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(5*R*)-3-(2-Chloroacetyl)-4-methyl-5-phenyl-1,3,4-oxadiazinan-2-one

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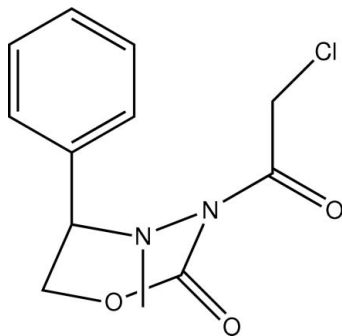
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Key indicators: single-crystal X-ray study; $T = 100$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; R factor = 0.020; wR factor = 0.054; data-to-parameter ratio = 14.5.

The 1,3,4-oxadiazinan-2-one ring in the title compound, $\text{C}_{12}\text{H}_{13}\text{ClN}_2\text{O}_3$, is in a distorted half-chair conformation. The phenyl and chloroacetyl groups occupy axial and equatorial positions, respectively, and lie to the opposite side of the molecule to the N-bound methyl substituent. Molecules are consolidated in the crystal structure by $\text{C}-\text{H}\cdots\text{O}$ interactions.

Related literature

For background to 1,3,4-oxadiazin-2-ones, see: Trepanier *et al.* (1968); Roussi *et al.* (1998, 1999, 2000); Casper *et al.* (2002*a,b*); Bonin *et al.* (2006). For a related structure, see: Zukerman-Schpector *et al.* (2009). For the synthesis, see: Rodrigues *et al.* (2005). For conformational analysis, see: Cremer & Pople (1975).



Experimental

Crystal data

$\text{C}_{12}\text{H}_{13}\text{ClN}_2\text{O}_3$
 $M_r = 268.69$
 Orthorhombic, $P2_12_12_1$
 $a = 9.4862$ (2) Å
 $b = 9.6237$ (2) Å
 $c = 13.2433$ (3) Å
 $V = 1209.01$ (5) Å³
 $Z = 4$
 Mo $K\alpha$ radiation
 $\mu = 0.32$ mm⁻¹
 $T = 100$ K
 $0.35 \times 0.30 \times 0.25$ mm

Data collection

Bruker APEXII CCD diffractometer
 29813 measured reflections
 2378 independent reflections
 2334 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.026$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.020$
 $wR(F^2) = 0.054$
 $S = 1.07$
 2378 reflections
 164 parameters
 H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.19$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.20$ e Å⁻³
 Absolute structure: Flack (1983), 993 Friedel pairs
 Flack parameter: 0.01 (5)

Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{C}2-\text{H}2\cdots\text{O}3^{\text{i}}$	1.00	2.60	3.5142 (15)	153
$\text{C}8-\text{H}8\cdots\text{O}2^{\text{ii}}$	0.95	2.60	3.2689 (16)	128
$\text{C}11-\text{H}11\text{b}\cdots\text{O}2^{\text{iii}}$	0.99	2.54	3.3675 (16)	141
$\text{C}12-\text{H}12\text{a}\cdots\text{O}2^{\text{iii}}$	0.98	2.57	3.5448 (16)	173

Symmetry codes: (i) $-x + \frac{1}{2}, -y, z - \frac{1}{2}$; (ii) $x + \frac{1}{2}, -y + \frac{1}{2}, -z + 2$; (iii) $-x, y - \frac{1}{2}, -z + \frac{3}{2}$.

Data collection: *APEX2* (Bruker, 2007); cell refinement: *SAINT* (Bruker, 2007); data reduction: *SAINT*; program(s) used to solve structure: *SIR97* (Altomare *et al.*, 1999); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3* (Farrugia, 1997) and *DIAMOND* (Brandenburg, 2006); software used to prepare material for publication: *MarvinSketch* (Chemaxon, 2010) and *publCIF* (Westrip, 2010).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HG5045).

