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Citation for published version:

Meier, T, Laniel, D, Pena-Alvarez, M, Trybel, F, Khandarkhaeva, S, Krupp, A, Jacobs, J, Dubrovinskaia, N & Dubrovinsky, L 2020, 'Nuclear Spin Crossover in Dense Molecular Hydrogen', *Nature Communications*, vol. 11, no. 1, 6334, pp. 1-7. https://doi.org/10.1038/s41467-020-19927-y

Digital Object Identifier (DOI):

10.1038/s41467-020-19927-y

Link:

Link to publication record in Edinburgh Research Explorer

Document Version: Peer reviewed version

Published In: Nature Communications

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1	Nuclear Spin Coupling Crossover in Dense Molecular Hydrogen
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13 Abstract

One of the most striking properties of molecular hydrogen is the coupling between molecular 14 rotational properties and nuclear spin orientations, giving rise to the spin isomers ortho- and 15 para-hydrogen. At high pressure, as intermolecular interactions increase significantly, the free 16 17 rotation of H₂ molecules is increasingly hindered, and consequently a modification of the coupling between molecular rotational properties and the nuclear spin system can be 18 anticipated. To date, high-pressure experimental methods have not been able to observe nuclear 19 spin states at pressures approaching 100 GPa^{1,2} and consequently the effect of high pressure on 20 21 the nuclear spin statistics could not be directly measured. Here, we present in-situ high-pressure 22 nuclear magnetic resonance data on molecular hydrogen in its hexagonal phase I up to 123 GPa 23 at room temperature. While our measurements confirm the presence of ortho-hydrogen at low 24 pressures, above 70 GPa, we observe a crossover in the nuclear spin statistics from a spin-1 25 quadrupolar to a spin-1/2 dipolar system, evidencing the loss of spin isomer distinction. These 26 observations represent a unique case of a nuclear spin crossover phenomenon in quantum solids.

27

28 Introduction

29 Changes in electronic spin statistics under changing thermodynamic conditions are an 30 established physical crossover phenomenon³. It has direct applications for spintronics⁴ and 31 enables the understanding of the stabilization of magnetospheres of rocky (Earth-like) planets⁵ 32 as well as gas- and ice-giants⁶. The degrees of freedom of the nuclei spins, however, are widely 33 approximated as fixed within the analysis of experiments, due to large shielding by core 34 electrons and the extremely short atomic distances necessary to induce such fundamental 35 changes.



Figure 1: **Overview of both spin-pairing and non-pairing regimes. a)** Schematic representation of the wave function overlap (red lines) of H₂ molecules. **b)** Schematic representation of the nuclear spin energy levels under the influence of an external magnetic field B₀ for the pairing (i.e. quadrupole interaction) and non-pairing (dipole-dipole interaction) regimes. **c)** Theoretical line shapes in the pairing and non-pairing regime. $\omega_0 = 2\pi f_0$ denotes the Larmor frequency of the hydrogen nuclei, ω_0 , η and C_q are the quadrupole frequency, the asymmetry parameter and the quadrupole coupling constant respectively.

36 Hydrogen, on the other hand, exhibits no core electrons and when bound contributes its 37 electron to the molecular bond. Furthermore, due to the low mass of the hydrogen nuclei, 38 quantum nuclear effects are considered to be significantly more pronounced compared to all 39 other elements. The combination of both effects results in a number of fascinating physical 40 phenomena in molecular H_2^{7-9} .

