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Abstract

Purpose: To determine how deamidation and partial loss of the N- and C-terminal extensions alter the heat stability of β B1-crystallin.

Methods: Human lens β B1, a deamidated β B1, Q204E, and α A-crystallins were expressed. Truncated β B1 was generated by proteolytic removal of part of its terminal extensions. The aggregation and precipitation of these proteins due to heating was monitored by circular dichroism and light scattering. The effect of heat on the stability of both monomers and oligomers was investigated. The flexibility of the extensions in wild type and deamidated β B1 was assessed by ^1H NMR spectroscopy.

Results: With heat, deamidated β B1 precipitated more readily than wild type β B1. Similar effects were obtained for either monomers or oligomers. Flexibility of the N-terminal extension in deamidated β B1 was significantly reduced compared to the wild type protein. Truncation of the extensions further increased the rate of heat-induced precipitation of deamidated β B1. The presence of the molecular chaperone, α A-crystallin, prevented precipitation of modified β B1s. More α A was needed to chaperone the truncated and deamidated β B1 than deamidated β B1 or truncated β B1.

Conclusions: Deamidation and truncation of β B1 led to destabilization of the protein and decreased stability to heat. Decreased stability of lens crystallins may contribute to their insolubilization and cataract formation.

Keywords

b1, human, crystallins, decreased, increased, stability, modified, requirement, chaperone, heat, beta

Disciplines

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Decreased heat stability and increased chaperone requirement of modified human β B1-crystallins

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Purpose: To determine how deamidation and partial loss of the N- and C-terminal extensions alter the heat stability of β B1-crystallin.

Methods: Human lens β B1, a deamidated β B1, Q204E, and α A-crystallins were expressed. Truncated β B1 was generated by proteolytic removal of part of its terminal extensions. The aggregation and precipitation of these proteins due to heating was monitored by circular dichroism and light scattering. The effect of heat on the stability of both monomers and oligomers was investigated. The flexibility of the extensions in wild type and deamidated β B1 was assessed by ¹H NMR spectroscopy.

Results: With heat, deamidated β B1 precipitated more readily than wild type β B1. Similar effects were obtained for either monomers or oligomers. Flexibility of the N-terminal extension in deamidated β B1 was significantly reduced compared to the wild type protein. Truncation of the extensions further increased the rate of heat-induced precipitation of deamidated β B1. The presence of the molecular chaperone, α A-crystallin, prevented precipitation of modified β B1s. More α A was needed to chaperone the truncated and deamidated β B1 than deamidated β B1 or truncated β B1.

Conclusions: Deamidation and truncation of β B1 led to destabilization of the protein and decreased stability to heat. Decreased stability of lens crystallins may contribute to their insolubilization and cataract formation.

Crystallins are the major structural proteins in the lens and form complex protein-protein interactions with each other. α -Crystallins form large assemblies predicted to contain as many as 33 subunits and act as molecular chaperones to prevent the aggregation and precipitation of the crystallins [1]. β -Crystallins contain aggregates ranging in size from dimers to octomers, while γ -crystallins exist as monomers. β B1-crystallin is a major subunit of the β -crystallins and comprises 9% of the total soluble crystallins in the human lens [2]. In the newborn, β B1 is predominantly found in the β High fraction, comprised of tetramers and octomers, after isolation by size exclusion chromatography [3]. Interaction of β B1 with other β -crystallins is likely to contribute to the stability of the β High assembly [4], and thus, the stability and packing of crystallins in the lens.

β B1, like all the basic β -crystallins, is characterized by having N- and C-terminal extensions. The N-terminal extension of β B1 is the longest of any crystallin, 57 amino acids. During aging, β B1 is extensively truncated [3,5]. Initial cleavage within the first year of life removes the first 15 residues from the N-terminal extension. Several additional cleavages occur between residues 33 and 41 [6]. Analysis of the water-insoluble proteins from older lenses suggests the presence of even more extensively degraded forms of β B1 [7]. During

aging, increasing amounts of truncated β B1 are found in the β Intermediate and β Light assemblies, therefore, loss of the N-terminal extension of β B1 may cause dissociation of β -assemblies [4].

Extensive deamidation (conversion of Gln/Asn to Glu/Asp) has also been reported in crystallins from older lenses [3,5]. The introduction of these negative charges into the crystallins during aging would be expected to destabilize native structure and possibly increase susceptibility to aggregation. This hypothesis is supported by the finding of an increased amount of deamidation in γ S-crystallin in human cataractous lenses [8]. Our earlier studies used two-dimensional electrophoresis and mass spectrometry to identify several distinct forms of deamidated β B1 in 55-year-old donor lenses [3]. Deamidation was also recently identified at Asn 157 in water-insoluble β B1 [7]. With recent improvements in sensitivity of mass spectrometry, confirmation and identification of more sites is anticipated. A useful strategy to determine the effect of these deamidations on protein structure and stability is to use mutagenesis to engineer them into recombinantly expressed proteins. While the effect of the sites of deamidation in β B1 have yet to be determined, a site at Gln 204 [3,9], in a hydrophobic region of β B1, did not significantly alter the secondary structure of the protein but did affect the shape of the β B1 dimer [10].