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supporting information

Acta Cryst. (2011). E67, o1571–o1572 [doi:10.1107/S1600536811020356]

(5*R*)-3-(2-Chloroacetyl)-4-methyl-5-phenyl-1,3,4-oxadiazinan-2-one

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S1. Comment

About forty years ago Trepanier and collaborators reported the first synthesis of some 3,4,5,6-tetrahydro-2*H*-1,3,4-oxadiazin-2-ones as candidates for central nervous system stimulant activity (Trepanier *et al.*, 1968). After three decades from the discovery of 1,3,4-oxadiazin-2-ones, Husson, Micouin and co-workers (Roussi *et al.*, 1998) successfully employed this class of compound as chiral auxiliaries in diastereoselective alkylations and in dipolar cycloadditions (Roussi *et al.*, 1999, 2000, Bonin *et al.*, 2006). In a different approach, Hitchcock and collaborators have been successfully applying 1,3,4-oxadiazinan-2-one derivatives as chiral auxiliaries in asymmetric aldol addition reactions (Casper *et al.*, 2002*b*). Besides their applications in asymmetric synthesis, this class of compounds show a very interesting conformational behaviour (Casper *et al.*, 2002*a*). To explore the adopted conformation of the title compound, (I), in the solid-state, an X-ray study was performed.

The crystal structure analysis of (I) confirms the *R* configuration at atom C2, Fig. 1, in accord with expectation from the synthesis. The 1,3,4-oxadiazinan-2-one ring is in a distorted half-chair conformation, as shown by the ring-puckering parameters: $q_2 = 0.364$ (1) Å, $q_3 = -0.333$ (1) Å, $Q = 493$ (1) Å and $\varphi_2 = 32.0$ (2) ° (Cremer and Pople, 1975). The deviations of the O1, C1, N1, N2, C2 and C3 atoms from their least-squares plane are 0.0424 (10) -0.0497 (13), -0.1285 (11), 0.3139 (11), -0.3212 (13) and 0.1430 (13) Å, respectively. The observed conformation contrasts the twisted chair conformation found in the only other single-ring 1,3,4-oxadiazinan-2-one structure known (Zukerman-Schpector *et al.*, 2009). The chloridoacetyl and phenyl groups lie to the same side of the molecule and opposite to that of the N-bound methyl group. The dihedral angles formed between the 1,3,4-oxadiazinan-2-one ring and the phenyl and chloridoacetyl (C1,O3,C10,C11) groups are 87.69 (6) and 14.41 (3) °, respectively, consistent with axial and equatorial substitution. Molecules are consolidated in the crystal packing by C—H···O interactions operating in three-dimensions, Table 1 and Fig. 2.

S2. Experimental

The starting (*R*)-4-methyl-5-phenyl-1,3,4-oxadiazinan-2-one was synthesized by using the same procedure as previously reported (Rodrigues *et al.* 2005). The chloroacetyl-1,3,4-oxadiazinan-2-one (I) species was prepared by the acylation reaction of 1,3,4-oxadiazinan-2-one. A solution of BuLi (2.00 *M* in hexane, 1.45 ml, 2.86 mmol) was added drop wise to a solution of 1,3,4-oxadiazinan-2-one (500 mg, 2.60 mmol) in dry THF (10 ml) at 195 K and the reaction was stirred for an additional 15 min. A chloroacetyl chloride (230 μ L, 2.86 mmol) solution in THF (1 ml) was added slowly to the reaction mixture. After 15 min, the light-yellow solution was warmed to RT for a further 30 min. The reaction was quenched with saturated aqueous ammonium chloride solution (5 ml). The mixture was concentrated under reduced pressure, taken up in water (5 ml) and extracted with DCM (3 times; 15 ml) then dried (MgSO₄). Evaporation of the solvent *in vacuo* gave the crude product which was purified by flash column chromatography on silica gel with 40%

