



Review

Mechanism of Electronegativity Heterojunction of Nanometer Amorphous-Boron on Crystalline Silicon: An Overview

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Version January 11, 2021 submitted to Crystals

Abstract: The discovery of the extremely shallow amorphous boron-crystalline silicon heterojunction occurred during the development of highly sensitive, hard and robust detectors for low-penetration-depth ionizing radiation, such as ultraviolet photons and low-energy electrons (below 1 keV). For many years it was believed that the junction created by the chemical vapor deposition of amorphous boron on n-type crystalline silicon was a shallow p-n junction, although experimental results could not provide evidence for such a conclusion. Only recently, quantum-mechanics based modelling revealed the unique nature and the formation mechanism of this new junction. Here, we review the initiation and the history of understanding the a-B/c-Si interface (henceforth called the “boron-silicon junction”), as well as its importance for the microelectronics industry, followed by the scientific perception of the new junctions. Future developments and possible research directions are also discussed.

Keywords: rectifying junction; photodiode; chemical vapor deposition; first principle molecular dynamics; electronegativity

1. Introduction: Initiation and history of boron-silicon junctions and importance of Si-based junctions/diodes in microelectronics

The first report about an ultra-shallow rectifying junction (diode) created by a pure boron atmospheric/low-pressure chemical vapor deposition (AP/LPCVD) on crystalline n-type silicon surface was reported in 2006 [1]. Initially, the application which led to the development of this novel rectifying junction was: a linear and high Q-factor varactor diode designed for the capacitance tuning of frequency in RF circuits [2,3]. The demonstrated good performance in the varactor application did not attract the expected attention. Fortunately, in 2006 it became clear that a different field of applications would benefit even more from the excellent electrical properties of this extremely shallow junction. This junction would prove useful as an accurate, stable and reliable detector for low-penetration depth radiation such as UV light and low-energy electrons, which are applied in UV optical lithography and scanning electron microscopes.

Since 2006 a significant amount of research has been completed in the following directions: (1) optimization of the critical junction creation process, i.e. the chemical vapor deposition (CVD) of amorphous boron on n-type crystalline silicon (in a method called the “PureB” process); (2) device

29 characterization and design optimization for a variety of applications; and (3) rendering the PureB
30 process CMOS-compatible. Initially it was believed that the excellent electrical properties of the
31 junction—especially the very low saturation currents which are typical for deep p-n junctions—were
32 defined by the p⁺ delta-doping of the n-type substrate which simultaneously occurs during the
33 boron CVD process [1]. It was assumed that the saturation current was mainly dominated by the
34 hole injection from the p⁺ region into the n-substrate, as governed by the Gummel Number (G_E)
35 of this region. The high level of electron injection typically dominating the current in the Schottky
36 diode counterpart was suppressed, although the actual p⁺ region was only a few nanometers thin
37 [4]. However, it was difficult to explain the very high effective G_E , keeping in mind the limited
38 solubility of boron in silicon in the applied CVD temperature range from 500–700 °C [5]. Electrical
39 measurements, as presented in [6], showed injection currents as low as a few 10^{-20} A/ μm^2 , which was
40 comparable to those achieved in deep, heavily doped junctions. This corresponded with a G_E in the
41 order of 10^{14} – 10^{15} atm/cm², which was orders of magnitudes higher than what would be expected
42 from nm-shallow junctions formed by bulk-doping the silicon. In the 700 °C PureB process the actual
43 doping of the Si-substrate contributed to G_E by roughly 10^{12} atm/cm², as documented in [4]. In
44 diodes that were formed solely by such a doping of the Si-substrate, the total current would approach
45 Schottky diode-like values.

46 The idea of the delta-doped p⁺ layer having any significant role in the junction formation was
47 completely abandoned when it was demonstrated that similar excellent electrical properties could be
48 achieved by boron CVD on n-type crystalline silicon substrates at temperatures as low as 400 °C, at
49 which no doping of boron in silicon is expected [7].

50 Later on, in order to overcome the inconsistencies in the above-mentioned concept for the
51 junction formation mechanism, it was suggested that the thickness of the amorphous boron layer was
52 directly responsible for the junction behavior [1,4]. The bulk properties of the amorphous boron layer
53 that could lead to the suppression of the electron injection included either: (i) a very short electron
54 diffusion length and low electron mobility, which could cause quenching of the electron transport
55 [4], or (ii) a wider bandgap than that of the Si, as proposed and supported by simulations in [8].
56 However, experimental results showed that even for a 1-second (s) boron deposition, where not even
57 a monolayer of boron could be deposited, the junctions were reported to contain an equally high hole
58 injection for both 700 °C and 500 °C depositions [6]. A closer look at the dependency between the
59 thickness of the boron layer and the G_E showed that after creating a very thin boron layer with a full
60 coverage of the silicon substrate surface, the injection current stopped decreasing. This observation
61 contradicted to the previous conclusion about the G_E -boron thickness relation, and raised serious
62 doubt about the dominant role of bulk amorphous boron for junction creation [6,8].

63 Thus, when the delta-doped p⁺ layer and the as-deposited boron layer thickness could no longer
64 be considered dominant factors in the junction creation, what remained was the search for an answer
65 in the physics behind the boron-silicon interface. In [6] the following proposition was made: “Based
66 on experimental evidence, the effectively high Gummel Number of the p⁺ region, which provides
67 low saturation currents despite the shallowness of the junctions, was related to the formation of a
68 virtually complete surface coverage of acceptor states as an interface property of boron on Si.” In
69 a later publication [9] this idea was further developed: “The results can be explained with a simple
70 model assuming a monolayer of acceptor states at the interface that fills with electrons to give a
71 monolayer of fixed negative charge. Furthermore, it can be assumed that the high resistivity of the
72 very thin PureB layer acts as a semi-insulating layer allowing an inversion layer of holes to be built
73 up. (...) The monolayer of n-charge represents a very high electric field that binds the holes to the
74 interface and limits their mobility, similar to the way a vertical electrical field attenuates the inversion
75 layer mobility in MOS devices”. However, this explanation of the junction properties has two major
76 weaknesses:

- 77 1. There is no explanation as to where the “monolayer of acceptor states” providing “a monolayer
78 of fixed negative charge” originates from;

