

# The Interactions of Cyanobacterial Cytochrome $c_6$ and Cytochrome $f$ , Characterized by NMR\*<sup>§</sup>

Received for publication, April 24, 2002, and in revised form, September 26, 2002  
Published, JBC Papers in Press, September 27, 2002, DOI 10.1074/jbc.M203983200

Peter B. Crowley<sup>‡</sup>, Antonio Díaz-Quintana<sup>§</sup>, Fernando P. Molina-Heredia<sup>§</sup>, Pedro Nieto<sup>§</sup>,  
Martin Sutter<sup>¶</sup>, Wolfgang Haehnel<sup>¶</sup>, Miguel A. De la Rosa<sup>§</sup>, and Marcellus Ubbink<sup>‡</sup>||

From the <sup>‡</sup>Leiden Institute of Chemistry, Leiden University, Gorlaeus Laboratories, P. O. Box 9502, 2300 RA Leiden, The Netherlands, <sup>§</sup>Centro de Investigaciones Científicas Isla de la Cartuja, Universidad de Sevilla y Consejo Superior de Investigaciones Científicas, Américo Vespucio s/n, 41092 Sevilla, Spain, and <sup>¶</sup>Institut für Biologie II, Albert-Ludwigs-Universität Freiburg, Schänzlestrasse 1, D-79104 Freiburg, Germany

**During oxygenic photosynthesis, cytochrome  $c_6$  shuttles electrons between the membrane-bound complexes cytochrome  $bf$  and photosystem I. Complex formation between *Phormidium laminosum* cytochrome  $f$  and cytochrome  $c_6$  from both *Anabaena* sp. PCC 7119 and *Synechococcus elongatus* has been investigated by nuclear magnetic resonance spectroscopy. Chemical-shift perturbation analysis reveals a binding site on *Anabaena* cytochrome  $c_6$ , which consists of a predominantly hydrophobic patch surrounding the heme substituent, methyl 5. This region of the protein was implicated previously in the formation of the reactive complex with photosystem I. In contrast to the results obtained for *Anabaena* cytochrome  $c_6$ , there is no evidence for specific complex formation with the acidic cytochrome  $c_6$  from *Synechococcus*. This remarkable variability between analogous cytochromes  $c_6$  supports the idea that different organisms utilize distinct mechanisms of photosynthetic intermolecular electron transfer.**

Electron transport between the membrane-bound complexes cytochrome  $bf$  and photosystem I (PSI)<sup>1</sup> is maintained by mobile electron carriers. In plants, this task is fulfilled by plastocyanin (Pc), whereas cytochrome  $c_6$  ( $cytc_6$ ) is the only carrier in certain cyanobacteria. There also exist eukaryotic algae and cyanobacteria, which have the capacity to replace Pc with  $cytc_6$  under copper-depleted conditions (1). Recently, a  $cytc_6$ -like protein has been discovered in the thylakoid lumen of *Arabidopsis* (2, 3). Despite being evolutionarily unrelated, Pc and  $cytc_6$  perform equivalent tasks with common reaction partners. This functional convergence suggests that similar interaction properties exist for both proteins. Not surprisingly, Pc and  $cytc_6$  have comparable redox potentials of around 350 mV. Further-

more, both proteins provide similar recognition information, demonstrated by the parallel variation of their isoelectric points, being acidic in green algae while ranging from acidic to basic in cyanobacteria (1, 4, 5).

To date there has been a considerable amount of kinetic and mutational analysis of the electron transfer reaction between both Pc and  $cytc_6$  and their partner PSI (6–11). A hierarchy of mechanisms for interprotein electron transport has emerged from this work. Reduction of PSI by Pc or  $cytc_6$ , isolated from different organisms, can follow an oriented collision mechanism (type I), a two-step mechanism requiring complex formation (type II), or a complex formation with rearrangement of the interface before electron transfer occurs (type III). A remarkable homology between the arrangement of charged and hydrophobic recognition patches on the surfaces of Pc and  $cytc_6$  has also been revealed (12, 13). In particular, a conserved arginine found in cyanobacterial Pc and  $cytc_6$  was shown to be critical for bimolecular association with PSI (9).

Two structures of the  $cytf$ -Pc complex from plant (14) and cyanobacterial (15) sources have been determined. A combination of charged and hydrophobic patches defines the complex interface between turnip  $cytf$  and spinach Pc. In contrast, the complex from *Phormidium laminosum*, a thermophilic cyanobacterium, was found to be predominantly hydrophobic. At present there are no kinetic or structural data for the interaction of  $cytc_6$  and  $cytf$ . Although mutagenesis studies have identified key residues for the reaction with PSI, there is no knowledge of the interaction site involved with  $cytf$ .

To address the question of molecular recognition in  $cytc_6$ , we have investigated complex formation with  $cytf$  using heteronuclear NMR. We have aimed to identify the surface features involved in complex formation and to draw comparisons with  $cytc_6$ -PSI interactions. Two cyanobacterial variants of  $cytc_6$  (Fig. 1), the basic protein from *Anabaena* sp. PCC 7119 (pI 9.0) and the acidic protein from *Synechococcus elongatus* (pI 4.8), have been studied. In this way, the role of electrostatics in protein interactions could be investigated explicitly.  $Cytf$  from *P. laminosum*, with a net charge of  $-14$ , was used as the partner protein. It is important to note that *P. laminosum cytf* shares 74 and 72% sequence identity with *Anabaena cytf* and *S. elongatus cytf*, respectively. Furthermore, the net charge is intermediate of  $cytf$  from *Anabaena* ( $-16$ ) and *S. elongatus* ( $-12$ ). This study reports the first structural characterization of the interactions of  $cytf$  and  $cytc_6$ .