EtOAc in hexanes to give the pure product as a colourless solid (572 mg, 82%). Colourless crystals of (I) were obtained by vapour diffusion from hexane/acetone at 298 K. mp = 408 – 410 K; $[\alpha]_D^{25} +52.5^\circ$ (*c* 1.02, CHCl₃); ¹H NMR (500 MHz, CDCl₃/TMS), δ (p.p.m.): 7.40–7.35 (m, 5H), 4.88 (dd, ²*J* = 11.5 Hz, ³*J* = 5.3 Hz, 1H), 4.77 (dd, ²*J* = 11.5 Hz, ³*J* = 7.7 Hz, 1H), 4.43 (*AB* spin system, *D*_n = 36.0 Hz, ²*J* = 15.3 Hz, 2H), 4.40 (dd, ³*J* = 7.7 Hz, ³*J* = 5.3 Hz, 1H), 2.81 (s, 3H). ¹H NMR (125 MHz, CDCl₃/TMS), δ (p.p.m.): 166.07, 149.54, 134.69, 129.16, 128.90, 127.02, 67.65, 61.54, 44.75, 42.12. Anal. calcd for C₁₂H₁₃ClN₂O₃: C, 53.64%; H, 4.88%; N, 10.43%. Found: C, 53.46%; H, 4.92%; N, 10.49%.

S3. Refinement

Carbon-bound H-atoms were placed in calculated positions (C—H 0.95 to 1.00 Å) and were included in the refinement in the riding model approximation, and with $U_{iso}(H) = 1.2U_{eq}(C)$ and $1.5U_{eq}(\text{methyl-C})$.

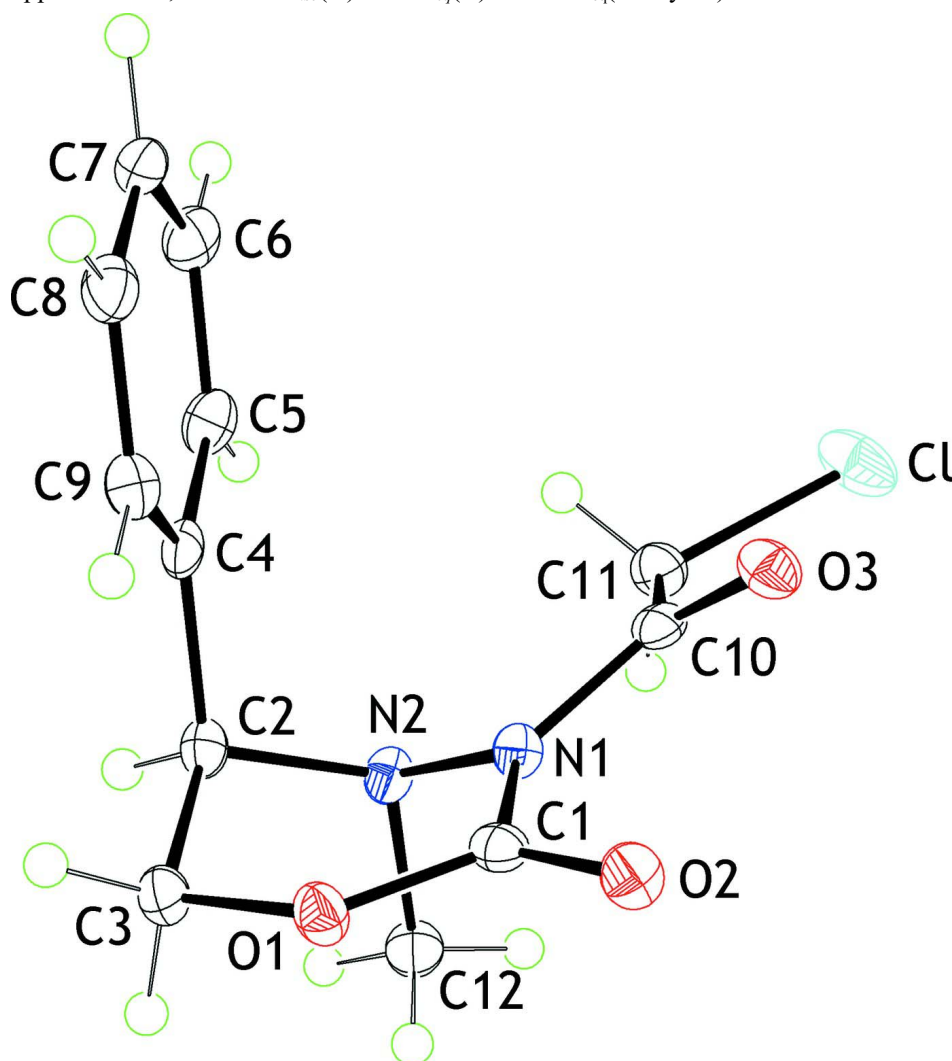


Figure 1

The molecular structure of (I) showing atom labelling scheme and displacement ellipsoids at the 50% probability level (arbitrary spheres for the H atoms).

