



Title	Estimation of submarine groundwater discharge in Plover Cove, Tolo Harbour, Hong Kong by 222Rn
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1 **Estimation of submarine groundwater discharge in Plover**

2 **Cove, Tolo Harbour, Hong Kong by ^{222}Rn**

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10
11 **Abstract**

12 Algal blooms in Tolo Harbour, Hong Kong have received much attention and
13 submarine groundwater discharge is speculated to be a significant pathway carrying
14 nutrients into the constricted estuary. Plover Cove, a small cove in the Harbour, was
15 selected for SGD analysis using ^{222}Rn budget. The volumetric SGD rates are estimated to
16 be about 8,000 m³/day for neap tide and about 17,000 m³/day for spring tide. Result of
17 nutrient analysis of the porewater indicates that the nutrient loading through this pathway
18 is speculated to be crucial for eutrophication in Tolo Harbour. Current practice for the
19 management of algal blooms in Hong Kong, in which nutrient loading through SGD was
20 ignored, has to be reviewed and the control measures of groundwater contamination are
21 obviously required.

22
23 Index terms: Geochemical tracers; Groundwater hydrology; Groundwater transport;
24 Hydrological cycles and budgets; Pollution: urban, regional and global.

25

26 **Introduction**

27 Tolo Harbour is located in the northeastern part of Hong Kong's New Territories
28 (Figure 1). It is susceptible to pollution because of the bottlenecked coastline
29 configuration as well as the prevailing northeasterly wind direction (Yin, 2003). Current
30 is low for the harbour and the estimated water residence times in the inner harbour range
31 from 16-42 days (Hodgkiss and Yim, 1995). All these factors result in preventing the
32 pollutants to be removed effectively. While the harbour is already under stress due to
33 natural factors, urban development since 1970s has further deteriorated the water quality.
34 Dramatic expansion of human population, from 70 000 in 1973 to 1 000 000 in 1990, has
35 degraded the environment.

36 As a consequence, algal bloom incidents increased from 1 per year in 1978 to over 40
37 in 1988 (Holmes, 1988; Environmental Protection Department (EPD), 2004). Holmes
38 (1988) has attributed the environmental degradation to reduction of the mangrove
39 coastline through the process of land reclamation. The mangrove coastline serves as an
40 effective sink for nutrients, so the loss of this natural resource shifts the primary
41 biological productivity to planktonic algae.

42 In order to alleviate the pollutant loading on Tolo Harbour, the government has
43 enforced the Tolo Harbour Action Plan in 1988 which included livestock waste control,
44 sewage treatment modification, effluent export scheme, legislation enforcement and
45 landfill restoration. An ecosystem model, which was developed for the Hong Kong
46 government, claimed that the concentrations of ammonia and nitrate could drop to zero
47 after the Action Plan was implemented (Holmes, 1988). After the action plan has been
48 implemented, the number of algal bloom incidents has been decreased to ~10 incidents
49 per year. The total phytoplankton densities, however, were still 2-6 times higher at

50 stations in Tolo Harbour and Channel than those in other water control zones over Hong
51 Kong (EPD, 2004). Xu et al. (2004) showed that the loadings of total phosphorus and
52 total nitrogen in Tolo Harbour even reached another peaks in 1996 and 1998 respectively.
53 It is speculated that the phytoplankton may be sustained by nutrients from other sources.

54 Hodgkiss and Ho (1997) suggested that implementation of the Tolo Harbour Action
55 Plan resulted in a lower N:P ratio which led to a shift in algal species composition from
56 diatoms to dinoflagellates and the dominance of dinoflagellates was the major cause of
57 the dramatic increase in red tide occurrence. This replacement of algal species was neither
58 found by Yung et al. (1997) nor supported by the long-term monitoring by EPD. Hu et al.
59 (2001) conducted a sediment diffusion experiment by placing undisturbed sediment core
60 into a settling column in laboratory and adding water sample collected from
61 corresponding site. They demonstrated that sediments can release 0.5 mmol of
62 orthophosphate phosphorus and 2.2 mmol of nitrate-nitrite nitrogen per square metre per
63 day.

64 Recently research in other coastal areas indicates that the direct discharge of
65 groundwater into the coastal zone may be a potentially significant pathway of dissolved
66 nutrients into the coastal environment (Laroche et al., 1997; Griggs et al., 2003; Miller
67 and Ullman, 2004). Terrestrial groundwater can discharge into the sea directly in response
68 to the hydraulic gradient, with groundwater head higher than sea level (Johannes, 1980).
69 Li and Jiao (2002) proposed a tide-induced seawater-groundwater circulation that SGD
70 happens even in the absence of net inland recharge of groundwater.

71 It is speculated that eutrophication in Tolo Harbour may be attributed to the nutrient
72 fluxes through the pathway of submarine groundwater discharge (SGD). Tolo Harbour is
73 enclosed by such a large catchment that the amount of SGD is believed to be comparable
74 to the river water flux. This study focused on Plover Cove that has no major river system.

75 Plover Cove is adjacent to a mountain range including Wong Leng (639 m) and Pat Sin
76 Leng (500-600 m), with a surface area of around 4 km² (Figure 1). Radium study was
77 conducted for Tolo Harbour (Tse, 2006) and based on this radium study, Plover Cove is
78 identified as a key area in Tolo Harbour receiving potentially significant amount of SGD
79 and details of the radium study can be found in Tse (2006). The current paper focuses on
80 radon study and the SGD flux was estimated by ²²²Rn following the approach described
81 by Burnett and Dulaiova (2003).

82 **Background of the Site**

83 The overall surface area for Tolo Harbour is 52 km² including the part of Tolo
84 Channel. In the inner harbour the water depth is less than 10m, while along the channel
85 the average depth is about 12 m. The mean sea level is 1.15 m above Principal Datum
86 (mPD), with average diurnal tidal range of 0.97 m (EPD, 1994). There are several rivers
87 entering Tolo Harbour and according to the monitoring data between 1998 and 2004 from
88 EPD, the total annual discharge rate is 3.61 x 10¹⁰ m³/yr. The mean annual rainfall is
89 2214 mm.

90 Tolo Harbour is enclosed by a large catchment with an area of 50 km². The catchment
91 is formed by three main mountainous blocks divided by the new towns Tai Po and Sha
92 Tin. The geology was described in detail by Tam (1980), Addison (1986) and Lai et al.
93 (1996). The district is formed mainly of Mesozoic stratified pyroclastic rocks sandwiched
94 between the Mesozoic sedimentary rocks, with sandstone, siltstone and conglomerate
95 above and marine mudstone and siltstone below.

96 The volcanic rocks are intruded by a complex granitic pluton at the age of Late
97 Jurassic to Early Cretaceous, which crops out at the low-lying areas. A series of faults
98 trending northeast and northwest was generated and one of the biggest faults is the Lai

99 Chi Kok – Tolo Channel Fault which belongs to a fault zone extending across Sha Tin
100 Valley to Tolo Channel.

101 The superficial soils, including the mantle of weathered rock, colluvium, alluvium and
102 beach sand are considered as shallow unconfined aquifer, which can be over 20 m thick
103 (Ruxton and Berry, 1957). The sandy deposit is subsequently replaced by silt or marine
104 mud in the estuary which serves as an extensive layer of aquitard. There are successive
105 layers of marine and alluvial deposits underlain the Holocene marine mud, which
106 indicates that channelized sand bodies can be found offshore as confined aquifers. These
107 channels are potentially fresh if they are hydraulically connected with the recharge area
108 onshore.

