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**Impact of sphingomyelin acyl chain (16:0 vs 24:1) on the interfacial properties of Langmuir monolayers: a PM-IRRAS study.**

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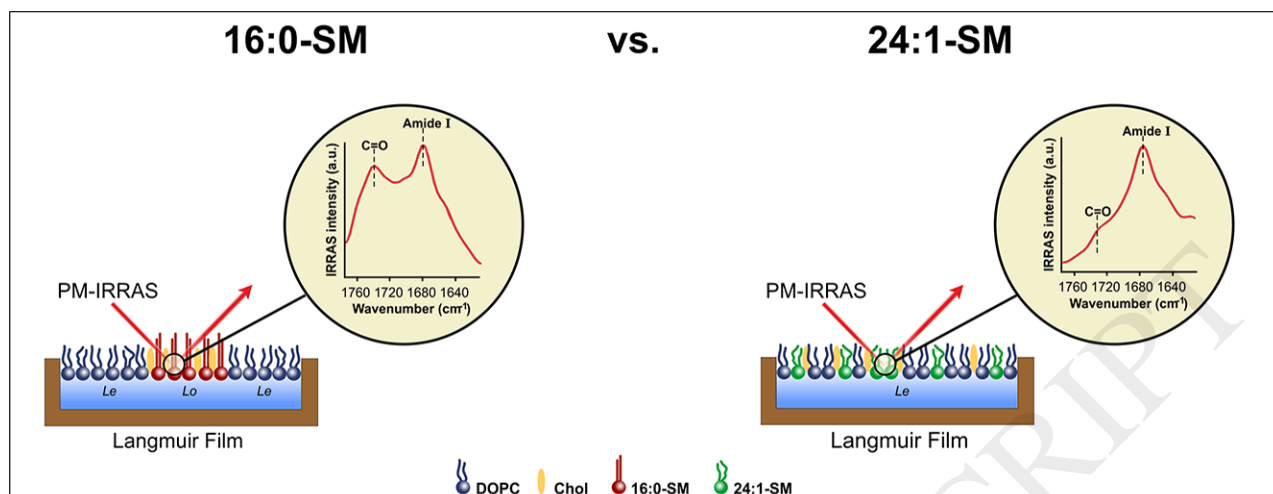
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**Graphical abstract**



## Highlights

- -Sphingomyelin (SM) structure is a key factor on membrane domain formation
- -PM-IRRAS spectra of 16:0 vs 24:1-SM and mixtures with DOPC and Chol were analyzed
- -Acyl chains packing and orientation of polar head groups were affected by the unsaturation in 24:1-SM
- -24:1-SM amide groups remained mostly as non-H-bonded when mixed with DOPC and Chol.
- These findings may play a key role in cell signaling and binding of biomolecules.

## ABSTRACT

Membrane structure is a key factor for the cell's physiology, pathology, and therapy. Evaluating the importance of lipid species such as N-nervonoyl sphingomyelin (24:1-SM) —able to prevent phase separation— to membrane structuring remains a formidable challenge. This is the first report in which polarization-modulated infrared reflection-absorption spectroscopy (PM-IRRAS) is applied to investigate the lipid-lipid interactions in 16:0 vs 24:1-SM monolayers and their mixtures with 1,2-Dioleoyl-sn-glycero-3-phosphocholine (DOPC) and cholesterol (Chol) (DOPC/SM/Chol 2:1:1). From the results we inferred that the *cis* double bond ( $\Delta 15$ ) in 24:1-SM molecule diminishes intermolecular H-bonding and chain packing density compared to that of 16:0-SM. In ternary mixtures containing 16:0-SM, the relative intensity of the two components of the Amide I band reflected changes in the H-bonding network due to SM-Chol interactions. In

contrast, the contribution of the main components of the Amide I band in DOPC/24:1-SM/Chol remained as in 24:1-SM monolayers, with a larger contribution of the non-H-bonded component. The most interesting feature in these ternary films is that the C=O stretching mode of DOPC appeared with an intensity similar to that of SM Amide I band in DOPC/16:0-SM/Chol monolayers (a two-phase [*Lo/Le*] system), whereas an extremely low intensity of the C=O band was detected in DOPC/24:1-SM/Chol monolayers (single *Le* phase). This is evidence that the unsaturation in 24:1-SM affected not only the conformational properties of acyl chains but also the orientation of the chemical groups at the air/water interface. The physical properties and overall H-bonding ability conferred by 24:1-SM may have implications in cell signaling and binding of biomolecules.

**Keywords:** membrane rafts; cholesterol-sphingomyelin interactions; Langmuir monolayers; PM-IRRAS.

## 1. INTRODUCTION

Cellular membranes are highly organized structures that separate the intracellular from the extracellular environment and create discrete compartments within living cells. The structures and functions of these membranes are primarily based on the intrinsic properties of membrane lipids. Current views on structural and dynamical aspects of biological membranes have challenged the idea of lipids forming a randomly organized fluid bilayer matrix [1]. The concept of an organization of the lipid components of membranes into domains has been introduced which constitutes the basis of the lipid raft hypothesis [2]. Sphingomyelin (SM) is one of the key constituents of a lipid raft [3], which is a segregated domain in the cell membrane where SMs, glycosphingolipids, cholesterol (Chol) and specific proteins are selectively assembled. Lipid rafts have been implicated in crucial cellular functions as membrane trafficking, cell signaling, and binding of biomolecules [4]. SM and related sphingolipids are believed to have important functional role(s) on membrane domain formation, *e.g.* for the Chol-rich liquid-ordered (*Lo*) phase [5, 6]. This phase is an intermediate physical state with tight lipid packing while maintaining rotational and lateral lipid mobility, leading to a thicker, but still liquid crystalline membrane [7]. The interaction of Chol with SM and phosphatidylcholine (PC)

has been studied in monolayers and bilayers as membrane models, in which the effects of Chol were found to depend on several factors, including the phospholipid headgroup, chain length, lipid backbone and the presence of saturated or unsaturated chains [8-10]. Chol and SM have been proposed to interact via hydrogen bonding involving the hydroxyl group of Chol and the amide group of the sphingosine, and through hydrophobic interactions between the rigid Chol rings and the SM acyl chains. As a consequence of these distinctive interactions, *Lo* phases enriched in SM and Chol coexist with liquid-disordered (*Ld*) phases in model membranes containing PC, SM, and Chol over a wide range of compositions [11].

