

29 NFC / PVB composite has three times more elongation at break, 13% more specific energy
30 absorbed with a half-tensile stress compared to an interlayer of PVB.

31 **1. Introduction**

32 In recent decades, petroleum-based plastics have emerged and diversified to reach a considerable
33 space in our daily life due to their numerous advantages, which allow to adapt to the desired
34 applications. According to a study by the University of Santa Barbara in California in 2017, global
35 production of plastics reached 9.1 billion tonnes, more than half of this volume, 5.4 billion tonnes,
36 has ended in the environment (Simon & Schulte, 2017). However, in addition to the sustainability
37 problems arising from the depletion of fossil fuels, the environmental impacts due to the wide use
38 of these non-renewable sustainable materials are seriously increasing, putting terrestrial and
39 aquatic ecosystems at risk (Bagheri, Radi, & Amiri, 2019; Floyd, 2016; Kumar Singla, Maiti, &
40 Ghosh, 2016; Simon & Schulte, 2017). Therefore, a rational use of biodegradable polymers
41 derived from renewable resources combined with an improvement in the quantity and quality of
42 recycling, are the key points of sustainable development of plastics materials (Floyd, 2016).
43 Nevertheless, the research on bio-based plastics produced from cellulose derivatives is challenging
44 because of the complexity in recycling strategies and far from providing a long-term sustainable
45 solution (Bagheri et al., 2019; Kumar Singla et al., 2016).

46 The 4-acetamido-2, 2, 6, 6 tetramethylpiperidine-1-oxyl (TEMPO) oxidized cellulosic nanofibrils,
47 NFC, are promising nanomaterials, widely used as attractive biopolymers for the production of
48 biobased films. The obtained films received reasonable attention because of their large specific
49 surfaces, rigidity, transparent nature and biodegradability (Bideau, Cherpozat, Loranger, &
50 Daneault, 2016; García, Gandini, Labidi, Belgacem, & Bras, 2016; Islam & Rahman, 2019; Niu,
51 Gao, & Wu, 2014; Rol, Belgacem, Gandini, & Bras, 2019; Syverud, Xhanari, Chinga-Carrasco,
52 Yu, & Stenius, 2011; Xhanari, Syverud, Chinga-Carrasco, Paso, & Stenius, 2011). However, the
53 non-flexibility and hydrophilic nature limits their potential applications in many fields and affects
54 their compatibility in most non-polar polymer matrices. Therefore, modifications of the surface
55 chemistry of NFCs were used in order to improve the performance of the films as well as their
56 compatibility with the polymer matrix (Islam & Rahman, 2019; Rol et al., 2019; Syverud et al.,
57 2011; Xhanari et al., 2011). The major challenge of modifying NFC is to increase the

58 hydrophobicity and flexibility of the films while keeping satisfactory mechanical and optical
59 properties (Tong, Chen, Tian, & He, 2020). The originality of our work is to synthesize flexible,
60 transparent and less hydrophilic nanofibrillated cellulose (NFC) films, but in a purely aqueous
61 medium.

62 In our study, NFC is chemically modified by radical polymerization of glycidyl methacrylate
63 (GMA) in aqueous medium. The monomer GMA, which is inexpensive and highly reactive
64 (Kocak, Solmaz, Tuncer, & Bütün, 2019), is often used to improve the mechanical properties and
65 hydrophobicity of films including cellulose-based films. It is also used as a compatibilizer and
66 coupling agent in the manufacture of composites (Abbasi et al., 2018; Cherifi, Boukoussa, Zaoui,
67 Belbachir, & Meghabar, 2018; Faria et al., 2019; Khan et al., 2018; Kocak et al., 2019; Reis et al.,
68 2009). Other studies have shown that adding plasticizers can improve the flexibility of films such
69 as, starch–chitosan films (Liu, Adhikari, Guo, & Adhikari, 2013), Na-alginate films (Bagheri et
70 al., 2019), silk fibroin films (Li et al., 2018), corn starch films (Šoltýs et al., 2019) and cellulose
71 acetate oleate films (Tedeschi et al., 2018). Glycerol is generally recognized as one of the most
72 suitable plasticizers (Bagheri et al., 2019; Li et al., 2018; Liu et al., 2013; Šoltýs et al., 2019).

73 The objective of the current research is to synthesize transparent and biodegradable films, which
74 would have satisfactory mechanical properties like resistance and elongation at break, and an
75 improvement of hydrophobic character compared to pure NFC film. Therefore, a simultaneous
76 modification of NFC oxidized TEMPO by GMA and glycerol was performed. According to the
77 literature, the amounts of monomer and plasticizer can significantly influence the characteristics
78 of films (Abbasi et al., 2018; Bagheri et al., 2019; Pracella, Haque, Paci, & Alvarez, 2016).
79 Therefore, a statistical response surface model was developed to study changes in the target
80 properties of the final product. This method has been widely used to optimize the process
81 conditions using a mathematical algorithm based on experimental results generated from
82 experiments designed by statistical analysis software. Response-surface design overcomes the
83 drawbacks of traditional optimization methods, which monitor the effect of a single parameter on
84 the process at a time regardless of the interactive effects among the parameters examined. In
85 addition, using statistical analysis requires less time and chemicals compared to the optimization
86 of each parameter separately.

