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# Effects of Wet N<sub>2</sub>O Oxidation on Interface Properties of 6H-SiC MOS Capacitors

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Abstract—Oxynitrides were grown on n- and p-type 6H-SiC by wet  $N_2O$  oxidation (bubbling  $N_2O$  gas through deionized water at 95°C) or dry  $N_2O$  oxidation followed by wet  $N_2O$  oxidation. Their oxide/SiC interfaces were investigated for fresh and stressed devices. It was found that both processes improve p-SiC/oxide but deteriorate n-SiC/oxide interface properties when compared to dry  $N_2O$  oxidation alone. The involved mechanism could be enhanced removal of unwanted carbon compounds near the interface due to the wet ambient, and hence a reduction of donor-like interface states for the p-type devices. As for the n-type devices, incorporation of hydrogen-related species near the interface under the wet ambient increases acceptor-like interface states. In summary, the wet  $N_2O$  oxidation can be used for providing comparable reliability for n- and p-SiC MOS devices, and especially obtaining high-quality oxide-SiC interface in p-type MOS devices.

Index Terms—Interface-state density, MOS capacitors, silicon carbide, wet  $N_2O$  oxidation.

## I. INTRODUCTION

HERE has been considerable progress in improving qualities of oxide on SiC and oxide/SiC interface, mainly based on nitridation processes. Although N2O-annealed oxides showed an increase in interface-state density [1], [2], NO-annealed oxides exhibited improved electrical properties: reduced interface-state density [1]-[4], enhanced reliability [5], and increased channel-carrier mobility [6]. Recently, oxynitrides directly grown in N<sub>2</sub>O [7] and NO [8] showed better oxide/SiC interface and bulk properties than their respective counterparts annealed in the same gas. However, an obvious problem is that nitridation-induced improvements of n-SiC and p-SiC interface characteristics are different, usually with the former larger than the latter. In this work, the effects of N<sub>2</sub>O oxidation in a wet ambient on the interface properties and reliability of n- and p-SiC MOS devices are investigated. When compared with the dry N<sub>2</sub>O oxidation, an opposite phenomenon is observed. Wet N<sub>2</sub>O oxidation improves p-SiC/oxide but deteriorates n-SiC/oxide interface properties. Moreover, experimental results show that a dry-wet N<sub>2</sub>O oxidation can be used to obtain comparable reliability for n-channel and p-channel SiC MOS

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devices. In this letter, relevant results are reported and physical mechanisms involved are analyzed.

#### **II. EXPERIMENTS**

N- and p-type (0001) Si-face 6H-SiC wafers, manufactured by CREE Research, were used in this study. The SiC wafers had a 5- $\mu$ m epitaxial layer grown on heavily doped substrates (i.e.,  $n/n^+$  and  $p/p^+$ ). The doping level of the epitaxial layer was  $4 \times 10^{15}$  cm<sup>-3</sup> for both types. The wafers were cleaned using the conventional RCA method followed by a 1-min dip in 5% HF, and then loaded into a quartz furnace at 800°C. After raising the temperature to 1150°C, a 4-h oxidation was carried out in a dry pure N<sub>2</sub>O (denoted as DN2OG sample) or a wet N<sub>2</sub>O ambient by bubbling pure N2O gas through de-ionized water at 95°C (denoted as WN2OG sample), both at a flow rate of 500 ml/min. The third sample was oxidized first in dry pure N<sub>2</sub>O for 2.5 h and then in wet N<sub>2</sub>O for 1.5 h at the same temperature and flow rate (denoted as DWN2OG). After cooling down in N<sub>2</sub> with a ramping rate of  $-1^{\circ}C/\min$  to 950°C, all samples were subjected to postoxidation annealing in  $N_2$  at a flow of 1.0 l/min for 0.5 h. Finally, aluminum was thermally evaporated, and then patterned to form gate electrodes, each with an area of  $1.14 \times 10^{-3}$  cm<sup>2</sup>. This step was followed by a forming gas anneal at 410°C for 30 min. High-frequency (HF, 1 MHz) capacitance–voltage (C-V) characteristics were measured using HP4284A LCR meter at room temperature with a bias sweep rate of 0.1 V/s and a small signal amplitude of 20 mV. Accumulation capacitance, flat-band voltage, equivalent oxide charge and average interface-state density were extracted from the HF capacitance-voltage (C-V) curves. Highfield stress (±7 MV/cm) applied by HP4156B precision semiconductor parameter analyzer, with the capacitors biased in accumulation, was used to examine device reliability in terms of the flat-band voltage shift ( $\Delta V_{\rm fb}$ ). Stress times were 10, 100, 500, 1000, 2000, 3000, and 5000 s. All measurements were carried out under a light-tight and electrically-shielded condition.

