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9-1997

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Fox, J. L., & Yelle, R. V. (1997). Hydrocarbon Ions in the Ionosphere of Titan. Geophysical Research Letters, 24 (17), 2179-2182. https://corescholar.libraries.wright.edu/physics/437

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# **Hydrocarbon ions in the ionosphere of Titan**

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**Abstract. We have constructed a new model of the ionosphere of Titan that includes 67 species and 626**  reactions. Although  $N_2^+$  is the major ion produced **over most of the ionosphere, the ionization flows to ions whose parent neutrals have lower ionization potentials and to ions formed from species with large proton affinities. In contrast to other models, which have predicted**  that  $HCNH<sup>+</sup>$  should be the major ion, our calculations **suggest that the major ions at and below the ion peak are hydrocarbon ions, and H, C, and N-containing ions. Our predicted peak electron density for a solar zenith**  angle of 60<sup>o</sup> is about  $7.5 \times 10^3$  cm<sup>-3</sup> at an altitude of **1040 km.** 

### **1. Introduction**

In addition to N<sub>2</sub> and CH<sub>4</sub>, the thermosphere of Ti**tan contains significant densities of H2, H, and nonmethane hydrocarbons such as acetylene, ethane, and ethylene [e.g., Hunten ei at., 1984, \$irobet ei al., 1992], and appears to be well-mixed to very high altitudes. The atmosphere may also contain as much as 14% At.**  The homopause in the model of Strobel et al. [1992] **is near 1050 km, where the eddy diffusion coefficient is**  about  $1 \times 10^9$  cm<sup>2</sup> s<sup>-1</sup>. The exobase is at about 1500 km, and the density profiles of H and H<sub>2</sub> are altered appre**ciably by thermal escape at the top of the atmosphere.** 

**Information about Titan's ionosphere is limited to the radio occultation measurements made by the Voyager 1 spacecraft. Lindal et al. [1983] reported only**  upper limits of  $(3 - 5) \times 10^3$  cm<sup>-3</sup> near the termina**tors as the spacecraft entered and exited the occultation region. Recently, Bird et at. [1995] reanalyzed the Voyager radio occultation data and reported a possible detection with a peak electron density of 2700 cm -3 at an altitude of 1190 km at the evening terminator.** 

**The ionosphere of Titan has been predicted to arise from both solar photoionization and from the interaction of the neutral atmosphere with energetic electrons** 

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**Paper number 97GL02051. 0094-8534/97/97GL-02051 \$05.00**  **from Saturn's magnetosphere, in which part of the or**bit of Titan is embedded [e.g., Strobel and Shemansky, **1983; Atreya, 1986]. \$trobel et al. [1991] found, how**ever, that the  $N^+$  1085 Å and  $N_2$  Lyman-Birge-Hopfield **band intensities that were measured by the Voyager 1 spacecraft during its Titan flyby in 1980 could be reproduced with the solar source alone. Cravens and co-workers [Keller et al., 1992; Gan et al., 1992] constructed a model of the ionosphere of Titan that included the interaction of energetic magnetospheric electrons with the thermosphere and found it to be a minor source. We model here the ionosphere of Titan produced by photoionization and photoelectron-impact ionization alone. Although we ignore the interaction of the thermosphere with Saturn's magnetosphere, we do not mean to imply that magnetospheric electrons are unimportant all the time or for every process. In all probability, however, they are unimportant for ion production near and for a substantial distance above the dayside ion peak.** 

