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2
3 **Compressed Earth Blocks Stabilised**
4 **with Glass Waste and Fly Ash**
5 **Activated with a Recycled Alkaline**
6 **Cleaning Solution**
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37 **Abstract**

38

39 The study reported in this paper focused on the physical-mechanical properties of compacted
40 earth blocks (CEB) stabilised with a sustainable alkali activated cement, completely
41 produced from wastes and residues, including coal fly ash and glass waste (from the
42 production of ophthalmic lenses) activated with an alkaline solution resulting from the
43 aluminium industry. A common Portuguese silty clay was used as the mineral skeleton of the
44 blocks, which were then evaluated based on the protocols of the UNE 41410 and DIN 18945-
45 47 standards. The results evidenced the effectiveness of the alkaline cementing agent in
46 forming a binding matrix for the soil particles, and the resulting material was used to
47 manufacture the earth-based masonry elements. After a careful optimisation of the
48 sustainable binder, an average compressive strength of 17.23 MPa, in unsaturated conditions,
49 was obtained for the blocks. The newly formed soil-binder structure was very capable to
50 withstand wetting and drying cycles, ice-thaw cycles and erosion. The microstructure of the
51 material was further analysed, using scanning electron microscopy and energy dispersive
52 spectroscopy. The results demonstrated the real possibility of using this type of cement as a
53 viable alternative to traditional soil stabilisation binders used in earth construction.

54

55

56 **Key words:** Alkali activated cements; sustainability; earth construction; soil stabilisation;
57 glass waste

58

59 **1. Introduction**

60

61 For millennia compacted earth has been one of the most used materials around the world,
62 mostly for housing construction, using two techniques known as “rammed earth” and
63 “adobe”. Rammed earth is a thick wall of soil rammed inside a wooden formwork, while
64 adobe is formed by blocks of compressed soil, usually slightly larger than a clay brick, they
65 may or may not include reinforcing fibres. Compacted earth blocks (CEB) are considered an
66 evolution from adobe (Pacheco-Torgal and Jalali, 2012), since the manufacturing follows the
67 same principle, only the compaction process has been improved and, additionally, chemical
68 stabilization is applied, thus improving mechanical properties and durability. The compaction
69 technique has also evolved, from the appearance of the first CINVA-RAM compactor
70 machine, to the current use of hydraulic presses, capable of significantly improving the
71 geometry and material properties of the CEB (Silva et al., 2015).

72

73 The chemical stabilization of soils to manufacture CEB is traditionally done with Portland
74 cement (OPC) or lime. However, in some cases, these traditional cements are not a viable
75 solution due to the financial cost of the material, if high cement contents are required to
76 obtain adequate properties. Therefore, it is justifiable to research alternative binders,
77 preferably capable of producing a more environmentally friendly material. In that context,
78 alternative binders based on industrial wastes or by-products are gaining acceptance, thus
79 becoming ideal for this particular application. From a relatively large pool of possibilities,
80 fly ash (FA) is a clear target, due to its extensive research in the last 20 years, which includes
81 applications as a soil stabilizer and as a precursor in alkaline activation reactions. This residue
82 from the combustion of coal, in thermoelectric powerplants, can act as a soil filler, reducing

83 the settlement of the resulting material, or can also be used together with a source of calcium,
84 taking advantage of its pozzolanic properties (Horpibulsuk et al., 2009; Siddiqua and Barreto,
85 2018).

86

87 Slags are also one of the most used wastes in this type of applications, and especially ground
88 granulated blast furnace slags (GGBS), because of its high content in calcium oxides and
89 hydraulic character, that makes them very similar to OPC. Oti et al. (2009a, 2009b) used
90 mixtures of GGBS and quick lime or hydraulic lime to stabilize a clayey soil, which was then
91 compacted with a hydraulic press to produce CEB. The GGBS/lime binder improved
92 compressive strength and especially the durability, in terms of ice/thaw cycles. Mixtures of
93 GGBS and OPC showed less effective results, which was attributed to the increased cationic
94 exchange capacity between the soil and the lime-based binder. In addition to FA and GGBS,
95 it is also worth mentioning the successful use, as chemical stabilizers for the manufacture of
96 BTC, of ashes from sugarcane bagasse and marble stone cutting muds. El-Mahllawy et al.
97 (2018) and Lima et al. (2012) mixed this type of waste with OPC and lime, in varied
98 percentages, effectively improving properties such as compressive strength and absorption
99 of the CEB.

100

101 Glass wastes (GW) are used, since the 1970s, in the manufacture of baked bricks. Due to its
102 very specific composition, they function as a melting material in the sintering process of clay
103 minerals (Hwang et al., 2006; Zhang et al., 2018). However, the use of glass waste as a
104 chemical stabilizer in the manufacture of BTC is not possible, because it is completely inert
105 if not in contact with a strong alkaline substance or a high acidic medium. Therefore, in
106 geotechnics, they are used mostly as a filler material, as a plasticity modifier or to increase

107 the internal friction angle of clay soil particles (Olufowobi et al., 2014). In order to improve
108 and use glass wastes as cementing materials, methods such as alkaline activation are showing
109 significant potential, mostly due to the capacity of the alkaline activators, mainly alkali metal
110 salts, to dissolve the previously pulverized glass. Several authors (Avila-López et al., 2015;
111 Espinoza and Escalante García, 1970; Lin et al., 2012; Novais et al., 2016; Pascual et al.,
112 2014; Rivera et al., 2018) have successfully used alkaline activated glass waste as a
113 cementitious material in various applications, demonstrating that pulverized glass waste can
114 be an alternative to traditional Portland cements, either when used as the sole precursor or
115 mixed with other aluminosilicate-rich precursors.

116

117 The alkaline activation technique is no stranger to the development of masonry elements
118 made of raw earth. This technique, combined with precursor residues or commercial
119 aluminosilicates, such as metakaolin, have been used to chemically stabilize soils for
120 subsequent compaction and manufacture of BTC. Very interesting results have been obtained
121 (Leitão et al., 2017; Omar Sore et al., 2018; Silva et al., 2015), showing the capacity of the
122 technique to greatly improve the mechanical properties, durability and stability in water of
123 the compacted elements; and even acquire properties comparable to the BTC stabilized with
124 OPC. Furthermore, a step forward was already given towards the implementation of
125 alternative, more environmentally friendly activators, with an optimised incorporated energy,
126 with the purpose of producing masonry with a very low carbon footprint. Recent research
127 (Cristelo et al., 2019; Fernández-Jiménez et al., 2017) has shown that the use of glass waste,
128 combined with precursors like fly ash and aluminium anodizing sludge, activated with a
129 recycled alkaline cleaning solution, can originate a sustainable alkaline cement.