# **III. RESULTS AND DISCUSSION**

Fig. 1(a) and (b) show typical HF C-V curves under the dark condition for n- and p-type 6H-SiC MOS capacitors fabricated using different processing conditions. Some parameters extracted from these HF C-V curves are listed in Table I, where equivalent oxide thickness  $t_{\text{ox}}$  is calculated from accumulation capacitance (i.e., oxide capacitance  $C_{\text{ox}}$ ) using the dielectric constant of SiO<sub>2</sub> (3.9), and flat-band voltage  $V_{\text{fb}}$  is determined

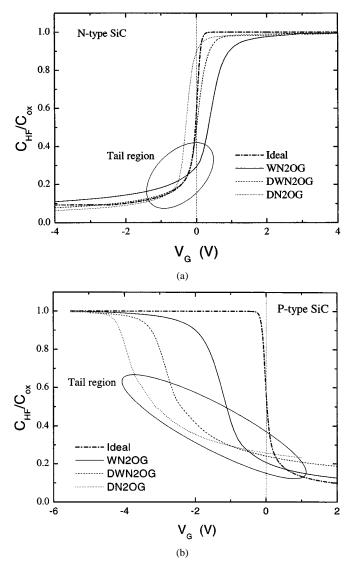


Fig. 1. Typical high-frequency C-V curves of the three kinds of capacitors under the dark condition at room temperature. (a) n-type SiC MOS and (b) p-type SiC MOS. Area of capacitor is  $1.14 \times 10^{-3}$  cm<sup>2</sup>. For clarity, the ideal curves are shown only for the WN2OG devices.

from the flat-band capacitance  $(C_{\rm fb})$  formula [9] modified by replacing the dielectric constant of Si with that of SiC:

$$\frac{C_{\rm fb}}{C_{\rm ox}} = \left(1 + \frac{150\sqrt{\frac{T}{300}}}{t_{\rm ox}\sqrt{N}}\right)^{-1}$$

with T the temperature in Kelvin and the carrier concentration N equal to the acceptor or donor doping as an approximation. Equivalent oxide-charge density is calculated as  $Q_{\rm ox} = -C_{\rm ox}(V_{\rm fb} - \phi_{\rm ms})/q$ , where the work-function difference  $\phi_{\rm ms}$ between aluminum and 6H-SiC is calculated to be 0.031 V for n-type SiC and -2.53 V for p-type SiC. The interface-state density ( $D_{\rm it}$ ) is taken as the average value between 0.5 and 0.6 eV away from the edge of conduction band (n-type SiC) or valence band (p-type SiC), estimated from the HF C-V curve using the Terman method [10]. From Table I and Fig. 1, three points can be summarized: 1) the growth rate of wet N<sub>2</sub>O oxidation is higher than that of dry N<sub>2</sub>O oxidation, probably due to

	N-type SiC			P-type SiC		
Sample	WN2OG	DWN2OG	DN2OG	WN2OG	DWN2OG	DN2OG
t <sub>ox (</sub> A)	328	280	246	317	272	240
Cfb/Cox	0.58	0.54	0.51	0.57	0.54	0.50
V <sub>fb</sub> (V)	0.42	0.04	-0.29	-1.23	-2.55	-3.13
$\begin{array}{c} Q_{ox} \times 10^{11} \\ cm^{-2} \end{array}$	-2.56	-0.07	2.81	-8.84	0.16	5.39
$D_{it} \times 10^{11}$ cm <sup>-2</sup> eV <sup>-1</sup>	4.5	3.1	2.2	3.0	4.2	5.7

higher oxidation rate in steam atmosphere than in dry ambient and enhanced out-diffusion of CO (one of the factors limiting the growth rate in wet oxidation [11]); 2) both wet and dry-wet N<sub>2</sub>O oxidations shift the C-V curves of the p-type SiC samples in the positive  $-V_G$  direction and hence reduce their  $|V_{\rm fb}|$ ; and 3) from the slope of the C-V curve in the depletion region and the size of the tail in the deep-depletion region, it can be deduced that compared with dry N<sub>2</sub>O oxidation, wet and dry-wet N<sub>2</sub>O oxidations improve the oxide/SiC interface quality for p-type device, but deteriorate the n-type oxide/SiC interface, similar to the results of wet O<sub>2</sub> oxidation on SiC [12], [13]. This is supported by the estimated  $D_{\rm it}$ 's in Table I.