We hypothesize that the post-translational modifications, deamidation and truncation, change the conformation of β B1. Indeed, in this study we use NMR spectroscopy to demon-

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strate that the N-terminal extension of deamidated β B1 has significantly reduced flexibility compared to the wild type protein. The changes in the structure, and hence stability of β B1 and its aggregates, may lead to insolubilization and cataract formation. Thus, the purposes of the present study were to extend our earlier work by first determining the stability of the β B1 monomers and homodimers to heat, and then to examine how post-translational modifications altered this stability. The effect of deamidation on the stability of crystallins has never been measured. We found that deamidation had a profound effect on β B1 stability and that truncation within the N- and C-terminal extensions further decreased stability. Truncation alone had little effect. Heating deamidated and/or truncated β B1s with an equal molar amount of α A resulted in precipitation of both proteins. However, additional α A prevented the precipitation. The role of hydrophilicity of β B1 in maintaining stability of β B1 and its complexes with α A-crystallin is discussed.

METHODS

Protein expression: Human recombinant wild type (wt) β B1, a deamidated mutant (Q204E) β B1, and wild type α A crystallins were expressed and purified as previously described [10]. Briefly, an RT-PCR product generated using gene-specific primers was inserted into pCR T7/CT TOPO vector (Invitrogen, San Diego, CA) and transformed into the cloning cells, Top 10F' (Invitrogen). Positive clones were transformed into the expression cells, BL21(DE3) Star (Invitrogen). Use of donor tissue for these experiments were approved by the Oregon Health and Science University institutional review board.

Proteins were purified as previously described [10]. Briefly, cell pellets were lysed by freeze-thawing and sonication in a phosphate buffer, pH 6.8. Cell lysates were centrifuged and the supernatants applied to a SP Sepharose Fast Flow cation exchange column (Amersham Biosciences, Piscataway, NJ). The purity of β B1 by this method has been previously demonstrated [10] and was checked by mass spectrometry. For NMR studies, β B1 was further applied to a size exclusion column (3x80 cm S-300 HR Sephacryl column, Amersham Biosciences) at a flow rate of 0.4 ml/min in 29 mM Na_2HPO_4 , 29 mM KH_2PO_4 , 100 mM KCl, 0.7 mM EDTA, 1 mM DTT, pH 6.8. α A was lysed from *E.coli* cell pellets and purified by size exclusion chromatography as described for β B1. This was followed by ion exchange (2.5x12 cm Macro Prep DEAE column, BioRad, Hercules, CA, USA) at a flow rate of 3 ml/min in 20 mM Tris/HCl, pH 7.4, 0.16 mM EDTA, 1 mM EGTA. Proteins were eluted using a linear 0-500 mM NaCl gradient.

Truncated β B1s were generated by cleaving purified wt and Q204E β B1s with recombinant calpain (Calbiochem, La Jolla, CA) at a ratio of 4 Units of Calpain (2.3 μ g) per mg β B1. Proteolysis was initiated by adding 2 mM calcium and the reaction was allowed to continue for 1-2 h at 37 °C. Calpain removes 47 amino acids off the N-terminus and 5 amino acids off the C-terminus of wt and Q204E β B1 ([10] and unpublished data). The reaction was stopped by removing the cal-

cium by buffer exchange at 4 °C.

Protein identity and purity were confirmed by mass spectrometry using an electrospray ionization mass spectrometer (ThermoQuest, San Jose, CA, USA) to compare predicted and calculated masses. Mass accuracy of the measurements was 0.01% as confirmed using a myoglobin standard. The purity of the preparations was also examined by electrophoresis using pre-cast polyacrylamide Nupage mini-gels (Novex, San Diego, CA). Proteins were visualized by staining with Coomassie blue G-250.

Circular dichroism: Initial denaturation experiments were performed at a rate of heating of 0.75 °C/min for wt β B1 from 30 to 80 °C. Due to precipitation of the protein at higher temperatures, no signal was recovered upon cooling. The CD signal was measured at 215 nm with an AVIV Circular Dichroism spectrophotometer, model 202 (Protein Solutions, Lakewood, NJ). A temperature below the temperature at which loss of secondary structure/precipitation occurred was chosen for the experiments below. The fraction unfolded protein to precipitated was calculated as the difference in signals (ellipticity) at a specified temperature and the minimum signal divided by the difference in minimum and maximum signals. Circular dichroism was performed at 0.1 and 0.4 mg/ml in 1 mm cells. Samples were dialyzed exhaustively against 10 mM phosphate buffer, pH 6.8, and 100 mM NaF. Protein concentrations were confirmed by amino acid analysis. Representative CD spectra from 180 to 260 nm for both proteins were previously published [10].

Light scattering: Light scattering was determined indirectly by measuring absorbance at 405 nm. Two different protocols were used, each optimized to investigate the effect of heat on either monomers or oligomers of β B1s. Based on previous data, a 0.1 mg/ml concentration contained predominantly monomers of β B1, while higher concentrations were predominantly dimers with small amounts of higher-ordered oligomers [10]. The samples at 0.1 mg/ml were heated at 55 °C for 725 min in a thermal jacketed cuvette with constant stirring (Cary 4 Bio UV-Visible spectrophotometer, Varian, Palo alto, CA). Incubations were performed in the same buffer as for circular dichroism, except the NaF was replaced by NaCl. No observable precipitate or change in protein concentration was seen following heating of wt β B1 under these conditions. Higher concentrations or temperatures resulted in precipitation of wt β B1.