## EXPERIMENTAL PROCEDURES

**Protein Preparation**—The soluble fragment of  $cytf$  was prepared according to previously published methods (15, 16). Unlabeled *Ana-cytc\_6* was produced in *Escherichia coli* GM119 (17) transformed with both pEAC-WT (18) and pEC86 (19). The culture was grown in LB

\* This work was supported by the Research Training Network "TRANSIENT" in the Human Potential Program of the European Commission (Grant HPRN-CT-1999-00095) as well as by the Spanish Ministry of Science and Technology (Grant BMC2000-0444) and the Andalusian Government (Grant PAI, CVI-0198). The costs of publication of this article were defrayed in part by the payment of page charges. This article must therefore be hereby marked "advertisement" in accordance with 18 U.S.C. Section 1734 solely to indicate this fact.

<sup>§</sup> The on-line version of this article (available at <http://www.jbc.org>) contains a table of assignments of the <sup>1</sup>H and <sup>15</sup>N backbone amide resonances of cytochrome  $c_6$  from *Anabaena* sp. PCC 7119 at pH 7.0.

|| To whom correspondence should be addressed. Tel.: 31-71-527-4628; E-mail: m.ubbink@chem.leidenuniv.nl.

<sup>1</sup> The abbreviations used are: PSI, photosystem I; Pc, plastocyanin;  $cytc$ , cytochrome  $c$ ;  $cytc_6$ , cytochrome  $c_6$ ;  $cytf$ , cytochrome  $f$ ; *Ana*, *Anabaena*; *Syn*, *Synechococcus*; HSQC, heteronuclear single quantum correlation spectroscopy; TOCSY, total correlation spectroscopy; NOESY, nuclear Overhauser enhancement spectroscopy.

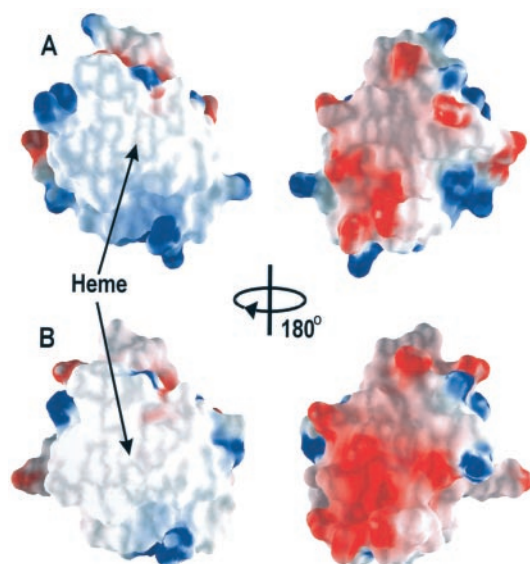


FIG. 1. Electrostatic potential surfaces of *Ana-cyt*<sub>6</sub> model (A) (see “Results and Discussion”) and *Syn-cyt*<sub>6</sub> (B) (25). All images were created with a color ramp for positive (blue) or negative (red) surface potentials saturating at 10 *kT*. Potentials were calculated, for formal charges only, and surfaces were rendered in GRASP (43). The arrows indicate the location of the exposed heme edge.

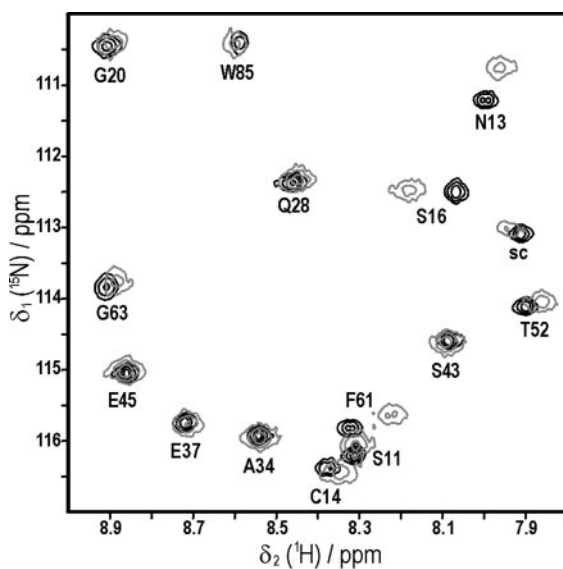


FIG. 2. Spectral region from overlaid <sup>1</sup>H-<sup>15</sup>N HSQC spectra of free *Ana-cyt*<sub>6</sub> (black) and *Ana-cyt*<sub>6</sub> in the presence of two equivalents of *cyt**f* (gray).

medium (20) supplemented with 100 μg/ml ampicillin, 10 μg/ml chloramphenicol, and 1 mM FeCl<sub>3</sub>. Aerobic growth was maintained at 37 °C for 18 h before harvesting. A similar expression system in *E. coli* JM109 was used for the production of uniformly <sup>15</sup>N-labeled *Ana-cyt*<sub>6</sub>. The culture was grown in M9 minimal media additionally supplemented with 1 g/liter <sup>15</sup>NH<sub>4</sub>Cl and 1 mM thiamine. Aerobic growth was maintained at 37 °C for 72 h before harvesting. Isolation and purification of *Ana-cyt*<sub>6</sub> were achieved as described previously (8, 18). The preparation and purification of <sup>15</sup>N, <sup>13</sup>C double-labeled *Syn-cyt*<sub>6</sub> have been reported previously (21).

