

# Northumbria Research Link

Citation: Ting, Matthew Zhi Yeon, Wong, Kwong Soon, Rahman, Muhammad and Meheron, Selowara Joo (2021) Deterioration of marine concrete exposed to wetting-drying action. Journal of Cleaner Production, 278. p. 123383. ISSN 0959-6526

Published by: Elsevier

URL: <https://doi.org/10.1016/j.jclepro.2020.123383>  
<<https://doi.org/10.1016/j.jclepro.2020.123383>>

This version was downloaded from Northumbria Research Link:  
<http://nrl.northumbria.ac.uk/id/eprint/45451/>

Northumbria University has developed Northumbria Research Link (NRL) to enable users to access the University's research output. Copyright © and moral rights for items on NRL are retained by the individual author(s) and/or other copyright owners. Single copies of full items can be reproduced, displayed or performed, and given to third parties in any format or medium for personal research or study, educational, or not-for-profit purposes without prior permission or charge, provided the authors, title and full bibliographic details are given, as well as a hyperlink and/or URL to the original metadata page. The content must not be changed in any way. Full items must not be sold commercially in any format or medium without formal permission of the copyright holder. The full policy is available online: <http://nrl.northumbria.ac.uk/policies.html>

This document may differ from the final, published version of the research and has been made available online in accordance with publisher policies. To read and/or cite from the published version of the research, please visit the publisher's website (a subscription may be required.)

1 Deterioration of marine concrete exposed to wetting-drying  
2 action

3 Matthew Zhi Yeon Ting;  
4 Kwong Soon Wong;  
5 Muhammad Ekhlaur Rahman\*;  
6 Selowara Joo Meheron

7  
8 \*Faculty of Engineering & Science  
9 Curtin University, Malaysia  
10 E-mail: [merahman@curtin.edu.my](mailto:merahman@curtin.edu.my)

11

12 **Matthew Zhi Yeon, Ting**

13 Department of Civil and Construction Engineering Faculty of Engineering and  
14 Science, Curtin University, Malaysia, CDT 250, 98009 Miri, Sarawak, Malaysia  
15 E-mail: [matthew.ting@postgrad.curtin.edu.my](mailto:matthew.ting@postgrad.curtin.edu.my)

16

17 **Kwong Soon, Wong**

18 Department of Civil and Construction Engineering Faculty of Engineering and  
19 Science, Curtin University, Malaysia, CDT 250, 98009 Miri, Sarawak, Malaysia  
20 E-mail: [wongkwongsoon@curtin.edu.my](mailto:wongkwongsoon@curtin.edu.my)

21

22 **Muhammad Ekhlaur, Rahman (Corresponding author)**

23 Department of Civil and Construction Engineering Faculty of Engineering and  
24 Science, Curtin University, Malaysia, CDT 250, 98009 Miri, Sarawak, Malaysia  
25 E-mail: [merahman@curtin.edu.my](mailto:merahman@curtin.edu.my)

26

27 **Selowara Joo, Meheron**

28 Department of Civil and Construction Engineering Faculty of Engineering and  
29 Science, Curtin University, Malaysia, CDT 250, 98009 Miri, Sarawak, Malaysia  
30 E-mail: [meheron.sj@curtin.edu.my](mailto:meheron.sj@curtin.edu.my)

31

32

1 **Abstract**

2 The adverse effects of hostile marine environment on concrete structure have inevitably  
3 resulted in huge economic loss and may contribute to catastrophic failure. Concrete is  
4 susceptible to weathering, particularly under wetting-drying action (WDA), although  
5 its current state of the art is well established. The diverse characteristics of WDA at  
6 different site locations have compromised the reliability of laboratory works. The  
7 objective of this study is to review the impact of WDA on concrete and to provide an  
8 overview of the research trend, aiming to identify the research gap. Concrete  
9 deterioration mechanisms in marine environment in respect of WDA are identified. The  
10 influential factors of WDA are analyzed. The physical and mechanical properties and  
11 corrosion resistance of concrete exposed to WDA are discussed. WDA aggravates  
12 concrete deterioration by hastening intrusion of inimical compounds such as chloride,  
13 sulphate and carbon dioxide. Chloride convection zone can be expanded by two to three  
14 times to cause a significant concrete cover loss. Physical damage of concrete starts with  
15 efflorescence staining, followed by mortar delamination, aggregate detachment and  
16 concrete spalling, leading to loss of mechanical properties. The use of mineral  
17 admixtures such as fly ash and silica fume improves concrete resistance against  
18 corrosion, but its refining effect may lead to over-accumulation of chloride, risking the  
19 long-term durability. Limited research works are identified on synergy between  
20 physical and chemical deteriorations, validation of simulated experiment, volume  
21 expansion, mass change and tensile strength of concrete.

22 **Keywords:** Wetting-drying action, Marine concrete, Deterioration mechanism,  
23 Concrete performance

24

1 **1. Introduction**

2 Phenomenal global warming and a consequent rise in seawater level are causing earth  
3 land to shrink. Approximately 70 % of the earth surface is covered by water which is  
4 mainly ocean (NOAA, 2018). Maritime structures are increasing in number because  
5 more coastal defense has to be constructed to counter the onslaught of sea action (Arns  
6 et al., 2017). Concrete has been widely used as construction material for maritime  
7 structures, such as breakwater, groin, seawall and harbour, due to better corrosion  
8 resistance than other materials like steel (Chilana et al., 2016). The proximity of coastal  
9 regions with marine environment can be up to 10 km from the coastline which is  
10 dependent on wind conditions (Pratolongo et al., 2019). Within this vicinity, aggressive  
11 marine environment will lead to concrete degradation. This can cause overall structural  
12 failure and catastrophic loss in the extreme case (Ibrion et al., 2020). Mitigation  
13 measures such as increase of concrete cover and strength, use of stainless steel and  
14 different repair strategies also greatly increase the life-cycle cost (Val and Stewart,  
15 2003). Thus, it is essential to investigate the deterioration mechanisms so that more  
16 holistic solutions can be discovered.

17 Maritime structure can be subjected to four exposure conditions, including submerged,  
18 splash, tidal and atmospheric zones as illustrated in Figure 1(a) (Song et al., 2008). In  
19 atmospheric and submerged zones, on account of low chloride content and less oxygen  
20 exposure respectively, concrete is less prone to corrosion. Concrete corrosion in tidal  
21 and splash zones occurs more intensely due to the high chloride and oxygen contents  
22 and the alternating wetting-drying process. In this context, wetting-drying action (WDA)  
23 facilitates the deterioration of concrete. It induces the chloride accumulation and  
24 increases the intrusion (Y. Chen et al., 2016). The WDA also increases moisture transfer  
25 into concrete, by which the elements brought in can cause material deterioration (Jun

1 Zhang et al., 2011). In tidal zone, concrete experiences a regular WDA caused by sea  
2 tide. The tidal range can vary from 0 m to 12 m (Trenhaile, 2011). In splash zone,  
3 concrete is subjected to more cycles of WDA caused by wave and is liable to more  
4 damage.

5 The WDA has also brought about weathering and erosion of coastal structures for  
6 centuries. Past research concentrated on topic related to wetting-drying effect of rock  
7 and soil used for coastal defense. Gökçeoğlu et al. (2000) found that WDA could  
8 weaken the intergranular bonds of clay-bearing rock. Rao et al. (2001) showed that  
9 lime-stabilized soil could be deteriorated by the same mechanism, causing it to lose  
10 cementation and increase porosity. During the last two decades, the studies have been  
11 focused on concrete structure due to the rapid advancement of concrete technology. The  
12 earlier research in this field analyzed the contributing factors. For example, Sahmaran  
13 et al. (2007) investigated the effect of exposure duration on concrete strength  
14 degradation. Medeiros et al. (2013) studied the effects of cycle number, height and  
15 position of the structure on chloride intrusion. Z. Yu et al. (2015) incorporated factors  
16 of temperature, relative humidity and exposure time to develop chloride diffusion  
17 model. The research was further extended to the cases with loading effects. F. Chen et  
18 al. (2017) studied the behavior of chloride intrusion into concrete subjected to  
19 compression. Fu et al. (2016) investigated the chloride intrusion into concrete under  
20 tensile loading. Gao et al. (2013) examined the cases with flexural loading. The recent  
21 studies enhanced concrete performance under WDA through incorporation of mineral  
22 admixtures such as granulated blast furnace slag (Qi et al., 2017), fly ash (Hoy et al.,  
23 2017), metakaolin (Valipour et al., 2017), palm oil fuel ash (Mohammadhosseini et al.,  
24 2017) and silica fume (Farahani et al., 2015). New research interest used fiber  
25 reinforced polymer (FRP) composite in concrete due to its higher corrosion resistance

1 and strength-to-weight characteristic. Dawei Zhang et al. (2016) strengthened concrete  
2 beam with carbon FRP fabrics, while W. Tang et al. (2020) used glass FRP sheets.  
3 Garzón-Roca et al. (2015) replaced steel bars with carbon FRP bars. There exists  
4 abundant research related to WDA on concrete, but little attempt has been made by  
5 researchers to summarize the findings in order to identify the trends and gaps in this  
6 field. This paper aims to present a summary of the impacts of WDA on concrete and to  
7 provide insight into the latest research.

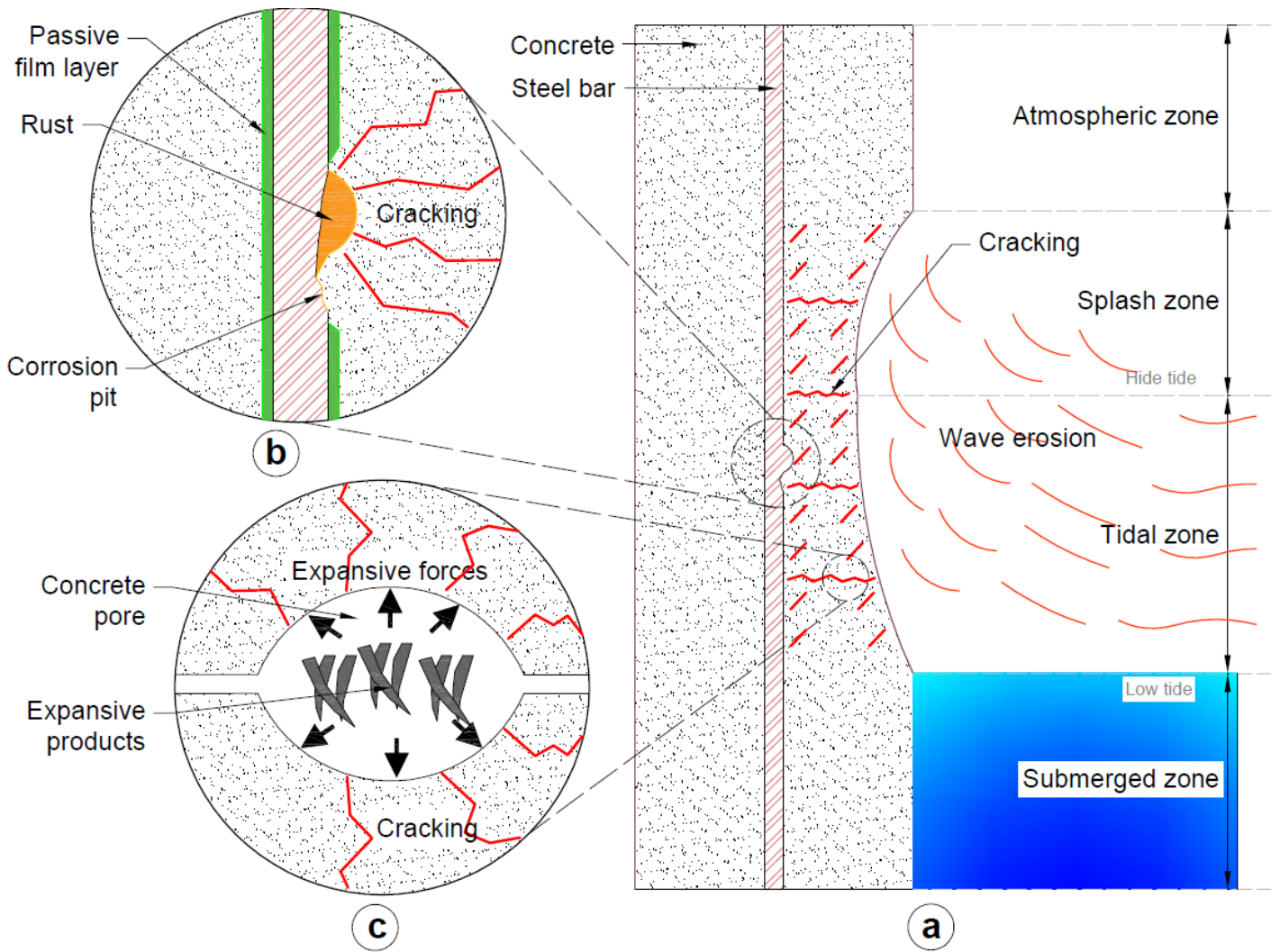
8 Marine concrete is designed for high strength and durability. The strength is required  
9 to resist abrasion and erosion. The durability, measured by permeability, has a bearing  
10 on concrete degradation related to WDA. There is little guideline in the existing  
11 standard to design high durability concrete. The standards provided by ACI (ACI211.1-  
12 91, 2002), BS (BS8110, 1997) and IS (IRC44, 2008) emphasize on the design of high  
13 strength concrete. High strength concrete is perceived to possess less porosity because  
14 of lower water demand. It contains less unreacted water that contributes to pore  
15 development and boosts durability (Kim et al., 2014). There is also little relevant  
16 guideline which can be relied on to carry out mix proportion design for marine concrete  
17 exposed to WDA. Designers are more inclined to do the design based on experimental  
18 results which is time-consuming and expensive. The difference in characteristics of  
19 WDA at different site locations also results in unique laboratory works. This review  
20 paper intends to give an overview of recent WDA experiment carried out by researchers  
21 including its influential factors to strengthen the knowledge in this domain.

22 Against this backdrop, the main objective of this paper is to review the impacts of WDA  
23 on concrete in marine environment. Figure 2 shows an overall diagram of the review.  
24 The concrete deterioration mechanisms and the influence of WDA on the mechanisms  
25 are identified and discussed. The influential factors of WDA on concrete are analyzed

1 and compared. The physical and mechanical properties and corrosion resistance of  
2 concrete exposed to WDA are also reviewed. Upon this review, problems related to  
3 WDA can be outlined to enhance marine concrete.