Heat induced aggregation of β B1 was also measured in the wells of microtiter plates. Samples at 1.5 mg/ml concentration were heated in a programmable thermal controller to 55°C and the temperature monitored with a probe. Samples were placed in the following buffer after chromatography by repeated concentration: 20 mM phosphate buffer, pH 6.8, 75 mM NaCl, 1 mM DTT, 0.5 mM EGTA. At set intervals, each tube was vortexed and 50 μ l transferred to a 96 well plate. Samples were shaken and absorbance read at 405 nm on a microtiter plate reader. All samples used in light scattering experiments were freshly prepared and stored in a protease inhibitor cocktail (Complete Inhibitor Tablets, Boehringer Mannheim GmbH, Mannheim, Germany).

Protein assays: Protein content was assayed by the Coomassie Plus-200 assay (Pierce, Rockford, IL) following manufacturer's instructions.

NMR spectroscopy: ^1H NMR spectra were acquired at 500 MHz and 25 °C using a Varian Inova-500 spectrometer (Varian, Inc., Palo Alto, CA). The acquisition conditions are outlined elsewhere [11]. Mixing times of 70 ms and 100 ms were used in the TOCSY and NOESY spectra respectively. In all experiments, WET methods were used for suppression of the large water resonance [12].

RESULTS

In order to test the stability of βB1 -crystallins, recombinant wild type βB1 and a deamidated βB1 mutant (Q204E) were heated from 30 to 80 °C at a rate of 0.75 °C per minute (Figure 1). Circular dichroism, followed by the signal at 215 nm, indicated no loss of secondary structure for either protein up to 62 °C. This was followed by irreversible loss of secondary structure and precipitation of the protein by 70 °C. Therefore, temperatures below 60 °C were chosen for experiments below.

As a measure of unfolding and aggregation, changes in absorbance due to light scattering at 405 nm were followed for dilute solutions of βB1 (0.1 mg/ml) upon heating at either 50 or 55 °C (Figure 2A). The molar mass of βB1 determined by multi-angle laser light scattering at this concentration was previously shown to be 30,000 Daltons, suggesting the protein is predominantly monomeric [10]. The deamidated mutant protein showed a significantly greater increase in light scattering at both temperatures, again demonstrating lower thermal stability. At 50 °C, the light scattering of deamidated βB1 increased rapidly with biphasic kinetics. The initial rate at a concentration of 0.1 mg/ml for deamidated βB1 was 0.015 min^{-1} as compared to 0.006 min^{-1} for wild type βB1 .

The ability of αA to prevent, via its chaperone action, the

aggregation of βB1 upon heating was also tested. The presence of 1:1 molar ratio of αA to wt βB1 significantly diminished the increase in light scattering seen upon heating wt βB1 (Figure 2B). In contrast, a 1:1 molar ratio of αA to deamidated βB1 produced stabilization against precipitation, followed by a sharp, dramatic increase in light scattering (Figure 2C). The sharp increase suggested precipitation of the Q204E βB1 and αA complex as was seen below (Figure 3B). This behavior could be prevented by increasing the concentration of αA . Thus, a 2:1 molar ratio of αA to deamidated βB1 prevented the precipitation of deamidated βB1 to a level similar to that seen for native wt βB1 in the presence of an equimolar amount of αA .

Due to instrumental sensitivity and our interest in the structure of the monomeric protein, the above studies were conducted at a very low concentration of βB1 . Because oligomerization of βB1 is concentration dependent [10,13] and oligomer state can influence thermal stability, we also examined heat denaturation at a higher βB1 concentration of 1.5 mg/ml (Figure 4). The results were similar to those observed at 0.1 mg/ml. Deamidated βB1 again exhibited greater light scattering than wt βB1 . As a control, heating αA did not result in any changes in absorbance (Figure 4A). Thermal denaturation at the higher concentration resulted in readily observable precipitation, and deamidated βB1 produced more precipitated protein than wt (Figure 3A).

Similar results were obtained after incubating αA with modified βB1 s at either 0.1 or 1.5 mg/ml. A 1:1 equal molar ratio of deamidated βB1 with αA did not prevent aggregation; in fact, precipitation was enhanced (Figure 4B). However, adding twice as much αA significantly inhibited precipitation (Figure 4B). SDS-PAGE of the proteins following centrifugation showed that the increased light scattering observed in Figure 4B was due to precipitation of both the deamidated βB1 and αA (Figure 3B, lanes 1 and 2). Additional αA increased the solubility of both proteins (Figure 3B, lanes 4 and 5).

The next set of experiments examined the role of the N- and C-terminal extensions on the heat stability of βB1 . Both N- and C-terminal extensions of βB1 and deamidated βB1 were cleaved by calpain. Calpain previously has been shown to remove 47 amino acids from the N-terminus and 5 amino acids from the C-terminus [10]. In contrast to deamidation, little difference in the rate of absorbance increase was observed between wt and truncated wt βB1 s during heating (Figure 2A and Figure 5A). However, there was a marked difference in the interaction of αA with the two proteins. A 1:1 ratio of truncated wt βB1 and αA resulted in a lag in the onset of aggregation, followed by a rapid increase in absorbance similar to that seen for deamidated βB1 . This was not observed with full-length wt βB1 (Figure 2B). A 2:1 ratio of αA to truncated wt βB1 increased the duration of the initial lag phase (Figure 5B).

