

Catalysis

Enantioselective Nickel-Catalyzed *anti*-Arylmethylative Cyclizations onto Acyclic KetonesHarley Green,^[a, b] Stephen P. Argent,^[b] and Hon Wai Lam^{*[a, b]}

Abstract: Domino reactions involving nickel-catalyzed additions of (hetero)arylboronic acids to alkynes, followed by cyclization of the alkenylnickel intermediates onto tethered acyclic ketones to give chiral tertiary-alcohol-containing products in high enantioselectivities, are described. The reversible *E/Z* isomerization of the alkenylnickel intermediates enables overall *anti*-arylmethylative cyclization to occur. The ring system of the products are substructures of certain diarylindolizidine alkaloids.

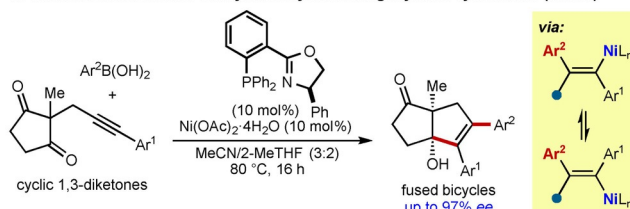
Stereogenic cyclic tertiary alcohols are important structural units that feature prominently in biologically active natural products and therapeutic compounds. Accordingly, new methods for the enantioselective construction of these units provide valuable tools for target synthesis. Of the strategies available, the catalytic asymmetric addition of carbon nucleophiles to ketones ranks highly in directness, versatility, and overall synthetic efficiency.^[1] One subset of these reactions are metal-catalyzed domino sequences initiated by the addition of an arylboron reagent to an alkyne, followed by enantioselective cyclization of the resulting alkenylmetal species onto a tethered ketone, which are applicable to the synthesis of diverse carbo- and heterocycles.^[2] Recently, nickel-catalyzed variants of these reactions have been developed in which reversible *E/Z* isomerization of the alkenylnickel intermediates is essential for cyclization.^[3] Application of this general method to achiral products has also been reported,^[4] and other related nickel-catalyzed processes have also appeared.^[5–7] In addition to nickel being much less expensive than comparable rhodium- or palladium-catalyzed reactions,^[2] the diverse reactivity of nickel catalysis^[5]

often enables unique transformations not available to other metal catalysts.

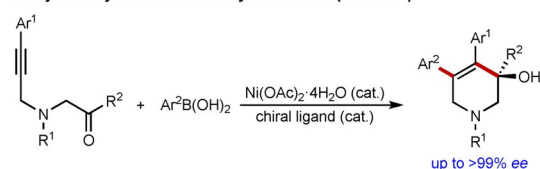
Our first contribution to this field included a study of enantioselective nickel-catalyzed desymmetrizations of cyclic 1,3-diketones, which give fused bicycles in high diastereo- and enantioselectivities (Scheme 1 A).^[3a] Although effective, the ability to use acyclic ketones in non-desymmetrizing cyclizations would also be valuable to significantly broaden the substrate scope and provide simpler, non-fused products. However, acyclic ketones are potentially less reactive than cyclic 1,3-diketones because of their greater conformational flexibility, and because they lack the activation from the second ketone through its electron-withdrawing effect as well as the electronic repulsion caused by having aligned dipoles. Two individual examples of non-symmetric nickel-catalyzed arylative cyclizations onto acyclic ketones have been reported recently,^[6d] but to our knowledge, corresponding enantioselective processes have yet to be described.

Herein, we describe the successful use of acyclic dialkyl and alkyl-aryl ketones in these reactions in the enantioselective preparation of aza- and carbocyclic tertiary alcohols

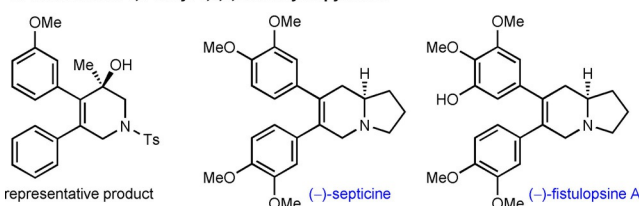
A. Enantioselective nickel-catalyzed desymmetrizing arylative cyclizations (ref. 3a)



B. Arylative cyclizations onto acyclic ketones (this work)



C. Relevance of 4,5-diaryl-1,2,3,6-tetrahydropyridines



Scheme 1. Enantioselective nickel-catalyzed arylative cyclizations onto ketones.

