1	Understanding the pharmacokinetics of synthetic cathinones:						
2	evaluation of the blood-brain barrier permeability of 13 related						
3	compounds in rats.						
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25 Abstract

26 Synthetic cathinones are the second most commonly seized new psychoactive 27 substance family in Europe. These compounds have been related to several intoxication 28 cases, including fatalities. Although the pharmacological effects, metabolism and 29 pharmacokinetics of cathinones have been studied, there is little information about the 30 permeability of these compounds through the blood-brain barrier (BBB). This is an 31 important parameter to understand the behaviour and potency of cathinones. In this work, 32 13 selected cathinones have been analysed in telencephalon tissue from Sprague-Dawley 33 rats intraperitoneally dosed at 3 mg/kg. Our results revealed a direct relationship between 34 compound polarity and BBB permeability, with higher permeability for the more polar 35 cathinones. The chemical moieties present in the cathinone had an important impact on 36 the BBB permeability, with lengthening of the α -alkyl chain or functionalization of the 37 aromatic ring with alkyl moieties resulting in lower concentration in telencephalon tissue. 38 Our data suggest that transport of cathinones is a carrier-mediated process, similar to 39 cocaine transport across the BBB.

40

41 Keywords Blood-brain barrier, new psychoactive substances, synthetic cathinones,
42 pharmacokinetics, pharmacology, toxicological analysis.

44 Introduction

45 The consumption of synthetic cathinones represents an important public health problem, according to the most recent report from the United Nations Office on Drugs 46 and Crime¹, which illustrates that this new psychoactive substance (NPS) family is one 47 of the most commonly seized worldwide, together with synthetic cannabinoids ¹. Most of 48 49 the cathinone seizures are powder, together with pills and similar products. These 50 compounds have also been found as adulterants in "classical" illegal drugs such as cocaine, illustrating that their prevalence of consumption could be underestimated ^{2,3}. In 51 52 addition to the data obtained by seizure analysis, the public health problem related to 53 cathinones is also illustrated by numerous intoxication cases related to these substances, and even some fatalities 4-6. The synthetic cathinones prevalence can also be illustrated 54 by analytical data obtained from wastewater analysis, illustrating that these compounds 55 are being consumed worldwide ^{7,8}. 56

57 It is almost impossible to ban all the cathinone derivatives existing nowadays due to 58 the continuous change in structure of new compounds appearing on the market. Besides, 59 new compounds that could replace banned ones surface in mere weeks, in a similar way to what occurs with synthetic cannabinoids ⁹. To face this public health problem, the 60 61 scientific community must be able to provide information about novel compounds, their 62 chemical, pharmacological and toxicological properties. Thus, a notable number of 63 papers have been published, as illustrated by the reviews available in literature about the metabolism of these substances ^{10–12}, the associated pharmacological behavior ^{13,14}, 64 toxicology 5,15 , and even their neurotoxicity 16 . 65

66 An important pharmacological issue to highlight is how cathinones affect endogenous 67 compounds, producing a psychoactive effect. Several studies have demonstrated that 68 cathinones act as non-selective monoamine uptake inhibitors, increasing the levels of

dopamine and serotonin 17,18 , producing effects similar to cocaine 19,20 . Thus, the potency 69 of cathinones and other NPS may be studied using *in vitro* approaches ^{21–23}, in a similar 70 way to synthetic cannabinoids ²⁴. Although *in vitro* studies provide valuable information 71 72 about the *intrinsic* potency of a compound, the *in vivo* effect must be determined by the 73 extent to which a compound reaches its site of action. One of the key barriers in this 74 context is the blood-brain barrier (BBB), modulating the exchange of compounds between the brain and the blood ²⁵. The BBB is a complex system that presents different 75 76 "entry routes" that can be used by drugs or hormones²⁵, such as passive diffusion (usually used by non-polar compounds such as steroids) and carrier-mediated influx ²⁵ (used by 77 some psychoactive substances such as cocaine ²⁶), whereby a specific transporter helps 78 79 the compound to cross the BBB and reach the brain. To complement the in vitro data and 80 better understand the pharmacokinetics (and in vivo potency) of cathinones, it is therefore essential to generate accurate data on the BBB permeability of these compounds ²⁶. 81