Heating the doubly modified truncated, deamidated βB1 led to a much more rapid increase in aggregation than heating deamidated βB1 (Figure 5A). This increase in light scattering did not fit parameters for biphasic kinetics as was observed

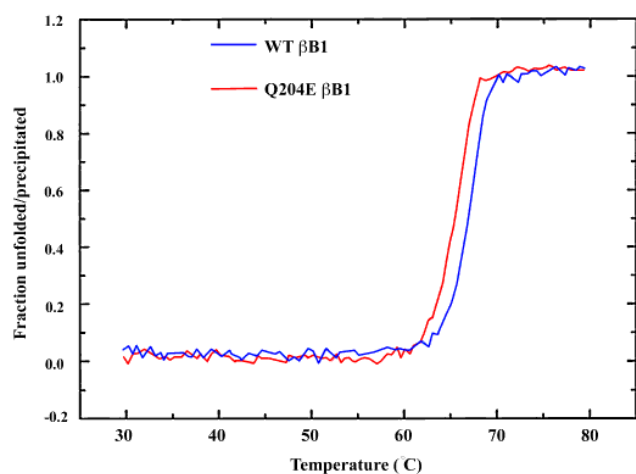


Figure 1. Percent of unfolded wt βB1 and Q204E βB1 . Percent of unfolded wt βB1 (red curve) and Q204E βB1 (blue curve) were measured by changes in the CD spectra at 215 nm. Protein concentrations are 0.4 mg/ml in 10 mM phosphate buffer, pH 6.8. Cell path length is 1 mm.

for deamidated β B1. Incubating an equal molar amount of α A with the truncated, deamidated β B1 only prevented an increase in light scattering for the first 100 min (Figure 5C). After this, a sharp increase occurred. The greatest and most rapid precipitation due to heating of all the proteins occurred with a 1:1 mixture of truncated, deamidated β B1 and α A. Additional α A at a 2:1 ratio did not prevent this increase (not shown), however a 4:1 ratio of α A: β B1 did. Thus, a greater amount of α A was required to prevent heat-induced aggregation of truncated, deamidated β B1 than for either truncated or deamidated β B1.

Previous ^1H NMR spectroscopic studies of other β -crystallin subunits, i.e. β B2, β A3, have shown that their N- and C-terminal extensions have great conformational flexibility compared to the domain core and adopt little or no preferred conformation (summarized in reference 13). In order to gain insight at the molecular level into the structural changes associated with deamidation of Q204 in β B1, ^1H NMR spectra were acquired of wt and deamidated β B1. Figure 6A shows the TOCSY NMR spectrum of cross-peaks arising from the NH protons of wt β B1 due to spin-spin coupled protons within the same amino acid. From analysis of this spectrum and a NOESY spectrum giving through-space connectivities between NH protons and the preceding α -CH protons, many of the cross-

peaks in the TOCSY spectrum could be assigned. From these data, it was apparent that these cross-peaks arise predominantly from the N- and C-terminal extensions of wt β B1 crystallin. The strong similarity of their chemical shifts to those for random coil peptides and hence lack of dispersion particularly in the α -CH chemical shifts, imply that the extensions have no ordered structure and have great conformational flexibility. The flexibility of the terminal extensions does not exclude the possibility of structure induced upon interaction with other proteins in the lens cell as would be suggested by NMR data on crystallin homogenates [14]. This lack of dispersion and the large number of residues in both extensions led to significant spectral overlap and hence the inability to assign all cross-peaks from the extensions of β B1.

The TOCSY spectrum of deamidated β B1 (Figure 6B) shows far fewer cross-peaks than in the spectrum of the wild type protein. Interestingly, cross-peaks from the C-terminal extension of β B1 were present in this spectrum but, in the main, those from the N-terminal extension were absent. Thus, introduction of the negative charge at residue 204, well distant from the N-terminal extension, leads to a significant loss of conformational mobility in the long N-terminal extension but does not affect flexibility of the much shorter C-terminal extension.