130

131 In the present work, the physical-mechanical characterization of chemically stabilized CEB
132 was carried out, integrating various types of industrial waste into the soil stabilization process
133 and manufacturing of the CEB, with the intention of replacing traditional precursors, such as
134 blast furnace slag or metakaolin, and alkaline activators such as hydroxides or silicates.

135

136

137 **2. Materials and methods**

138

139 *2.1 Materiales*

140

141 The soil was collected in the area of Chaves, a city in the Northern Region of Portugal. The
142 fly ash was generated by the Portuguese thermo-electric powerplant of Pego. According to
143 ASTM C618 (2012), it was classified as a class F. The glass waste powder results from the
144 production of optical lenses at the Portuguese company POLO. The chemical composition of
145 the soil and precursors is presented in Table 1, obtained by x-ray fluorescence with a
146 PHILLIPS PW-1004 spectrometer.

147

148 The recycled cleaning solution acting as the alkaline activator has a high pH, between 12 and
149 14, with an approximate density of 1.3 g/cm³. It previously underwent a homogenization
150 process. Table 2 shows the elementary composition of the cleaning solution, determined by
151 Inductively Coupled Plasma Atomic Emission Spectrometer (ICP-AES), with plasma power
152 of 1.4 kW, plasma gas flow of 15.00 l / min and gas nebulizer flow of 0.85 l/min, with a
153 reading time of 5s.

154

155 Table 1. Chemical composition of the Fly ash, Soil and Waste glass (%w)

Element (oxides)	Fly ash	Soil	Waste glass
SiO ₂	56.11	61.74	58.37
Al ₂ O ₃	21.44	23.27	3.94
Fe ₂ O ₃	8.20	5.68	0.15
CaO	1.31	0.33	6.13
MgO	1.46	1.65	0.49
TiO ₂	1.15	0.08	-
MnO	0.07	0.89	2.73
Na ₂ O	1.12	0.58	8.75
K ₂ O	2.81	5.12	4.66
P ₂ O ₅	0.29	0.09	5.07
LOI	5.05	0.33	1.85

156

157 Table 2. Elemental composition of the cleaning solution

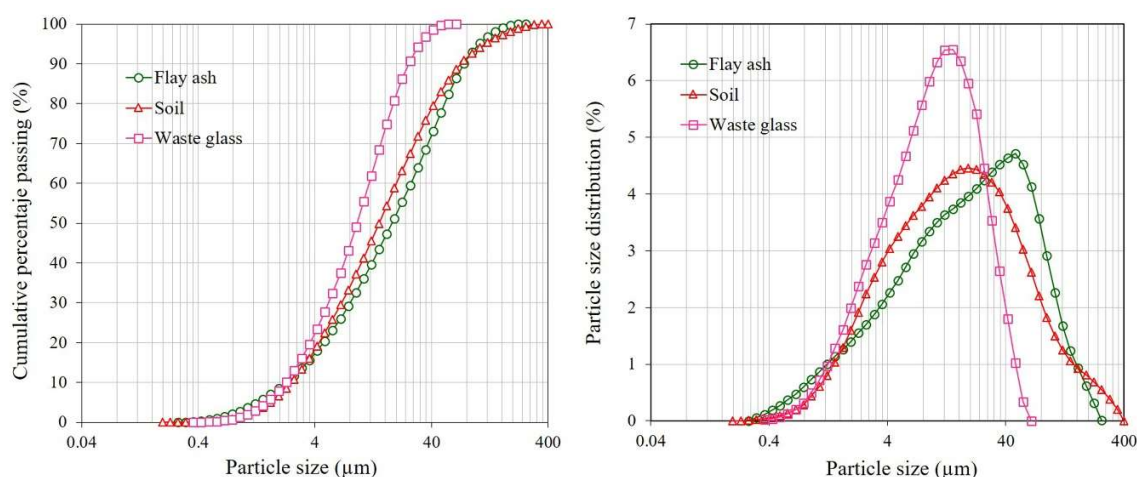
Al ₂ O ₃	Na ₂ O	SO ₃	H ₂ O	pH	^a [OH] ⁻	Density (g/cm ³)
7.14	12.13	1.18	79.5	12-14	5.3	1.30

158 ^a Acid-base reaction with 5 N HCL (Panrea S.A.)

159

160 Figure 1 shows the particle size distribution of the soil and both precursors. The curves were
 161 obtained by laser diffraction, using a Sympatec Helos BF particle size analyser, with a
 162 measuring range between 0.9 mm and 175 mm. The particle size below 45µm was 75.6, 81.3
 163 and 99.2% for fly ash, soil and glass waste, respectively.

164



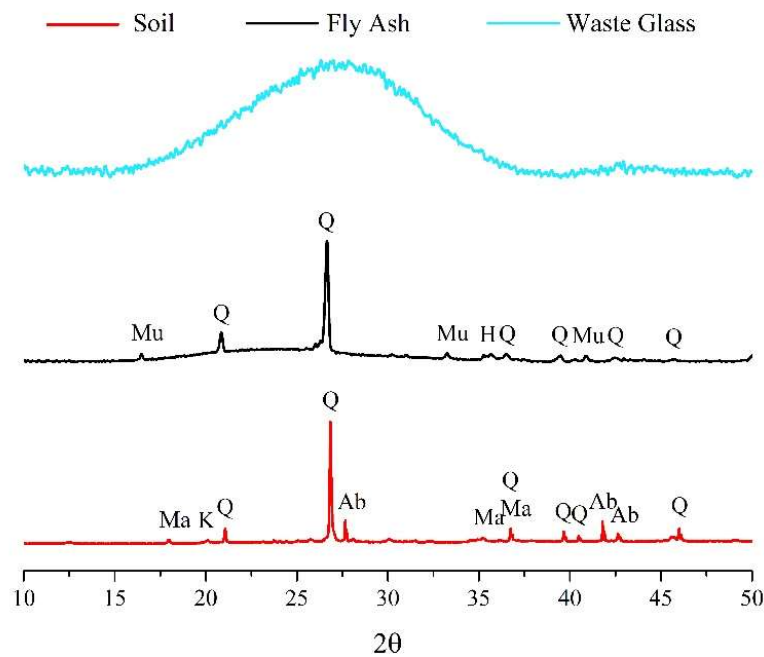
165

166 Figure 1. Cumulative particle size and particle size distribution of the Fly ash, Soil and Waste glass

167

168 The mineralogical composition of the soil and precursors and the soil was determined using
169 X-ray diffraction (XRD), with a PANalytical X'Pert Pro MPD, with CuK α radiation at
170 40Kev and 30mA, equipped with a detector X'Celerator and a secondary monochromator.
171 The diffractograms are presented in Figure 2. The glass residue is a highly amorphous
172 material, as evidenced by the large halo between the 17°(2 θ) to 37°(2 θ) angles. The fly ash
173 is a semi-crystalline material, also with some amorphous content, represented by a smaller
174 halo between angles 18°(2 θ) and 32°(2 θ). It is mineralogy composed essentially by quartz,
175 mullite and hematite. As expected, the soil is a totally crystalline material, composed of
176 several minerals such as quartz, kaolinite magnetite and albite.