#### **2. The Model**

**For the background atmosphere, the density profiles of N• and CH4 are taken from Strobet et al. [1992], and**  mixing ratios of  $C_2H_2$ ,  $C_2H_4$ ,  $C_2H_6$ , and  $C_4H_2$  are taken **from Yung et al. [1984]. The mixing ratio of methane has been estimated as 2-10% in the Titan atmosphere [Hunten et al., 1984], but we have adopted a mixing ratio of 3%, consistent with the preferred model of Strobel et al. [1992]. The CO mixing ratio is adopted from the**  ground-based measurements of Gurwell and Muhleman **[1995]. The density profiles of these species are shown in Figure 1. The densities of 32 neutral species and 35**  ions are computed in the model, including H, H<sub>2</sub>, C, CH, CH<sub>2</sub>, <sup>1</sup>CH<sub>2</sub>, CH<sub>3</sub>, C<sub>2</sub>, C<sub>2</sub>H, C<sub>2</sub>H<sub>3</sub>, C<sub>2</sub>H<sub>5</sub>, C<sub>4</sub>H<sub>3</sub>,  $N, N(^{2}D), NH, NH_2, NH_3, CN, HCN, H_2CN, CNN,$ CHCN, C<sub>2</sub>N<sub>2</sub>, HC<sub>2</sub>N<sub>2</sub>, C<sub>3</sub>N, HC<sub>3</sub>N, H<sub>2</sub>C<sub>3</sub>N, CH<sub>3</sub>CN,  $CH_3NH_2$ ,  $C_4N_2$ , O, CO,  $H^+$ ,  $H_2^+$ ,  $H_3^+$ ,  $C^+$ ,  $CH^+$ ,  $CH_2^+$ ,  $CH_3^+$ ,  $CH_4^+$ ,  $CH_5^+$ ,  $C_2^+$ ,  $C_2H_7^+$ ,  $C_2H_2^+$ ,  $C_2H_3^+$ ,  $C_2H_5^+$ ,  $C_2H_6^+$ ,  $C_2H_7^+$ ,  $C_4H_3^+$ ,  $C_xH_y^+$ ,  $N^+$ ,  $NH^+$ ,  $NH_2^+$ ,  $NH_3^+$ ,  $NH_4^+$ ,  $HCN^+$ ,  $H_2CN^+$ ,  $CN^+$ ,  $C_2N^+$ ,  $C_xH_yN_z^+$ ,  $N_2^+$ ,  $N_2H^+$ ,  $O^+$ ,  $CO^+$ ,  $HCO^+$ , and  $NO^+$ . The temperature profile is taken from Strobel et al. [1992], and **is nearly isothermal at the exospheric temperature of approximately 175 K above about 1000 km. The elec**tron temperatures have been computed by Gan et al.

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**Figure 1. Neutral density profiles in the thermosphere of Titan that were adopted in these calculations.** 

**[1992] and by Roboz and Nagy [1994]. We adopt here**  the profile of *Gan et al.* [1992].

**The model includes photoabsorption and photoion**ization of  $N_2$ ,  $CH_4$ ,  $C_2H_2$ ,  $C_2H_4$ ,  $C_2H_6$ , N, C, H and  $H<sub>2</sub>$ , photodissociation of HCN, and photoionization of **CH3 by Lyman alpha. The solar spectrum we have adopted is the high solar activity F79050N spectrum of**  Hinteregger [private communication, see also Torr et al., 1979], and the solar zenith angle is 60°. Photoelectronimpact ionization is included for  $N_2$ ,  $CH_4$ ,  $H$ ,  $H_2$ ,  $N$  and **C. The most important ion production rate profiles are**  shown in Figure 2. The major ion produced is  $N_2^+$  from **600 to 1800 kin, and CH4 + above that altitude. Other**  important ions produced include  $N^+$ ,  $\text{CH}_3^+$ ,  $\text{C}_2\text{H}_2^+$ , and  $C_2H_4^+$ .

**The ion density profiles were computed including 626 chemical reactions, most of which were taken from the compilations of Anicich and Huntress [1986] and Ani-**



**Figure 2. Computed production rate profiles for the most important ions produced in the Titan ionosphere.** 

**cich [1993]. For neutral-neutral reactions, rate coeffi**cients were taken from the model of Yung et al. [1984] and from models of the interstellar medium [e.g., Millar et al., 1988]. For many of the most important dis**sociative recombination coefficients, the rates are unknown; where no measurements are available, values**  of  $(3 - 3.5) \times 10^{-7} (300/T_e)^{0.5}$  cm<sup>3</sup> s<sup>-1</sup> were assumed. **Very recent measurements, which have not been incorporated into this model, have shown that dissociative recombination coefficients for protonated alkanes do not increase with the size of the ion, but remain in the range**   $(5-8) \times 10^{-7}$  cm<sup>3</sup> s<sup>-1</sup> at 300 K for C<sub>1</sub>- to C<sub>8</sub>- ions **[Lefaoui et al., 1997]. We have also assumed that, in the absence of other information, the available channels in dissociative recombination reactions are populated with equal probability. Eddy and molecular diffusion**  for neutrals, and ambipolar diffusion of ions were in**cluded in our model, for which the lower and upper**  boundaries were at 600 km and 3000 km, respectively. **A preliminary model based on the same chemistry was Presented by Foz and Yelle [1995] and by Foz [1996a,b].** 

**The predicted densities of the major ions are shown**  in Figure 3. The peak electron density is about  $7.5 \times$  $10^3 \text{ cm}^{-3}$  near 1030 km. This is larger than either the **Voyager upper limits and the possible detection of Bird et al. [1995], but those measurements are for the terminator region, whereas our model is for the dayside at a**  solar zenith angle of 60<sup>°</sup>.