109 The bedrock, according to the degree of decomposition, varies from fresh rock to
110 residual soil. Fracture zones exist along the rockhead below the decomposed rock which
111 form a relatively deep confined aquifer (Jiao et al., 2005; Jiao et al., 2006).

112 Grant (1989) investigated the permeability in Lam Tsuen and Tolo Harbour Areas and
113 in his study, the permeabilities of alluvial plain and terrace lands are 1.17×10^{-4} m/s and
114 2.17×10^{-4} m/s respectively.

115

116 **Methodology**

117 The SGD flux was estimated by ^{222}Rn following the approach described by Burnett
118 and Dulaiova (2003). In general, ^{222}Rn concentration in the system is influenced by
119 various sources and sinks, such as ingrowth from ^{226}Ra dissolved in the water, tidal effect,
120 atmospheric loss, diffusion from sediments, mixing loss to the open sea and SGD (Figure
121 2). An increase or decrease of ^{222}Rn concentration over a time interval is referred to the
122 net balance between these sources and sinks during that period. ^{222}Rn concentrations in
123 coastal waters are measured continuously to determine the difference between two

124 successive measurements. This is then corrected for all other sources and sinks to obtain
125 the ^{222}Rn flux attributed to SGD. With this flux divided by ^{222}Rn concentration in fresh
126 groundwater or porewater, the SGD flux is computed.

$$127 \quad F_{SGD} = F_t - F_{sed} - F_o + F_{atm} + F_i + F_m \quad (1)$$

128 where F_{SGD} is the ^{222}Rn flux attributed to SGD

129 F_t is the difference in concentrations of excess ^{222}Rn between two successive
130 hours

131 F_{sed} is the flux diffused from sediments

132 F_o is the flux leaving with the outgoing tide

133 F_{atm} is the flux into the atmosphere

134 F_i is the flux entering with the incoming tide

135 F_m is the flux out of the system by mixing

136 A site in Plover Cove was selected for ^{222}Rn analysis in July 2005. ^{222}Rn in coastal
137 waters was monitored continuously by a commercially available radon-in-air monitoring
138 system called RAD7 produced by DurrIDGE Co., Inc. for four days, with 48 hours during
139 the neap tide (14-16 July) and 48 hours during the spring tide (21-23 July). In the
140 meantime, water depth was estimated from the iron framework where a height indicator
141 was made. Water samples were collected bihourly for ^{226}Ra analysis. Sediments were
142 collected from the sea bottom to estimate ^{222}Rn in porewater. Wind speed, salinity, air and
143 water temperatures were measured manually every 10 minutes. At the same time, a
144 continuous heat-type automated seepage meter (Taniguchi and Iwakawa, 2001) was also
145 deployed for direct SGD measurement. It was pushed into the sea bottom and
146 programmed to take readings every 6 s. During the sampling period, monitoring of ^{222}Rn
147 in coastal waters was suspended twice: from 00:23, July 16 onwards to measure ^{222}Rn in
148 ambient air, and from 05:00, July 22 onwards because of a storm. ^{222}Rn in groundwater

149 from two private wells (Po Sum Pai and Chim Uk) was collected and measured in March
150 2006. Po Sum Pai private well is situated on Quaternary deposits of volcanic rocks with a
151 depth of 2.7 m, while Chim Uk private well is situated on granitic rocks with a depth of
152 3.3 m.

153 **Continuous Monitoring of ^{222}Rn**

154 In order to monitor ^{222}Rn in coastal waters continuously, an iron framework was
155 placed on the sea bottom on which the submersible pump was fastened so that it was
156 fixed at 0.5 m above the sea bottom. Seawater was pumped out and filtered through a 1
157 μm cartridge filter to screen out the particulates. It was then sparged into an air-water
158 exchanger where radon was distributed from the running flow of water to a closed air
159 loop until the two phases reached equilibrium. The air stream was then fed to the RAD7
160 for measurement, which was converted to ^{222}Rn in the water by the following equation
161 (DurrIDGE Co., Inc., 2001),

$$162 \qquad \qquad \qquad \alpha = 0.105 + 0.405e^{-0.050T} \qquad \qquad \qquad (2)$$

163

164 where α is the partition coefficient (concentration ratio of water to air)

166 T is the water temperature in $^{\circ}\text{C}$, which was measured by a temperature probe
167 inserted into the air-water exchanger

168 The RAD7 was programmed to integrate counts every hour.

169 **^{226}Ra analysis**

170 The water from the air-water exchanger was fed into a 50 L water tank for radium
171 extraction. After 50 L of the water had been collected, it was forced to flow through 30 g
172 of Mn-fiber described by Moore (1976) to extract the radionuclide. The flow rate was
173 controlled below 1 L/min so that sufficient time was allowed for the adsorption. In the
174 laboratory, ^{226}Ra was extracted from the Mn-fiber by refluxing with HCl, and the filtrate

175 was co-precipitated with 10 mL of saturated $\text{Ba}(\text{NO}_3)_2$ and 25 mL of 7 M H_2SO_4 . The
176 precipitate [$\text{Ba}(\text{Ra})\text{SO}_4$] was then filtered out by 0.45 μm glass-fiber filter and washed
177 with 3 M HCl and water to remove the Mn remains. Finally the precipitant was air-dried,
178 stored in a small vial for 3-4 weeks for equilibrium, and measured by gamma ray
179 spectrometer (Rutgers van der Loeff and Moore,1999).

180 **Determination of ^{222}Rn in Groundwater**

181 Groundwater samples were pumped out from two private wells (Figure 1) and
182 collected in 250 mL collection vials. As radon in groundwater can be quickly distributed
183 into air, narrow tubing was attached to the pump and inserted into the bottom of the
184 collection vial which was placed inside a 1 L plastic beaker. The vial was filled from
185 bottom with fresh sample until the water overflowed into the beaker and the water in the
186 beaker rose well above the vial. In this way, the vial was flushed with fresh sample
187 without exposure to air and it was capped while still under the water.

188 The samples were then measured by RAD7 with the RAD- H_2O accessory. RAD- H_2O
189 aerates the sample for 5 minutes to deliver ^{222}Rn to the RAD7. The system will wait a
190 further 5 minutes for the equilibrium between ^{222}Rn and ^{218}Po . The air stream is then
191 measured by RAD7 in 4 runs of five-minute period and the result is given as the average
192 of the 4 runs.

193 **Determination of ^{222}Rn in Porewater**

194 Concentration of ^{222}Rn in porewater was determined by sediment equilibration
195 experiment which was described by Corbett et al. (1998). Sediments from the sea bottom
196 and seawater were collected for the experiment. In the laboratory, 85 g sediments were
197 mixed with 300 mL seawater in a 500-mL Erlenmeyer flask. The flask was sealed and
198 agitated for 1 month until ^{222}Rn in head space, water and sediments reach equilibrium.
199 The water was then pumped out and transferred into a 40 mL collection vial, with the

200 special sampling technique employed to prevent air contact. The sample was then
201 measured by RAD7 with the RAD-H₂O accessory.

