1	Title: Propagating ice front induces gas bursts and ultrasonic acoustic emissions from
2	freezing xylem
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25	coefficient of CO <sub>2</sub> ; respiration; winter embolism

#### 26 Abstract

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Ice formation and propagation in the xylem of plants is a complex process. During freezing of 28 xylem sap, gases dissolved in liquid sap are forced out of the ice lattice due to their low 29 solubility in ice, and supersaturation of xylem sap as well as low water potential ( $\Psi$ ) are induced 30 at the ice-liquid interface. Supersaturation of gases near the ice front may lead to bubble 31 32 formation and potentially to cavitation and/or to burst of gases driven out from the branch. In 33 this study, we investigated the origin and dynamics of freezing-related gas bursts and ultrasonic 34 acoustic emissions (AEs), which are suggested to indicate cavitation. Picea abies and Salix *caprea* branch segments were exposed to frost cycles in a temperature test chamber, and CO<sub>2</sub> 35 efflux (indicating gas bursts) and AEs were recorded. On freezing, two-thirds of the observed 36 gas bursts originated from the xylem and only one third from the bark. Simultaneously with gas 37 bursts, AEs were detected. Branch  $\Psi$  affected both gas bursts and AEs, with high gas burst in 38 saturated and dry samples but relevant AEs only in the latter. Repeated frost cycles led to 39 decreasing gas burst volumes and AE activity. Experiments revealed that the expanding ice 40 front in freezing xylem was responsible for observed gas bursts and AEs, and that branch  $\Psi$ 41 42 influenced both processes. Results also indicated that gas bursts and cavitation are independently induced by ice formation, though both may be relevant for bubble dynamics 43 during freezing. 44

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## 48 **1. Introduction**

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Ice formation in wood is a complex dynamic process. At subzero temperatures, xylem sap can 50 be in a doubly metastable state, when it is under tension and supercooled (Caupin et al. 2015). 51 While increasing tension may induce cavitation, decreasing temperatures will lead to ice 52 nucleation, after which ice will propagate longitudinally and radially within the xylem (Neuner 53 54 et al. 2010, Pramsohler et al. 2012, Charrier et al. 2017). Freezing can damage plants at the cellular level by plasmolysis or intracellular ice formation, which is lethal for cells (Ristic & 55 Ashworth 1993, Pearce 2001, Ruelland et al. 2009, Charrier et al. 2013). Furthermore, frost 56 57 cycles can cause embolism and, as a consequence, loss of xylem hydraulic conductivity (Tyree & Sperry 1989, Sperry & Sullivan 1992, Tyree et al. 1994, Mayr et al. 2007, Charrier et al. 58 2014). Following the classical theory of freeze-thaw-induced embolism, embolism occurs 59 60 because gases, dissolved in unfrozen xylem sap, are hardly soluble in ice. During freezing, gases segregate and form bubbles within the ice (Sucoff 1969, Sperry & Sullivan 1992, Mayr 61 & Sperry 2010, Sevanto et al. 2012, Charra-Vaskou et al. 2015). Upon thawing, these bubbles 62 collapse and gases dissolve back into the sap. However, in the case of large bubbles and/or 63 tension in the surrounding xylem sap, the bubbles can expand, fill the conduit, and lead to 64 65 embolism (Pittermann & Sperry 2006, Mayr & Améglio 2016).

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A visualization of bubble formation in freezing xylem sap has not yet been possible, but several studies reported ultrasonic acoustic emissions (AEs) to occur during ice formation (Raschi et al. 1989, Kikuta & Richter 2003, Mayr et al. 2007, Mayr & Zublasing 2010, Charrier et al. 2015). Under drought, AEs enable the detection of cavitation, i.e. the rapid tension release in the conduits when liquid water at low water potential ( $\Psi$ ) is replaced by water vapor near vacuum pressure (Tyree & Dixon 1983, Salleo & Lo Gullo 1986, Mayr & Rosner 2011, Charrier

et al. 2015). During freezing, the mechanism producing AEs is less clear, but previous studies 73 indicated that AEs during freezing are caused by cavitation in the xylem sap due to low  $\Psi$  at 74 75 the ice-liquid interface (Mayr et al. 2007, Mayr & Sperry 2010, Charrier et al. 2015, 2017). This suggests that there might be two embolism mechanisms involved in embolism formation 76 77 of tree stems during frost cycles: the freeze-thaw-induced embolism (according to classical 78 theory; Sucoff 1969, Sperry & Sullivan 1992, Mayr & Sperry 2010, Sevanto et al. 2012, Charra-79 Vaskou et al. 2015) and freeze-induced cavitation at the propagating ice front (producing AEs; Mayr et al. 2007, Mayr & Sperry 2010, Charrier et al. 2015, 2017). 80