4 The review of WDA research has provided a better understanding of the concrete  
5 deterioration mechanisms. Mitigation measures have been identified and presented in  
6 Section 2.3, which contributes to the sustainability in two aspects. The first aspect is  
7 the use of more sustainable materials, such as wastes and recycled materials, to improve  
8 marine concrete. This enhances the sustainability by reducing the depletion of  
9 conventional resources for concrete production. The other sustainability aspect of this  
10 study is to increase the resistance of concrete against deterioration and to prolong the  
11 life span of structure. The repair and reconstruction of structures can be minimized,  
12 which reduces maintenance costs and thus the carbon footprint.

13



2 **Figure 1: Concrete exposed to deterioration in marine environment: (a) Element**  
 3 **of concrete structure, (b) steel corrosion and (c) damage by expansion (Mehta**  
 4 **and Monteiro, 2006)**

5



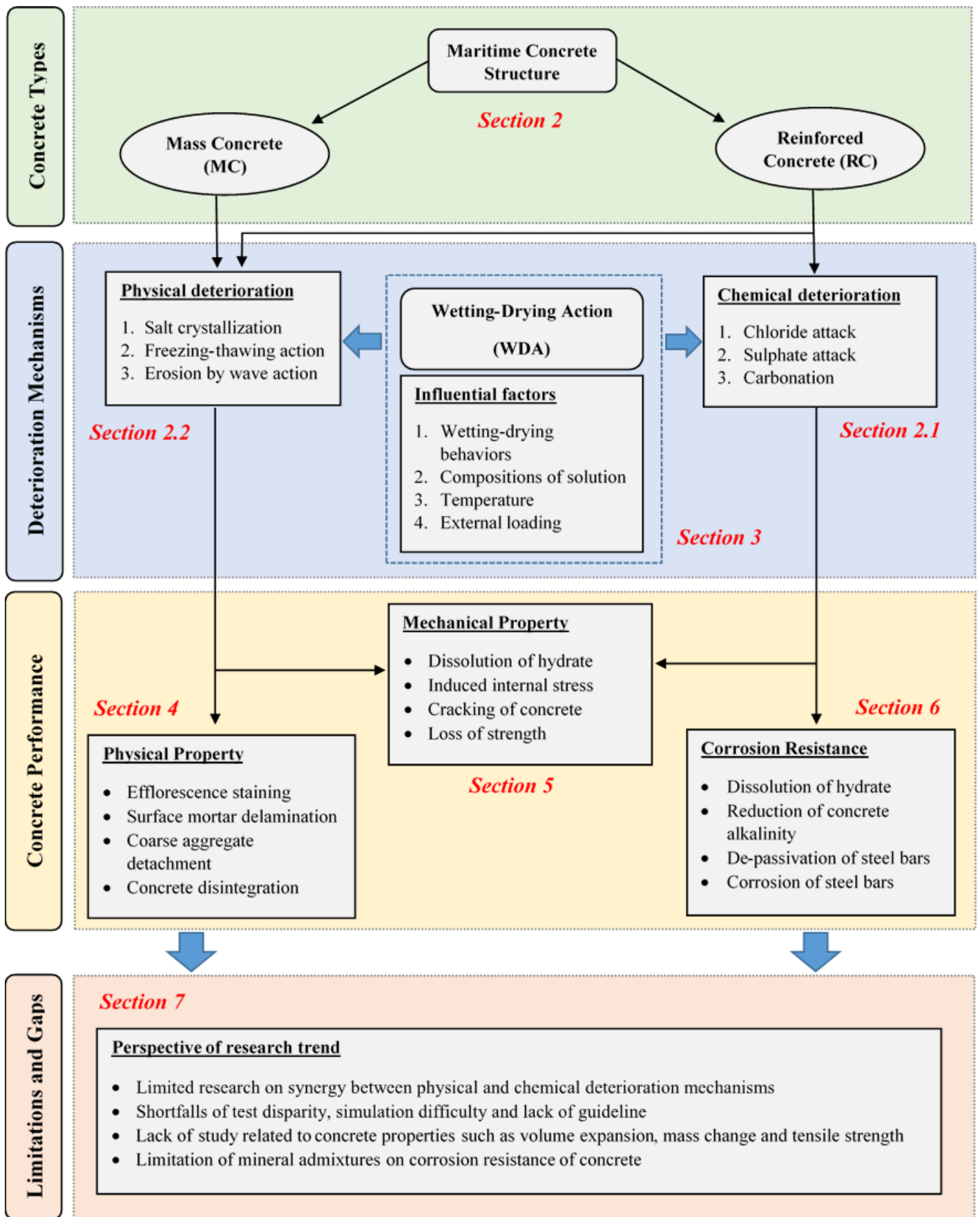


Figure 2: Flow chart of reviewing concrete deterioration exposed to WDA

## 1 **2. Concrete deterioration mechanism in marine environment**

2 From the durability perspective and based on its application, marine concrete can be  
3 classified into mass concrete (MC) and reinforced concrete (RC) (Alexander and  
4 Nganga, 2016). MC is a plain concrete with no reinforcement, while RC is reinforced  
5 with steel bars. MC is used for gravity-stabilizing purpose such as coastal armoring  
6 feature, which is only subjected to physical deterioration. The durability issue of MC is  
7 associated with deterioration of concrete and is critical if its strength is compromised.  
8 RC portrays more severe durability problem due to corrosion of embedded steel. The  
9 intrusion of deleterious compounds, themselves and subsequent reactions with concrete  
10 matrix which cause property changes, can reduce the passivity of steel and lead to  
11 corrosion (R. Zhang et al., 2010). Concrete performance deteriorates in aggressive  
12 marine environment and WDA can further aggravate it. The chemical and physical  
13 deterioration mechanisms of concrete are further elaborated.

14

### 15 *2.1 Chemical deterioration*

#### 16 *2.1.1. Chloride attack*

17 Chloride from seawater causes corrosion of steel in RC. In alkaline environment of  
18 concrete, the surface of steel is surrounded by a thin passive oxide film which stabilizes  
19 the metal and prevents ionization. If the concentration of invading chloride reaches a  
20 threshold level, it reacts with and destroys the oxide film, exposing the steel to oxygen  
21 and water to cause corrosion (Khan et al., 2017). The deterioration process commences  
22 with localized corrosion and formation of rust. It causes approximately 2 to 6 times  
23 volume expansion of steel to exert stress which causes concrete cracking and  
24 performance loss as illustrated in Figure 1(b) (Apostolopoulos et al., 2019). Both bound  
25 and free chlorides exist in concrete. Bound chloride is immobilized in alkaline

1 environment and is not deleterious to steel, while free chloride penetrates into concrete  
2 to cause corrosion.

3 The WDA increases the rate of chloride attack on concrete. Significant increment of  
4 chloride concentration can be found at the concrete surface after long-term exposure to  
5 WDA (Qi et al., 2018). The increased chloride concentration on concrete surface results  
6 in expansion of convection zone which is also referred to the depth of maximum  
7 chloride concentration. This zone is equivalent to loss of concrete cover for steel  
8 protection. Table 1 compares the convection zones of concrete under WDA and full  
9 immersion exposures which have been determined by researchers. Full immersion  
10 refers to the test conducted in fully immersed environment without being subjected to  
11 drying phase. The result shows comparatively deeper convection zone for concrete  
12 under WDA exposure than the fully immersed condition. The intrusion rate of chloride  
13 is increased by WDA through the following mechanisms: (i) chloride penetrates into  
14 concrete by sorption during initial wetting stage, (ii) water evaporates during drying  
15 stage while chloride only permeates by means of diffusion, and (iii) chloride further  
16 accumulates and penetrates through capillary suction during the subsequent wetting  
17 stage. The alternating WDA has consequently induced a steep concentration gradient  
18 which amplifies the diffusion rate.

19 **Table 1: Convection zone for chloride intrusion into concrete**

Literature	Convection zone (WDA exposure)	Convection zone (full immersion)*
Chrisp et al. (2002)	25 – 30 mm	N.D.
Simčič et al. (2015)	4 – 8 mm	N.D.
Ye, Jin, Jin, et al. (2016)	5 – 12 mm	N.D.
J. Wu et al. (2016)	10 – 15 mm	5 – 10 mm
Paul et al. (2016)	10 – 15 mm	0 – 5 mm
Qi et al. (2018)	2.5 – 7.5 mm	N.D.
Y. Wang et al. (2018)	2 – 6 mm	N.D.

\*Concrete was fully immersed throughout testing period without being subjected to drying phase.  
N.D. – Not determined in the study

1 2.1.2. *Sulphate attack*

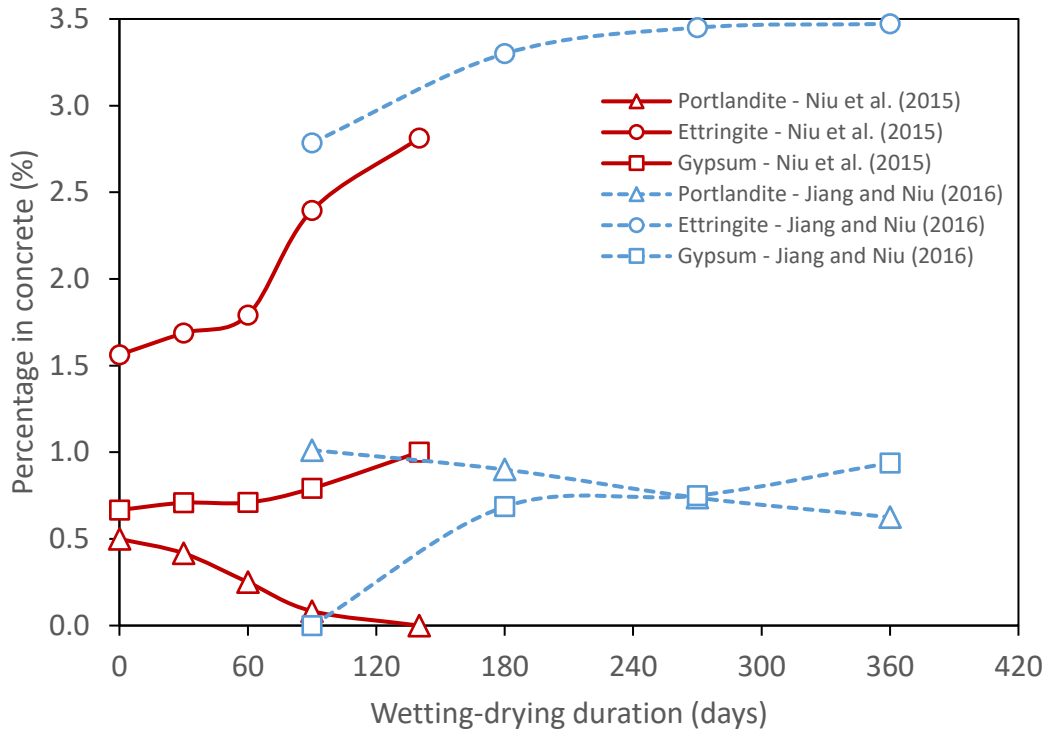
2 Sulphate is another element of seawater which brings about durability problem to  
3 concrete. It reacts with calcium aluminate hydrate (C-A-H) and calcium hydroxide (CH)  
4 to form ettringite and gypsum which are expansive (Rozière et al., 2009). Figure 3  
5 quantifies the reaction of sulphate attack. The ettringite and gypsum increase with  
6 duration of WDA, while the portlandite decreases. The accumulation of ettringite and  
7 gypsum exerts stress on concrete pore, causing it to crack and disintegrate as illustrated  
8 in Figure 1(c). The dissolution of CH and C-A-H also leads to loss of concrete strength.  
9 It can reduce concrete alkalinity and thereby reduces the corrosion resistance of steel.  
10 Concrete experiences more severe deterioration in the presence of magnesium sulphate  
11 (Jiang and Niu, 2016). Cation-exchange reaction can occur, by which magnesium (Mg)  
12 ion replaces Ca ion from calcium silicate hydrate (C-S-H) to form magnesium silicate  
13 hydrate (M-S-H). M-S-H does not provides cementation which results in further  
14 strength loss (Cheng, Shui, Gao, Lu, et al., 2020).

15 The WDA intensifies the sulphate attack on concrete deterioration. It increases the rate  
16 of capillary absorption and diffusion which facilitates sulphate penetration. The drying  
17 stage from WDA also causes concrete shrinkage and then micro-cracking. However, Z.  
18 Wu et al. (2017) and X.-t. Yu et al. (2018) claimed that the subsequent wetting stage  
19 could partially heal the cracks. Once re-wetted, the damaged C-S-H absorbs water, and  
20 swells to close the cracks. The crystallization of sulphate-based salt due to WDA also  
21 causes concrete damage. The precipitated salts exist in the form of thenardite and  
22 mirabilite crystals (Zhongya et al., 2019). These crystals are needle-shaped which exert  
23 stress in the confined concrete pores (Jiang and Niu, 2016).

24 The deterioration caused by sulphate attack exhibits in two stages which are enhancing  
25 and weakening stages. The findings have been confirmed in the separate studies carried

1 out by Gao et al. (2013), Jinrui Zhang et al. (2017) and X.-t. Yu et al. (2018). The  
2 enhancing stage is attributed to the filling of concrete pores by ettringite and gypsum  
3 which improves the mechanical properties. The weakening stage is caused by  
4 expansion of ettringite and gypsum which induces concrete damage. Some studies,  
5 however, describe the sulphate attack as three stages of deterioration mechanism. A  
6 damaging stage which occurred prior to the enhancing stage was noted by Qi et al.  
7 (2017). The damaging stage was due to negative impact of WDA such as shrinkage  
8 cracking caused by repeated drying. After weakening stage, Müllauer et al. (2013)  
9 found that in the third stage, concrete would not undergo further deterioration. This was  
10 ascribed to exhaustion of reactants for formation of expansive products. Zhongya et al.  
11 (2019) discovered an incubation stage in between the enhancing and weakening stages  
12 under WDA in the actual environment. The stress generated by gradual formation of  
13 expansive products within this period was not sufficient to cause deterioration. This  
14 indicated that the accelerated laboratory test could alter the deterioration process.