202 **Determination of Nutrients in Porewater**

203 Wet sediment at a depth of 0.5 m was collected along the foreshore during lowest tide.
204 It was immediately brought to the laboratory where the porewater was separated from the
205 sediment by centrifuge. Nutrients in porewater (NO₂-N, NO₃-N, NH₃-N, PO₄-P and silica)
206 were then analyzed by a spectrophotometer.

207

208 **Result and Discussion**

209 The results of the continuous ²²²Rn measurements in the water column, together with
210 the observed water depth, during the neap and spring tides are shown in Figures 3 and 4,
211 respectively. The concentrations of ²²²Rn were high during low tide, and low during high
212 tide, which fluctuated between 110 and 688 Bq/m³ with an average of 222 Bq/m³. Tidal
213 period cyclicity of the ²²²Rn data was generally accepted to reflect dilution of offshore
214 waters at flood tide, mixing offshore and most importantly, SGD variation (Kim and
215 Hwang, 2002; Burnett and Dulaiova, 2003; Lambert and Burnett, 2003). At low tide,
216 recirculated seawater drains out due to tidal pumping. Simultaneously the hydrostatic
217 pressure is lowered and the hydraulic gradient between seawater and groundwater is
218 increased, which contributes to a larger SGD flux. At high tide, recirculated seawater
219 seeps into the seabed sediments due to tidal pumping. Simultaneously the hydrostatic
220 pressure is increased and the hydraulic gradient between seawater and groundwater is
221 decreased, which contributes to a smaller SGD flux.

222 **Tidal Effects**

223 In order to account for the dilution effect during flood tide, Lambert and Burnett
224 (2003) introduced the concept of excess ²²²Rn inventory to eliminate the effect. Excess

225 ^{222}Rn inventory is defined as the product of excess ^{222}Rn in water (Concentration of
226 ^{222}Rn – Concentration of ^{226}Ra) and the water depth (h). During flood tide, a larger h is
227 multiplied so that the dilution effect can be compensated. Apart from that, ^{226}Ra which is
228 about 6 Bq/m^3 was also subtracted from ^{222}Rn in excess ^{222}Rn inventory so that the radon
229 supported by ^{226}Ra was also corrected.

230 Excess ^{222}Rn inventory can actually be interpreted as the excess ^{222}Rn in a water
231 column within an area of 1 m^2 . From this definition, it is deduced that inventory is still
232 subject to changes in tidal height: ^{222}Rn is removed from the water column with the
233 outgoing waters on the ebb tide (F_o) while extra ^{222}Rn is added to the water column with
234 the incoming waters on the flood tide (F_i). Correction is required to remove this tidal
235 effect. The excess ^{222}Rn inventory is corrected by an addition of the removed ^{222}Rn
236 inventory ($\Delta h \times$ concentration of ^{222}Rn in the study domain) at low tide or a subtraction of
237 the extra ^{222}Rn inventory ($\Delta h \times$ concentration of ^{222}Rn in offshore waters) at high tide,
238 where Δh is the difference between the two successive tidal heights.

239 **Atmospheric Loss**

240 ^{222}Rn is slightly soluble gas in water, exchange across the air-water interface is
241 possible if ^{222}Rn in the two phases are in disequilibrium. At equilibrium,

$$242 \quad \quad \quad 243 \quad \quad \quad C_w = \alpha C_a \quad \quad \quad (3)$$

244 where C_w is the concentration of ^{222}Rn in water

245 C_a is the concentration of ^{222}Rn in air

246 α is the partition coefficient

247
248 When $C_w > \alpha C_a$, ^{222}Rn will diffuse across the air-water interface and according to
249 MacIntyre et al. (1995), the diffusive flux is,

250

251
$$F_{atm} = k(C_w - \alpha C_a) \quad (4)$$

252 where F_{atm} is the diffusive flux across the air-water interface
 253

254 k is the gas transfer velocity

255 Considerable effort has gone into determining empirical relationship between the gas
 256 transfer velocity and wind speed, which was based on five experiments on lakes with
 257 deliberate tracers SF₆ (MacIntyre et al., 1995; Lambert and Burnett, 2003),

258
$$k_{600} = \begin{cases} 0.45\mu^{1.6} \left(\frac{Sc}{600}\right)^{-0.5} & \text{for } \mu > 3.6 \text{ m/s} \\ 0.45\mu^{1.6} \left(\frac{Sc}{600}\right)^{-0.6667} & \text{for } \mu \leq 3.6 \text{ m/s} \end{cases} \quad (5)$$

260 where μ is the measured wind speed (m/s)
 261

262 Sc is the Schimidt number

263 k_{600} is the gas transfer velocity normalized to the Schimidt number of CO₂ at 20
 264 °C in freshwater (cm/hr)

265 At wind speeds of less than 1.5 m/s, the value for k is assumed to be 0.91 cm/hr
 266 (Lambert and Burnett, 2003). This value is calculated based on the k value for CH₄ (0.75
 267 ± 0.54 cm/hr) measured by Happell et al. (1995) at zero wind speed.

268 The Schmidt number for ²²²Rn in seawater is given by Pilson (1998) as a function of
 269 the water temperature.

270 Five measurements were made from 23:40, July 15 to the end of neap tide period for
 271 ²²²Rn in air and the average (30 Bq/m³) is used for calculation (Table 1). Figures 5 and 6
 272 show the temporal variations of wind speed and the diffusive flux of ²²²Rn across the air-
 273 water interface during the two sampling periods.

274 **Diffusive Flux from Seabed Sediments**

275 Similar to the air-water interface, exchange across the sediment-water interface is
276 possible if concentration of ^{222}Rn in porewater is greater than that in the overlying water.

277 The diffusive flux across the sediment-water interface is given by Martens et al. (1980),

$$278 \quad \quad \quad 279 \quad \quad \quad F_{sed} = (\lambda D_s)^{0.5} (C_{eq} - C_o) \quad (6)$$

280 where F_{sed} is the diffusive flux across the sediment-water interface

282 λ is the decay constant of ^{222}Rn , which is 0.181 day^{-1}

283 D_s is the effective wet bulk sediment diffusion coefficient

284 C_{eq} is the concentration of ^{222}Rn in porewater

285 C_o is the concentration of ^{222}Rn in the overlying water

286 Ullman and Aller (1981) pointed out that the effective wet bulk sediment diffusion
287 coefficient is approximately equal to the product of porosity and the molecular diffusivity
288 coefficient of ^{222}Rn . Molecular diffusivity coefficient was described by Peng et al. (1974)
289 as a function of temperature,

$$290 \quad \quad \quad 291 \quad \quad \quad D_o = 10^{-\left[\left(\frac{980}{T+273}\right)^{+1.59}\right]} \quad (7)$$

292 where D_o is the molecular diffusivity coefficient

294 T is the temperature in $^{\circ}\text{C}$

295 Particle size analysis classified the seabed sediments as loose uniform sand. The
296 porosity estimated from soil analysis is around 0.41.