87 **2. Experimentation**

88 **2.1. Materials**

89 The pulp used in this work is a bleached kraft pulp of resinous wood from the paper company
90 Domtar. The catalyst 4-acetamido-TEMPO is purchased from Sigma-Aldrich (Canada) and the
91 sodium bromide from Fisher Scientific (Canada). The sodium hypochlorite used is 6 %
92 concentrated as found in supermarkets. Poly (vinyl butyral-co-vinyl, alcohol-co-vinyl acetate)
93 powder (Co PVB,>90 %, Sigma), glycidyl methacrylate (GMA, 97%, Sigma) and glycerol (99%,
94 Sigma) are used as received. The other chemicals and solvents are laboratory grade and supplied
95 by Sigma-Aldrich (USA) without further purification.

96 **2.2. Preparation of NFC by TEMPO-mediated oxidation**

97 The preparation of TEMPO oxidized NFCs is carried out based on a previously published protocol
98 from our research group (Loranger, Paquin, Daneault, & Chabot, 2011; Paquin, Loranger,
99 Hannaux, Chabot, & Daneault, 2013; Rattaz, Mishra, Chabot, & Daneault, 2011). The basic
100 principle consists in the oxidation of cellulose fibers by adding NaClO to aqueous cellulose
101 suspensions in the presence of catalytic amounts of 2, 2, 6, 6-tetramethyl-1-piperidinyloxy
102 (TEMPO) and NaBr at pH 10–11 at room temperature. Ultrasounds are optional and may be used
103 to further reduce the amounts of reactive used without any effect on the final properties of the
104 nanofibrillated cellulose. The NFCs obtained had a carboxyl content of 1700 mmol / kg. The
105 obtained NFC gel is dispersed in a homogenizer as optimized by Loranger et al. (Loranger, Piché,
106 & Daneault, 2012).

107 **2.3. Preparation of modified NFC film**

108 The TEMPO oxidized NFC gel (3.3%) is diluted 50% up to 1.65 % with distilled water and then
109 centrifuged for 15 minutes at 13,000 rpm. A transparent aqueous dispersion is obtained (0.1%) by
110 recovering the supernatant and removing the micro suspensions. Then, the aqueous NFC solution
111 is concentrated up to 0.9% using a rotavapor.

112 The modification consists in synthesizing poly (glycidyl methacrylate)-co-NFC (PGMA-co-NFC)
113 followed by the addition of a plasticizer. An aqueous solution of NFC (200 ml) is stirred
114 magnetically with ammonium persulfate (APS) (0.05 ± 0.0001 g) to initiate the in situ

115 polymerization of GMA. Then different volumes of glycidyl methacrylate (GMA) are added at
116 40°C. After stirring for 48 hours, different amounts of glycerol as a plasticizer are added at 70°C
117 for 5 hours. Based on a Response-surface design as explained earlier, Table 1 shows the
118 composition of the mixtures prepared in this study. Finally, the solutions are poured into aluminum
119 cups and dried in an oven at 30°C for 2 days.

120 **2.4. Preparation of composite NFC/PVB**

121 The PVB / NFC composite is prepared by a coating method developed by our group (Maury,
122 Loranger, & Daneault, 2016). First, the PVB is mixed with anhydrous ethanol (1% by weight) and
123 stirred for 30 minutes until being completely dissolved. Then, the solution is poured into an
124 aluminum cup already containing a previously dried NFC film. Finally, the cup is dried in an oven
125 at 30°C for 2 days.

126 **2.5. Characterization**

127 **Fourier-transform infrared spectroscopy (FTIR).** The functional groups of native and modified
128 NFC are analyzed by infrared spectroscopy FTIR in the range of 4500–600 cm⁻¹ from 16 scans
129 with a resolution of 4 cm⁻¹. FT-IR Spectra are obtained using a Nicolet™ IS10 FT-IR™
130 spectrometer (ThermoScientific™, USA) at room temperature.

131 **Mechanical properties.** The mechanical properties of the films are measured by universal testing
132 machine Instron 4201™ equipped with a 500 N load cell. Rectangular shaped samples (30 mm
133 length, 15 mm width) are stretched at a rate of 10 mm min⁻¹. All the stress-strain curves are
134 recorded in a controlled room at 25°C and a relative humidity (RH) of 50%. Three measurements
135 are carried out for each sample. As the mean and median values are very close, only the mean is
136 reported. The Young's modulus, the stress, the elongation at break and the absorbed energy values
137 are extracted from the stress-strain curves. A specific energy absorbed is calculated by the ratio
138 between the energy absorbed at automatic break and the cross area of the sample. The thicknesses
139 are measured with a Lhomargy™ micrometre (± 0.01 mm).

140 **Transparency.** Light transmission of films is measured by a Tint Meter Inspector Model 200™.

141 **Scanning electron microscopy (SEM).** Scanning electron micrographs of the cross-section and
142 surfaces of the samples are obtained by scanning electron microscopy (SEM) with a JEOL JSM
143 T300™ microscope. An acceleration voltage of 5 KV and magnification of 2500× (cross section)

144 and 1000× (surface) are used to observe the morphology of samples. Before analysis, all samples
145 are deposited on a steel plate and coated with a mix of gold and platinum.

146 **Contact angle.** In order to characterize the hydrophobicity of the film surface, a FTA4000™
147 contact angle measuring system (First Ten Angstroms) is used. Water contact angles are measured
148 with the drop method at room temperature at five different locations on each surface. One drop
149 (0.8 µl ± 0.07) of purified water (milli Q) is deposited on the surfaces and 300 images are captured
150 within 90 s.