Naturally occurring SMs have a phosphocholine headgroup esterified to the C1 hydroxyl group of a long-chain sphingoid base. The most common long chain base in mammalian SM is sphingosine. Typically, the acyl chains amide-linked to carbon 2 of the sphingoid base are long (16–24 carbon atoms) and saturated, but 24:1 has also been recognized as a common N-linked acyl chain in natural SMs [12]. 24:1-SM is the most common unsaturated SM species, accounting for 20–25% of SM in bovine brain and HL-60 cells [13, 14]. Changing levels of 24:1-SM have been linked to processes in the central nervous system such as myelin maturation [15], pointing to the physiological importance of acyl compositional changes in SMs. The influence of 24:1-SM on the lateral organization and mechanical stability of model and natural membranes has been examined [16], where confocal and atomic force microscopy images showed that unlike 16:0-SM and 24:0-SM, 24:1-SM does not induce phase segregation in ternary lipid mixtures with dioleoyl-PC (DOPC) and Chol. It seems that 24:1-SM is able to accommodate both DOPC and Chol in a single phase. The 24:1-SM behavior reinforces the idea by Epanand and Epanand that the driving force behind creation of larger clusters or domains is not only SM-Chol interactions [17]. In fact, the exact nature of the molecular interactions that lead to lipid immiscibilities in membranes is a question partly unresolved. The distinct features in the molecular structure of SM may have substantial effects on its interactions with other membrane components such as sterols and glycerophospholipids. For example, the capacity of SM to form intra- and intermolecular hydrogen bonds has an important role to play in the phase behavior of SMs [18, 19].

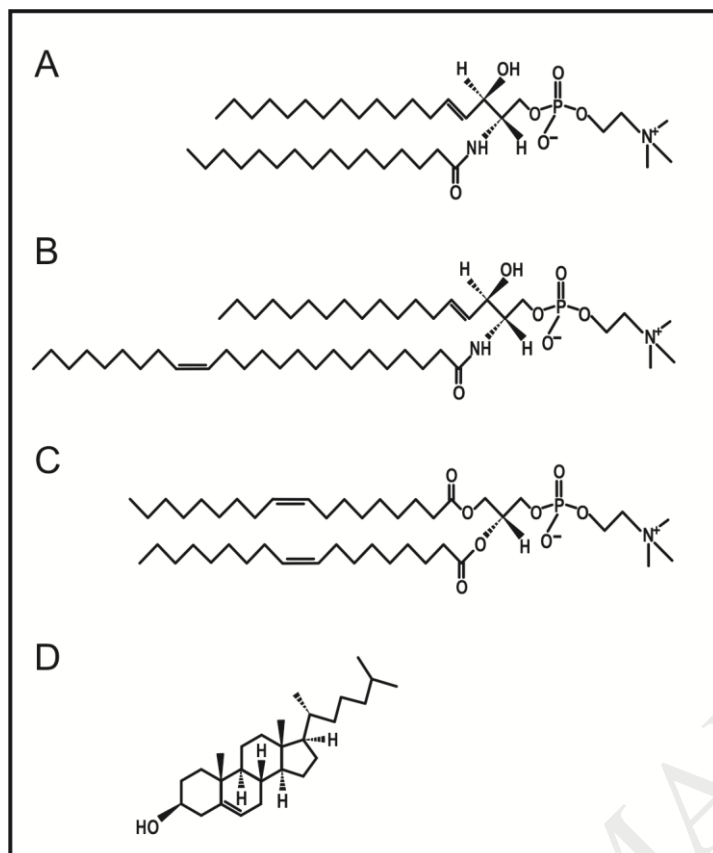
In this study we further characterize SM-SM (24:1 vs 16:0) and SM/PC/Chol monolayers to understand lipid interactions that drastically alter membrane structure and functions.

Specifically, we performed PM-IRRAS measurements on lipid monolayers to characterize the conformational properties and intermolecular interactions in pure lipid films and in two ternary lipid mixtures (DOPC/16:0-SM/Chol and DOPC/24:1-SM/Chol)—which are relevant as mammalian cell membrane models. We analyze the importance of the acyl chain composition of SM in establishing SM-colipid interactions for the formation of membrane domains. PM-IRRAS and BAM results showed that the presence of the *cis* double bond in 24:1-SM molecule affects not only the conformational properties of acyl chains but also the orientation of the chemical groups at the air/water interface, weakening the intermolecular interactions between neighboring molecules and inducing a disordering effect on DOPC/24:1-SM/Chol ternary systems, preventing phase segregation. We advocate that investigating lipid interactions and phase behavior will bring us closer to understanding how domains and rafts are formed in biological membranes. The different properties of the two ternary systems assayed in this work could have significant implications for membrane proteins functions and binding of biomolecules.

## 2. EXPERIMENTAL SECTION

### 2.1- Materials.

1,2-Dioleoyl-sn-glycero-3-phosphocholine (DOPC), N-palmitoyl-D-erythro-sphingosyl phosphorylcholine (16:0-SM), N-nervonoyl-D-erythro-sphingosyl phosphorylcholine (24:1-SM), and cholesterol (Chol) were purchased from Avanti Polar Lipids (Birmingham, AL, USA) and used without further purification. The molecular structures of lipids used in this study are shown in Fig. 1. Tris buffer, NaCl, CaCl<sub>2</sub> and other reagents, all analytical-grade, were acquired from Sigma–Aldrich (St. Louis, MO, USA), unless otherwise stated. HPLC-grade chloroform was purchased from Merck (Darmstadt, Germany). The surface tension and resistivity of the ultrapure water used were 72.2 mN/m and 18.2 MΩ cm at 23 °C, respectively.



**Fig. 1** Chemical structure of lipid molecules used. SMs with a chain length of  $n=16:0$  (palmitoyl-SM) (A) and  $n=24:1$  (nervonoyl-SM) (B), 1,2-dioleoyl-*sn*-glycero-3-phosphocholine (DOPC) (C) and Chol (D). The sphingosine backbone of SM has double-bonded carbons at the 4<sup>th</sup> and 5<sup>th</sup> positions, whilst the fatty acyl-chain is bonded to the backbone by an amide bond. The amide and hydroxyl groups of SM serve as both hydrogen bond donors and acceptors.

## 2.2. Surface-pressure ( $\pi$ ) experiments.

Surface pressure measurements were carried out with a Langmuir trough by NIMA Model 102A (NIMA Technology, Coventry, UK) with a Wilhelmy platinum plate as the  $\pi$  sensor. The aqueous subphase consisted of 20 mM Tris, 150 mM NaCl, 2 mM CaCl<sub>2</sub>, pH 7.4 (TBS Buffer) prepared in ultrapure water. The lipid dissolved in chloroform was gently spread over the subphase surface with a Hamilton microsyringe. After 15 min of solvent evaporation and monolayer equilibration, the films were compressed with two symmetric moving barriers at a compression rate of 5 cm<sup>2</sup>.min<sup>-1</sup>. All the surface pressure measurements were performed at  $23 \pm 1$  °C and repeated at least three times to ensure reproducibility. The in-plane elasticity of the

films was analyzed by calculating the compressibility modulus ( $C_s^{-1}$ ) from the  $\pi$ -Area isotherms according to Eq. 1 [20]:

$$C_s^{-1} = -A (\partial\pi/\partial A) \quad \text{Eq. (1)}$$

where A is the mean molecular area (MMA) at the surface pressure  $\pi$ .