15



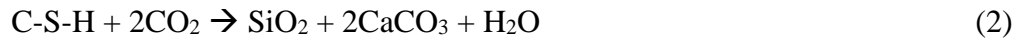
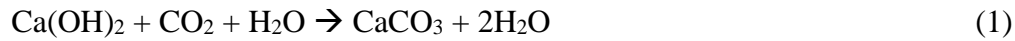
1

2 **Figure 3: Relative composition of ettringite, gypsum and portlandite in concrete**  
 3 **(Jiang and Niu, 2016; Niu et al., 2015)**

4

5 *2.1.3. Carbonation*

6 In humid marine environment, moisture absorbed by concrete contains dissolved  
 7 carbon dioxide (CO<sub>2</sub>) which initiates carbonation. CO<sub>2</sub> is highly reactive with calcium  
 8 elements such as CH and C-S-H in hydrated cementitious matrix. The chemical  
 9 reactions of carbonation is presented as Eq. (1) and Eq. (2) (Samimi et al., 2018). The  
 10 products of these two reactions are calcium carbonate. This leads to dissolution of  
 11 portlandite and depletion of bonding medium, resulting in instability of anhydrous  
 12 phase of cement paste (C.-F. Chang and Chen, 2006). The alkalinity of concrete  
 13 gradually decreases with the reduction of hydroxide concentration which destroys the  
 14 passivity of embedded steel, leading to corrosion.

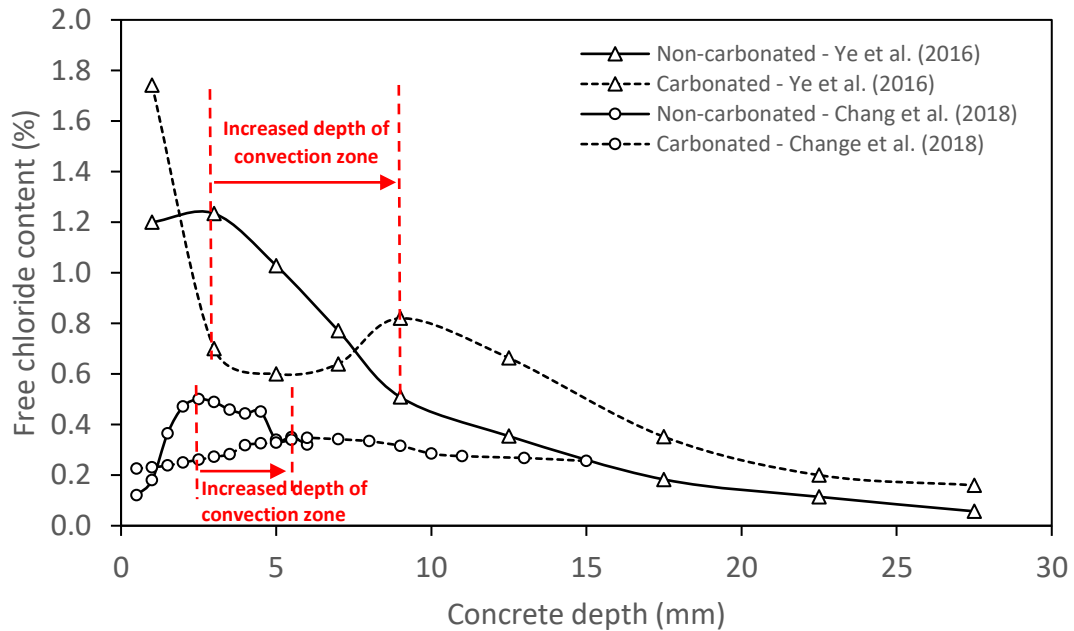


1 Research shows that the WDA intensifies deteriorating effect of concrete carbonation.  
2 Jerman et al. (2019) found that WDA caused microstructure of cementitious matrix to  
3 become coarser which speeded up intrusion of CO<sub>2</sub>. H. Chang et al. (2018) showed that  
4 the accelerated capillary suction induced by WDA and the direct penetration of CO<sub>2</sub>  
5 into concrete during drying period caused more severe carbonation. Ye, Jin, Fu, et al.  
6 (2016) discovered that the carbonation was speeded up in chloride environment.  
7 However, Kuosa et al. (2014) demonstrated that the presence of chloride could retard  
8 concrete carbonation. The hygroscopic characteristic of chloride created water film to  
9 block concrete pores. Salt crystallization caused by WDA also filled up concrete pore  
10 and slowed down carbonation.

11 Carbonation can alter concrete property and negatively affect deterioration from  
12 chloride attack (H. Chang, 2017). Carbonation reduces chloride binding capacity of  
13 concrete which converts bound chloride to free chloride and accelerates steel corrosion.  
14 The bound chloride, existed as Friedel's salt in concrete, is decomposed during  
15 carbonation and is prevented from formation. This is due to the reduction of pH which  
16 gradually alters the surface charge of C-S-H and reduces its ability to bind chloride  
17 (Saillio et al., 2014). The carbonation also widens the chloride convection zone as  
18 shown in Figure 4. Ye, Jin, Fu, et al. (2016) found pronounced increment in the depth  
19 of chloride convection zone in concrete after being carbonated. H. Chang et al. (2018)  
20 demonstrated that under accelerated carbonation, the location of chloride maximum  
21 progressed deeper into concrete. Within the carbonated zone, concrete pores are refined  
22 which reduces chloride intrusion, but this tends to accumulate chloride at the

1 carbonation front. The bound chloride is also decomposed into free chloride. This  
2 consequently leads to high concentration gradient and hastens chloride diffusion.

3



4

5 **Figure 4: Chloride ingress profile before and after carbonation (H. Chang et al.,**  
6 **2018; Ye, Jin, Fu, et al., 2016)**

7

## 8 2.2 Physical deterioration

### 9 2.2.1. Salt crystallization

10 In the absence of chemical interaction, salt crystallization occurs in saline environment  
11 and physically damages concrete. The damage is caused by internal stress arising from  
12 the formation of salt crystals in concrete pores when the pore solution becomes  
13 supersaturated (Thaulow and Sahu, 2004). Salt crystallization happens in two  
14 circumstances which are evaporation of pore solution with the capillary rise and WDA  
15 (Nadelman and Kurtis, 2019). A schematic illustration of the first mechanism is shown  
16 in Figure 5. Concrete in the close vicinity of salt source becomes saturated as the  
17 moisture is transferred through capillary suction. Concrete surface becomes drier when



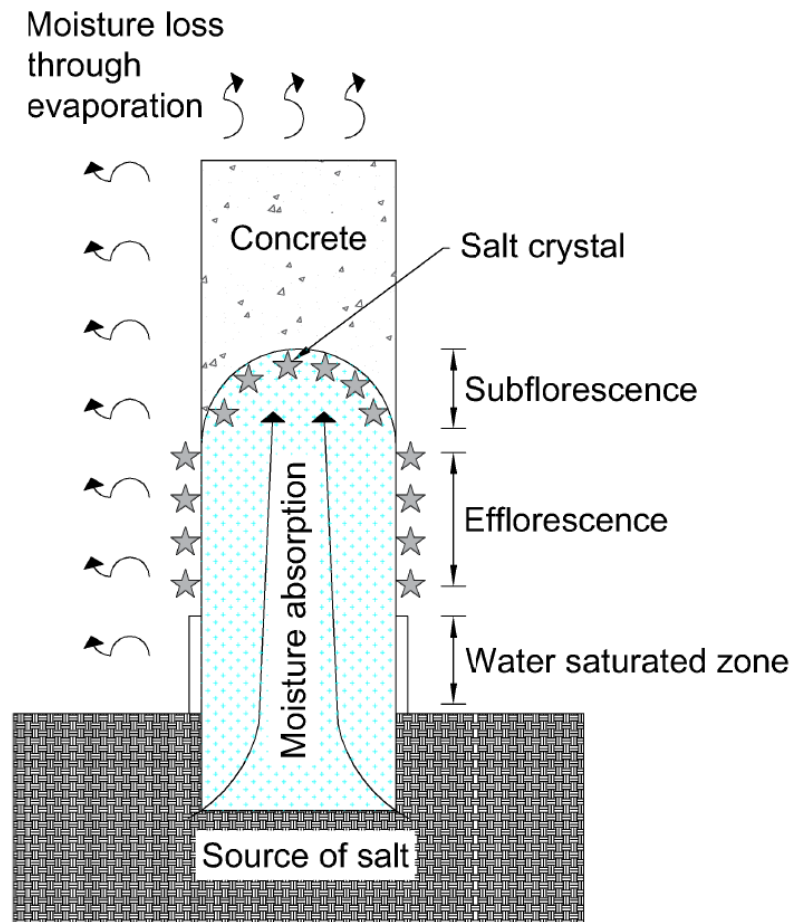
1 the surface is further away from the source as a result of evaporation. The concentration  
2 of pore solution increases and salt begins to precipitate after reaching the saturation  
3 point. Alternating cycles of WDA, on the other hand, causes repetitive exchange of pore  
4 solution and accumulates salt solution in concrete whereby crystallization takes place  
5 after saturation point. The salt precipitated on concrete surface is termed as  
6 efflorescence, while that formed within concrete pore is subflorescence which is the  
7 cause of concrete distress (Lee and Kurtis, 2017). Concrete is damaged through bulging  
8 when the pressure exerted by subflorescence exceeds tensile strength.

9 Saline environments such as salt-bearing soil, saline groundwater and seawater mainly  
10 contains chloride and sulphate salts. The damage by sulphate salt is more severe than  
11 chloride salt and it is receiving greater attention. Nadelman and Kurtis (2019) noticed  
12 the formation of micro-cracks and layers of efflorescence on mortar surface which were  
13 caused by crystallization of sulphate salt. Steiger and Asmussen (2008) found that  
14 sodium sulphate crystallized to form thenardite and mirabilite which caused damage to  
15 concrete. The phase conversion between thenardite and mirabilite could also result in  
16 concrete scaling (Haynes et al., 2008).

17 Some studies, on the contrary, shows that other types of salt cause more severe  
18 deterioration of concrete. Gentilini et al. (2012) found more significant destruction of  
19 specimen by chloride salt, while the sulphate salt appeared to provide stiffening effect.  
20 This was due to the chemical reaction of sulphate with cement hydrate, forming  
21 ettringite and gypsum, which strengthened concrete by filling up pores. Lee and Kurtis  
22 (2017) discovered that the damage of mortar caused by nitrate salt is more severe than  
23 sulphate salt. The samples exposed to sulphate salt were not damage, although heavy  
24 efflorescence was observed. This could be due to insufficient exposure duration to the  
25 salt. The result discrepancy could also be due to different source of salt used which

1 exhibits different phase-transition characteristic with distinct crystallization behavior  
2 (Haynes et al., 2010).

3



4

5 **Figure 5: Salt crystallization by capillary rise and evaporation (Lee and Kurtis,**  
6 **2017; Nadelman and Kurtis, 2019)**

7

### 8 2.2.2. Freezing-thawing action

9 Freezing-thawing action (FTA) is a physical concrete deterioration mechanism occurs  
10 in wet and cold climate. Its occurrence is caused by crystallization of solvent inside  
11 capillary cavity of concrete. Liquid in pore at freezing temperature transforms to ice  
12 and expands to exert pressure on concrete. The damage depends on concrete pore  
13 characteristics such as size, volume and continuity. FTA is less destructive if the

1 concrete possesses adequate pores which are interconnected and big enough to provide  
2 room for uniform expansion and thus less stress in concrete (H. Cai and Liu, 1998).  
3 Otherwise, internal cracking and spalling of concrete can occur, resulting in loss of  
4 mechanical properties and even further intrusion of deleterious substances (Z. Wang et  
5 al., 2014).

6 Salt-containing seawater can alter deterioration mechanism of FTA on concrete. Miao  
7 et al. (2002) found more damage was experienced by concrete exposed to FTA with  
8 sulphate solution. Sun et al. (2002) found the damage caused by FTA with chloride  
9 solution was twice than that with water. The differential thermal expansion of salt  
10 solution and cementitious matrix causes superficial spalling of concrete (Z. Wang et al.,  
11 2014). Chemical deterioration of concrete, such as sulphate attack, can also take place  
12 which is in synergy with FTA. However, Jiang et al. (2015) discovered that FTA with  
13 salt solution caused less damage than that with fresh water. The salt lowered freezing  
14 point of pore solution which made the damage less severe.

15 In actual environment, concrete experiences FTA during winter and WDA during rest  
16 of the year. Hamze (2014) found that the combined FTA and WDA had significant  
17 effect on causing micro-cracks. F. Liu et al. (2018) demonstrated that the combined  
18 FTA, sulphate attack and carbonation caused more severe damage to concrete with  
19 increased pore degradation. Nevertheless, the study related to combined effect of WDA  
20 and FTA is still limited and indeed required investigation.

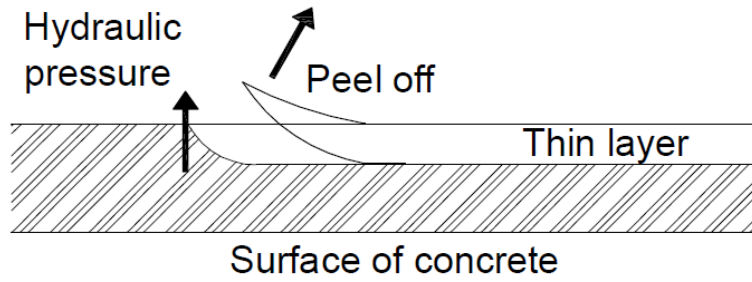
21

### 22 2.2.3. *Erosion by wave action*

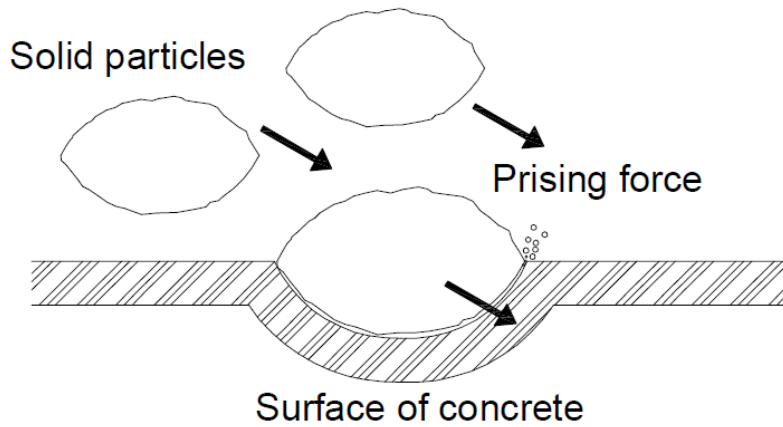
23 Wave action can also cause physical concrete damages such as wear and tear. Wave  
24 often carries debris such as silt, sand, gravel, rock and ice. It can cause two types of  
25 action which are hydrodynamic force and abrasion by water-borne debris (Dandapat

1 and Deb, 2016). The latter action causes more damage to concrete (Kryżanowski et al.,  
2 2009). The erosion of concrete due to abrasion occurs in three stages as shown in Figure  
3 6. During initial stage, concrete is subjected to weak pre-abrasion by hydraulic pressure  
4 of flowing water, whereby concrete surface is slightly stripped off. Drifting water-borne  
5 debris follows on to exert prising force which generates more drastic damage such as  
6 superficial cracking. Combination of both actions abrades and scours off the mortar  
7 layer, disintegrating coarse aggregate and interfacial transition surface. Cavitation force  
8 from implosion of air bubble during wave striking can also induce destructive wearing  
9 damage on concrete (Elżbieta Horszczaruk, 2005). Concrete consequently loses  
10 mechanical strength and shield against corrosion.

