

NEW MULTIFUNCTIONAL DIVALENT METAL-COORDINATED SULFOPHOSPHONATES: STRUCTURES AND PROTON CONDUCTIVITIES

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Abstract: Metal phosphonate-based coordination polymers are structurally versatile multifunctional compounds that may contain a number of acidic groups such as P-OH; SO₃H, COOH, N⁺-H...These features result in formation of extended H-bond networks and confer proton conducting properties.

In this work, the crystal structures resulting from the combination of the amino-sulfophosphonate ligand (H₂O₃PCH₂)₂-N-(CH₂)₂-SO₃H with different divalent metal ion, are reported. Optimal synthesis conditions were implemented by microwave methodology and high through-put screening. For cupper derivatives, single-crystal data were employed. While for Mn(II) derivative the crystal structure was solved *ab initio* from synchrotron X-ray powder diffraction data. The arrangement of the sulfonic groups determines a wide variety of metal-ligand coordination modes and the creation of efficient hydrogen bonding networks for proton transport, as has been observed for other lanthanide derivatives¹. For copper derivatives the presence of an auxiliary ligand (1,10-phenanthroline, 2,2'-bipyridine) were required to obtain a crystalline compound. As a consequence of this structural variability, different H bond networks can be generated leading to a wide range of proton conductivity values.

Key words: coordination polymers, sulfophosphonates, proton-conductivity

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