177



178

179 Figure 2. XRD Pattern of the original materials . Fly ash: Mu=Mullite, Q= quartz, H=Hematite; Soil:
180 Q=quartz, K=Kaolinite, Ma=Magnetite, Ab= Albite

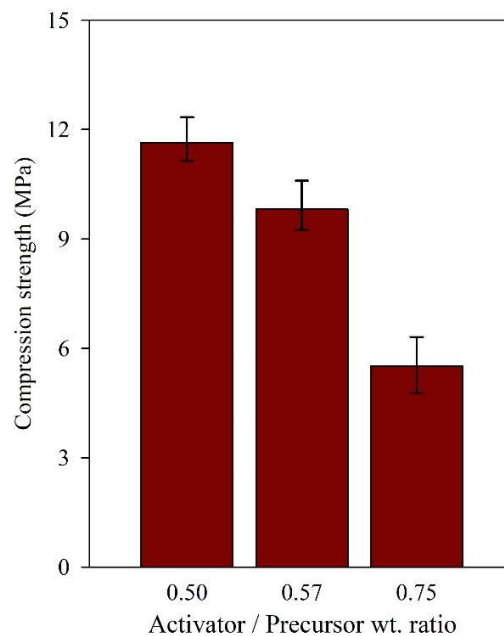
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182 *2.2 Definition of the alkaline cement*

183

184 Prior to the stabilization of the soil, it was necessary to define the composition of the alkaline
185 activated cement (AAC) that would be used as a binder. This AAC was prepared with the
186 described cleaning solution (CS), which was used to activate the glass waste (GW) and the
187 fly ash (FA), acting as the precursor, in a 50/50 weight ratio. The CS was added according
188 with pre-defined activator/precursor weight ratios of 0.50, 0.57 and 0.75. The lowest of these
189 values was defined based on the minimum workability requirements. The pastes were then
190 moulded in 4 cm cubic specimens, which were cured at 50°C and 10% relative humidity in a
191 climatic chamber, for 7 days. Figure 3 presents the uniaxial compressive strength (UCS)
192 obtained by each of the AAC tested. The highest UCS (11.6 MPa) was obtained with the
193 lower activator/precursor ratio of 0.50.

194



195

196

Figure 3. Compressive strength, after 7 days curing, of the 50% FA + 50% GW precursor paste

197

198 *2.3 Definition of the AAC/soil composition*

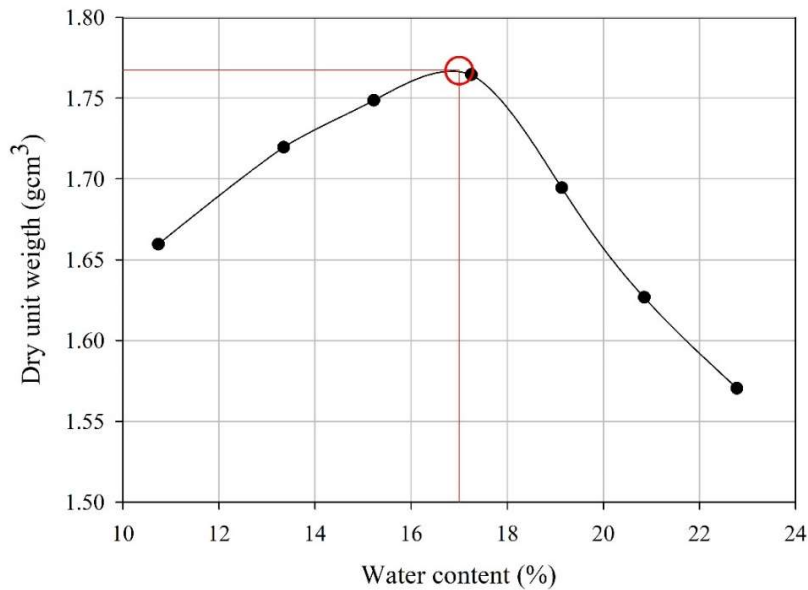
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200 To define the most adequate activator content, in terms of dry density, a Standard Proctor
201 test was carried out based on the methodology proposed by the EN13286-42 (2004), using
202 the AAC with an activator/precursor ratio of 0.50 as a starting point. The test was performed
203 with 3 layers and 12 blows per layer, on pastes with a precursor/soil weight ratio of 30/70. A
204 curve was obtained (Figure 4) by adding more or less activator to the solids (precursor +
205 soil), from which the maximum dry density and optimum moisture content were identified
206 as 1.77 g/cm³ and 17%, respectively.

207

208 However, when the CEB was compacted, significant expansion was registered, which was
209 attributed to excessive moisture. To avoid this effect during the fabrication of the CEB, it
210 was decided to further optimise the moisture content in the AAC/soil composition, aiming at
211 a value lower than 17%, without reducing the density of 1.77 g/cm³ obtained during the
212 Proctor test. This was possible due to the higher energy applied by the earth block compaction
213 machine, compared with the dynamic energy transmitted by the Proctor hammer.