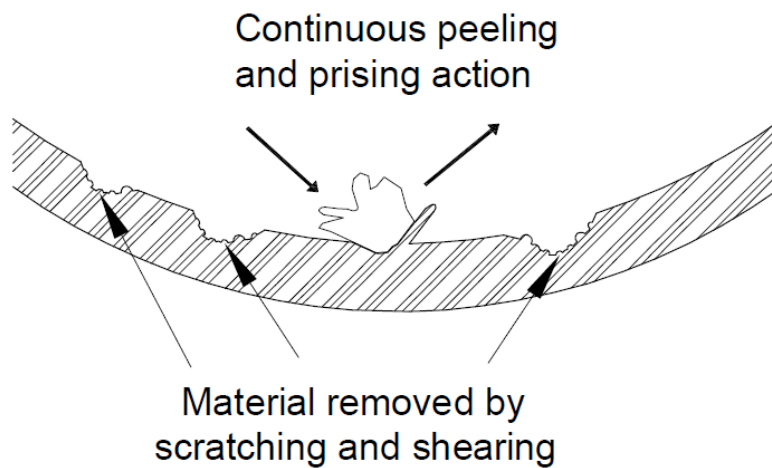
11 Research in this domain has made assessment on hydraulic structures subjected to  
12 abrasion. Abrasion resistance is often related to the strength of concrete. Kryżanowski  
13 et al. (2009) used compressive strength, tensile strength, elastic modulus and aggregate  
14 strength to assess abrasion resistance of concrete. Elżbieta Horszczaruk (2005)  
15 demonstrated that concrete compressive strength could accurately estimate the abrasion  
16 resistance. Y.-W. Liu et al. (2006) found that splitting tensile strength was more  
17 effective in predicting abrasion. The abrasion resistance of concrete also depends on  
18 other factors. Dandapat and Deb (2016) showed that concrete made from coarse  
19 aggregates having larger size, less angularity and flakiness exhibited stronger abrasion  
20 resistance. Xinhua Cai et al. (2016) discovered that wave abrasion force increased when  
21 impinging angle increased from 0 to 90° as this caused greater concrete deformation.  
22 Higher impinging velocity of wave could further aggravate the damage. The  
23 degradation of concrete can occur together with other types of mechanism. Rosenqvist  
24 et al. (2017) showed that both leaching and FTA could cause superficial concrete  
25 damage and debilitate the abrasion resistance.



(a) Erosion peel



(b) Prising by solid particles



(c) Abrasive wear

1

2

**Figure 6: Mechanism of abrasion from water-borne solid on concrete surface**

3

(Y.-W. Liu et al., 2006; Plum and Xufei, 1996)

1    2.3 *Remarks and improvement approach*

2    MC is prone to deterioration only if the mechanisms, such as sulphate attack, salt  
3    crystallization, FTA and erosion, causes loss of mechanical properties. The main  
4    concern for RC is corrosion of steel bar whereby carbonation and chloride attack come  
5    into play. However, other deterioration mechanisms that are critical to MC also have  
6    an impact on providing means for intrusion of chloride and carbon dioxide.

7    The deleterious compounds, which include chloride, sulphate and carbon dioxide, enter  
8    the concrete through capillary pore to actuate deterioration. The intrusion of these  
9    compounds can be minimized by reducing size, number and continuity of pore. The  
10    damage caused by physical deterioration can be mitigated by improving the concrete  
11    mechanical properties, especially the tensile strength, which prevents disintegration of  
12    concrete. Some of the existing methods for mitigating concrete deteriorations are also  
13    summarized in Table 2.

14    The WDA has great influence on the deterioration mechanisms of concrete. It increases  
15    intrusion rate of deleterious compounds into concrete which escalates the deterioration  
16    processes. Research concentrates on a single deterioration mechanism. In actual  
17    weathering scenario, however, there is occurrence of synergy between two or more  
18    mechanisms. It is recommended to study synergetic effect of various combinations,  
19    especially the combination of chemical and physical deteriorations.

20

**Table 2: Concrete deterioration mitigation methods**

Deterioration mechanism	Mitigation methods	Remark	Reference
<u>Chemical deterioration</u>			
1. Chloride attack	i) Additions of supplementary cementitious material (SCM) such as blast furnace slag, fly ash and silica fume	• Densify and refine pore structure through pozzolanic reaction	(Chrisp et al., 2002; Ye, Jin, Jin, et al., 2016)
	ii) Use of early carbonation curing method	• Pre-carbonate concrete in CO <sub>2</sub> environment to densify pore structure	(Duo Zhang and Shao, 2016)
2. Sulphate attack	i) Addition of SCM such as blast furnace slag and fly ash	• Densify and refine pore structure through pozzolanic reaction	(Gao et al., 2013; Qi et al., 2017)
	ii) Use of low tricalcium aluminate (C <sub>3</sub> A) cement	• Reduce hydrates of C <sub>3</sub> A to form ettringite to cause deterioration	(Haynes et al., 2008)
	iii) Use of recycle coarse aggregate (RCA)	• Higher water absorption of RCA lowers w/b ratio to increase strength	(Qi et al., 2017)
3. Carbonation	i) Addition of SCM such as blast furnace slag, fly ash and silica fume	• Densify and refine pore structure through pozzolanic reaction • Require prudent adding SCM since higher content reduces portlandite and consequently carbonation resistance	(Backus et al., 2013; Ye, Jin, Fu, et al., 2016)
	ii) Lower water-to-binder (w/b) ratio	• Reduce concrete pore and intrusion rate	(H. Chang et al., 2018)
	iii) Addition of $\gamma$ -dicalcium silica ( $\gamma$ -C <sub>2</sub> S) for CO <sub>2</sub> sequestration	• $\gamma$ -C <sub>2</sub> S reacts with CO <sub>2</sub> to form highly dense SiO <sub>2</sub> gel to fill up concrete pore	(Z. Chen et al., 2019)
<u>Physical deterioration</u>			
1. Salt crystallization	i) Lower w/b ratio	• Increase concrete strength against crystallization pressure • Reduce concrete pore and bring down flux of salt solution	(Lee and Kurtis, 2017)
	ii) Use of Portland limestone cement (PLC)	• Fine PLC reduces size and interconnectivity of concrete pore	(Nadelman and Kurtis, 2019)
2. Freezing-thawing action	i) Addition of air-entraining admixtures	• Increase volume and connectivity of concrete pore to increase escape boundary and reduce expansion stress	(Aygörmez et al., 2020; Hamze, 2014)
	ii) Steel fibre reinforcement	• Increase concrete ductility and retard cracking	(Miao et al., 2002; Sun et al., 2002)

3. Wave erosion

- |   |   |  |
|---|---|--|
| i) Use of round, non-flaky and large coarse aggregate | • Produce stockier, less brittle and mechanically stronger concrete           | (Dandapat and Deb, 2016; Y.-W. Liu et al., 2006) |
| ii) Steel fibre and polypropylene fibre reinforcement | • Provide better adhesion to cementitious matrix and reduce crack propagation | (EK Horszczaruk, 2009; Kryżanowski et al., 2009) |
| iii) Addition of SCM such as fly ash and silica fume  | • Increase concrete strength through pozzolanic reaction                      | (Xinhua Cai et al., 2016)                        |
- 

1

2



1 **3. Influential factors of WDA**

2 The characteristics of WDA have bearing on the intrusion of deleterious compounds  
3 into concrete. The characteristics vary with site location, climate and exposure  
4 conditions. Researchers have reproduced the characteristics in laboratory with the aim  
5 to study the influences of WDA on concrete performance. Accelerated test is also used  
6 to shorten the study period. This can lead to inaccurate prediction if irrelevant method  
7 is used. The influential factors of WDA are reviewed based on the experiments  
8 performed by various researchers.

9

10 *3.1. Phenomenon of wetting and drying*

11 The alternating wetting and drying of marine concrete is caused by tidal phenomenon.  
12 The characteristics of tide, particularly tidal cycle and tidal range, has great influence  
13 on WDA behavior. Tidal phenomenon can be classified into three types which are  
14 diurnal, semidiurnal and mixed tides. Diurnal tide exhibits one cycle of low and high  
15 tides per day, while semidiurnal tide displays two cycles per day (NOAA, 2019). Mixed  
16 tide is the case of semidiurnal tide having different amplitude of crest and trough. The  
17 tidal range can vary from near 0 to 12 m (Trenhaile, 2011). Considering the effects of  
18 tidal cycle and range, wetting and drying periods can vary in different tidal situations.

19 Table 3 summarizes the wetting and drying periods adopted by several researchers in  
20 their laboratory tests to simulate WDA at the field. J. Wu et al. (2016) conducted  
21 wetting-drying test by following the tidal conditions of Bohai Sea, East Sea and Yellow  
22 Sea in China, in which concrete was exposed to 8 hours of both wetting and drying for  
23 one tidal cycle. X.-t. Yu et al. (2018) used wetting-drying period from Chinese Standard:  
24 GB/T 50082-2009 (Standard, 2009) which adopted 16 hours of wetting, 2 hours of room  
25 temperature drying and 6 hours of oven drying at 60 °C. Niu et al. (2015) also used the

1 same laboratory condition for simulating deterioration of shotcrete in a tunnel  
2 construction site of Western China. This exposure condition has included an oven  
3 drying period which is used to boost water evaporation and shorten test duration.  
4 Backus et al. (2013) used period of 1-day wetting and 6-day drying which they claimed  
5 this scenario as splash zone. Zhongya et al. (2019) adopted a regime of 8-day wetting  
6 and 7-day drying and compared with continuous immersion and actual wetting-drying  
7 environment of Chongqing, China. The study showed that the adopted conditions for  
8 laboratory test could shorten test period, but slightly change the actual degradation  
9 process.

10

### 11 *3.2. Ion type and concentration*

12 Concrete deterioration by WDA is also influenced by compositions of seawater  
13 especially chloride and sulphate, and their concentrations. Chloride and sulphate  
14 contents vary with salinity of seawater. Salinity is the total amount of non-carbonate  
15 salts which exist in combination with sodium, potassium and magnesium ions (Effler  
16 et al., 1986). The seawater has an average salinity of 3.5 % and the salt exists mostly as  
17 sodium chloride (Lehigh, 2011). Researchers thus use this concentration of chloride  
18 solution in their studies as shown in Table 3. The commonly used sulphate  
19 concentration is 5 %. It is assumed that this concentration can result in noticeable  
20 damage to concrete and is suitable for laboratory test. The ASTM (C1012, 2018)  
21 standard also recommends this concentration of sulphate solution for testing. Some  
22 studies also increase ion concentration to perform accelerated test. The chloride and  
23 sulphate concentrations have been increased up to 10 % and 15 % respectively. Yuan  
24 et al. (2016) showed the test duration was reduced by a factor of two if sulphate  
25 concentration at saturation point had been used. Different types of sulphate also cause

1 different extent of concrete degradation. Jiang and Niu (2016) and Cheng, Shui, Gao,  
2 Yu, et al. (2020) demonstrated that concrete damage by  $MgSO_4$  solution was more  
3 severe than  $Na_2SO_4$  solution. An additional deterioration caused by cation-exchange  
4 reaction which replaced Ca ion in C-S-H with Mg ion, resulting in strength loss of  
5 concrete.

6 Researchers also use a composite solution containing chloride and sulphate salts as  
7 occurred in seawater. Jiang and Niu (2016) used composite solution of 10 %  $Na_2SO_4$   
8 and 3.5 % NaCl which reduced concrete deterioration. The presence of chloride reduced  
9 diffusion of sulphate as the Friedel's salt was formed from chloride to fill up pores in  
10 concrete which delayed sulphate attack. Cheng, Shui, Gao, Yu, et al. (2020) also found  
11 reduced chloride intrusion with the co-existence of sulphate. The intruded chloride  
12 amount could be further reduced if the sulphate concentration increased (Y. Chen et al.,  
13 2016). The modification of charge-equilibrium and mass-equilibrium status of ions in  
14 composite solution reduced the activity of chloride ion. The formation of ettringite by  
15 sulphate also refined and densified concrete pores, limiting penetration of chloride. The  
16 findings show that the concrete deterioration by composite solution, which better  
17 represent the actual situation, is lower than that with a single salt.

18

### 19 *3.3. Temperature in drying stage*

20 Concrete gradually loses its moisture through water evaporation in drying stage of  
21 WDA. As the ambient temperature increases, the relative humidity decreases which  
22 induces faster drying of concrete. The continual sun exposure raises concrete  
23 temperature up to 30–50 °C (Guan, 2011; Tan et al., 2014). Researchers also include an  
24 oven drying stage in their experiments to consider this effect. The temperature used can

1 range from 45 to 80 °C as shown in Table 3. Higher temperature is used to increase  
2 water evaporation rate which substantially reduces test period.

