

Electronic structure of the parent compound of superconducting infinite-layer nickelates

M. Hepting^{1†}, D. Li¹, C. J. Jia¹, H. Lu¹, E. Paris², Y. Tseng², X. Feng¹, M. Osada¹, E. Been¹, Y. Hikita¹, Y.-D. Chuang³, Z. Hussain³, K. J. Zhou⁴, A. Nag⁴, M. Garcia-Fernandez⁴, M. Rossi¹, H. Y. Huang⁵, D. J. Huang⁵, Z. X. Shen¹, T. Schmitt², H. Y. Hwang¹, B. Moritz¹, J. Zaanen⁶, T. P. Devereaux¹, and W. S. Lee^{1*}

¹Stanford Institute for Materials and Energy Sciences, SLAC National Accelerator Laboratory and Stanford, Menlo Park, California 94025, USA

²Photon Science Division, Swiss Light Source, Paul Scherrer Institut, CH-5232 Villigen PSI, Switzerland

³Advanced Light Source, Lawrence Berkeley National Laboratory

⁴Diamond Light Source, Harwell Science and Innovation Campus, Didcot, Oxfordshire OX11 0DE, United Kingdom

⁵NSRRC, Hsinchu Science Park, Hsinchu 30076, Taiwan

⁶Leiden Institute of Physics, Leiden University, 2300 RA Leiden, The Netherland

Correspondence to: leews@stanford.edu

Present address:

†Max Planck Institute for Solid State Research, 70569 Stuttgart, Germany

The search for oxide materials with physical properties similar to the cuprate high T_c superconductors, but based on alternative transition metals such as nickel, has grown and evolved over time [1-10]. The recent discovery of superconductivity in doped infinite-layer nickelates $RNiO_2$ (R = rare-earth element) [11,12] further strengthens these efforts. With a crystal structure similar to the infinite-layer cuprates – transition metal oxide layers separated by a rare-earth spacer layer – formal valence counting suggests that these materials have monovalent Ni^{1+} cations with the same $3d$ electron count as Cu^{2+} in the cuprates. Here, we use x-ray spectroscopy in concert with density functional theory to show that the electronic structure of $RNiO_2$ (R = La, Nd), while similar to the cuprates, includes significant distinctions. Unlike cuprates with insulating spacer layers between the CuO_2 planes, the rare-earth spacer layer in the infinite-layer nickelate supports a weakly-interacting three-dimensional $5d$ metallic state. This three-dimensional metallic state hybridizes with a quasi-two-dimensional, strongly correlated state with $3d_{x^2-y^2}$ symmetry in the NiO_2 layers. Thus, the infinite-layer nickelate can be regarded as a sibling of the rare earth intermetallics [13-15], well-known for heavy Fermion behavior, where the NiO_2 correlated layers play an analogous role to the $4f$ states in rare-earth heavy Fermion compounds. This unique Kondo- or Anderson-lattice-like “oxide-intermetallic” replaces the Mott insulator as the reference state from which superconductivity emerges upon doping.

While the mechanism of superconductivity in the cuprates remains a subject of intense research, early on it was suggested that the conditions required for realizing high T_c superconductivity are rooted in the physics of a two-dimensional electron system subject to strong local repulsion [16, 17]. This describes the Mott (charge-transfer) insulators in the stoichiometric

parent compounds, characterized by spin $\frac{1}{2}$ Heisenberg antiferromagnetism, from which superconductivity emerges upon doping. A long-standing question regards whether these “cuprate-Mott” conditions can be realized in other oxides; and extensive efforts to synthesize and engineer nickel oxides (nickelates) have promised such a realization [1-10]. Infinite-layer NdNiO₂ became the first such nickelate superconductor following the recent discovery of superconductivity in Sr-doped samples [11]. The undoped parent compound, produced by removing the apical oxygen atoms from the perovskite nickelate NdNiO₃ using a metal hydride-based soft chemistry reduction process [10, 18-20], appears to be a close sibling of the cuprates—it is isostructural to the infinite-layer cuprates with monovalent Ni¹⁺ cations and possesses the same $3d^9$ electron count as that of Cu²⁺ cations in undoped cuprates. Yet, as we will reveal, the electronic structure of the undoped RNiO₂ ($R = \text{La}$ and Nd) remains distinct from the Mott, or charge-transfer, compounds of undoped cuprates, and even other nickelates.

As a reference, we first discuss the electronic structure of canonical nickelates, NiO and LaNiO₃. Rock salt NiO is a charge-transfer insulator, as characterized in the Zaanen-Sawatzky-Allen scheme [21], whose charge-transfer energy Δ (promoting charge from oxygen ligands to Ni d orbitals) lies below the Coulomb interaction scale U on Ni sites. The valence Ni d orbitals strongly hybridize with oxygen ligands, yielding wavefunctions with mixed character $\alpha|3d^8\rangle + \beta|3d^9L\rangle$ ($\alpha^2 + \beta^2 = 1$), with $\beta^2 \sim 0.2$ [22, 23] per in a NiO₆ octahedron, where L denotes a ligand hole on the oxygens. Such Ni-O ligand hybridization gives rise to a pre-peak in x-ray absorption spectroscopy (XAS) near the O K -edge (Fig. 1a). In addition, a large band gap set by Δ appears in the oxygen partial density of states (PDOS) obtained both experimentally (Fig. 1b) and from LDA+U calculations (Fig. 1e). In the perovskite RNiO₃ where formal valence counting would give Ni³⁺ ($3d^7$), both theoretical and experimental studies indicate that the perovskite structure leads to

a decrease of Δ , such that it becomes effectively negative [24]. Under such a scenario, electrons from oxygen ligands spontaneously transfer to Ni cations, giving rise to “self-doped” holes on the ligands, and a pre-peak in the O *K*-edge XAS (Fig. 1a). As expected for a negative charge-transfer metal, no band gap appears in the oxygen PDOS (Figs. 1c and f) [25].

The O *K*-edge XAS tells a very different story for the infinite-layer nickelates LaNiO₂ and NdNiO₂, as shown in Fig. 1a. The lack of a pre-edge peak suggests that the oxygen ligands carry significantly less weight in the ground state wave function, signaling a weaker effective mixing between oxygen and the Ni¹⁺ cations. Unlike NiO and LaNiO₃, the oxygen PDOS (Fig. 1d) exhibits a diminished weight near the Fermi energy, also indicating that oxygen 2*p* orbitals carry less weight in the states near the Fermi energy by comparison; all of which is consistent with the calculated oxygen PDOS from LDA+U (Fig. 1g).

While the oxygen electronic structure deviates significantly from other nickelates and cuprates [26], we examine the electronic structure of the Ni cation in RNiO₂ using both XAS and resonant inelastic x-ray scattering (RIXS) at the Ni *L*₃-edge (a core-level 2*p* to valence 3*d* transition). As shown in Fig. 2a, while XAS for both NiO and LaNiO₃ exhibit distinct multi-peak structures originating from $2p^63d^8-2p^53d^9$ and $2p^63d^8L^n-2p^53d^9L^n$ multiplet transitions, respectively [23, 24], XAS for the infinite-layer nickelates shows a main absorption peak (denoted A), which closely resembles the single peak associated with the $2p^63d^9-2p^53d^{10}$ transition in cuprates [27]. In particular for LaNiO₂, the XAS exhibits an additional lower energy shoulder A', at which the RIXS spectrum consist a lower energy feature of ~ 0.6 eV (Fig. 2b and e). Note that this feature is absent in the RNiO₃ (*R* = La, Nd) compounds (Fig. 2c, e, and Ref. 24). Using exact diagonalization (see Method), we reproduce the general features from XAS and RIXS (Figs. 2f-h), including the A' features, which highlights the hybridization between the Ni

$3d_{x^2-y^2}$ and La $5d$ orbitals. Thus, in configuration interaction, the Ni state can be expressed as a combination of $|3d^9\rangle$ and $|3d^8R\rangle$ where R denotes a charge transfer to the rare-earth cation (See [Method](#) and [Supplementary Table 2](#)). In NdNiO₂, the ~ 0.6 eV feature due to the Nd-Ni hybridization also exists in RIXS ([Figs. 2d, e](#)), but its resonance energy (A') almost coincides with the main absorption peak A . As a consequence, the A' feature cannot be resolved in XAS ([Fig. 2a](#)).