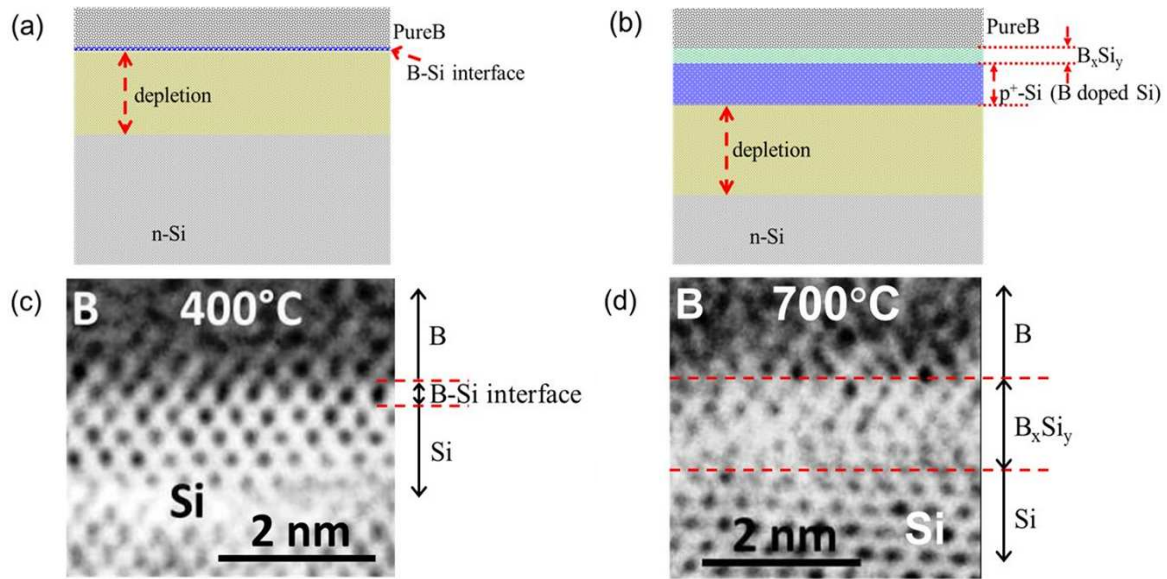


Figure 1. Schematic illustrations and HRTEM images of the cross-section of the layer stack: a) schematic illustration of the cross-section of the B-Si junction processed at 400 °C; b) Schematic illustration of the cross-section of the B-Si junction processed at 700 °C; c) HRTEM image of the cross-section of the B-Si junction processed at 400 °C; d) HRTEM image of the cross-section of the B-Si junction processed at 700 °C [10].

79 2. It is not clear what made the charge “fixed”.

80 Despite the fact that the junction formation was most probably correctly allocated, the failure to
 81 explain its mechanism led to a “dose of despair”. In [5] we read: “However, even if the chemical
 82 bonding structure of the interface was well known, translating it into an electrical structure is no
 83 straightforward task as can be appreciated evaluating the enormous number of studies devoted to
 84 understanding the metal-semiconductor interfaces of Schottky diodes”.

85 Then, in 2017 a completely new concept regarding junction formation with the PureB process
 86 was introduced [10]. It was proposed that this junction should not be considered a p-n type, and
 87 it should also not be assigned to any existing types of heterojunctions. That would clarify why the
 88 existing “instrumentarium” in semiconductor physics used to explain and predict the properties of
 89 known rectifying junctions could not be used successfully here. Instead, a deeper dive into solid-state
 90 physics and material science was proposed using a more powerful “scientific weapon”— the theory
 91 of quantum mechanics. An analysis of the junction formation was reported in [10], which concluded
 92 that the chemical interaction between the surface atoms of crystalline silicon and the first atomic layer
 93 of the as-deposited amorphous boron was the dominant factor leading to the formation of a depletion
 94 zone in the crystalline silicon originating from the surface. A first-principles quantum mechanics
 95 molecular dynamics simulation showed a very strong electric field across the a-B/c-Si interface
 96 systems where the charge transfer occurred mainly from the interface Si atoms to the neighboring
 97 B atoms. This electric field appeared to be responsible for the creation of a depletion zone in the
 98 n-silicon, resulting in a rectifying junction formation. A more detailed introduction of this hypothesis
 99 is provided in Section 3. Before that, in Section 2, information is provided on the PureB process and
 100 the most attractive electrical and optical characteristics of the boron-silicon junction as a radiation
 101 detector.

Table 1. Main characteristics and related parameters of an ideal silicon-based photodetector.

Main characteristics	Related diode parameters	Ideal parameter values
Responsivity	Passivation layer thickness	0
	Depletion depth	0
	Depletion width	Optimized by the photon attenuation length
Resolution	Dark current	0
	Shunt resistance	∞
Stability	Passivation Layer	Oxide-free, high-conductivity, protective
Operational speed	Series resistance	0
	Junction capacitance	0

2. The PureB process and characterization of the boron-silicon junction as a radiation detector

2.1. The PureB process and temperature effects

Amorphous pure boron (a-B) thin films can be deposited on crystalline Si substrates by using chemical vapor (e.g. cold-wall reactors or hot-wall furnaces) and physical vapor techniques (e.g. molecular beam epitaxy [11,12]) and sputtering [13] tools in ultra clean chambers with high purity gasses.

Chemical vapor deposition (CVD) of the precursor molecule diborane (B_2H_6) is mostly employed to obtain high-quality ultra-thin (2–10 nm) a-B films. The CVD tools operate in a range of 400–800 °C, with pressure ranging from tens of Torr to atmospheric pressure (760 Torr). For CVD at temperatures higher than 400 °C, the diborane molecules decompose at the surface of the substrate into gas-phase boron hydrates. The most common resulting species is BH_3 [14].

At the beginning of the deposition, the Si surface atoms have exposed dangling bonds to which the BH_3 molecules react, creating a Si-B bond and the release of hydrogen. If the Si dangling bonds are passivated with H atoms, the latter must be desorbed before any Si-B bonds can be formed. In fact, it has been observed that the presence of H_2 gas (generally used as a carrier and diluting gas) in the reactor reduces the a-B deposition rate [15]. When the coverage of the surface Si atoms by the BH_x species is accomplished, the BH_3 molecules react with the exposed B dangling bonds, which then generates B-B bonds (along with H/ H_2 release). The growth of the a-B film advances according to the adsorption-reaction mechanisms presented in [16].