3 Some studies, however, indicate that the elevated temperature can lead to negative  
4 consequences. It causes concrete shrinkage and then micro-cracking which affects  
5 intrusion behaviors and test result (X.-t. Yu et al., 2018). The elevated concrete  
6 temperature also induces greater moisture transfer and increases chloride content and  
7 penetration depth (Oh and Jang, 2007). Farahani et al. (2015) discovered that chloride  
8 diffusion rate increased by two times when temperature was raised from 28.5 to 34.5  
9 °C. The increased temperature provided more energy for chloride diffusion. Alhozaimy  
10 et al. (2012) found that steel corrosion became more severe when temperature was  
11 increased from 30 to 40 °C, but decreased corrosion for the temperature range of 40 to  
12 50 °C. The increased corrosion was due to catalytic effect provided by higher  
13 temperature for chloride to destroy steel passivity. The decreased corrosion was  
14 attributed to the lower oxygen solubility and the increased concrete pore discontinuity  
15 at this high temperature. Thus, postulation of using elevated temperature in WDA  
16 experiment remains controversial as it can alter the deterioration mechanism.

17

### 18 *3.4. External loading*

19 Concrete in service is subjected to stresses such as compression and tension which can  
20 affect the WDA test results. The stresses modify concrete pore characteristics and  
21 change the intrusion behavior. The resultant strain and the induced crack allow for  
22 further penetration of deleterious components into concrete.

23 Studies have also incorporated the loading effect on deterioration of concrete subjected  
24 to WDA. Gao et al. (2013) found that, with flexural loading, more severe deterioration  
25 was experienced by concrete subjected to WDA. The concrete deterioration also

1 increased with flexural stress ratio (Ye, Jin, Fu, et al., 2016). This was due to the  
2 enlarged concrete pore under tensile strain increased the ingress of corrosive  
3 compounds. The damage caused by long-term flexural loading was also more severe  
4 than the short-term loading since the former caused permanent deformation such as  
5 creep (F. Chen et al., 2017). Fu et al. (2016) showed that cyclic uniaxial tensile loading  
6 had caused fatigue damage to concrete which increased the chloride intrusion by a  
7 factor of three. Paul et al. (2016) stated that the cracked concrete was intruded with  
8 significantly higher chloride than non-cracked concrete. The development of micro-  
9 cracks caused by flexural and tensile stress had speeded up the intrusion of corrosive  
10 compounds.

11 The effect of compressive stress in WDA on concrete has also been studied. Jinrui  
12 Zhang et al. (2017) pointed out that in the compression zone of concrete under flexural  
13 loading, concrete was actually strengthened. J. Wu et al. (2016) also observed similar  
14 scenario when concrete was compressively loaded. The compression closed the  
15 concrete pores and reduced permeability. But, the growth of salt crystal in pores could  
16 induce higher stress under compressive loading in the long term, causing greater  
17 damage to concrete (F. Chen et al., 2017).

18

### 19 3.5. *Remarks*

20 The influential factors of WDA such as wetting-drying periods, ion type, ion  
21 concentration, temperature and condition of external loading have significant impact  
22 on concrete deterioration mechanisms. Table 3 summarizes these influential factors that  
23 are used by researchers to duplicate WDA in actual environment. The wetting period  
24 can range from 0.3 hour to 1080 hours, while the drying period at ambient temperature  
25 ranges from 0.7 hour to 1172 hours. Some studies include an oven drying period of 5

1 hours to 168 hours in order to accelerate the test. The oven drying temperature varies  
2 from 45 to 80 °C. As for ion type in corrosive solution, the study mainly includes NaCl,  
3 Na<sub>2</sub>SO<sub>4</sub> and MgSO<sub>4</sub> and composite of them. The simulated external loadings are  
4 compression, tension and flexural bending.

5 The WDA tests performed by researchers have achieved the objectives of respective  
6 studies. However, the laboratory experiments have inherent limitations such as  
7 difficulties in simulating the actual site situation, shortening the test period using  
8 accelerated test and predicting the long-term concrete durability. Further research is  
9 still required to improve the experiments.

10

**Table 3: Laboratory-simulated field exposure conditions of WDA on concrete**

Literature	Phenomenon of wetting and drying			Ion type and concentration		Drying temperature		External loading		Site location (which WDA was duplicated)
	Wetting period (hours)	Room drying period (hours)	Oven drying period (hours)	Solution type	Concentration (%)	Room temperature (°C)	Oven temperature (°C)	Loading type	Stress ratio	
Ye, Jin, Jin, et al. (2016)	24	48	-	NaCl	5	35 ± 0.5	-	-	-	-
Chrisp et al. (2002)	48	1176	-	NaCl	5.8	22 ± 2	-	-	-	-
Simčič et al. (2015)	0.3	0.7	-	NaCl	10	20 ± 2	-	-	-	-
J. Wu et al. (2016)	8	8	-	NaCl	3	-	-	Compressive	0, 0.3, 0.5, 0.7	Bohai Sea, East Sea and Yellow Sea in China
Paul et al. (2016)	72	96	-	NaCl	3.5	-	-	-	-	-
Qi et al. (2018)	21	6	45	NaCl	3.5	20 ± 2	60	Flexural	0, 0.3, 0.5	-
Y. Wang et al. (2018)	12	12	-	NaCl	3.5	20 ± 2	-	-	-	-
Qi et al. (2017)	21	6	45	Na <sub>2</sub> SO <sub>4</sub>	5	20 ± 2	60	-	-	-
J. Tang et al. (2018)	16	2	6	Na <sub>2</sub> SO <sub>4</sub>	0, 5, 10	20 ± 2	80 ± 5	-	-	-
Y. Chen et al. (2016)	21	6	45	Na <sub>2</sub> SO <sub>4</sub>	5	20 ± 3	60	-	-	-
Yuan et al. (2016)	16	2	8	(Na <sub>2</sub> SO <sub>4</sub> + NaCl)	(10 + 5), (10 + 5)	-	70	-	-	-
Xiaopei Cai et al. (2019)	12	6	6	Na <sub>2</sub> SO <sub>4</sub>	5	20	45	-	-	-
Jiang and Niu (2016)	168	192	-	Na <sub>2</sub> SO <sub>4</sub>	10	20 ± 3	-	-	-	-
				MgSO <sub>4</sub>	10					
				(MgSO <sub>4</sub> + NaCl)	(10 + 3.5)					
Zhongya et al. (2019)	192	-	168	Na <sub>2</sub> SO <sub>4</sub>	0, 2.1, 15	-	45 ± 2	-	-	General use, Chongqing China
H. Zhang et al. (2019)	24	144	-	Na <sub>2</sub> SO <sub>4</sub>	5	20	-	-	-	-
X.-t. Yu et al. (2018)	16	2	6	Na <sub>2</sub> SO <sub>4</sub>	0, 5	20 ± 2	80 ± 2	-	-	Standard (GB/T 50082-2009)

Niu et al. (2015)	16	2	6	Na <sub>2</sub> SO <sub>4</sub>	10	20 ± 2	60	-	-	Tunnel construction site in Western China
Jinrui Zhang et al. (2017)	72	96	-	Na <sub>2</sub> SO <sub>4</sub>	5	23 ± 1	-	Flexural	0.4	-
F. Chen et al. (2017)	21	6	45	(Na <sub>2</sub> SO <sub>4</sub> + NaCl)	(5 + 5)	20	60	Short-term flexural	0, 0.35, 0.5	-
Cheng, Shui, Gao, Yu, et al. (2020)	18	1	5	NaCl	3.3	20 ± 2	60	-	-	-
Cheng, Shui, Gao, Lu, et al. (2020)				(NaCl + MgCl <sub>2</sub> )	(2.7 + 0.5)					
				(NaCl + Na <sub>2</sub> SO <sub>4</sub> )	(3.3 + 0.4)					
				(NaCl + MgCl <sub>2</sub> + Na <sub>2</sub> SO <sub>4</sub> )	(3.3 + 0.5 + 0.4), (3.2 + 0.5 + 0.4)					
Li et al. (2020)	720-1080	0.5	96	Na <sub>2</sub> SO <sub>4</sub>	0, 3, 5, 10	20 ± 2	40 ± 2	-	-	-
Gao et al. (2013)	21	6	45	Na <sub>2</sub> SO <sub>4</sub>	5	20 ± 3	60	Flexural	0, 0.2, 0.4, 0.6	-
Jerman et al. (2019)	48-72	48-72	-	Water	-	25	-	-	-	-
Malheiro et al. (2014)	24	144 (CO <sub>2</sub> environment)	-	NaCl	3.5	20	-	-	-	-
Ye, Jin, Fu, et al. (2016)	96	48, 48 (CO <sub>2</sub> environment), 96 (CO <sub>2</sub> environment)	48	NaCl	15	20 ± 5	60	Flexural	0, 0.3, 0.6	-
Backus et al. (2013)	24	144	-	NaCl	3.2	20 ± 1	-	-	-	Extreme splash zone
H. Chang et al. (2018)	24	144, 144 (CO <sub>2</sub> environment)	-	NaCl	3.5	20 ± 1	-	-	-	-



## 1 **4. Physical properties of concrete exposed to WDA**

### 2 *4.1. Physical appearance*

3 Concrete with aesthetic exterior look is indicative of its reasonably good performance.  
4 Assessment of concrete physical appearance is beneficial in early detection of concrete  
5 deterioration and hence mitigation can be expediently provided for prior to catastrophic  
6 failure.

7 Researchers assess concrete deterioration through its physical appearance. X.-t. Yu et  
8 al. (2018) found that mortar surface was severely damaged after being exposed to 270  
9 days of WDA in 5 %  $\text{Na}_2\text{SO}_4$  solution. The surface was covered with white powdery  
10 efflorescence and erosion pit, which gradually led to spalling and delamination. Jiang  
11 and Niu (2016) investigated the effect of solutions respectively comprising 10 %  
12  $\text{Na}_2\text{SO}_4$  (S1), 10 %  $\text{MgSO}_4$  (S2) and composite of 10 %  $\text{Na}_2\text{SO}_4$  plus 3 %  $\text{NaCl}$  (S3)  
13 on concrete physical appearance. Concrete exposed to S1 experienced some surface  
14 delamination and cracks near the edges. The damage on concrete exposed to S2 was  
15 more severe, wherein the surface was peeled off, uncovering coarse aggregate. Minimal  
16 scaling of surface was observed in concrete exposed to S3. The magnesium sulphate  
17 caused more severe damage to concrete, but chloride if combined with sulphate could  
18 reduce the effect. Y. Chen et al. (2016) also found that the concrete exposed to  
19 composite solution containing chloride was not damaged. The chloride ion was capable  
20 of delaying formation of corrosion products as discussed in Section 3.2.

21 Effect of fly ash (FA) and silica fume (SF) on physical appearance was studied by  
22 Zhongya et al. (2019). The concrete incorporated with FA and SF suffered more severe  
23 physical damage than normal concrete. The damages such as spalling and chipping of  
24 mortar occurred along corners and edges of concrete. Although addition of FA and SF

1 refined concrete pore structure, salt crystals formed had indeed increased pressure in  
2 the pores and caused more damage.

3

#### 4 *4.2. Volume expansion*

5 Concrete swells and expands after being deteriorated in marine environment. Zhongya  
6 et al. (2019) studied the concrete expansion exposed to full immersion (E1), indoor  
7 WDA (E2) and actual WDA (E3) using  $\text{Na}_2\text{SO}_4$  solution for 720 days. The length of  
8 concrete increased with exposure time and the highest volume expansion was found to  
9 be 1.0 %, 1.6 % and 1.4 % for E1, E2 and E3 respectively. The WDA aggravated  
10 volume expansion of concrete. Expansion in E2 was higher than E3 because in the  
11 actual field situation, less cycles of WDA had been experienced by concrete. X.-t. Yu  
12 et al. (2018) investigated the effect of four exposure conditions on the mortar expansion  
13 for 270 days which were Condition 1 (0 %  $\text{Na}_2\text{SO}_4$  with full immersion), Condition 2  
14 (0 %  $\text{Na}_2\text{SO}_4$  with WDA), Condition 3 (5 %  $\text{Na}_2\text{SO}_4$  with full immersion) and  
15 Condition 4 (5 %  $\text{Na}_2\text{SO}_4$  with WDA). There was no noticeable volume change  
16 observed from mortar in Condition 1, while the mortar under Condition 2 experienced  
17 slight but minimal reduction of volume which was caused by shrinkage during drying.  
18 Expansions of mortar were 0.8 % and 1.4 % for Condition 3 and Condition 4  
19 respectively. The WDA had pronounced effect in accelerating concrete deterioration  
20 and then increased volume expansion.

21

#### 22 *4.3. Mass change*

23 Mass change of concrete is also measured to determine extent of concrete deterioration  
24 caused by WDA. Y. Chen et al. (2016) reported the concrete mass increased with a  
25 maximum gain of 1.75 %. The mass gain was caused by salt crystallization and

1 formation of ettringite and gypsum. Cheng, Shui, Gao, Yu, et al. (2020) also noticed  
2 mass gain of mortar up to 2.74 %. The mass gain was caused by continuous hydration  
3 produced more C-S-H which had higher surface area and increased water imbibition up  
4 to its saturation point. Qi et al. (2017) substituted aggregate of concrete with recycled  
5 coarse aggregate. The recycled aggregate concrete (RAC) at replacement level up to  
6 50 % exhibited smaller mass gain than normal concrete, but significantly higher mass  
7 gain at higher replacement level. RAC possessed higher water absorption property  
8 whereby higher replacement level could increase mass gain. At low replacement level,  
9 RCA absorbed mixing water which in turn reduced effective w/b ratio and concrete  
10 strength increased. The concrete was less vulnerable to deterioration and hence  
11 exhibited lower mass gain.

12 Mass of concrete could also decrease when subjected to WDA. J. Tang et al. (2018)  
13 studied mass change of concrete exposed to 0 %, 5 % and 10 % Na<sub>2</sub>SO<sub>4</sub> solutions. The  
14 concrete exposed to 5 % and 10 % Na<sub>2</sub>SO<sub>4</sub> solutions exhibited mass gain, the mass  
15 reduced in 0 % solution. The mass loss was due to surface damages such as peeling and  
16 chipping of mortar layer caused by WDA. The mass gain in 5 % and 10 % Na<sub>2</sub>SO<sub>4</sub>  
17 solution was ascribed to formation of ettringite and gypsum. Jiang and Niu (2016)  
18 noticed that concrete mass loss occurred in two stages which were steady and escalated  
19 stages. Concrete experienced constant mass loss during steady stage and after 150 days  
20 of exposure, it lost mass rapidly. Mortar layer gradually disintegrated to cause weight  
21 loss in escalated stage.