To further analyze the electronic structure, we turn to density functional theory. The LDA+U scheme [28] has a long track record of reproducing correctly the gross features of correlated electronic structure for transition metal oxides. While generally first principle, one cannot be certain about the value of the local Coulomb interaction U ; however, we can put bounds on it. The infinite-layer nickelates are undoubtedly less good metals than elemental nickel, characterized by $U \sim 3$ eV, which we can take as a lower bound. From O K -edge XAS, the Coulomb interaction should be smaller than that of the large band gap charge-transfer insulator NiO, where $U \sim 8$ eV. Here, we choose $U = 6$ eV in our calculations for LaNiO₂ (with a lowest energy antiferromagnetic solution, see [Method](#) for details), revealing some salient features that correlate with experimental observations: (a) As shown in [Fig. 1g](#) (and [Fig. 3a](#)), when compared to other nickelates, the oxygen $2p$ bands lie significantly further away from the Fermi energy, signaling reduced oxidation and implying a charge-transfer energy Δ that exceeds U . This places the RNiO₂ infinite-layer nickelates within the Mott-Hubbard regime of the Zaanen-Sawatzky-Allen scheme [21]. (b) The density of states near E_F is dominated by the half-filled Ni $3d_{x^2-y^2}$ states, which appear isolated from the occupied Ni $3d$ bands. The characteristic lower and upper Hubbard bands ([Fig. 3a](#)), at least in part, signal a textbook single-band Hubbard model, all but confirming that the Ni cation should be in a very nearly monovalent $3d^9$ state, consistent with the Ni L -edge

XAS and RIXS (Fig. 2). (c) The density of states at E_F is actually finite, but small, as shown upon closer inspection of both Figs. 3a and 3b. Near the Γ point, a Fermi surface pocket forms of mainly La $5d$ character (Fig. 3b); it is quite extended and three-dimensional (see the wavefunction at k_F , Fig. 3c, and Fermi surface, Fig. 3d). This contrasts with the two-dimensional (2D) nature of the correlated $3d_{x^2-y^2}$ Ni states (Fig. 3b). In other words, the electronic structure of the infinite layer nickelate consists of a low density three-dimensional (3D) metallic rare-earth band coupled to a 2D Mott system.

A minimal model for these materials would look like

$$H = \sum_{k,\sigma} (\varepsilon_k^R n_{k,\sigma}^R + \varepsilon_k^{Ni} n_{k,\sigma}^{Ni}) + U \sum_i n_{i,\uparrow}^{Ni} n_{i,\downarrow}^{Ni} + \sum_{k,i,\sigma} (V_{k,i} c_{k,\sigma}^\dagger d_{i,\sigma} + h.c.),$$

where the first term describes the non-interacting rare-earth (R) and Ni bands with energies ε_k^R and ε_k^{Ni} , respectively, the second term represents the usual on-site Hubbard interaction with strength U in the quasi-two-dimensional Ni layer, and the third term describes the coupling with strength $V_{k,i}$ between the R and Ni subsystems. Here, $n_{k,\sigma}^R$ and $n_{k,\sigma}^{Ni}$ represent the usual number operators for the R and Ni subsystems, while $c_{k,\sigma}^\dagger$ ($c_{k,\sigma}$) and $d_{k,\sigma}^\dagger$ ($d_{k,\sigma}$) create (annihilate) electrons in the 3D metallic R and 2D Hubbard-like Ni subsystems, respectively. This model resembles the Anderson-lattice (or Kondo-lattice) model for the rare-earth intermetallics [13-15], but with the notable addition of a weakly hybridized single-band Hubbard-like model for the Ni layer, rather than strongly interacting $4f$ states (or localized spin moments). We can take this a step further by “downfolding” the band structure to a minimal model that should be the starting point for the unusual correlated electron physics in this system. Figure 4a shows the band structure for LaNiO_2 obtained from LDA without a Hubbard U (see Supplementary Information for details). Here, consistent with previous calculations [2], two bands cross the Fermi level: a fully three-dimensional band with predominantly La $5d$ character, and a quasi-two-dimensional band with Ni

$3d$ character. Wannier downfolding [29] produces one extended orbital with $d_{3z^2-r^2}$ symmetry centered on La (Fig. 4b) and another orbital confined primarily to the NiO₂ planes with $d_{x^2-y^2}$ symmetry centered on Ni (Fig. 4c), which are fully consistent with the expected orbital arrangements given the crystal and ligand field symmetries for this material and the LDA+U results shown in Fig. 3. Full details about the downfolded model, including effective model parameters, can be found in the [Supplementary Table 3](#).

This downfolded model is to the best of our knowledge unique to this particular system. Viewed theoretically, this is uncharted territory and it is a natural question to ask what happens to the basic single-band Hubbard model when its states weakly hybridize with a metallic band. For example, do the spins in the NiO₂ layers order antiferromagnetically or will the Kondo effect strongly screen the local moments and give rise to electronic band hybridizations in analogy to the case of heavy Fermions [13-15]? Note that unlike the rare-earth intermetallics, here, Ni spins interact via the strong short range super-exchange interaction, which replaces the RKKY interactions in the heavy Fermion compounds. More importantly, can superconductivity emerge in this model by introducing doped charge carriers? Apparently, experimental information, particularly about the Fermi surface, magnetic susceptibility, and information about other elementary excitations, such as spin, charge, and phonon excitations, will be required to gain further insights. Nevertheless, our results have provided a first glimpse into the novel electronic structure of the parent compounds of superconducting infinite-layer nickelates, which appear to serve as a birthplace of superconductivity upon doping.

Methods

Materials

LaNiO₃ films with 12 and 50 nm thicknesses were grown on top of 5 × 5 mm² TiO₂-terminated SrTiO₃ (001) substrates by pulsed laser deposition (PLD) using a 248nm KrF excimer laser. Prior to growth, SrTiO₃ substrates were pre-annealed at an oxygen partial pressure (p_{O_2}) of 5×10^{-6} torr for 30 min at 950 °C to achieve sharp step-and-terrace surfaces. The films were subsequently grown at a substrate temperature T_g of 575 °C and $p_{O_2} = 34$ mtorr, using 1.4 J cm⁻² laser fluence and 4 mm² laser spot size on the target. The growth was monitored by reflection high-energy electron diffraction (RHEED) intensity oscillations. After the growth, the samples were cooled down to room temperature in the same oxygen environment. Characterization by x-ray diffraction (XRD) scans with Cu K α radiation indicated the presence of the perovskite phase of (001)-oriented LaNiO₃ and high epitaxial quality for all as-grown films. AFM topographic scans showed atomically flat surfaces. Reducing conditions [30] were adapted to remove apical oxygen for producing both the (001)-oriented LaNiO_{2.5} and LaNiO₂ phases. For reduction experiments, each LaNiO₃ sample was cut into 2 pieces of 2.5 × 5 mm² size. The 2.5 × 5 mm² sample was then vacuum-sealed together with blocks of CaH₂ powder in a Pyrex glass tube (pressure < 0.1 mtorr). The tube was heated to 240 °C at a rate of 10 °C/min and kept at this temperature for 30-120 mins, before cooled down to room temperature at a rate of 10 °C/min. After the annealing process, remnant CaH₂ powder on sample surface was rinsed off by 2-Butanone. The XRD scans in [Supplementary Fig. 1a](#) show the characteristic Bragg peaks of the 12 nm LaNiO₃ film and the ~ 50 nm LaNiO₂ film used in the XAS and RIXS measurements of the main text. Additionally, a ~ 50 nm LaNiO_{2.5} film was characterized as a reference sample. The 2θ peak position of these three

films coincide with that of similar films on SrTiO₃ [30]. The *c*-axis lattice constants extracted from the XRD scans are 3.809, 3.771, and 3.407 Å for the LaNiO₂, LaNiO_{2.5}, and LaNiO₂ film, respectively. In comparison to LaNiO₂ powder [17, 30], the *c*-axis lattice constant of the film is slightly expanded due to the compressive strain induced by the SrTiO₃ substrate.