**NMR Samples**—Protein interactions were investigated for the diamagnetic species only, and reducing conditions were maintained in the presence of sodium ascorbate. Protein solutions were concentrated to the required volume using ultrafiltration methods (Amicon; YM3 membrane) and exchanged into 10 mM potassium phosphate, pH 6.0, 10% D<sub>2</sub>O, 1.0 mM sodium ascorbate. Protein concentrations were determined spectrophotometrically using an ε<sub>553</sub> of 26.2 mM<sup>-1</sup>cm<sup>-1</sup> for the ferrous form of both *Ana-cyt*<sub>6</sub> (22) and *Syn-cyt*<sub>6</sub> and an ε<sub>556</sub> of 31.5 mM<sup>-1</sup>cm<sup>-1</sup> for the ferrous form of *cyt**f*. For the assignment of *Ana-cyt*<sub>6</sub>, 0.7 mM

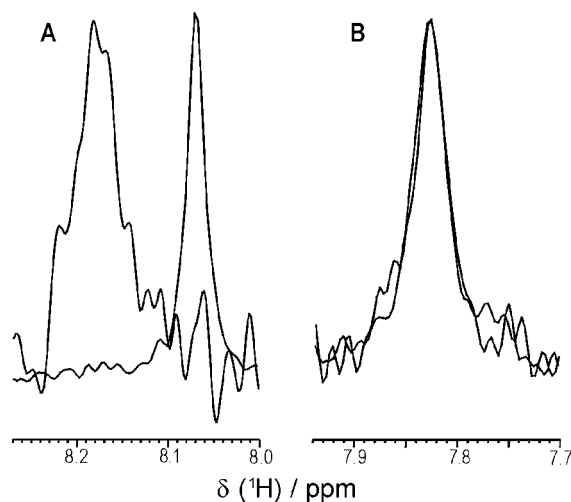


FIG. 3. Line broadening effects. A, cross-sections along the *F*<sub>2</sub> dimension through the <sup>1</sup>H<sup>N</sup> resonance of Ser-16 in *Ana-cyt*<sub>6</sub>. The resonance in the free protein (8.07 ppm) has a line width at half-height of 13 Hz. In the presence of two equivalents of *cyt**f*, the resonance shifts to 8.18 ppm, and the line width at half-height increases to 34 Hz. B, cross-sections along the *F*<sub>2</sub> dimension through the <sup>1</sup>H<sup>N</sup> resonance of Gly-12 in *Syn-cyt*<sub>6</sub>. Despite the presence of two equivalents of *cyt**f*, the resonance is unperturbed (compare Ala-12 of *Ana-cyt*<sub>6</sub>, Fig. 4), and the line width at half-height increases by only 2 Hz.

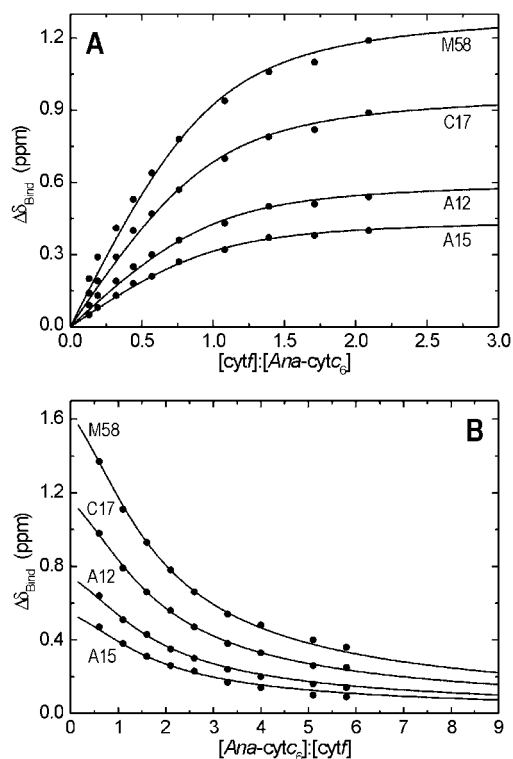


FIG. 4. Binding curves for the interaction of *Ana-cyt*<sub>6</sub> and *cyt**f*. As shown in A, a 0.3 mM sample of <sup>15</sup>N-*Ana-cyt*<sub>6</sub> was titrated with a 1.9 mM solution of *cyt**f*. The data were simultaneously fitted (non-linear, least-squares) to a 1:1 model (27), yielding a binding constant of  $\sim 1 \times 10^4$  M<sup>-1</sup>. As shown in B, in the reverse titration, a 0.5 mM sample of *cyt**f* was titrated with a 3.0 mM *Ana-cyt*<sub>6</sub> stock. Fits to the 1:1 model yielded a binding constant of  $8 (\pm 2) \times 10^3$  M<sup>-1</sup>.

<sup>15</sup>N-labeled and 1.0 mM unlabeled samples were prepared. To investigate complex formation between *Ana-cyt*<sub>6</sub> and *cyt**f*, microliter aliquots of a 1.9 mM *cyt**f* stock solution were titrated into an NMR sample containing 0.3 mM <sup>15</sup>N-*Ana-cyt*<sub>6</sub>. A reverse titration was also performed in which a sample containing 0.2 mM <sup>15</sup>N-*Ana-cyt*<sub>6</sub> and 0.5 mM *cyt**f* was titrated with a 3.0 mM stock of unlabeled *Ana-cyt*<sub>6</sub>. To investigate complex formation between *Syn-cyt*<sub>6</sub> and *cyt**f*, a 0.35 mM sample