One property intrigued physicists in particular: the nature of the nuclear spin of the H₂ molecule and the resulting coexistence of the spin isomers ortho- (ortho-H₂) and para-hydrogen (para-H₂). Following Pauli's exclusion principle, in order for the total H₂ molecular wave function to be antisymmetric under exchange of atomic positions, demands for the rotational ground state J=0, that the corresponding total nuclear wave function is antisymmetric (singlet state of I=0, i.e. para-H₂). Analogously, for the J=1 rotational state, the total nuclear wave





function is required to be symmetric (triplet state of I=1, i.e. ortho-H₂). Therefore, the spin 47 48 allotropic isomerism of the H₂ molecule originates in the coupling of both rotational state and nuclear spin. It has been argued^{10,11} that at high pressure (P) ortho- and para-hydrogen spin 49 isomers remain stable up to the dissociative Wigner-Huntington transition at P > 400 GPa^{12,13}. 50 This can only be assumed for weak or moderate intermolecular interactions, i.e. when nearest 51 neighbour distances (r_n) are much shorter (≈ 0.7 Å at ambient conditions) than next-nearest 52 neighbour distances ($r_{nn} \approx 3 \text{ Å}$ at ambient conditions), allowing for sufficient intramolecular 53 wave function overlap (left side of Fig. 1a). 54

55 Under high enough densities, however, intermolecular interactions increase 56 significantly as r_{nn} decreases by ~70% within 100 GPa^{14,15}. At these pressures, r_{nn} approaches 57 r_n and collective nuclear quantum fluctuations increase rapidly¹⁶. For decades, theoretical^{8,17,18} and experimental⁹ studies indicated that under such extreme pressures odd values of J become unstable, rapidly decaying into even states, which leads to a potential indistinguishability of the hydrogen spin isomers on experimental timescales."

The only experimental technique in high-pressure research to directly study the H_2 61 nuclear spin states is nuclear magnetic resonance (NMR) spectroscopy, detecting the linear 62 63 response of the nuclear spin system upon radio frequency excitation in a magnetic field B_0 . The spin singlet state of para-hydrogen is NMR silent, whereas application of B_0 lifts the three-fold 64 degeneracy of the ortho states and allows for an excitation of nuclear spin transitions (Fig. 1b). 65 66 Nuclear spin pairing in ortho-hydrogen leads, furthermore, to a finite electric quadrupole 67 moment, eQ, interacting with the local charge distribution defined by the structural arrangement 68 of hydrogen molecules. Thus, quadrupolar coupling can be considered the dominant spin interaction, resulting in characteristic NMR line shapes¹⁹ (Fig. 1c). 69

Here, we present ¹H-NMR data of dense molecular hydrogen up to 123 GPa at room temperature and found a distinct crossover in the nuclear spin statistics of molecular hydrogen indicating a loss of ortho-para spin isomer distinctionDetails on experimental conditions, spectral simulations as well as data analysis are provided in the Methods Section.

74

75 **Results**

Two NMR-DACs equipped with diamond anvils of 250 μ m and 100 μ m culets were loaded with molecular H₂. At low pressure (below 60 GPa), intense ¹H resonances of roughly 500 kHz width were detected. With increasing *P*, the resonance signals broadened significantly approaching 750 kHz at 68 GPa (Fig. 2). Above 68 GPa, we observed a resonance narrowing accompanied by the emergence of two Pake doublets²⁰ with increasing splitting upon compression.

For the quadrupolar nature expected for ortho-H₂ (I=1), the electric field V(r), defined 82 by the local charge distribution based on the crystal structure of phase I, should influence the 83 shape of the observed resonance lines. Calculated line shapes for a *I*=1 spin system are shown 84 85 in Figure 2 at pressures of up to 68 GPa. The order of magnitude of the quadrupolar interaction energy was considered small relative to the nuclear Zeeman energy¹⁹ and consequently treated 86 87 as a first order perturbation (see Methods and Materials Section for computational details). Up to 68 GPa, the measured ¹H-NMR spectra are well described by calculated line shapes 88 89 broadened by first order quadrupole interaction. The line shape is mainly controlled by two parameters: (i) the quadrupole coupling constant C_q describing the coupling between eQ and 90

91 $V(\mathbf{r})$ as well as (ii) the electric field gradient asymmetry parameter η accounting for the 92 geometry of $V(\mathbf{r})$.