NdNiO₂ films grown on a SrTiO₃ substrate with a thickness of ~10 nm were prepared using the conditions described in Ref. 10. NdNiO₂ films with and without a capping layer of 5 unit cell (u.c.) SrTiO₃ were measured, which show the same spectral properties. As a reference, we also measured a NdNiO₃ film grown on a SrTiO₃ substrate and capped with a 5 u.c. SrTiO₃ film.

[Supplementary Fig. 1b](#) displays the resistivity as a function of temperature of the LaNiO₃ film in a four-probe geometry, which shows metallic behaviour down to 2 K. The LaNiO₂ film exhibits higher resistivity than LaNiO₃ at 300 K, which increases further with decreasing temperature [[Supplementary Fig. 1b](#)]. Similar transport properties were reported in Refs. 19, 30, 31.

Commercially available NiO powder with $\geq 99.995\%$ purity (Sigma-Aldrich) was used for the measurements.

XAS and RIXS measurements

XAS and RIXS spectra of the La-based nickelate samples were measured at the ADRESS beamline with the SAXES spectrometer at the Swiss Light Source (SLS) of the Paul Scherrer Institute [32]. For the RIXS measurements the scattering angle was fixed to 130° and the combined instrument resolution was approximately 100 meV at the Ni *L*₃-edge. The scattering plane coincided with the crystallographic *ac* (*bc*) plane with a grazing incident angle $\theta = 15^\circ$. The XES and RIXS spectra shown in [Fig. 1](#) of the main text were measured with π -polarized incident photons. Due to the strong fluorescence signal from the STO substrate, the XES of LaNiO₃ and LaNiO₂ shown in [Fig.](#)

1 were obtained from the fluorescence signal identified in RIXS incident-photon-energy-and emission-energy map across the oxygen pre-edge (incident photon energy from ~ 525 eV to ~ 530 eV). The elastic line and weak Raman-like excitations (in LaNiO_2) were removed for clarity.

The XAS and XES spectra of NiO shown in Fig. 1a were measured at beamline BL8.0 using the q-RIXS endstation of the Advanced Light Source (ALS) of the Lawrence Berkeley National Laboratory. For the RIXS/XES measurements the scattering angle was fixed to 130° and the combined instrument resolution was approximately 300 meV at the Ni L_3 -edge and approximately 200 meV at the O K -edge. The XAS at the O K -edge for NdNiO_3 and NdNiO_2 (Fig. 1a) were taken at 41A BlueMagpie beamline at Taiwan Photon Source. The XAS and RIXS map at the Ni L -edge for the NdNiO_2 were taken at I21 beamline at the Diamond Light Source. The RIXS spectrometer is set at 146° , with a resolution of approximately 50 meV. The scattering plane coincided with the crystallographic ac (bc) plane with a grazing incident angle $\sim 10^\circ$. π -polarized incident photons were used for this measurement.

All XAS at O K -edge (Fig. 1) were taken in fluorescence yield mode with a grazing incident angle of 10 and 20 degrees for the La-based and Nd-based nickelates, respectively. The grazing incident geometry is used to reduce the signal arising from the STO substrate. The spectrum is normalized such that the intensity at the pre-edge and the post-edge is 0 and 1, respectively.

All XAS at the Ni L -edge (Fig. 2) were taken in fluorescence yield mode with a normal incident geometry. These XAS are normalized such that the intensity at the pre-edge and the post-edge are 0 and 1, respectively. For the XAS of the La-based nickelates, the intense La M_4 -line centered around 850.5 eV (Supplementary Fig. 2a) was fitted by a Lorentzian peak profile and subtracted from the LaNiO_3 and LaNiO_2 XAS to correct for the overlap between the tail of the La M_4 -line and the Ni L_3 -edge. The resulting spectra are shown in Fig. 2 of the main text.

Theory calculations

For the oxygen partial density of states (PDOS) as shown in Fig. 1 e-g and the electronic structure of LaNiO₂ shown in Figure 3, LDA+U calculations were performed using the GGA method and the simplified version by Cococcioni and de Gironcoli [33], as implemented in Quantum ESPRESSO [34]. We find that an antiferromagnetic solution, with wave vector (π, π, π) , leads to the lowest energy, with a two Ni, body centered tetragonal (BCT) unit cell and corresponding Brillouin zone.

The Ni L_3 -edge RIXS calculations [Fig. 2] were performed using an exact diagonalization technique [35, 36], which accounts for the full overlap of the many-body wavefunctions. The microscopic Hamiltonian used for these calculations includes both material-specific on-site energies and hybridizations as encoded in a Wannier downfolding of the bandstructure [37] and the full set of Coulomb interactions as expressed in terms of Slater integrals. The Wannier downfolding parameters for paramagnetic LaNiO₂ (a one Ni, tetragonal unit cell), as shown in [Supplementary Table 1](#) were obtained from Wannier90 [38] for 12-orbital (O $p_x/p_y/p_z$, Ni $d_z^2/d_x^2-y^2/d_{xy}/d_{xz}/d_{yz}$, La d_z^2) Wannier downfolding was used in the Ni L -edge RIXS calculation for LaNiO₂, where the relevant parameters appear in [Supplementary Table 1](#). The Slater integrals for Ni $3d$ in the LaNiO₂ calculations were: $F^0 = 0.5719$ eV, $F^2 = 11.142$ eV, and $F^4 = 6.874$ eV. The Slater integrals for Ni $3d$ - $2p$ interactions are: $F^0_{p,d} = 0.148$ eV, $F^2_{p,d} = 6.667$ eV, $G^1_{p,d} = 4.922$ eV, and $G^3_{p,d} = 2.796$ eV. The values of F^2 , F^4 , $F^2_{p,d}$, $G^1_{p,d}$ and $G^3_{p,d}$ are taken from Ref [37]. We take 0.7 as a screening factor for the non-monopole terms. A core-level spin-orbit coupling of 12.5 eV has been used for the Ni $2p$ core electrons. The resulting weight of the Ni wave function is shown in [Supplementary Table 2](#).

The two-orbital, low energy model for the physics of LaNiO₂ appears in [Fig. 4](#). This model, obtained once again by Wannier downfolding the DFT paramagnetic solution for LaNiO₂ in the one Ni, tetragonal unit cell (the same method as that used to obtain the noninteracting part of the Hamiltonian for the LaNiO₂ RIXS calculation, but only for the two bands that cross E_F), yields the independent hopping parameters listed in [Supplementary Table 3](#), cutoff for absolute values smaller than 0.008 eV. The two Wannier orbitals are shown in [Fig. 4b](#) of the main text: (1) a very extended orbital, centered on La, with $d_{3z^2-r^2}$ character, which makes-up the majority character of the three-dimensional band; and (2) a more localized orbital, centered on Ni and primarily confined to the NiO₂ plane, with $d_{x^2-y^2}$ character, which makes-up the majority character of the quasi-two-dimensional band. Note that this paramagnetic solution in the tetragonal Brillouin zone has one large quasi-two-dimensional hole-like Fermi surface from the Ni-centered orbital and two smaller three-dimensional electron-like Fermi surfaces center at the Γ - and A-points from the La-centered orbital. The low energy, antiferromagnetic bandstructure from LDA+U [[Fig. 3](#)] would result from a (π,π,π) band-folding of the La-centered band, which moves the A-point to the Γ -point, formation of upper and lower Hubbard Ni-centered bands, gapping-out the large hole Fermi surface, and a shift in chemical potential to compensate for the loss of carriers, which leaves a single electron pocket at the Γ -point.