[a] H. Green, Prof. H. W. Lam
The GlaxoSmithKline Carbon Neutral Laboratories for Sustainable Chemistry, University of Nottingham Jubilee Campus, Triumph Road, Nottingham, NG7 2TU (UK)
E-mail: hon.lam@nottingham.ac.uk
Homepage: <http://www.nottingham.ac.uk/~pczh/>

[b] H. Green, Dr. S. P. Argent, Prof. H. W. Lam
School of Chemistry, University of Nottingham University Park, Nottingham, NG7 2RD (UK)

Supporting information and the ORCID identification numbers for the authors of this article can be found under: <https://doi.org/10.1002/chem.202100143>.

© 2021 The Authors. Published by Wiley-VCH GmbH. This is an open access article under the terms of the Creative Commons Attribution License, which permits use, distribution and reproduction in any medium, provided the original work is properly cited.

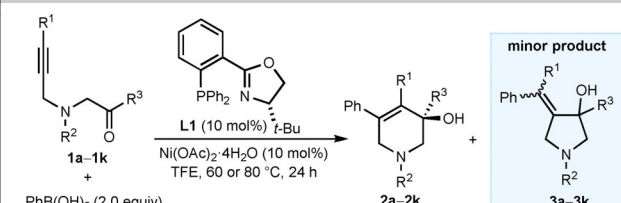
(Scheme 1B). Most of the products contain the 4,5-diaryl-1,2,3,6-tetrahydropyridine ring system, which appears in indolizidine natural products such as (–)-phylosteminine,^[8] (–)-septicine,^[9] and (–)-fistulopsine A^[10] (Scheme 1C).

This investigation began with the reaction of PhB(OH)₂ with acyclic substrates **1**, which contain an alkyne tethered to a ketone through a sulfonamide (Table 1). Chiral phosphine-oxazoline (PHOX) ligands have proven to be excellent ligands in related studies^[3] and we found (*S*)-*t*-Bu-PHOX (**L1**) to be highly effective in these reactions. Heating a mixture of the substrate **1** and PhB(OH)₂ (2.0 equiv) in the presence of 10 mol% each of Ni(OAc)₂·4H₂O and **L1** in TFE (2,2,2-trifluoroethanol) at 60 or 80 °C for 24 h provided azacycles **2a–2k** in 44–90% yield and

up to 99% *ee*.^[11] Small quantities of minor arylative cyclization products **3**, resulting from initial phenylnickelation of the alkyne with the regioselectivity opposite to that required for the formation of the major products **2**, were also observed by ¹H NMR spectroscopy, but with the exception of the reaction producing **2k** and **3k**, these were not isolated. A range of aromatic ketones are tolerated in these reactions, with substrates containing phenyl (**2a**), 4-chlorophenyl (**2b**), (3-trifluoromethyl)phenyl (**2c**), or 2-methylphenyl ketones (**2d**) readily undergoing arylative cyclization. Simple dialkyl ketones are also competent electrophiles, with ketones containing methyl (**2e**, **2f**, **2i**, and **2j**), ethyl (**2k**), isopropyl (**2g**), or methyl propanoate (**2h**) groups reacting successfully. Regarding the alkynyl substituent, the process is tolerant of phenyl (**2a–2h**), (4-carbomethoxy)phenyl (**2i**), and 3-methoxyphenyl groups (**2j**). A substrate with a vinyl group on the alkyne also reacted smoothly, but the enantiomeric excess of the resulting product **2k** (45% *ee*) was lower than in the other cases. Replacing the *para*-toluenesulfonamide with a 4-nitrophenylsulfonamide group is also possible (**2f**), which has implications for subsequent product manipulation because 4-nitrophenylsulfonyl groups are more readily deprotected than tosyl groups.

