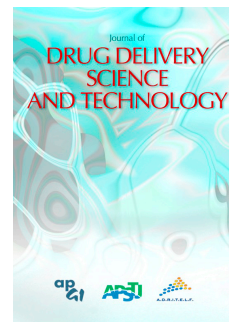


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The influence of surfactant on the properties of albendazole-bile salts particles designed for lung delivery

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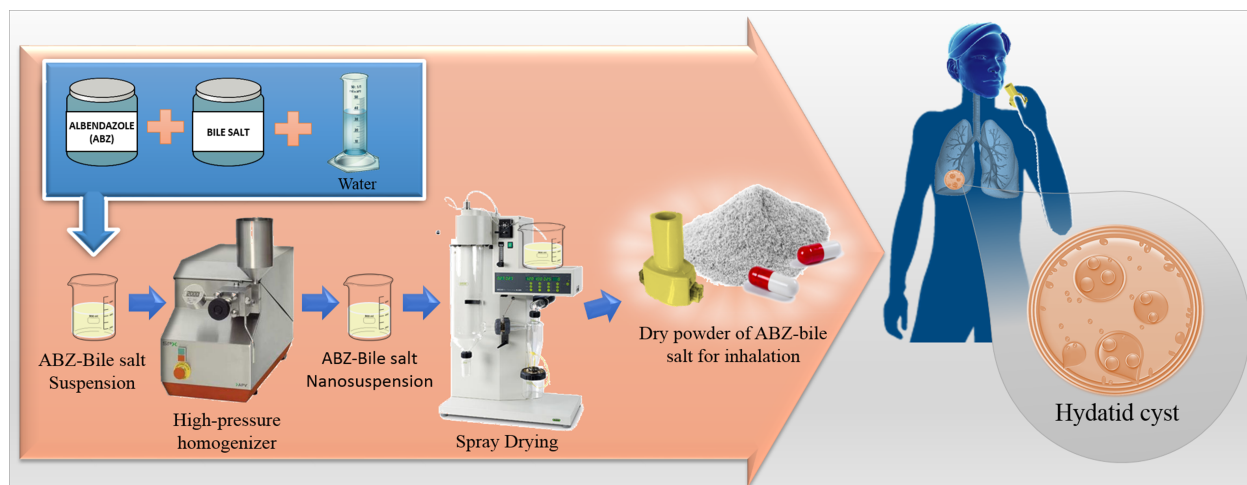
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25 Abstract

26 Albendazole is a first line drug for the treatment of several parasitic diseases in humans.
27 Some parasites target the lungs, however lung delivery of albendazole has so far not been
28 reported. We have developed albendazole-based powders suitable for pulmonary
29 delivery, and studied the impact of surfactant on the formulation properties. High-
30 pressure homogenization followed by spray drying was used to produce inhalable
31 particles of albendazole containing the bile salts sodium taurocholate and sodium
32 glycocholate. The process resulted in porous microparticles, that exhibited good spray
33 drying yield (> 50%), low moisture contents (< 1%) and aerodynamic $D_{90} < 5 \mu\text{m}$. The
34 particles showed adequate aerosolization performance either for normal conditions
35 (respiratory fraction > 71%) or conditions that simulate decreased respiratory capacity
36 (respiratory fraction > 49%). The powders did not disturb lung surfactant function. The
37 comparison between both formulations has revealed that the properties of the surfactants
38 affect mainly the particle size of the suspensions and the porosity of the powders. The
39 higher porosity of the albendazole-sodium taurocholate powder led to an enhanced
40 aerodynamic performance of the formulation compared to albendazole-sodium
41 glycocholate. The developed albendazole powders may pave a way for local lung
42 treatment of parasitic diseases.

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50 Keywords

51 Nanoaggregates; high-pressure homogenization; spray drying; sodium taurocholate; sodium
52 glycocholate, dry powder inhaler.

53 1. Introduction

54 Hydatidosis or human cystic echinococcosis (CE) is a cosmopolitan zoonotic disease that
55 can be found in people and livestock infected with the larval stage of the nematode
56 *Echinococcus granulosus* [1]. In humans, the outcome of this infection is cyst
57 development in different organ systems, with the liver (65%) and the lungs (25%) being
58 the most commonly affected organs [2, 3]. Since lungs have high elasticity, pulmonary
59 cysts grow very fast [4].

60 Albendazole (ABZ) is the drug of choice for chemotherapeutic treatment of CE [5, 6]. It
61 is known that the success of the chemotherapeutic treatment of CE depends on the
62 capacity of the drug to access to the hydatid cyst at adequate concentrations for sufficient
63 periods of time [7]. Since drugs administered by the oral route, once absorbed, access to
64 the liver through the portal vein to be metabolized, this route is useful for treatment of
65 hepatic hydatid cysts. Treatment of pulmonary CE by local administration of ABZ
66 through the pulmonary route has so far been an unexplored alternative. There are several
67 advantages of delivering drugs directly to the lungs, it allows the use of lower doses than
68 those required orally, while it reduces possible side effects and, at the same time, high
69 local concentrations are reached at the site of action [8].

70 Designing suitable aerosols for inhalation is challenging. Particle size is one of the most
71 important characteristics in dry powder inhaler formulations, together with shape,
72 porosity, density, electrical charge and hygroscopicity [9]. In order to reach the lower
73 respiratory tract and optimize pulmonary drug deposition, the formulations need to have
74 aerodynamic diameters between 0.5 and 5 μm [10]. Once in the respiratory parts of the
75 lungs, particles can interact with components of the lung surfactant (LS) film and this
76 interaction depends on the physical properties of the particles [11].

77 A widely used method to produce particles for pulmonary delivery is spray drying (SD)
78 [9]. One of the main advantages of this technique is that the characteristics of the
79 resulting particles can be controlled by adjusting the formulation and process parameters
80 [12, 13]. This scalable technology is able to process a variety of liquids. If the water

81 solubility of drugs is very low preparation of solutions using organic solvents or aqueous
82 suspensions are some of the available alternatives. The main disadvantage of organic
83 solvents is their toxicity related to the presence of residual solvents in the final product,
84 therefore, aqueous suspensions are preferred [9].

85 Since suspensions are unstable heterogeneous mixtures, they must be stabilized. One
86 way to stabilize suspensions is by using surfactants. In the literature on pulmonary
87 delivery, bile salts, such as salts of cholate, deoxycholate, glycocholate,
88 glycodeoxycholate, taurocholate and taurodeoxycholate are used [14-21]. Bile salts have
89 other important roles in drug absorption and the aerodynamic properties of powders [14].
90 Therefore, the selection of the surfactant is a relevant formulation factor.