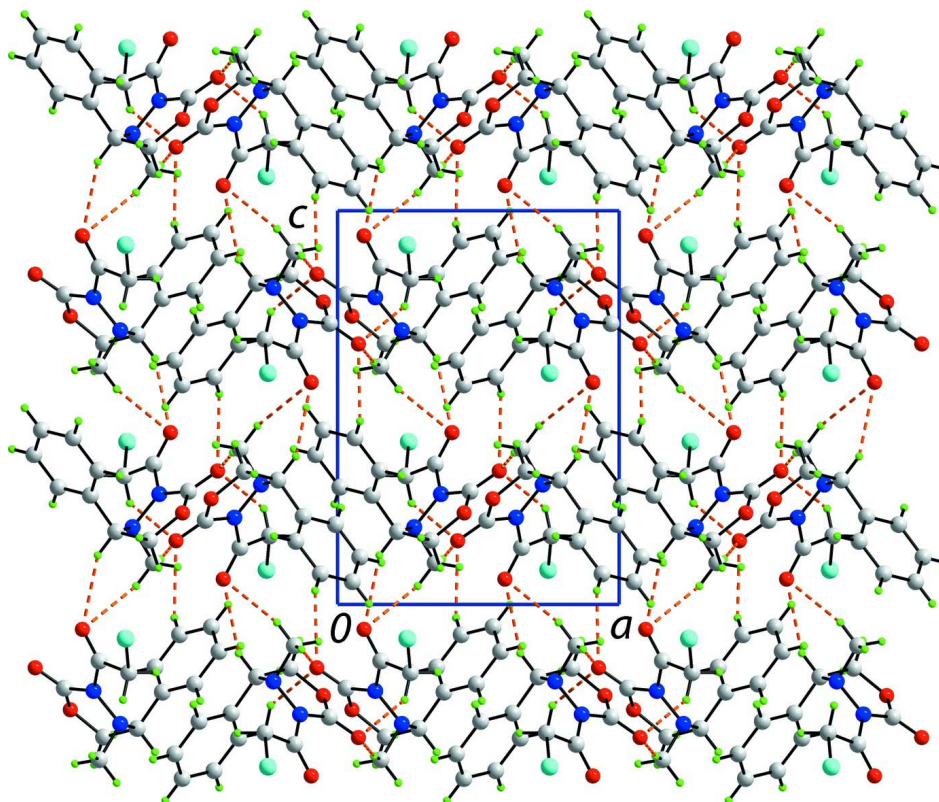


Figure 2

A view of the unit-cell contents in (I) viewed in projection down the b axis showing the C—H...O interactions as orange dashed lines.

(5R)-3-(2-Chloroacetyl)-4-methyl-5-phenyl-1,3,4-oxadiazinan-2-one

Crystal data

$C_{12}H_{13}ClN_2O_3$

$M_r = 268.69$

Orthorhombic, $P2_12_12_1$

Hall symbol: P 2ac 2ab

$a = 9.4862$ (2) Å

$b = 9.6237$ (2) Å

$c = 13.2433$ (3) Å

$V = 1209.01$ (5) Å³

$Z = 4$

$F(000) = 560$

$D_x = 1.476$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 2003 reflections