297 Concentration of ^{222}Rn in porewater was determined by sediment equilibration
298 experiment. From the experiment, the activity of ^{222}Rn released from the wet sediments is
299 0.6 Bq/kg. Concentration of ^{222}Rn in porewater is calculated by,

300

301

$$C_{eq} = \frac{{}^{222}\text{Rn released from the wet sediment} \times \rho_{wet}}{n} \quad (8)$$

302 where n is the porosity (0.41)

303 ρ_{wet} is the wet bulk density, which is measured as 2086 kg/m³, which is based on

304 excavation method and sand displacement test.

305 From the sediment equilibration experiment, concentration of ²²²Rn in porewater is

306 estimated to be 3052 Bq/m³. Concentration of ²²²Rn in overlying water was monitored

307 continuously during the two sampling periods. The diffusive flux of ²²²Rn across the

308 sediment-water interface is calculated to be 0.4 Bq/m²·hr, which is insignificant in

309 comparison to the total flux.

310 **Mixing Loss**

311 Figures 7 and 8 show the net ²²²Rn flux after correcting for atmospheric loss and

312 sediment diffusion, which should be a balance between supply from SGD and mixing loss

313 to the open sea. Burnett and Dulaiova (2003) and Lambert and Burnett (2003) chose the

314 negative net ²²²Rn fluxes as conservative estimates of the mixing loss. Indeed a larger

315 mixing loss is possible to be balanced by a larger supply from SGD, but the conservative

316 estimates of the mixing loss provide a good guess for the minimum SGD flux. It is

317 represented by the dashes line shown in Figures 7 and 8.

318 The estimated mixing losses are between 29.4 to 89.0 Bq/m²·h r during neap tide and

319 between 64.0 to 125.0 Bq/m²·h r during spring tide.

320 **SGD Flux**

321 As mentioned, the net ²²²Rn flux is a balance between supply from SGD and mixing

322 loss to the open sea. After correcting for mixing loss, the ²²²Rn flux is solely attributed to

323 SGD. The average ²²²Rn flux solely attributed to SGD is 52.8 Bq/m²·h r, which is about

324 130 times larger than the flux diffused from sediment. Although Hu et al. (2001) and Xu

325 et al. (2004) suggested that significant amount of nutrients was diffused from seabed
326 sediments and supplied to the phytoplankton in Tolo Harbour, nutrients discharged with
327 SGD are conceived to be much more significant than diffusion from sediment in Plover
328 Cove.

329 In order to convert ^{222}Rn fluxes to SGD fluxes, the ^{222}Rn fluxes solely attributed to
330 SGD have to be divided by the concentration of ^{222}Rn in the SGD fluid. Determination of
331 the SGD fluid requires an understanding of the discharge characteristic: If slow seepage
332 through sediment is dominant, ^{222}Rn in porewater will be a good guess of that in SGD;
333 otherwise ^{222}Rn in groundwater will be more representative if fast groundwater flow is
334 dominant. ^{222}Rn in porewater was determined by sediment equilibration experiment to be
335 3052 Bq/m^3 . Groundwater was collected from two private wells near Plover Cove, the
336 concentrations of ^{222}Rn are 6858 Bq/m^3 ($n = 28$; $\sigma = 3728$) for Po Sum Pai private well
337 and 16790 Bq/m^3 ($n = 5$; $\sigma = 4188$) for Chim Uk private well.

338 The SGD fluxes are plotted on Figures 9 and 10 and summarized in Table 2. The
339 inverse relationship between tidal height and SGD is verified. Precisely the peaks of SGD
340 coincide with the transitions from flood tide to ebb tide. With porewater being selected to
341 represent the SGD fluid, the mean SGD fluxes are 30.3 and 63.0 cm/day for neap and
342 spring tides respectively. The mean SGD fluxes are 7.8 and 16.3 cm/day for neap and
343 spring tides with groundwater from the two wells averaged to represent the SGD fluid.

344 The SGD fluxes based on terrestrial groundwater are chosen for calculating the
345 nutrient loading to Plover Cove through SGD so as to obtain the most conservative
346 estimates of SGD and the corresponding nutrient loading to Plover Cove. These values
347 are also comparable to the mean SGD fluxes measured directly by seepage meter, which
348 are 8.0 and 9.3 cm/day for neap and spring tides respectively.

349 The width of the seepage face is difficult to estimate. During the field studies, neap
350 tide, the exposed sandy beach face during neap tide is about 30 m from the high water
351 mark. In the following calculation, it is assumed that the width of the seepage face is 30 m.
352 With the length of the shoreline approximately about 3.5 km, the amount of SGD should
353 be around 8,000 m³/day for neap tide and 17,000 m³/day for spring tide. Result in nutrient
354 analysis and nutrient loading to Plover Cove through SGD are given in Table 3.

355 **Conclusion**

356 Within Tolo Harbour, Plover Cove has been selected for SGD study. SGD flux has
357 been estimated via continuous ²²²Rn measurements and seepage meter. ²²²Rn flux
358 attributed to SGD has been obtained after correcting for tidal effect, atmospheric loss,
359 sediment diffusion and mixing loss to the open sea. With terrestrial groundwater being
360 selected, the amount of SGD is 8,000 m³/day during neap tide and 17,000 m³/day during
361 spring tide. It has to point out that the estimated SGD include both components of
362 terrestrial groundwater discharge as well as recirculated seawater.

363 Management of algal blooms has emphasized on external nutrient loading from rivers.
364 Even most of the nutrient loading from rivers has been removed through the Tolo
365 Harbour Action Plan, the total phytoplankton densities are still the highest among Hong
366 Kong waters. In this study, nutrients discharged with SGD are conceived to be much
367 more significant than all the other pathways in Plover Cove. Taking the width of the
368 seepage area to be 30 m, nutrient loading on Plover Cove through SGD during neap tide
369 and spring tide are: Total inorganic nitrogen 1,241 and 2,599 mol/day; Orthophosphate
370 phosphorus 27.8 and 58.2 mol/day; Silica 185.9 and 388.5 mol/day. In terms of nutrient
371 loadings per unit area, the amount of nitrate-nitrite nitrogen loading through SGD is about
372 twice of that released from sediment diffusion obtained by Hu et al. (2001).
373 Transformation along the groundwater flow path in the sediments may attenuate nutrient

374 composition in the SGD fluid but still, based on the result from this study, the current
375 practice for the management of algal blooms in Tolo Harbour has to be reviewed and the
376 control measures of groundwater contamination are obviously required.