214



215

216 Figure 4. Proctor test results for the stabilised soil, using the activator as the liquid phase and a precursor/soil
 217 weight ratio of 30/70

218

219 This optimisation was developed through an experiment design based on the response surface
 220 methodology (RSM). A Central Composite Design (CCD) was configured with the aim of
 221 finding a relation between the independent variables and the response variable (compressive
 222 strength) capable of maximize the latter (Montgomery and Runger, 2014), therefore defining
 223 the most effective AAC/soil combination. Table 3 presents the independent variables
 224 considered (i.e. soil/precursor and activator/solids), as well as their respective range. The
 225 commercially available software *Minitab 17* was used which, based on the number of
 226 independent variables assumed, established a total of 13 random runs. The soil/precursor
 227 range was defined based on previous experience by the research team (Corrêa-Silva et al.,
 228 2019; Cristelo et al., 2018; Miranda et al., 2017; Rios et al., 2016); while the activator/solid
 229 range was defined after the maximum and minimum CS content in the paste that presented
 230 the most adequate workability. This range was on the ‘dry side’ of the Proctor curve, thus
 231 below the 17% humidity threshold that produced the expansion of the paste. The response

232 variable was assessed through compressive strength tests on compacted specimens with 70
 233 mm in diameter and 140 mm in height, cured at 20 ± 1 °C for 28 and 180 days.

234

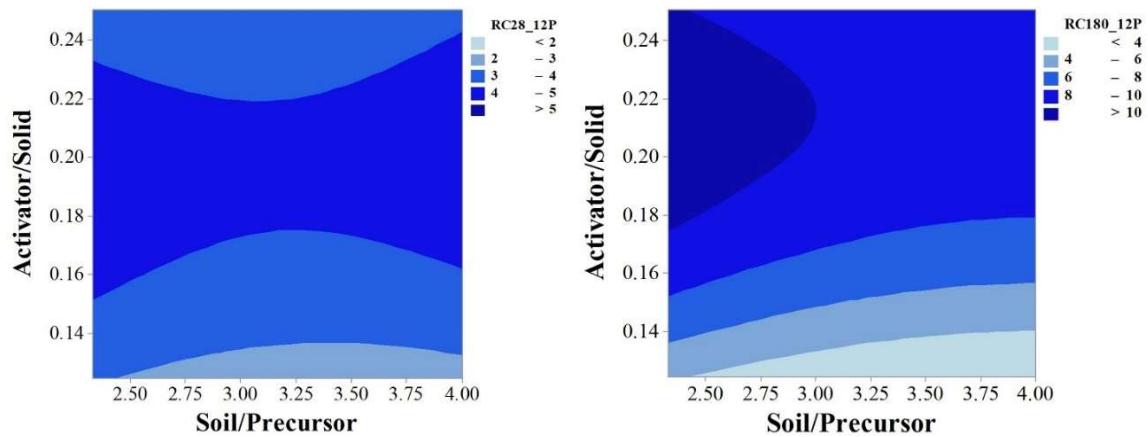
235 The influence of each independent variable and the corresponding compressive strength is
 236 presented in Figure 5. For the 28-day curing period, higher UCS were attained when the
 237 pastes showed the highest CS content, regardless of the precursor content; while for 180-day
 238 period the higher UCS was recorded when the CS and precursor contents simultaneously
 239 assumed their higher values.

240

241 Table 3. Definition of the independent variables and corresponding value range

Variable	Level	Value
Activator/Solids	Lower (-)	0.12
	Upper (+)	0.25
	Central Point	0.18
Soil/Precursor	Lower (-)	2.3
	Upper (+)	4.0
	Central Point	3.1

242



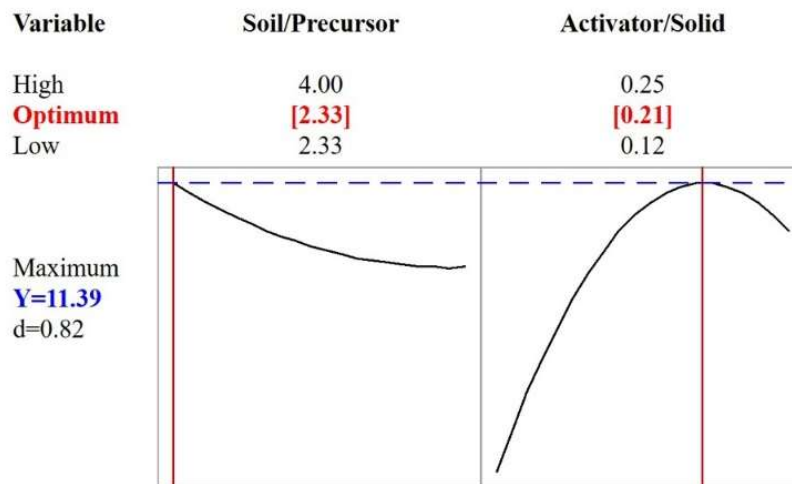
243

244 Figure 5. Contour diagrams showing the relation between the independent variables and the UCS after 28
 245 days (left) and 180 days curing (right) on the AAC/soil

246

247

248 The final values of each independent variable, based on the compressive strength data of 11.4
 249 MPa (desirability of 0.82), obtained after 180 days (Figure 6), were set at soil/precursor =
 250 2.3 and activator/solids = 0.21.
 251



252
 253 Figure 6. Optimised independent variables and resulting UCS and desirability values

254
 255 *2.4 Fabrication of the Compressed Earth Blocks*

256
 257 The CEB were fabricated with the optimised AAC-soil combination. The main steps towards
 258 the production of each block are presented in Figure 7, and involve, after a thorough mixture
 259 of the soil with the binder, the filling of two simultaneous moulds, in a fully manually
 260 operated CEB machine, their subsequent compression and, finally, the curing and drying
 261 stage. Considering that the 180-day curing result was chosen as the reference, and to avoid
 262 such prolonged waiting, the CEBs were cured at 85°C for 20h, to accelerate the dissolution
 263 stage and the production of binding gel (Bakharev, 2005; Criado et al., 2010; Fernández-
 264 Jiménez et al., 2008, 2006; Torres-Carrasco et al., 2014; Torres-Carrasco and Puertas, 2015).
 265



266

267 Figure 7. Main steps in the fabrication process of the CEBs, including, after mixture of the soil with the
268 alkaline cement, the filling of the moulds (a and b), the subsequent compression (c) and the curing and drying
269 of the resulting blocks (d)

270

271 The physical-mechanical characterization of the CEBs was carried out with compressive
272 strength tests, water absorption tests, wetting-drying tests, ice-thawing tests and water
273 erosion tests. Furthermore, the microstructural characterization of the AAC-soil was
274 performed using Fourier Transform Infra-Red Spectroscopy (FTIR) and Scanning Electron
275 Microscopy (SEM).