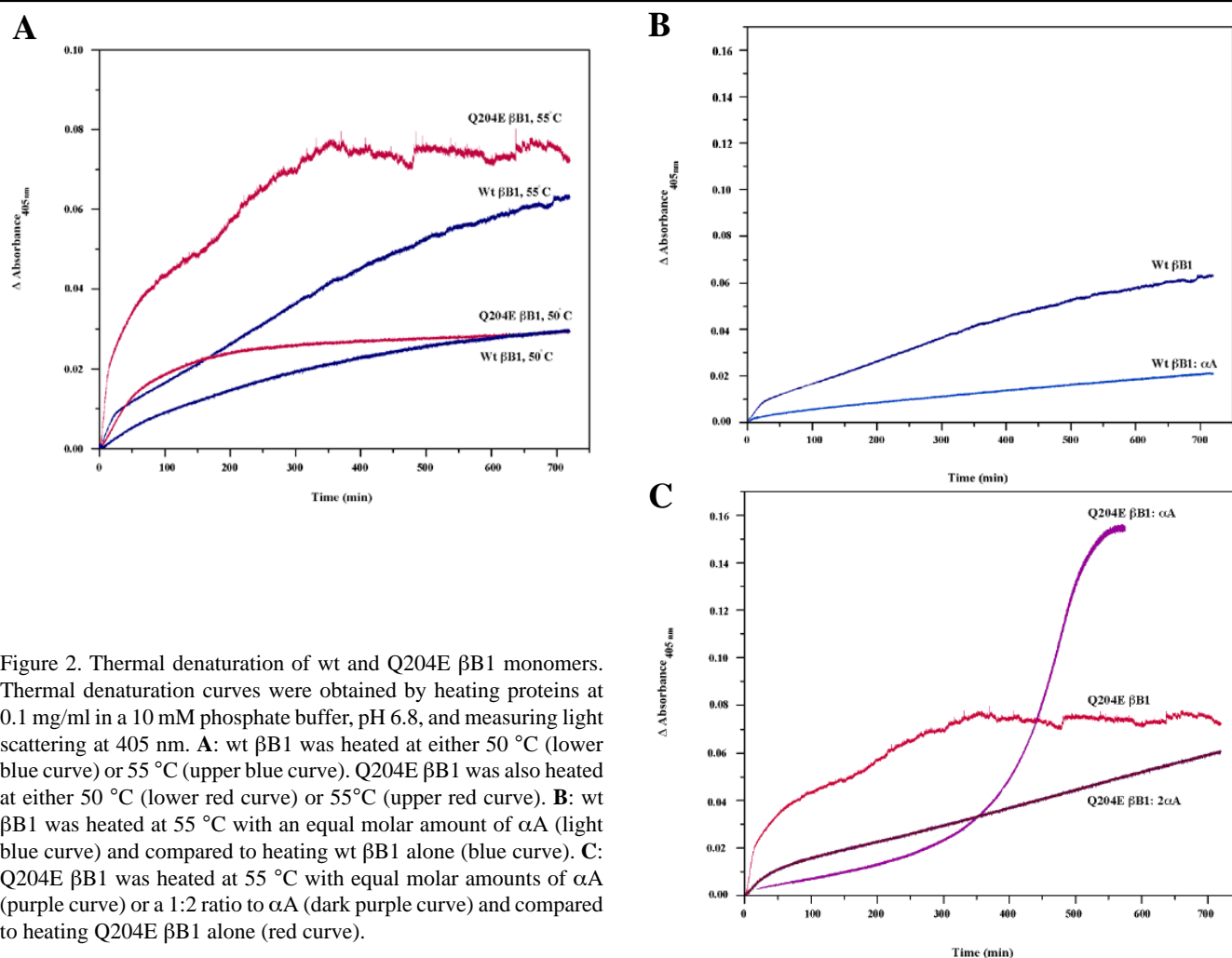


Figure 2. Thermal denaturation of wt and Q204E β B1 monomers. Thermal denaturation curves were obtained by heating proteins at 0.1 mg/ml in a 10 mM phosphate buffer, pH 6.8, and measuring light scattering at 405 nm. **A:** wt β B1 was heated at either 50 °C (lower blue curve) or 55 °C (upper blue curve). Q204E β B1 was also heated at either 50 °C (lower red curve) or 55°C (upper red curve). **B:** wt β B1 was heated at 55 °C with an equal molar amount of α A (light blue curve) and compared to heating wt β B1 alone (blue curve). **C:** Q204E β B1 was heated at 55 °C with equal molar amounts of α A (purple curve) or a 1:2 ratio to α A (dark purple curve) and compared to heating Q204E β B1 alone (red curve).

DISCUSSION

Proteolysis and deamidation are major post-translational modifications of lens crystallins in the aging human lens [3,5,7,15,16]. We previously reported that introducing a deamidation at Gln 204 in the C-terminus or truncation of the extensions of β B1 causes a change in the shape of the protein as evidenced by multi-angle laser light scattering [10]. The NMR data reported herein are consistent with this observation. The N-terminal extension in deamidated β B1 has reduced flexibility compared to its wild type counterpart possibly due to electrostatic interactions between one or more of its posi-

tively charged residues (K5, K21 and K23), and the introduced negative charge at residue 204. In this study we explored whether these modifications also decreased the heat stability of β B1.

We found that human recombinant β B1 was very heat stable, and retained its native secondary structure up to 62–65 °C (Figure 1). The stability of the β B1 secondary structure was most likely responsible for the resistance of the protein to heat-induced aggregation. Deamidation decreased the thermal stability of β B1 as measured by changes in light scattering, while truncation of the extensions alone did not effect the heat stability of β B1.

Deamidation decreased the heat-stability of both the monomer and oligomer forms of β B1. Previous light scattering measurements have estimated the size of β B1 at the concentrations used in this study to be a monomer at 0.1 mg/ml and an oligomer of predominately dimers at 1.5 mg/ml [10,13]. Similarly, at the relatively high concentrations used for the NMR studies, β B1 is present as an oligomer. These concentrations are considerably lower than the high concentrations found in the lens, but allow for comparison of the behavior of modified proteins. By homology to the β B2 X-ray crystal structure [17], the site of deamidation in β B1 is predicted to be at the interface between the two subunits of the dimer. Deamidation alters the shape of the dimer [10] by reducing