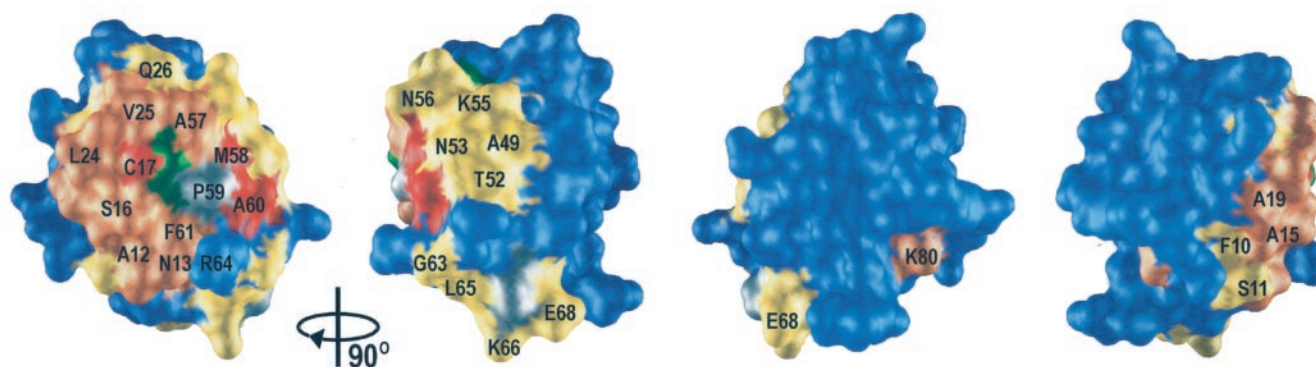


FIG. 5. Chemical-shift map of *Ana-cytc*<sub>6</sub> in the presence of two equivalents of *P. laminosum cytf*. Residues are colored according to their largest  $\Delta\delta_{\text{Bind}}$  (ppm): blue,  $< 0.03$  ( $^1\text{H}$ ) and  $< 0.1$  ( $^{15}\text{N}$ ); yellow,  $\leq 0.06$  ( $^1\text{H}$ ) and  $\leq 0.3$  ( $^{15}\text{N}$ ); orange,  $\leq 0.10$  ( $^1\text{H}$ ) and  $\leq 0.6$  ( $^{15}\text{N}$ ); red,  $\leq 1.0$  ( $^1\text{H}$ ) and  $\leq 2.0$  ( $^{15}\text{N}$ ). Prolines are indicated in gray, and the heme is colored green. Each view represents a  $90^\circ$  rotation with respect to the previous orientation.

of  $^{15}\text{N}$ ,  $^{13}\text{C}$ -*Syn-cytc*<sub>6</sub> was titrated with a 1.9 mM *cytf* solution. After each addition of protein, the pH of the samples was verified, and  $^1\text{H}$ - $^{15}\text{N}$  HSQC spectra were recorded.

**NMR Spectroscopy**—For sequence-specific assignment of the backbone resonances of *Ana-cytc*<sub>6</sub>, two-dimensional  $^1\text{H}$ - $^{15}\text{N}$  HSQC, three-dimensional  $^1\text{H}$ - $^{15}\text{N}$  NOESY-HSQC (100 ms mixing time), and three-dimensional  $^1\text{H}$ - $^{15}\text{N}$  TOCSY-HSQC (80 ms mixing time) spectra were recorded on a Bruker DRX 500 NMR spectrometer. For assignment of the side-chain protons, two-dimensional homonuclear NOESY and TOCSY spectra were recorded. XWINNMR was used for spectral processing, and the assignment was performed in XEASY (23). Complete assignments of the  $^1\text{H}$  and  $^{15}\text{N}$  resonances were determined for all backbone amides (see Supplementary material, Table S1). The non-exchangeable side-chain protons were also assigned. This assignment is consistent with those reported previously for *Monoraphidium braunii cytc*<sub>6</sub> (24) and *Syn-cytc*<sub>6</sub> (25).

Measurements on samples containing *cytf* and either  $^{15}\text{N}$ -*Ana-cytc*<sub>6</sub> or  $^{15}\text{N}$ ,  $^{13}\text{C}$ -*Syn-cytc*<sub>6</sub> were performed on a Bruker DMX 600 NMR spectrometer operating at 300 K. Two-dimensional  $^1\text{H}$ - $^{15}\text{N}$  HSQC spectra (26) were recorded with spectral widths of 30.0 ppm ( $^{15}\text{N}$ ) and 13.9 ppm ( $^1\text{H}$ ). Analysis of the chemical-shift perturbation ( $\Delta\delta_{\text{Bind}}$ ) with respect to the free protein was performed in XEASY.

**Binding Curves**—Titration curves were obtained by plotting  $\Delta\delta_{\text{Bind}}$  against the molar ratio (*R*) of [*cytf*]:[*Ana-cytc*<sub>6</sub>]. For the reverse titration,  $\Delta\delta_{\text{Bind}}$  was plotted versus [*Ana-cytc*<sub>6</sub>]:[*cytf*]. Non-linear least-squares fits to a one-site model (27) were performed in Origin (Originlab, Northampton, MA). This model explicitly treats the concentration of both proteins with *R* and  $\Delta\delta_{\text{Bind}}$  as the independent and dependent variables, respectively (27). The binding constant ( $K_a$ ) and the maximum chemical-shift change ( $\Delta\delta_{\text{Max}}$ ) were the fitted parameters. A global fit was performed in which the curves were fitted simultaneously to a single  $K_a$  value, whereas  $\Delta\delta_{\text{Max}}$  was allowed to vary for each resonance.

## RESULTS AND DISCUSSION

**The Complex of *cytf* and *Ana-cytc*<sub>6</sub>**—Comparison of  $^1\text{H}$ - $^{15}\text{N}$  HSQC spectra of free *Ana-cytc*<sub>6</sub> and *Ana-cytc*<sub>6</sub> in the presence of *cytf* revealed distinct differences arising from complex formation (Fig. 2). When *cytf* was titrated into *Ana-cytc*<sub>6</sub>, 33 of the backbone amides experienced chemical-shift perturbation,  $\Delta\delta_{\text{Bind}} \geq 0.03$  ppm ( $^1\text{H}$ ) and/or  $\geq 0.10$  ppm ( $^{15}\text{N}$ ). A single averaged resonance was observed for each backbone amide, indicating that the free and bound forms of *cytc*<sub>6</sub> were in fast exchange on the NMR time scale. In addition to chemical-shift perturbation, the presence of *cytf* caused a general broadening of about 20 Hz of the amide resonances (Fig. 3A), as expected for complex formation (28).