276

277

278 3. Results and discussion

279

280 3.1 Mechanical performance of the CEBs

281

282 The compressive strength is mostly accepted as a universal property to determine the quality
283 of the CEBs. In general, the compressive strength is related to the type of soil; type and
284 amount of stabilizer; compaction pressure and process. Table 4 summarizes the compressive
285 strength results performed on the CEBs in both dry and saturated conditions, together with
286 the percentage of water absorption until saturation. The maximum compressive strength
287 reached by the CEB (17.23 MPa) was significantly higher than the uniaxial compressive
288 strength developed by the optimum AAC/soil formulation (11.4 MPa). Such increase, of
289 more than 50%, is not attributed to the geometry differences between the two types of
290 specimen, but mostly to the thermal curing process that was implemented for the CEB.
291 According to some authors (Bakharev, 2005; Singh and Subramaniam, 2019) fly ash-based
292 alkaline cements generate mechanical strength very slowly at room temperature. On the
293 contrary, the dissolution of the amorphous phase of the fly ash is improved if the curing
294 occurs under an increased temperature (Rivera et al., 2018; Torres-Carrasco and Puertas,
295 2015), generating aluminosilicate gel in shorter periods and, thus, increasing the strength
296 development rate of the material (Ryu et al., 2012; Šimonová et al., 2018; Singh and
297 Subramaniam, 2019).

298

299 Table 4. Compressive strength and water absorption of the CEBs (UNE 41410, 2008)

Property		Type of specimen	
		<i>Unsaturated CEB</i>	<i>Saturated CEB</i>
Compressive strength (MPa)	Average	17.23	7.45
	Range	14.94 - 19.39	6.35 - 9.29
	N° of specimens	4	4
Water absorption (%)	Average	-	16.21
	Range	-	15.92 - 16.49
	N° of specimen	-	4

300

301 The UNE 41410 (2008) standard classifies the CEBs in three categories, according to its
302 respective compressive strength: CEB1, CEB3 and CEB5. Each category corresponds to the
303 minimum compressive strength of the CEB, e.g. category CEB5 corresponds to the blocks
304 with a strength of at least 5 MPa. According to this standard, it is possible to classify the
305 present CEB in the CEB5 category. ASTM C62-17 (2017) specifies the minimum
306 requirements of solid masonry made of compacted clay or similar materials, sintered at high
307 temperatures. Based on this standard, the CEB designed in this research can be classified as
308 a ‘MW’ grade brick, with the advantage that in the manufacturing process of the block no
309 high temperatures were used for sintering its components. This means that good mechanical
310 properties were achieved with an amount of incorporated energy well below the energy
311 required to fabricate a cooked brick.

312

313 *3.2 Durability tests*

314

315 Figure 8 shows images of the evolution of the wetting-drying test, using a total of 7 blocks,
316 following the contents of the UNE 41410 standard. After detailed visual inspection, no
317 significant deterioration was observed after the planned 6 cycles of wetting-drying. This
318 corroborates that the stabilization of the soil and the cementation of its particles was very
319 effective, to the point where the physical integrity of the blocks was not affected by the
320 volume changes intentionally induced by the test.

321

322 The ice-thaw test used for BTCs was done based on DIN 18945-47 (2013). The durability
323 assessment of the blocks is very important since they mostly measure the longevity of the

324 CEB. Moreover, sequential submission to freezing and thawing is one of the most aggressive
325 environmental conditions for this class of compacted earth masonry elements. The humidity
326 lodged in the structure of the compacted piece, at the time of the freezing, generates internal
327 pressure in the pores due to the increase in volume of the frozen water, which can even cause
328 a structural collapse (Jamshidi et al., 2016; Mak et al., 2016). Figure 9 presents the
329 temperature profile used during the test, comprising a total of 15 cycles. The exposed surface
330 of the block was moistened and submitted to the temperature variations. At the end of the 15
331 cycles, no significant damage was observed, including gaps, cracks, fractures or delamination
332 of the material, demonstrating that the structure of the stabilized CEB was strong enough to
333 withstand the stresses caused by the internal pressure of the frozen water during the test.

334

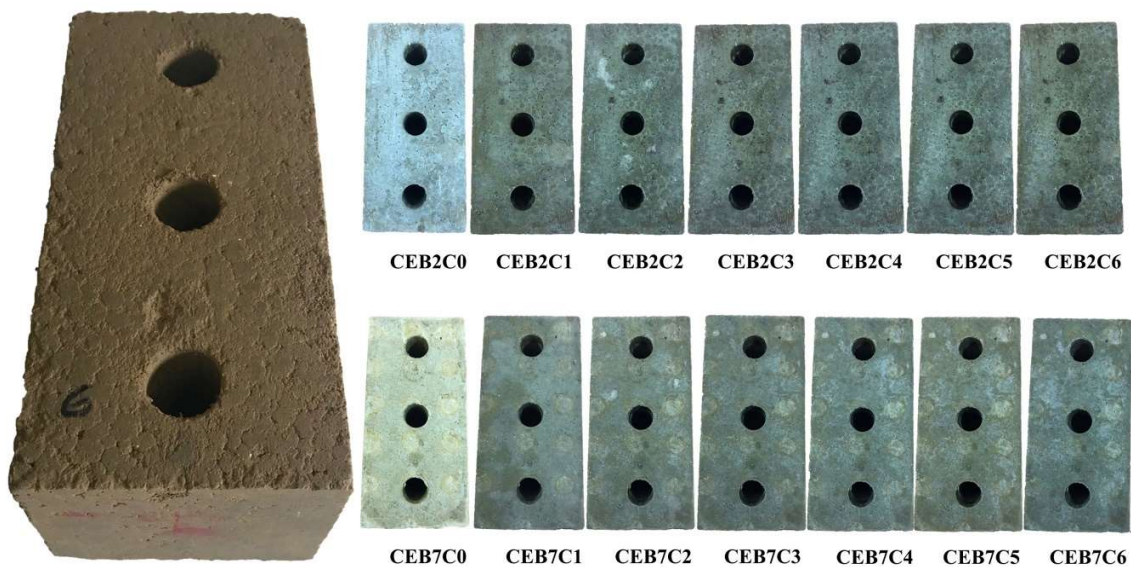
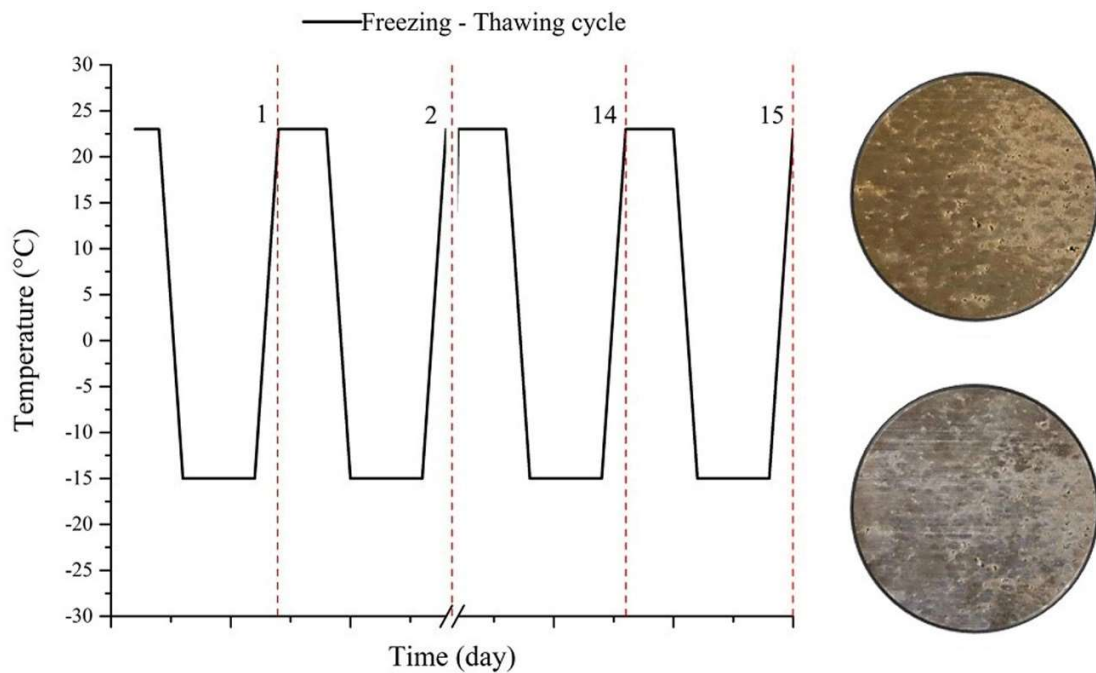