It is believed that the interface states in the upper half and lower half of the band gap are quite different (acceptor-like and donor-like, respectively), and originate from dangling bonds and carbon compounds at the interface, respectively [13]. (For acceptor-like interface states, there could be other possible origins, e.g., Si-Si bonds at the interface which induce symmetric defect states at the two band edges [4].) Therefore, as compared with the dry N<sub>2</sub>O-grown sample, the improved interface properties of p-type MOS capacitors prepared using wet or dry-wet N<sub>2</sub>O oxidation may be associated with the effective removal of unwanted carbon compounds (donor-like interface states) due to two possible effects of the wet ambient: 1) enhanced out-diffusion of CO as mentioned previously and 2) enhanced removal of interstitial carbon as well as carbon clusters by nitridation [14]. On the other hand, the inferior interface quality of n-type MOS capacitors could be attributed to the creation of more acceptor-like interface states in the wet oxidation ambient which is speculated to result from hydrogen-related species such as -OH bonds in the steam [12], [13], [15]. These could be considered as the main reason for the positive flat-band shift of the wet N<sub>2</sub>O-oxidized samples, because the increase in the negative fixed oxide charge is negligible compared with the change in the interface states during wet oxidation [12]. Therefore, the more negative  $Q_{\text{ox}}$  in the WN2OG sample in Table I does not indicate a real increase in the negative fixed oxide charge, but instead indicates a reduction in the donor-like interface states for p-type devices or an increase in the acceptor-like interface states for n-type devices.

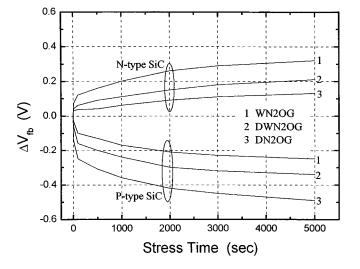


Fig. 2. Flat-band voltage shifts of all samples under high-field stressing ( $\pm 7$  MV/cm) at room temperature, with capacitors biased in accumulation. Area of capacitor is  $1.14 \times 10^{-3}$  cm<sup>2</sup>.

Fig. 2 is the flat-band voltage shift of all samples during high-field stressing  $(\pm 7 \text{ MV/cm})$  with their gates biased in accumulation at room temperature. The stress field is determined as  $(V_G - V_{\rm fb})/t_{\rm ox}$  to account for the difference in the flat-band voltages of the samples under comparison. It should be noted that the  $|\Delta V_{\rm fb}|$ 's of all these nitrided samples are approximately  $10 \times$  smaller than that of a conventional O<sub>2</sub>-grown sample [7]. In agreement with the improvement of the interface quality in Fig. 1, the WN2OG sample exhibits the best and worst reliability for p-type and n-type substrates, respectively, with the DWN2OG sample lying in between. Also, there is an opposite change in  $V_{\rm fb}$  for p-type and n-type devices, indicating that donor (acceptor)-like interface states and positive (negative) oxide charges are generated in p (n)-type device under the high-field stressing condition. A smaller  $|\Delta V_{\rm fb}|$  for p-type WN2OG and DWN2OG samples than the p-type DN2OG sample implies stronger oxide/SiC interface in the former two samples, and thus less generation of donor-like interface states and positive oxide charges. This indicates that a wet ambient is beneficial for the effective removal of carbon-related species as well as the creation of more strong  $Si \equiv N$  and  $N \equiv O$  bonds. In addition, from Fig. 2, it can be observed that  $\Delta V_{\rm fb}$  of both DWN2OG samples always lie between those of WN2OG and DN2OG samples. Therefore, it is possible for the dry-wet N<sub>2</sub>O oxidation to produce high-quality p- and n-channel SiC MOS devices with equal reliability, when the times for dry and wet oxidations are properly adjusted.

#### IV. SUMMARY

Improved p-SiC/oxide interface quality was demonstrated in MOS devices with wet or dry-wet N<sub>2</sub>O-grown oxynitride as the gate dielectric. Enhanced reliability of the two p-type SiC MOS devices was also observed under high-field stress. These facts indicate that wet N<sub>2</sub>O oxidation is a promising process for the fabrication of high-performance p-SiC MOS devices. Moreover, when the wet N<sub>2</sub>O oxidation is combined with dry N<sub>2</sub>O oxidation, p-type SiC MOS devices still show improved interface properties, with acceptable degradation in its n-type counterpart. As a result, this can be an efficient approach for improving the interface properties of p- and n-channel SiC MOS devices, and even for achieving identical reliability for both devices.

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