Upon adsorption, the BH_3 molecules diffuse along the surface before forming a stable bond. At temperatures between 400–600 °C, their surface migration is limited and the a-B grows into islands with the surface reactive sites [16–18]. The diffusion is strongly temperature-dependent. Fig. 1 illustrates a schematic cross-section of the layer stack of the B-Si junction and the corresponding High-Resolution Tunneling Electron Microscope (HRTEM) images at deposition temperatures of 400 °C and 700 °C [10]. The HRTEM images confirm the lack of boron-silicide (B_xSi_y) layer for the samples obtained using the low temperature process (400 °C, Section 1), while for the samples prepared at the high temperature (700 °C, Section 1) a 1-2 nm-thick boron-silicide (B_xSi_y) can be observed. It was also reported that the silicide forms a uniform layer at deposition temperatures higher than 750 °C [15]. For deposition at ~ 500 °C, the a-B contains significant portions of hydrogen, due to incomplete precursor dissociation, which eventually coalesce into a rough and continuous film [16,17]. When the deposition temperatures are higher than 700 °C, the surface diffusivity of the adsorbed precursors is enhanced allowing a smooth, continuous film to be deposited of minimal thickness (a few nm) [16].

The diffusion of B atoms in bulk Si takes place during the CVD of amorphous B, but the diffusion becomes significant at temperatures above 750 °C. The experiments also showed that the diffusion rate of a-B at 600 °C is the same for both Si{0 0 1} and Si{1 1 1} surfaces, while at 800 °C, B atoms diffuse faster into the Si{0 0 1} subsurface [17].

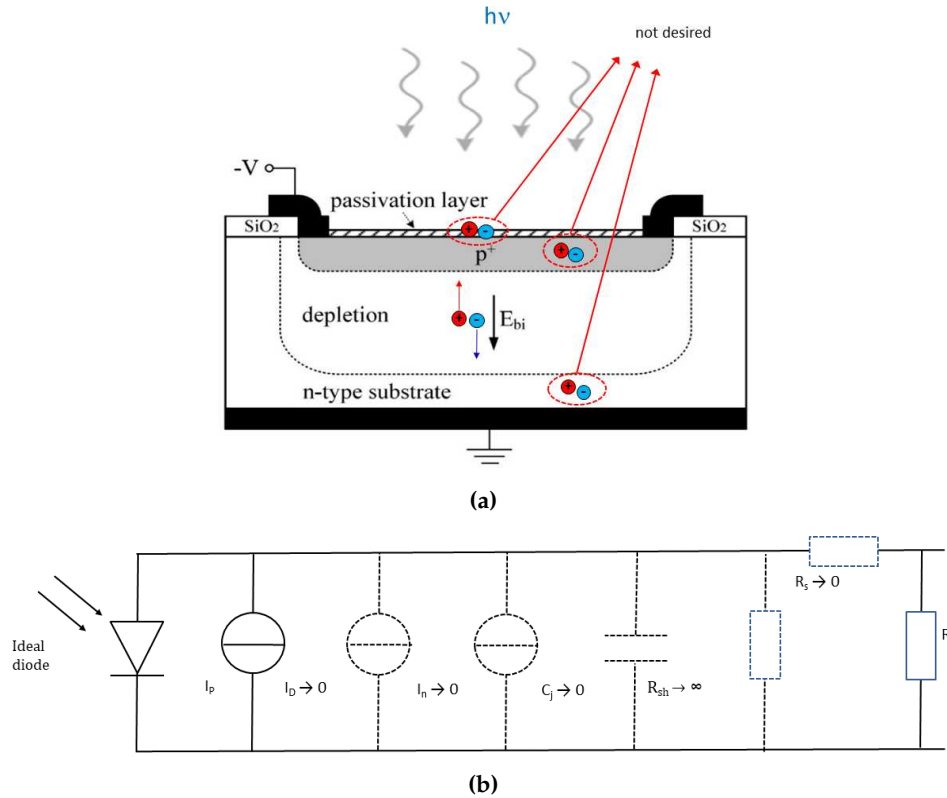


Figure 2. Vertical cross section (a), and electrical equivalent circuit (b) of a silicon p-n junction photodetector, where I_D is the dark current, I_p is the photogenerated current, I_n is the shot noise associated with the dark current, C_j is the junction capacitance, R_{sh} is the shunt resistance, R_s is the series resistance and R_L is the load resistance [19].

138 2.2. Characterization of the boron-silicon junction as a radiation detector

139 In the Introduction we revealed the main application driving the development of the PureB
 140 process: the detection of low-penetration depth radiation such as UV photons and low-energy
 141 electrons. Here we shall present the most attractive characteristics of the boron-silicon junction, in
 142 its application as a photodetector (photodiode). For this purpose we shall use as a reference the
 143 characteristics of an ideal photodetector, as presented in Fig. 2 and Table 1 [19]. Fig. 2 shows the
 144 vertical cross section of a silicon p-n junction photodetector and its electric circuit equivalent. Table 1
 145 presents the parameter values of an ideal silicon photodiode, which provide the best performance
 146 with respect to: responsivity, resolution, speed and stability.

147 2.2.1. Responsivity

148 The structure of the boron-silicon junction satisfies in an excellent way the three parameters
 149 affecting the responsivity: passivation layer thickness, depletion depth and width (see Table 1).
 150 The junction can be essentially created by a single layer of boron atoms deposited on the n-type
 151 silicon substrate, in such a way that chemical bonds are formed between the boron atoms and
 152 the surface silicon atoms. For reliable protection of the underlying silicon from oxidation and
 153 potentially detrimental environment conditions, a few extra boron layers with a total thickness
 154 of a few nanometers are deposited in practice. The depletion region, where the photogenerated
 155 electron-hole pairs can be separated and collected, starts from the very first atomic layer of the
 156 underlying n-type silicon substrate. The lost radiation absorbed in the passivation layer is minimal
 157 due to its thinness. Fig. 3 shows the measured spectral responsivity of a boron-silicon junction with a
 158 ~ 5 nm amorphous boron protection layer in the extreme ultra violet (EUV) spectral range, compared
 159 with the theoretically attainable values for an ideal Si-based photodetector and a commercial n^+p