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It was also shown that freezing can lead to emission of gases from xylem tissues—Lintunen et 82 al. (2014) demonstrated for the first time, that gases were pushed out from conifer stems during 83 ice propagation both in laboratory experiments and under field conditions. These results suggest 84 that a proportion of gases dissolved in the xylem sap can be emitted from the xylem during 85 86 freezing and is not trapped in ice. However, it is still unknown, from which tissue(s) the observed gas bursts originate and the underlying processes are not yet understood. It is likely 87 that gas concentrations in front of the propagating ice front increase (Sevanto et al. 2012) and 88 89 thus create a large concentration gradient between gases inside conduits and inter-conduit spaces (as well as ambient air). This gradient may accelerate the diffusion of gases out from the 90 stem until the entire sap is frozen. In addition to this diffusion process, gases may be driven out 91 from the vascular system by pressure-driven mass flow (Lintunen et al. 2014). The volume 92 increase upon water-ice transition creates positive pressure (Robson & Petty 1987), which will 93 94 force gases to escape from the conduits much faster than by diffusion only. Theoretical calculations of gas volumes (Lintunen et al. 2014) suggest that bursts might reduce the 95 likelihood of embolism formation during the freezing process by decreasing the sap gas content. 96

In this study, we monitored the spatial and dynamic patterns of gas bursts and AEs of branches 98 99 to analyze the effects of branch water status and the role of xylem and bark. Also, we studied potential links between gas bursts and AEs. Branch segments of Picea abies were exposed to 100 101 frost cycles in a temperature test chamber, and gas bursts and AEs of saturated and dehydrated stems and peeled xylem as well as detached bark samples were measured. Measurements were 102 also performed during repeated frost cycles, and additional freeze-thaw experiments on a 103 104 broad-leaved species (Salix caprea). We hypothesized that (i) gas bursts (measured as CO<sub>2</sub> efflux) originating from the xylem are larger than from the bark as proposed based on 105 theoretical calculations by Lintunen et al. (2014); (ii) gas bursts and AEs occur simultaneously; 106 and (iii) the volume of gas bursts and the number of AEs depend on branch  $\Psi$ , with the highest 107 gas bursts and AEs recorded in medium dry samples. In fact, we expect freezing-induced 108 cavitation to be absent in saturated samples, while in completely dehydrated samples, the 109 remaining volume of the xylem sap should be too small for relevant gas bursts or AEs. Finally, 110 111 (iv) repeated frost cycles were expected to cause a decrease in gas bursts and AEs as embolism formation would reduce the probability of cavitation and gas bursts would reduce sap gas 112 concentrations in consecutive frost cycles. Based on this first simultaneous analyses of gas 113 114 efflux (measured as CO<sub>2</sub> efflux) and AEs, the identification of stem tissues producing the gas burst and the analysis of the influence of water potentials, this study should enable new insights 115 into the complex process of freezing in plant xylem. 116

- 117
- 118 **2. Material and methods**

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<sup>120 &</sup>lt;u>2.1. Plant material</u>

Picea abies (L.) H. Karst. branches (1 to 1.5 m long) were collected from mature trees growing 122 123 at Natters (47°11'N, 11°12'E, 838 m a.s.l.) near the Institute of Botany in Innsbruck, Austria. Sixteen branches were collected during the first week of March 2018 and the first 2 weeks of 124 April 2018, before the start of the vegetation period. In addition, three branches of Salix caprea 125 L. were collected during the second week of April at the alpine timberline (Praxmar, Tyrolean 126 Central Alps, Austria; 47°09'N, 11°07'E; 1680 m a.s.l.). P. abies and S. caprea species were 127 128 selected as they are typical temperate/boreal species growing in extreme habitats, and thus are exposed to numerous frost cycles during winter. 129