91 During the spray drying process, solid drugs in suspension aggregate and form dried
92 particles with a larger size than the starting material [9]. Hence, in order to obtain
93 microparticles suitable for pulmonary deposition by SD, suspensions need to be
94 previously processed to obtain particles in the nanometric range (e.g. by high-pressure
95 homogenization, HPH).

96 As mentioned above, to date, no formulations targeting ABZ to the lung have been
97 reported. Therefore, the present study has addressed the production, characterization and
98 comparison of new dry powder formulations based on albendazole and two different bile
99 salts, namely sodium taurocholate (STC) and sodium glycocholate (SGC), for pulmonary
100 delivery. The novel formulations were obtained by processing ABZ-bile salts
101 suspensions by high-pressure homogenization followed by spray drying. Both
102 formulations were characterized and compared in terms of particle size, surface tension,
103 morphology, powder density, porosity and aerodynamic properties. The effect of the
104 ABZ-bile salt formulations on LS function was also evaluated.

105 The ABZ-bile salt formulations manufactured and characterized in the present article
106 have the potential for filling the therapeutic niche of treating pulmonary parasitic
107 diseases, such as CE.

108

109 **2. Materials and methods**

110 *2.1. Materials*

111 ABZ and lactose monohydrate (both pharmaceutical grade), as well as gelatin capsules
112 number 3 were purchased from Saporiti (Buenos Aires, Argentina). Sieved lactose (+100
113 ASTM mesh) with $D_{10}= 167.8 \mu\text{m}$, $D_{50}= 234.9 \mu\text{m}$ and $D_{90}= 336.4 \mu\text{m}$ was used as a
114 carrier. STC and SGC (both analytical grade), were purchased from Sigma Aldrich (St.
115 Luis, MO, USA). Curosurf was produced by Chiesi (Parma, Italy). The unidose RS01
116 high resistance inhaler (Plastiapae, Milano, Italy) and the multidose Turbuhaler device
117 (AstraZeneca, Gothenburg, Sweden) were used as model inhalers. Ethyl alcohol 96%
118 (Pharmacopoeia quality, Porta, Argentina) and bidistilled water were also used.

119

120 *2.2. Characterization of the ABZ-bile salt suspensions*

121 *2.2.1. Preparation and particle size analysis of suspensions*

122 Aqueous solutions containing either STC or SGC (0.03% w/v) were prepared. ABZ
123 suspensions were obtained by adding ABZ (1% w/v) to the surfactant solution under
124 magnetic stirring for 30 min. The bile salt concentration in the suspension was selected
125 based on stability [22] and safety considerations [17, 21] (for more details, please see
126 Section 1.S in the Supplementary material).

127 After dispersion, the suspensions were subjected to a first size reduction step using a
128 PRO200 homogenizer (Tecnolab, Argentina) at 30,000 rpm (10 min for a sample of 500
129 mL). Then, a second reduction/deagglomeration step using sonication (Cole Parmer
130 ultrasonic cleaner) (10 min for a sample of 500 mL, at a frequency of 40 kHz) was
131 performed. Samples were withdrawn after each step to evaluate particle size
132 distributions.

133 The particle size distributions were determined by using laser diffraction (LA 950V2,
134 Horiba, Kyoto, Japan, Liquid method). For the measurement, 2 mL of suspensions were
135 added to 200 mL of recirculating bidistilled water from the laser diffraction equipment.

136 The average size distribution was reported as median volumetric diameter (D_{50}) and the
137 distribution width was informed as span (see Eq.1).

138

$$139 \quad \text{Span} = \frac{D_{90} - D_{10}}{D_{50}} \quad (1)$$

140

141 In Eq. 1, D_{10} , D_{50} and D_{90} represent the diameters where the 10%, 50% and 90% of the
142 population is below each value, respectively. Span values below 2 indicate relatively
143 narrow distributions [23].

144

145 2.2.2. Surface tension determination

146 In order to analyze the effect of the type of bile salt over the surface tension of the
147 suspensions, the surface tension of bidistilled water, STC solution (0.03% w/v), SGC
148 solution (0.03% w/v), ABZ-STC suspension after homogenization/sonication and ABZ-
149 SGC suspension after homogenization/sonication was determined.

150 The measurements were performed with a ring Krüss tensiometer (Krüss GmbH,
151 Hamburg, Germany). All determinations were carried out in duplicate at a temperature of
152 20.7 ± 0.4 °C.

153

154 2.3. Preparation and particle size analysis of nanosuspensions

155 The suspensions were further processed by HPH technique (APV homogenizer, Soeborg,
156 Denmark) to obtain nanosuspensions. The coarse suspensions (Section 2.2.1) were
157 subjected to HPH for 10, 20 and 30 cycles at 800 bar. By using an external heat
158 exchanger, the suspension temperature was maintained at around 30°C. The intensity-
159 weighted average hydrodynamic diameter and polydispersity index (PdI) of the aqueous
160 nanosuspensions were determined by photon correlation spectroscopy (PCS) using a
161 Zetasizer Nano ZS (Malvern Instruments, UK). For this measurement, the
162 nanosuspensions were diluted with bidistilled water up to an ABZ final concentration of

163 0.25% w/v. Mean particle diameters were reported as “Z-average” diameters. All
164 determinations were carried out in triplicate.

165

166 2.4. *Spray drying*

167 The aqueous nanosuspensions were atomized in a lab-scale spray drier (Mini Spray
168 Dryer B-290, BÜCHI, Flawil, Switzerland) equipped with a high performance cyclone.
169 A two-fluid nozzle with a cap-orifice diameter of 0.5 mm was used. The nanosuspensions
170 were spray dried under constant stirring to keep the samples homogeneous. The
171 following conditions were used during the procedure: air inlet temperature (co-current
172 flow): 110°C; drying air flowrate: 35 m³/h; liquid feed flowrate: 1.5 mL/min and
173 atomization air flowrate: 670 L/h. These conditions were selected based on preliminary
174 studies. The process yield was calculated as the ratio of the powder weight obtained in
175 the product collection vessel with respect to the total solids fed to the system. All
176 determinations were carried out in duplicate.

177

178 2.5. *Characterization of the SD products*

179 2.5.1. *Moisture content*

180 The moisture content of the SD products was measured using a halogen moisture
181 analyzer (MB45, Ohaus, Pine Brook, United States). The determination was carried out
182 using samples of around 500 mg. They were heated up to 80°C until the weight changes
183 were less than 1 mg in 60 s.