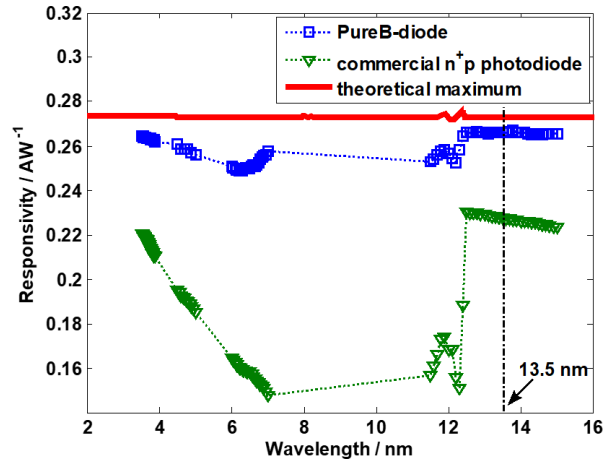


Figure 3. Measured spectral responsivity (squares) of boron-silicon photodiode in the EUV spectral range, compared with a commercial n+p photodiode and the theoretically attainable values for an ideal Si-based photodetector [20].

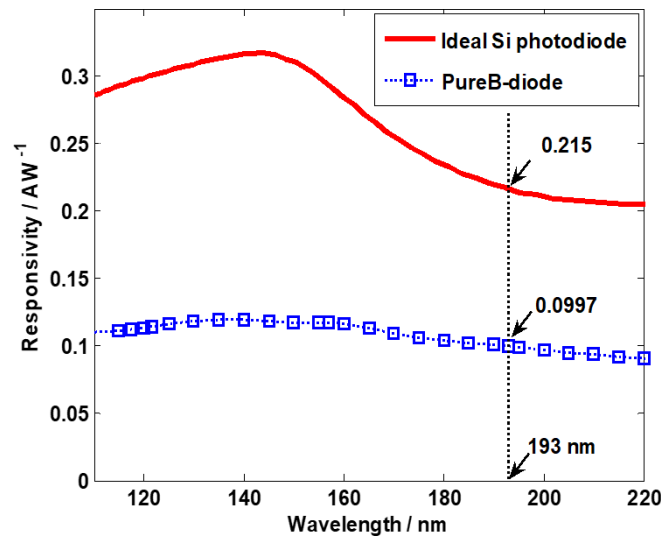


Figure 4. Measured responsivity of a boron-silicon photodiode with ~ 2 nm boron layer, in the DUV/VUV spectral ranges, compared with the theoretically attainable values for an ideal Si-based photodetector [19].

160 photodiode (SXUV from ODC) [19,20]. The measured responsivity above the silicon edge (12.4 nm) is
 161 0.265 A/W, which is very close to that of an ideal lossless system (0.27 A/W), indicating 100% internal
 162 quantum efficiency. The slight drop in responsivity at wavelengths shorter than the silicon edge can
 163 be assigned to a very thin silicon absorbing layer above the depletion zone. However, most probably
 164 another phenomenon is playing a dominant role here: after the silicon edge, the penetration depth
 165 of the photons decreases significantly leading to the absorption of more photons close to the detector
 166 surface. The kinetic energy of the freed electrons is high enough to allow them to overcome the
 167 internal electric field and to move in an arbitrary direction, with some being lost when they recombine
 168 with the holes, or some even escaping from the surface of the detector as secondary electrons.

169 Fig. 4 shows the measured spectral responsivity of a boron-silicon photodiode in the deep
 170 ultra violet (DUV) and vacuum ultra violet (VUV) spectral ranges [19]. As indicated in the figure,
 171 based on the measured responsivity at a 193 nm wavelength (0.0997 A/W) and the theoretical
 172 value (0.215 A/W), the quantum efficiency is: $QE = 0.0997/0.215 \approx 0.46$. Considering the
 173 nearly 100% quantum efficiency measured at a 13.5 nm wavelength (Fig. 4), the loss due to the
 174 photon-generated electron-hole pairs recombining in the diode depletion region can be regarded as

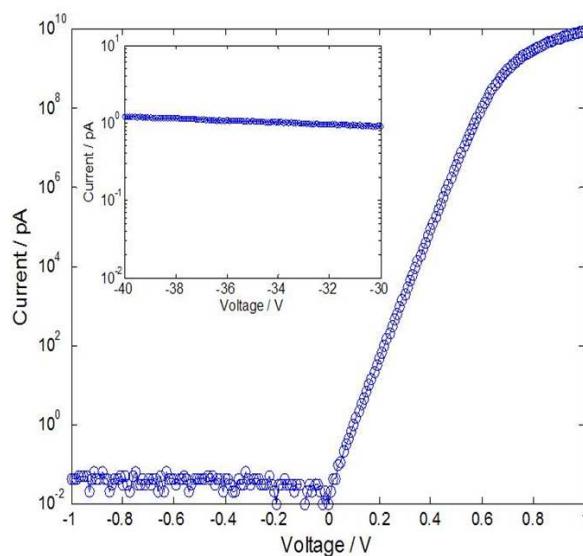


Figure 5. I-V characteristic of a boron-silicon diode with active area $300 \times 300 \mu\text{m}^2$. The inset shows a dark current of 1 pA with a reverse bias up to -40V, corresponding to current density: $\sim 11 \text{ nA}$ per square micrometer) [22].

175 negligible. The main reason for the quantum efficiency drop in the VUV/DUV spectral range, besides
176 the reflection-induced photon loss on the diode surface, is the extremely low penetration depth of the
177 photons in the boron and silicon: only a few nanometers [21]. Because of this, even a 2-nm-thick
178 boron layer will absorb a substantial part of the incident radiation.

179 2.2.2. Resolution

180 For achieving a very high resolution it is very important the interface between the silicon
181 substrate and the passivation layer on top of it to be defect free. This is achieved by the PureB
182 CVD process, during which the boron atoms form strong chemical bonds with the surface silicon
183 atoms. The number of silicon dangling bonds (acting as recombination centers) is extremely small.
184 This results in an excellent I-V characteristic typical for a high-quality deep p-n junction, with a very
185 low reverse-bias (dark) current and an ideality factor very close to 1 (Fig. 5) [22], despite the fact the
186 depletion region starts literally from the boron-silicon interface. A low dark current means a very
187 low shot noise associated with it. The low value of the dark current maintained at high reverse-bias
188 voltage is an evidence for a very high shunt resistance R_{sh} .