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131 Immediately after collection, branches were covered with black plastic bags and saturated in a 132 bucket filled with tap water for at least 24 h. After saturation,  $\Psi$  was measured from 2–4 end twigs (ca 10 cm) per branch with a pressure chamber (PMS, model 1505D, Albany, NY, USA). 133 A 10 cm long segment was cut under water from the base of each branch, and the remaining 35 134 135 cm long branch segment, without side branches, was further used for experiments (later referred to as sample). After cutting, the ends of the sample were immediately sealed with instant 136 adhesive (Loctite 382 Instant Adhesive, Henkel, Düsseldorf, Germany) to minimize drying of 137 138 the sample during the experiment.

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## 140 <u>2.2. Ultrasonic acoustic emissions</u>

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Ultrasonic AE measurements were performed with a PCI-8-based system (PAC Micro-II Express Digital AE System) and 150 kHz resonance sensors (R15) connected to a preamplifier set to 40 dB (Physical Acoustics, Wolfegg, Germany). The threshold was set to 45 dB and the gain to 40 dB (Mayr et al. 2007, Mayr & Rosner 2011, Charrier et al. 2015). With these settings, AEs can be detected over several centimeters in conifers. Registration and analysis of AEs were

performed with AEwin software (Mistras Holdings Corp., Princeton, NJ, USA). Ultrasonic sensors were attached to both ends of the samples (Fig. 1). On stem samples, a small square (1  $\times$  1 cm) of the bark was removed to attach the sensors, and silicone grease was used to allow good contact between the sensors and the xylem surface.

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## 152 <u>2.3. Gas exchange measurements</u>

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A cuvette (outside length 10 cm; inside length 5.5 cm; diameter varied with branch size and mounting) made of transparent polyethylene plastic was wrapped around the central part of the sample, sealed at both ends with foam, and tightened with cable ties (Fig. 1). The joint of the cuvette was sealed with transparent tape. To allow constant air circulation, gas inlet and outlet tubes were connected to the cuvette and a battery-powered fan (1 lpm) was inserted into the cuvette (Fig. 1).

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The gas exchange of the sample was measured with a mobile CO<sub>2</sub> and H<sub>2</sub>O open-flow cuvette 161 system that includes a high-resolution LI-840A gas analyzer (LI-COR, Lincoln, NE, USA). A 162 163 Raspberry Pi computer was used to communicate with the gas analyzer, via an 8-channel 164 transmitter (Nokeval, Nokia, Finland) and user-defined measuring protocols. The system was powered by a 48 Ah lead acid battery (Optima Batteries, WI, USA) and the measurement 165 system was dynamic, such that it allowed supply air from a single source (ambient or cylinder 166 167 air) to enter two cuvettes and a reference line (Fig. 2). Supply air was continuously passed through the cuvettes at a constant rate, and measurements were based on the concentration of 168 the air stream flowing out of the cuvette and reference airline. The system was designed to 169 measure three air samples in the following alternating order: air sample from the first cuvette, 170 air sample from the reference line (that fed substitute air for compensating the flush and sample 171

172 flows from the cuvettes), and air sample from the second cuvette. The flows through the 173 cuvettes and reference lines were controlled via valves, which were turned on and off following 174 a measurement cycle. The flow rates for measurements and flushing during the supply air 175 measurement were set to 0.56 lpm.

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## 177 <u>2.4. Freezing experiments</u>

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To study CO<sub>2</sub> bursts and AEs during freezing, we conducted freezing experiments in a 179 temperature test chamber (Binder model MK 53, Tuttlingen, Germany). Gas exchange cuvettes 180 181 and AE sensors were first installed onto samples. Further, two T-type thermocouples were positioned in the xylem located inside the cuvette with 2 cm distance from each other, and two 182 thermocouples were positioned in the xylem located outside the cuvette (on both ends of the 183 184 branch segment; see Fig. 1). Thermocouples allowed constant detection of xylem temperature and thus enabled to record when the xylem sap froze. Freezing releases thermal energy due to 185 the phase change from water into ice (Burke et al. 1976) and, therefore, it is possible to detect 186 sudden increases in local xylem temperature (i.e. freezing exotherm), which indicate freezing 187 xylem sap. Xylem temperatures were measured at 10 s intervals and logged with a data logger 188 189 (Campbell Scientific, Logan, UT, USA).