82 This work is the first to quantify an extensive series of cathinones in brain samples 83 from rats intraperitoneally injected with these compounds, with the objective of relating 84 the permeability through the BBB with their structure. To this aim, we have developed and validated ^{27–30} advanced analytical methodology based on ultra-high performance 85 86 liquid chromatography coupled to tandem mass spectrometry (UHPLC-MS/MS) for the 87 determination of 13 cathinones (Figure 1) in Sprague-Dawley rats' telencephalon tissue. 88 UHPLC coupled to mass spectrometry (MS) plays an essential role in cathinone analysis, 89 using both high-resolution MS (HRMS) and low-resolution MS/MS. Thus, most studies about the identification of novel cathinones ^{31–33} and the elucidation of their metabolites 90 ^{34–36} have utilized UHPLC-HRMS, taking profit of the full-spectrum acquisition and high 91 92 mass accuracy provided by this technique. UHPLC-MS/MS is the preferred technique for 93 the accurate and sensitive determination of a predetermined list of compounds, and has 94 often been used for studying the pharmacokinetics and pharmacodynamics of cathinones
 95 ^{20,37-40}.

96 As shown in Figure 1, taking pentedrone as a template, the cathinones investigated in 97 this study differ in the amine functionalization, aromatic ring substitutions and/or length 98 of the alkyl chain of the alpha carbon atom. The studied compounds were thoroughly 99 selected in order to cover most of the possible combinations of moiety changes usually 100 appearing in cathinones. Using a human in vitro BBB permeability model, Simmler et al. 101 readily assessed the transendothelial transport of a series of cathinones ¹⁷, whereas the 102 permeability of three cathinones with different alkyl chain length (methylone, butylone and pentylone) was evaluated by Grecco et al. using Sprague-Dawley rats ⁴¹. Our work 103 104 further elaborates on this by testing in vivo an extensive set of well-chosen cathinones 105 with different changes in the moieties of the molecule.

106

107 Materials and methods

108 **Reagents and chemicals**

109 Research chemicals containing cathinones were provided by Energy Control (ABD 110 Foundation, Barcelona, Spain). All the compounds were characterized and purity-111 tested by UHPLC-HRMS and nuclear magnetic resonance, following the same procedures already reported in literature ^{42,43}. Cathinone stock solutions were prepared 112 113 at approximately 1 mg/mL in methanol (0.01 mg accuracy). HPLC-grade water was 114 obtained by purifying demineralized water using a Milli-Q system from Millipore 115 (Bedford, MA, USA). HPLC-grade methanol, HPLC-grade acetonitrile, HPLC-grade 116 ethanol, formic acid and acetone were purchased from Scharlau (Scharlab, Barcelona, 117 Spain). Physiological saline solution was purchased from Laboratorios ERN 118 (Barcelona, Spain).

119 Animal testing

120 For animal experiments, dose and brain dissection time were selected based on the information available in literature. In this work, a realistic 3 mg/kg dose ⁴⁰ was preferred 121 over the very large doses (around 20 mg/kg) used in other pharmacokinetic studies 39,41 . 122 123 This is relevant because extremely high doses could affect the metabolome, metabolic 124 routes or active transport - among other parameters - that could be involved in the 125 pharmacology and pharmacokinetics of these compounds. It should be noted that, 126 depending on the compound, the doses reported by users are vastly different. For 127 example, in the case of pentedrone, the compound used as reference in this work, 30 to 128 60 mg has been described as a typical dose when using intravenous route (roughly 0.5 to 1.0 mg/kg)⁴⁴. However, for the aim of this work, it seemed reasonable to use similar 129 130 doses for all the tested compounds, in order to facilitate the interpretation of the results 131 and avoid the influence of different cathinone concentrations on the data obtained.

Peters and colleagues ⁴⁰ quantified three cathinones in rat brains by LC-MS/MS, dosing 132 133 the rats at 3 mg/kg for methylone and mephedrone, and at 1 mg/kg for MDPV. 134 Pharmacokinetic data obtained revealed that the highest concentration was achieved 20 135 min after administration. Also another study ⁴⁵, studying the variation of monoamine 136 transporters in brain tissue from rats injected at 1, 3 and 10 mg/kg, reported the highest 137 concentration in brain tissue 20 min after injection. Based on these studies, 3 mg/kg was 138 selected as the dosage to be administered to the rats for all the compounds, and the brain 139 was dissected 20 min post-injection.