Figure 8. Evolution of two different CEB specimens submitted to the wetting and drying durability test cycles



338

339

Figure 9. Freeze-thaw temperature cycles

340

341 The durability of the CEBs was also evaluated through their resistance to water erosion,

342 tested according to the Swinburne test (SAET), as established in the Spanish standard UNE

343 41410 (2008). This test was specifically developed for earthen materials and simulates the

344 accelerated erosion action of hitting rain. The test setup consists of a deposit, elevated 1 m

345 above the exposed surface of the CEB, which continuously drops water on the surface of the

346 CEB from an outlet of 5 mm diameter, while keeping a constant water head of 0.5 m (Figure

347 10a). The surface of the CEB is inclined 27° relatively to the horizontal plane. The water

348 dropping lasts 10 minutes, after which the specimen is opened and the pitting depth caused

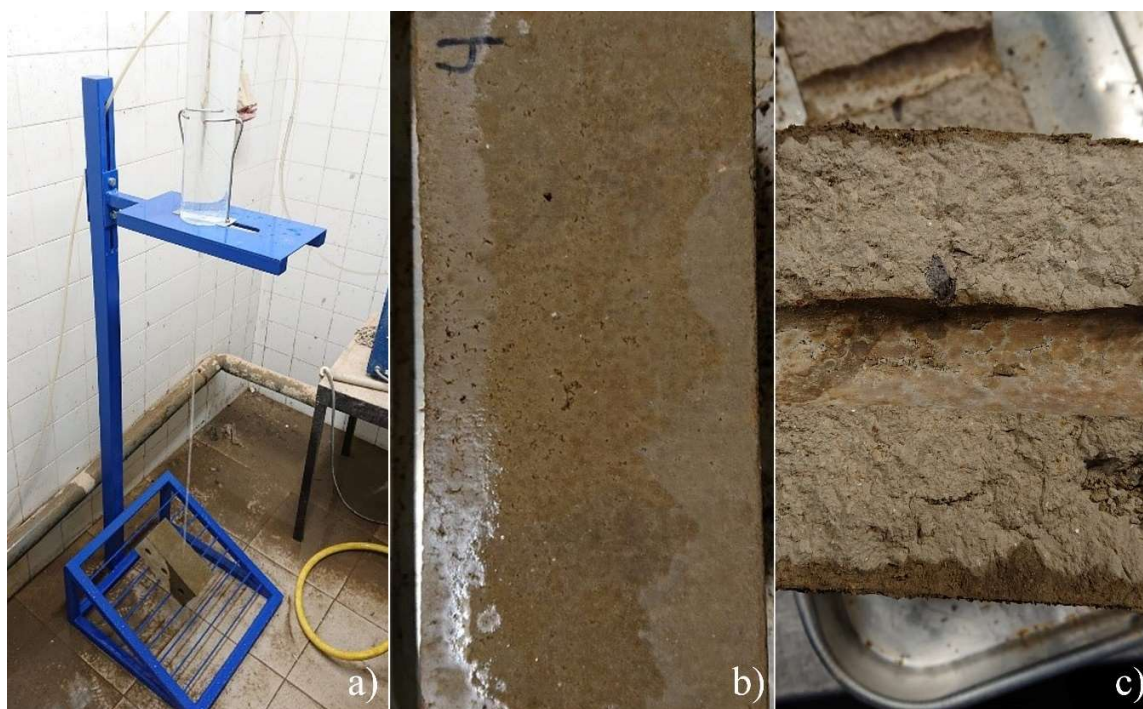
349 by the water stream is measured, using a 3 mm diameter probe. According to the contents of

350 the mentioned standard, the CEBs are suitable if the pitting depth is not higher than 10 mm.

351

352 The results of the three specimens tested are presented in Table 5. As can be seen in Figure
 353 10b, the water stream did not cause any surface degradation, meaning that the stabilisation
 354 solution used is highly effective in providing water erosion resistance to the CEBs. The
 355 moisture penetration depth was additionally measured, by breaking the specimens into 2
 356 halves along the longitudinal section hit by the stream (Figure 10b). An average value of
 357 9 mm was observed, although all specimens showed some variation in the penetration depth
 358 (Figure 10c), which was caused by the presence of some clogs formed during the mix located
 359 near the surface, facilitating the water intake.