Figure 3: Extracted ¹H-NMR data of molecular H₂ at pressures up to 123 GPa at room temperature. a) Top panel: quadrupole coupling constant C_q determined from NMR data (squares). The rose line denotes theoretical ortho-para conversion rates from electric quadrupole interaction²⁸. Bottom panel: asymmetry parameter η in the spin pairing regime (P < 60 GPa). The dashed line represents the inferred n based on the hcp structure of Phase I from diffraction experiments. The shading of the squares at P > 60 GPa highlights the crossover to the non-pairing I=1/2 regime. b) Top panel: next nearest neighbour distances rnn. Green circles are based on DFT computations¹⁵. Blue circles are extracted values of rn and rnn from the NMR spectra in the non-pairing I=1/2 regime. Middle panel: nearest neighbour distances r_n (blue circles) and DFT calculations ¹⁵(red circles). The discontinuity at P > 125 GPa in the DFT calculations indicates a transition from the hcp based to a monoclinic structure. Bottom panel: comparison between the equations of state derived from ab-initio computations²⁴ (yellow line) and diffraction data^{14,21} along with the unit cell volumes (blue dots) derived from r_{nn} and r_n extracted from the *NMR* experiments. *c)* Extracted isotropic chemical shift values δ_{iso} after homo-nuclear Lee-Goldberg decoupling. Error bars are within the symbol size. The orange dotted line shows the room temperature Raman shift of the H_2 vibron²⁶. d) Selected Raman spectra of the H₂ vibron at increasing pressure. Spectra taken from [25] Error bars in a) and b) were taken from spectral simulation and comparison with experimental data. Error bars in c) are taken from Voigtian spectral line shape fitting to the experimental data. Error bars of the NMR data after Lee-Goldburg decoupling (blue points in c)) were within the symbol size.

Figure 3a (top panel) shows estimated values of C_q which increase from 28.1(6) kHz at 20 GPa to 61.9(7) kHz at 58 GPa. This increase is likely originated in the high compressibility and rapidly reducing next-nearest neighbour distances between molecular H₂ units, enhancing 96 quadrupolar coupling. The asymmetry parameter η (Fig. 3a, bottom panel) was found to be 97 almost constant within experimental errors varying between 0.44(6) at 20 GPa and 0.49(9) at 98 58 GPa. Based on the hexagonal crystal structure of phase I^{14,21}, η can be expected to be close 99 to 0.5, which is in excellent agreement with values derived from the analysis of our NMR 100 measurements.

101 Above $P \approx 68$ GPa, however, we observed a sudden decrease in both C_q and η 102 coinciding with a resonance peak splitting. Since no structural rearrangement of H₂ molecules 103 is reported by diffraction methods²¹ or Raman spectroscopy²² at room temperature in this *P*-104 range, effects based on a modification of the H₂ nuclear spin system should be considered.

105 This pressure indicates a turning point in the behaviour of the ¹H spin system, as the 106 observed peak splitting devolves from having dominantly quadrupolar characteristics towards 107 a system controlled by nuclear dipole-dipole coupling, resulting in pronounced I=1/2 line shapes²³ with a frequency difference between spectral density function singularities directly 108 correlated to the distances between hydrogen atoms. Considering that in this case both, the 109 110 interaction with the nearest and next nearest neighbours will result in a dipolar NMR pattern, respectively, a superposition of signals as shown in Figure 1c can be expected. Computationally 111 obtained values of nearest and next-nearest neighbour distances¹⁵ are $r_n \approx 0.731$ Å and $r_{nn} \approx$ 112 1.342 Å at 120 GPa. Calculating the distances from the parameter-set obtained through analysis 113 of the NMR spectra for such a mixed scenario resulted in: $r_{\rm n} \approx 0.727$ Å and $r_{\rm nn} \approx 1.27(8)$ Å 114 115 at 123 GPa, in excellent agreement with the computational estimates by Labet et al.¹⁵. Figure 116 2 shows the comparison between experimental spectra and calculated I=1/2 line shapes between 117 71 GPa and 123 GPa. The respective values derived for both nearest (top panel) and nextnearest (middle panel) neighbour distances can be found in Figure 3b. Additionally, the 118 comparison between the equation of state derived from *ab-initio* computations ²⁴ and diffraction 119 data^{14,21}, along with the unit cell volumes (blue dots) derived from r_n and r_{nn} extracted from the 120 analysis of the NMR spectra are shown in the bottom panel of Figure 3b. 121