151 **Thermogravimetric Analysis (TGA).** Analysis of the thermal stability of the samples is carried
152 out in a Perkin-Elmer™ Thermogravimetric analyzer TGA 8000™ (Pyris Series). The samples
153 (Table1) are heated in platinum pans from 50 to 575°C, under a nitrogen atmosphere, at a heating
154 rate of 5°C / min. Then the samples are heated from 575 to 900°C of 10°C / min under a nitrogen
155 flow of 20 ml / min.

156 **Statistical analysis.** To assess the effect of the amounts of monomer (GMA) and plasticizer
157 (glycerol) on the properties of the film, a central composite design-response surface model (CCD)
158 is used. Preliminary results in the laboratory showed that for an amount of 200 ml of NFC, if the
159 volumes of GMA and glycerol are increased to more than 4.5 and 1 ml respectively, no film can
160 be obtained. Therefore, a volume of GMA in the range of 0 to 4.5 ml and a volume of glycerol in
161 the range of 0 to 1 ml were introduced to the JMP® Start Statistics 2007 (SAS Institute Inc.™,
162 Cary, NC, USA) to design the experiments (Table1). Consequently, 11 different random
163 experimental combinations of variables including 3 central points were proposed and their
164 corresponding responses were measured.

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Table 1. *Response-Surface design array of experiments*

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Test number	Sample name	Volume of GMA (ml)	Volume of Plasticizer (ml)
1	NFC 1	2.25	1
2	NFC 2	2.25	0.5
3	NFC 3	4.5	0
4	NFC 4	4.5	0.5
5	NFC 5	4.5	1
6	NFC 6	2.25	0.5
7	NFC 7	2.25	0
8	NFC 8	0	1
9	NFC 9	2.25	0.5
10	NFC 10	0	0.5
11	NFC 11	0	0

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3. Results

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3.1. Fourier-transform infrared spectroscopy (FTIR)

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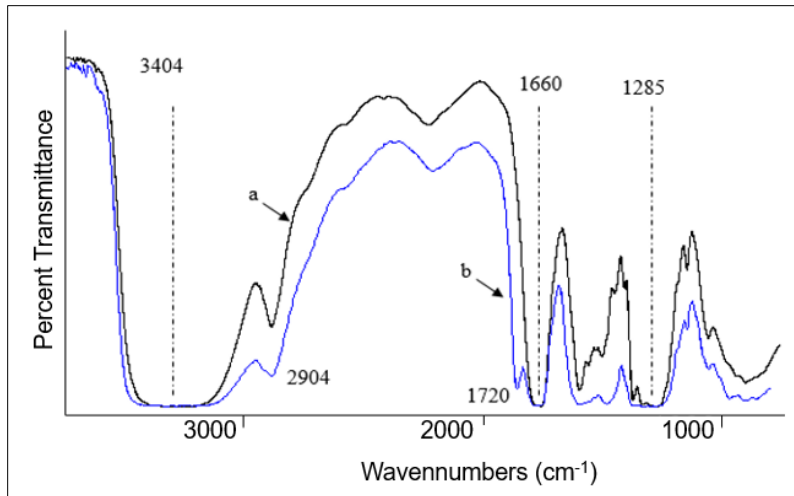
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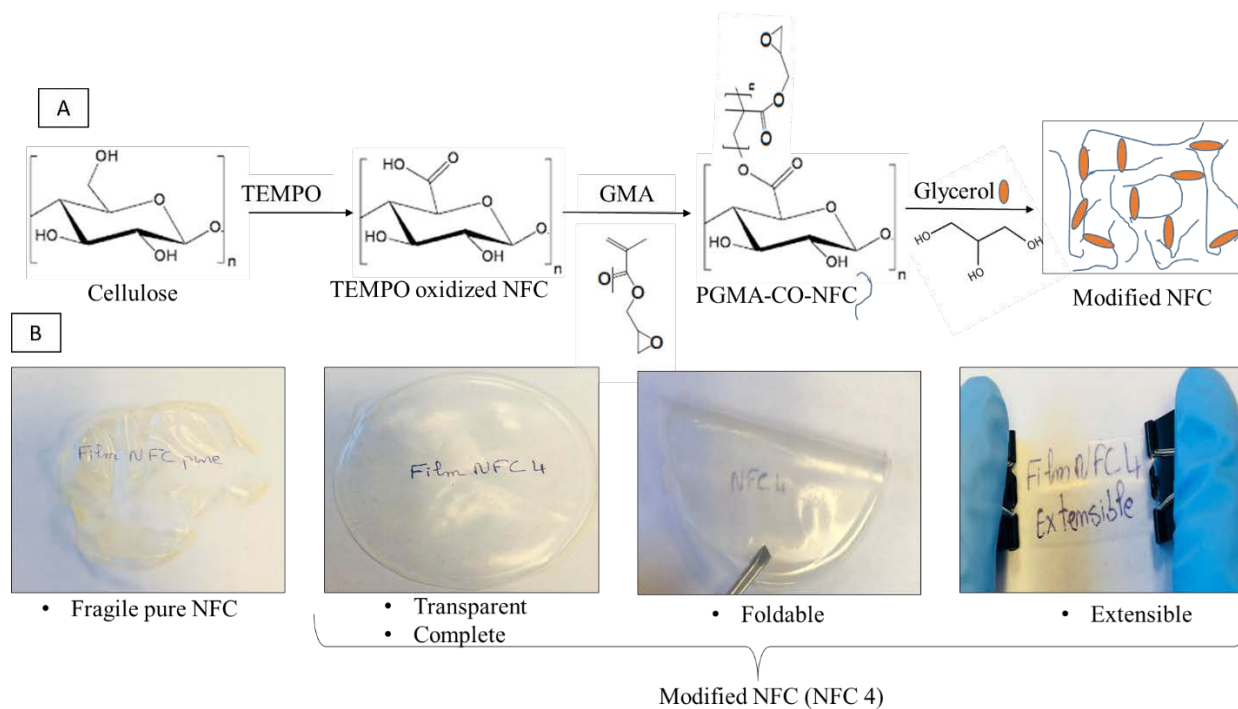
The FTIR spectra (600-4000 cm^{-1}) of pure NFC 11, modified NFC 3 films are shown in Figure 1. The analysis of the spectra shows that the chemical structure of the NFC does not change during the modification process, as characteristic bands typical of the original form were observed. The band located at 3404 cm^{-1} is attributed to OH stretching vibrations, the band at 2904 cm^{-1} to CH_2 stretching vibrations and the band at 1660 cm^{-1} to carboxylic functional groups of native NFC film. The band at 1720 cm^{-1} and the weak broad band at 1285 cm^{-1} are attributed to the C=O and C-O stretching of GMA, respectively (Abbasi et al., 2018; Faria et al., 2019). The schematic of modifying the cellulose to obtain a modified NFC film is shown in Figure 2A while photographs of NFC 4 films are found in Figure 2B for comparison purposes.