184

185 2.5.2. *Particle size analysis*

186 The particle size distribution (PSD) of powders obtained by spray drying (SD ABZ-STC
187 and SD ABZ-SGC) was obtained by laser diffraction (LA 950V2, Horiba, Kyoto, Japan),
188 using the dry powder method. Since the resulting powders were cohesive, the SD
189 samples were dispersed in relative coarse lactose (sample:lactose =1:4) to improve the
190 flow from the feed hopper to the measuring cell [24]. The lactose PSD did not interfere

191 with the PSD of the microparticles obtained by spray drying. The PSDs for ABZ and
192 lactose were also measured in triplicate.

193

194 2.5.3. Scanning electron microscopy (SEM)

195 The morphology of ABZ raw material and the SD powders ABZ-STC and ABZ-SGC
196 was analyzed by scanning electron microscopy (SEM), using a EVO 40-XVP, LEO
197 scanning electron microscope (Oberchoken, Germany). The samples were metalized with
198 a thin layer of gold using a sputter coater (PELCO 91000, TellPella, Canada). The
199 scanning electron microscope was operated at an acceleration voltage of 10 kV.

200

201 2.5.4. Bulk and tap density

202 Bulk and tap density of ABZ raw material, SD powders (ABZ-STC and ABZ-SGC) and
203 lactose were measured by using a 10 mL graduated cylinder. Bulk density (δ_{bulk}) was
204 calculated as the ratio of the weight of the powder poured in the cylinder with respect to
205 the volume occupied by the powder. Tap density (δ_{tap}) was calculated by tapping the
206 cylinder until no measurable change in volume was observed. All determinations were
207 made in triplicate. Carr Index (*CI*) was calculated using Eq.(2) and the results were
208 interpreted according USP classification [25].

209

$$210 \quad CI (\%) = \frac{\delta_{tap} - \delta_{bulk}}{\delta_{tap}} \times 100 \quad (2)$$

211

212 2.5.5. Porosity

213 Pore size distribution of the SD ABZ-STC and SD ABZ-SGC powders was determined
214 in a Autosorb iQ gas sorption analyzer (Quantachrome Instruments, Boynton Beach, FL,
215 USA) by using the Brunauer-Emmet-Teller (BET) method (relative pressure range: 0.05-
216 0.30). Samples were degassed at 50°C for 24 h before analysis. Pore volume
217 distribution was calculated with the Barrett-Joyner-Halenda (BJH) model.

218

219 *2.5.6. Aerodynamic performance*

220 The aerodynamic particle size distribution of SD ABZ-STC and SD ABZ-SGC powders
221 was determined using a Next Generation Impactor (NGI, Coplay, Nottingham, UK) [26,
222 27] equipped with a pre-separator (PS).

223 To ensure an adequate dispersibility of the particles, the samples were mixed with lactose
224 in a ratio 1:2 (SD powder:lactose w/w) and 1:1 (SD powder:lactose w/w). It is known
225 that during inhalation, the drug particles are detached from the surface of lactose, the
226 carrier particles impact in the oropharynx and the upper airways and they are swallowed,
227 while the inhalable drug particles go to the lower parts of the respiratory tract [28].

228 A 3-size gelatin capsule was filled with 80 mg of the powder. The doses used represent
229 3.25% or 5% of the maximum dose administered orally in adults (400 mg twice per day).

230 The filled capsule was placed in a RS01 high resistance inhaler (Plastiaple). The NGI
231 stages were precoated with glycerin to avoid particle re-entrainment.

232 Pressure drops of 2 and 4 kPa across the NGI were assayed, being the circulating air
233 flowrate 42.5 and 60 L/min, respectively. Although USP specifies as a standard test
234 condition a pressure drop of 4 kPa, the formulations have also been tested at 2 kPa to
235 evaluate the effect of the inspiration rate on the aerosolization performance [26]. For
236 both flowrates analyzed, the aerodynamic cutoff diameters of each stage of the NGI were
237 calculated as previously described [25-27, 29].

238 The powders deposited on the NGI stages, inhaler, induction port, mouthpiece adapter
239 and PS were recovered with an appropriate volume of ethanol. The ABZ content in each
240 stage or NGI component was determined by UV spectrophotometry at 217 nm.

241 Experiments were done in triplicate. The emitted fraction (EF), the fine particle fraction
242 (FPF) and respirable fraction (RF) were calculated as follows [30]:

243

$$244 \quad EF\% = \frac{\text{drug mass deposited on induction port, pre-separator and all the NGI stages}}{\text{total drug recovered}} \times 100 \quad (3)$$

245

$$246 \quad FPF\% = \frac{\text{drug mass deposited on stages 3-7 and micro-orifice collector}}{\text{drug mass deposited on induction port, pre-separator and all the NGI stages}} \times 100 \quad (4)$$

247

$$248 \quad RF\% = \frac{\text{drug mass deposited on stages 3-7 and micro-orifice collector}}{\text{total drug recovered}} \times 100$$

249 (5)

250

251 The mass median aerodynamic diameter ($MMAD$, D_{50}), D_{10} and D_{90} as well as the
252 percentage of particles with aerodynamic diameters below $5 \mu\text{m}$, $3 \mu\text{m}$ and $1 \mu\text{m}$ were
253 determined by using polynomial equations that adjusted for the passing volume
254 cumulative distribution (built considering the drug mass collected in NGI-1-7 stages and
255 the micro-orifice collector).

256 The geometric standard deviation (GSD), which represents the spread of an aerodynamic
257 particle size distribution, was calculated as $(D_{84}/D_{16})^{1/2}$, where D_{84} and D_{16} represent the
258 diameters at which 84% and 16% of the drug mass are recovered from the NGI 1-7
259 stages and micro-orifice collector, respectively.

260

261 2.5.7. Constrained drop surfactometer analysis

262 The effects of ABZ-STC and ABZ-SGC powders on the LS function were analyzed by
263 the constrained drop surfactometer (CDS) method (BioSurface Instruments, Honolulu,
264 HI) under the operating conditions previously described by Sørli et al [31-33]. Curosurf
265 (porcine lung surfactant, LS) was diluted in water to achieve a final concentration of 2.5
266 mg/mL. For exposure experiments, a drop ($10 \mu\text{L}$) of the curosurf solution was placed on
267 a pedestal. The drop was compressed/expanded less than 30% of its initial area, at a
268 cycling frequency of 40 cycles/min to simulate the movements of the lung during
269 breathing. For each experiment, a baseline (for the unexposed LS drop) was determined.
270 The powders were continuously introduced into the chamber by a Venturi tube connected
271 to a multidose aerosol device Turbuhaler®. The deposited doses were measured with a

272 quartz crystal microbalance (Vitreocell, Waldkirch, Germany) in real time. The dose was
273 increased from 0 to approximately 11 μg of ABZ-STC and ABZ-SGC powders per mg of
274 LS. The minimal surface tensions of the drop before and during exposure were analyzed
275 for the whole range of doses using ADSA (Axisymmetric Drop Shape Analysis) [33].
276 The experiments were performed five times (N=5).