Homonuclear Lee-Goldburg decoupling sequences²⁵ have been used to suppress quadrupolar and dipolar line broadening in order to resolve isotropic chemical shifts, δ_{iso} . Figure 3c shows the evolution of δ_{iso} : initially decreasing from 8.6 to 5.9 ppm between 20 and 59 GPa, δ_{iso} has an inflection point at ~60 GPa and raises under further compression to 22.7 ppm at 123 GPa. Comparison with Raman data²⁶ suggests that the minimum in δ_{iso} coincides with the well-known turn-over in the Raman shift of the H₂ vibron caused by a weakening of intramolecular and increased intermolecular interactions²⁷.

129

132 **Discussion:**

133 The presented data analysis leads to the following interpretation of the observed effects: 134 At P < 60 GPa, ¹H-NMR data is characteristic for an I=1 quadrupolar spin system as expected for ortho-H₂. In this regime, individual nuclear spin angular momenta couple with their nearest 135 neighbours (separated on average by r_n), leading to a significant wave function overlap within 136 the molecular units and a stabilisation of the spin isomers. The excellent agreement between 137 138 NMR derived values for the electric field gradient asymmetry parameter η and values inferred from X-ray diffraction data^{14,21} strengthens this assessment. Following the theoretical study of 139 Strzhemechny et al.²⁸, this compression driven enhancement of the quadrupolar coupling 140 constant C_q in this *P*-regime may be interpreted as an experimental evidence for the mechanism 141 142 of ortho-para conversion through electric quadrupole interaction.

143 At P > 60 GPa, quadrupolar coupling rapidly diminishes despite the absence of a structural 144 rearrangement of the molecular H₂ units. Starting from about 70 GPa, spectral features 145 characteristic of homonuclear dipole-dipole coupling between nearest and next-nearest 146 neighbours become apparent. Provided the good agreement between NMR data with DFT¹⁵ and experimentally^{14,21} derived intermolecular and interatomic distances, this shift in behaviour 147 implies intramolecular coupling of nuclear spins to become increasingly perturbed. The 148 inflection-point in the isotropic chemical shift δ_{iso} strengthens this hypothesis as the increasing 149 nuclear de-shielding above 60 GPa indicates a shift of electron density away from individual 150 151 molecular centres towards intermolecular regions.

¹⁴N-NMR on molecular nitrogen at P = 3 GPa (see Methods and Materials Section) supports this argument, as the nitrogen spin system shows clear characteristics of a nuclear spin triplet state anticipated within the non-pairing regime contrary to the quintuplet state stabilised in the molecular spin pairing regime.

156 In this work, in-situ High-Pressure Nuclear Magnetic Resonance was used to investigate 157 the nature of the nuclear spin statistics of molecular hydrogen up to 123 GPa in Phase I at room 158 temperature. It was found that even at moderately high pressures (< 100 GPa) intramolecular 159 nuclear spin coupling broke down and the hydrogen spin system adopted an average dipolar I=1/2 value. Crossovers of the nuclear spin statistics of a quantum solid such as hydrogen have 160 so far not been observed and given the large compressibility of hydrogen in conjunction with 161 162 strong nuclear quantum effects, this crossover phenomenon might only be experimentally 163 observable in molecular H₂. Nuclear spin statistics of similar diatomic molecules (e.g. N₂) are likely to be best described as non-pairing nuclear spins due to enhanced atomic masses as wellas reduced compressibilities due to the presence of core electrons.

166 This nuclear spin crossover may have far reaching consequences for understanding 167 different phenomena such as the stabilisation of magnetospheres of gas and ice giant planets 168 containing large quantities of molecular H₂.