377

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Table 1 Measurements of ^{222}Rn in ambient air on 16th July 2005

Time	Conc. of ^{222}Rn (Bq/m ³)
01:23	47.73
02:23	14.80
03:23	29.23
04:23	18.50
05:23	37.37
Average	29.53

Table 2 Estimated SGD fluxes

During neap tide period

Approach	SGD fluxes	Mean (cm/day)
<i>^{222}Rn (SGD fluid represented by)</i>		
• Porewater	0-67.5	30.3
• Groundwater from nearshore wells	0-17.4	7.8
<i>Seepage meter</i>	0.9-40.1	8.0

During spring tide period

Approach	SGD fluxes	Mean (cm/day)
<i>^{222}Rn (SGD fluid represented by)</i>		
• Porewater	0-236	63.0
• Groundwater from nearshore wells	0-60.8	16.3
<i>Seepage meter</i>	1.0-144	9.3

Table 3 Nutrients in porewater and nutrient loading to Plover Cove through SGD

Nutrients	Concentration ($\mu\text{mol/L}$)	Nutrient Loading to Plover Cove through SGD (mol/day)	
		Neap Tide	Spring Tide
Nitrite Nitrogen	0.8	6.6	13.7
Nitrate Nitrogen	78.6	643.7	1345.2
Ammonia Nitrogen	72.1	590.5	1234.0
Orthophosphate Phosphorus	3.4	27.8	58.2
Silica	22.7	185.9	388.5

Figures

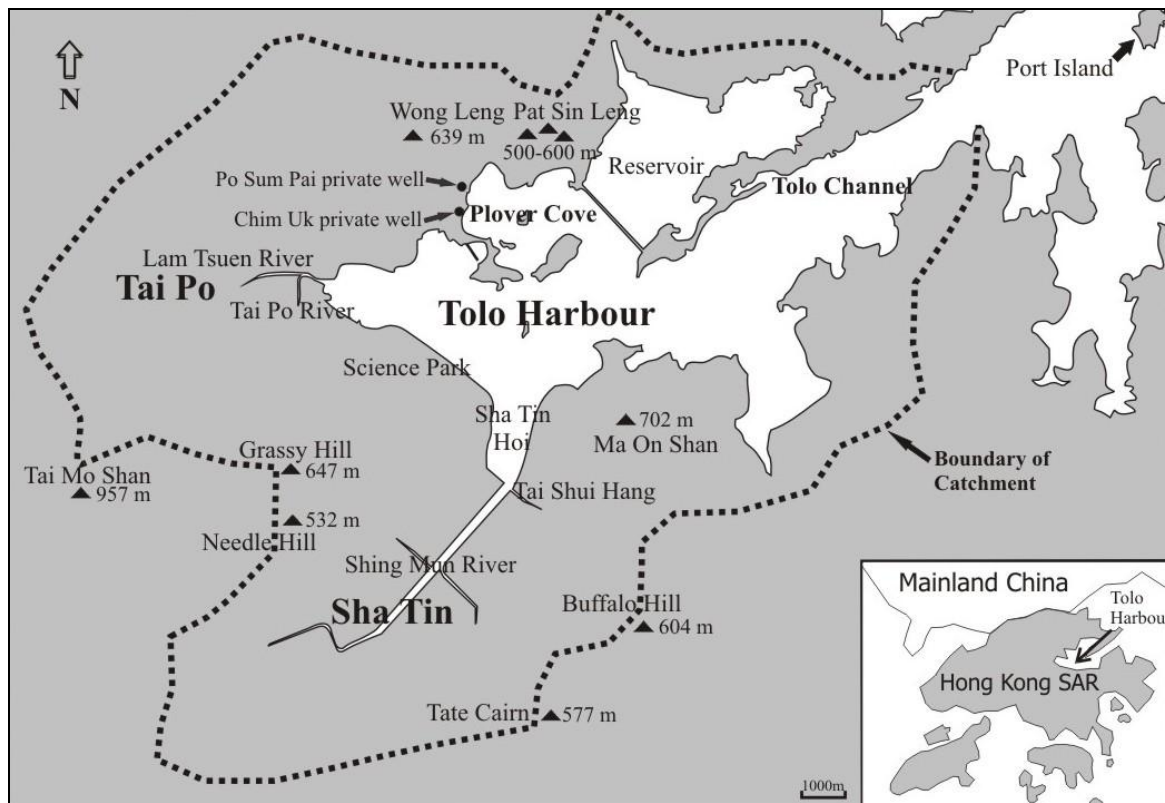


Figure 1 Map of Tolo Harbour

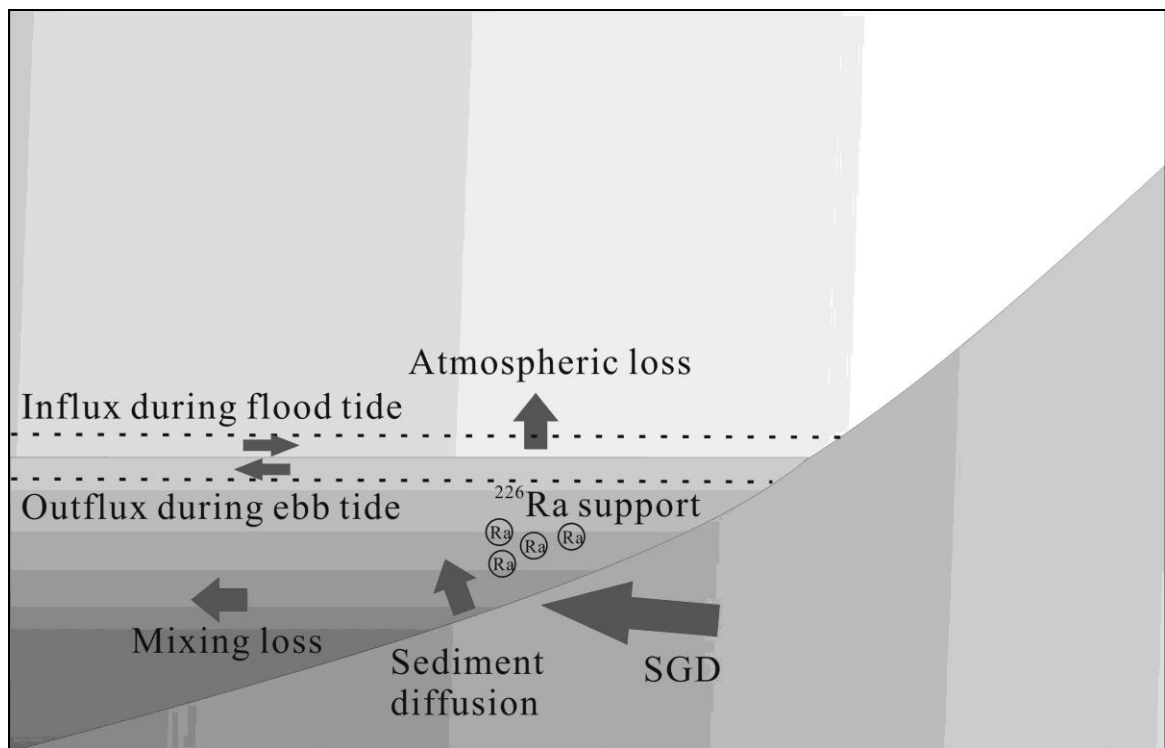


Figure 2 Sources and sinks of ^{222}Rn in coastal waters (modified from Lambert and Burnett 2003)