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Figure 1. ATR-FTIR spectra of pure NFC 11 (a), modified NFC 3 (b) films



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Figure 2.A) Schematic of cellulose modifications, B) Photographs of the pure NFC and modified NFC

(Example: NFC 4) films placed on a white background paper to demonstrate their properties

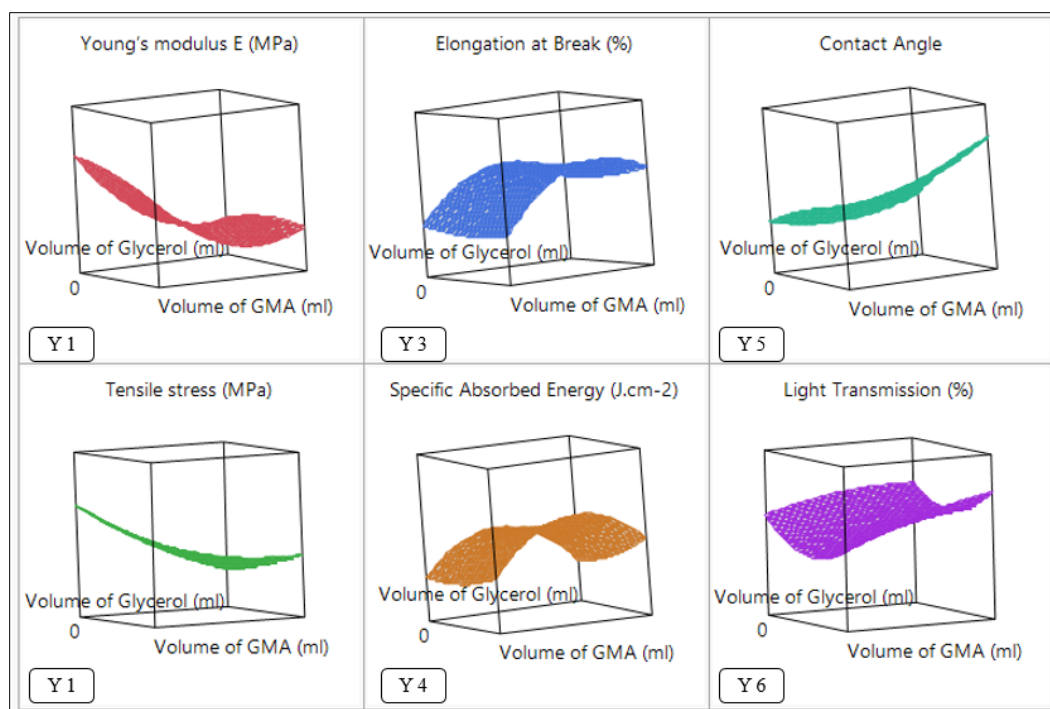
3.2. Effect of monomer and plasticizer quantities

A central composite design-response surface model is applied for modeling and optimizing the influence of GMA quantity (X1) and glycerol quantity (X2) on the properties of the films: Young's

205 modulus E (Y1), tensile stress (Y2), elongation at break ϵ (Y3), specific absorbed Energy (Y4),
 206 contact angle (Y5) and light transmission (Y6). Results from experiments are illustrated in Table
 207 2. Analysis of variance is used to identify the significance of factors with a P value < 0.05. Based
 208 on the regression coefficient (R^2) and adjusted regression coefficient ($adjR^2$), polynomial models
 209 for each response are determined. Thus, the execution of the model provides Iso-response profilers,
 210 which help determine the variation of responses for each combination of factors (Figure 3).