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170 Methods

171 <u>Diamond Anvil Cell Preparation</u>

Two diamond anvil cells, equipped with pairs of diamond anvils with a culets size of 250 μ m and 100 μ m, were prepared. Rhenium gaskets were pre-indented to 25 μ m and 10 μ m respectively, and 80 μ m and 40 μ m diameter holes were laser drilled in the centre of the indentation to form the sample cavities, resulting in sample volumes of about 125 pl and 13 pl respectively.

177 The diamond anvils were coated with a 1 μ m thick layer of copper using physical vapour 178 deposition²⁹. Double³⁰ (in the case of the 250 μ m diamonds) and triple³¹ (for the 100 μ m 179 diamonds) stage Lenz-lens radio-frequency resonators were produced by using focused ion 180 beam milling. To ensure electrical insulation and avoid hydrogen diffusion into the rhenium, 181 the gaskets were coated by 500 nm thick layers of Al₂O₃. Radio-frequency excitation coils were 182 made from 100 μ m thick, teflon insulated, copper wire and arranged such that a Helmholtz coil 183 pair is formed.

Hydrogen loading was conducted at the ESRF at and pressure was increased at cryogenic temperatures to avoid rapid hydrogen diffusion into the diamond anvils. Pressure was calibrated by means of the diamond edge Raman scale^{32,33}. Comparison of the vibron frequencies of the H₂ samples at elevated pressures shows a slight systematic offset of less than 5 GPa at the highest pressures where Raman data was collected³⁴.

189

190 <u>NMR Experiments</u>

All NMR experiments were conducted using a solid-state NMR spectrometer from *Tecmag Inc.* (Redstone) equipped with a 100 W pulse amplifier. To polarize the nuclear spin system, we used a sweepable electromagnet with an average magnetic field of 1 T and sufficiently high homogeneity. Intense ¹H-NMR signals were recorded at frequencies of 45.26 MHz, corresponding to an external magnetic field strength of about 1063 mT. Using nutation experiments, we found optimal excitation pulses between 1 - 1.2 μ s for both cells, in reasonable agreement with earlier experiments^{29–31,35}. Free induction decays were excited using a single pulse of 833 kHz to 1 MHz bandwidth. The spectrometer was blanked off for 1 μ s to avoid damage to the pre-amplifier. Supplementary Figures S2 and S3 show all ¹H-NMR spectra recorded by this method. 25000 scans were accumulated for each spectrum (Figure 2).

In order to resolve isotropic chemical shifts, δ_{iso} , a Lee-Goldburg pulse for homonuclear decoupling was used²⁵. The resulting narrowed NMR spectra had a FWHM line width of about 3 ppm, thus the resolution accuracy of δ_{iso} was found to be in the order of 0.1 ppm (Figure S4). Resonance frequencies were referenced towards an aqueous solution of tetramethylsilane in a similar DAC at ambient pressure conditions.

207

208 <u>Computation of NMR Lineshapes and AsymmetryPparameters of the Electric Field Gradient</u>

Calculation of the NMR line shapes was carried out following the analytical method outlined
 by Bloembergen and Rowland³⁶, Pake²⁰ and Hughes and Harris³⁷:

Using the standard expressions for the resonance frequency distribution ω for both first order
quadrupole interaction as well as homonuclear dipole-dipole interaction:

213
$$\omega(\alpha,\beta,m) = \omega_Q \cdot (m+1/2) \cdot \left(\frac{3\cos^2\beta - 1}{2} - \frac{\eta}{2}\sin^2\beta\cos(2\alpha)\right), \qquad (1)$$

214
$$\omega_{i}(\alpha) = d_{i} \cdot \left(\frac{3\cos^{2}\beta - 1}{2}\right), \qquad (2)$$

215

216 with

217

218
$$\omega_Q = \frac{6\pi}{2I(2I+1)} \cdot C_q, \tag{3}$$

219
$$C_{\rm q} = \frac{e^2 q Q}{h}, \qquad (4)$$

220
$$d_{\rm i} = \frac{\mu_0 \gamma_n^2 \hbar}{8\pi^2 r_{\rm i}^3},$$
 (5)