189 2.2.3. Stability

190 The strong chemical bonds formed between the silicon and the boron atoms provide an excellent
191 radiation shield, as they cannot be destroyed by UV photons or low-energy electrons. Experiments
192 with extensive exposure up to 220 kJ/cm^2 of a boron-silicon photodiode at 13.5 nm radiation
193 could not reveal a measurable degradation of the responsivity [23]. Only an insignificant amount
194 of responsivity degradation was observed in the VUV spectrum. Fig. 6 shows the responsivity
195 degradation of three boron-silicon junctions exposed to 121 nm radiation (radiation around 120 nm
196 is considered the most challenging in the VUV spectrum) [19]. The difference between the three samples
197 is the oxygen content on the surface, expressed as a thickness in nanometers. In this experiment high
198 exposure levels are not necessary, as any available drop in responsivity is evident almost immediately
199 at the start of the VUV exposure, subsequently settling to its lower level. The presence of oxygen is
200 assigned to local oxidation of silicon through pin holes in the thin boron layer which only partially
201 covers silicon. With VUV exposure the oxidized silicon surface is positively charged due to the
202 secondary electron emission, temporarily reducing the responsivity. As can be seen in Fig. 6, with

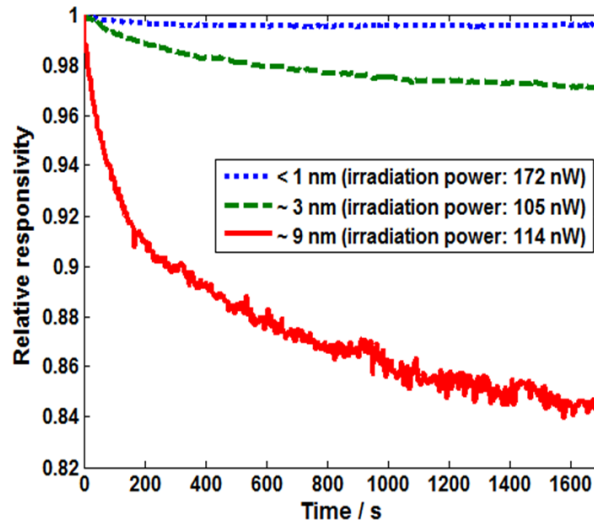


Figure 6. Responsivity degradation of 3 boron-silicon photodiodes with varying parasitic oxide content on the surface, at a 121-nm wavelength. The oxide content was measured by ellipsometry and is expressed as a thickness in nanometers [19].

203 1 nm oxide the degradation is extremely small – within the margin of uncertainty of the measurement
 204 equipment. It is important to mention that this kind of degradation is recoverable with time as the
 205 positive charge dissipates very slowly.

206 Another factor influencing the photodiode stability is the working environment. In this aspect,
 207 the boron-silicon junction demonstrates very high robustness to harsh working conditions. Boron
 208 itself is a very stable material at room temperature. Furthermore, a nanometer-thin amorphous boron
 209 layer, when completely covering the underlying silicon, acts as an excellent barrier protecting the
 210 silicon substrate from detrimental environmental elements such as hydrogen radicals and oxygen
 211 plasma, used for surface cleaning purposes. Extensive exposure to such elements has not resulted in
 212 noticeable deterioration of the electrical or optical characteristics of the boron-silicon photodiode [23].

213 2.2.4. Operational speed

214 For a fast reaction to pulsed radiation, the photogenerated charge must be removed quickly from
 215 the depletion region of the photodetector and delivered to the interface electronics. For this purpose
 216 the time constant of the detector—defined by the junction capacitance C_j and the series resistance
 217 R_s —must be small (Fig. 2b). The value of the series resistance is dominated by the sheet resistance
 218 of the surface of the detector. This is because after separation, the photogenerated charge must reach
 219 the top ring electrode (Fig. 2a) by moving along the surface of the detector. However, due to the high
 220 resistivity of the very thin boron layer and the fact that the depletion region starts from the silicon
 221 surface, the sheet resistance is very high. This makes the boron-silicon photodiode extremely slow.
 222 Furthermore, unlike a typical p-n junction detector—where the time constant can be decreased by
 223 reducing C_j with the application of a higher reverse-bias voltage, the same approach does not work
 224 well with the boron-silicon detector. Just the opposite, the time constant of the boron-silicon detector
 225 increases with a higher reverse-bias voltage despite the reduction in the junction capacitance. This is
 226 because apparently the series resistance increases faster than the decrease in the capacitance [24].

227 A solution to this problem is to trade some responsivity for the benefit of the time constant. For
 228 example, deploying a metal Al grid on top of the boron layer, which covers just 1% of the surface,
 229 leads to a dramatic reduction in the sheet resistance and the time constant, respectively [19].

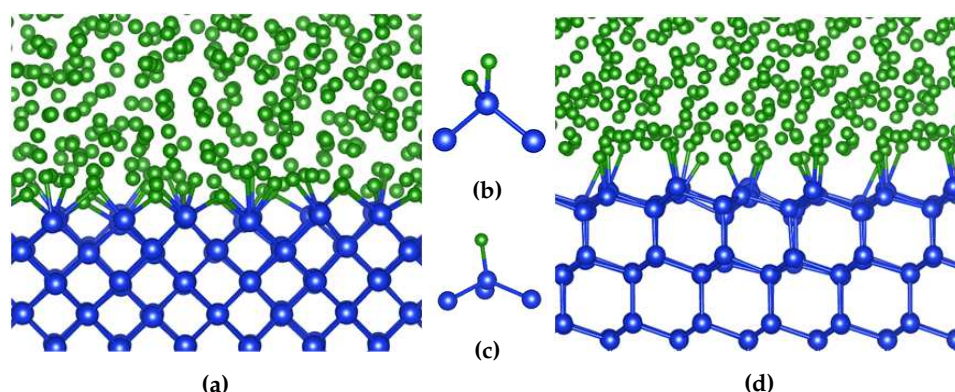


Figure 7. Snapshots of (a, d) the equilibrated interface and (b, c) related typical Si coordination for the Si{0 0 1}/a-B and the Si{1 1 1}/a-B interfaces, respectively. The green spheres represent B and blue Si.