The non-interacting bands of the effective low energy model can be written in tight-binding form as

$$\begin{aligned}
\varepsilon_k^R = \varepsilon_0^R &+ 2 t_{[0,0,1]}^R \cos(k_z) + 2 t_{[0,0,2]}^R \cos(2 k_z) + 2 t_{[0,0,3]}^R \cos(3 k_z) \\
&+ 2 t_{[1,0,0]}^R [\cos(k_x) + \cos(k_y)] \\
&+ 4 t_{[1,0,1]}^R [\cos(k_x) + \cos(k_y)] \cos(k_z) \\
&+ 4 t_{[1,0,2]}^R [\cos(k_x) + \cos(k_y)] \cos(2 k_z) \\
&+ 4 t_{[1,1,0]}^R \cos(k_x) \cos(k_y) \\
&+ 8 t_{[1,1,1]}^R \cos(k_x) \cos(k_y) \cos(k_z) \\
&+ 8 t_{[1,1,2]}^R \cos(k_x) \cos(k_y) \cos(2 k_z) \\
&+ 8 t_{[1,1,3]}^R \cos(k_x) \cos(k_y) \cos(3 k_z) \\
&+ 4 t_{[2,0,1]}^R [\cos(2 k_x) + \cos(2 k_y)] \cos(k_z) \\
&+ 8 t_{[2,1,1]}^R [\cos(2 k_x) \cos(k_y) + \cos(k_x) \cos(2 k_y)] \cos(k_z)
\end{aligned}$$

$$\begin{aligned}
\varepsilon_k^{Ni} = \varepsilon_0^{Ni} &+ 2 t_{[1,0,0]}^{Ni} [\cos(k_x) + \cos(k_y)] \\
&+ 4 t_{[1,1,0]}^{Ni} \cos(k_x) \cos(k_y) \\
&+ 2 t_{[2,0,0]}^{Ni} [\cos(2 k_x) + \cos(2 k_y)] \\
&+ 2 t_{[0,0,1]}^{Ni} \cos(k_z) \\
&+ 8 t_{[1,1,1]}^{Ni} \cos(k_x) \cos(k_y) \cos(k_z)
\end{aligned}$$

where the appropriate matrix elements can be found in [Supplementary Table 3](#).

References:

1. Anisimov, V. I., Bukhvalov, D. & Rice, T. M. Electronic structure of possible nickelate analogs to the cuprates. *Phys. Rev. B* **59**, 7901-7906 (1999).
2. Lee, K.-W. & Pickett, W. E. Infinite-layer LaNiO₂: Ni¹⁺ is not Cu²⁺. *Phy. Rev. B* **70**, 165109 (2004).
3. Botana, A. S., Pardo, V. & Norman, M. R. Electron doped layered nickelates: Spanning the phase diagram of the cuprates. *Phys. Rev. Materials* **1**, 021801(R) (2017).
4. Chaloupka, J. & Khaliullin, G. Orbital Order and Possible Superconductivity in LaNiO₃/LaMO₃ Superlattices. *Phys. Rev. Lett.* **100**, 016404 (2008).
5. Hansmann, P. et al. Turning a Nickelate Fermi Surface into a Cuprate-like One through Heterostructuring. *Phys. Rev. Lett.* **103**, 016401 (2009).
6. Middey, S. et al. Physics of Ultrathin Films and Heterostructures of Rare-Earth Nickelates. *Annu. Rev. Mater. Res.* **46**, 305-334 (2016).
7. Boris, A. V. et al. Dimensionality Control of Electronic Phase Transitions in Nickel-Oxide Superlattices. *Science* **332**, 937-940 (2011).
8. Benckiser, E. et al. Orbital reflectometry of oxide heterostructures. *Nat. Mater.* **10**, 189-193 (2011).
9. Disa, A. S. et al. Orbital Engineering in Symmetry-Breaking Polar Heterostructures. *Phys. Rev. Lett.* **114**, 026801 (2015).
10. Zhang, J. et al. Large orbital polarization in a metallic square-planar nickelate. *Nat. Phys.* **13**, 864-869 (2017).
11. Li, D. et al. Superconductivity in an infinite-layer nickelate. *Nature* **572**, 624-627 (2019).
12. Sawatzky, G. A. Superconductivity seen in a nickel oxide. *Nature* **572**, 592-593 (2019).

13. Fisk, Z., Ott, H. R., Rice, T. M. & Smith, J. L. Heavy-electron metals. *Nature* **320**, 124-129 (1986).
14. Coles, B. R. Heavy-fermion intermetallic compounds. *Contemp. Phys.* **28**, 143-157 (1987).
15. Stewart, G. R. Non-Fermi-liquid behavior in *d*- and *f*-electron metals. *Rev. Mod. Phys.* **73**, 797-855 (2001).
16. Mott, N. F. & Peierls, R. Discussion of the paper by de Boer and Verwey. *Proc. Phys. Soc.* **49**, 72-73 (1937).
17. Lee, P. A., Nagaosa, N. & Wen, X.-G. Doping a Mott insulator: Physics of high-temperature superconductivity. *Rev. Mod. Phys.* **78**, 17-85 (2006).
18. Crespin, M. O., Isnard, O., F. Dubois, F., Choisnet, J. & Odier, P. LaNiO₂: Synthesis and structural characterization. *J. Solid State Chem.* **178**, 1326-1334 (2005).
19. Hayward, M. A., Green, M. A., Rosseinsky, M. J. & Sloan, J. Sodium hydride as a powerful reducing agent for topotactic oxide deintercalation: Synthesis and characterization of the nickel(I) oxide LaNiO₂. *J. Am. Chem. Soc.* **121**, 8843-8854 (1999).
20. Ikeda, A., Krockenberger, Y., Irie, H., Naito, M. & Yamamoto, H. Direct observation of infinite NiO₂ planes in LaNiO₂ films. *Appl. Phys. Express* **9**, 061101 (2016).
21. Zaanen, J., Sawatzky, G. A. & Allen, J. W. Band Gaps and Electronic Structure of Transition-Metal Compounds. *Phys. Rev. Lett.* **55**, 418 (1985).
22. Kuiper, P., Kruizinga, G., Ghijsen, J. & Sawatzky, G. A. Character of Holes in Li_xNi_{1-x}O and Their Magnetic Behavior. *Phys. Rev. Lett.* **62**, 221 (1989).
23. Kurmaev, E. Z. et al. Oxygen x-ray emission and absorption spectra as a probe of the electronic structure of strongly correlated oxides. *Phys. Rev. B* **77**, 165127 (2008).
24. Bisogni, V. et al. Ground-state oxygen holes and the metal-insulator transition in the negative charge-transfer rare-earth nickelates. *Nat. Commun.* **7**, 13017 (2016).