221

where the Euler angles α and β describe the orientation of the crystallites with respect to the external magnetic field. γ_n is the gyromagnetic ratio of the hydrogen nuclei, m the nuclear spin quantum number (m = 1, 0, -1) and r_i the average distance between interacting hydrogen nuclei, r_n or r_{nn} , respectively. η describes the asymmetry of the electric field gradient tensor (V_{ij}) in the principal axis system as:

227
$$\eta = \frac{V_{yy} - V_{xx}}{V_{zz}}, |V_{zz}| > |V_{xx}| > |V_{yy}|.$$
 (6)

228

229 The line shape function, $P(\omega)$, for quadrupolar spin interactions, is given by:

231
$$P(\omega) = \sum_{m} \int_{-1}^{1} \frac{\mu}{4\pi} \sin(\beta(\omega, \alpha, m)) \cdot \left(\left\| \frac{\partial \beta(\omega, \alpha, m)}{\partial \omega} \right\| \right) d(\cos(2\alpha)),$$
(7)

 $P(\omega) = \sum_{i} \int_{-1}^{1} \frac{\mu}{4\pi} \sin(\beta(\omega_{i}, \alpha)) \cdot \left(\left\| \frac{\partial \beta(\omega_{i}, \alpha)}{\partial \omega_{i}} \right\| \right) d(\cos(2\alpha)),$

232

where $\beta(\omega, \alpha, m)$ denotes the inverse function of eq. (1) with respect to β , and μ accounts for the multiplicity of spectral functions. For the dipolar interaction $P(\omega)$, is given by

(8)

236

237 238 where $\beta(\omega_i, \alpha)$ denotes the inverse function of eq. (2) with respect to β , and μ accounts for the

239 multiplicity of spectral functions.

Cut-off frequencies of the resulting spectral line functions were chosen according to the Heaberlein convention for NMR shift tensors³⁸. Spectral line broadening was accounted for by convolution of the total line shape function with a Voigtian line of defined Lorentzian and Gaussian widths. In order to fit the experimental data, the respective line shape function $P(\omega)$ is optimized by varying C_q and η for quadrupolar coupling and r_n and r_{nn} for dipolar coupling. The corresponding Python scripts are available from the authors upon request. Table S1 summarises all fit parameters.

In order to calculate the asymmetry parameter η of the electric field gradient tensor in the spin-pairing regime, we used the second derivative of the electric potential, $V(\mathbf{r})$, defined by the molecular center of gravity positions from diffraction measurements¹⁴:

250
$$V(\mathbf{r}) = \frac{e}{4\pi\epsilon_0} \sum_{i} \frac{1}{\sqrt{(x-x_i)^2 + (y-y_i)^2 + (z-z_i)^2}},$$
 (9)

251

 $V_{ij} = \frac{\partial V(r)}{\partial x_i \partial y_j}.$ (10)

Using eq. (6) under consideration of the ordering of the components of V_{ij} in the principal axis system allows computation of η from crystallographic data.

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263 Table 1: Fitting parameters of ¹H-NMR spectra. C_q is the quadrupole coupling constant, η the asymmetry 264 parameter of the electric field gradient tensor in the principle axis system, r_n and r_{nn} are the nearest and second

- 265 nearest neighbour distances, respectively. The isotropic chemical shift, δ_{iso} , was derived after homonuclear Lee-
- 266 *Goldburg decoupling.*
- 267