Thus, thirty female Sprague-Dawley rats (8 weeks of age), weighing between 280 and 320 g, were purchased from Janvier Labs (Le Genest-Saint-Isle, France). Animals were housed in groups of 3 animals in polypropylene plastic cages under controlled temperature (24 ± 2 °C) and lighting conditions (12h:12h; lights ON at 8 am), with *ad libitum* access to food and water. Before drug administration, animals
were handled and habituated to the experimental room for one week. Experiments
were conducted in accordance with the standard ethical guidelines (European
Communities Directive 86/60-EEC) and approved by the Valencian Region
government ethical committee (*Generalitat Valenciana, Direcció General d'Agricultura, Ganaderia i Pesca*, ref. 2019/VSC/PEA/0048).

150 Two rats were dosed per compound and the brains were pooled in order to avoid 151 possible animal differences. In total, twenty-six rats received intraperitoneal injections 152 of the 13 different cathinones at a dose of 3 mg/kg in 300 µL of physiological saline 153 solution containing 5% ethanol – the latter for increasing the solubility of the 154 synthetic cathinones. Four additional animals were injected with the same volume of 155 the vehicle and used to obtain blank brain tissue samples, to be used to prepare quality 156 control samples (QCs) and matrix-matched calibration curves. After 20 min, rats were 157 anesthetized with CO₂ and decapitated immediately. The brain was dissected 158 (avoiding blood that could contaminate it), and the telencephalon (both cerebral 159 hemispheres) was isolated, quickly frozen in liquid nitrogen and stored at -23 °C until 160 analysis.

161 Sample treatment

Brain tissue samples were homogenized and crushed with dry ice (Praxair, Valencia, Spain) using an electric grinder, followed by a -23 °C overnight storage for CO₂ evaporation. After that, approximately 250 mg were accurately weighted (\pm 0.1 mg) in 1.5 mL polypropylene tubes, and 750 µL of acetonitrile containing 1% of formic acid were added. Samples were extracted for 30 min under agitation using a vortex (Velp Scientifica, Usmate Velate, Italy) at 1200 rpm. After keeping extracts for 30 min at -23 °C, samples were centrifuged at 12000 rpm for 10 min. Finally, the supernatant was diluted with ultrapure water for UHPLC-MS/MS analysis: in the case of samples used for
method validation, the supernatant was 10-fold diluted, whereas for the brain samples
obtained after administration of cathinones, supernatant was 1000-fold diluted.

172 The sample treatment procedure was adapted from literature ¹⁹, with the only difference 173 being the homogenization procedure. In the present study, homogenization was 174 performed using an electric grinder and dry ice, followed by extraction with acetonitrile 175 and 1% formic acid, a freezing step as clean-up and dilution of the supernatant with 176 HPLC-grade water. Sample weight, extraction volumes and dilutions were designed according to information available in literature ⁴⁰, with some modifications to improve 177 178 method sensitivity. For a detailed description on analytical methodology validation and 179 the results obtained, see **Supporting Information**.

180

181 Instrumentation

Samples were analyzed using an Acquity UPLCTM H-Class liquid chromatography system (Waters Corp, Mildford, MA, USA) coupled to a Xevo TQ-S mass spectrometer (Waters Corp, Manchester, UK) equipped with a triple quadrupole mass analyzer, using a Z-Spray electrospray interface (ESI). Further information about the UHPLC-MS/MS instrument, the conditions employed, and its optimization can be found in **Supporting Information**.

188

189 **Results**

190 Cathinone concentrations found in brain displayed wide differences, from 762 ng/g 191 brain tissue for *N*,*N*-dimethylpentylone to 10596 ng/g for N-ethyl-pentylone, the 192 concentrations for most of the remaining compounds ranging between 1000 and 4000 193 ng/g. In all cases, the concentrations were well above the analytical performance, in terms of sensitivity and limits of quantification of our methodology. In addition, reliability of the analytical methodology was supported by analysis of quality control (QC) samples in duplicate, spiked at 1 and 10 ng/g, included in the sample batch. Recoveries between 70 and 120% were obtained, confirming the correct quantification of the cathinones in telencephalon tissue.