Next, different boronic acids were investigated in reactions with substrate **1f**, and we were pleased to observe that arylative cyclization products **2i–2p** were obtained in up to 79% yield and uniformly high enantioselectivities (98% to >99% *ee*, Table 2). Various substituted phenylboronic acids are compatible with this process including, notably, 3-hydroxyphenylboronic acid (**2m**). 2-Naphthylboronic acid (**2o**) and 3-thienylboronic acid (**2p**) also readily underwent the reaction.

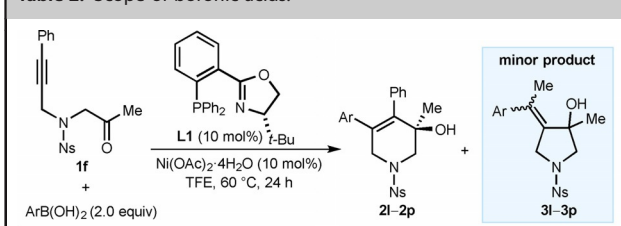
Table 1. Scope of alkynyl-tethered ketones.^[a]



2a 84% (11:1), 94% <i>ee</i> ^[b]	2b 78% (16:1), 98% <i>ee</i> ^[c]	2c 76% (16:1), 98% <i>ee</i> ^[b]
2d 72% (10:1), >99% <i>ee</i> ^[b]	2e 70% (16:1), 99% <i>ee</i> ^[c]	2f 86% (>19:1), 99% <i>ee</i> ^[b]
2g 80% (12:1), 97% <i>ee</i> ^[b]	2h 56% (13:1), 99% <i>ee</i> ^[c]	2i 44% (8:1), 98% <i>ee</i> ^[b]
2j 46% (14:1), 98% <i>ee</i> ^[c]	2k 90% (12:1), 45% <i>ee</i> ^[c,d]	

[a] Reactions were conducted using 0.30 mmol of **1** in TFE (3 mL). Yields are of isolated products. Values in parentheses refer to the ratio of **2:3** as determined by ¹H NMR analysis of the crude reactions. Unless stated otherwise, the minor isomers **3** were not evident in the isolated products. Enantiomeric excesses were determined by HPLC analysis on a chiral stationary phase. [b] At 80 °C. [c] At 60 °C. [d] Product **2k** was obtained as an inseparable 12:1 mixture together with the minor product **3k** in 90% combined yield.

Table 2. Scope of boronic acids.^[a]

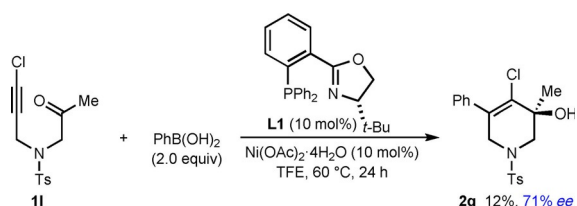


2i 78% (>19:1), 99% <i>ee</i>	2m 67% (16:1), >99% <i>ee</i>	2n 79% (19:1), >99% <i>ee</i>
2o 51% (>19:1), >99% <i>ee</i>	2p 64% (>19:1), 98% <i>ee</i>	

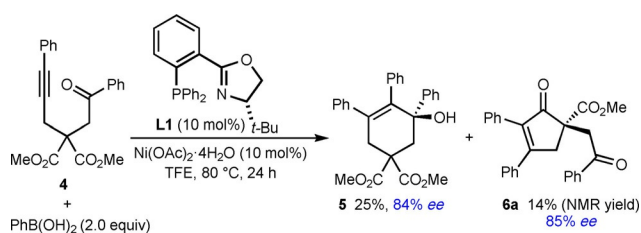
[a] Reactions were conducted using 0.30 mmol of **1f** in TFE (3 mL). Yields are of isolated products. Values in parentheses refer to the ratio of **2:3** as determined by ¹H NMR analysis of the crude reactions. Unless stated otherwise, the minor isomers **3** were not evident in the isolated products. Enantiomeric excesses were determined by HPLC analysis on a chiral stationary phase.