3. Boron-silicon junction formation premise based on *ab initio* modeling

To gain insight into the preparation processes and the structural and electronic properties of the boron-silicon heterojunction, we modeled the decomposition of B_2H_6 molecules and the deposition of BH_n ($n = 1-4$) molecular/radicals on a Si substrate at the early stages of the PureB processes [10, 25,26], and a-B/c-Si interfaces [10,27]. Here we briefly review our recent theoretical work on the local structure, chemistry and electronic properties of a-B/c-Si interfaces. At present, Si{0 0 1} wafers are used to prepare boron-silicon heterojunctions. The unusual structure of the Si{1 1 1} surfaces and commercial availability of Si{1 1 1} wafers stimulate us to include a-B/Si{1 1 1} interfaces in our study, as well.

Supercells were built for *ab initio* molecular dynamics simulations. A tetragonal supercell with $a = 3a_0 = 16.45 \text{ \AA}$ (a_0 is the lattice parameter of cubic Si at the simulation temperature), $c = 19.18 \text{ \AA}$ for a-B/Si{0 0 1}. This cell contains 444 atoms (144 c-Si and 300 a-B). For a-B/Si{1 1 1}, we built a hexagonal cell with $a = 2\sqrt{2}a_0 = 15.15 \text{ \AA}$, $c = 26.65 \text{ \AA}$, which contains 492 atoms (192 Si and 300 B). These large supercells are required in order to prevent the interaction between the two surfaces and eliminate the risk of the a-B artificial crystallization.

All simulations were performed using the first-principles code VASP (Vienna *Ab initio* Simulation Package). This code is based on a pseudo-potential plane-wave approach within the density-functional theory (DFT) [28]. It employs the projector augmented-wave (PAW) method [29], and it allows variable fractional occupation number, which works well for interfaces between insulators and metals [28,30]. An *ab initio* molecular dynamics (AIMD) simulation employs the finite-temperature density-functional theory of the one-electron states, where the exact energy minimization and calculation of the exact Hellmann-Feynman forces occur after each MD step using both the preconditioned conjugate techniques, and the Nose dynamics to generate a canonical NVT ensemble [28]. The exchange and correlation terms are described using the generalized gradient approximation (GGA-PBE) [31]. For electronic structure calculations, we used cut-off energies of 400.0 eV for the wave functions, 550.0 eV for the augmentation functions, and dense grids in the irreducible Brillouin zone (BZ) of the cells [32]. For the AIMD simulations we used cut-off energies of 250.0 eV and the Γ -point in the BZs. This is due to the whole system lacking periodicity in such crystal/amorphous interfaces [28,33].

We created amorphous B by first equilibrating the samples at 3000 K for 2000 iterations (at 1.5 fs per iteration, which totals 3 ps) and then cooling systems to the desired temperature. Next the obtained a-B samples were placed on the crystalline Si substrates (c-Si), forming a-B/c-Si interfaces for the subsequent AIMD simulations. The prepared a-B/c-Si systems were then allowed to equilibrate at 1000 K with the Si atoms in the substrate pinned. Then, all the atoms including the substrate were allowed to relax at 1000 K over a period of 6 ps. Finally we relaxed the atoms at 0 K to eliminate internal forces and stress.

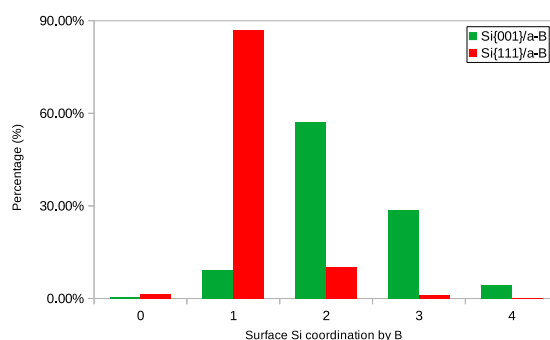


Figure 8. Distribution of the numbers of the coordination number of the interfacial Si atoms by boron at The Si{0 0 1}/a-B (circles) and Si{1 1 1}/a-B (squares) interfaces.

266 3.1. Local Chemistry of the c-Si/a-B Interfaces

267 Fig. 7 displays snapshots of the relaxed a-B/Si{0 0 1} and a-B/Si{1 1 1} interfaces from the AIMD
 268 simulations with the inputs, settings and treatments presented in the previous section. From Fig. 7
 269 we find the following features common to both interfaces:

- 270 1. The Si atoms in the substrates are positioned in an orderly fashion, whereas the B atoms remain
 271 disordered;
- 272 2. There is a spacing separating the crystalline Si and amorphous B at both the Si{0 0 1}/a-B (Fig. 7a)
 273 and the Si{1 1 1}/a-B (Fig. 7d) interfaces;
- 274 3. There is a certain amount of disordering for the surficial Si atoms at both substrates.

275 A closer look reveals subtle differences between the two interfaces. The spacing between the
 276 c-Si substrate and a-B at a-B/Si{1 1 1} is apparently larger than that at a-B/Si{0 0 1}. Moreover, the
 277 surficial Si atoms at a-B/Si{0 0 1} have more B neighbors than those at a-B/Si{1 1 1}. We analyzed the
 278 Si-B bonding at both interfaces with about 20 interfaces each. The cut-off of the Si-B bonds is 2.28 Å,
 279 which is 10 % longer than the average value of the B-B bond (1.79 Å) and the Si-Si bond-length (2.35
 280 Å) in the elemental solids, respectively, taking into account the exponential decay of bond strength as
 281 a function of interatomic distance [34]. The results are plotted in Fig. 8.