					Lee-Goldburg
	1st Order Quadrupole		Dipole-Dipole		Decoupling
	Interaction		Interaction		δ_{iso}
<i>P</i> in GPa	C_q in kHz	η	<i>r</i> _n in Å	r_{nn} in Å	in ppm
20	28.1(6)	0.44(6)			8.665(112)
24	27.9(8)	0.43(4)			7.363(112)
36	30.0(7)	0.50(7)			6.429(112)
42	32.7(5)	0.52(4)			5.951(112)
47	35.2(6)	0.44(6)			5.928(125)
50	44.8(4)	0.59(3)			5.905(114)
54	48.0(3)	0.46(7)			5.924(150)
58	61.9(7)	0.49(8)			6.139(120)
68	43.9(9)	0.37(9)			6.670(173)
71	24.5(8)	0.20(9)	0.736(5)	1.509(14)	7.280(127)
77	18.6(7)	0.15(7)	0.733(5)	1.457(13)	8.479(195)
85	16.6(3)	0.14(6)	0.732(5)	1.430(11)	10.612(149)
97	15.0(4)	0.10(5)	0.732(5)	1.340(12)	15.351(100)
106	20.9(6)	0.04(1)	0.731(5)	1.307(14)	18.895(153)
115	19.2(9)	0.05(7)	0.729(5)	1.270(11)	21.323(147)
123	19.5(9)	0.02(7)	0.727(5)	1.270(18)	22.673(154)

268

269 <u>¹⁴N-NMR of Molecular Nitrogen at 3 GPa</u>

Molecular nitrogen was measured using natural isotopic composition, where the majority of molecules can be expected to be pairs of ¹⁴N nuclei. As ¹⁴N nuclei have a nuclear spin of *I*=1, one can expect a spin pairing scenario similar to molecular D₂: the para-N₂ states consist of a quintuplet subsystem with *I*=2 whereas the ortho-N₂ states are a triplet subsystem.

The electric field gradient asymmetry parameter η was estimated according to diffraction data¹⁴ to be around 0.23. Recorded ¹⁴N-NMR spectra (Figure S2; right panel) do not show pronounced shoulder, expected for a *I*=2 quadrupolar powder pattern in absence of $m_{-2\to-1}$ and $m_{1\to2}$ transitions. In fact, the spin system is well described by a *I*=1 spin system using the estimated value for η (Figure S2; left panel).

According to structural data³⁹, r_n can be estimated to be around 1.2 Å at this pressure; four times longer than the thermal de-Broglie wavelength of a single ¹⁴N atom. Therefore, the

281	wave	ve function overlap should be negligible in molecular nitrogen at these pressures and nuclear					
282	spins considered unpaired.						
283							
284	Data availability						
285	The data supporting the findings of this study are publicly available from the corresponding						
286	author upon request.						
287							
288	Code availability						
289	he used custom Python code for simulating the NMR spectra is available from the						
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379 Acknowledgements

380 We thank Nobuyoshi Miyajima for help with the FIB milling. We are very thankful to Graeme

381 Ackland and Gerd Steinle-Neumann for fruitful discussions.

382

383 Funding

- The authors thank the German Research Foundation (Deutsche Forschungsgemeinschaft, DFG, 384 Project Nos. DU 954/11-1, DU 393/13-1, DU 393/9-2, STE 1105/13-1 and ME 5206/3-1) and 385 the Federal Ministry of Education and Research, Germany (BMBF, Grant No. 05K19WC1) for 386 387 financial support. D.L. thanks the Alexander von Humboldt Foundation for financial support. M. P. A. would like to acknowledge the support of the European Research Council (ERC) Grant 388 389 "Hecate" reference No. 695527 secured by Graeme Ackland. N.D. thanks the Swedish 390 Government Strategic Research Area in Materials Science on Functional Materials at 391 Linköping University (Faculty Grant SFO-Mat-LiU No. 2009 00971). 392 393 **Author Contributions** 394 T.M. and L.D. designed the experiment. T.M., S.K., A.K. and J.J. prepared the DACs and 395 NMR resonators. T.M., D.L., M.P.A., F.T. and A.K. performed and analysed the experiments. 396 T.M., M.P.A., D.L., F.T., N.D. and L.D. analysed the results and wrote the manuscript. 397 398 **Competing Interests**
- 399 The authors declare that they have no competing interests.
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