Titration curves of  $\Delta\delta_{\text{Bind}}$  versus the molar ratio of *cytf*:*Ana-cytc*<sub>6</sub> were plotted for the  $^{15}\text{N}$  nuclei of the four most strongly shifted resonances (Fig. 4A). The curves clearly illustrate that the chemical-shift perturbation increases as a function of the *cytf* concentration. Despite the addition of two equivalents of *cytf*, however, saturation of the chemical-shift changes was not observed. The binding curves were fitted to a 1:1 model with a

binding constant of  $\sim 1 \times 10^4 \text{ M}^{-1}$ . The quality of the fit is poor, particularly at the beginning of the curve, which is perhaps due to a small systematic error in the determination of the protein ratio. To investigate this further, a reverse titration was performed in which a sample of *cytf* was titrated with *Ana-cytc*<sub>6</sub>. In this case, the binding curves were fitted satisfactorily with a binding constant of  $8 (\pm 2) \times 10^3 \text{ M}^{-1}$  (Fig. 4B). It can be concluded that the binding affinity of *cytf* for *Ana-cytc*<sub>6</sub> is  $\sim 10^4 \text{ M}^{-1}$ , which is about 2 orders of magnitude greater than the affinity for the physiological partner, *P. laminosum Pc* (15, 29). From the ratio of the observed  $\Delta\delta_{\text{Bind}}$  to the fitted  $\Delta\delta_{\text{Max}}$  (Fig. 4B), it was calculated that 63% of the *Ana-cytc*<sub>6</sub> was bound in the first point of the reverse titration.

Six structures of *cytc*<sub>6</sub>, from green algal (30–32), red algal (33), and cyanobacterial (12, 25) sources, have been determined previously. All of the known structures exhibit high structural homology. At present, the structure of *Ana-cytc*<sub>6</sub> is unavailable, and therefore, a model was built in Swiss-MODEL (34) using the NMR structure of *Syn-cytc*<sub>6</sub> (25) as a template. The chemical-shift map in Fig. 5 illustrates the location of the affected residues in this model with each residue colored according to its observed  $\Delta\delta_{\text{Bind}}$ . The complex interface consists of a well defined patch, which surrounds the exposed methyl groups (methyl 3, thioether 4 methyl, and methyl 5) of the heme. This patch is composed mainly of three stretches of the primary structure, residues 9–19, 23–26, and 51–61. The first of these includes the heme-binding motif CXYCH, whereas the sixth ligand, Met-58, occurs in the third stretch. Val-25, which was shown to be important for the interaction between *cytc*<sub>6</sub> and PSI (8), is found in the second stretch. Cys-17, Met-58, and Ala-60 experience the largest shifts in the complex. Alanine and asparagine are the most abundant residues in the interface, accounting for one-third of all the affected residues. Although 60% of the interface can be classified as hydrophobic, only 4 residues are charged (Lys-55, Lys-66, Glu-68, and Lys-80). Notably the conserved arginine, Arg-64, which has been implicated in the reaction with PSI (9), was not shifted in the complex. Fragata (35) has produced a theoretical model of the complex formed between *M. braunii cytc*<sub>6</sub> and turnip *cytf* (Protein Data Bank accession code 1jx8). There is good agreement between the binding site on *cytc*<sub>6</sub> identified in this model and the experimentally observed interaction site on *Ana-cytc*<sub>6</sub>.

The complex of *cytf* and *Ana-cytc*<sub>6</sub> was also investigated at 50, 100, and 200 mM NaCl. The observed  $\Delta\delta_{\text{Bind}}$  for the 33 affected amides is plotted as a function of the salt concentration in Fig. 6. As the salt concentration was increased,  $\Delta\delta_{\text{Bind}}$  decreased. At 200 mM NaCl, the  $\Delta\delta_{\text{Bind}}$  is zero for most residues. This suggests that the binding constant is considerably re-



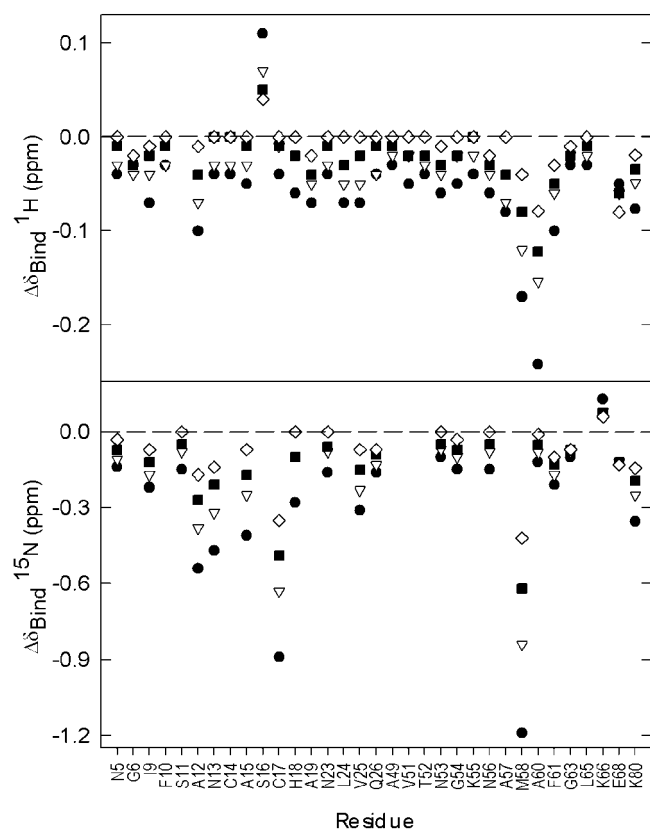


FIG. 6. Salt dependence of  $\Delta\delta_{\text{Bind}}$  for all 33 affected backbone amide resonances, observed in the complex of *Ana-cytc*<sub>6</sub> and *cytf* at 0 mM NaCl (●), 50 mM NaCl (▽), 100 mM NaCl (■), and 200 mM NaCl (◇).

duced, illustrating the importance of attractive electrostatic interactions in complex formation. The more strongly affected residues such as Cys-17 and Met-58 still experience appreciable shifts at 200 mM NaCl. Furthermore, although the line broadening decreased with increasing salt, there remains ~5 Hz broadening at 200 mM NaCl. These observations are indicative of a significant interaction even at high ionic strength.