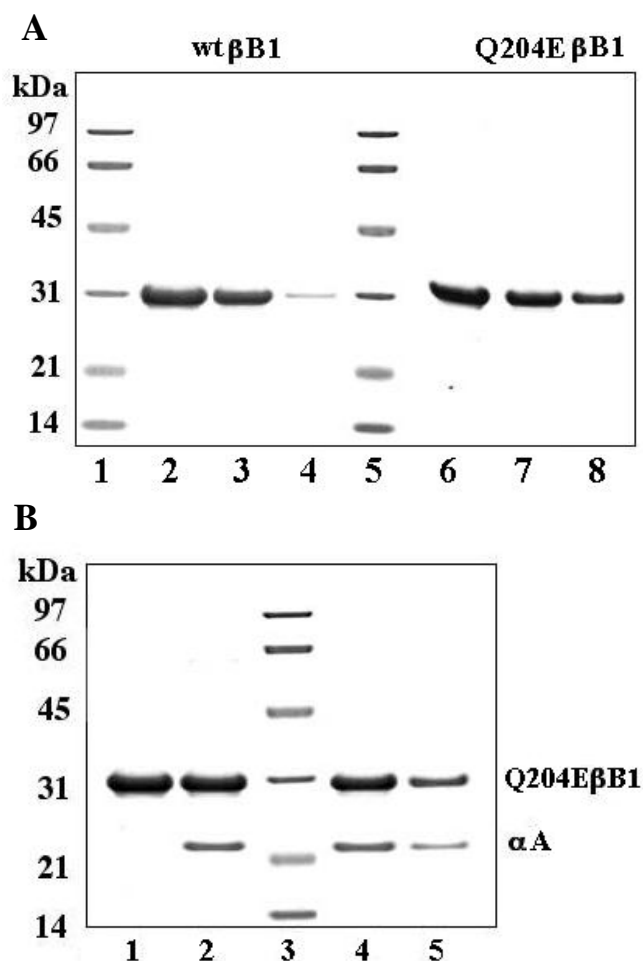


Figure 3. SDS-PAGE of β B1 samples from Figure 4. SDS-PAGE of wt β B1 and Q204E β B1 samples from Figure 4. **A**: Soluble and precipitated proteins were separated by centrifugation after heating for 180 min. Lanes 1 and 5 show molecular weight markers. Lanes 2 and 6 show wt β B1 and Q204E β B1 before heating. Lanes 3 and 7 show soluble wt β B1 and Q204E β B1. Lanes 4 and 8 show precipitated wt β B1 and Q204E β B1. **B**: Soluble and precipitated Q204E β B1 after heating in the presence of α A. Lanes 1 and 4 show soluble Q204E β B1 and α A after heating at equal molar concentrations (lane 1) and a 1:2 ratio of Q204E β B1 to α A (lane 4). Lanes 2 and 5 show precipitated Q204E β B1 and α A after heating at equal molar concentrations (lane 2) and 1:2 ratio of Q204E β B1 to α A (lane 5). Lane 3 shows molecular weight markers.

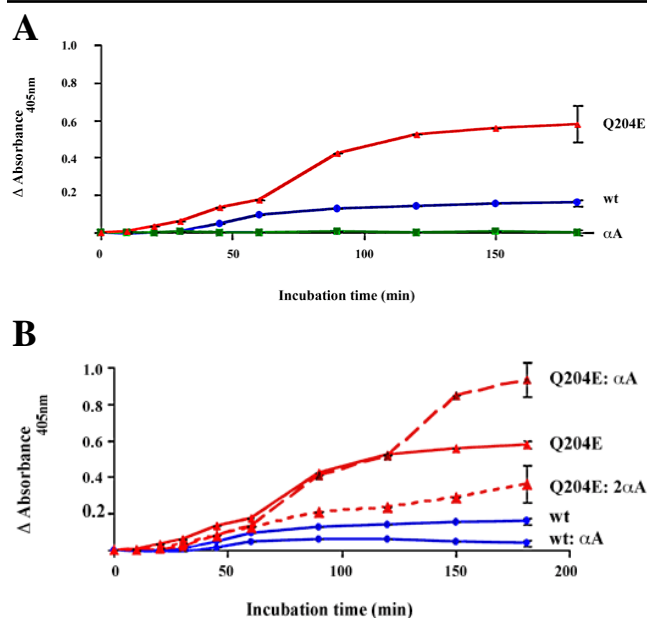


Figure 4. Thermal denaturation of wt and Q204E β B1 oligomers. Thermal denaturation of wt and Q204E β B1 oligomers was assessed at 55 °C. Thermal denaturation curves were obtained by heating proteins at a concentration of 1.5 mg/ml in a 20 mM phosphate buffer, pH 6.8, and measuring light scattering at 405 nm. Error bars are standard deviation, N=4. **A**: wt β B1 (blue curve), Q204E β B1 (red curve), and α A (green curve) were heated at a concentration of 1.5 mg/ml for wt and Q204E β B1 and 1.1 mg/ml for α A. **B**: Wt β B1 (blue curves) was heated with or without an equal molar amount of α A. Q204E β B1 (red curves) was heated with or without equal molar amounts of α A or a 1:2 ratio to α A.

the flexibility of the N-terminal extension (Figure 6). As a result, the stability of the dimer is lowered under heating stress. At the low concentrations used for monomer studies it is not possible to study the protein using NMR spectroscopy to ascertain whether the N-terminal extension has reduced flexibility in monomeric deamidated β B1. The reduced temperature stability of the mutant would imply, however, that similar structural features are present in both monomeric and oligomeric deamidated β B1.