211 Table 2. *Experimental design and results of different dependent variables*

Test number	X1 (ml)	X2 (ml)	Y1 (MPa)	Y2 (MPa)	Y3 (%)	Y4 (J/cm ²)	Y5 (°)	Y6 (%)
1	2.25	1.0	32.0	1.2	5.1	0.1	83.0	93.0
2	2.25	0.50	67.9	4.6	13.7	1.0	46.3	73.0
3	4.50	0.00	1016.0	7.5	1.3	0.1	51.8	70.0
4	4.50	0.50	156.4	8.1	13.2	2.5	76.6	90.0
5	4.50	1.00	22.0	3.3	16.4	0.9	81.7	93.0
6	2.25	0.50	79.3	5.1	16.1	1.8	50.1	75.0
7	2.25	0.00	2324.8	11.6	1.6	0.2	52.3	34.0
8	0.00	1.00	23.7	0.9	12.3	0.3	37.8	93.0
9	2.25	0.50	66.4	4.8	17.9	1.6	43.2	74.0
10	0.00	0.50	47.4	2.6	12.5	0.7	34.7	85.0
11	0.00	0.00	2168.7	16.5	2.2	0.3	38.3	93.0



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214 Figure 3. Iso-response profilers predicting effects of GMA (horizontal axis) and glycerol (vertical axis)
 215 volumes on Young's modulus E (Y1), tensile stress (Y2), Elongation at break ϵ (Y3), Specific Absorbed
 216 Energy (Y4), Contact Angle (Y5) and Light Transmission (Y6)

217 3.2.1. Mechanical properties

218 Tensile stress-strain curves of pure NFC 11, modified NFC 3, 4, 5 and 6, Films are shown in Figure
 219 4. Pure NFC film (NFC 11) is a fragile material, which fractures directly when it is subjected to
 220 mechanical stress. Previous research has shown that the rupture of brittle materials is due to the
 221 presence of initial defects (Hild, 1992). This is confirmed by the shape of the stress-strain curve.
 222 The detection of breaking point for this type of material is very difficult so, we had to stop the
 223 testing manually.

224 The modification of NFC increases the elongation at break to around 17.9 % (Film 9) to be
 225 compared with an elongation at break of 2.2% for pure NFC film. Contrariwise, the tensile strain
 226 values of the purest natural polymer-based film are around 15% (Huang, Zhong, Zhang & Cai,
 227 2017; Tong, Chen, Tian, & He, 2020). NFC films modified by GMA and glycerol simultaneously
 228 exhibit ductile behavior, which undergoes great deformation before rupture. The photographs
 229 shown in Figure 2 confirm these results. The modified film (Example film 4) can be rolled, folded

230 and stretched without cracking. These advantages can lead to a much-improved NFC-based film
231 for applications that require transparent and flexible films.

232 The results from the statistical analysis show that the change in the volume of glycerol has a
233 significant effect on the mechanical properties of the film while the amount of GMA has no. As
234 shown in Figure 3, when the glycerol volume increases, the Young's modulus E and tensile stress
235 decrease, however, the elongation at break ϵ and absorbed energy increase. This is explained by
236 the plasticizing effect of the glycerol (Li et al., 2018; Liu et al., 2013). Therefore, a minimum level
237 of glycerol volume is determined to achieve maximum rigidity, whereas a maximum level is
238 determined as optimum to achieve maximum flexibility. These results are confirmed by tensile
239 stress-strain curves presented on Figure 4. Indeed, for a same volume of GMA (4.5 ml), when the
240 quantity of glycerol increases from 0.25, 0.5 to 1 ml, we can clearly see the change in the shape of
241 the curves towards a more ductile behavior of NFC 3, 4, 5 films respectively. The model equations
242 are:

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$$Y1 \text{ (Young's modulus E (MPa))} = 83.492 - 905.308X_2 + 847.696X_2^2$$

$$(R^2 = 87\%, R^2_{adj} = 84\%)$$

$$Y2 \text{ (Tensile stress (MPa))} = 6,011 - 5,057X_2$$

$$(R^2 = 69\%, R^2_{adj} = 66\%)$$

$$Y3 \text{ (Elongation at break } \epsilon \text{ (%))} = 14,683 + 4,778X_2 - 8,175X_2^2$$

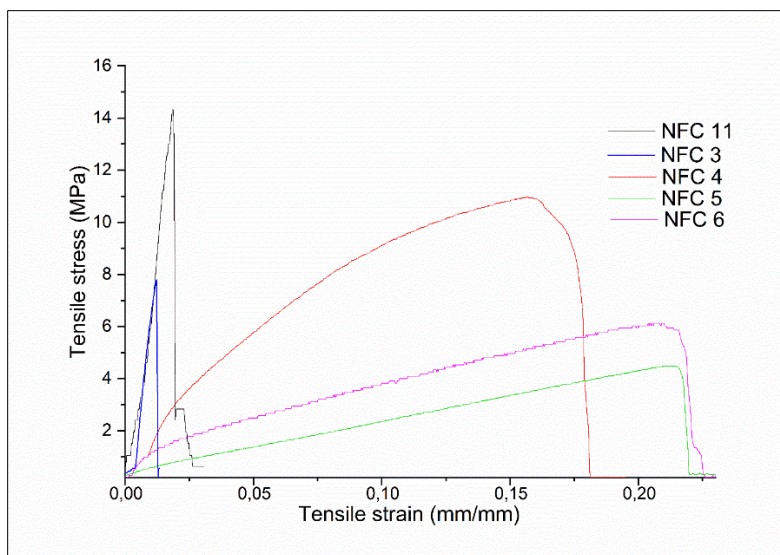
$$(R^2 = 79\%, R^2_{adj} = 74\%)$$

$$Y4 \text{ (Specific Absorbed Energy (J. cm}^{-2}\text{))} = 1,524 + 0,127X_2 - 1,206X_2^2$$