Although  $N_2^+$  is the major ion produced over much of **the ionosphere, it is destroyed by reaction with methane** 

$$
N_2^+ + CH_4 \rightarrow CH_2^+ + N_2 + H_2 \qquad (R1a)
$$

$$
\rightarrow CH_3^+ + N_2 + H, \qquad (R1b)
$$

and through reaction with  $H_2$ :

$$
N_2^+ + H_2 \rightarrow N_2H^+ + H. \tag{R2}
$$



**Figure 3. Computed steady-state densities of the most important ions in the Titan ionosphere. The curve labeled "e" is the electron density profile.** 

**At altitudes below about 1900 km, reactions (Rla)**  and (R1b) dominate; above 1900, H<sub>2</sub> becomes the ma**jor constituent, and reaction (R2) is more important.**  The column integrated production rate of  $N_2^+$  is 7.3  $\times$  $10^8$  cm<sup>-2</sup> s<sup>-1</sup>, and the column integrated rates for reactions (R1) and (R2) are  $6.7 \times 10^8 \text{ cm}^{-2} \text{ s}^{-1}$ , and  $4.2 \times 10^7$  cm<sup>-2</sup> s<sup>-1</sup>, respectively.

The CH<sub>3</sub> ion reacts with methane to produce  $C_2H_5^+$ **via** 

$$
CH_3^+ + CH_4 \rightarrow C_2H_5^+ + H_2, \qquad (R3)
$$

with a column integrated rate of  $5.9 \times 10^8$  cm<sup>-2</sup> s<sup>-1</sup>. Once formed,  $C_2H_5^+$  reacts with HCN to produce  $HCNH^+$ **through** 

$$
C_2H_5^+ + HCN \rightarrow HCNH^+ + C_2H_4. \qquad (R4)
$$

**The column integrated rate for (R4) in our model is**   $1 \times 10^8$  cm<sup>-2</sup> s<sup>-1</sup>.

Production of N<sup>+</sup> also leads to HCNH<sup>+</sup> formation via **the reaction:** 

$$
N^+ + CH_4 \rightarrow HCNH^+ + H_2, \qquad (R5)
$$

which has a column integrated rate of  $3.7 \times 10^7$  cm<sup>-2</sup> s<sup>-1</sup>. This reaction dominates HCNH<sup>+</sup> production at altitudes above 1500 km, whereas the sequence (R1-R4) **is more important below that altitude. Reactions (R4) and (RS) account for 63% and 25%, respectively, of HCNH + production. Other reactions that contribute significantly to HCNH + production include** 

$$
HCN^+ + CH_4 \rightarrow HCNH^+ + CH_3, \qquad (R6)
$$

$$
CH_3^+ + NH \rightarrow HCNH^+ + H_2, \qquad (R7)
$$

$$
C_2H_3^+ + HCN \rightarrow HCNH^+ + C_2H_2, \qquad (R8)
$$

$$
N^+ + C_2H_4 \rightarrow HCNH^+ + CH_2. \tag{R9}
$$

The total HCNH<sup>+</sup> column production rate is  $1.6 \times$  $10^8$  cm<sup>-2</sup> s<sup>-1</sup>.

**The reaction sequences discussed above have been included in previous models of Titan's ionosphere [e.g., Ip, 1990; Keller et al., 1992; Keller and Cravens, 1994].**  These calculations predicted that HCNH<sup>+</sup> should be **the major ion at the ionospheric peak. Our model differs from earlier work in the sequence of reactions following HCNH + production. Specifically, we include**  the reaction of  $HCNH<sup>+</sup>$  with minor neutral species that **have larger proton affinities than HCN or HNC. According to our model, HCNH + is lost primarily through**  reactions with NH<sub>3</sub>, C<sub>4</sub>H<sub>2</sub>, and HC<sub>3</sub>N:

 $HCNH<sup>+</sup> + NH<sub>3</sub> \rightarrow NH<sub>4</sub><sup>+</sup> + HCN,$  (R10)

$$
HCNH+ + C4H2 \rightarrow C4H3+ + HCN, \qquad (R11)
$$

$$
\quad \text{and} \quad
$$

**and** 

$$
HCNH^{+} + HC_{3}N \rightarrow CHCCNH^{+} + HCN, \qquad (R12)
$$

which are characterized by rate coefficients of  $2.4\times10^{-9}$ .  $1.8 \times 10^{-9}$ , and  $3.4 \times 10^{-9}$  cm<sup>3</sup> s<sup>-1</sup>, respectively, and column integrated rates of  $1.6 \times 10^7$ ,  $7.5 \times 10^7$ , and  $0.7 \times 10^7$  cm<sup>-2</sup> s<sup>-1</sup>, respectively.