**The Interactions of *cytf* and *Syn-cytc*<sub>6</sub>**—In contrast to the case of *Ana-cytc*<sub>6</sub>, titration of *cytf* into *Syn-cytc*<sub>6</sub> produced only minor effects in the <sup>1</sup>H-<sup>15</sup>N HSQC spectra. Despite the presence of two equivalents of *cytf*, the amide resonances of *Syn-cytc*<sub>6</sub> did not experience chemical-shift perturbation. Line broadening effects on the order of 2 Hz were observed (Fig. 3B), consistent with a weak and highly dynamic interaction (36). Increasing the ionic strength by the addition of 200 mM NaCl had no effect on the protein interactions. In contrast to the clearly defined complex between *cytf* and *Ana-cytc*<sub>6</sub>, there is no evidence for specific complex formation with *Syn-cytc*<sub>6</sub>.

**Biological Implications**—From our results, it is clear that *Ana-cytc*<sub>6</sub> and *cytf* form a well defined complex and that the amount of complex formed is dependent on the ionic strength. It has been shown previously that *Ana-cytc*<sub>6</sub> forms a complex with *Anabaena* PSI, the affinity of which decreases with increasing ionic strength (7). Although *Ana-cytc*<sub>6</sub> has a net charge close to zero, the presence of positive residues in the vicinity of the heme group promotes favorable electrostatic docking to *cytf*. The role of electrostatics in this complex is therefore analogous to the complex formed between plant *cytf* and *Pc* *in vitro* (14, 27, 37–39). As illustrated in Fig. 5, the interaction site of *Ana-cytc*<sub>6</sub> is composed mainly of hydrophobic residues, which is necessary to achieve specific complex formation between the two proteins. Such a hydrophobic site is sim-

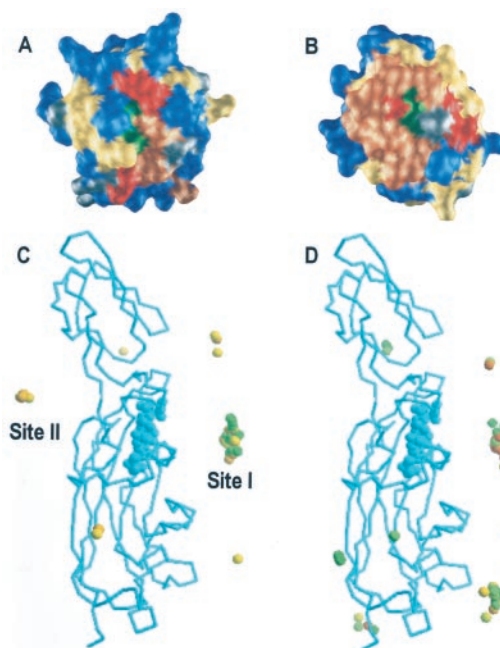


FIG. 7. Comparison of the chemical-shift maps of yeast *cytc* (29) (A) and *Ana-cytc*<sub>6</sub> (B) in the presence of *cytf*, with color coding as in Fig. 5. The experimentally ranked docking solutions generated by BiGGER (42) are illustrated in the lower panel. *Cytf* is depicted as the C<sup>α</sup> trace with the heme group in spacefill. The geometric centers of *cytc* (C) and *Ana-cytc*<sub>6</sub> (D) are represented by spheres in each of the top 50 docking orientations. Color coding from green to red indicates the ranking position, with red being more favorable.

ilar to that proposed previously for the interaction of *Ana-cytc*<sub>6</sub> with PSI (8).

A type I mechanism has been reported for the reduction of PSI by *Syn-cytc*<sub>6</sub> (21). In this mechanism, the electron transfer rate is proportional to the number of collisions in which the redox centers of both proteins are aligned. The NMR titration of *cytf* into *Syn-cytc*<sub>6</sub> provides no evidence for specific complex formation but suggests a highly dynamic interaction between these partners, in agreement with a type I mechanism. The optimal orientation for productive collisions is facilitated by the prominent acidic patch on the backside of *Syn-cytc*<sub>6</sub> (Fig. 1) since the front face will be the energetically more favorable approach with the acidic partners, *cytf* and PSI. Apparently, *cytc*<sub>6</sub> from different organisms utilizes different mechanisms for the electron transfer reaction with its partners. A similar conclusion has been reached for the interactions of *cytf* and *Pc* from different organisms (15). The source of this different reactivity can be traced to the nature of the protein surfaces and ultimately to variations in the primary structures. The sequences of *Ana-cytc*<sub>6</sub> and *Syn-cytc*<sub>6</sub> share 67% identity. Of the 33 residues identified in the *Ana-cytc*<sub>6</sub> interaction site, 22 are conserved in the *Syn-cytc*<sub>6</sub> sequence. Only 6 residues, located around the periphery of the complex interface, are significantly different in *Syn-cytc*<sub>6</sub>. Ala-19 and Gln-26 are replaced by methionines, which have been implicated as endogenous antioxidants (25, 40). Ala-49 is replaced by a tyrosine, whereas glutamine and histidine replace Thr-52 and Asn-53, respectively. This cluster of three variations in *Syn-cytc*<sub>6</sub> results in a bulkier surface, which may hinder interactions with *cytf*. The replacement of Lys-66 by a threonine in *Syn-cytc*<sub>6</sub> is representative of the most striking difference between the two sequences. Five additional lysines in *Ana-cytc*<sub>6</sub> contribute to the significantly higher pI of this protein and enable attractive electrostatic interactions with the acidic *cytf*.