$\theta = 2.8$ – 16.3°

$\mu = 0.32$ mm⁻¹

$T = 100$ K

Block, colourless

$0.35 \times 0.30 \times 0.25$ mm

Data collection

Bruker APEXII CCD

diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

ω scans

29813 measured reflections

2378 independent reflections

2334 reflections with $I > 2\sigma(I)$

$R_{int} = 0.026$

$\theta_{max} = 26.0^\circ$, $\theta_{min} = 2.6^\circ$

$h = -11 \rightarrow 11$

$k = -11 \rightarrow 11$

$l = -16 \rightarrow 16$

*Refinement*Refinement on F^2

Least-squares matrix: full

 $R[F^2 > 2\sigma(F^2)] = 0.020$ $wR(F^2) = 0.054$ $S = 1.07$

2378 reflections

164 parameters

0 restraints

Primary atom site location: structure-invariant
direct methodsSecondary atom site location: difference Fourier
mapHydrogen site location: inferred from
neighbouring sites

H-atom parameters constrained

 $w = 1/[\sigma^2(F_o^2) + (0.028P)^2 + 0.320P]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{\max} = 0.001$ $\Delta\rho_{\max} = 0.19 \text{ e } \text{\AA}^{-3}$ $\Delta\rho_{\min} = -0.20 \text{ e } \text{\AA}^{-3}$ Absolute structure: Flack (1983), 993 Friedel
pairs

Absolute structure parameter: 0.01 (5)

Special details

Geometry. All s.u.'s (except the s.u. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell s.u.'s are taken into account individually in the estimation of s.u.'s in distances, angles and torsion angles; correlations between s.u.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell s.u.'s is used for estimating s.u.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > 2\sigma(F^2)$ is used only for calculating R -factors(gt) etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
Cl	0.24739 (4)	-0.34782 (3)	0.91280 (2)	0.02998 (10)
O1	0.05085 (9)	0.22385 (9)	0.73203 (7)	0.0182 (2)
O2	-0.07381 (9)	0.09934 (10)	0.83862 (7)	0.0196 (2)
O3	0.09385 (10)	-0.08909 (10)	0.93619 (7)	0.0200 (2)
N1	0.13328 (11)	0.00514 (11)	0.77991 (8)	0.0146 (2)
N2	0.22794 (11)	0.00168 (11)	0.69603 (7)	0.0156 (2)
C1	0.02961 (13)	0.10834 (13)	0.78683 (9)	0.0153 (2)
C2	0.28807 (12)	0.14314 (13)	0.68462 (9)	0.0165 (2)
H2	0.3517	0.1421	0.6244	0.020*
C3	0.16885 (13)	0.24374 (14)	0.66273 (9)	0.0184 (3)
H3A	0.2041	0.3402	0.6689	0.022*
H3B	0.1359	0.2301	0.5925	0.022*
C4	0.37768 (13)	0.17604 (13)	0.77675 (9)	0.0163 (2)
C5	0.49224 (13)	0.08997 (14)	0.79745 (10)	0.0200 (3)
H5	0.5116	0.0132	0.7545	0.024*
C6	0.57847 (14)	0.11524 (15)	0.88025 (11)	0.0228 (3)
H6	0.6556	0.0553	0.8942	0.027*
C7	0.55191 (14)	0.22803 (15)	0.94258 (10)	0.0221 (3)
H7	0.6113	0.2458	0.9988	0.027*
C8	0.43859 (15)	0.31474 (13)	0.92274 (10)	0.0212 (3)
H8	0.4203	0.3921	0.9654	0.025*
C9	0.35165 (14)	0.28841 (13)	0.84028 (10)	0.0185 (3)
H9	0.2737	0.3477	0.8272	0.022*