### 2.3. Brewster Angle Microscopy (BAM).

Langmuir monolayers were prepared using a KSV Minitrough equipment (KSV, Helsinki, Finland) mounted on the stage of the microscope. Film imaging by BAM was performed with an imaging ellipsometer (Nanofilm EP3sw imaging ellipsometer, Accurion GmbH, Gottingen, Germany) equipped with a 532 nm laser, a 20 $\times$  objective, and a CCD camera operating at a resolution of 2  $\mu$ m in the BAM mode. Minimum reflection was set with a polarized laser beam ( $\lambda = 532$  nm) incident on the bare aqueous surface at the experimentally calibrated Brewster angle ( $\sim 53.1^\circ$ ). The lipid monolayers were spread onto the aqueous surface and compressed to the desired pressure. After monolayer formation and during compression, the reflected light was collected through a 20x objective to a CCD camera. The gray level at each pixel of the BAM images can be converted to reflectivity values with calibration factors tested for each individual experiment. The reflectivity obtained from BAM measurements is related to both the film thickness and refractive index of the film [21]. Therefore, the liquid-expanded (*Le*) phases of lipid films appear as dark gray areas whilst the more compact and thicker *Lo* phase is observed as light gray domains.

### 2.4. Polarization-Modulated Infrared Reflection-Absorption Spectroscopy (PM-IRRAS) experiments.

PM-IRRAS measurements were performed using a KSV PMI 550 instrument (KSV, Biolin Scientific Oy, Espoo, Finland) with the experimental setup described by Pavinatto *et al* [22]. In the PM-IRRAS technique, the polarized reflectivities for the parallel ( $R_p$ ) and perpendicular ( $R_s$ ) directions to the plane of incidence are measured simultaneously by using an incoming light that is continuously modulated at a high frequency between *s*- and *p*-polarization. The absorption of the *p*-polarized light comes mainly from vertically oriented dipoles, while that of the *s*-polarized light comes from the horizontally oriented ones. A differential reflectivity spectrum  $\Delta R = (R_p -$



$R_s)/(R_p + R_s)$  is then obtained which provides information on the orientation of the vibrational dipoles at the surface. PM-IRRAS measurements were performed at an incident angle of  $80^\circ$ , at which positive bands indicate a transition moment oriented preferentially on the surface plane, whereas negative bands indicate preferential orientation perpendicular to the surface. An average of 600 scans was collected for each spectrum at a resolution of  $8\text{ cm}^{-1}$ . The spectra of the monolayers were normalized to the respective subphase spectra. The lipid monolayers were obtained by spreading the pure lipids or the lipid mixtures (DOPC/16:0-SM/Chol or DOPC/24:1-SM/Chol, at 2:1:1 molar ratio) dissolved in chloroform on the subphase (TBS Buffer) surface. After solvent evaporation and monolayer equilibration, the monolayers were compressed at  $5\text{ cm}^2\cdot\text{min}^{-1}$  until the desired lateral pressure was achieved and PM-IRRAS spectra were collected 15 min after the monolayer equilibration. Measurements were carried out in a class 10,000 clean room at  $23 \pm 1\text{ }^\circ\text{C}$ . Representative PM-IRRAS spectra of two independent experiments are shown.

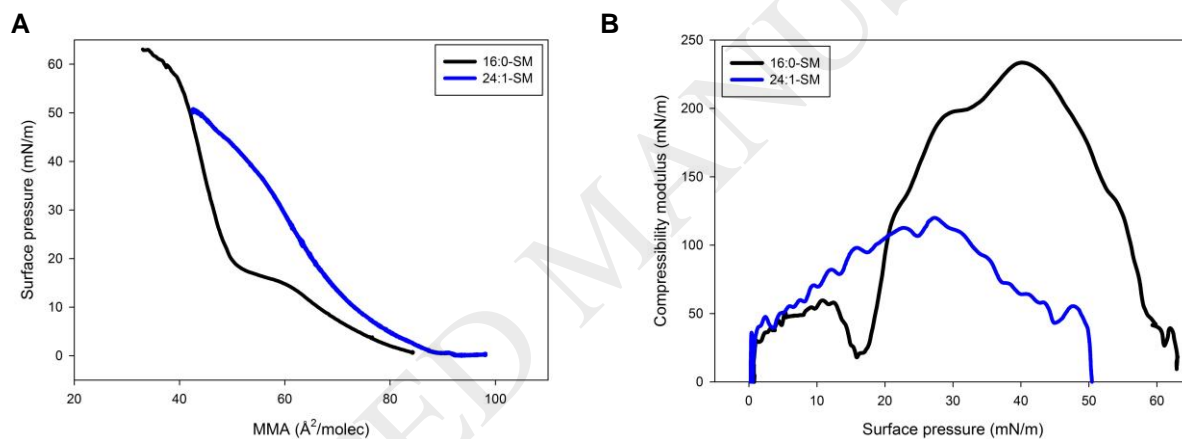
### 3. RESULTS AND DISCUSSION

The main aim in this paper is to investigate by means of PM-IRRAS measurements the influence of unsaturation in the hydrophobic chain of 24:1-SM for its monolayer properties, in addition to comparing the ternary mixtures involving DOPC, Chol and SMs. The results are therefore organized into two subsections, dealing with the neat 16:0-SM and 24:1-SM monolayers and those made with the ternary mixtures. In each subsection, we start the analysis with a full characterization of the monolayer compression isotherms, their stability and rheological properties, so as to link those results to the data obtained from the PM-IRRAS measurements.

#### 3.1. Neat 16:0 and 24:1-SM systems

The  $\pi$ -A isotherms on panel A in Fig. 2 for pure 16:0-SM and 24:1-SM monolayers are similar to those reported in the literature with collapse at  $62.1 \pm 2.0\text{ mN/m}$  and  $50.0 \pm 0.6\text{ mN/m}$ , respectively [23, 24]. The larger MMA values of 24:1-SM along the whole isotherm reflect a lateral packing density significantly lower than for 16:0-SM. Pure 16:0-SM monolayers exhibit a plateau at  $15\text{ mN/m}$  typical of a phase transition, a feature that was not observed for 24:1-SM.

