HEC or CMC alone (Figures 5a and 5b) present a rather  
407 similar and homogeneous morphology with an apparent low porosity, which contrasts to  
408 the much less dense structure and higher porosity of the non-cross-linked homologues  
409 (data not shown). When CMC and HEC are mixed and cross-linked together, the  
410 porosity and heterogeneity increases (Figure 5c) and a “fibrillar”-like morphology is  
411 obtained for the highest CA concentration used (Figure 5d). Such an aggregated fibrillar  
412 structure is expected to account for the high elasticity of the wet films (Figure 3) and  
413 poor elongation to break (Figure 2). The increased heterogeneity in the film, with  
414 visible fracture points, may be responsible for the observed brittleness and decrease in  
415 thermal stability (Figure 4).

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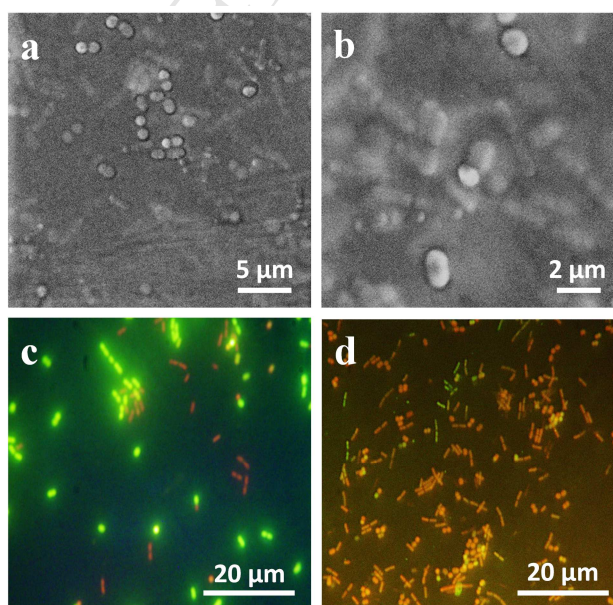
### 417 3.2. Probiotic encapsulation and viability

418 In order to evaluate the ability of the different cellulose-based films to work as  
419 matrices for probiotic entrapment, LGG was selected as model bacteria. As described in  
420 the experimental section, the bacteria were entrapped in the films following two distinct

421 methods. Briefly, in a first approach, LGG was mixed directly with all compounds and  
422 casted in the Petri dishes. A typical SEM example of LGG embedded in the film matrix  
423 is shown in Figure 6a with a zoomed area presented in Figure 6b. However, even if the  
424 curing conditions for the cross-linking reaction are far milder than the standard  
425 procedure reported in literature, the entrapped bacteria are found to have a quite low  
426 viability, as qualitatively evaluated by fluorescence microscopy using a Dead/Alive kit  
427 (Figure 6d). In this test, the viable bacteria appear as green rods while the dead bacteria  
428 are red. As it is clearly visible, the curing conditions (i.e. 50 °C for 24h) are still too  
429 harsh with the majority of the bacteria being found dead; 3 Log CFU/mL (CFU: colony-  
430 forming units) is estimated from the plate counting technique. Although lower  
431 temperatures are beneficial for probiotic viability, the films do not form below 50 °C.

432 It has been speculated that some mechanisms behind the probiotic health effects  
433 may not entirely dependent on the viability of the cells but on other factors such as  
434 probiotic adhesion. It has been argued that probiotic adhesion to host tissues facilitates  
435 the host-microbial interactions, such as the effects of microbes on the immune system of  
436 the host. In fact, some reports suggest that viable and non-viable probiotics are equally  
437 adherent to intestinal mucus (Lahtinen, 2012). Apart from the hypothetical effect of non-  
438 viable probiotics, further studies are planned to improve the synthesis conditions in  
439 order to guarantee a higher probiotic viability in the films.

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452 **Figure 6.** SEM of probiotic bacteria entrapped in CMC-based films (a and b) and the  
453 corresponding fluorescent microscopy image using the Dead/Alive kit (d). The control  
454 is LGG in the polymer matrix before curing (c).



455

456 An alternative approach was tested to endow the films with viable cells. The  
 457 method is based on probiotic adsorption onto the films after soaking them in a LGG  
 458 solution for 30 min at room temperature. The films were subsequently rinsed to remove  
 459 any adsorbed bacteria at the film surface. The film ability to release the trapped bacteria  
 460 into an aqueous media of different pH was evaluated. As can be seen in Figures 7a and  
 461 8b, the majority of the bacteria appear as green, which is a strong qualitative indication  
 462 of high viability even after exposure to the different pH conditions. Note that the images  
 463 in Figure 7 are not as clear as those in Figure 6 since the cross-linked mixture of CMC  
 464 and HEC results in films with poorer transparency, which contributes to blur the  
 465 fluorescence images. Additionally, during the cell diffusion into the film matrix, it is  
 466 expected that bacteria accumulate at the larger pores and structural defects in the films  
 467 thus resulting in a less homogeneous distribution of the bacteria. This can be inferred  
 468 from Figure 7b where a rather large green area is observed. The viability of LGG after  
 469 entrapment in the films was also quantitatively checked by plate counting (Figure 8).