The concentrations found in telencephalon samples for all the compounds are shown in **Tables 1-3**. The differences observed in the cathinone levels suggest that the BBB permeability of these compounds is structure-dependent, being associated with their polarity, as discussed further.

203

204 **Discussion**

205 Cathinone penetration through the blood-brain barrier

The main objective of our study was to evaluate the relationship between the structure of cathinones and their BBB permeability, in order to get better acquainted with the pharmacological behaviour of these substances.

Pronounced concentration differences were observed in the telencephalon for different cathinones, ranging from 762 ng/g (*N*,*N*-dimethylpentylone) to 10596 ng/g (*N*-ethylpentylone). This difference is surprising given the very high structural similarity of these two compounds, which only differ in the amine functionalization (dimethyl *vs*. ethyl).

Table 1 shows the concentrations found in telencephalon tissue for cathinones that differ by the functionalization of the amine moiety (*N*). In the two groups (those without aromatic ring substitution and those with a 3,4-methylenedioxy substituent), cathinones with an *N*-methyl (pentedrone and pentylone) moiety were found at a higher concentration than those with a pyrrolidine ring (α -PVP and MDPV). Regarding compounds with an *N*-ethyl group, *N*-ethyl-pentedrone had lower permeability than the N-methyl analogue whereas *N*-ethyl-pentylone had a higher permeability than the *N*methyl analogue. It is also remarkable that the cathinone with a *N*,*N*-dimethyl group (*N*,*N*dimethylpentylone) seemed to have the lowest permeability of the BBB.

222 N-ethyl-pentylone was, by far, the cathinone with the highest concentration in 223 telencephalon tissue. Also known as ephylone or bk-EBDP, this substance is a recently reported cathinone that has been involved in numerous recent intoxication cases ^{46,47} 224 including 151 deaths between 2014 and 2018⁴⁸, which raises high concerns regarding the 225 226 toxicity of this compound. The high N-ethyl-pentylone concentration found in 227 telencephalon tissue is in line with a recent study about the pharmacokinetic behavior of this cathinone, which also suggested a high BBB permeability ³⁹, which could explain its 228 229 elevated toxicity.

230 Another common modification seen in cathinone analogs is altering the length of the 231 alkyl chain. In this study, we evaluated three pairs of cathinones that only differed from 232 each other in the length of the alkyl chain. (Table 2): buphedrone and pentedrone, N-233 ethyl-pentedrone and N-ethyl-hexedrone, and butylone and pentylone. In all three cases, 234 lengthening the alkyl chain led to a reduction of the BBB permeability, as shown in Table 235 2. These results are in concordance with data reported in a similar study ⁴¹, where the 236 permeability of methylone, butylone and pentylone through the BBB was evaluated. The 237 reported concentrations in cerebrospinal fluid were around 13 mg/L for butylone and 7 238 mg/L for pentylone after dosing Sprague-Dawley rats at 20 mg/kg. These results are 239 coherent with those of the present study, where around 6,000 and 3,700 ng/g butylone 240 and pentylone, respectively, were found in telencephalon for rats dosed at 3 mg/kg. Based 241 on these data, the increment of the non-polarity of the cathinones due to the increase of 242 the alkyl chains produces a reduction of the BBB permeability. Strangely, the most potent 243 cathinone analogs in terms of dose reported by consumers are those that have a three244 carbon alkyl chain: MDPV, pentylone, α -PVP, pentedrone, etc., with the dose being higher if the length is shortened or increased further in most cases ⁴⁹. This could indicate 245 246 that the mechanisms of toxicity of these compounds are not directly linked to their BBB 247 permeability. As can be seen in the case studies for bk-EBDP intoxications, users 248 frequently report a long duration of action for this compound, which is not so common 249 for other cathinones. Perhaps the duration of effects indicates that bk-EBDP lingers in the 250 body for an unusually long amount of time, and some of the toxicity may stem from this 251 phenomenon.