22 The contradiction in concrete mass change (i.e. mass gain or loss) described in the  
23 above literature is attributed to the difference in concrete mechanical properties. Higher  
24 strength concrete can resist damage caused by sulphate attack and salt crystallization,  
25 particularly the internal stress resulted from expansive products. Concrete mass

1 increases with the accumulation of the products. Low strength concrete is more prone  
2 to the damage. The expansive products cannot be accumulated in the concrete, but can  
3 cause severe physical scaling and weight loss.

4

#### 5 *4.4. Remarks*

6 The sulphate attack and salt crystallization under the influence of WDA are the main  
7 deterioration mechanisms which change the physical properties of concrete. Concrete  
8 surface is physically damaged with forming of efflorescence initially, followed by  
9 gradual delamination and peeling of mortar, which further cause detachment of coarse  
10 aggregate. This leads to cover loss and further disintegration of concrete.

11 Concrete can also experience volume expansion after exposing to the long-term  
12 deterioration mechanisms. Review of the literature shows this type of expansion is  
13 minimal with value in the range of 0.8–1.6 %. But, the expansion of concrete volume  
14 in a restrained structure can be detrimental as it induces stress to cause concrete  
15 cracking.

16 Mass change of concrete caused by the deterioration mechanisms depends on its  
17 strength. Concrete with high strength experiences mass increase, but low strength  
18 concrete suffers mass loss. The change of concrete physical properties in respect of  
19 their appearance, volume and mass changes can be used as material deterioration  
20 indicators.

21

## 1 **5. Mechanical properties of concrete exposed to WDA**

### 2 *5.1 Compressive strength*

3 The influence of WDA deteriorates the mechanical properties of concrete, of which the  
4 compressive strength is the most important assessment criterion used for concrete  
5 performance. The following investigates the deterioration of compressive strength  
6 associated with WDA.

7 X.-t. Yu et al. (2018) investigated compressive strength of mortar under the influence  
8 of WDA and sulphate solution. The compressive strength increased for 150 days but  
9 decreased significantly afterward. Li et al. (2020) found that the strength reduction  
10 started after 6 months of WDA. The formation of ettringite and gypsum from sulphate  
11 solution densified microstructure initially, but their subsequent accumulation caused  
12 strength reduction. The reduction was more severe under WDA. The WDA also caused  
13 formation of salt crystals in the pores when the solution became supersaturated during  
14 drying process. Internal stress was induced, causing damage and hence loss of strength.  
15 Zhongya et al. (2019) also studied compressive strength of concrete exposed to full  
16 immersion (E1), indoor WDA (E2) and actual WDA (E3) using sulphate solution. The  
17 strength of concrete exposed to E1 increased throughout exposure period, but slightly  
18 reduced towards the end of test period. The reduction was more prominent with the use  
19 of higher sulphate concentration. Concrete with E3 exposure exhibited similar trend as  
20 E1, but the strength loss occurred earlier with higher severity. The WDA accelerated  
21 the strength loss. The concrete exposed to E2 deteriorated two times faster than that in  
22 E3 exposure. The actual WDA could be imitated in laboratory with accelerated rate in  
23 the aspect of compressive strength.

24 J. Tang et al. (2018) found that the compressive strength of concrete exposed to WDA  
25 using tap water reduced by 29 % after 140 days. Although the sulphate deterioration

1 was not involved in the test, the damage could be caused by shrinkage of concrete  
2 during drying cycles. Tap water also contained mineral by which salt crystallization  
3 had occurred to deteriorate concrete. Jiang and Niu (2016) investigated the effect of  
4 different sulfate solutions which were  $\text{Na}_2\text{SO}_4$  (S1),  $\text{MgSO}_4$  (S2) and composite of  
5  $\text{Na}_2\text{SO}_4$  plus  $\text{NaCl}$  (S3) on concrete compressive strength. In all cases, the strength  
6 slightly increased initially. It was followed by a constant reduction, then an accelerated  
7 reduction towards the end of test period. The strength reduced in the increasing order  
8 of S3, S1 and S2. The S3 contained chloride which delayed sulphate attack. Mg ion  
9 from S2 induced cation-exchange reaction which reduced C-S-H for bonding.

10

### 11 *5.2 Tensile strength*

12 Tensile strength is an essential mechanical property to prevent cracking of concrete. It  
13 can be assessed from both splitting test and flexural test. Limited research related to  
14 this property has been done. J. Tang et al. (2018) investigated the effect of WDA on  
15 concrete flexural strength. The flexural strength reduced when concrete was exposed to  
16 WDA using sulphate solution. The flexural strength reduction was greater than that of  
17 compressive strength. Concrete cracking as a result of sulphate attack and salt  
18 crystallization caused more loss of concrete capacity to resist tensile stress.

19

### 20 *5.3 Modulus of elasticity*

21 Relative dynamic modulus of elasticity (RDME) is also used to assess weathering of  
22 concrete subjected to WDA. It is the elasticity of concrete under dynamic loads such as  
23 longitudinal and flexural vibrations.

24 Jiang and Niu (2016) studied the effect of WDA on RDME of concrete exposed to  
25 sulphate solution. The RDME loss occurred in three stages which were steady-

1 decreasing, constant and rapid-decreasing stages. The damage in first stage was caused  
2 by WDA and salt crystallization. In the second stage, concrete pores were densified by  
3 formation of ettringite and gypsum. The accumulation of these products induced stress  
4 which accelerated RDME loss in the last stage. J. Tang et al. (2018) determined the  
5 effect of sulphate concentration on concrete RDME. The RDME loss was higher when  
6 the sulphate concentration became higher as this produced more corrosion products.  
7 The study also showed that the RDME loss reduced with the use of coral reef sand as  
8 fine aggregate. Higher porosity of the sand had provided more space for ettringite and  
9 gypsum expansion.

10 Qi et al. (2017) used recycle coarse aggregate (RCA) to improve the concrete resistance  
11 against sulphate attack subjected to WDA. The incorporation of RCA at low  
12 replacement level of 30 % to 50 % reduced the RDME loss of concrete. The higher  
13 water absorption of RCA had reduced the effective w/b ratio of concrete and increased  
14 strength. The RDME loss increased for the replacement level higher than 50 %. The  
15 high porosity of RCA overwhelmed the positive effect and reduced the mechanical  
16 strength. The study also showed that the addition of mineral admixtures such as  
17 grounded blast furnace slag (GBFS) and fly ash (FA) reduced concrete permeability  
18 and hence loss of RDME. Y. Chen et al. (2016) demonstrated that the incorporation of  
19 50 % GBFS and 30 % FA reduced RDME loss. The GBFS and FA improved the RDME  
20 of concrete by 15.7 % and 18.7 % respectively as the cementitious matrix was densified  
21 by pozzolanic reaction and secondary hydration.

22 Gao et al. (2013) studied the effect of long-term flexural loading on concrete subjected  
23 to sulphate attack under influence of WDA. The application of 40 % ultimate flexural  
24 loading caused slight reduction of concrete RDME. Higher loading level at 60 %  
25 resulted in concrete cracking which escalated the deterioration and hence further

1 RDME loss. The incorporation of GBFS into concrete slightly improved the RDME,  
2 while FA incorporation showed better improvement. F. Chen et al. (2017) also found  
3 greater RDME loss at higher level of applied flexural loading. The difference of RDME  
4 between tension zone and compression zone was compared. The RDME loss in tension  
5 zone was greater than that in compression zone. The tension induced cracking to cause  
6 concrete degradation, whereas compression prevented cracking.

7

#### 8 *5.4 Remarks*

9 Concrete mechanical properties such as compressive strength, tensile strength and  
10 elastic modulus are damaged under the influence of WDA in marine environment. The  
11 deteriorating effect is mainly caused by sulphate attack and salt crystallization. The loss  
12 of mechanical properties is also highly dependent on applied external stress. Tensile  
13 stress induces concrete cracking, while compressive stress closes concrete pore to  
14 reduce damage. The incorporation of GBFS and FA improves the mechanical properties  
15 and shields concrete against weathering.

16 The published literature concerning the effect of WDA on tensile strength is limited.  
17 The tensile strength is perceived to be closely related to compressive strength as  
18 concrete having high compressive strength usually possesses higher tensile strength.  
19 This leads to misconception that WDA has similar impact on tensile strength to that on  
20 compressive strength. It is recommended to further conduct research in this area since  
21 their relationship is not linear.

22

### 23 **6. Corrosion resistance of concrete exposed to WDA**

24 In wetting-drying marine environment, concrete resistance to corrosion of embedded  
25 steel is important for ensuring long-term durability. The deterioration process is divided



1 into two stages which are de-passivation and propagation stages (Chrisp et al., 2002).  
2 The de-passivation refers to the loss of steel passivity as a result of concrete property  
3 change in response to exposure of aggressive environment. The de-passivation of steel  
4 is caused by chloride attack and carbonation. This subsequently initiates the corrosion  
5 of steel and causes volume expansion. The concrete experiences damages such as  
6 cracking, reduction of bonding strength with steel and loss of steel mechanical  
7 properties throughout the propagation stage. This consequently leads to issues of  
8 structural safety, high maintenance cost and reduction of building lifespan. Therefore,  
9 the corrosion resistance of concrete subjected to WDA is reviewed.

10

11 *6.1 Review of published works*

12 Chrisp et al. (2002) studied the mechanism of WDA on corrosion resistance of concrete  
13 through determination of electrical conductivity in cover zone. The conductivity of  
14 concrete in wet condition was higher than that in dry condition. The difference of  
15 conductivity between two conditions diminished with increasing depth into concrete  
16 until it became zero. The conductivity in this zone was influenced by WDA and was  
17 referred as convection zone of concrete where maximum content of chloride was found.  
18 The convection zone of concrete extended to 30 mm in wetting-drying environment.  
19 Ye, Jin, Jin, et al. (2016) determined the chloride ingress of concrete subjected to salt  
20 fog WDA whereby a solution containing 5 % NaCl was sprayed to wet concrete  
21 periodically. The chloride ingress depended on orientation of concrete exposure surface.  
22 The highest chloride ingress was found on concrete when the surface was placed  
23 perpendicularly to the salt fog spray. The convection zone was found at the depth  
24 ranged between 5 and 12 mm.

1 J. Wu et al. (2016) studied the influence of combined WDA and compressive loading  
2 on chloride ingress. The WDA accelerated moisture transfer, by which chloride  
3 accumulated during drying and migrated further into concrete during wetting. The  
4 chloride ingress reduced with increase of compressive loading as this closed concrete  
5 pores and reduced permeability. Paul et al. (2016) performed rapid chloride migration  
6 test to investigate the effect of crack on chloride ingress of strain hardening cement  
7 composite (SHCC). Cracks were formed on beam specimens by bending at deformation  
8 level of 33 % and 66 % with respect to ultimate deformation. The chloride migration  
9 coefficient of SHCC increased with deformation level. The enlarged crack width of  
10 SHCC at higher deformation level provided more means of chloride penetration.

11 Effect of chloride ingress on mortar was also evaluated by Backus et al. (2013). Lower  
12 concentration of chloride was found on the mortar surface, but peak value was  
13 encountered at 5 mm depth inside the mortar after being exposed to WDA for 24 weeks.  
14 Occurrence of carbonation reduced the pH value of mortar which released bound  
15 chloride into free form at the carbonation front. H. Chang (2017) evaluated the chloride  
16 binding capacity of cement paste in three conditions which were C-I (carbonated and  
17 then exposed to chloride), C-II (exposed to chloride and then carbonated) and C-III  
18 (chloride introduced during paste mixing and then carbonated). The chloride binding  
19 capacity of paste reduced in the order C-I, C-III and C-II. In both C-II and C-III,  
20 chloride was absorbed into cement paste and then exposed to carbonation. Carbonation  
21 occurred to decompose Friedel's salt to form free chloride which substantially increased  
22 chloride content. C-S-H also lost its chloride binding capacity due to drop of pH value  
23 and change of surface charge. Cement paste in C-I exhibited higher binding capacity  
24 because carbonation occurred prior to chloride contact and hence it was not affected by

1 carbonation. In actual situation, C–II is more representative of concrete deterioration,  
2 showing more acute damage caused by mutual effect of chloride attack and carbonation.  
3 Backus et al. (2013) also studied and found that the incorporation of mineral admixtures  
4 such as GBFS, FA and SF improved chloride penetration resistance. The mineral  
5 admixtures increased chloride binding capacity of mortar. Qi et al. (2018) demonstrated  
6 that the incorporation of GBFS and FA refined and densified concrete pore structure  
7 through pozzolanic reaction. Simčič et al. (2015) showed that the use of FA with 20 %  
8 and 50 % replacement level provided filling effect on concrete through pozzolanic  
9 reaction. Although mineral admixture reduced chloride ingress by refinement of pore,  
10 considerable amount of chloride could accumulate on concrete surface (Ye, Jin, Jin, et  
11 al., 2016). This gradually induced high concentration gradient which facilitated  
12 chloride ingress in the long term.

13

14 *6.2 Remarks*

15 Concrete corrosion resistance is aggravated by WDA. The WDA promotes more rapid  
16 transfer of moisture into concrete to accelerate chloride penetration. Convection zone  
17 is created inside concrete with the surface exposed to WDA. A long-term exposure to  
18 WDA increases the depth of convection zone and causes loss of concrete cover. The  
19 synergy between chloride attack and carbonation reduces chloride binding capacity,  
20 leading to more severe corrosion. The addition of mineral admixtures such as GGBS,  
21 FA and SF can improve concrete shielding against chloride ingress. The admixtures  
22 refine and densify pore structure of concrete through pozzolanic reaction. But, chloride  
23 accumulates on concrete surface and increases its concentration gradient, jeopardizing  
24 long-term corrosion resistance of concrete.

25

## 1 **7. Perspective of research trend**

2 Research gaps and limitations related to concrete deterioration under the influence of  
3 WDA are identified. The research trends are recommended as follows.