The different roles of electrostatics and hydrophobics can be

further illustrated by comparison with the recently characterized, non-physiological complex of yeast cytochrome *c* (cyt<sub>c</sub>) and cyt<sub>f</sub> (29). It was found that cyt<sub>f</sub> binds two equivalents of cyt<sub>c</sub> with binding constants of  $\sim 2 \times 10^4 \text{ M}^{-1}$  and  $4 \times 10^3 \text{ M}^{-1}$ . The higher binding affinity, as compared with the cyt<sub>f</sub>-*Ana*-cyt<sub>c</sub><sub>6</sub> complex, probably arises from the strong electrostatic attraction between cyt<sub>f</sub> and cyt<sub>c</sub>, which has a pI of 9.7 (41). Fewer residues experienced chemical-shift perturbation in this complex, resulting in a less extensive interaction site (29) (Fig. 7, A and B). Furthermore, the chemical-shift perturbation was on average 50% smaller than that observed in the complex with *Ana*-cyt<sub>c</sub><sub>6</sub>. The prevalence of electrostatic attractions may reduce the amount of time spent in a single orientation, as determined by hydrophobic contacts, and consequently, fewer and smaller chemical-shift changes are observed. Alternatively, the diminished interaction site observed for cyt<sub>c</sub> may be explained in terms of "goodness of fit." It can be assumed that the hydrophobic patch surrounding the heme is better adapted in *Ana*-cyt<sub>c</sub><sub>6</sub> than in yeast cyt<sub>c</sub> for binding to cyt<sub>f</sub>. This complementarity favors closer contact between the protein surfaces and thus a larger interaction site.

Protein docking simulations, using an NMR filter implemented in BiGGER (42), identified the cyt<sub>c</sub> binding sites as the front (Site I) and back (Site II) faces of the heme region of cyt<sub>f</sub> (29) (Fig. 7C). A similar docking simulation using the *Ana*-cyt<sub>c</sub><sub>6</sub> model gave slightly different results. Although the front face of the heme remains the favored site of interaction, there is a significant fraction of favorable docking orientations beneath the heme region in the large domain of cyt<sub>f</sub> (Fig. 7D). Notably, there is significant overlap between the top-ranking docking configuration found for *Ana*-cyt<sub>c</sub><sub>6</sub> in BiGGER and the docking orientation of *M. braunii* cyt<sub>c</sub><sub>6</sub> in the model of Fragata (35). In this model, between the algal cyt<sub>c</sub><sub>6</sub> and plant cyt<sub>f</sub>, there are favorable electrostatics between complementary charged patches, a ridge of lysines on the small domain of cyt<sub>f</sub> and a cluster of acidic residues on the side of cyt<sub>c</sub><sub>6</sub>. This results in a slightly different orientation of cyt<sub>c</sub><sub>6</sub> in comparison with the cyanobacterial complex, which does not possess such well defined complementary charged patches.

**Acknowledgments**—P. B. Crowley is grateful to Dr. J. A. R. Worrall for helpful discussions and to C. Erkelens for assistance with the NMR facilities. A. Díaz-Quintana and P. Nieto thank Dr. M. Bruix for help with recording NMR spectra for the assignment of *Ana*-cyt<sub>c</sub><sub>6</sub>. Dr. M. Hervás is kindly acknowledged for helpful discussions and for critical reading of the manuscript.