C10	0.15311 (13)	-0.09421 (13)	0.85593 (9)	0.0157 (2)
C11	0.25837 (14)	-0.20626 (13)	0.82638 (9)	0.0194 (3)
H11A	0.3550	-0.1674	0.8271	0.023*
H11B	0.2378	-0.2394	0.7571	0.023*
C12	0.15380 (14)	-0.04923 (14)	0.60530 (9)	0.0195 (3)
H12A	0.1286	-0.1472	0.6146	0.029*
H12B	0.2155	-0.0399	0.5464	0.029*
H12C	0.0680	0.0056	0.5947	0.029*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.0483 (2)	0.01763 (15)	0.02398 (16)	0.00875 (16)	-0.00658 (16)	0.00243 (12)
O1	0.0178 (4)	0.0164 (4)	0.0205 (4)	0.0022 (4)	0.0018 (4)	0.0041 (4)
O2	0.0179 (4)	0.0206 (5)	0.0202 (4)	0.0031 (4)	0.0031 (4)	0.0011 (4)
O3	0.0249 (5)	0.0204 (4)	0.0148 (4)	0.0010 (4)	-0.0004 (4)	-0.0004 (4)
N1	0.0149 (5)	0.0139 (5)	0.0150 (5)	-0.0001 (4)	0.0019 (4)	-0.0008 (4)
N2	0.0170 (5)	0.0159 (5)	0.0138 (5)	-0.0026 (4)	0.0035 (4)	-0.0031 (4)
C1	0.0170 (6)	0.0150 (6)	0.0137 (5)	0.0002 (5)	-0.0030 (5)	-0.0017 (5)
C2	0.0177 (6)	0.0157 (6)	0.0162 (5)	-0.0024 (5)	0.0035 (4)	0.0002 (5)
C3	0.0196 (6)	0.0191 (6)	0.0164 (6)	-0.0026 (5)	0.0014 (5)	0.0025 (5)
C4	0.0157 (6)	0.0163 (6)	0.0170 (5)	-0.0050 (5)	0.0035 (5)	0.0016 (5)
C5	0.0160 (6)	0.0195 (6)	0.0245 (7)	-0.0014 (5)	0.0050 (5)	-0.0024 (5)
C6	0.0142 (6)	0.0263 (7)	0.0280 (7)	-0.0014 (5)	0.0016 (5)	0.0039 (6)
C7	0.0200 (6)	0.0271 (7)	0.0193 (6)	-0.0099 (5)	-0.0015 (5)	0.0034 (5)
C8	0.0270 (7)	0.0181 (6)	0.0186 (6)	-0.0052 (5)	0.0019 (5)	-0.0011 (5)
C9	0.0210 (6)	0.0153 (6)	0.0192 (6)	-0.0001 (5)	0.0029 (5)	0.0012 (5)
C10	0.0169 (6)	0.0141 (6)	0.0160 (6)	-0.0031 (5)	-0.0046 (5)	-0.0024 (5)
C11	0.0210 (6)	0.0156 (6)	0.0218 (6)	0.0012 (5)	-0.0027 (5)	0.0017 (5)
C12	0.0233 (7)	0.0199 (6)	0.0154 (6)	-0.0024 (5)	-0.0006 (5)	-0.0034 (5)

Geometric parameters (Å, °)

C1—C11	1.7823 (13)	C4—C5	1.3937 (18)
O1—C1	1.3428 (15)	C5—C6	1.3895 (19)
O1—C3	1.4601 (15)	C5—H5	0.9500
O2—C1	1.2001 (15)	C6—C7	1.387 (2)
O3—C10	1.2034 (16)	C6—H6	0.9500
N1—C1	1.4006 (16)	C7—C8	1.386 (2)
N1—C10	1.4012 (16)	C7—H7	0.9500
N1—N2	1.4287 (14)	C8—C9	1.3917 (19)
N2—C12	1.4760 (15)	C8—H8	0.9500
N2—C2	1.4838 (16)	C9—H9	0.9500
C2—C3	1.5167 (18)	C10—C11	1.5209 (17)
C2—C4	1.5203 (17)	C11—H11A	0.9900
C2—H2	1.0000	C11—H11B	0.9900
C3—H3A	0.9900	C12—H12A	0.9800
C3—H3B	0.9900	C12—H12B	0.9800