- 4 1. The current research focuses on a single deterioration mechanism and multiple  
5 mechanisms but restricted to chemical types such as between chloride attack and  
6 carbonation as well as between chloride attack and sulphate attack. Physical  
7 weathering of concrete can occur simultaneously to cause scaling and cracking of  
8 its surface, degrading the shielding layer and making it more susceptible to chemical  
9 deterioration. For example, salt crystallization and wave erosion cause significant  
10 physical and mechanical damages on concrete. It is recommended to perform  
11 research on synergy between physical and chemical deteriorations of concrete.
- 12 2. Researchers have performed experiments to simulate wetting-drying environment,  
13 of which the behaviors of WDA vary differently from their respective laboratory  
14 works. Validation has not been conducted to prove the relevance and accuracy of  
15 the simulated environment. It is suggested to also conduct experiments based on  
16 real time wetting-drying phenomenon, so as to establish the relationship between  
17 laboratory and actual conditions.
- 18 3. The change of concrete physical properties is a part of deterioration process. The  
19 current research is performed using simple test specimens such as cubes and prisms.  
20 The volume expansion and mass change of concrete can cause deformation in  
21 structure, leading to problems in the aspects of stability and integrity. Future  
22 research is recommended to expand to larger scale of testing to match the actual  
23 situation in the field.
- 24 4. Further investigation is recommended to study the effect of WDA on tensile  
25 strength of concrete as the research is limited in the current literature.

1 5. The incorporation of mineral admixtures such as GBFS, FA and SF reduces  
2 concrete permeability. It also accumulates and increases chloride concentration  
3 gradient on concrete surface to facilitate chloride penetration. The impact on the  
4 long-term corrosion resistance of concrete is not comprehensively studied. More  
5 extensive research which includes longer test period is recommended.

6

## 7 **8. Conclusion**

8 The impacts of WDA on marine concrete are reviewed. Concrete deterioration  
9 mechanisms are identified and the influences of WDA on the processes are discussed.  
10 This research provides deeper insight into the effect of WDA on the physical and  
11 mechanical properties as well as on corrosion resistance of concrete. Upon the review,  
12 the following conclusions can be drawn.

13 1. The WDA aggravates chemical deteriorations of concrete such as chloride attack,  
14 sulphate attack and carbonation. It increases the intrusion of harmful substances by  
15 accumulating them at the wetting front, resulting in steep concentration gradient  
16 which increases the diffusion. Physical concrete deterioration includes salt  
17 crystallization, freezing-thawing action and wave erosion. In the actual  
18 environment, synergy exists between two mechanisms in which the physical  
19 mechanism has a catalytic effect on exacerbation of chemical deterioration.

20 2. Concrete deterioration is greatly influenced by wetting-drying characteristics, ion  
21 type, ion concentration, temperature and external loading. The wetting and drying  
22 periods used by researchers range from 0.3–1080 hours/cycle and 0.7–1172  
23 hours/cycle respectively. A drying period with a temperature range of 45–80 °C is  
24 also used to accelerate water evaporation, but it can change the degradation process  
25 such as micro-cracking in concrete. The use of composite solution made from NaCl,

1 Na<sub>2</sub>SO<sub>4</sub> and MgSO<sub>4</sub> better simulates the actual environment. The damage caused is  
2 less severe than the individual component due to mutual interaction. External tensile  
3 action develops crack to speed up intrusion of deleterious compounds, while the  
4 external compressive action prevents the intrusion by closing the concrete pores and  
5 cracks.

6 3. The WDA, together with sulphate attack and salt crystallization, has a significant  
7 impact on the physical properties of concrete. The deterioration begins with  
8 formation of efflorescence, and is followed by delamination of mortar and coarse  
9 aggregate detachment, which eventually leads to disintegration of concrete.  
10 Concrete can experience volume expansion up to 1.6 %. The mass change of  
11 concrete depends on its strength, wherein strong concrete experiences mass increase  
12 whilst weak concrete suffers loss.

13 4. Mechanical properties of concrete deteriorate progressively with WDA. The main  
14 deterioration mechanisms are salt crystallization and sulphate attack. Internal stress  
15 exerted by the formation of salt crystals, ettringite and gypsum damages the  
16 concrete. But, filling of pores by the products can cause a temporary strengthening  
17 of concrete. The addition of mineral admixture such as GBFS and FA can minimize  
18 the strength degradation.

19 5. Corrosion resistance of concrete is deleteriously affected by chloride attack and  
20 carbonation. WDA promotes increased transfer of moisture which increases  
21 chloride intrusion. This expands the depth of chloride convection zone in concrete  
22 which causes a significant loss of cover and accelerates the corrosion of steel.  
23 Although the addition of mineral admixtures improves concrete resistance against  
24 corrosion, its refining effect accumulates chloride and increases the concentration  
25 to accelerate intrusion in the long-term.

1 Shortfalls exist in the laboratory tests for duplicating WDA of the actual environment.  
2 Disparity of test results ensues due to simulation difficulties and lack of unified  
3 guidelines. Nevertheless, the findings still can provide a promising trend in the WDA  
4 research. The laboratory simulation of WDA can be further improved with the  
5 acquisition of large-scale field data.

6

## 7 **Acknowledgement**

8 The authors wish to acknowledge the Novakey Developer Sdn Bhd for providing  
9 financial support to perform this research project. Special thanks to Professor Lau  
10 Hieng Ho and Mr. Ting Seng Kung for their encouragement in the research.

11

## 12 **Reference**

- 13 ACI211.1-91. 2002. Standard Practice for Selecting Proportions for Normal, Heavyweight and  
14 Mass Concrete.
- 15 Alexander, M., & Nganga, G. 2016. Introduction: Importance of marine concrete structures  
16 and durability design *Marine Concrete Structures* (pp. 1-13): Elsevier.
- 17 Alhozaimy, A., Hussain, R. R., Al-Zaid, R., & Al-Negheimish, A. 2012. Coupled effect of ambient  
18 high relative humidity and varying temperature marine environment on corrosion of  
19 reinforced concrete. *Construction and Building Materials*, 28(1), 670-679.
- 20 Apostolopoulos, C. A., Koulouris, K. F., & Apostolopoulos, A. C. 2019. Correlation of Surface  
21 Cracks of Concrete due to Corrosion and Bond Strength (between Steel Bar and  
22 Concrete). *Advances in Civil Engineering*, 2019.
- 23 Arns, A., Dangendorf, S., Jensen, J., Talke, S., Bender, J., & Pattiaratchi, C. 2017. Sea-level rise  
24 induced amplification of coastal protection design heights. *Scientific reports*, 7, 40171.
- 25 Aygörmez, Y., Canpolat, O., Al-mashhadani, M. M., & Uysal, M. 2020. Elevated temperature,  
26 freezing-thawing and wetting-drying effects on polypropylene fiber reinforced  
27 metakaolin based geopolymer composites. *Construction and Building Materials*, 235,  
28 117502.
- 29 Backus, J., McPolin, D., Basheer, M., Long, A., & Holmes, N. 2013. Exposure of mortars to cyclic  
30 chloride ingress and carbonation. *Advances in Cement Research*, 25(1), 3-11.
- 31 BS8110. 1997. Structural use of concrete - Part 1: code of practice for design and construction.  
32 London: British Standards Institution.
- 33 C1012, A. S. 2018. Standard test method for length change of hydraulic-cement mortars  
34 exposed to a sulfate solution. West Conshohocken, PA: American Society for Testing  
35 and Materials.
- 36 Cai, H., & Liu, X. 1998. Freeze-thaw durability of concrete: ice formation process in pores.  
37 *Cement and Concrete Research*, 28(9), 1281-1287.

- 1 Cai, X., He, Z., Tang, S., & Chen, X. 2016. Abrasion erosion characteristics of concrete made  
2 with moderate heat Portland cement, fly ash and silica fume using sandblasting test.  
3 *Construction and Building Materials*, 127, 804-814.
- 4 Cai, X., Zhang, Y., Gao, L., Wang, J., & Peng, H. 2019. Deterioration of cement asphalt pastes  
5 with polymer latexes and expansive agent under sulfate attack and wetting-drying  
6 cycles. *Engineering Failure Analysis*, 104252.
- 7 Chang, C.-F., & Chen, J.-W. 2006. The experimental investigation of concrete carbonation  
8 depth. *Cement and Concrete Research*, 36(9), 1760-1767.
- 9 Chang, H. 2017. Chloride binding capacity of pastes influenced by carbonation under three  
10 conditions. *Cement and Concrete Composites*, 84, 1-9.
- 11 Chang, H., Mu, S., & Feng, P. 2018. Influence of carbonation on “maximum phenomenon” in  
12 surface layer of specimens subjected to cyclic drying-wetting condition. *Cement and  
13 Concrete Research*, 103, 95-109.
- 14 Chen, F., Gao, J., Qi, B., Shen, D., & Li, L. 2017. Degradation progress of concrete subject to  
15 combined sulfate-chloride attack under drying-wetting cycles and flexural loading.  
16 *Construction and Building Materials*, 151, 164-171.
- 17 Chen, Y., Gao, J., Tang, L., & Li, X. 2016. Resistance of concrete against combined attack of  
18 chloride and sulfate under drying–wetting cycles. *Construction and Building Materials*,  
19 106, 650-658.
- 20 Chen, Z., Lee, Y., Cho, H., Lee, H., & Lim, S. 2019. Improvement in Carbonation Resistance of  
21 Portland Cement Mortar Incorporating  $\gamma$ -Dicalcium Silicate. *Advances in Materials  
22 Science and Engineering*, 2019.
- 23 Cheng, S., Shui, Z., Gao, X., Lu, J., Sun, T., & Yu, R. 2020. Degradation progress of Portland  
24 cement mortar under the coupled effects of multiple corrosive ions and drying-  
25 wetting cycles. *Cement and Concrete Composites*, 103629.
- 26 Cheng, S., Shui, Z., Gao, X., Yu, R., Sun, T., Guo, C., & Huang, Y. 2020. Degradation mechanisms  
27 of Portland cement mortar under seawater attack and drying-wetting cycles.  
28 *Construction and Building Materials*, 230, 116934.
- 29 Chilana, L., Bhatt, A. H., Najafi, M., & Sattler, M. 2016. Comparison of carbon footprints of  
30 steel versus concrete pipelines for water transmission. *Journal of the Air & Waste  
31 Management Association*, 66(5), 518-527.
- 32 Chrisp, T., McCarter, W., Starrs, G., Basheer, P., & Blewett, J. 2002. Depth-related variation in  
33 conductivity to study cover-zone concrete during wetting and drying. *Cement and  
34 Concrete Composites*, 24(5), 415-426.
- 35 Dandapat, R., & Deb, A. 2016. A probability based model for the erosive wear of concrete by  
36 sediment bearing water. *Wear*, 350, 166-181.
- 37 Effler, S. W., Schimel, K., & Millero, F. J. 1986. Salinity, chloride, and density relationships in  
38 ion enriched Onondaga Lake, NY. *Water, Air, and Soil Pollution*, 27(1-2), 169-180.
- 39 Farahani, A., Taghaddos, H., & Shekarchi, M. 2015. Prediction of long-term chloride diffusion  
40 in silica fume concrete in a marine environment. *Cement and Concrete Composites*,  
41 59, 10-17.
- 42 Fu, C., Ye, H., Jin, X., Yan, D., Jin, N., & Peng, Z. 2016. Chloride penetration into concrete  
43 damaged by uniaxial tensile fatigue loading. *Construction and Building Materials*, 125,  
44 714-723.
- 45 Gao, J., Yu, Z., Song, L., Wang, T., & Wei, S. 2013. Durability of concrete exposed to sulfate  
46 attack under flexural loading and drying–wetting cycles. *Construction and Building  
47 Materials*, 39, 33-38.
- 48 Garzón-Roca, J., Sena-Cruz, J. M., Fernandes, P., & Xavier, J. 2015. Effect of wet-dry cycles on  
49 the bond behaviour of concrete elements strengthened with NSM CFRP laminate  
50 strips. *Composite Structures*, 132, 331-340.



- 1 Gentilini, C., Franzoni, E., Bandini, S., & Nobile, L. 2012. Effect of salt crystallisation on the  
2 shear behaviour of masonry walls: an experimental study. *Construction and Building*  
3 *Materials*, 37, 181-189.
- 4 Gökceoğlu, C., Ulusay, R., & Sönmez, H. 2000. Factors affecting the durability of selected weak  
5 and clay-bearing rocks from Turkey, with particular emphasis on the influence of the  
6 number of drying and wetting cycles. *Engineering Geology*, 57(3-4), 215-237.
- 7 Guan, K. K. 2011. Surface and ambient air temperatures associated with different ground  
8 material: a case study at the University of California, Berkeley. *Environmental Science*,  
9 196, 1-14.
- 10 Hamze, Y. 2014. Concrete durability in harsh environmental conditions exposed to freeze thaw  
11 cycles. *Physics Procedia*, 55, 265-270.
- 12 Haynes, H., O'Neill, R., Neff, M., & Kumar Mehta, P. 2010. Salt weathering of concrete by  
13 sodium carbonate and sodium chloride. *ACI Materials Journal*, 107(3), 258.
- 14 Haynes, H., O'Neill, R., Neff, M., & Mehta, P. K. 2008. Salt weathering distress on concrete  
15 exposed to sodium sulfate environment. *ACI Materials Journal*, 105(1), 35.
- 16 Horszczaruk, E. 2005. Abrasion resistance of high-strength concrete in hydraulic structures.  
17 *Wear*, 259(1-6), 62-69.
- 18 Horszczaruk, E. 2009. Hydro-abrasive erosion of high performance fiber-reinforced concrete.  
19 *Wear*, 267(1-4), 110-115.
- 20 Hoy, M., Rachan, R., Horpibulsuk, S., Arulrajah, A., & Mirzababaei, M. 2017. Effect of wetting–  
21 drying cycles on compressive strength and microstructure of recycled asphalt  
22 pavement–Fly ash geopolymer. *Construction and Building Materials*, 144, 624-634.
- 23 Ibrion, M., Paltrinieri, N., & Nejad, A. R. 2020. Learning from failures: Accidents of marine  
24 structures on Norwegian continental shelf over 40 years time period. *Engineering*  
25 *Failure Analysis*, 104487.
- 26 IRC44. 2008. Guidelines for cement concrete mix design for pavement. Delhi: Indian Road  
27 Congress.
- 28 Jerman, M., Scheinherrová, L., Medved', I., Krejsová, J., Doleželová, M., Bezdička, P., & Černý,  
29 R. 2019. Effect of cyclic wetting and drying on microstructure, composition and length  
30 changes of lime-based plasters. *Cement and Concrete Composites*, 104, 103411.
- 31 Jiang, L., & Niu, D. 2016. Study of deterioration of concrete exposed to different types of  
32 sulfate solutions under drying-wetting cycles. *Construction and Building Materials*,  
33 117, 88-98.
- 34 Jiang, L., Niu, D., Yuan, L., & Fei, Q. 2015. Durability of concrete under sulfate attack exposed  
35 to freeze–thaw cycles. *Cold Regions Science and Technology*, 112, 112-117.
- 36 Khan, M. U., Ahmad, S., & Al-Gahtani, H. J. 2017. Chloride-induced corrosion of steel in  
37 concrete: an overview on chloride diffusion and prediction of corrosion initiation time.  
38 *International Journal of Corrosion*, 2017.
- 39 Kim, Y.-Y., Lee, K.-M., Bang, J.-W., & Kwon, S.-J. 2014. Effect of W/C ratio on durability and  
40 porosity in cement mortar with constant cement amount. *Advances in Materials*  
41 *Science and Engineering*, 2014.
- 42 Kryžanowski, A., Mikoš, M., Šušteršič, J., & Planinc, I. 2009. Abrasion resistance of concrete in  
43 hydraulic structures. *ACI Materials Journal*, 106(4), 349-356.
- 44 Kuosa, H., Ferreira, R., Holt, E., Leivo, M., & Vesikari, E. 2014. Effect of coupled deterioration  
45 by freeze–thaw, carbonation and chlorides on concrete service life. *Cement and*  
46 *Concrete Composites*, 47, 32-40.
- 47 Lee, B. Y., & Kurtis, K. E. 2017. Effect of pore structure on salt crystallization damage of  
48 cement-based materials: Consideration of w/b and nanoparticle use. *Cement and*  
49 *Concrete Research*, 98, 61-70.
- 50 Lehigh. (2011). Chloride and salinity. Retrieved from  
51 <http://www.ei.lehigh.edu/envirosoci/watershed/wq/wqbackground/chloridebg.html>