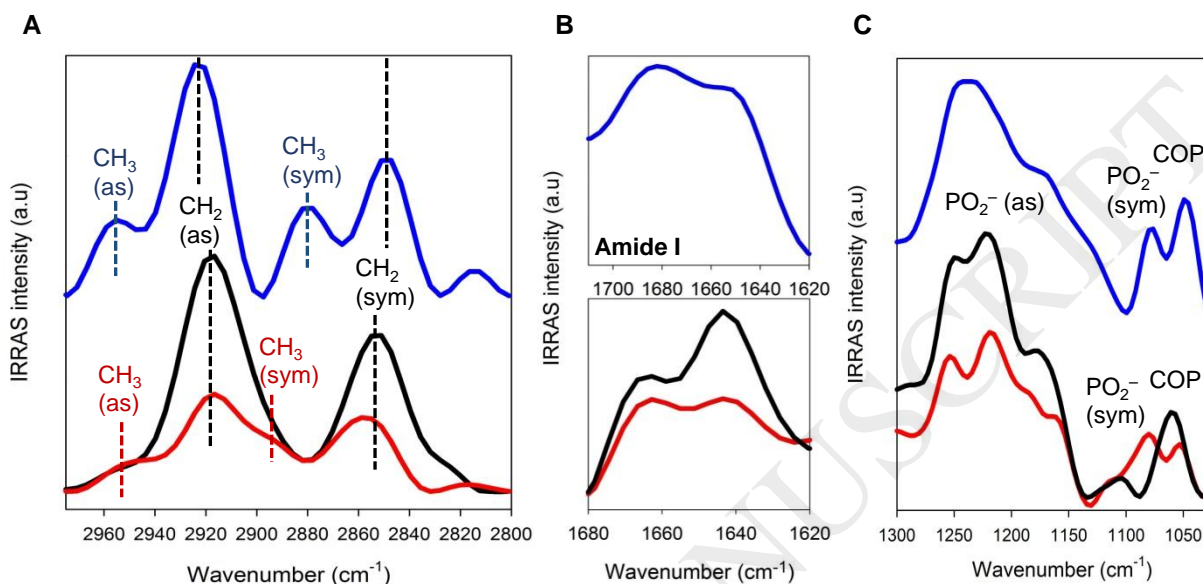
The compressibility modulus ( $C_s^{-1}$ ) provides insights into the lateral packing elasticity, i.e., ease/resistance to lateral compression, within the monolayer. Typical  $C_s^{-1}$  values for *Le* phases are  $\sim 100$  mN/m or lower, while higher values denote liquid condensed (*Lc*) (and *Lo*) phases. The surface elasticity data on panel B in Fig. 2 indicate a transition between *Le* and *Lc* phases at  $\sim 15$  mN/m for 16:0-SM films, whereas 24:1-SM monolayers formed *Le* films with no indication of 2D phase transitions. At high surface pressures, such as those that mimic membrane-like conditions [25],  $C_s^{-1}$  values for 24:1-SM were consistently lower than those of 16:0-SM. At a surface pressure of 30 mN/m,  $C_s^{-1} = 84 \pm 18$  mN/m indicated a *Le* phase state for 24:1-SM monolayer while a  $C_s^{-1} = 203 \pm 22$  mN/m for 16:0-SM corresponds to a *Lc* phase. These results indicate that changes in the acyl composition of SMs markedly affect their monolayer in-plane elasticity.



**Fig. 2.** Compression isotherms of pure SM monolayers. Panel A: Surface-pressure–Area isotherms of 16:0- and 24:1-SM. Panel B: Compressibility moduli ( $C_s^{-1}$ ) corresponding to each isotherm, calculated according to Eq. 1.

The differences between pure 16:0 and 24:1-SM monolayers are likely to be a consequence of a closer lipid packing and higher H-bonding ability with neighboring SMs for 16:0-SM in comparison to 24:1-SM. We performed PM-IRRAS measurements to characterize the conformational properties of these SMs, and Fig. 3 shows the spectra taken at 30 mN/m. For 16:0-SM the spectrum at 10 mN/m was also obtained to analyze the differences in the IR absorption characteristics of 16:0-SM monolayers in the *Lc* (chain-ordered) and *Le* (chain-disordered) phases. In the CH-region of the PM-IRRAS spectra of pure 16:0-SM monolayers in the *Le* phase (10 mN/m) two main bands appeared at 2916 and 2858  $\text{cm}^{-1}$  (Fig. 3, panel A)

assigned to CH<sub>2</sub> antisymmetric and symmetric stretching modes, respectively. Bands assigned to antisymmetric and symmetric stretching of CH<sub>3</sub> also appeared at 2953 and 2887 cm<sup>-1</sup>. Upon



compression to 30 mN/m, the 16:0-SM films (*L<sub>c</sub>* phase) displayed two well-defined, intense peaks at 2916 and 2851 cm<sup>-1</sup> corresponding to CH<sub>2</sub> antisymmetric and symmetric stretching modes, respectively, and a small shoulder at 2953 cm<sup>-1</sup> is due to CH<sub>3</sub> stretching. The large increase in the CH<sub>2</sub>-to-CH<sub>3</sub> intensity ratio correlates with an increased order in the acyl chains when 16:0-SM monolayers underwent the *Le*-to-*L<sub>c</sub>* phase transition. In contrast, the spectra of pure 24:1-SM monolayers at 30 mN/m show CH<sub>2</sub> bands at 2920 (antisymmetric stretching) and 2854 cm<sup>-1</sup> (symmetric stretching) and intense bands at 2954 and 2878 cm<sup>-1</sup> owing to CH<sub>3</sub> antisymmetric and symmetric stretching modes, respectively. As expected, 16:0-SM formed highly ordered lipid monolayers at 30 mN/m while the higher frequency of CH<sub>2</sub> bands and the presence of CH<sub>3</sub> bands in 24:1-SM films spectra are characteristic of conformational disorder in the acyl chains [26-28].

**Fig. 3.** PM-IRRAS spectra of pure SM monolayers. CH region of the PM-IRRAS spectra for 16:0-SM and 24:1-SM films (panel A). Panel B: Amide I region of 24:1-SM (upper panel) and 16:0-SM (lower panel). Panel C: Phosphate regions of 16:0 and 24:1-SM monolayers. The spectra were collected at 10 mN/m (red trace) and 30 mN/m (black trace) for 16:0-SM monolayers and 30 mN/m for 24:1-SM films (blue trace). Measurements were performed at 23 ± 1 °C.

In the Amide I region of 16:0-SM films (Fig. 3B, lower panel), two bands were observed at 1662 and 1643  $\text{cm}^{-1}$  at 10 mN/m (*Le* phase), which could be assigned to non- or weakly-H-bonded and H-bonded amide groups, respectively [29-31]. For 16:0-SM monolayers in the *Lc* phase, the two bands remained but with an increment in intensity of the 1643  $\text{cm}^{-1}$  band, thus indicating a higher level of intermolecular H-bonded lipids at 30 mN/m. Deconvolution of the Amide I band of 16:0-SM indicated a 50% of H-bonded groups at 10 mN/m, whereas this component increased to 66% when monolayers were compressed to 30 mN/m (Figs. S1A and S1B in the *Supplementary Material*). This result is in line with the intensity increase of the high wavenumber component with temperature, along the transition from the gel to the *Ld* phase reported for egg sphingomyelin liposomes [30]. As for 24:1-SM films at 30 mN/m (upper panel in Fig. 3B), the asymmetric Amide I band had two components at 1682 and 1650  $\text{cm}^{-1}$ , with the high wavenumber component associated with non-H-bonded SM being more intense than that of the H-bonded one. After deconvolution of this band, 73% of the amide groups were estimated to remain as non-H-bonded in these films at 30 mN/m (Fig. S1C in the *Supplementary Material*). This remarkable difference in intermolecular H-bonding between 16:0 and 24:1-SM molecules is consistent with calorimetric studies of the hydration of the polar groups of pure SM bilayers [32] where 16:0-SM was found to bind only 5.5 molecules of  $\text{H}_2\text{O}$ /lipid compared to 22/24-SM that binds 9 molecules of  $\text{H}_2\text{O}$ /molecule at comparable levels of hydration.