$$(R^2 = 64\%, R^2_{adj} = 55\%)$$

244 Even though the statistical analysis shows no significant effect on addition of GMA, visual tests
245 support that this step is very important to obtain a complete NFC films, maintaining similar
246 mechanical properties of pure NFC films, which are fragile and crack down easily while handling
247 (Figure 2). This is particularly the case when comparing NFC 11 to NFC modified with GMA only
248 NFC 3. In addition, tensile stress-strain curves of NFC 6 and 4 films (Figure 4) show that, for the
249 same amount of glycerol (0.5 ml), when we increase the quantity of the monomer (GMA) from
250 2.25 to 4.5 ml respectively, tensile stress increases, while tensile strain decreases.

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253 *Figure 4. Tensile stress-strain curves of pure NFC 11, modified NFC 3, 4, 5 and 6, Films*

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3.2.2. Contact angle measurements

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It is clear from the Table 1 and Iso-response profilers (Figure 3) that both glycerol and GMA volumes have significant effects on contact angle which is increased from 38.3° (pure film 11) to 83° (modified film 1). To support our results, Figure 5 shows the droplet profiles and the evaluation of the contact angle on different types of films: pure NFC (NFC 11), NFC modified with only GMA (Example NFC 3), NFC modified with only glycerol (Example NFC 8) and NFC modified with both GMA and glycerol respectively. Previous studies have shown that the grafting of epoxy chains onto allyl cellulose increases the hydrophobicity of the films (Tong, Chen, Tian, & He, 2020). These observations are confirmed by our study, for films 3 (51.8°) and films 7 (52.3°). On the other hand, the addition of only glycerol has an opposite effect and the contact angle of the films always remains small, e.g. film 8 (37.8°) and film 10 (34.7°). These results are also confirmed by previous studies (Coupland, Shaw, Monahan, O'Riordan, & O'Sullivan). However, the contact angle results drawn from this study show that the polymerization by GMA followed by the addition of plasticizer presents an excellent compromise for increasing the hydrophobic character of the films as shown by film 1 (83°), film 4 (76.6°) and film 5 (81.7°) (Figure 5). The outstanding flexible and transparent cellulose films show contact angle values higher than that of nanocellulose film (47°) (Fukuzumi, Saito, Iwata, Kumamoto, & Isogai, 2009) and modified nanocellulose film, of which contact angle does not exceed 79° (Tong, Chen, Tian, & He, 2020). Ultimately, the model equation for Contact Angle Y5 is given by:

$$Y5 (\text{Contact Angle}) = 54,173 + 16,555X1 + 10,029X2 + 7,595X1X2$$

$$(R^2 = 77\%, R^2_{\text{adj}} = 68\%)$$

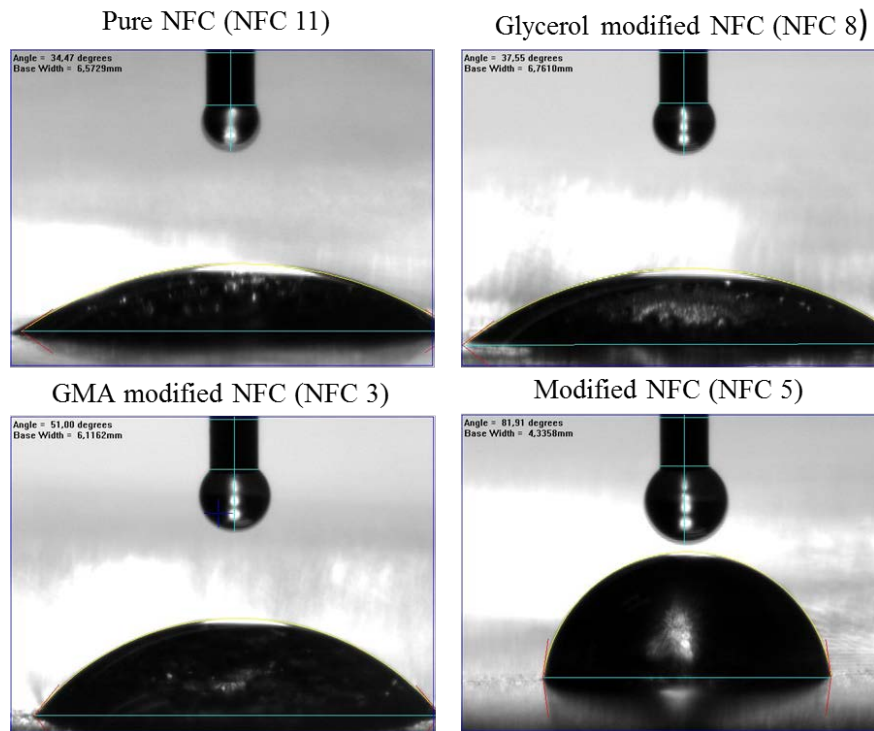


Figure 5. Water contact angles of pure NFC 11, Glycerol modified NFC (NFC 8), GMA modified NFC (NFC 3) and modified NFC (NFC 5)