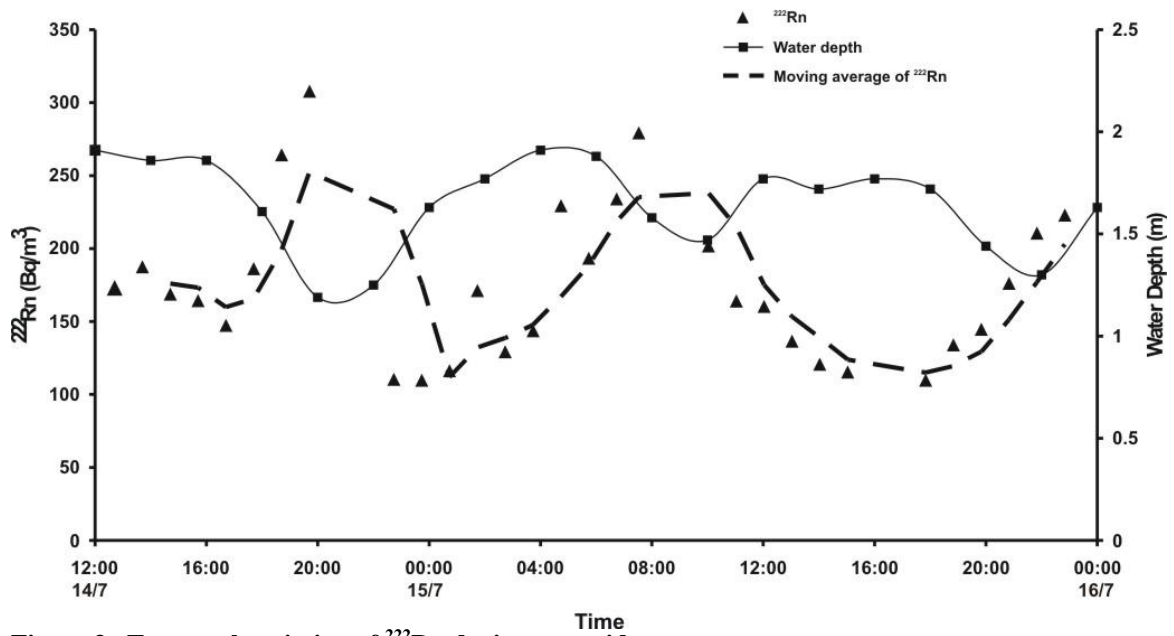


Figure 3 Temporal variation of ^{222}Rn during neap tide

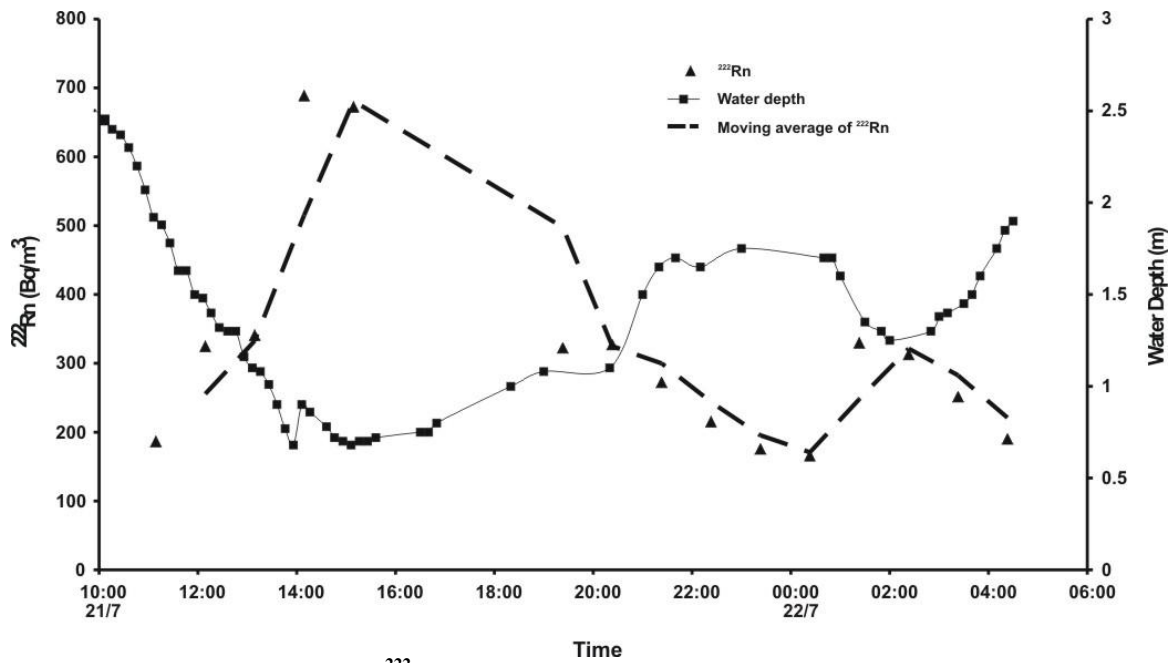


Figure 4 Temporal variation of ^{222}Rn during spring tide

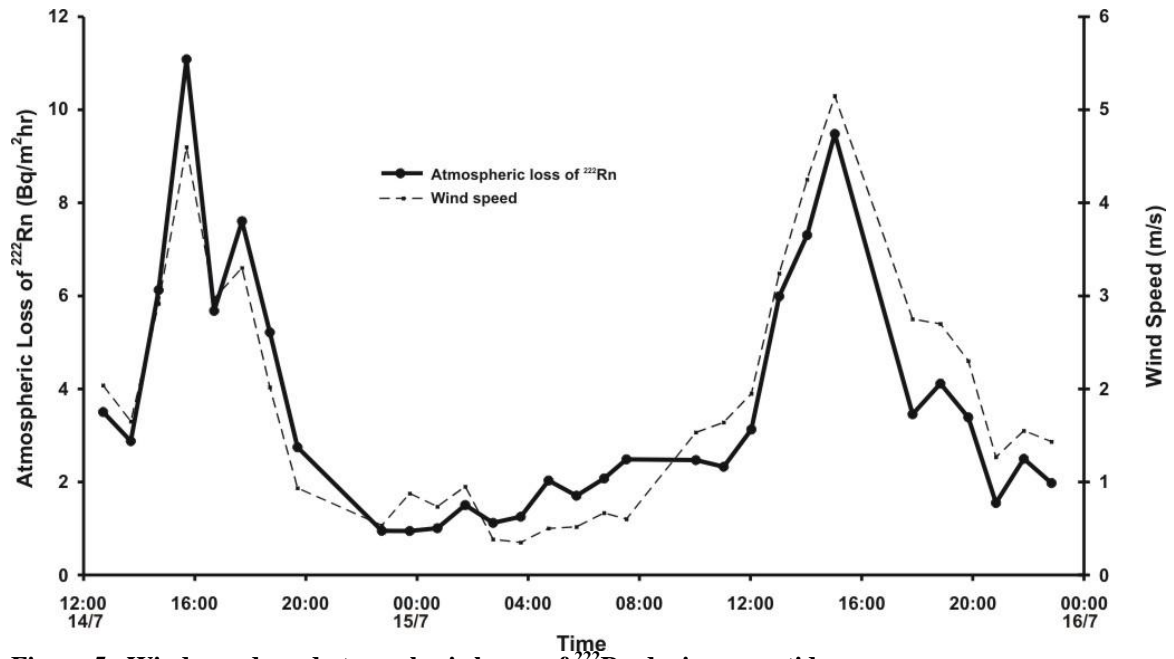


Figure 5 Wind speeds and atmospheric losses of ^{222}Rn during neap tide