25. Zaanen, J. & Sawatzky, G. A. Systematics in band gaps and optical spectra of 3D transition metal compounds. *J. Solid State Chem.* **88**, 8-27 (1990).
26. Chen, C. T. et al. Electronic states in $\text{La}_{2-x}\text{Sr}_x\text{CuO}_{4+\delta}$ probed by soft-x-ray absorption. *Phys. Rev. Lett.* **66**, 104 (1991).
27. Grioni, M. et al. Studies of copper valence states with Cu *L3* x-ray-absorption spectroscopy. *Phys. Rev. B* **39**, 1541 (1989).
28. Anisimov, V. I. Zaanen, J. & Anderson, O. K. Band theory and Mott insulators: Hubbard U instead of Stoner I. *Phys. Rev. B* **44**, 943 (1991).
29. Haverkort, M. W., Zwierzycki, M. & Andersen O. K. Multiplet ligand-field theory using Wannier orbitals. *Phys. Rev. B* **85**, 165113 (2012).

Acknowledgments

This work is supported by the U.S. Department of Energy (DOE), Office of Science, Basic Energy Sciences, Materials Sciences and Engineering Division, under contract DE-AC02-76SF00515. X. F. and D. L. acknowledge partial support from the Gordon and Betty Moore Foundation's EPIQS Initiative through Grant GBMF4415. Part of the synchrotron experiments have been performed at the ADDRESS beamline of the Swiss Light Source (SLS) at the Paul Scherrer Institut (PSI). The work at PSI is supported by the Swiss National Science Foundation through the NCCR MARVEL (Research Grant 51NF40_141828) and the Sinergia network Mott Physics Beyond the Heisenberg Model - MPBH (Research Grant CRSII2_160765/1). Part of research was conducted at the Advanced Light Source, which is a DOE Office of Science User Facility under contract DE-AC02-05CH11231. We acknowledge preliminary XAS characterization at BL13-3, SSRL by J. S. Lee in the early stage of the project.

Author contributions

W.S.L., M.H. and H.Y.H. conceived the experiment. M.H., H.L., E.P., Y.T., T.S., and W.S.L. conducted the experiment at SLS. H.L., W.S.L., Z.H. and Y.D.C. conducted the experiment at ALS, USA. H.L., W.S.L., A.N. and K.J.Z. conducted XAS measurement at Diamond Light Source, UK. M.R., W.S.L., H.H., and D.J.H. conducted XAS measurements at NSRRC, Taiwan. J.S.L. contributed to XAS characterization of samples at an early stage of the work. M.H., H.L. and W.S.L. analyzed the data. C.J.J., B.M., J. Z. and T.P.D. performed the theoretical calculations. D.L., X.F., Y.H., and H.Y.H. synthesized and characterized the nickelate samples using transport and XRD. M.H., Z.X.S. and W.S.L. prepared and aligned samples for x-ray spectroscopy measurements. M.H., B.M. J. Z. and W.S.L. wrote the manuscript with input from all authors.

Competing financial interests

The authors declare no competing financial interests.

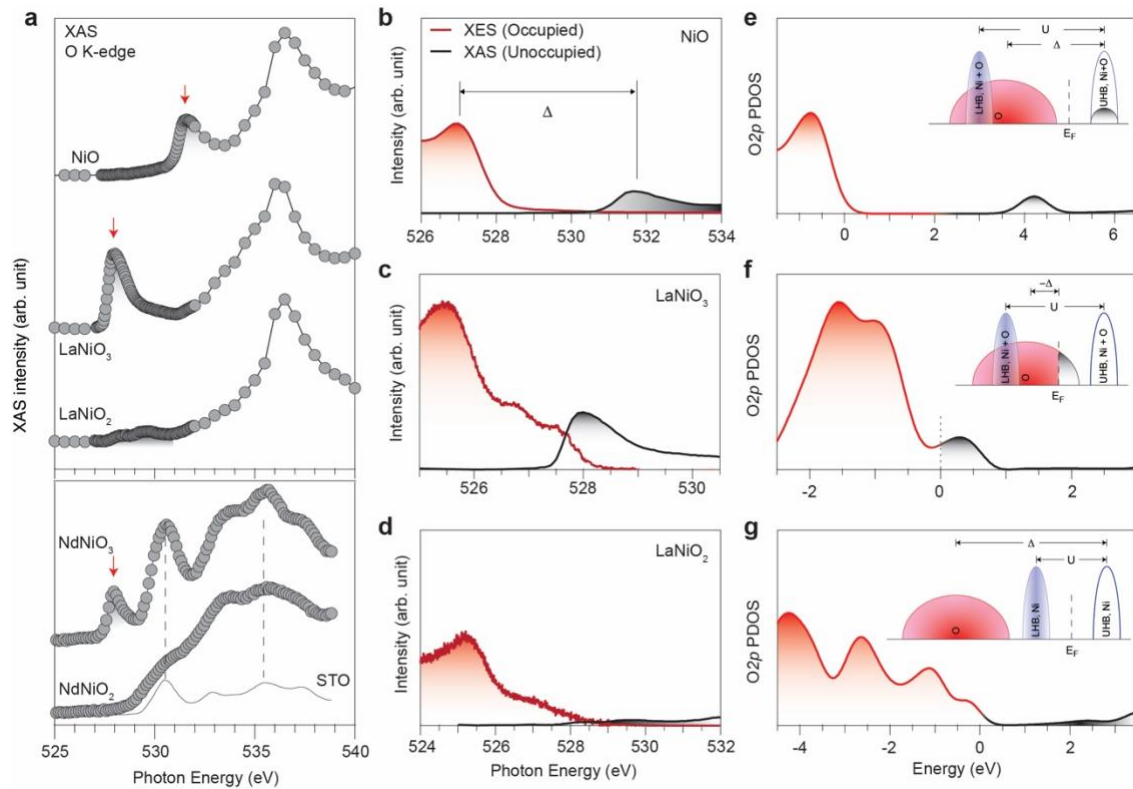


Figure 1 | X-ray spectroscopy near the O K-edge and LDA+U calculation. **a**, X-ray absorptions spectra (XAS) of NiO, LaNiO₃, and LaNiO₂. Red arrows mark the pre-edge peaks indicative of Ni-O hybridization. The lower panel shows the XAS of NdNiO₃ and NdNiO₂. Dashed vertical lines indicate features of the SrTiO₃ (STO) substrate (solid grey line) in the XAS of NdNiO₃ and NdNiO₂, due to thinner film thickness than that of the La-based films shown in the upper panel. Spectra are vertically offset for clarity. **b-d**, X-ray emission spectrum (XES) and XAS in the pre-edge region, roughly reflecting the occupied (red shading) and unoccupied (black shading) oxygen PDOS, respectively. Vertical lines illustrate the band gap projected in the oxygen density of states, corresponding to the effective charge transfer energy Δ in NiO and LaNiO₃. **e-g**: LDA+U calculations for the PDOS with O 2p orbital character. ($U=8\text{eV}$ for NiO and LaNiO₃, $U=6\text{eV}$ for LaNiO₂). Red and black shadings indicate the occupied and unoccupied oxygen PDOS, respectively. Insets are sketches of the relationship between U and Δ in the Zaanen-Sawatzky-Allen scheme for each compound.