252 The last typical change in the cathinone structure is functionalization of the aromatic 253 ring. As can be observed in Table 3, the functionalizations studied were the addition of a 254 3,4-methylenedioxy moiety, a methyl group, a 3,4-dimethoxy group, and the addition of 255 an halogen atom (in this case, a fluorine). Three of the four cathinone couples 256 with/without a 3,4-methylenedioxy moiety (buphedrone and butylone, pentedrone and 257 pentylone, and α -PVP and MDPV) presented a reduction of the permeability through the 258 BBB when this moiety was added to the molecule (Table 3), and it could also be related 259 to the increment of the non-polarity of the compound by this modification. The 260 remarkably low brain tissue concentration (860 ng/g) for MDPV is in line with previously reported concentrations for MDPV in rat brain, quantified around 260 ng/g at 30 min 261 262 when dosing a rat at 1 mg/kg 38 . Only for the couple *N*-ethyl-pentedrone and *N*-ethylpentylone, the cathinone with the 3,4-methylenedioxy moiety presented a higher 263 264 concentration in telencephalon tissue. Similar to the results obtained when analyzing the 265 N-functionalization (Table 1), N-ethyl-pentylone produced an unexpected result when 266 compared to the other cathinones. The higher telencephalon concentration of 4-267 fluoropentedrone, when compared with pentedrone, suggests that the presence of a 268 halogen atom may increase BBB permeability, potentially due to an increment of the

269 compound's polarity. The reason for the increment of the permeability observed when 270 adding a methyl group (*N*-ethyl-pentedrone vs *N*-ethyl-4-methylpentedrone) or a 3,4-271 dimethoxy group (α -PVP vs 3,4-dimethoxy- α -PVP) is unclear. It is possible that the 272 presence of these terminal methyl groups allows for easier passing through the BBB. In 273 order to confirm this, more cathinones with these aromatic ring changes should be 274 evaluated.

275 The concentration differences discussed above, when changing the N-276 functionalization, alkyl chain length and aromatic ring substitution, point at a positive 277 correlation between polarity and BBB permeability, as also suggested by others ⁴¹. 278 However, based on an in vitro model using TY09 conditionally immortalized human 279 brain capillary endothelial cells, Simmler and colleagues suggested the opposite: these 280 authors reported that a decrease in polarity of cathinones produces an increment of the 281 permeability of the BBB, with non-polar cathinones presenting a particularly high 282 transendothelial permeability ¹⁷. Although these *in vitro* data apparently contradict our findings and those of the previous literature ^{38,41}, the use of live animals instead of a cell 283 284 culture is a closer representation of the real pharmacokinetic behavior of these compounds 285 in a process as complex as BBB permeability.

286 In fact, the BBB is composed not just of endothelial cells, but also includes associated 287 cell elements such as astrocyte endfeet, pericytes and microglia. There are several important routes of transport across the BBB ²⁵, passive diffusion and the carrier-288 289 mediated influx being the most common ones related to psychoactive substances. The 290 coexistence of both influx processes for cocaine through the BBB has been reported in 291 literature using an *in vivo* model with Swiss mice; here, the carrier mediated influx rate was 3.4 times greater than its passive diffusion rate ²⁶. The same publication indicates that 292 293 MDPV is also a substrate for the cocaine transporter. Based on this evidence and in line

with the data presented in this work as well as in literature ⁴¹, it can be deduced that the penetration of cathinones through the BBB is a carrier-mediated process. An exhaustive study about the solute carrier transporters involved in cathinone transport should be performed in order to confirm this hypothesis.

298 In addition to compound polarity and moieties present in the distinct structures, 299 different physicochemical properties of the compounds such as logP and logD, 300 topological polar surface area, number of rotatable bonds, fraction of sp3 carbons, heavy 301 atom count, among other parameters, can be also be related to BBB permeability. For the 302 compounds studied in this work, topological polar surface area values (obtained from 303 PubChem) were evaluated for 11 compounds. No relationship between this parameter and 304 BBB permeability was found. No additional parameters were found in compound 305 databases, consequently those should be determined experimentally or theoretically to 306 evaluate their contribution to BBB permeability.