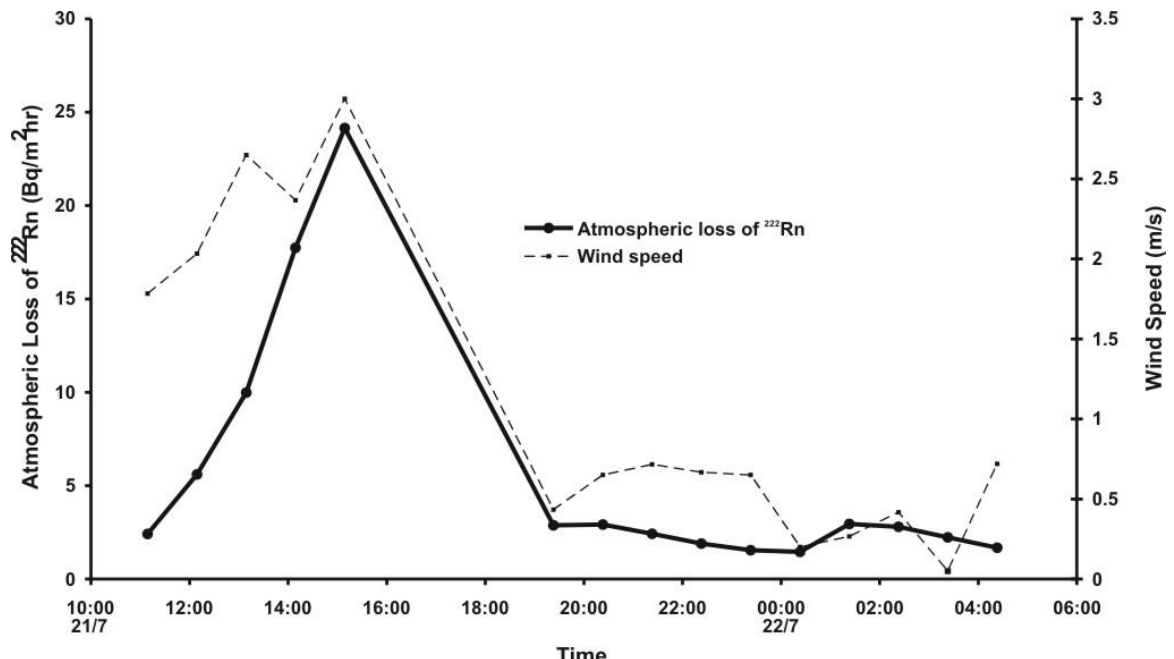


Figure 6 Wind speeds and atmospheric losses of ^{222}Rn during spring tide

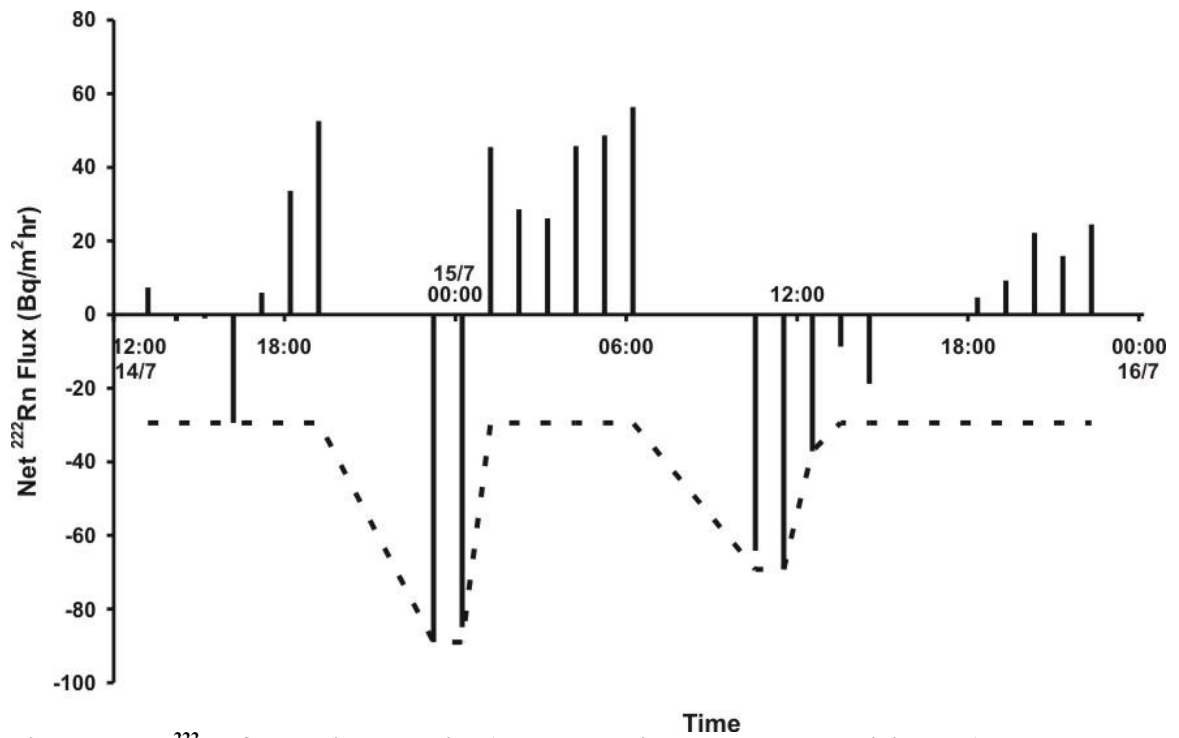


Figure 7 Net ^{222}Rn flux during neap tide (The dashed line represents the mixing loss)