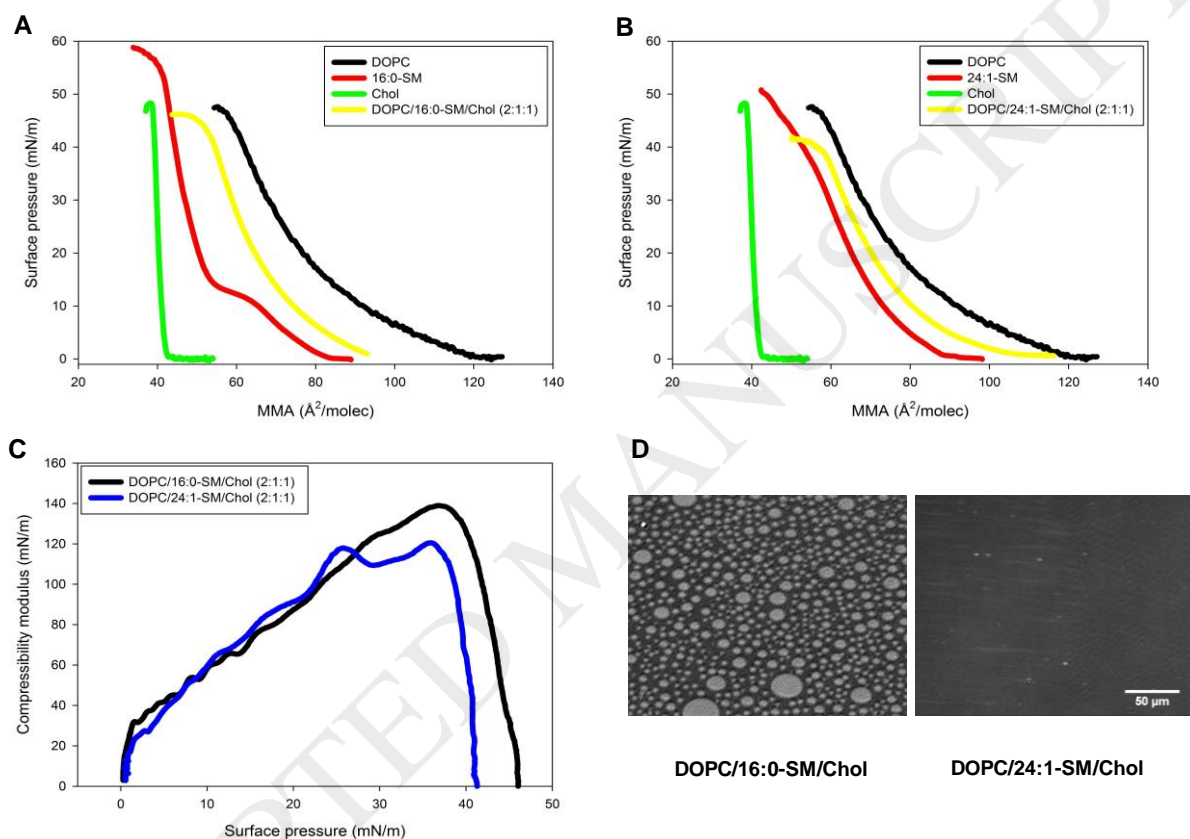
The phosphate group region between 1000 and 1300  $\text{cm}^{-1}$  on panel C in Fig. 3 was also inspected. In agreement with other studies [30, 33] for 16:0-SM, the band at 1230  $\text{cm}^{-1}$  assigned to antisymmetric  $\text{PO}_2^-$  stretching had two components at 1250 and 1219  $\text{cm}^{-1}$  at the two surface pressures studied, corresponding to dehydrated and hydrated  $\text{PO}_2^-$  groups, respectively. The symmetric  $\text{PO}_2^-$  stretching band appeared at 1080  $\text{cm}^{-1}$  at 10 mN/m which did not appear at 30 mN/m, while a small band at 1106  $\text{cm}^{-1}$  remained. The diester phosphate stretching band at 1054  $\text{cm}^{-1}$  for 16:0-SM films in the *Le* phase increased its intensity and shifted to 1061  $\text{cm}^{-1}$  in the *Lc* phase, which could be a consequence of changes in orientation of the phospholipid headgroup [34]. In contrast, 24:1-SM monolayers at 30 mN/m showed a broad band encompassing the two  $\text{PO}_2^-$  antisymmetric stretching at  $\sim 1230 \text{ cm}^{-1}$ , reflecting a more heterogeneous level of hydration of the  $\text{PO}_2^-$  groups. The two main bands in the 1100-1000  $\text{cm}^{-1}$  region were assigned to  $\text{PO}_2^-$  symmetric stretching (1076  $\text{cm}^{-1}$ ) and C-O-P-O-C stretching (1045  $\text{cm}^{-1}$ ).

In summary, the results in Figs. 2 and 3 can be interpreted as follows. The *cis* double bond between carbons 15 and 16 in 24:1-SM molecule diminishes intermolecular H-bonding and chain packing density. This explains differences in transition temperature observed for liposomes from gel to liquid-crystalline phases ( $T_m$ ), which are 41°C and  $23 \pm 3$  °C for 16:0- and 24:1-SM, respectively [35].  $T_m$  of a molecular species of SM is determined by the balance between intermolecular van der Waals and H-bonding interactions, both of which depend on packing constraints in the hydrocarbon tails of the bilayer [36]. Clearly, these should be key factors that determine the miscibility of SM with PC and Chol.

### 3.2. Ternary lipid mixtures

Monolayers of the canonical raft mixture (PC/SM/Chol) were investigated for the two SMs, whose  $\pi$ -Area isotherms are shown on panels A and B of Fig. 4. The isotherms for the neat lipid monolayers are consistent with those in the literature [23, 24]. The Chol monolayer displayed a very low compressibility with steep increase in  $\pi$  while neat DOPC showed a *Le* behavior. The isotherms of ternary lipid mixtures showed *Le*-like behavior as indicated by the low values of  $C_s^{-1}$  (Fig. 4, panel C) with MMA values intermediate between those of SM and DOPC. 16:0-SM-containing mixtures collapsed at lower surface pressures ( $45.8 \pm 0.4$  mN/m) than for neat 16:0-SM ( $62.1 \pm 2.0$  mN/m), at values close to the collapse pressure of DOPC ( $46.2 \pm 0.9$  mN/m). It is worth mentioning that monolayers of DOPC/16:0-SM/Chol exhibit a two-phase (*Lo/Le*) coexistence, with DOPC being the major component of the *Le* phase while a SM- and Chol-enriched *Lo* phase appeared as domains in the ternary mixture [37, 38]. The *Le* behavior obtained for DOPC/16:0-SM/Chol films indicated that monolayer compressibility was not highly influenced by the presence of *Lo* domains; instead, the continuous phase compression properties determined the rheological properties of the two-phase system, a feature that has been reported in other systems where two phases coexist [39]. DOPC/24:1-SM/Chol mixtures formed less stable films, without phase coexistence, which collapsed at lower surface pressures ( $41.3 \pm 0.1$  mN/m) than the pure lipid films, thus indicating unfavorable interactions between the film components. BAM images of the ternary lipid mixtures at 30 mN/m confirmed the presence of phase coexistence in DOPC/16:0-SM/Chol films (Fig. 4, panel D, left image) with a thick phase (light gray) forming circular domains immersed in a thinner phase (dark gray). Previous results