282 As shown in Fig. 8, most surficial Si (88%) at the a-B/Si{1 1 1} interfaces have only one B neighbor.
 283 Another 10% of the interfacial Si atoms are coordinated with two B atoms. The Si coordination of the
 284 surficial Si atoms at the a-B/Si{0 0 1} interfaces is more complex. The surficial Si atoms with two B
 285 neighbors are dominant at the a-B/Si{0 0 1} interface (57%). 29% of the interfacial Si atoms have three
 286 B neighbors, a relatively small amount of the interfacial Si (9%) have one B neighbor, and only 4% of
 287 the surface Si atoms have four B neighbors. The larger variety of Si coordination at the a-B/Si{0 0 1}
 288 interfaces is related to the reduced symmetry constraint from the Si substrates, as each superficial Si
 289 is bonded to only two Si atoms at the subsurface. The different local Si-B bonding indicates variation
 290 in the B arrangements at the interfaces. The dominant surficial Si atoms with one B and three Si atoms
 291 at a-B/Si{1 1 1}, and those Si with two B neighbors and two Si neighbors at a-B/Si{0 0 1}, satisfy the
 292 sp^3 type hybridization for Si [35].

293 The statistical analysis also produced the averaged spacing between surficial Si and the
 294 neighboring B atoms. A larger spacing (2.0 Å) was obtained at the a-B/Si{1 1 1} interface compared
 295 to that at the a-B/Si{0 0 1} interface (1.2 Å).

296 3.2. Electronic Properties of the c-Si/a-B Interfaces

297 Electronic structure calculations were performed for the relaxed interfaces. Fractions of the
 298 obtained electron density distributions are shown in Fig. 9. Based on the electron densities in the
 299 interface systems, we analyzed the charges at each atomic site using the Bader charge model [36].
 300 The charges obtained at the atomic sites at both interfaces are plotted in Fig. 10.

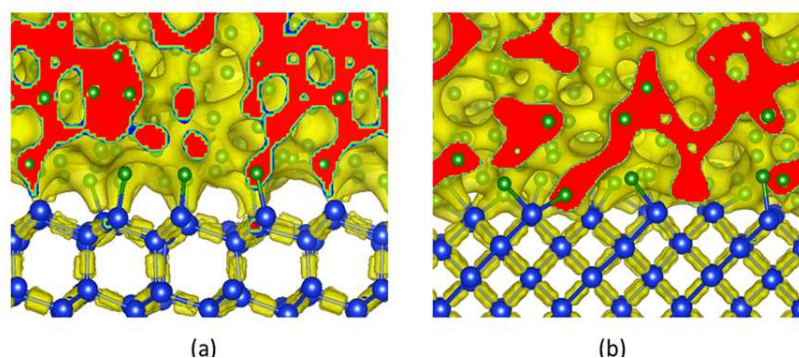


Figure 9. Electron density distribution (iso-surfaces $\rho_0(r) = 0.075 e/\text{\AA}^3$) and charges at the atomic sites at the (a) Si{0 0 1}/a-B and (b) Si{1 1 1}/a-B interfaces, respectively. The blue squares represent Si sites, the blue spheres the B sites and the filled black spheres for the averaged charge values. The broken red lines represents the centre of surficial Si and the broken green lines for the centres of interfacial a-B.

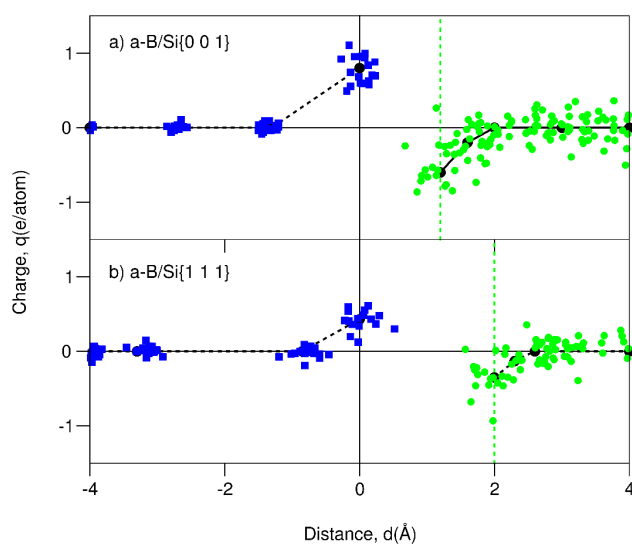


Figure 10. Charges at the a-B/c-Si interfaces. The blue squares represent charges at the Si sites and green spheres represent charges at the B sites. The solid black lines indicate the peak of the outmost Si, while the dotted green lines represent the first peak of B atoms.

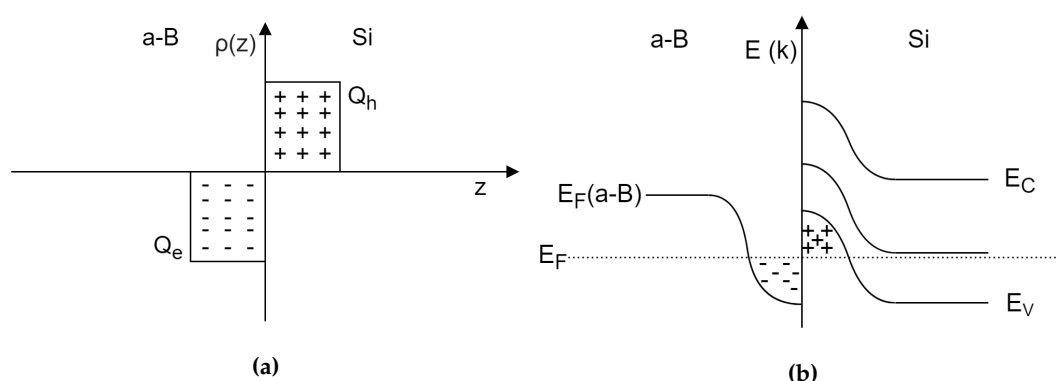


Figure 11. A schematic drawing of (a) charging and (b) band bending at the ideal Si/a-B. charge density is $4.7 \cdot 10^{18} \text{ e/m}^2$ at Si{0 0 1}/a-B and $2.7 \cdot 10^{18} \text{ e/m}^2$ at Si{1 1 1}/a-B based on the first-principles calculations.