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Samples were exposed to a temperature cycle inside the test chamber. The temperature was first set to  $+2^{\circ}$ C for 45 min before cooling for a period of 1 h to reach a temperature of  $-8^{\circ}$ C. Then, the temperature was kept at  $-8^{\circ}$ C for 1 h before warming for 1 h to reach back  $+2^{\circ}$ C. As last step, the temperature was kept constant at  $+2^{\circ}$ C for a minimum of 30 min (see Supporting Information). As with previous freezing studies, similar freezing/thawing rates were used to reduce the time needed for each experiment (Mayr et al. 2007, Mayr & Sperry 2010, Lintunen et al. 2014). The temperature inside the gas exchange cuvette was always higher than in the surrounding test chamber but the temperature difference never exceeded +3.5 °C. This was enabled by positioning the supply air tubing on the test chamber floor (loops with a total length of ca 2m) to allow heat transfer between the metallic test chamber and the tubing before it reached the cuvette. Drying of the samples during the experiments was expected to be small, because the cut ends of the branches were sealed and temperature of the test chamber decreased to  $+2^{\circ}$ C fast in the beginning of the experiment.

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Based on these settings, several freezing experiments were conducted. In experiment 1, we 205 206 tested for the origins of freezing-related CO<sub>2</sub> bursts (xylem, bark, or both). The bark was carefully detached from three saturated P. abies samples as a single piece (i.e., bark samples 207 included the phloem and outer bark). The detached bark was then positioned around a plastic 208 209 tube and stabilized with small cable ties to prevent excess drying from the inner surface of the 210 bark. Parallel to the detached bark sample, the peeled xylem sample was measured in the second cuvette. In experiment 2, we analyzed the influence of  $\Psi$  on the CO<sub>2</sub> bursts and AEs. After 211 saturation, prior to exposure to the temperature cycles, P. abies samples were dehydrated to a 212  $\Psi$  between -3.2 MPa and -3.4 MPa (hereafter called "dry"; 3 samples) and to a  $\Psi$  below -4 213 214 MPa (hereafter called "very dry"; 3 samples). Four branches were kept saturated ( $\Psi$  between -0.3 MPa and -0.5 MPa). In experiment 3, we studied the dynamics of CO<sub>2</sub> bursts and AEs 215 216 during three repeated frost cycles with four saturated P. abies samples. In other words, the same cycle (from  $+2^{\circ}$ C to  $-8^{\circ}$ C) was repeated three times, and between cycles the temperature was 217 kept for 1 h at  $+2^{\circ}$ C. In experiment 4, we studied CO<sub>2</sub> bursts in three saturated samples of S. 218 *caprea* L. 219

Freezing was artificially induced in all samples to decrease the variability of the freezing 221 222 temperature and to control the point of ice nucleation. The tip of a copper nail, which was previously cooled via liquid nitrogen, was used to touch a  $1 \times 2$  cm debarked section of the 223 branch (see also Fig. 1) at the sample base. Freezing was induced when the xylem temperature 224 at the stem base (outside the cuvette) was around -1.5°C in the saturated samples and around -225  $3.0^{\circ}$ C in the dehydrated samples to mimic their natural freezing temperatures (Lintunen et al. 226 227 2018). In experiment 3, artificial freezing was induced at the stem base at -1.5°C in the first two cycles, while freezing occurred spontaneously in the third cycle (due to the dehydrated 228 surface of the debarked section, induction of freezing was not possible). 229

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# 231 2.5. Calculation of the freezing-related $CO_2$ burst and effective diffusion coefficient for $CO_2$