Heating each of the expressed β B1 proteins examined here resulted in an increase in light scattering due to protein aggregation. At longer times, higher temperatures, and higher concentrations, heat-induced aggregation resulted in extensive precipitation that was clearly observed after centrifugation. The molecular chaperone, α A-crystallin, was added to determine if it could prevent the heat-induced aggregation and precipitation of the β B1-crystallins. A 1:1 molar ratio of α A was able to chaperone wt β B1 at 55 °C at both high and low β B1 concentrations (Figure 2). However, after a significant lag period, heating a 1:1 ratio of deamidated β B1 to α A resulted

in an accelerated precipitation of both proteins. α A-crystallin is a very efficient chaperone at these elevated temperatures, which is most likely related to its enhanced subunit exchange rate [18]. The poor chaperoning ability of α A with deamidated β B1 must therefore arise from the structural changes in deamidated β B1 compared to the wt protein. α A and the destabilized β B1 proteins form a complex as a result of the former's chaperone action. However, at stoichiometric amounts of both proteins, the complex containing deamidated β B1 eventually becomes insoluble as the hydrophobic deamidated β B1 saturates the binding sites on α A. Similar behavior is observed during the chaperone action of destabilized forms of α B-crystallin [19] and a related small heat-shock protein, Hsp25 [20]. Thus, R120G α B is a poor chaperone because it has a disordered structure. Interaction with a target protein leads to complexation and rapid precipitation of both proteins [19]. For Hsp25, deletion of its flexible and polar C-terminal extension leads to similar behavior when it complexes to a target protein, α -lactalbumin [20]. It would seem, therefore, that for deamidated β B1, the reduction in flexibility of its N-terminal

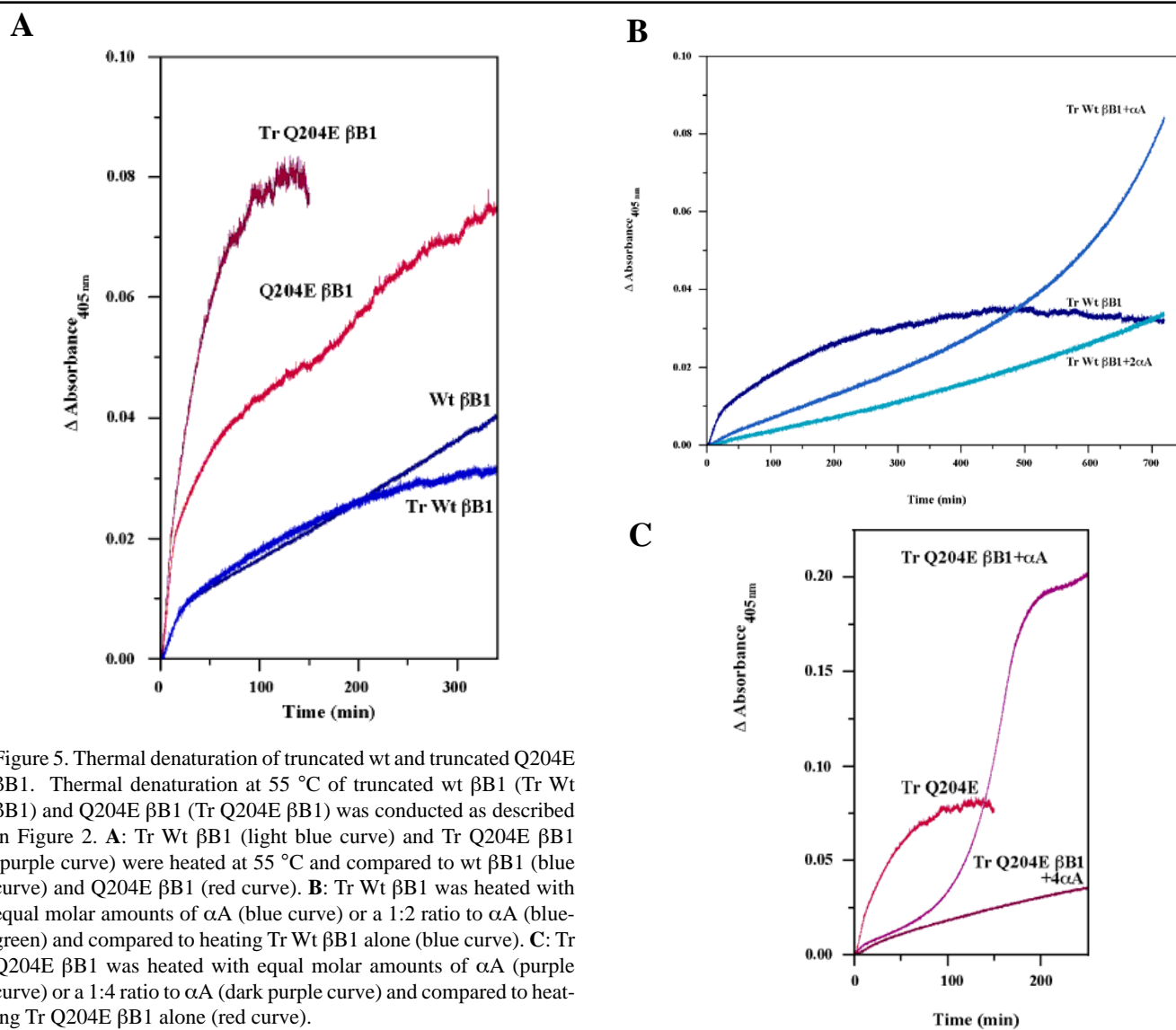


Figure 5. Thermal denaturation of truncated wt and truncated Q204E β B1. Thermal denaturation at 55 °C of truncated wt β B1 (Tr Wt β B1) and Q204E β B1 (Tr Q204E β B1) was conducted as described in Figure 2. **A:** Tr Wt β B1 (light blue curve) and Tr Q204E β B1 (purple curve) were heated at 55 °C and compared to wt β B1 (blue curve) and Q204E β B1 (red curve). **B:** Tr Wt β B1 was heated with equal molar amounts of α A (blue curve) or a 1:2 ratio to α A (blue-green) and compared to heating Tr Wt β B1 alone (blue curve). **C:** Tr Q204E β B1 was heated with equal molar amounts of α A (purple curve) or a 1:4 ratio to α A (dark purple curve) and compared to heating Tr Q204E β B1 alone (red curve).