C4—C9	1.3923 (18)	C12—H12C	0.9800
C1—O1—C3	124.28 (10)	C7—C6—C5	119.99 (12)
C1—N1—C10	122.08 (11)	C7—C6—H6	120.0
C1—N1—N2	120.58 (10)	C5—C6—H6	120.0
C10—N1—N2	117.28 (10)	C8—C7—C6	119.96 (12)
N1—N2—C12	109.95 (9)	C8—C7—H7	120.0
N1—N2—C2	107.42 (9)	C6—C7—H7	120.0
C12—N2—C2	113.88 (10)	C7—C8—C9	119.92 (13)
O2—C1—O1	119.42 (11)	C7—C8—H8	120.0
O2—C1—N1	124.07 (12)	C9—C8—H8	120.0
O1—C1—N1	116.50 (10)	C8—C9—C4	120.70 (12)
N2—C2—C3	108.57 (10)	C8—C9—H9	119.7
N2—C2—C4	108.92 (10)	C4—C9—H9	119.7
C3—C2—C4	115.94 (11)	O3—C10—N1	122.96 (12)
N2—C2—H2	107.7	O3—C10—C11	124.27 (12)
C3—C2—H2	107.7	N1—C10—C11	112.77 (11)
C4—C2—H2	107.7	C10—C11—C1	109.77 (9)
O1—C3—C2	111.58 (10)	C10—C11—H11A	109.7
O1—C3—H3A	109.3	C1—C11—H11A	109.7
C2—C3—H3A	109.3	C10—C11—H11B	109.7
O1—C3—H3B	109.3	C1—C11—H11B	109.7
C2—C3—H3B	109.3	H11A—C11—H11B	108.2
H3A—C3—H3B	108.0	N2—C12—H12A	109.5
C9—C4—C5	118.75 (12)	N2—C12—H12B	109.5
C9—C4—C2	123.20 (12)	H12A—C12—H12B	109.5
C5—C4—C2	118.04 (12)	N2—C12—H12C	109.5
C6—C5—C4	120.68 (12)	H12A—C12—H12C	109.5
C6—C5—H5	119.7	H12B—C12—H12C	109.5
C4—C5—H5	119.7		
C1—N1—N2—C12	-74.26 (13)	C3—C2—C4—C9	1.96 (17)
C10—N1—N2—C12	108.57 (12)	N2—C2—C4—C5	59.96 (14)
C1—N1—N2—C2	50.16 (13)	C3—C2—C4—C5	-177.28 (11)
C10—N1—N2—C2	-127.01 (11)	C9—C4—C5—C6	0.36 (19)
C3—O1—C1—O2	-176.89 (11)	C2—C4—C5—C6	179.64 (11)
C3—O1—C1—N1	3.76 (17)	C4—C5—C6—C7	-0.77 (19)
C10—N1—C1—O2	-22.68 (18)	C5—C6—C7—C8	0.57 (19)
N2—N1—C1—O2	160.29 (11)	C6—C7—C8—C9	0.02 (19)
C10—N1—C1—O1	156.64 (11)	C7—C8—C9—C4	-0.42 (19)
N2—N1—C1—O1	-20.39 (16)	C5—C4—C9—C8	0.23 (19)
N1—N2—C2—C3	-61.59 (12)	C2—C4—C9—C8	-179.00 (11)
C12—N2—C2—C3	60.42 (12)	C1—N1—C10—O3	-8.18 (18)
N1—N2—C2—C4	65.49 (12)	N2—N1—C10—O3	168.94 (11)
C12—N2—C2—C4	-172.50 (10)	C1—N1—C10—C11	172.73 (11)
C1—O1—C3—C2	-18.86 (16)	N2—N1—C10—C11	-10.15 (15)
N2—C2—C3—O1	47.44 (13)	O3—C10—C11—C1	14.29 (16)
C4—C2—C3—O1	-75.51 (14)	N1—C10—C11—C1	-166.64 (8)

N2—C2—C4—C9 -120.81 (13)

Hydrogen-bond geometry (Å, °)

<i>D—H...A</i>	<i>D—H</i>	<i>H...A</i>	<i>D...A</i>	<i>D—H...A</i>
C2—H2...O3 ⁱ	1.00	2.60	3.5142 (15)	153
C8—H8...O2 ⁱⁱ	0.95	2.60	3.2689 (16)	128
C11—H11b...O2 ⁱⁱⁱ	0.99	2.54	3.3675 (16)	141
C12—H12a...O2 ⁱⁱⁱ	0.98	2.57	3.5448 (16)	173

Symmetry codes: (i) $-x+1/2, -y, z-1/2$; (ii) $x+1/2, -y+1/2, -z+2$; (iii) $-x, y-1/2, -z+3/2$.