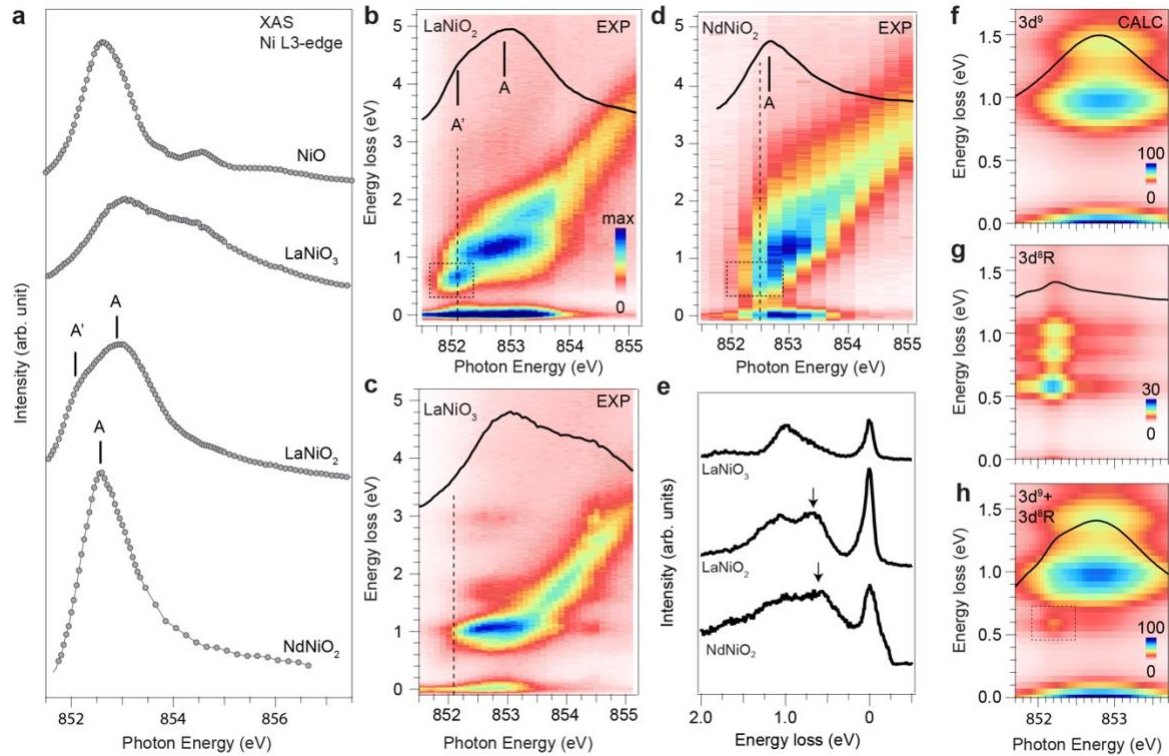


Figure 2 | X-ray absorption spectroscopy (XAS) and resonant inelastic x-ray scattering (RIXS) at the Ni L₃-edge. **a**, Normalized absorption spectra across the Ni L₃-edge of NiO, LaNiO₃, LaNiO₂, and NdNiO₂. The La M₄-line was subtracted from the LaNiO₂ and LaNiO₃ spectra (see [Supplementary Fig. 2](#)). The markers A indicate the main peak for LaNiO₂ and NdNiO₂. A' labels a lower energy shoulder in the XAS of LaNiO₂. Spectra are vertically offset for clarity. **b-d**, RIXS intensity map of LaNiO₂, LaNiO₃, and NdNiO₂ measured as a function of incident photon energy at T = 20 K. The corresponding XAS is superimposed as a solid black line in each map. The dashed box highlights the ~0.6 eV features of LaNiO₂ and NdNiO₂ that are associated with the Ni-La and Ni-Nd hybridization, respectively. **e**, RIXS energy loss spectra of LaNiO₃, LaNiO₂, and NdNiO₂ at incident energies indicated by vertical dashed lines in **b-d**. The black arrows highlight the 0.6 eV features of LaNiO₂ and NdNiO₂. **f-g**, Calculated RIXS maps and absorption spectra (solid black lines) of LaNiO₂ for a 3d⁹, 3d⁸R, and 3d⁹+3d⁸R (R denote a charge transfer to the rare-earth cation) ground state, respectively. The lightly dashed box in panel **f** highlights the same feature as the box in panel **b**.

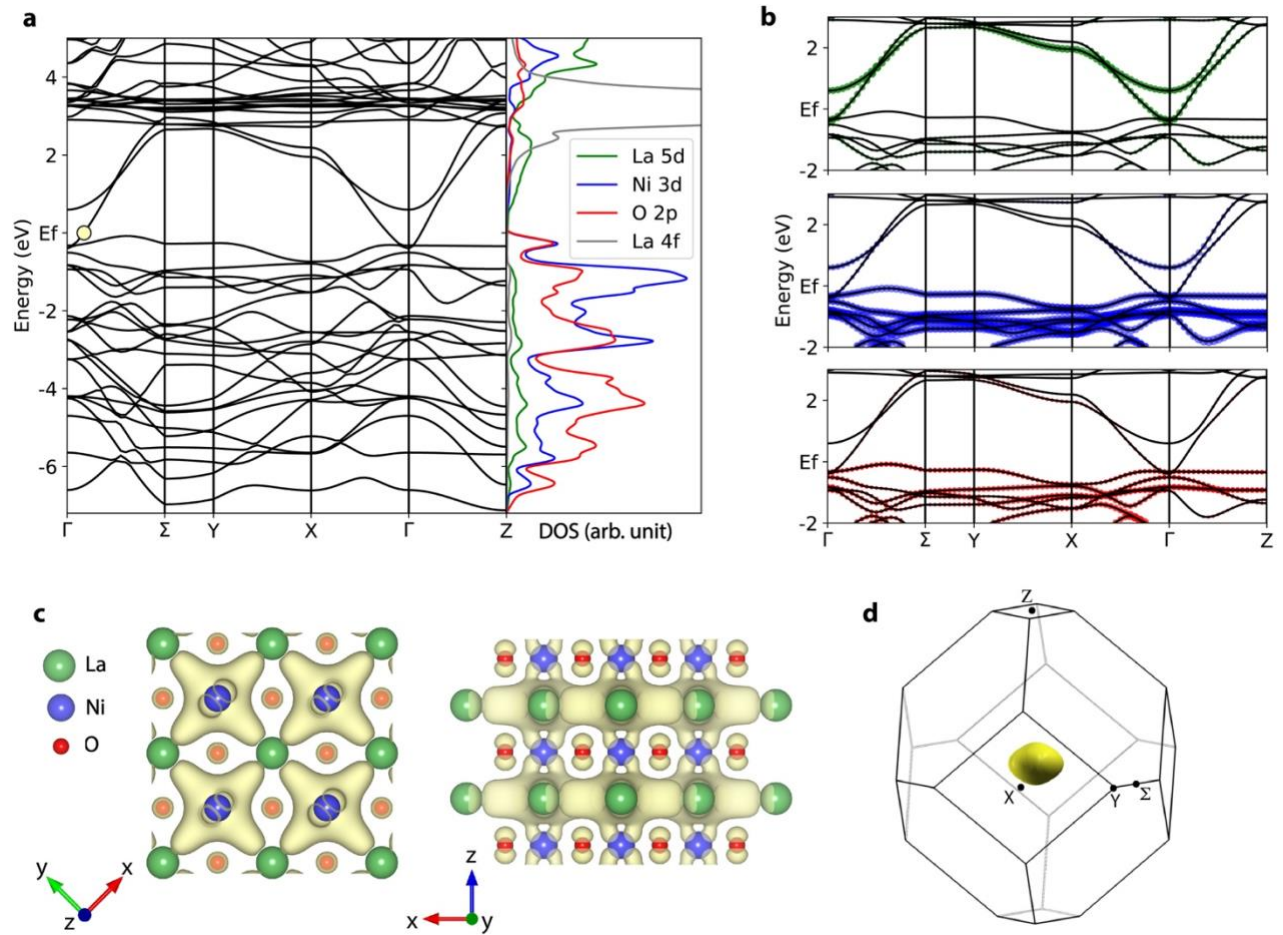


Figure 3 | Electronic structure of LaNiO₂. Theoretical calculations of the electronic structure in the LDA+U framework with $U = 6$ eV (antiferromagnetic solution). **a**, Band structure of LaNiO₂ along high symmetry directions in the body centered tetragonal (BCT) Brillouin zone (BZ). The BZ with labeled high symmetry points is also shown in **d**. The right-hand side shows the La 5d (green), Ni 3d (blue), O 2p (red) and La 4f (grey) partial density of states with a smaller energy broadening than that used in Figs. 1e-g. **b**, Orbital-projected band structure of LaNiO₂ near E_F . The color code is identical to that used in panel a, representing the projection onto orbitals with different atomic character. **c**, Top- and side-views of an electron density contour for the single-particle wavefunction at k_F along Γ - Σ (yellow marker in panel a). **d**, Fermi surface (closed electron pocket) around Γ with dominant La 5d character in the first BCT BZ with labeled high symmetry points.