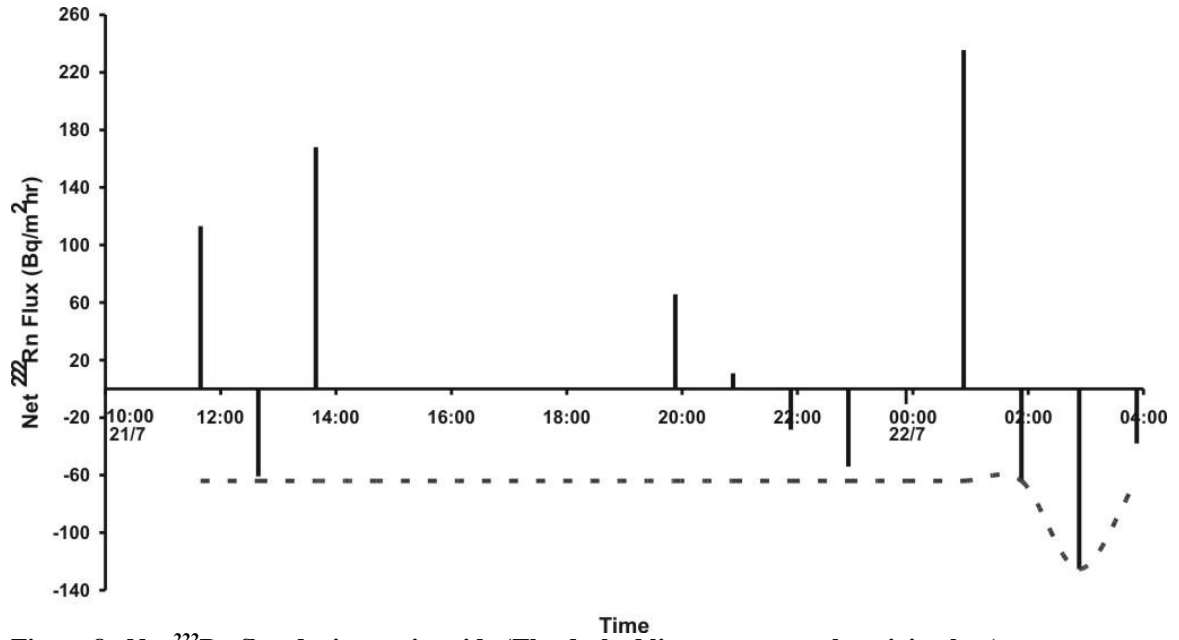


Figure 8 Net ^{222}Rn flux during spring tide (The dashed line represents the mixing loss)

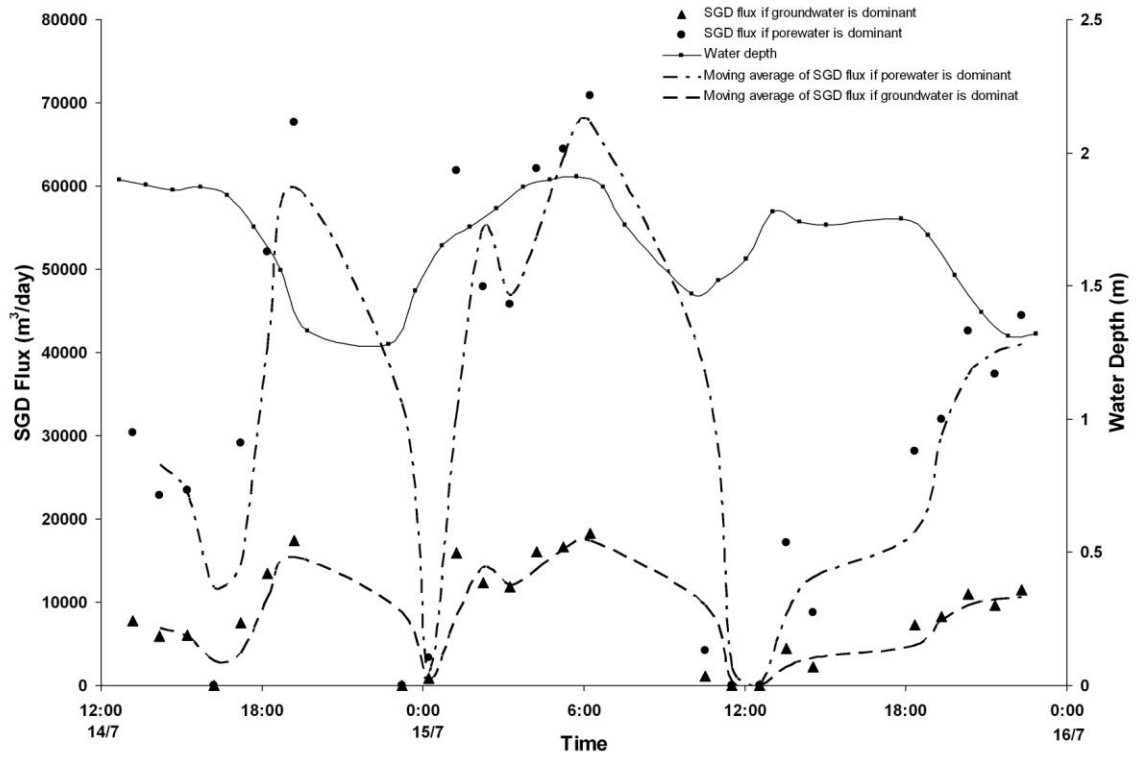


Figure 9 SGD fluxes during neap tide

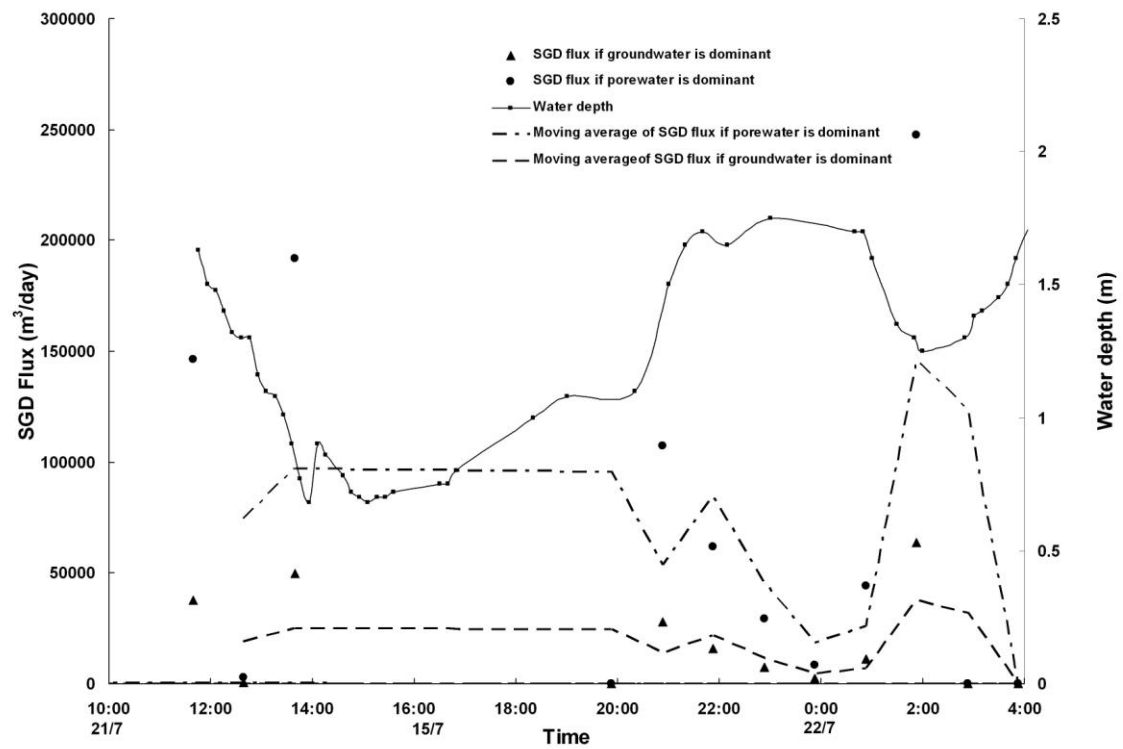


Figure 10 SGD fluxes during spring tide