360



361

362 Figure 10. Swinburn test setup (a); surface of the CEB after exposure (b); water penetration depth (c)

363

364 Table 5. Results of the Swinburn test (CoV in brackets)

Specimen	Pitting depth (mm)	Moisture penetration depth (mm)
S1	0	9
S2	0	10

S3	0	8
Average	0	9 (16%)

365

366 3.3 FTIR

367

368 Figure 11 shows the IR spectra of the raw material with which the CEBs were manufactured.

369 The FA and WG residues have several signals in common – bands centred around 455, 774

370 and 776 cm^{-1} , associated with Si-O (quartz) functional groups. The main band of the FA is

371 located around 1023 cm^{-1} , characteristic of Si-O-T (T = Al, Si) type links. This spectrum is

372 typical of a material rich in aluminosilicates (Criado et al., 2007; Gao et al., 2014; Panias et

373 al., 2007; Rivera et al., 2019). The GW has a main band centred around 959 cm^{-1} , and the

374 signals located between 950 - 1000 cm^{-1} are associated with vibration modes of Si-O-Na

375 functional groups in condensed units of type Q2 and Q3 silicates, typical of a calcium sodium

376 glass (Khalil et al., 2010; Rivera et al., 2018; Varma et al., 2009; Véron et al., 2013). The

377 characteristic spectrum of the soil reveals signals at 455 cm^{-1} , 748 cm^{-1} and 998 cm^{-1} wave

378 lengths, associated with vibration modes of Si-O-Si and Si-O functional groups (i.e. quartz).

379 The band signal around 529 cm^{-1} can be attributed to stretching vibrations of Fe-O, Fe₂O₃

380 and Si-O-Al type bonds, while bands in the region 850-950 cm^{-1} are characteristic of O-H-

381 Al type links, and the band identified at 1633 cm^{-1} is attributed to bending vibration modes

382 of O-H-O type bonds of water molecules (Kaufhold et al., 2012; Nayak and Singh, 2007;

383 Saikia and Parthasarathy, 2010).

384

385 The cleaning solution, being a highly alkaline aqueous solution contaminated by the

386 aluminium foundry industry, is basically composed by products from the reaction between

387 caustic soda and aluminium waste. In its IR spectrum, a signal centred at 527 cm^{-1} can be

388 observed, attributed to non-condensed octahedral species of AlO_6 (these octahedral species
389 can present signals between $400 - 530 \text{ cm}^{-1}$). The region between $700 - 900 \text{ cm}^{-1}$ is
390 characteristic of AlO_4 tetrahedral species (Tarte, 1967). Therefore, it is not surprising that
391 ionic species of $\text{Al}(\text{OH})_4^-$ aluminates in solution were identified in this same region. These
392 aluminate species are predominant in pH values between 8-12, with vibrations of Al-OH type
393 bonds (Li et al., 2014; MA et al., 2007). According to Ram (2001), the bands in the IR
394 spectrum of the cleaning solution around 1019 and 1420 cm^{-1} could be considered as
395 vibration signals of double bond stretching of $\text{Al}=\text{O}$ of amorphous compounds. In short, and
396 as expected, signals of aluminate compounds and aluminium hydroxides were found in the
397 spectrum of the cleaning solution.

398

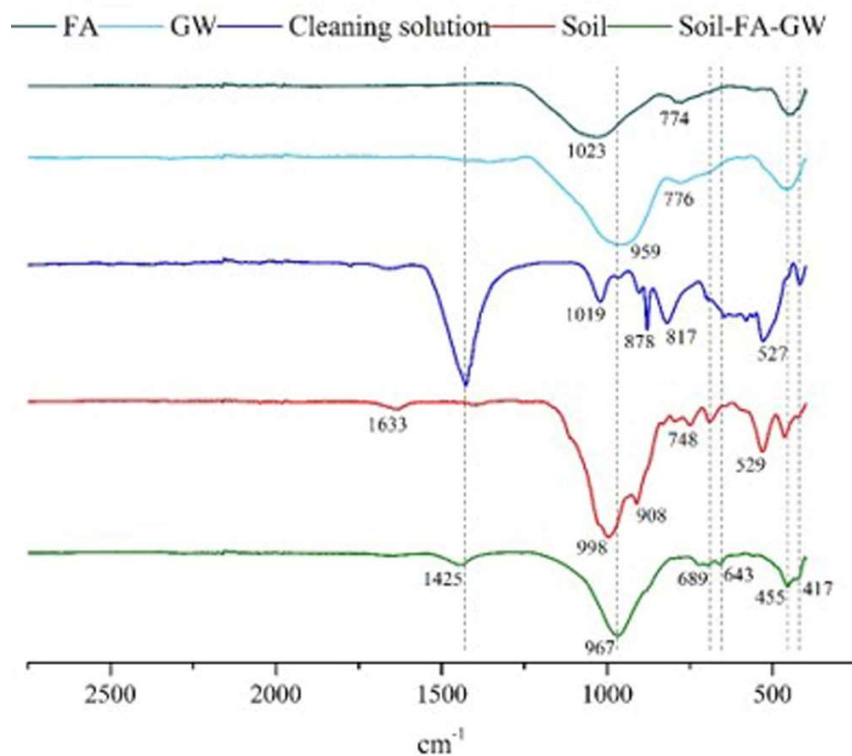
399 When stabilizing the soil-FA-GW mixture with the cleaning solution, some changes in the
400 IR spectrum of the stabilized material were noted. The cleaning solution, being highly
401 alkaline (pH between 12-14), attacks the amorphous acid oxides included in the precursor
402 materials (FA and GW), and even oxides such as FeO and Fe_2O_3 can be affected by the
403 alkaline attack (Lemougna et al., 2013; Lloyd et al., 2009; van Deventer et al., 2007). Some
404 signals are preserved, such as those from the crystalline quartz 455 cm^{-1} .

405

406 On the other hand, bands around 643 and 689 cm^{-1} can be attributed to new reaction products.
407 According to Criado et al. (2007), in the region between $500 - 800 \text{ cm}^{-1}$ the typical signals of
408 various cyclic structures of aluminosilicates formed by the union of SiO_4 and AlO_4
409 tetrahedra, linked by oxygen atoms resulting from the alkaline attack, are identified. The
410 compounds identified in the cleaning solution apparently now form part of the new structure
411 of the cementing gel, since the signals of the $\text{Al}(\text{OH})_4^-$ aluminates and AlO_4 tetrahedra are

412 not noticeable in the new stabilized material, and even the signal identified as representing
 413 the Al=O link (1425 cm^{-1}) has decreased almost until disappearing. The main band of the
 414 stabilized material located around 967 cm^{-1} , can represent different types of overlapping links
 415 and structures. Generally the region between $950\text{-}1250\text{ cm}^{-1}$ of the FTIR spectrum
 416 corresponds to Si-O-T type bonds (T = Si, Al) which indicates that the main reaction product
 417 generated by the dissolution of the precursors is an aluminosilicate gel. This region is also
 418 related to signals from reactive terminals of the type Si-O-Na plus whichever ions are
 419 responsible for generating the aluminosilicate chains (Lee and Van Deventer, 2003). The
 420 precise location of the band in this region will depend on the type of generated structure, of
 421 alkaline activator used and of the composition of the binding gel (Criado et al., 2007).