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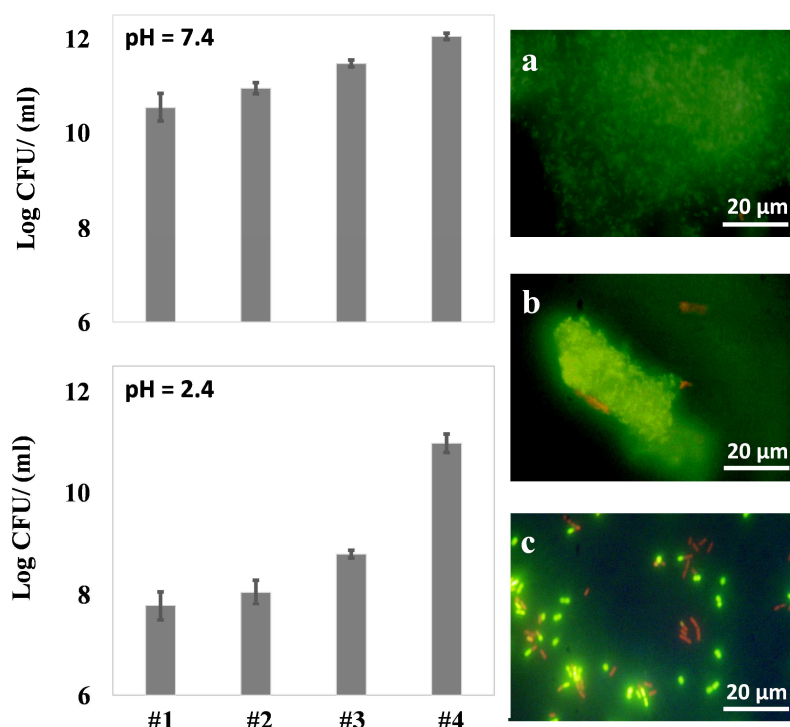
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485 **Figure 7.** Left: Viability counts in Log CFU/mL of the different systems developed  
 486 after the bacteria have been released in PBS, pH 7.4 (top) and in an aqueous solution of  
 487 HCl, pH 2.4 (bottom). #1: 2 wt% CMC + 5 wt% CA; #2: 2 wt% HEC + 1 wt% CMC +  
 488 5 wt% CA; #3: 1 wt% HEC + 1 wt% CMC + 10 wt% CA; #4: naked bacteria.

489 Fluorescence microscopy images of the 1 wt% HEC + 1 wt% CMC + 5 wt% CA film  
490 after bacteria soaking and later exposure to entrapped bacteria of the CMC + HEC +  
491 10% CA films after their exposure to a) pH 7.4 and b) pH 2.4. Image c) refers to naked  
492 bacteria at pH 7.4. The films were stained with the Live and Dead viability kit. The  
493 error bars represent the standard deviation.

494

495 There is a significant difference ( $p < 0.05$ ) in the viability when the CMC+HEC+10 wt%  
496 CA films are compared to all other samples at pH 7.4. However, no significant  
497 difference is observed at pH 2.4 among the films. The larger the content of CA, the  
498 higher is the cell counting. This observation is most likely related to the fact that the  
499 film defects increase with CA (see Figure 5). Note that the HEC film was not selected  
500 for these tests since the mechanical properties of the wet film are rather poor. While at  
501 pH 7.4 the viability for all films tested is above ca. 10 log CFU/mL, it is noticeable that  
502 at 2.4 the viability count drops significantly to ca. 7-8 log CFU/mL. This viability  
503 counting is expected to be underestimated due to the presence of remaining entrapped  
504 bacteria in the film matrix. The fluorescence data suggests that not all the bacteria is  
505 released, which might be due to the poor swelling of the film. On the other hand, at high  
506 pH, film swelling is favoured and the bacteria are more efficiently released thus  
507 resulting in a higher viability counting.

508 In general, viable LGG bacteria can be effectively entrapped in the cross-linked  
509 cellulose based films and their release is strongly dependent on the film composition  
510 and pH of the media.

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#### 512 **4. Conclusions**

513 Both CMC and HEC can be successfully cross-linked using CA and the reaction can  
514 be followed by FTIR. The film properties can be controlled by the cross-linker amount.  
515 An increase in CA concentration leads to a decrease in swelling ratio of the films,  
516 increased rigidity and substantial decrease in the surface area. However, higher CA  
517 content also induces the formation of fibrillar-like aggregates and structural defects in  
518 the films. Viable LGG could be effectively entrapped in the films after soaking them in  
519 bacteria medium. The direct mixing of bacteria with the film components followed by  
520 casting and curing was found to be very harsh for LGG. Overall, the films developed  
521 proved to be promising matrices for bacteria entrapment in, for instance, food  
522 applications where the mechanical, swelling and release properties can be tuned by

523 HEC/CMC ratio and amount of cross-linker. The systems developed represent  
524 promising advances in the search for new applications of edible cellulose based films  
525 and coatings as carriers of diverse probiotics and open new possibilities for the  
526 development of novel food probiotic products. Additionally, these films may inspire the  
527 formation of related systems, such as micro-beads as carriers for probiotic bacteria and  
528 delivery in the gastrointestinal tract. The use of plasticizers is also expected to improve  
529 the mechanical performance of the films and this will be tried in the future.

530

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**Research Highlights**

- Cellulose-based films cross-linked with citric acid were developed.
- Novel edible coatings for probiotic bacteria entrapment and delivery are proposed.
- Suitable thermal, swelling and mechanical properties identified.
- Probiotic bacteria successfully entrapped into the films with acceptable viability.