277

278 2.6. Statistical analysis

279 Statistical analysis was performed using multiple t-tests, using GraphPad Prism software
280 v6.05 (GraphPad Software Inc, San Diego, Ca, USA). *p*-values lower than 0.05 were
281 considered statistically significant. Data represent the mean value \pm standard deviation.
282 The number of repetitions of the experiments is indicated in each section of Material and
283 methods.

284

285 3. Results and discussion

286 3.1. Characterization of the suspensions of ABZ with bile salts

287 Table 1 shows information about the particle size distributions of the raw ABZ powder
288 and the suspensions before and after the homogenization and sonication processes. ABZ,
289 as supplied, presented a median size of 76.31 μm . By producing a suspension of ABZ
290 with bile salts, an important decrease in the D_{50} value of ABZ could be observed.
291 Interestingly, the D_{50} of the initial ABZ-STC suspension was about 3 times larger than
292 the D_{50} exhibited by the initial ABZ-SGC suspension, suggesting that SGC is more
293 effective as surfactant than STC. A similar behavior was found for suspensions of
294 tobramycin with STC and SGC [34].

295 To reduce particle size, homogenization and sonication of the dispersions were
296 performed prior to HPH (Table 1). The homogenization process produced a 52%
297 reduction in the D_{50} of the ABZ-STC system, while it did not change the D_{50} of ABZ-
298 SGC suspension. On the other hand, sonication significantly reduced (around 50%) the
299 particle size of both suspensions, obtaining D_{50} of 11.87 and 6.08 μm for the ABZ-STC

300 and ABZ-SGC systems, respectively. The span values, were close to or higher than 2,
301 indicating that all the particle size distributions were dispersed before HPH process.
302 In order to confirm the higher capacity of SGC as surfactant, the surface tensions (ST) of
303 water, the solutions of STC and SGC in water, and the ABZ-STC and ABZ-SGC
304 suspensions were measured. The initial surface tension of water was 73.3 mN/m. By
305 adding SGC, the surface tension of the solution decreased to 54.0 mN/m, whereas the
306 addition of STC reduced the water surface tension to 63.4 mN/m. These results
307 confirmed that SGC is more efficient at reducing the surface tension of water than STC.
308 When the surface tension of the solutions of STC and SGC in water and the suspensions
309 of ABZ containing SGC or STC were compared, an increase in the surface tension was
310 observed for the ABZ-SGC suspension (64.4 mN/m), while the value remained almost
311 unchanged for the ABZ-STC suspension (67.9 mN/m). These results suggest that the
312 SGC is interacting strongly with the ABZ particles in suspension [35], which explains
313 the increase in the water surface tension, while the interaction of STC with the ABZ
314 particles is low, leading to a similar water surface tension in the ABZ suspension.
315 It is well known that the amidation of the carboxyl group with amino acids, such as
316 glycine or taurine, contributes to changes in hydrophilic-lipophilic balance (HLB) of bile
317 salts. Thus, tauroderivatives are more hydrophilic than glycoderivatives, which are more
318 hydrophilic than the original salts [36]. Housaindokht et al.[37] synthesized
319 nanostructured inorganic particles with various surfactants, having different HLB values
320 and they found that particle size and PSD increased with increasing surfactant HLB
321 values. These results are in agreement with our data, which showed that the suspension
322 containing SGC has a significantly lower particle size than the suspension stabilized with
323 STC (with higher HLB) during the whole process.

324

325 *3.2. Influence of the HPH processing on the particle size of the nanosuspensions*

326 After homogenization and sonication, the suspensions of ABZ-SGC and ABZ-STC were
327 processed by HPH and particle size distributions were analyzed as a function of the

328 applied homogenization cycles (Section 2S and Table S1, Supplementary material). After
329 10 homogenization cycles (considered appropriate to produce the ABZ
330 nanosuspensions), the Z-average diameters were 454 and 490 nm for ABZ-SGC and
331 ABZ-STC, respectively. Results from other authors showed that after HPH process, SGC
332 was more suitable to stabilize tobramycin nanosuspensions, compared with STC, leading
333 to nanosuspensions of lower particle size [34]. In the case of our formulations, although
334 the particle size of the ABZ-SGC nanosuspensions tended to be smaller compared to
335 ABZ-STC in all the homogenization cycles performed, the differences were not
336 statistically significant.

337

338 *3.3. Spray-drying process yield and outlet air temperature*

339 The nanosuspensions were then processed by spray drying. Table 2 shows the SD yield
340 and outlet air temperature for both systems. The outlet air temperature was
341 approximately 70 °C, well below the degradation temperature reported for ABZ (140 °C)
342 [38]. FT-Infrared Spectroscopy results confirmed that processing did not affect the
343 chemical stability of the drug (Section 3.S, Supplementary material). The process yield
344 was 50% for SD ABZ-STC and 60% for SD ABZ-SGC. The statistical analysis has
345 shown that the differences in the yields of the formulations are not significant and that
346 they can be considered acceptable for lab-scale driers [39].

347

348 *3.4. Characterization of the SD powders*

349 *3.4.1. Particle size distribution and powder moisture*

350 Table 2 presents the moisture content, D_{10} , D_{50} , D_{90} and span of the SD ABZ-STC and
351 SD ABZ-SGC powders. Considering that a moisture content of 5% w/w is acceptable for
352 inhalation powders [40], the combination of the selected operating conditions and
353 formulations allowed the production of particles with very low water contents (< 1%).
354 The comparison between formulations shows that the moisture content of SD ABZ-SGC
355 is significantly lower than SD ABZ-STC powder. According to Seville et al., a reduction

356 in the ST decreases the moisture content of the dried particles due to the greater
357 permeation of heat into the atomized liquid feed [8]. Our results (Section 3.1) showed
358 that solutions and suspensions developed with SGC have lower surface tension than
359 solutions and suspensions of STC. Thus, the lower moisture content of SD ABZ-SGC
360 formulations compared with SD ABZ-STC can be due to the higher capacity of SGC to
361 reduce the ST of ABZ formulations.