422



423

424

Figure 11. FT-IR spectra of the original and stabilised materials

425

426 3.4 SEM

427

428 The microstructure of the stabilized soil was examined with scanning electron microscopy,
429 complemented with energy dispersive spectroscopy, for chemical assessment of the binding
430 gel developed. Some of the images obtained (the analysis was performed on a sample
431 collected from one of the previously tested CEBs) are presented in Figure 12, showing the
432 overall morphology of the stabilised soil (Figure 12a), where the binding effect of the
433 material synthesized from alkaline activated residues is clear, resulting in a homogeneous
434 and fairly compacted structure with apparently well cemented soil particles. Figure 12b
435 presented the cemented soil particles in higher detail, where it is possible to detect both
436 precursor and soil particles embedded in what appears to be the cementitious matrix. The
437 nature of the reaction products was identified as a sodium aluminosilicate gel (N-A-S-H),
438 which corroborates the characterization done with FTIR. Indeed, it was previously proposed
439 that the nature of the reaction products was a sodium aluminosilicate gel, due to the nature
440 of the precursors and the alkaline activator.

441

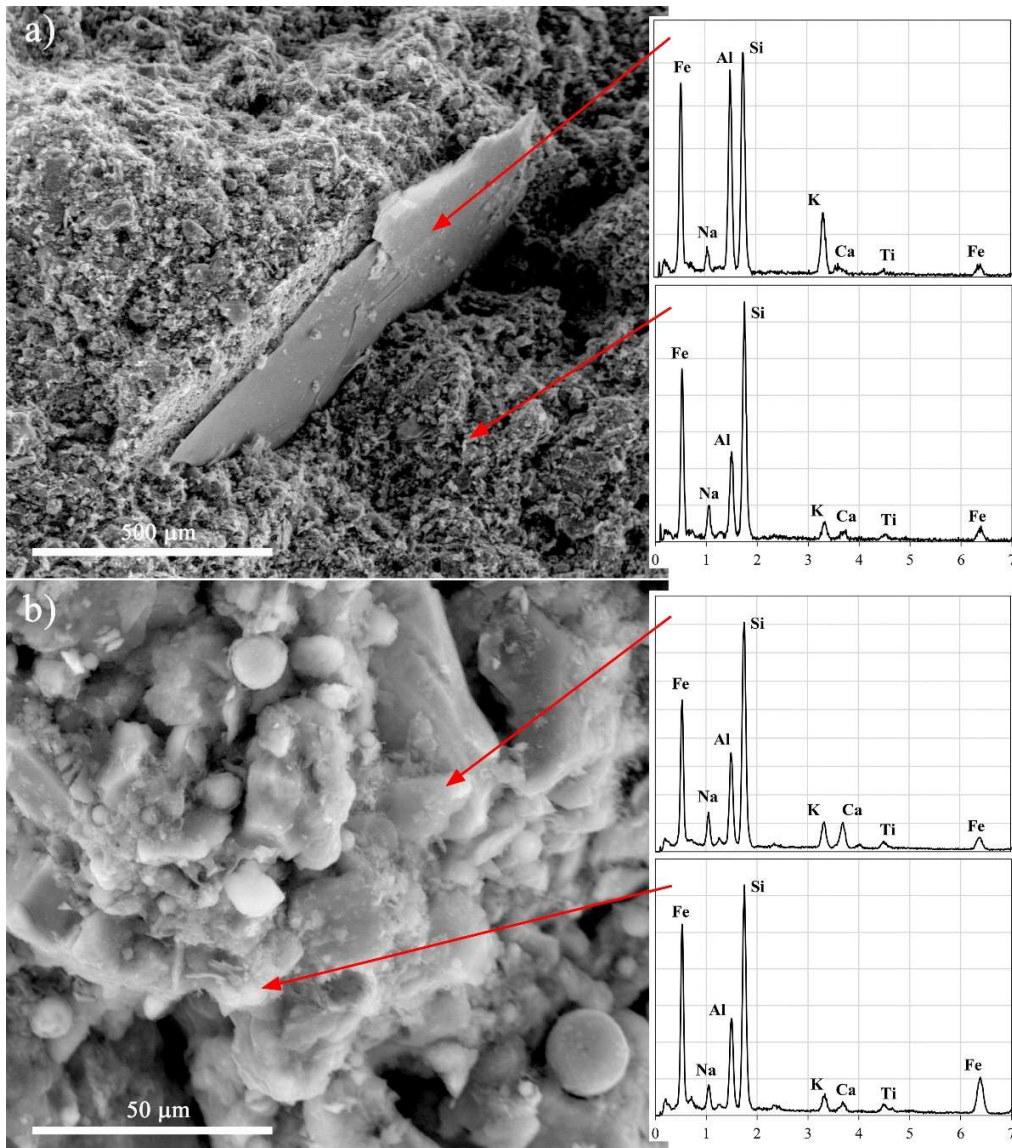


Figure 12. SEM micrographs of the CEB structure

442

443

444

445

446 4. Conclusions

447

448 This research work demonstrated the viability of alternative and environmentally friendly

449 soil stabilisation, with the specific purpose of manufacturing masonry elements commonly

450 known as “compressed earth blocks”, or CEBs. The alternative synthesized cements are

451 100% based on industrial waste, and its applicability to soil stabilisation was thoroughly
452 assessed through strength and durability tests. The results showed that such material can be
453 an effective alternative to traditional chemical stabilisation of soils, namely to calcium-based
454 cements, such as lime or Portland cement. The manufacture of compacted earth masonry
455 elements using this type of cementitious material suggests both a technical and environmental
456 advantage since, in theory, the amount of incorporated energy in waste-based binding agents
457 is significantly lower than the energy associated with traditional binders. However, and in
458 order to fully corroborate this, it will be necessary to perform a comparison between both
459 types of binders, in order to characterised and make evident and all the advantages and
460 disadvantages.

461

462

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464

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468

469

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480

481

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