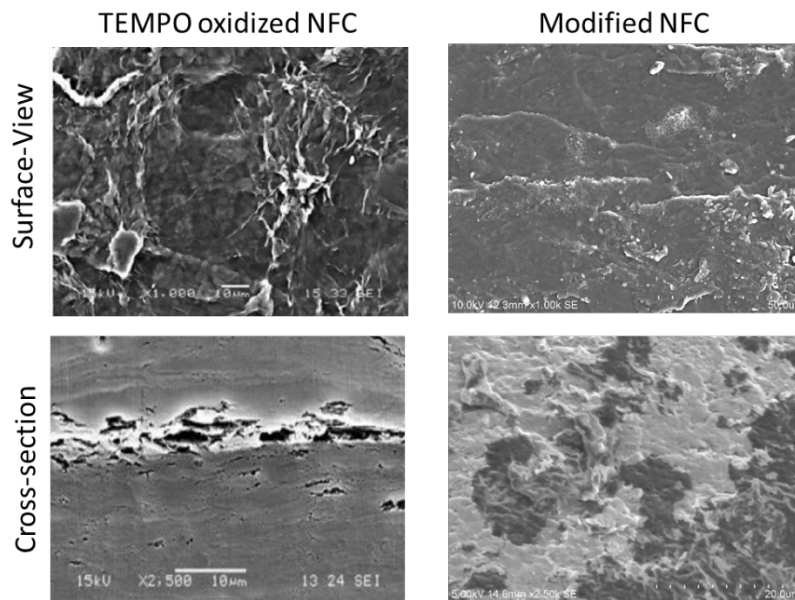
273 3.2.3. Optical properties

274 The optical transmittance values of our films are very promising. Hence, the majority of the films
 275 obtained (Table 2) show high light transmittance values (optimal 93%) (Figure 2). They are
 276 comparable to those of commercial cellophane (85%), Nanopaper (90%) (Tong, Chen, Tian, &
 277 He, 2020) and ginger nanofiber (82%) (Abraal et al., 2020) films. However, the film NFC 3 (70%)
 278 and film NFC 7 (34%) have moderately to low light transmission values. Statistical analyses with
 279 p value < 0.05 do not indicate which factor has a significant effect on the optical transmittance of
 280 films. Nevertheless, based on Iso-response profilers (Figure 3) we can notice that the presence of
 281 glycerol increases the transparency of the films. In fact, studies have shown that the glycerol
 282 plasticizer improves the homogeneous dispersion in the polymer matrix by interfacial interactions
 283 (Figure 2 (A)) (Li et al., 2018).

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285 3.3. Morphological characterization

286 Figure 6 shows the comparison of SEM micrographs of the cross section and surfaces of pure NFC
287 and modified NFC with both GMA and glycerol. The surface morphology of the modified NFC is
288 more homogeneous and clearly different, supporting successful modification of TEMPO oxidized
289 NFC. As it has been demonstrated by Liu et al. (Liu, Adhikari, Guo, & Adhikari, 2013), the
290 formation of plasticized and rubbery films is supported by the formation of a homogeneous and
291 smooth surface. The cross-section micrographs show that the pure NFC appears as a layer with a
292 few pores and cracks whereas the modified NFC seems to be strongly entangled and less porous
293 showing that probably more interactions between modified NFC. In conclusion, SEM images show
294 a strong entanglement of the multiple layers of NFC.



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296 *Figure 6. SEM micrographs of the cross section and surfaces of pure NFC and modified NFC*

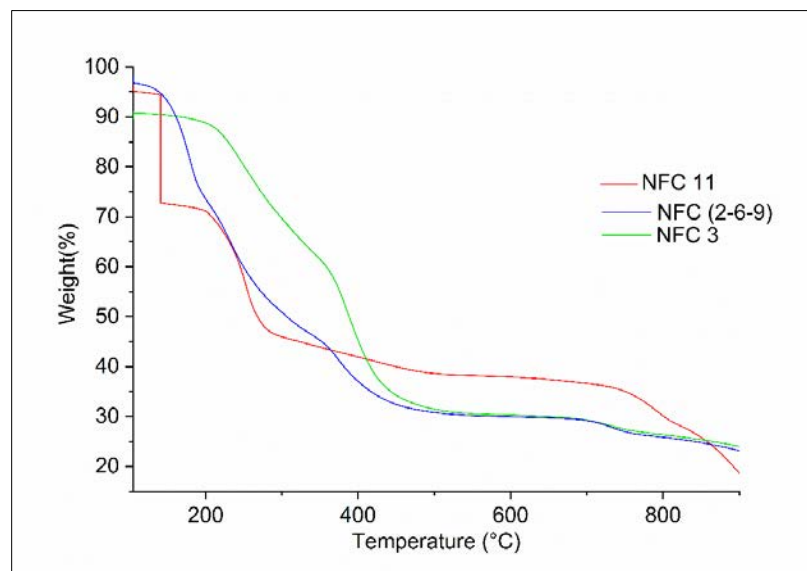
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298 3.4. Thermogravimetric Analysis