362 The SD ABZ-SGC and ABZ-STC powders also presented suitable geometric median
363 diameters (lower than 4 μm) for pulmonary administration. The differences in the D_{50}
364 values of the SD ABZ-SGC and SD ABZ-STC powders were not statistically significant.
365 According to the span values, the SD ABZ-STC particle size distribution could be
366 considered narrow, while it was slightly more disperse for the SD ABZ-SGC.

367

368 *3.4.2. Morphology*

369 The morphology of the ABZ raw material and SD powders (ABZ-STC and ABZ-SGC)
370 was analyzed by scanning electron microscopy (SEM) (Figure 1).

371 As shown in Figure 1A, the ABZ raw material exhibited a laminar structure [60]. The
372 ABZ raw material presented some particles that were much larger than those observed
373 for the SD powders. This is in agreement with data obtained by laser diffraction for the
374 ABZ raw material ($D_{50} = 76.31 \mu\text{m}$, Table 1). On the other hand, the SEM micrographs
375 of the SD ABZ-STC and SD ABZ-SGC powders (Fig. 1B and 1C, respectively) suggest
376 that the particles were agglomerates of small particles with a final size of approximately
377 4 μm , also in agreement with the previous laser diffraction results of the SD ABZ-STC
378 and SD ABZ-SGC powders ($D_{50} = 3.72 \mu\text{m}$ and 3.86 μm , respectively, Table 2). In
379 addition, the SD ABZ-STC and SD ABZ-SGC particles seemed to present pores, due to
380 the agglomeration of the nanoparticles.

381

382 *3.4.3. Flow properties*

383 A common indicator of powder flowability is the Carr's compressibility index (*CI*). The
384 flow of ABZ raw material, SD ABZ-STC, SD ABZ-SGC and lactose was evaluated by
385 means of the *CI*. According to USP, *CI* values higher than 30 indicate poor powder
386 flowability and values lower than 25 indicate good flow characteristics [25]. As shown
387 in Table 3, the *CI* of ABZ raw material was about 35%, exhibiting poor flow properties
388 according to the USP. The SD powders also showed poor flowability properties (*CI*
389 values of about 31% and 28% for SD ABZ-SGC and SD ABZ-STC systems,
390 respectively).

391 The two main factors that generally affect powders flowability are particle shape and
392 size. Spherical particles ($> 50 \mu\text{m}$) with smooth surfaces often flow better than rough
393 non-spherical particles [41]. In addition, when particle size decreases, the flowability of
394 the powder is reduced due to interparticle forces (primarily the Van der Waals attractive
395 forces), which become larger in relation to gravitational and drag forces. Therefore, fine
396 powders with median diameters smaller than $30 \mu\text{m}$ often exhibit poor flow and a
397 tendency to agglomerate [42]. Since particles suitable for pulmonary delivery have to be
398 very small, they are usually cohesive. Although the SD ABZ-STC and SD ABZ-SGC
399 particles were more rounded than the raw material, the SD formulations presented
400 particle sizes of $3.72 \mu\text{m}$ and $3.86 \mu\text{m}$ respectively (Table 2) and the poor flowability of
401 the tested powders was expected.

402 Dry powder inhalers are usually formulated as a powder mixture of coarse carrier
403 particles [43] to improve drug particle flowability, thus improving dosing accuracy and
404 minimizing the dose variability observed with drug formulations alone [28]. Hence, in
405 this work the SD samples were mixed with lactose with excellent flow properties
406 ($CI=3.13$, Table 3).

407

408 *3.4.4. Porosity determination*

409 The surface areas of SD ABZ-STC and SD ABZ-SGC samples were 10.8 m²/g and 8.8
410 m²/g, respectively. These results could be attributed to both particle size and porosity.
411 According to the pore volume distribution, the use of STC as surfactant resulted in
412 enhanced product porosity. Figure 2 shows that both SD particle systems have similar
413 pore size distributions, SD ABZ-STC has pore diameters slightly larger than SD ABZ-
414 SGC. Moreover, SD ABZ-STC has more mesopores (2–50 nm) and macropores (> 50
415 nm) than SD ABZ-SGC [44].

416 It has been reported that when suspensions are dried by SD, agglomerate-like particles
417 are formed. Typically, the transformation of the nanosuspensions into aggregates by
418 spray drying employed pharmaceutical excipients, such as polymers, sugars and
419 surfactants. They act by forming “excipient bridges”, interconnecting the nanoparticles
420 [45]. The evaporation of water from sprayed droplets produces the agglomeration of
421 particles, generating porous structures which enable further moisture evaporation through
422 pores and capillaries [46]. These data agree with our analysis of surface area
423 determination (Figure 2) and with the micrographs (Figure 1), where some pores are
424 visible between the agglomerated crystals in both formulations. Moreover, these results
425 are in accordance with the low moisture levels in the obtained products (Table 2).

426 The differences in porosity found in SD ABZ-STC and SD ABZ-SGC formulations
427 could be attributed to the interaction between ABZ and the surfactants used. In this
428 sense, based on the surface tension results (see Section 3.1), we concluded that SGC
429 showed higher interaction with the suspended ABZ particles than STC. Therefore, SGC
430 could generate more interparticle bridges, which would increase the nanoparticle
431 proximity; and thus, it would decrease the porosity of the final microparticles.

432

433 *3.4.5. Aerosolization performance*

434 In order to analyze the quality of SD ABZ-STC and SD ABZ-SGC for pulmonary
435 application, the aerosolization performance of the particle systems was evaluated by
436 using the NGI cascade impactor.

437 Several articles have demonstrated the effects of STC and SGC as enhancers of the
438 absorption of hydrophobic drugs and peptides-proteins, both through the pulmonary
439 route [17] and other routes of drug administration [18, 47-51]. However, to date, the
440 capability of the bile salts as enhancers of the aerosolization properties has not been
441 exhaustively explored. Li et al. [15] found that if STC was incorporated into non-viral
442 gene therapy formulations prior to SD of the solutions, this resulted in a powder with
443 cavities, with reduced deposition in the throat and stage 1 in the evaluation of *in vitro*
444 deposition of dry powders. The results showed that STC improved the aerodynamic
445 properties by deaggregation of the powder agglomerates [15].