- 1 Li, J., Xie, F., Zhao, G., & Li, L. 2020. Experimental and numerical investigation of cast-in-situ  
2 concrete under external sulfate attack and drying-wetting cycles. *Construction and*  
3 *Building Materials*, 249, 118789.
- 4 Liu, F., You, Z., Yang, X., & Wang, H. 2018. Macro-micro degradation process of fly ash concrete  
5 under alternation of freeze-thaw cycles subjected to sulfate and carbonation.  
6 *Construction and Building Materials*, 181, 369-380.
- 7 Liu, Y.-W., Yen, T., & Hsu, T.-H. 2006. Abrasion erosion of concrete by water-borne sand.  
8 *Cement and Concrete Research*, 36(10), 1814-1820.
- 9 Malheiro, R. L. M. C., Camões, A., Ferreira, R. M., Meira, G., & Amorim, M. (2014). *Effect of*  
10 *carbonation on the chloride diffusion of mortar specimens exposed to cyclic wetting*  
11 *and drying*. Paper presented at the XIII DBMC, International Conference on Durability  
12 of Building Materials and Components.
- 13 Medeiros, M., Gobbi, A., Réus, G., & Helene, P. 2013. Reinforced concrete in marine  
14 environment: Effect of wetting and drying cycles, height and positioning in relation to  
15 the sea shore. *Construction and Building Materials*, 44, 452-457.
- 16 Mehta, P. K., & Monteiro, P. J. 2006. *Concrete microstructure, properties and materials*, The  
17 *McGraw-Hill Companies, Inc.*
- 18 Miao, C., Mu, R., Tian, Q., & Sun, W. 2002. Effect of sulfate solution on the frost resistance of  
19 concrete with and without steel fiber reinforcement. *Cement and Concrete Research*,  
20 32(1), 31-34.
- 21 Mohammadhosseini, H., Yatim, J. M., Sam, A. R. M., & Awal, A. A. 2017. Durability  
22 performance of green concrete composites containing waste carpet fibers and palm  
23 oil fuel ash. *Journal of cleaner production*, 144, 448-458.
- 24 Müllauer, W., Beddoe, R. E., & Heinz, D. 2013. Sulfate attack expansion mechanisms. *Cement*  
25 *and Concrete Research*, 52, 208-215.
- 26 Nadelman, E., & Kurtis, K. 2019. Durability of Portland-limestone cement-based materials to  
27 physical salt attack. *Cement and Concrete Research*, 125, 105859.
- 28 Niu, D.-t., Ma, R., Wang, J.-b., & Xu, S.-h. 2015. Experiment study on the failure mechanism of  
29 dry-mix shotcrete under the combined actions of sulfate attack and drying-wetting  
30 cycles. *Construction and Building Materials*, 81, 74-80.
- 31 NOAA. (2018). How much water is in the ocean. Retrieved from  
32 <https://oceanservice.noaa.gov/facts/oceanwater.html>
- 33 NOAA. (2019). Tides and water level. Retrieved from  
34 [https://oceanservice.noaa.gov/education/tutorial\\_tides/tides07\\_cycles.html](https://oceanservice.noaa.gov/education/tutorial_tides/tides07_cycles.html)
- 35 Oh, B. H., & Jang, S. Y. 2007. Effects of material and environmental parameters on chloride  
36 penetration profiles in concrete structures. *Cement and Concrete Research*, 37(1), 47-  
37 53.
- 38 Paul, S. C., van Zijl, G. P., Babafemi, A. J., & Tan, M. J. 2016. Chloride ingress in cracked and  
39 uncracked SHCC under cyclic wetting-drying exposure. *Construction and Building*  
40 *Materials*, 114, 232-240.
- 41 Plum, D., & Xufei, F. 1996. A rock and a hard place. *International water power & dam*  
42 *construction*, 48(7), 30-33.
- 43 Pratolongo, P., Leonardi, N., Kirby, J. R., & Plater, A. 2019. Temperate coastal wetlands:  
44 morphology, sediment processes, and plant communities *Coastal Wetlands* (pp. 105-  
45 152): Elsevier.
- 46 Qi, B., Gao, J., Chen, F., & Shen, D. 2017. Evaluation of the damage process of recycled  
47 aggregate concrete under sulfate attack and wetting-drying cycles. *Construction and*  
48 *Building Materials*, 138, 254-262.
- 49 Qi, B., Gao, J., Chen, F., & Shen, D. 2018. Chloride penetration into recycled aggregate concrete  
50 subjected to wetting-drying cycles and flexural loading. *Construction and Building*  
51 *Materials*, 174, 130-137.

- 1 Rao, S., Reddy, B., & Muttharam, M. 2001. The impact of cyclic wetting and drying on the  
2 swelling behaviour of stabilized expansive soils. *Engineering Geology*, 60(1-4), 223-  
3 233.
- 4 Rosenqvist, M., Pham, L.-W., Terzic, A., Fridh, K., & Hassanzadeh, M. 2017. Effects of  
5 interactions between leaching, frost action and abrasion on the surface deterioration  
6 of concrete. *Construction and Building Materials*, 149, 849-860.
- 7 Rozière, E., Loukili, A., El Hachem, R., & Grondin, F. 2009. Durability of concrete exposed to  
8 leaching and external sulphate attacks. *Cement and Concrete Research*, 39(12), 1188-  
9 1198.
- 10 Sahmaran, M., Erdem, T., & Yaman, I. 2007. Sulfate resistance of plain and blended cements  
11 exposed to wetting–drying and heating–cooling environments. *Construction and  
12 Building Materials*, 21(8), 1771-1778.
- 13 Saillio, M., Baroghel-Bouny, V., & Barberon, F. 2014. Chloride binding in sound and carbonated  
14 cementitious materials with various types of binder. *Construction and Building  
15 Materials*, 68, 82-91.
- 16 Samimi, K., Kamali-Bernard, S., & Maghsoudi, A. A. 2018. Durability of self-compacting  
17 concrete containing pumice and zeolite against acid attack, carbonation and marine  
18 environment. *Construction and Building Materials*, 165, 247-263.
- 19 Simčič, T., Pejovnik, S., De Schutter, G., & Bosiljkov, V. B. 2015. Chloride ion penetration into  
20 fly ash modified concrete during wetting–drying cycles. *Construction and Building  
21 Materials*, 93, 1216-1223.
- 22 Song, H.-W., Lee, C.-H., & Ann, K. Y. 2008. Factors influencing chloride transport in concrete  
23 structures exposed to marine environments. *Cement and Concrete Composites*, 30(2),  
24 113-121.
- 25 Standard, C. 2009. Test Methods of Long-Term Performance and Durability of Ordinary  
26 Concrete, GB/T 50082–2009.
- 27 Steiger, M., & Asmussen, S. 2008. Crystallization of sodium sulfate phases in porous materials:  
28 the phase diagram Na<sub>2</sub>SO<sub>4</sub>–H<sub>2</sub>O and the generation of stress. *Geochimica et  
29 Cosmochimica Acta*, 72(17), 4291-4306.
- 30 Sun, W., Mu, R., Luo, X., & Miao, C. 2002. Effect of chloride salt, freeze–thaw cycling and  
31 externally applied load on the performance of the concrete. *Cement and Concrete  
32 Research*, 32(12), 1859-1864.
- 33 Tan, C. L., Wong, N. H., & Jusuf, S. K. 2014. Effects of vertical greenery on mean radiant  
34 temperature in the tropical urban environment. *Landscape and Urban Planning*, 127,  
35 52-64.
- 36 Tang, J., Cheng, H., Zhang, Q., Chen, W., & Li, Q. 2018. Development of properties and  
37 microstructure of concrete with coral reef sand under sulphate attack and drying-  
38 wetting cycles. *Construction and Building Materials*, 165, 647-654.
- 39 Tang, W., Li, S., Lu, Y., & Li, Z. 2020. Combined effects of wetting–drying cycles and sustained  
40 load on the behaviour of FRP-strengthened RC beams. *Engineering Structures*, 213,  
41 110570.
- 42 Thaulow, N., & Sahu, S. 2004. Mechanism of concrete deterioration due to salt crystallization.  
43 *Materials Characterization*, 53(2-4), 123-127.
- 44 Trenhaile, A. S. 2011. Cliffs and rock coasts.
- 45 Val, D. V., & Stewart, M. G. 2003. Life-cycle cost analysis of reinforced concrete structures in  
46 marine environments. *Structural safety*, 25(4), 343-362.
- 47 Valipour, M., Shekarchi, M., & Arezoumandi, M. 2017. Chlorine diffusion resistivity of  
48 sustainable green concrete in harsh marine environments. *Journal of cleaner  
49 production*, 142, 4092-4100.

- 1 Wang, Y., Wu, L., Wang, Y., Liu, C., & Li, Q. 2018. Effects of coarse aggregates on chloride  
2 diffusion coefficients of concrete and interfacial transition zone under experimental  
3 drying-wetting cycles. *Construction and Building Materials*, 185, 230-245.
- 4 Wang, Z., Zeng, Q., Wang, L., Yao, Y., & Li, K. 2014. Corrosion of rebar in concrete under cyclic  
5 freeze-thaw and Chloride salt action. *Construction and Building Materials*, 53, 40-47.
- 6 Wu, J., Li, H., Wang, Z., & Liu, J. 2016. Transport model of chloride ions in concrete under loads  
7 and drying-wetting cycles. *Construction and Building Materials*, 112, 733-738.
- 8 Wu, Z., Wong, H., & Buenfeld, N. 2017. Transport properties of concrete after drying-wetting  
9 regimes to elucidate the effects of moisture content, hysteresis and microcracking.  
10 *Cement and Concrete Research*, 98, 136-154.
- 11 Ye, H., Jin, N., Jin, X., Fu, C., & Chen, W. 2016. Chloride ingress profiles and binding capacity of  
12 mortar in cyclic drying-wetting salt fog environments. *Construction and Building  
13 Materials*, 127, 733-742.
- 14 Ye, H., Jin, X., Fu, C., Jin, N., Xu, Y., & Huang, T. 2016. Chloride penetration in concrete exposed  
15 to cyclic drying-wetting and carbonation. *Construction and Building Materials*, 112,  
16 457-463.
- 17 Yu, X.-t., Chen, D., Feng, J.-r., & Zhang, Y. 2018. Behavior of mortar exposed to different  
18 exposure conditions of sulfate attack. *Ocean Engineering*, 157, 1-12.
- 19 Yu, Z., Chen, Y., Liu, P., & Wang, W. 2015. Accelerated simulation of chloride ingress into  
20 concrete under drying-wetting alternation condition chloride environment.  
21 *Construction and Building Materials*, 93, 205-213.
- 22 Yuan, J., Liu, Y., Tan, Z., & Zhang, B. 2016. Investigating the failure process of concrete under  
23 the coupled actions between sulfate attack and drying-wetting cycles by using X-ray  
24 CT. *Construction and Building Materials*, 108, 129-138.
- 25 Zhang, D., & Shao, Y. 2016. Effect of early carbonation curing on chloride penetration and  
26 weathering carbonation in concrete. *Construction and Building Materials*, 123, 516-  
27 526.
- 28 Zhang, D., Zhao, Y., Ueda, T., Li, X., & Xu, Q. 2016. CFRP strengthened RC beams with pre-  
29 strengthening non-uniform reinforcement corrosion subjected to post-strengthening  
30 wetting/drying cycles. *Engineering Structures*, 127, 331-343.
- 31 Zhang, H., Ji, T., & Liu, H. 2019. Performance evolution of the interfacial transition zone (ITZ)  
32 in recycled aggregate concrete under external sulfate attacks and dry-wet cycling.  
33 *Construction and Building Materials*, 229, 116938.
- 34 Zhang, J., Gao, Y., & Han, Y. 2011. Interior humidity of concrete under dry-wet cycles. *Journal  
35 of Materials in Civil engineering*, 24(3), 289-298.
- 36 Zhang, J., Sun, M., Hou, D., & Li, Z. 2017. External sulfate attack to reinforced concrete under  
37 drying-wetting cycles and loading condition: numerical simulation and experimental  
38 validation by ultrasonic array method. *Construction and Building Materials*, 139, 365-  
39 373.
- 40 Zhang, R., Castel, A., & François, R. 2010. Concrete cover cracking with reinforcement  
41 corrosion of RC beam during chloride-induced corrosion process. *Cement and  
42 Concrete Research*, 40(3), 415-425.
- 43 Zhongya, Z., Xiaoguang, J., & Wei, L. 2019. Long-term behaviors of concrete under low-  
44 concentration sulfate attack subjected to natural variation of environmental climate  
45 conditions. *Cement and Concrete Research*, 116, 217-230.

46