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233 The exchange of  $CO_2$  between the stem and its environment can be directly measured, while it is necessary to exclude respiration effects from the measured CO<sub>2</sub> efflux to calculate the 234 magnitude of the freezing-related CO<sub>2</sub> burst. At above freezing temperatures, stem CO<sub>2</sub> 235 exchange and respiration are tightly coupled to each other, (albeit with a time lag; see Teskey 236 & McGuire 2007, Bloemen et al. 2013), while they are clearly decoupled during and after 237 238 freezing. Very little information exists on the temperature dependence of the respiration rate in a frozen stems, therefore, we assumed similar temperature dependence before and after freezing 239 (i.e., the temperature dependency of respiration was extrapolated to below freezing 240 temperatures). The magnitude of the freezing-related  $CO_2$  burst is thus affected by  $CO_2$ 241 produced by respiration (i.e., the concentration gradient between the inside and outside of the 242 stem), the effective radial diffusion coefficient for CO<sub>2</sub>, and the partial pressure difference 243 between the inside and outside of the stem. We used a previously published dynamic model of 244 CO<sub>2</sub> mass balance and transport within the stem (Hölttä & Kolari 2009, Lintunen et al. 2014) 245

to separate the freezing-related CO<sub>2</sub> burst from the total stem CO<sub>2</sub> efflux and to estimate the effective radial CO<sub>2</sub> diffusion coefficient for the samples. The model (Hölttä & Kolari 2009) solves the CO<sub>2</sub> concentration profile within the stem by taking into account the CO<sub>2</sub> produced in respiration, its partitioning between the liquid and gaseous phase, and its radial diffusion according to the concentration gradient within the stem. The change in the amount of CO<sub>2</sub>  $(\Delta N_{CO2}, \mu mol)$  inside the stem during a time interval  $\Delta t$  (s) can be expressed as:

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$$\Delta N_{\rm CO2} / \Delta t = R(T_{\rm xylem}) - (C_{\rm CO2} - C_{\rm CO2, amb}) d_{\rm eff}$$
(1)

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where  $d_{\rm eff}$  (m<sup>3</sup> s<sup>-1</sup>) is an effective diffusion coefficient (Heitjans & Kärger 2005, including 255 256 information on size, geometry, and water content of the object, i.e., it is the constant of proportionality between the flux of CO<sub>2</sub> due to diffusion and the difference in the concentration 257 between the inside of the stem and the ambient air) and R respiration rate ( $\mu$  mol s<sup>-1</sup>), C<sub>CO2</sub> is the 258  $CO_2$  concentration in the air phase inside the stem, and  $C_{CO2,amb}$  is the ambient  $CO_2$ 259 concentration inside the climate chamber. We modelled R to be exponentially dependent on 260 temperature inside the unfrozen stem ( $T_{xylem}$ , °C): A Q10 value of 2.5 was used for the 261 temperature dependency of respiration, similarly to the original parameterization.  $R(T_{xylem})$  was 262 263 fitted separately to each experiment based on the measurements before freezing. We assumed both gaseous (see Gartner et al. 2004) and liquid phases to account for 25% of stem volume. 264 The relationship between CO<sub>2</sub> concentration in gaseous and liquid phase was calculated by the 265 Henry's law (Seinfeld & Pandis 1998), which strongly depends on temperature. Model 266 parameterization was kept as described in Lintunen et al. (2014). 267

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The difference between the total measured  $CO_2$  efflux (which includes respired  $CO_2$  and freezing-related  $CO_2$  burst) and the modeled  $CO_2$  efflux (includes only respired  $CO_2$ ) represents the burst of  $CO_2$  released from the stem due to the freezing process. The volume of the freezingrelated  $CO_2$  burst was calculated as an integral of this difference from the moment of ice nucleation until the difference went to zero or the apoplastic water thawed at about 0°C (in some cases the burst was not over when the stem thawed). Values of the radial diffusion coefficient and absolute respiration were both fitted so that the dynamics and absolute values of the modeled  $CO_2$  efflux rate matched the measured  $CO_2$  efflux rate until the stem was frozen.

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We also calculated the volume of freezing-related  $CO_2$  bursts normalized to the respiration rate of the sample at +5°C, (i.e., the volume of the burst was divided by the respiration efflux of the sample at +5°C; hereafter called "normalized  $CO_2$  burst"). In this way, samples with different respiration levels, and thus a different amount of stored  $CO_2$ , could be compared.