446 Our results for different SD powder:lactose ratios showed that the EFs were above 90%
447 for both SD powders (Table 4). On the other hand, the FPF and RF were significantly
448 higher for SD ABZ-STC than for SD ABZ-SGC. According to these results, the
449 performance of SD ABZ-STC was better than that of SD ABZ-SGC, which is consistent
450 with the data previously reported. Porous particles contain a high void space, producing
451 particles with low density, leading to a low aerodynamic diameter, and consequently,
452 high respiratory fractions [52]. As shown in Figure 2, the SD ABZ-STC presented higher
453 porosity than SD ABZ-SGC, which may explain the differences in the percentages of
454 respiratory fraction of each formulation.

455 For a SD powder:lactose ratio of 1:2, the *MMAD* for SD ABZ-STC was about 1.9 μm ,
456 whereas the *MMAD* for SD ABZ-SGC was about 2.3 μm (Table 5). The differences
457 between the *MMAD* values of both formulations were not statistically significant, but a
458 tendency to lower *MMAD* for the SD ABZ-STC, compared to SD ABZ-SGC, was found.

459 In order to administer a suitable drug dose by the pulmonary route, without significantly
460 decreasing the flow properties, a SD powder:lactose ratio of 1:1 was used. The EF, FPF
461 and RF values (Table 4) and the percentage of particles smaller than 5 μm and 3 μm
462 (Table 5) for both formulations were similar to that obtained with a SD powder:lactose
463 ratio of 1:2. Present results allow us to conclude that the incorporation of a high dose of

464 SD powder does not affect the aerodynamic behavior, and it would be a good approach
465 for the administration of an effective dose of ABZ.

466 For both SD powders and different powder:lactose ratios, the RF values were higher than
467 70% (Table 4), and 90% of the particles presented an aerodynamic diameter below 5 μm
468 (Table 5). The GSD values obtained (lower than 3) showed that the aerodynamic
469 diameter distributions were narrow [53]. Therefore, the formulations could be adequately
470 deposited throughout all the regions of the lungs and effectively reach the lower airways
471 [54].

472

473 *3.4.6. Aerodynamic behavior under special conditions*

474 Dyspnea is a frequent symptom in patients with pulmonary CE [4, 55]. In order to
475 determine the performances of the SD ABZ-SGC and SD ABZ-STC powders under
476 conditions that simulate reduced respiratory capacity, cascade impactor experiments
477 were performed using a low pressure drop of 2 KPa (Section 2.5.6).

478 As shown in Table 4, the RF of SD ABZ-STC:lactose (1:1) and SD ABZ-SGC:lactose
479 (1:1) formulations were of 69.00% and 48.98%, respectively (percentages lower than RF
480 values obtained under normal respiratory conditions). In addition, *MMAD* values and the
481 percentage of particles smaller than 5 μm were similar to those obtained under normal
482 conditions (Table 5). Even though the special conditions affected some aerodynamic
483 parameters, all the values obtained are adequate for ABZ administration by inhalation.
484 On the other hand, in agreement with our previous result under normal pressure drop
485 (Section 3.4.5), the RF of SD ABZ-STC was significantly higher than SD ABZ-SGC.
486 Again, it can be attributed to the higher porosity of SD ABZ-STC than SD ABZ-SGC
487 [52].

488

489 *3.4.7. Effects of ABZ dry powders on LS function*

490 LS plays an important role in ensuring lung functionality [56]. The analysis of the
491 minimum surface tension is a useful tool to study LS damage, given that a significant

492 increase in this parameter is indicative of deterioration in the surface film and could lead
493 to alveolar collapse. It has been proposed that, if the *in vivo* minimum surface tension
494 exceeds 10 mN/m, this may lead to atelectasis [57].

495 In this work, the effects of the SD ABZ-STC and SD ABZ-SGC formulations on LS
496 functionality were studied by constrained drop surfactometer (CDS). Sørli et al. analyzed
497 the applicability of the CDS *in vitro* model as a predictor for lung toxicity *in vivo* both
498 for impregnation products and for potential pharmaceutical enhancers (bile salts,
499 including the two used in the present study) [31, 58]. For impregnation products the *in*
500 *vitro* - *in vivo* correlation demonstrated that all the analyzed products that induced acute
501 respiration toxicity in mice upon inhalation also inhibited the LS function *in vitro*.
502 Therefore, they concluded that this method may greatly reduce the number of animals
503 used for toxicity testing and formulation of new products [58]. The bile salts induced
504 rapid shallow breathing upon inhalation by mice, the concentration that induced rapid
505 shallow breathing was ranked, and this rank was the same as when compared to the
506 concentration that inhibited LS function [31]. The bile salt concentrations tested in the
507 paper by Sørli et al (2018) was much higher than in the present work.

508 Figure 3 shows that when the SD ABZ-STC and SD ABZ-SGC powders were
509 aerosolized and introduced into the CDS chamber, the minimum surface tension
510 increased slightly compared to the baseline values, but they were all well below 10
511 mN/m. Neither formulation affected the LS function at the analyzed doses, which
512 suggests that the SD powders do not disrupt the LS function.

513

514 **4. Conclusions**

515 In the present work, two formulations containing the hydrophobic drug ABZ and the bile
516 salts STC and SGC, for local delivery of ABZ through the pulmonary route, were
517 developed and compared. Both products were obtained by processing aqueous
518 suspensions using simple and scalable techniques, such as high-pressure homogenization
519 and spray drying. Thereby, the use of organic solvents was avoided.

520 Our results demonstrate that although SGC is a better suspension stabilizer than STC, the
521 formulation containing STC presented a better aerodynamic behavior, as a result of the
522 higher porosity of the powder, compared with SD ABZ-SGC. Thus, we conclude that the
523 bile salt type can differently affect the porosity of the nanoaggregates obtained, leading
524 to formulations with different aerodynamic profiles, which is relevant for lung drug
525 delivery applications.

526 Nevertheless, both SD powders presented high yields, low moisture content and high
527 percentages of respirable fraction (under normal and special conditions) and they did not
528 affect the lung surfactant functionality. Thus, both formulations could be attractive
529 strategies to target ABZ for the treatment of parasitic diseases affecting the respiratory
530 system. However, drug dissolution and *in vivo* studies should be performed to establish a
531 correlation between the *in vitro* results obtained in this work and the efficacy of the
532 formulations.

533

534 **Conflicts of interest**

535 The authors report no conflicts of interest.

536

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546

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760

761

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763

764 **Figure captions**

765

766 **Figure 1.** Morphology of ABZ raw material (A) and the SD powders of ABZ-STC (B) and
767 ABZ-SGC (C) by scanning electron microscopy (SEM) at a magnification of 8500 X and 20000
768 X.