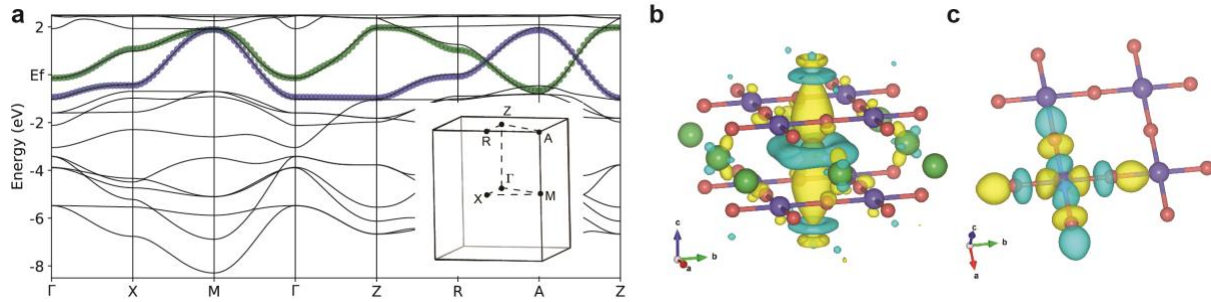


Figure 4 | Modeling the rare-earth infinite layer nickelates. **a**, Band dispersion of LaNiO_2 , highlighting two bands which cross E_F in the paramagnetic LDA calculation. The inset shows the high symmetry points in the tetragonal BZ. **b**, Isosurface plots for an extended La-centered $d_{3z^2-r^2}$ -like and essentially planar Ni-centered $d_{x^2-y^2}$ -like Wannier orbital for the minimal low-energy model of LaNiO_2 . These two orbitals produce the three-dimensional band (La, green) and quasi-two-dimensional band (Ni, blue) highlighted in **a**.

Supplementary Information
for
Electronic structure in the parent compound of superconducting
infinite layer nickelates

M. Hepting^{1,7}, D. Li¹, C. J. Jia¹, H. Lu¹, E. Paris², Y. Tseng², X. Feng¹, M. Osada¹, E. Been¹, Y. Hikita¹, Y.-D. Chuang³, Z. Hussain³, K. J. Zhou⁴, A. Nag⁴, M. Garcia-Fernandez⁴, M. Rossi¹, H. Y. Huang⁵, D. J. Huang⁵, Z. X. Shen¹, T. Schmitt², H. Y. Hwang¹, B. Moritz¹, J. Zaanen⁶, T. P. Devereaux¹, and W. S. Lee^{1*}

¹Stanford Institute for Materials and Energy Sciences, SLAC National Accelerator Laboratory and Stanford,
Menlo Park, California 94025, USA

²Photon Science Division, Swiss Light Source, Paul Scherrer Institut, CH-5232 Villigen PSI, Switzerland

³Advanced Light Source, Lawrence Berkeley National Laboratory

⁴Diamond Light Source, Harwell Science and Innovation Campus, Didcot, Oxfordshire OX11 0DE, United
Kingdom

⁵NSRRC, Hsinchu Science Park, Hsinchu 30076, Taiwan

⁶Leiden Institute of Physics, Leiden University, 2300 RA Leiden, The Netherland

Current affiliation:

⁷Max-Planck-Institute for Solid State Research, Heisenbergstraße 1, 70569 Germany

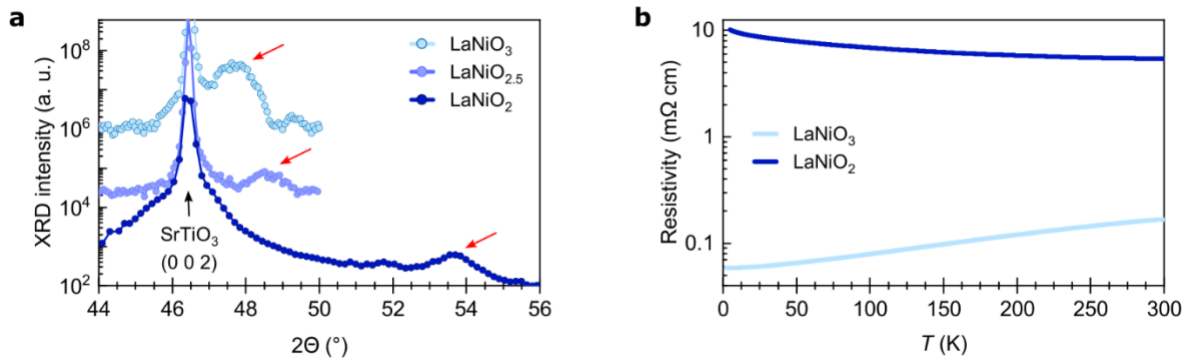
Correspondence to: leews@stanford.edu

This Supplementary Information Contains:

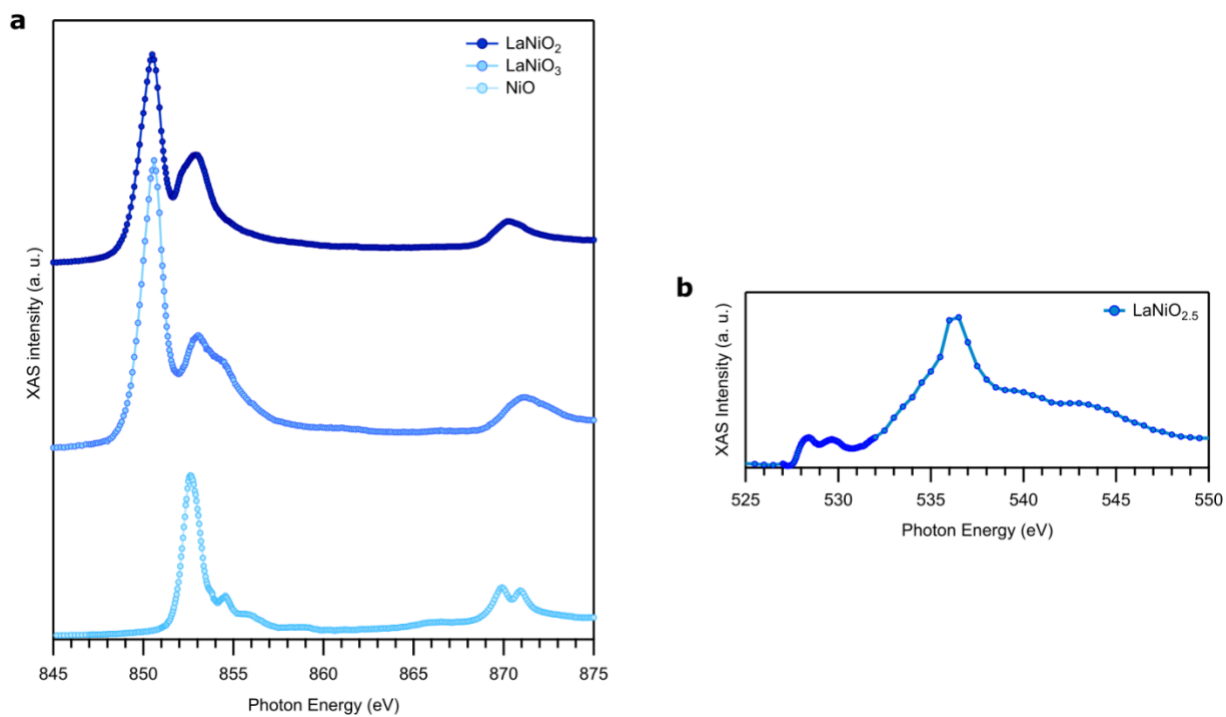
Supplementary Fig. 1-2

Supplementary Table 1-3

Supplementray Reference 30 - 38



Supplementary Figure 1 | XRD characterization and electrical transport measurements. a, XRD pattern of the LaNiO₃, LaNiO_{2.5}, and LaNiO₂ films grown on SrTiO₃ (001) substrates, measured with Cu K α radiation. The red arrows indicate the nickelate film peaks and the black arrows the (002) SrTiO₃ substrate peak. The film peak shifts to higher 2 θ values as a function of apical oxygen reduction. The curves are offset in vertical direction for clarity. **b,** Resistivity vs. temperature of the LaNiO₃ and LaNiO₂ film.