Further experiments to explore the scope of this process in the reactions of various other substrates with PhB(OH)_2 are shown in Schemes 2–6. Changing the alkynyl substituent to a chloride was only moderately successful; substrate **11** reacted to give chloroalkene-containing tetrahydropyridine **2q** in 12% yield and 71% *ee*, with the remainder of the material being predominantly unreacted **11** (Scheme 2). Next, the preparation of carbocyclic products was attempted by changing the sulfonamide connecting the alkyne and the ketone to a malonyl group. Interestingly, the reaction of substrate **4** did give the six-membered product **5** in 25% yield and 84% *ee*, but the cyclopent-2-enone **6a** resulting from cyclization of the intermediate alkenylnickel species onto one of the ester groups was also obtained in 14% yield^[12] and 85% *ee* (Scheme 3). The enantioselective formation of cyclopent-2-enones in this manner was described by our group previously,^[3c] and in this case, it appears that despite the lower electrophilicity of the methyl esters in **4** compared with the phenyl ketone, the kinetic preference to form a five-membered ring over a six-membered ring makes the formation of **6a** competitive with **5**. An attempt to prepare a five-membered cycloalkanol by the reaction of PhB(OH)_2 with substrate **7** was unsuccessful, and gave a complex mixture of unidentified products (Scheme 4). However, it should be noted that the formation of five-membered rings by *anti*-carbometallative cyclizations onto cyclic 1,3-diketones was successful in our previous work (see Scheme 1 A).^[3a]

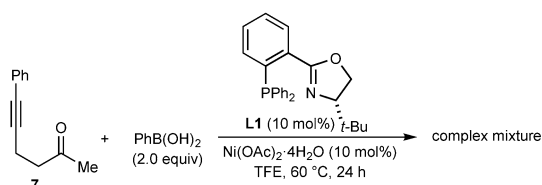
Attempts to form seven-membered ring products are shown in Schemes 5 and 6. Substrate **8** reacted with PhB(OH)_2 to give



Scheme 2. Reaction of chloroalkyne **11**.



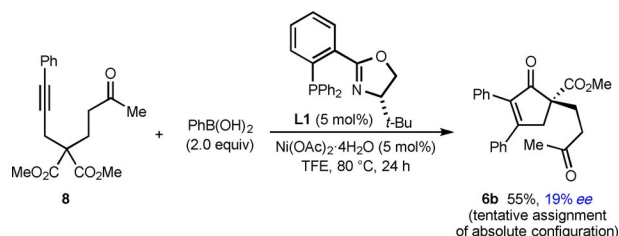
Scheme 3. Formation of carbocyclic products **5** and **6a**.



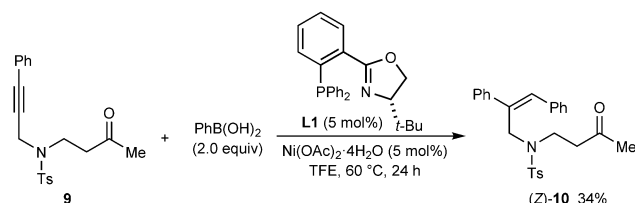
Scheme 4. Attempted reaction of substrate **7**.

cyclopent-2-enone **6b** in 55% yield and 19% *ee*^[13] but no product resulting from cyclization onto the ketone was observed (Scheme 5). In addition, the reaction of substrate **9** with PhB(OH)_2 led to the trisubstituted alkene (*Z*)-**10** resulting from alkyne hydroarylation as the only isolable product in 34% yield (Scheme 6). The identity of the remainder of the material in this reaction was not clear; although this did not appear to contain an appreciable quantity of the corresponding *E*-isomer of **10**, we cannot rule out its presence resulting from *E/Z* isomerization of the alkenylnickel intermediate.