extension and potentially the subtle structural changes associated with deamidation, lead to a protein that is unable to be chaperoned efficiently by α A. If deamidation results in partial unfolding of the core domains, this would then lead to a more hydrophobic protein. However, in the case of wt β B1, the hydrophobicity of the domain core of the protein is counteracted by the flexibility of its charged extensions to the extent that, under heating stress, the complex it makes with α A remains soluble.

Truncation of N- and C-terminal extensions did not affect the thermal stability of β B1 (Figure 5), implying that they are not required to be fully intact in order to maintain solubility of the protein under heating stress. However, when interacting with α A, the truncation of the extensions has a similar effect to that of deamidation, i.e. the protein is more hydrophobic, which leads to precipitation of the complex between both proteins. Thus, removal of a portion of the flexible extensions in β B1 makes the complex with α A less stable. Other studies have shown the importance of the flexible terminal extensions of α -crystallin and other sHsps in maintaining solubility of the chaperone-target protein complex [21,22]. The results presented herein suggest that hydrophobicity, and possibly flexibility in the target protein, in this case β B1, are also important in solubilization of the complex. The extensions have been suggested to be important in stabilizing the interactions of β B1 with other β subunits [4]. This is supported by our pre-

vious report of a decreased heat stability when the β High-crystallins from bovine lenses were proteolyzed [23]. The extensions may function to allow oligomerization with other β subunits instead of increasing the stability of their secondary structure.

The combined effect of truncation and deamidation leads to an enhanced rate of precipitation under heat stress (Figure 5C), which is also consistent with the notion that increasing the hydrophobicity of the β B1 molecules promotes their mutual association. Furthermore, the results are consistent with the NMR data implying an interaction of the N-terminal extension with the introduced charge at Q204E. Coupled with this, deamidation could cause β B1 to unfold more readily with the resulting partly folded protein being less soluble without its hydrophilic extensions. Interestingly, *in vivo*, truncation and deamidation of β B1 occur together during aging [3,5].

In summary, we have shown that a deamidated mutant of β B1 crystallin, Q204E, displays decreased thermal stability compared to wt β B1. This decrease in thermal stability was observed both at concentrations where the proteins are predominantly monomers or dimers. NMR spectroscopy indicates that the deamidated mutant has a marked reduction in flexibility of its N-terminal extension. The structural alteration in the protein leads to a reduction in its thermal stability. Both deamidated β B1 and truncated β B1 have altered conformations and interaction with α A-crystallin, compared to the wt protein, requiring higher concentrations of this important chaperone to maintain solubility upon heat stress. The combination of truncation and deamidation diminished thermal stability to a greater degree than either modification alone. The decreased ability of α A to act as a chaperone of deamidated β B1 and truncated, deamidated β B1 may be another mechanism whereby deamidation and proteolysis could increase susceptibility of crystallins to insolubilization contributing to cataract formation.

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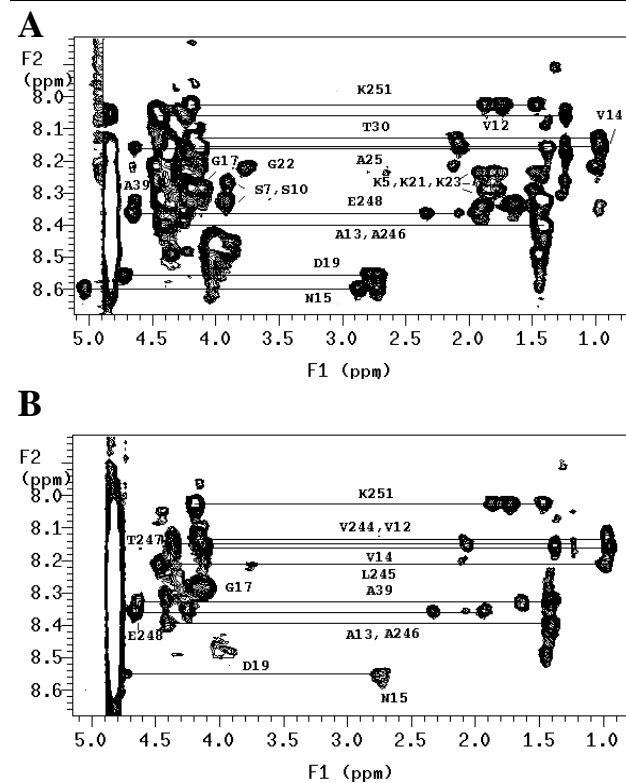


Figure 6. TOCSY NMR spectra of β B1 crystallins. TOCSY NMR spectra of (A) wt β B1 and (B) deamidated β B1. The spin lock mixing period in each experiment was 70 ms. Cross-peaks arising from the NH protons are shown. Those cross-peaks that could be assigned from NOESY spectra are indicated. A: B:

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