301 As shown in Fig. 9, the electron clouds form regular shapes and are concentrated around the
 302 Si-Si bonds in the substrates. This corresponds to the crystalline structures and their covalent nature.
 303 Meanwhile, electron clouds in the a-B part show irregular forms of high electron density, which
 304 corresponds to the local disordering. At the interfacial region, there are clear clouds for both interfaces
 305 between the interfacing Si-B atoms, which indicate chemical bonding. T3 also shows that the electron
 306 clouds are more present at the B atoms, which is an indication of charge transfer from the Si atoms to
 307 the B atoms. At the a-B/Si{1 1 1} interface, each surficial Si is coordinated to one B with the electron
 308 clouds forming regular shapes, whereas at a-B/Si{0 0 1}, each Si has two or three B neighbors with
 309 dense electron clouds (see Fig. 9 for both).

310 Fig. 10 includes the charges at the atomic sites at both interfaces. Clearly, the Si and a-B atoms
 311 located away from the interfacial layers are electronically neutral. Charge transfer only occurs from
 312 interfacial Si atoms to interfacial B atoms. Analysis results revealed an average amount of charge
 313 transfer of 0.75 e/Si ($4.7 \times 10^{18} \text{ e/m}^2$) at a-B/Si{0 0 1}, and 0.40 e/Si ($2.7 \times 10^{18} \text{ e/m}^2$) at a-B/Si{1 1 1}.
 314 These values correspond to the number of Si-B bonds at the interfaces. These values are smaller than
 315 those from the ionic model (Si^{2+} at a-B/Si{0 0 1} and Si^+ at a-B/Si{1 1 1}). This is indicative of a bond
 316 of a strong covalent nature between the interfacing Si and B atoms (electronegativity value is 2.04 on
 317 Pauling scale for B and 1.90 for Si).

318 The charge transfer from the interfacial Si to B induces charge barriers at the a-B/c-Si interfaces.
 319 The formation of the charge barriers is essential for the heterojunctions/diodes. It also causes band
 320 bending of the heterojunctions.

321 3.3. Band Bending for the Electronegativity Junctions

322 Our AIMD simulations and electronic structure calculations for the a-B/c-Si interfaces revealed
 323 the formation of well-separated Si-B interfaces. Charge transfer occurs from the interfacial Si to B,
 324 forming $\text{Si}^{+q}/\text{B}^{-q}$ polar plates. Moreover, our study also showed that amorphous B located away
 325 from the interfaces is intrinsically a 'bad' metal with localized defect states. Therefore, at a-B/c-Si
 326 interfaces, the Fermi level of a-B will be changed near the interface due to extra electrons from Si
 327 filling of the defect states. Based on the results above and the semiconducting nature of bulk Si,
 328 we can build a band bending model for the boron-silicon heterojunctions. Together with the charge
 329 model, it is schematically shown in Fig. 11.

330 Fig. 11a schematically represents the charge model with the $Q_h = 0.70 \text{ e/Si}$ at the a-B/Si{0 0 1},
 331 and $Q_h = 0.45 \text{ e/Si}$ at a-B/Si{1 1 1} interfaces. The spacing between the positively charged Si and the
 332 negatively charged B has an averaged value of about 1.2 \AA at a-B/Si{0 0 1} and 2.0 \AA at a-B/Si{1 1 1},
 333 respectively. Based on these data, we obtained charge barriers of about 11 V for a-B/Si{0 0 1} and 10

334 V for a-B/Si{1 1 1}. These barrier values correspond to the balance between the Si-B interaction and
335 anisotropy of the Si surfaces.

336 The charge transfer and corresponding charge barriers cause band bending at the a-B/c-Si
337 interfaces, as schematically shown in Fig. 11b. The boron-silicon heterojunction properties essentially
338 originate from the charge transfer occurring at the interfaces due to the differences between the
339 electronegativities of Si and B.

340 The abundance of positive charge in the top atomic layer of the n-type silicon substrate acts
341 as a highly doped p-region in a p-n junction, attracting free electrons from the bulk n-type silicon,
342 and leading to the formation of a depletion region [10,27]. The significant amount of positive static
343 charge at the surface of the silicon n-type substrate explains the high Gummel Number and the low
344 saturation current of the boron-silicon junction, which is typical for high-quality deep p-n junctions.

345 4. Conclusions

346 Many research endeavors throughout the history of science, aside from achieving their primary
347 research goals, have managed to generate unexpected additional new knowledge. We believe that
348 this is also the case with the development of PureB technology. For many years the research efforts
349 have been mainly focused on the development of the process itself. Only recently has the rectifying
350 junction created through the PureB process—demonstrating properties not typical for a shallow
351 p-n junction and any other existing rectifying junction—attracted more attention. In this paper we
352 presented the main features of the PureB process and the qualities of the boron-silicon junction as a
353 radiation detector. We also discussed how the understanding of the boron-silicon junction formation
354 has evolved. The recently proposed quantum-mechanical junction formation mechanism presented
355 proves the power of quantum-mechanics-based approaches to solid-state physics and materials
356 science, offering a world of rich variety.

357 Overall, our understanding is that the chemical interaction between the surface atoms of
358 crystalline silicon and the first atomic layer of the amorphous boron is the dominant factor leading to
359 the rectifying function of boron-silicon junctions. Obviously, boron doping is not present in this model
360 and thus the boron-silicon junction does not belong to p+n type junctions. Although a-B exhibits a
361 high density of states in the band gap of Si, these states are of a localized nature. The electrons in a-B
362 are conducted via a hopping mechanism, thereby disqualifying the a-B/c-Si diodes from the category
363 of Schottky junctions. Furthermore, the a-B/c-Si diodes cannot be classified into any of the existing
364 types of heterojunctions in semiconductor physics.

365 The new junction has the application potential of reaching far beyond its primary target. The
366 new junction formation mechanism may lead to a technology breakthrough in a number of scientific
367 fields such as: semiconductor wide-bandgap material processing, electron microscopy, optics, space
368 exploration, chemical engineering, quantum mechanics, nano-materials, etc. Furthermore, this type
369 of junction may trigger many innovations in industry, e.g. direct ultraviolet detection and imaging,
370 energy harvesting (solar cells), etc.

371 New possible research directions may include: (i) better understanding of the quantum
372 phenomena responsible for the formation of the boron-silicon junction; (ii) creation of an analytical
373 model for the boron-silicon junction and extending this model to other semiconductor materials, such
374 as SiC and wide-bandgap materials; (iii) study of the electrical, optical and mechanical properties of
375 devices developed for different applications, based on the boron-silicon junction technology.

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