769

770 **Figure 2.** Barrett-Joyner-Halenda (BJH) pore size distribution from N₂ adsorption isotherm
771 expressed as pore volume.

772

773 **Figure 3.** Minimum surface tension of the lung surfactant (LS) after administration of the spray
774 drying products ABZ-SGC (A) and ABZ-STC (B) at doses between 1 and 11 ug/mg of LS. Data
775 represent the mean value \pm standard deviation. The experiments were performed five times
776 (n=5).

777

Table 1. Median volumetric diameter (D_{50}) and distribution width ($Span$) of ABZ-STC and ABZ-SGC suspensions (initial suspensions and suspensions obtained after homogenization and sonication operations).

Particles size	Raw material	Suspension ABZ-STC			Suspension ABZ-SGC		
	ABZ	Initial suspension	Homogenization	Homogenization/ Sonication	Initial Suspension	Homogenization	Homogenization/ Sonication
D_{50} (μm)	76.31 \pm 2.99	44.97 \pm 1.55	21.65 \pm 0.16	11.87 \pm 0.79	14.90 \pm 0.12 (**)	15.57 \pm 0.25 (**)	6.08 \pm 0.00 (**)
$Span$	11.4	1.95	2.52	2.28	4.51	4.31	2.53

Asterisks (*) indicate significant differences between D_{50} values of ABZ-STC and ABZ-SGC suspensions at each stage of the process. **, $p < 0.01$.

Table 2. Particle size (D_{10} , D_{50} and D_{90}), outlet temperature, yield and moisture of SD ABZ-STC and SD ABZ-SGC powders.

	Yield (%)	Outlet temperature ($^{\circ}\text{C}$)	Moisture (%)	D_{10} (μm)	D_{50} (μm)	D_{90} (μm)	$Span$
ABZ-STC	50 \pm 10	68 \pm 1	0.70 \pm 0.04	1.10 \pm 0.09	3.72 \pm 0.06	8.38 \pm 0.40	1.96
ABZ-SGC	60 \pm 6	69 \pm 1	0.56 \pm 0.02 (*)	1.06 \pm 0.02	3.86 \pm 0.30	9.56 \pm 0.80	2.20

Asterisk (*) indicate significant differences between moisture values of ABZ-STC and ABZ-SGC SD powders. *, $p < 0.05$.

Table 3. Bulk density, Tap density and Carr Index determinations of ABZ raw material, SD powders and lactose.

Material	ABZ raw material	ABZ-STC	ABZ-SGC	Lactose
δ_{bulk} (g/mL)	0.34 \pm 0.01	0.28 \pm 0.01	0.27 \pm 0.00	0.66 \pm 0.00
δ_{tap} (g/mL)	0.53 \pm 0.02	0.38 \pm 0.01	0.40 \pm 0.01	0.68 \pm 0.01
Carr Index	35.43 \pm 1.54	27.61 \pm 0.83	31.00 \pm 1.41	3.13 \pm 1.48
Flow (Classification)	Poor	Poor	Poor	Excellent

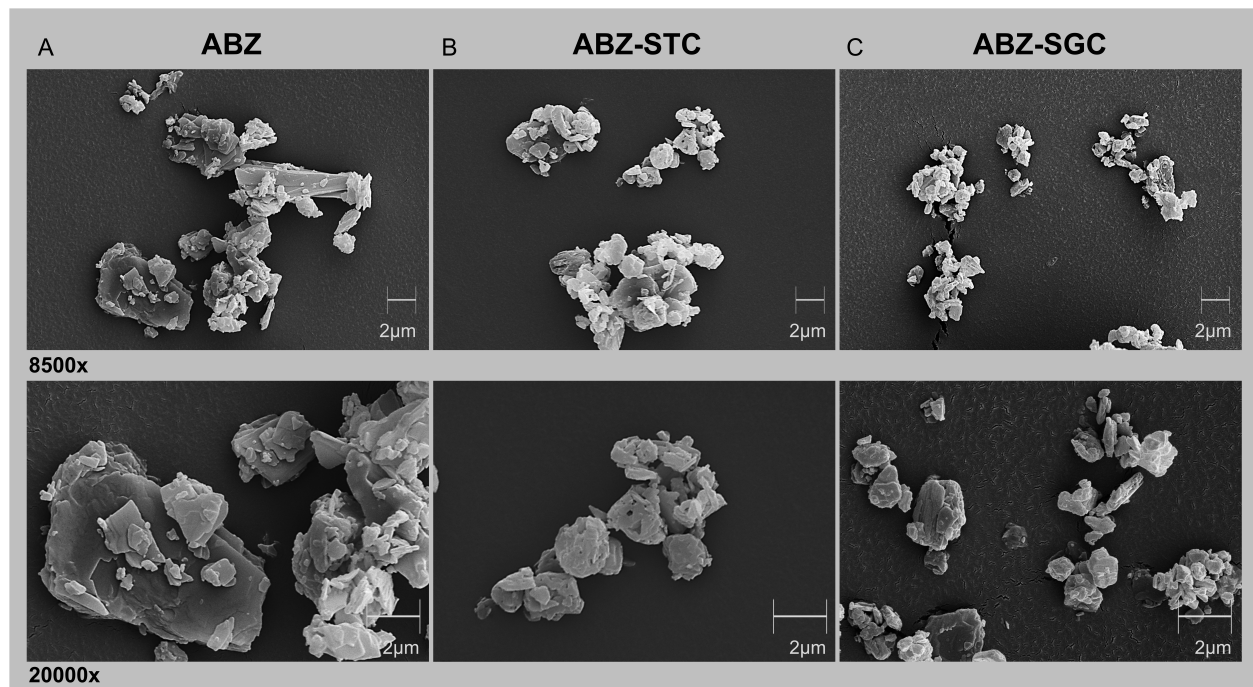
Table 4. Emitted fraction (EF), fine particle fraction (FPF) and respirable fraction (RF) of SD ABZ-STC and SD ABZ-SGC powders under different operating conditions.