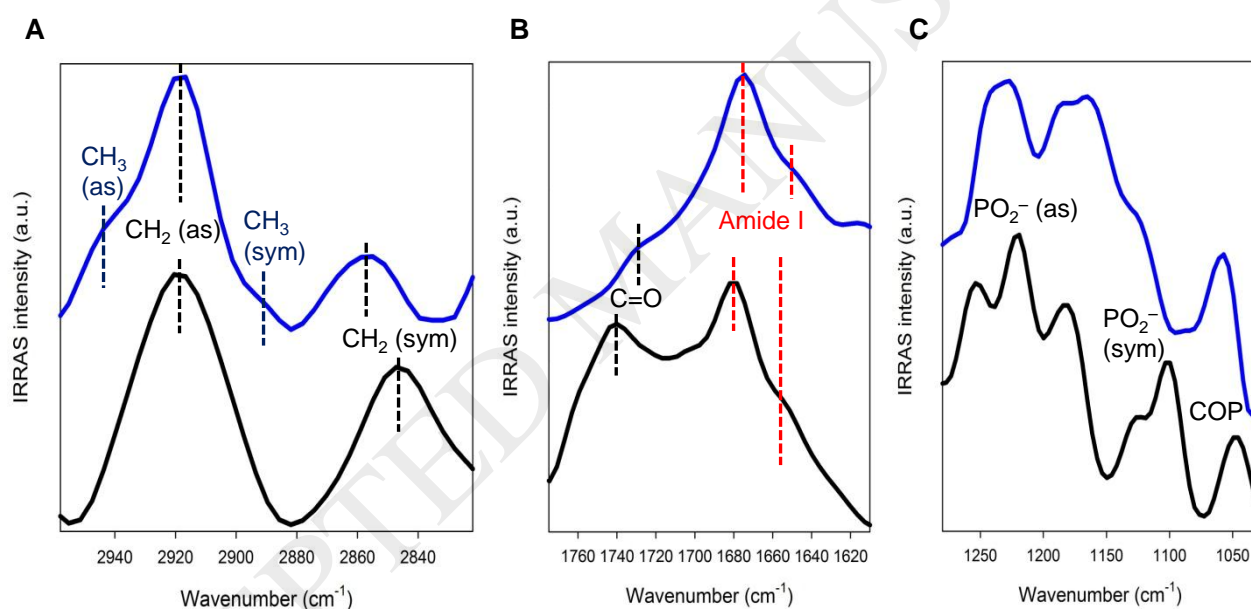
by Fanani *et al.* [38] support the assignment of a *Lo* character to the thick phase and *Le* character to the thinner continuous phase. On the other hand, the right image in Fig. 4, panel D, shows a completely different microscopic organization and texture of the ternary films containing 24:1-SM, characterized by a homogenous (*Le*) phase.



**Fig. 4.** Compression isotherms of ternary lipid mixtures. Surface pressure-Area isotherms for neat lipid components and ternary lipid mixtures of DOPC/16:0-SM/Chol (2:1:1 molar ratio) (panel A) and DOPC/24:1-SM/Chol (2:1:1 molar ratio) (panel B). The compressibility modulus ( $C_s^{-1}$ ) was calculated for each ternary mixture using Eq. 1 (panel C). Panel D: BAM images of DOPC/16:0-SM/Chol (left image) and DOPC/24:1-SM/Chol (right image) monolayers at 30 mN/m. Scale bar is 50  $\mu\text{m}$  for both images.

Considering the phase behavior of both ternary lipid mixtures monolayers at 30 mN/m, their PM-IRRAS spectra shown in Fig. 5 should reflect the presence of both phases (*Le/Lo*) in 16:0-SM-containing monolayers and of a single *Le* phase in 24:1-SM-containing films. The CH region for DOPC/16:0-SM/Chol monolayers presented two main bands at 2920 and 2847  $\text{cm}^{-1}$

assigned to CH<sub>2</sub> antisymmetric and symmetric stretching vibrations, respectively (Fig. 5, panel A). The CH<sub>2</sub> bandwidth for this ternary mixture is larger than for the CH region of pure 16:0-SM owing to the presence of different CH<sub>2</sub> components, as expected for a two lipid-phase system [31]. For DOPC/24:1-SM/Chol films, CH<sub>2</sub> bands are observed at 2916 and 2854 cm<sup>-1</sup>, with CH<sub>3</sub> bands appearing as shoulders at 2943 (CH<sub>3</sub> antisymmetric) and 2890 cm<sup>-1</sup> (CH<sub>3</sub> symmetric stretching). These results are consistent with 24:1-SM-containing monolayers presenting a higher conformational disorder in the acyl chains region, in comparison to 16:0-SM-containing ones. Notwithstanding, the conformational chain order in DOPC/24:1-SM/Chol films was higher than in pure 24:1-SM monolayers, indicating that incorporation of DOPC and Chol has an ordering effect over 24:1-SM.



**Fig. 5.** PM-IRRAS spectra of ternary lipid mixtures at 30 mN/m. CH (panel A), Amide I (panel B) and phosphate (panel C) regions of the PM-IRRAS spectra of DOPC/16:0-SM/Chol (black trace) and DOPC/24:1-SM/Chol (blue trace) films. All of the spectra were collected at a lateral pressure of 30 mN/m. Measurements were performed at  $23 \pm 1$  °C.

The 1800-1600 cm<sup>-1</sup> region contains information regarding carbonyl and amide I groups, where the carbonyl bands relate to the properties of the phosphatidylcholines at the lipid-water interface (in this case, DOPC) while the Amide I band should depend on the SM-water interface [30]. These bands are strong and occur in a region where Chol itself does not have any major

absorption bands. The spectrum for DOPC/16:0-SM/Chol films is markedly different from that of neat 16:0-SM monolayer, featuring an overlap of the two components of the Amide I band at  $1682\text{ cm}^{-1}$  with a shoulder at  $1650\text{ cm}^{-1}$ , which could be due to H-bonded amide groups (Fig. 5, panel B). An intense band at  $1740\text{ cm}^{-1}$  is assigned to C=O stretching of DOPC. The wavenumbers at which the main bands appeared correspond to non-hydrated DOPC C=O groups and mainly non-H-bonded 16:0-SM amide groups. The relative intensity of the low frequency-to-high frequency components of the Amide I band was inverted compared to pure 16:0-SM (*cf.* Fig. 3B). The analysis of the deconvoluted Amide I band indicated that 45% of the main band corresponded to H-bonded groups in the ternary films (Fig. S2A in the *Supplementary Material*). In this ternary mixture, 16:0-SM is packed with Chol forming *Lo* domains. Hence, in addition to SM-SM intermolecular interactions there are also SM-Chol interactions, increasing the heterogeneity of the H-bonding network. This could be the reason for the lower intensity of the H-bonded band, although conformational changes should not be discarded. On the other hand, the *Le* phase is composed mainly by DOPC whose C=O stretching mode had an intensity similar to that of SM Amide I band. This band was shifted to higher wavenumbers compared to the case of neat DOPC monolayers at  $30\text{ mN/m}$  for which C=O vibration was observed at  $1732\text{ cm}^{-1}$  (Fig. S3 in the *Supplementary Material*), indicating that the addition of SM and Chol affected the behavior of DOPC at the interface which C=O groups appeared in a more hydrophobic environment.