299 Thermal behaviours of NFC based-films are studied using Thermogravimetric analysis (TG) under
300 nitrogen atmosphere from 50-900°C (Figure 7). In all the curves, the first part of the analysis from

301 50 to 105 ° C was removed because it corresponds to the evaporation of solvent (water). Thus,
302 depending on the water content, the curves may not start at 100% in weight. Initially, the modified
303 films (NFC 2, 3) show a thermal stability from 50 up to 200°C more than the native film (NFC
304 11). The first decrease in the mass of pure NFC 11 film shows up to 140° C, which could be
305 attributed to the release of moisture and a weakly bonded water, while the modified films show a
306 lower weight loss due to its more hydrophobic characteristics (Ashori, Babae, Jonoobi, &
307 Hamzeh, 2014). In addition, the modification of NFC by grafting only PGMA (NFC 3) was found
308 to introduce thermal degradation behavior in several stages. The first and second decomposition
309 steps (240.07°C and 384.03°C) can be attributed to the decomposition of the glycidyl and carboxyl
310 groups of the GMA respectively (Abbasi et al., 2018; Cherifi et al., 2018). For the modification by
311 grafting of PGMA followed by the addition of plasticizer (NFC 2-6-9), we note the appearance of
312 a new degradation step at 172.98°C, which can be attributed to the decomposition of glycerol. As
313 shown in Figure 7, the proposed modification of NFC has increased the thermal degradation
314 resistance of the films at higher temperature.

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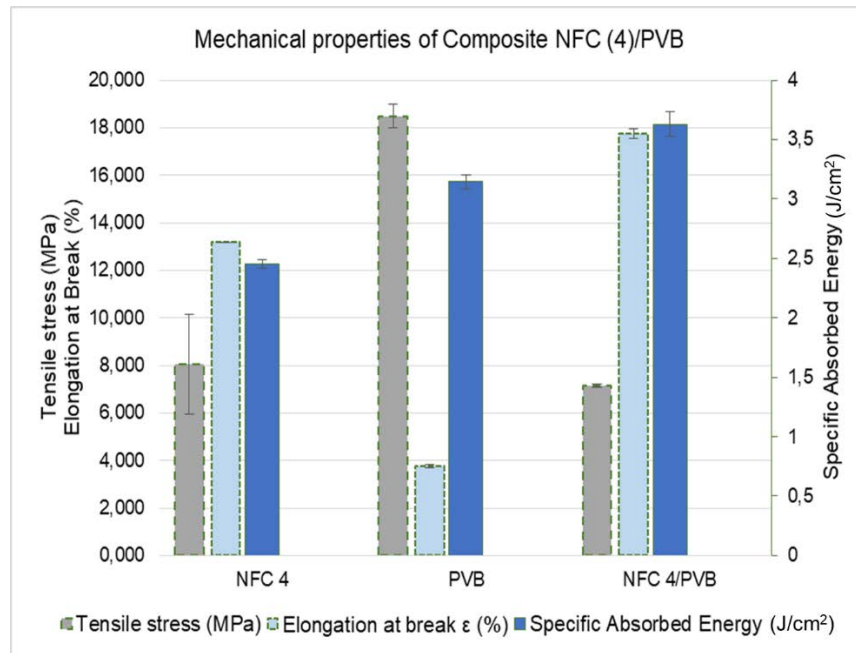
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317 *Figure 7. Thermogravimetric analysis of pure NFC 11, NFC (2-6-9) and NFC 3 films*

318 **3.5. Discussion: NFC/PVB composite**

319 Today, safety glass is laminated using a thermoplastic polymer, polyvinyl butyral (PVB) which is
320 petroleum-made, non-biodegradable and expensive. NFC 4 film was chosen to synthesize a NFC
321 / PVB composite as a candidate interlayer for safety glass and bulletproof glazing because of its

322 interesting properties such as flexibility, transparency and above all, the highest specific absorbed
323 energy (2.45 J/cm^2). The results presented in Figure 8 show that the composite exhibits attractive
324 mechanical characteristics compared to a PVB film alone. Despite a drastic reduction in tensile
325 stress, there is a strong increase in the elongation at break property, which is very important for
326 resistance in the case of impact (Petroudy, 2017). In addition, there was a 13% increase of energy
327 absorbed at break.
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330 *Figure 8. Mechanical properties of Composite NFC (4)/PVB*

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332 4. Conclusions

333 In the present study, different films were prepared by modifying the NFC oxidized TEMPO by
334 different amounts of GMA and glycerol. The synthesized films have interesting properties such as
335 transparency, flexibility, resistance and a less hydrophilic nature. The response surface
336 methodology (RSM) was successfully applied to model and optimize the performance of modified
337 NFC-based films. The optimal conditions for modification depend on the objectives. For example,
338 to get the best flexibility independently of the other parameters, it is necessary to work with 2.25
339 ml of GMA and 0.5 ml of glycerol (NFC 9 Film). If we are looking for a good compromise between

340 flexibility, hydrophobicity, and transparency, NFC 5 is a good candidate with volumes of GMA
341 and glycerol of 4.5 and 1 ml respectively. The enhancement of the properties can make these films
342 a potential alternative to petroleum-based films depending on the target applications such as
343 electronic substrates, cars, packaging, sports equipment, etc. It could also position this solution as
344 candidates for interlayers in security glazing.

345 **5. Acknowledgments**

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349 **6. References**

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