307 In summary, the relationship between cathinone structure and their ability to cross the 308 blood-brain barrier was studied in this work. To this aim, telencephalon tissues from 309 Sprague-Dawley rats dosed at 3 mg/kg with 13 different cathinones were analyzed by a 310 validated UHPLC-MS/MS procedure. The results obtained showed that permeability of 311 cathinones is related to their polarity, with better crossing of the BBB with increasing 312 polarity. These findings are in accordance with previously published data using rats ⁴¹ but differ from those obtained using *in vitro* experiments ¹⁷, demonstrating the importance of 313 314 not solely relying on *in vitro* data. Less polar N functionalizations, such as the presence 315 of a pyrrolidine ring, reduced the cathinone transport through the BBB. In a similar way, 316 cathinones with longer alkyl chain were less able to cross the BBB. Non-polar aromatic 317 ring substitutions such as 3,4-methylenedioxy reduced BBB permeability, while the 318 presence of a fluorine atom increased BBB transport. All this data, together with

information available in literature from similar studies ⁴¹ and the BBB transport ²⁶ suggest 319 320 that cathinones cross the BBB through a carrier-mediated process. Additionally, this 321 study shows that studying the pharmacology and pharmacokinetics of cathinones, and 322 NPS as a whole, is crucial for a better understanding of the *in vivo* potency of these 323 compounds, complementing other studies such as dopamine and serotonin uptake 324 inhibition. Our future work will be focused on the study of additional cathinones that will 325 appear on the continuously evolving NPS market, in order to support the role of carrier-326 mediated processes in the BBB passage of cathinones.

327

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339

340 **Competing Interests**

341 The authors declare that they have no competing interests.

342

343 Author contribution

344	D.F-S., J.V.S and M.I. conceived the work. M.B-M. and F.M-G. performed animal
345	experiment and brain dissection. D.F-S. and M.I. performed sample treatment,
346	instrumental analysis and data process. D.F-S., J.V.S, M.B-M., F.M-G, X.C, M.V. and
347	M.I. interpreted and discussed the results. F.H. and F.M-G contributed with new reagents
348	and analytical tools. D.F-S and M.I. wrote the first draft of the manuscript. J.V.S, M.B-
349	M., F.M-G, X.C, M.V., C.S. and F.H. provided useful comments and feedback for the

350 manuscript.

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525 Figure captions



530 Tables

Table 1. Concentration levels found in rat telencephalon tissue for cathinones differing
 on the amine (*N*) functionalization (rats dosed at 3 mg/kg).

Change in the N functionalization			
Compound	Brain tissue conc. (ng/g)	Ratio found/dosed (µg/mg)	Structure
Pentedrone	3,718	1.03	O HZ
α-ΡVΡ	3,054	0.84	
N-Ethyl-pentedrone	1,432	0.40	C H
Pentylone	3,113	0.86	OT T
MDPV	863	0.24	STOR N
N-Ethyl-pentylone	10,596	2.94	
<i>N</i> , <i>N</i> -Dimethylpentylone	762	0.21	STORY -

533

Table 2. Concentration levels found in rat telencephalon tissue for cathinones differing 536 on the alkyl chain length (rats dosed at 3 mg/kg).

	Alkyl chain length		
Compound	Brain tissue conc. (ng/g)	Ratio found/dosed (µg/mg)	Structure
Buphedrone	6,067	1.69	O HZ
Pentedrone	3,718	1.03	o H
N-Ethyl-pentedrone	1,432	0.40	C L L
N-Ethyl-hexedrone	1,160	0.32	
Butylone	4,659	1.29	
Pentylone	3,113	0.86	

Table 3. Concentration levels found in rat telencephalon tissue for cathinones differing540 on the aromatic ring substitutions (rats dosed at 3 mg/kg).

Aromatic ring substitution			
Compound	Brain tissue conc. (ng/g)	Ratio found/dosed (µg/mg)	Structure
Buphedrone	6,067	1.69	O T T T
Butylone	4,659	1.29	C L L
Pentedrone	3,718	1.03	
Pentylone	3,113	0.86	or the second se
4-Fluoropentedrone	4,806	1.34	F C C C C C C C C C C C C C C C C C C C
N-Ethyl-pentedrone	1,432	0.40	O H N N
N-Ethyl-pentylone	10,596	2.94	
<i>N</i> -Ethyl-4-methylpentedrone	2,541	0.71	O H N N
α-PVP	3,054	0.85	
MDPV	863	0.24	
3,4-dimethoxy-α-PVP	2,083	0.58	