## REFERENCES

- Wood, P. M., (1978) *Eur. J. Biochem.* **87**, 9–19
- Wastl, J., Bendall, D. S., and Howe, C. J. (2002) *Trends Plant Sci.* **7**, 244–245
- Gupta, R., He, Z. Y., and Luan, S. (2002) *Nature* **417**, 567–571
- Navarro, J. A., Hervás, M., and De la Rosa, M. A. (1997) *J. Biol. Inorg. Chem.* **2**, 11–22
- Kerfeld, C. A., and Krogmann, D. W. (1998) *Annu. Rev. Plant Physiol.* **49**, 397–425
- Hervás, M., Navarro, J. A., Díaz, A., Bottin, H., and De la Rosa, M. A. (1995) *Biochemistry* **34**, 11321–11326
- Hervás, M., Navarro, J. A., Díaz, A., and De la Rosa, M. A. (1996) *Biochemistry* **35**, 2693–2698
- Molina-Heredia, F. P., Díaz-Quintana, A., Hervás, M., Navarro, J. A., and De la Rosa, M. A. (1999) *J. Biol. Chem.* **274**, 33565–33570
- Molina-Heredia, F. P., Hervás, M., Navarro, J. A., and De la Rosa, M. A. (2001) *J. Biol. Chem.* **276**, 601–605
- Hippler, M., Drepper, F., Haehnel, W., and Rochaix, J. D. (1998) *Proc. Natl. Acad. Sci. U. S. A.* **95**, 7339–7344
- Illerhaus, J., Altschmied, L., Reichert, J., Zak, E., Herrmann, R. G., and Haehnel, W. (2000) *J. Biol. Chem.* **275**, 17590–17595
- Frazão, C., Soares, C. M., Carrondo, M. A., Pohl, E., Dauter, Z., Wilson, K. S., Hervás, M., Navarro, J. A., De la Rosa, M. A., and Sheldrick, G. M. (1995) *Structure* **3**, 1159–1169
- Ullmann, G. M., Hauswald, M., Jensen, A., Kostic, N. M., and Knapp, E. W. (1997) *Biochemistry* **36**, 16187–16196
- Ubbink, M., Ejdeback, M., Karlsson, B. G., and Bendall, D. S. (1998) *Structure* **6**, 323–335
- Crowley, P. B., Otting, G., Schlarb-Ridley, B. G., Canters, G. W., and Ubbink, M. (2001) *J. Am. Chem. Soc.* **123**, 10444–10453
- Schlarb, B. G., Wagner, M. J., Vijgenboom, E., Ubbink, M., Bendall, D. S., and Howe, C. J. (1999) *Gene (Amst.)* **234**, 275–283
- Marinus, M. G., and Morris, N. R. (1973) *J. Bacteriol.* **14**, 1143–1150
- Molina-Heredia, F. P., Hervás, M., Navarro, J. A., and De la Rosa, M. A. (1998) *Biochem. Biophys. Res. Commun.* **243**, 302–306
- Arslan, E., Schulz, H., Zufferey, R., Künzler, P., and Thöny-Meyer, L. (1998) *Biochem. Biophys. Res. Commun.* **251**, 744–747
- Sambrook, J., Fritsch, E. F., and Maniatis, T. (1989) *Molecular Cloning: A Laboratory Manual*, 2nd Ed., Cold Spring Harbor Laboratory, Cold Spring Harbor, NY
- Sutter, M., Sticht, H., Schmid, R., Hörth, P., Rösch, P., and Haehnel, W. (1995) in *Photosynthesis: From Light to Biosphere* (Mathis, P. ed.), Vol. II, pp. 563–566, Kluwer Academic Publishers, The Netherlands,
- Medina, M., Louro, R. O., Gagnon, L., Peleato, M. L., Mendes, J., Gómez-Moreno, C., Xavier, A. V., and Teixeira, M. (1997) *J. Biol. Inorg. Chem.* **2**, 225–234
- Bartels, C., Xia, T.-H., Billeter, M., Güntert, P., and Wüthrich, K. (1995) *J. Biomol. NMR* **5**, 1–10
- Banci, L., Bertini, I., De la Rosa, M. A., Kouloughiotis, D., Navarro, J. A., and Walter, O. (1998) *Biochemistry* **37**, 4831–4843
- Beißinger, M., Sticht, H., Sutter, M., Ejchart, A., Haehnel, W., and Rösch, P. (1998) *EMBO J.* **17**, 27–36
- Andersson, P., Gsell, B., Wipf, B., Senn, H., and Otting, G. (1998) *J. Biomol. NMR* **11**, 279–288
- Kannt, A., Young, S., and Bendall, D. S. (1996) *Biochim. Biophys. Acta-Bioenergetics* **1277**, 115–126
- Zuiderweg, E. R. P. (2002) *Biochemistry* **41**, 1–7
- Crowley, P. B., Rabe, K. S., Worrall, J. A. R., Canters, G. W., and Ubbink, M. (2002) *ChemBioChem* **3**, 526–533
- Kerfeld, C. A., Anwar, H. P., Interrante, R., Merchant, S., and Yeates, T. O. (1995) *J. Mol. Biol.* **250**, 627–647
- Schnackenberg, J., Than, M. E., Mann, K., Wiegand, G., Huber, R., and Reuter, W. (1999) *J. Mol. Biol.* **290**, 1019–1030
- Yamada, S., Park, S.-Y., Shimizu, H., Koshizuka, Y., Kadokura, K., Satoh, T., Suruga, K., Ogawa, M., Isogai, Y., Nishio, T., Shiro, Y., and Oku, T. (2000) *Acta Crystallogr. Sect. D Biol. Crystallogr.* **56**, 1577–1582
- Sawaya, M. R., Krogmann, D. W., Serag, A., Ho, K. K., Yeates, T. O., and Kerfeld, C. A. (2001) *Biochemistry* **40**, 9215–9225
- Guex, N., and Peitsch, M. C., (1997) *Electrophoresis* **18**, 2714–2723
- Fragata, M. (2002) *Biophys. J.* **82**, 1618
- Worrall, J. A. R., Liu, Y., Crowley, P. B., Nocek, J. M., Hoffman, B. M., and Ubbink, M. (2002) *Biochemistry* **41**, 11721–11730
- Soriano, G. M., Ponamarev, M. V., Piskorowski, R. A., and Cramer, W. A. (1998) *Biochemistry* **37**, 15120–15128
- Bergkvist, A., Ejdeback, M., Ubbink, M., and Karlsson, G. (2001) *Protein Sci.* **10**, 2623–2626
- Soriano, G. M., Ponamarev, M. V., Tae, G.-S., and Cramer, W. A. (1996) *Biochemistry* **35**, 14590–14598
- Levine, R. L., Mosoni, L., Berlett, B. S., and Stadtman, E. R. (1996) *Proc. Natl. Acad. Sci. U. S. A.* **93**, 15036–15040
- Park, K. S., Frost, B. F., Shin, S., Park, I. K., Kim, S., and Paik, W. K. (1988) *Arch. Biochem. Biophys.* **267**, 195–204
- Palma, P. N., Krippahl, L., Wampler, J. E., and Moura, J. J. G. (2000) *Proteins* **39**, 372–384
- Nicholls, A., Sharp, K., and Honig, B. (1991) *Proteins* **11**, 281–296

## The Interactions of Cyanobacterial Cytochrome $c_6$ and Cytochrome $f$ , Characterized by NMR

Peter B. Crowley, Antonio Di?az-Quintana, Fernando P. Molina-Heredia, Pedro Nieto,  
Martin Sutter, Wolfgang Haehnel, Miguel A. De la Rosa and Marcellus Ubbink

*J. Biol. Chem.* 2002, 277:48685-48689.

doi: 10.1074/jbc.M203983200 originally published online September 27, 2002

---

Access the most updated version of this article at doi: [10.1074/jbc.M203983200](https://doi.org/10.1074/jbc.M203983200)

### Alerts:

- [When this article is cited](#)
- [When a correction for this article is posted](#)

[Click here](#) to choose from all of JBC's e-mail alerts

### Supplemental material:

<http://www.jbc.org/content/suppl/2002/12/06/277.50.48685.DC1>

This article cites 42 references, 6 of which can be accessed free at  
<http://www.jbc.org/content/277/50/48685.full.html#ref-list-1>