In DOPC/24:1-SM/Chol monolayers, an intense Amide I band at  $1674\text{ cm}^{-1}$  with a slight shoulder at  $1647\text{ cm}^{-1}$  was detected. These wavenumbers correspond to non-H-bonded and H-bonded amide groups, respectively. A higher proportion of non-H-bonded groups was confirmed by deconvolution of the amide I band where 77% of the integrated area corresponded to the non-H-bonded component (Fig. S2B in the *Supplementary Material*). A strikingly low intensity of the C=O band of DOPC was observed, detected as a shoulder at  $1728\text{ cm}^{-1}$ . This red shift could reflect formation of H-bonds and/or conformational changes that displace these groups to microenvironments of different polarities. In this context, it is worth mentioning that—as stated in the *Experimental Section*—the intensity of the PM-IRRAS signal is sensitive to the orientation of the vibration moments. Upward bands are observed when the vibration moments lie parallel to the interface. Thus, the low intensity of the C=O band in 24:1-SM-containing films indicates



that the orientation of DOPC polar groups is different from the one adopted in 16:0-SM-containing monolayers. 24:1-SM is able to accommodate both DOPC and Chol, forming a single phase in monolayers and bilayers [16, 24]; therefore, in contrast to what happens in ternary mixtures containing 16:0-SM, in 24:1-SM-containing monolayers no preferential interactions take place between the lipid components. The components are brought together in a single phase, in a process that must involve the adoption of different orientations of the individual lipids compared to the 16:0-SM-containing monolayers.

In the  $\text{PO}_2^-$  region, two well-defined bands ( $1253, 1219 \text{ cm}^{-1}$ ) were observed in DOPC/16:0-SM/Chol ternary monolayers spectrum (Fig. 5, panel C) corresponding to the  $\text{PO}_2^-$  antisymmetric stretching, in a similar trend as for pure 16:0-SM monolayers with a minor shift of the high frequency component. A band at  $1100 \text{ cm}^{-1}$  with a shoulder at  $1119 \text{ cm}^{-1}$  was detected for the symmetric stretching in the ternary mixture with higher intensity than for pure 16:0-SM, while the phosphodiester stretching appeared red-shifted compared to 16:0-SM at  $1050 \text{ cm}^{-1}$ . For DOPC/24:1-SM/Chol monolayers, a band at  $1226 \text{ cm}^{-1}$  is assigned to antisymmetric stretching, with the same contour characteristics as in the 24:1-SM monolayer spectra. Thus, the presence of DOPC did not modify the band shape for the phosphate group of 24:1-SM which may indicate that the phosphate group of DOPC could participate in H-bonding with donor groups, most likely the -OH group of Chol and/or the -OH group of SM, since the Amide I band appeared mainly as non-H-bonded groups [33]. This interaction could explain the conformational changes at the interface in this mixture (*cf.* Fig. 5B). In contrast to pure 24:1-SM monolayers, no clear bands assigned to  $\text{PO}_2^-$  symmetric stretching were detected while the band corresponding to phosphodiester stretching mode appeared shifted to higher frequencies, at  $1057 \text{ cm}^{-1}$ . In both ternary mixtures, the band at  $\sim 1165 \text{ cm}^{-1}$  could be attributed mainly to the ester C-O asymmetric stretching of DOPC [40].

The results from compression isotherms and PM-IRRAS measurements revealed that the *cis* double bond in 24:1-SM produced a loosening of lipid chain packing—as evidenced by the higher area per molecule and acyl chain disorder in the ternary lipid films containing 24:1-SM (*cf.* Fig. 4 and 5)—with a pronounced effect on the level of hydration at the interface. In the interfacial region, the C=O and  $\text{PO}_2^-$ -stretching vibrations are sensitive to the local environment and the frequencies at which these bands appear are influenced by the level of hydration. For

DOPC/24:1-SM/Chol monolayers, the C=O stretching band of DOPC at  $1728\text{ cm}^{-1}$  corresponds to fully hydrated carbonyl groups while in DOPC/16:0-SM/Chol films this band appeared at  $1740\text{ cm}^{-1}$  owing to highly dehydrated C=O groups at the interface. The analysis of the antisymmetric  $\text{PO}_2^-$  bands also revealed a high level of hydration at the interface for 24:1-containing ternary films with the band centered at  $\sim 1226\text{ cm}^{-1}$  appearing due to hydrated  $\text{PO}_2^-$  groups whereas in DOPC/16:0-SM/Chol films a second population of highly dehydrated  $\text{PO}_2^-$  groups was detected as a band at  $1253\text{ cm}^{-1}$ . Thus, the disordering effect caused by unsaturation in 24:1-SM prevented phase segregation in the ternary mixtures leading to the formation of *Le* films with high levels of hydration.

Results obtained from PM-IRRAS measurements also revealed a different orientation of DOPC when mixed in ternary mixtures with Chol and the unsaturated asymmetric 24:1-SM than that observed when 16-SM is present. The extremely low intensity of the C=O band in the 24:1-SM-containing films indicates a reorientation of the C=O transition moments with respect to the surface plane. The orientational properties of the C=O groups reflect a different conformation of the polar region of DOPC molecules in these ternary mixtures compared to that adopted in DOPC/16:0-SM/Chol and neat DOPC films. In the latter, the intense positive C=O stretching bands point to a preferential orientation of C=O transition moments on the surface plane. Therefore, the interaction of DOPC with 24:1-SM and Chol in the single *Le* phase of DOPC/24:1-SM/Chol monolayers seems to force the polar groups of DOPC to adopt a different conformation from that in pure DOPC monolayers or in systems where this PC is the major component of one of the lipid phases—as is the case of DOPC/16:0-SM/Chol mixtures.

#### 4. CONCLUSIONS

The most important conclusion to infer from the surface pressure isotherms and PM-IRRAS spectra is associated with the importance of acyl chain composition of SM in monolayers with components roughly simulating rafts. Not only were the conformational properties of acyl chains and interfacial hydration affected by 24:1-SM unsaturation but also the orientation of the chemical groups at the air/water interface. This implies that—to obtain a more accurate perspective of membrane lipid phase behavior—it is necessary to examine the properties of different molecular species of lipids from biological sources. In the case of 24:1-SM, the

properties conferred by unsaturation have implications for the formation of lipid domains in cell membranes. Unlike 16:0-SM, when 24:1-SM is mixed with DOPC and Chol, a single *Le* phase is obtained in monolayers with 24:1-SM amide groups being mostly non-H-bonded. In this regard, due to their H-bonding capability, sphingolipids constitute important molecules involved in cell signaling events [41] and also serve as receptors for various extracellular ligands. SM, for instance, has been found to constitute the binding site for viral proteins [42] and several pore-forming toxins [43-48] with the H-bonding groups (2-NH and the 3-OH) being important for such binding. Results obtained here indicate that the incorporation of 24:1-SM affects the monolayer physical properties and enhances the overall H-bonding ability. The latter may play a key role in cell signaling and binding of biomolecules. We hope that the approach and results presented here may inspire further studies of membrane proteins that depend on the surrounding lipid environment for optimal function.