In summary, we have described enantioselective nickel-catalyzed *anti*-arylmetallative cyclizations of (hetero)arylboronic acids with substrates containing an alkyne tethered to an acyclic ketone, which proceed to give chiral tertiary alcohols with high enantioselectivities in most cases (often $\geq 99\%$ *ee*). Compared with a previous study,^[3a] this work demonstrates a substantial increase in scope of ketones that can be used as electrophiles. The products are 4,5-diaryl-1,2,3,6-tetrahydropyridines, a ring system that is seen in certain indolizidine alkaloids. The formation of carbocyclic products is also possible.^[14]



Scheme 5. Attempted formation of a seven-membered carbocycle.



Scheme 6. Attempted formation of a seven-membered azacycle.

Acknowledgements

This work was supported by the Engineering and Physical Sciences Research Council and AstraZeneca [Industrial CASE Studentship, grant number EP/S513854/1]; the University of Nottingham; and GlaxoSmithKline.

Conflict of interest

The authors declare no conflict of interest.

Keywords: asymmetric catalysis · cyclization · isomerization · ketones · nickel

- [1] For selected, general reviews on the enantioselective synthesis of tertiary alcohols via nucleophilic additions to ketones, see: a) M. Hatano, K. Ishihara, *Synthesis* **2008**, 1647–1675; b) M. Shibasaki, M. Kanai, *Chem. Rev.* **2008**, *108*, 2853–2873; c) S. Adachi, T. Harada, *Eur. J. Org. Chem.* **2009**, 3661–3671; d) J. F. Collados, R. Solà, S. R. Harutyunyan, B. Macià, *ACS Catal.* **2016**, *6*, 1952–1970; e) J. Rong, T. Pellegrini, S. R. Harutyunyan, *Chem. Eur. J.* **2016**, *22*, 3558–3570; f) Y. Yamashita, T. Yasukawa, W.-J. Yoo, T. Kitanosono, S. Kobayashi, *Chem. Soc. Rev.* **2018**, *47*, 4388–4480; g) Y.-L. Liu, X.-T. Lin, *Adv. Synth. Catal.* **2019**, *361*, 876–918.
- [2] For representative examples, see: a) R. Shintani, K. Okamoto, Y. Otsu, K. Ueyama, T. Hayashi, *J. Am. Chem. Soc.* **2005**, *127*, 54–55; b) R. Shintani, A. Tsurusaki, K. Okamoto, T. Hayashi, *Angew. Chem. Int. Ed.* **2005**, *44*, 3909–3912; *Angew. Chem.* **2005**, *117*, 3977–3980; c) J. Song, Q. Shen, F. Xu, X. Lu, *Org. Lett.* **2007**, *9*, 2947–2950; d) Y. Li, M.-H. Xu, *Org. Lett.* **2014**, *16*, 2712–2715; e) A. Groves, J. Sun, H. R. I. Parke, M. Callingham, S. P. Argent, L. J. Taylor, H. W. Lam, *Chem. Sci.* **2020**, *11*, 2759–2764; f) A. Selmani, S. Darses, *Org. Lett.* **2020**, *22*, 2681–2686. g) For examples that use aldehydes as the electrophiles instead of ketones, see: X. Han, X. Lu, *Org. Lett.* **2010**, *12*, 108–111.
- [3] a) C. Clarke, C. A. Incerti Pradillos, H. W. Lam, *J. Am. Chem. Soc.* **2016**, *138*, 8068–8071; b) C. Yap, G. M. J. Lenagh-Snow, S. N. Karad, W. Lewis, L. J. Diorazio, H. W. Lam, *Angew. Chem. Int. Ed.* **2017**, *56*, 8216–8220; *Angew. Chem.* **2017**, *129*, 8328–8332; c) S. N. Karad, H. Panchal, C. Clarke, W. Lewis, H. W. Lam, *Angew. Chem. Int. Ed.* **2018**, *57*, 9122–9125; *Angew. Chem.* **2018**, *130*, 9260–9263; d) Z. Lu, X.-D. Hu, H. Zhang, X.-W. Zhang, J. Cai, M. Usman, H. Cong, W.-B. Liu, *J. Am. Chem. Soc.* **2020**, *142*, 7328–7333.