Although reactions  $(R10-R12)$  ensure that  $HCNH<sup>+</sup>$ **will not account for a large fraction of the total ion density, it is not clear what the dominant ion will be, or if there will be a single major ion. In fact, it is likely that a large number of ions may have comparable densities. Hydrocarbon ions, once formed, quickly react with hydrocarbon neutrals to produce larger hydrocarbon ions. When nitrogen is present, many of the ions also may incorporate one or more nitrogen atoms. The chemistry of such ions is largely incomplete, and tracking of only a few individual species is possible at this point. We**  therefore adopt two pseudo-ions:  $C_xH_v^+$ , which repre**sents a hydrocarbon ion with 3 or more carbon atoms;**  and  $C_xH_yN_z^+$ , which represents a large ion containing **C, H, and N. This allows us to track the essential character of the ionosphere, although detailed information is of necessity lost in this process. Similar pseudo-ions have been used in models of the ionosphere of Jupiter [e.g., Kim and Fox, 1994], and Titan [Ip, 1990; Keller et al., 1992].** 

In our model,  $C_xH_v^+$  is the most important ion over **most of the ionosphere. The major production reactions**  for  $C_xH_y^+$  are reactions of  $C_1$ - and  $C_2$ - hydrocarbon ions with acetylene, C<sub>2</sub>- hydrocarbon ions with methane and a few reactions of C<sub>1</sub>- or C<sub>2</sub>- hydrocarbon ions with eth**ylene and ethane. Some of the more important reactions are listed below:** 

$$
CH_3^+ + C_2H_2 \to C_3H_3^+ + H_2 \qquad (R13)
$$

$$
C_2H_4^+ + C_2H_2 \to C_3H_3^+ + CH_3 \qquad (R14)
$$

$$
C_2H_4^+ + C_2H_2 \to C_4H_5^+ + H \qquad (R15)
$$

$$
C_2H_5^+ + C_2H_2 \rightarrow C_3H_3^+ + CH_4 \qquad (R16)
$$

$$
C_2H_5^+ + C_2H_2 \to C_4H_5^+ + H_2 \qquad (R17)
$$

$$
C_2H_2^+ + CH_4 \rightarrow C_3H_5^+ + H \qquad (R18)
$$

$$
C_2H_3^+ + CH_4 \rightarrow C_3H_5^+ + H_2 \qquad (R19)
$$

$$
C_2H_4^+ + C_2H_4 \to C_3H_5^+ + CH_3 \qquad (R20)
$$

$$
C_2H_5^+ + C_2H_4 \rightarrow C_3H_5^+ + CH_4 \qquad (R21)
$$

$$
CH_3^+ + C_2H_6 \rightarrow C_3H_7^+ + H_2 \qquad (R22)
$$

Although  $C_xH_y^+$  does not represent a single ion, the computed peak density of HCNH<sup>+</sup> is almost an order **of magnitude less than the total electron density. Thus H CNH + is clearly less important than other models have found it to be.** 

#### **3. Discussion and Conclusions**

**The ionization peak on Titan appears roughly at the homopause, which is near 1050 km in the Strobel [1992] model. There the major neutral species are N2 and CHq. Although hydrocarbon and related ions comprise only a small fraction of the ions produced by photoion**ization and photoelectron-impact ionization, our model

**shows that such ions may be abundant none-the-less near and below the major ion peak, as well as in the topside ionosphere. The major ions produced in any ionosphere will be transformed by ion-molecule reactions if they are formed in the presence of sufficient densities of neutral species with which they can react. In reducing environments, ionization flows from species whose parent neutrals have smaller proton affinities to species whose parent neutrals have larger proton affinities.** 

**A major uncertainty in the ionospheric model is the accuracy of the neutral model, including the density profiles computed by Yung et al. [1984]. We note here that the crucial C4H2 density was underestimated in the middle atmosphere in the Yung et al. model compared to measured values. It seems that, if anything, the actual values will be larger than those we have as**sumed. Moreover, the conclusion that  $HCNH<sup>+</sup>$  does not **dominate the ion density profile depends only on the presence (with a large enough density) of some species that has a higher proton affinity than HNC or HCN, or of some other photochemical pathway to the protonated species. With a fairly large thermospheric mixing ratio of methane, and the presence of solar ultraviolet photons, that is all but certain.** 

**Another interesting result of the calculations is that hydrocarbon ions dominate the ionosphere at high altitudes as well as at the ionospheric peak. The exobase**  of Titan is near 1500 km, where both  $C_2H_5^+$  and  $C_xH_5^+$ **have densities that are on the order of several hundred cm -3. Although the processes are not included in our model, ions above the exobase may escape into Saturn's magnetosphere. These processes have been investigated by Keller and Cravens [1994] using earlier**  chemical models that predicted HCNH<sup>+</sup> to be the major ion. The results presented here suggest that  $C_2H_5^+$ and  $C_xH_v^+$  may also be the dominant ions supplied to **Saturn's magnetosphere.** 

**Acknowledgments. Partial support of JLF has been provided by NASA grants NAGW-2958 and NAGW-2483 to the State University of New York at Stony Brook. RVY**  acknowledges support from the Cassini project, and NASA **grant NAGW-4923.** 

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**<sup>(</sup>Received May 7, 1997; accepted June 30, 1997.)**