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## REFERENCES

1. F.M. Goñi, *Biochim Biophys Acta*, 2014. 1838(6): 1467-1476.
2. K. Simons and G. van Meer, *Biochemistry*, 1988. 27(17): 6197-6202.
3. K. Simons and E. Ikonen, *Nature*, 1997. 387: 569-572.
4. R.E. Brown, *J Cell Sci*, 1998. 111: 1-9.
5. P.J. Quinn, *Prog Lipid Res*, 2010. 49(4): 390-406.
6. J.V. Busto, J. Sot, J. Requejo-Isidro, F.M. Goñi and A. Alonso, *Biophys J*, 2010. 99: 1119-1128.
7. M.C. Giocondi, D. Yamamoto, E. Lesniewska, P.E. Milhiet, T. Ando and C. Le Grimmelc, *Biochim Biophys Acta*, 2010. 1798: 703-718.
8. T. Rog, M. Pasenkiewicz-Gierula, I. Vattulainen and M. Kartunnen, *Biochim Biophys Acta*, 2009. 1788: 97-121.
9. S. Jaikishan, A. Björkbom and J.P. Slotte, *Biochim Biophys Acta*, 2010. 1798: 1987-1994.
10. J. Juhasz, J.H. Davis and F.J. Sharom, *Biochem J*, 2010. 430: 415-423.
11. S. Jaikishan and J.P. Slotte, *Biochim Biophys Acta*, 2011. 1808: 1940-1945
12. J.P. Slotte, *Prog Lipid Res*, 2013. 52: 206-219.
13. J.P.J. Boegheim, M. van Linde, J.A.F. op den Kamp and B. Roelofsen, *Biochim Biophys Acta*, 1983. 735: 438-442.
14. V. Fitzgerald, M.L. Blank and F. Snyder, *Lipids*, 1995. 30: 805-809.
15. F. Babin, P. Sarda, B. Limasset, B. Descomps, D. Rieu, F. Mendy and A. Crastes de Paulet, *Lipids*, 1993. 28: 627-630.
16. S. Maté, J.V. Busto, A.B. García-Arribas, J. Sot, R. Vazquez, V. Herlax, C. Wolf, L. Bakás and F.M. Goñi, *Biophys J*, 2014. 106(12): 2606-2616.
17. R.M. Epand and R.F. Epand, *Chem Phys Lipids*, 2004. 132: 37-46.
18. P.S. Niemelä, M.T. Hyvönen and I. Vattulainen, *Biochim Biophys Acta*, 2009. 1788(1): 122-135.
19. J.P. Slotte, *Biochim Biophys Acta*, 2016. 1858(2): 304-310.
20. G.L. Gaines, *Insoluble Monolayers at Liquid-Gas Interfaces*. Interscience Publishers, 1966.
21. C. Lheveder, J. Meunier, H. Henon and W.N.E. A. Baszkin, *Brewster Angle Microscopy*. Physical Chemistry of Biological Interfaces, Marcel Dekker Inc., NY., 2000.
22. F.J. Pavinatto, C.P. Pacholatti, E.A. Montanha, L. Caseli, H.S. Silva, P.B. Miranda, T. Viitala and O.N. Oliveira, *Langmuir*, 2009. 25(17): 10051-10061.
23. X.M. Li, J.M. Smaby, M.M. Momsen, H.L. Brockman and R.E. Brown, *Biophys J*, 2000. 78: 1921-1931.
24. S.M. Maté, R.F. Vazquez, S.V. Herlax, D.-M. M.A., M.L. Fanani, M.E. Vela, B. Maggio and L. Bakás, *Biochim Biophys Acta*, 2014. 1838: 1832-1841.
25. D. Marsh, *Biochim Biophys Acta*, 1996. 1286: 183-223.
26. R.G. Snyder, A.L. Aljibury and H.L. Strauss, *J Chem Phys*, 1984. 81(12): 5352-5361.
27. R. Mendelsohn, J.W. Brauner and A. Gericke, *Annu Rev Phys Chem*, 1995. 46: 305-334.
28. C.L. Leverette and R.A. Dluhy, *Colloids Surf. A*, 2004. 243: 157-167.
29. M. Rujoi, D. Borchman, D.B. DuPré and M.C. Yappert, *Biophys J*, 2002. 82(6): 3096-3104.
30. Z. Arsov and L. Quaroni, *Biochim Biophys Acta*, 2008. 1778(4): 880-889.
31. B. Boulgaropoulos, Z. Arsov, P. Laggner and G. Pabst, *Biophys J*, 2011. 100(9): 2160-2168.

32. M. Kodama, M. Abe, Y. Kowasaki, S. Ohira, H. Nozaki and C. Katagiri, *Thermochim Acta*, 2004. 416: 105-111.
33. J. Villalaín, A. Ortiz and J.C. Gómez-Fernández, *Biochim Biophys Acta*, 1988. 941(1): 55-62.
34. F.M. Goñi and J.L. Arrondo, *Faraday Discuss Chem Soc*, 1986. 81: 117-126.
35. N. Jiménez-Rojo, A.B. García-Arribas, J. Sot, A. Alonso and F.M. Goñi, *Biochim Biophys Acta*, 2014. 1838: 456-464.
36. P.J. Quinn and C. Wolf, *Biochim Biophys Acta*, 2009. 1788(9): 1877-1889.
37. R.F. de Almeida, A. Fedorov and M. Prieto, *Biophys J*, 2003. 85(4): 2406-2416.
38. M.L. Fanani and B. Maggio, *J Phys Chem*, 2011. 115: 41-49.
39. N. Wilke, F. Vega Mercado and B. Maggio, *Langmuir*, 2010. 26(13): 11050-11059.
40. J.E. Fookson and D.F. Wallach, *Arch Biochem Biophys*, 1978. 189(1): 195-204.
41. N. Bartke and Y.A. Hannun, *J Lipid Res*, 2009. 50 (Suppl.): 91-96.
42. L. Weng, Y. Hirata, M. Arai, M. Kohara, T. Wakita, K. Watashi, K. Shimotohno, Y. He, J. Zhong and T. Toyoda, *J Virol*, 2010. 84(22): 11761-11770.
43. A.W. Bernheimer and L.S. Avigad, *Proc Natl Acad Sci U.S.A.*, 1976. 73(2): 467-471.
44. T. Tomita, K. Noguchi, H. Mimuro, F. Ukaji, K. Ito, N. Sugawara-Tomita and Y. Hashimoto, *J Biol Chem*, 2004. 279(26): 26975-26982.
45. H. Shogomori and T. Kobayashi, *Biochim Biophys Acta*, 2008. 1780(3): 612-618.
46. L. Garcia-Ortega, J. Alegre-Cebollada, S. Garcia-Linares, M. Bruix, A. Martinez-Del-Pozo and J.G. Gavilanes, *Biochim Biophys Acta*, 2011. 1808(9): 2275-2288.
47. T. Takara, T. Nakagawa, M. Isobe, N. Okino, S. Ichinose, A. Omori and M. Ito, *Biochim Biophys Acta*, 2011. 1811(5): 323-332.
48. L. Pedrera, M.L. Fanani, U. Ros, M.E. Lanio, B. Maggio and C. Alvarez, *Biochim Biophys Acta*, 2014. 1838(7): 1752-1759.