- [4] a) X. Zhang, X. Xie, Y. Liu, *Chem. Sci.* **2016**, *7*, 5815–5820; b) G. R. Kumar, R. Kumar, M. Rajesh, M. S. Reddy, *Chem. Commun.* **2018**, *54*, 759–762; c) S. M. Gillbard, C.-H. Chung, S. N. Karad, H. Panchal, W. Lewis, H. W. Lam, *Chem. Commun.* **2018**, *54*, 11769–11772.
- [5] For reviews on nickel-catalyzed difunctionalization of alkynes, see: a) S. E. Bottcher, L. E. Hutchinson, D. J. Wilger, *Synthesis* **2020**, *52*, 2807–2820; b) W. Liu, W. Kong, *Org. Chem. Front.* **2020**, *7*, 3941–3955.
- [6] a) M. Hari Babu, G. Ranjith Kumar, R. Kant, M. Sridhar Reddy, *Chem. Commun.* **2017**, *53*, 3894–3897; b) M. Rajesh, M. K. R. Singam, S. Puri, S. Balasubramanian, M. Sridhar Reddy, *J. Org. Chem.* **2018**, *83*, 15361–15371; c) N. Iqbal, N. Iqbal, D. Maiti, E. J. Cho, *Angew. Chem. Int. Ed.* **2019**, *58*, 15808–15812; *Angew. Chem.* **2019**, *131*, 15955–15959; d) M. K. R. Singam, A. Nagireddy, M. Rajesh, V. Ganesh, M. S. Reddy, *Org. Chem. Front.* **2020**, *7*, 30–34; e) J. Chen, Y. Wang, Z. Ding, W. Kong, *Nat. Commun.* **2020**, *11*, 1882; f) Z. Zhou, W. Liu, W. Kong, *Org. Lett.* **2020**, *22*, 6982–6987; g) Z. Zhou, J. Chen, H. Chen, W. Kong, *Chem. Sci.* **2020**, *11*, 10204–10211; h) Z. Ding, Y. Wang, W. Liu, Y. Chen, W. Kong, *J. Am. Chem. Soc.* **2021**, *143*, 53–59.
- [7] T. Igarashi, S. Arai, A. Nishida, *J. Org. Chem.* **2013**, *78*, 4366–4372.
- [8] A. Cave, M. Leboeuf, H. Moskowitz, A. Ranaivo, I. R. C. Bick, W. Sinchai, M. Nieto, T. Sevenet, P. Cabalion, *Aust. J. Chem.* **1989**, *42*, 2243–2263.
- [9] For examples of the synthesis of (–)-septicine, see: a) D. L. Comins, X. Chen, L. A. Morgan, *J. Org. Chem.* **1997**, *62*, 7435–7438; b) S. Hanessian, A. K. Chattopadhyay, *Org. Lett.* **2014**, *16*, 232–235.
- [10] a) V. A. Yap, M. E. Qazzaz, V. J. Raja, T. D. Bradshaw, H.-S. Loh, K.-S. Sim, K.-T. Yong, Y.-Y. Low, K.-H. Lim, *Phytochem. Lett.* **2016**, *15*, 136–141; b) N. V. G. Moorthy, S. V. Pansare, *Tetrahedron* **2018**, *74*, 1422–1429.
- [11] The absolute configurations of products **2 f** and **2 j** were determined by X-ray crystallography, and those of the remaining products were assigned by analogy. Deposition number 2040015 contains the supplementary crystallographic data for this paper. These data are provided free of charge by the joint Cambridge Crystallographic Data Centre and Fachinformationszentrum Karlsruhe Access Structures service.
- [12] Cyclopenten-2-one **6 a** was accompanied by inseparable, unidentified impurities, and the yield was therefore determined by ¹H NMR analysis using an internal standard.
- [13] Because of the low ee value of **6 b**, the assignment of its absolute configuration, which was made by analogy to the results reported in ref. [3c], is tentative.
- [14] The research data associated with this publication can be found at: <https://doi.org/10.17639/nott.7098>.

Manuscript received: January 13, 2021

Accepted manuscript online: February 3, 2021

Version of record online: March 5, 2021