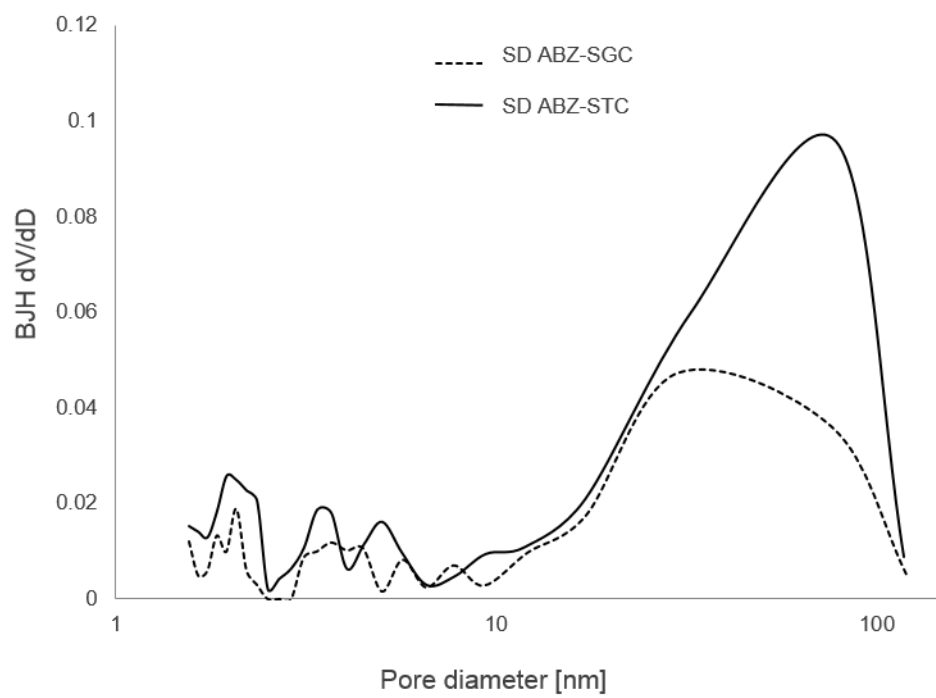
	SD/Lactose ratio	Pressure drop (Kpa)	EF (%)	FPF (%)	RF (%)
ABS-STC	1:2	4	94.13 ± 1.46	83.40 ± 5.36 (*)	78.45 ± 3.83
	1:1	4	96.80 ± 2.40	79.90 ± 0.71 (*)	77.10 ± 1.06 (**)
	1:1	2	93.27 ± 2.41	73.95 ± 2.55 (**)(#)	69.00 ± 4.15 (**)(#)
ABZ-SGC	1:2	4	96.50 ± 0.51	74.20 ± 1.44	70.50 ± 3.45
	1:1	4	93.50 ± 1.08	76.50 ± 1.57	71.50 ± 2.29
	1:1	2	90.75 ± 0.78	53.97 ± 3.72 (###)	48.98 ± 2.94 (###)

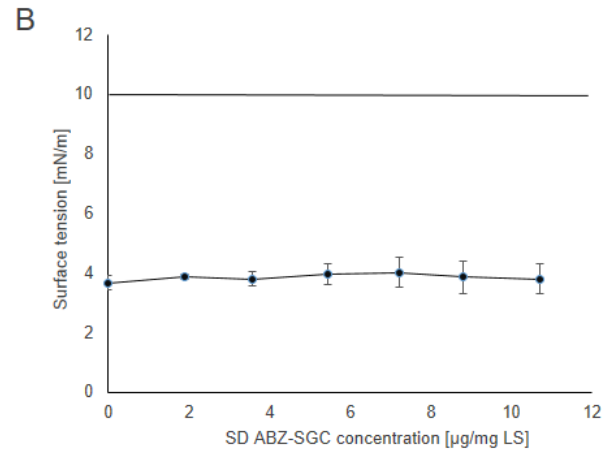
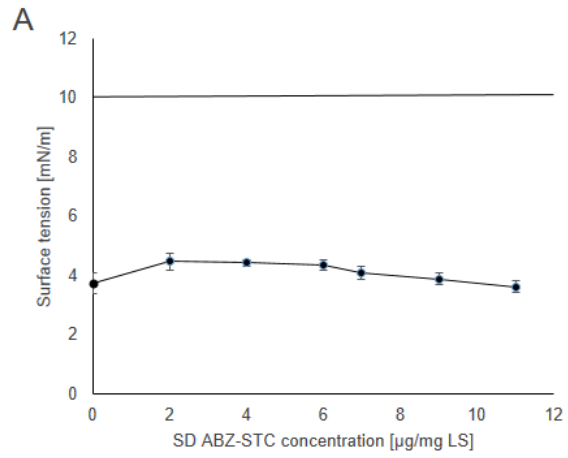
Asterisks (*) indicate significant differences of FPF-RF values from SD ABZ-STC and ABZ-SGC, when comparing the same experimental conditions. Hashtags (#) indicate significant differences of FPF-RF values obtained when comparing the pressure drops of 4 kpa and 2 kpa, for each formulation. One symbol indicates $p < 0.05$; two symbols indicate $p < 0.01$ and three symbols indicate $p < 0.001$.

Table 5. Aerodynamic diameter (D_{10} , D_{50} , D_{90} and D_{99}), GSD and percentage of particles whose size are lower than 5 μm , 3 μm and 1 μm of SD ABZ-STC and SD ABZ-SGC powders.

	SD/Lactose ratio	Pressure drop (Kpa)	D_{10} (μm)	D_{50} (μm)	D_{90} (μm)	D_{99} (μm)	Particles < 5 μm (%)	Particles < 3 μm (%)	Particles < 1 μm (%)	GSD
	1:2	4	0.65 \pm 0.03	1.87 \pm 0.20	4.05 \pm 0.09	8.00 \pm 0.84	91 \pm 3	75 \pm 4	23 \pm 2	2.13 \pm 0.04
ABS-STC	1:1	4	0.80 \pm 0.01	2.21 \pm 0.01	4.35 \pm 0.06	8.02 \pm 0.01	89 \pm 1	67 \pm 3	10 \pm 1	1.99 \pm 0.01
	1:1	2	0.76 \pm 0.00	2.07 \pm 0.01	4.07 \pm 0.07	4.89 \pm 0.15	92 \pm 1	74 \pm 1	10 \pm 1	1.98 \pm 0.01
	1:2	4	0.77 \pm 0.06	2.25 \pm 0.14	4.56 \pm 0.33	7.92 \pm 0.07	90 \pm 1	64 \pm 6	13 \pm 1	2.05 \pm 0.03
ABZ-SGC	1:1	4	0.77 \pm 0.02	2.21 \pm 0.05	4.46 \pm 0.11	7.89 \pm 0.01	89 \pm 2	65 \pm 3	10 \pm 1	2.03 \pm 0.00
	1:1	2	0.75 \pm 0.04	2.20 \pm 0.16	4.96 \pm 0.51	6.59 \pm 0.04	85 \pm 4	65 \pm 2	12 \pm 1	2.15 \pm 0.04







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