Supplementary Figure 2 | Ni L-edge and O K-edge x-ray absorption spectra (XAS). A, Normalized XAS of LaNiO₃ and LaNiO₂ across the Ni $L_{3,2}$ -edge before subtraction of the La M_4 line. The XAS of NiO powder is also shown. Spectra are offset in vertical direction for clarity. **B,** Normalized XAS of LaNiO_{2.5} across the O K-edge.

Unit eV		Ni [0,0,0]					O[0, ½, 0]			O[½, 0, 0]			La [½,½,½]
		d _{z2}	d _{x2-y2}	d _{xy}	d _{xz}	d _{yz}	p _z	p _x	p _y	p _z	p _x	p _y	d _{z2}
Ni [0,0,0]	d _{z2}	-0.04							-0.3		0.3		-0.5
	d _{x2-y2}		0.70						-1.2		-1.2		
	d _{xy}			-0.1				0.7				0.7	
	d _{xz}				0					-0.8			-0.1
	d _{yz}					0	0.8						0.1
O [0, ½, 0]	p _z					0.8	-2.32	0.2					0.4
	p _x			0.7			0.2	-2.34			0.3	0.2	-0.2
	p _y	-0.3	-1.2						-3.26		0.6	0.3	0.4
O [½, 0, 0]	p _z				-0.8					-2.32		-0.2	0.4
	p _x	0.3	-1.2					0.3	0.6		-3.25		-0.4
	p _y			0.7				0.2	0.3	-0.2		-2.35	0.2
La [½,½,½]	d _{z2}	-0.5			-0.1	0.1	0.4	-0.2	0.4	0.4	-0.4	0.2	2.42

Supplementary Table 1 Materials parameters for LaNiO₂ obtained from Wannier downfolding.

The diagonal terms represent on-site energies. The off-diagonal terms represent the hopping between two orbitals. The shaded area only shows those parameters whose absolute value is larger than 0.1. All values are in units of eV. The triplet [i,j,k] appearing next to each orbital shows its relative position within a one Ni, LaNiO₂ tetragonal unit cell along the unit a, b, and c axes, respectively, with Ni at [0,0,0].

Orbital configuration		Approximate percentage	Dipole transition spin up (minority spin)	Dipole transition spin down (majority spin)
d^9	$d^4 \uparrow d^5 \downarrow$	56%	Yes	No
d^8R	$d^3R \uparrow d^5 \downarrow$	24%	Yes	No
	$d^4 \uparrow d^4R \downarrow$	14%	Yes	Yes
d^7R^2	$d^3R \uparrow d^4R \downarrow$	6%	Yes	Yes

Supplementary Table 2: Orbital configurations and their approximate percentage shown in the many-body ground state calculation. Here, R represents a rare earth electron in the La cage surrounding each Ni, in analogy to the standard ligand hole (*L*), due to hybridization between Ni and La. This is akin to a charge transfer, but from Ni to the rare earth, instead of oxygen to transition metal more commonly encountered in oxides.

Hopping Parameters from Wannier Downfolding			
i	j	k	$t_{[i,j,k]}^R$
0	0	0	1.132
0	0	1	-0.022
0	0	2	-0.112
0	0	3	0.019
1	0	0	-0.028
1	0	1	-0.164
1	0	2	0.033
1	1	0	-0.062
1	1	1	0.024
1	1	2	0.013
1	1	3	-0.019
2	0	1	0.009
2	1	1	0.009
i	j	k	$t_{[i,j,k]}^{Ni}$
0	0	0	0.267
1	0	0	-0.355
1	1	0	0.090
2	0	0	-0.043
0	0	1	-0.043
1	1	1	0.013
i	j	k	$t_{[i,j,k]}^{R-Ni}$
2	0	0	-0.026
2	0	2	0.013

Supplementary Table 3: Tight-binding model parameters for the two-orbital model for LaNiO₂.

The elements in the table show all the independent hopping parameters with a absolute magnitude larger than 0.008 eV. The triplet of integers [i,j,k] represents hopping between unit cells with a relative separation $r = i a + j b + k c$, where a, b, and c are unit vectors in the respective directions. Here, for simplicity we take a=b=c=1 and measure all momenta k_x , k_y , and k_z , accordingly. Tetragonal symmetry dictates that for the rare earth- (R-) and Ni-band parameters, the hopping with triplet [i,j,k] would be equivalent to [-i,j,k], [i,-j,k], [i,j,-k], [-i,-j,k], [-i,j,-k], [i,-j,-k], [-i,-j,-k], and all combinations with $i \leftrightarrow j$, with a phase factor of 1 or -1 depending on the symmetry of the orbitals. The triplet [0,0,0] represents the orbital site energy ϵ_0 . The La-centered and Ni-centered Wannier orbitals have relative unit cell coordinates [0.5, 0.5, 0.5] and [0.0, 0.0, 0.0], respectively,

which only affects the equivalent triplets for the cross-orbital-hybridization terms t^{R-Ni} . The fermi energy E_F is at 0 eV.

Supplementary References

30. Kawai, M. et al. Reversible changes of epitaxial thin films from perovskite LaNiO₃ to infinite-layer structure LaNiO₂. Appl. Phys. Lett. 94, 082102 (2009).
31. Ikeda, A., Manabe, T. & Naito, M. Improved conductivity of infinite-layer LaNiO₂ thin films by metal organic decomposition. Physica C 495, 134-140 (2013).
32. Strocov, V. N. et al. High-resolution soft X-ray beamline ADDRESS at the Swiss Light Source for resonant inelastic X-ray scattering and angle-resolved photoelectron spectroscopies. J. Synchrotron Radiat. 17, 631–643 (2010).
33. Cococcioni, M. and de Gironcoli, S., Linear response approach to the calculation of the effective interaction parameters in the LDA+U method, Physical Review B 71, 035105 (2005)
34. Giannozzi, P. et al. QUANTUM ESPRESSO: a modular and open-source software project for quantum simulations of materials. Journal of Physics: Condensed Matter 21, 395502 (2009).
35. Jia, C. J. et al. Persistent spin excitations in doped antiferromagnets revealed by resonant inelastic light scattering. Nature Communications 5, 3314 (2014).
36. Jia, C., Wohlfeld, K., Wang, Y., Moritz, B. & Devereaux, T. P. Using RIXS to Uncover Elementary Charge and Spin Excitations. Physical Review X 6, 021020 (2016).
37. Haverkort, M. W., Zwierzycki, M. & Andersen, O. K. Multiplet ligand-field theory using Wannier orbitals. Physical Review B 85, 165113 (2012).
38. Mostofi, A. A. et al. An updated version of wannier90: A tool for obtaining maximally-localised Wannier functions. Computer Physics Communications 185, 2309-2310 (2014).