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### 283 <u>2.6. Statistical analysis</u>

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Statistical analysis was performed with the GLM procedure in SAS ver. 9.4 (SAS Institute, 285 Cary, NC, USA), which uses the method of least squares to fit general linear models. Analysis 286 of variance was used to compare the volume of the freezing-related CO<sub>2</sub> bursts, number of AEs, 287 respiration at +5°C, and effective diffusion coefficient of CO<sub>2</sub> between samples of different  $\Psi$ , 288 samples with peeled xylem, detached bark or both (i.e. intact branch), and samples of different 289 species. In the analysis comparing samples with different  $\Psi$ , data were unbalanced due to 290 different sample sizes between treatments. Therefore, a Type III test was used in GLM, because 291 292 this test is independent of the number of observations per treatment combination. All tests were performed at a probability level of 5%. Values are given as mean  $\pm$  standard error. 293

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#### 295 **3. Results**

## 297 <u>3.1. Freezing-related CO<sub>2</sub> bursts from peeled xylem and detached bark</u>

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In experiment 1, freezing-related CO<sub>2</sub> bursts were detected in both peeled xylem samples and 299 detached bark samples of P. abies (Fig. 3A, B). In detached bark, the volume of absolute CO<sub>2</sub> 300 bursts was about 3.4-fold lower (P = 0.043) and the volume of normalized CO<sub>2</sub> bursts about 7-301 302 fold (P = 0.010) lower than that of intact samples (Fig. 4A). Neither the volume of absolute nor normalized CO<sub>2</sub> bursts differed between peeled xylem and intact samples (Fig. 4A). CO<sub>2</sub> efflux 303 per surface area at +5°C did not differ between detached bark, peeled xylem or intact samples 304 305 (not shown). The effective CO<sub>2</sub> diffusion coefficient was higher in detached bark than in peeled 306 xylem or intact samples (P < 0.001), whereas there was no significant difference between peeled xylem and intact samples (Fig. 4A). 307

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# 309 <u>3.2. CO<sub>2</sub> efflux and acoustic emission dynamics at different water potential</u>

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In experiment 2, samples showed decreasing CO<sub>2</sub> efflux with decreasing xylem temperature 311 until freezing occurred (Fig. 5A). The average level of CO<sub>2</sub> efflux at  $+5^{\circ}$ C was  $0.52 \pm 0.06$ 312  $\mu$  mol m<sup>-2</sup> s<sup>-1</sup> and it decreased down to 0.30  $\pm$  0.03  $\mu$  mol m<sup>-2</sup> s<sup>-1</sup> prior to freezing. CO<sub>2</sub> efflux 313 increased in all samples simultaneously with freezing, which was indicated by a freezing 314 exotherm (Fig. 5A). The average maximum CO<sub>2</sub> efflux during the freezing exotherm was 0.42 315  $\pm$  0.03 µmol m<sup>-2</sup> s<sup>-1</sup>. There was no statistical difference between saturated and dehydrated 316 samples in the CO<sub>2</sub> efflux at +5°C or in the freezing-related CO<sub>2</sub> bursts. However, the volume 317 of the freezing-related normalized CO<sub>2</sub> bursts in saturated and dry samples was 3.4-fold and 318 3.1-fold higher than in very dry samples, respectively (Fig. 6A). In contrast, the effective 319 diffusion coefficient of CO<sub>2</sub> was 6.9-fold and 2.9-fold higher in very dry samples than in 320

saturated and dry samples, respectively (Fig. 6B, the sample with  $\Psi$  of -6.7 MPa was dropped from the analysis as an outlier).

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Freezing-related CO<sub>2</sub> bursts were not always completed before thawing. This was particularly obvious in saturated stems, which also showed long-lasting freezing exotherms. However, the CO<sub>2</sub> efflux was always highest soon after the beginning of the exotherm and decreased toward the onset of thawing. Thus, the potential bias in the calculated burst volume is expected to be small.

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Besides CO<sub>2</sub> efflux, we also observed AEs during freezing (Fig. 5B). At the stem base (where ice nucleation was induced), the onset of AEs was always registered parallel to the start of the exotherm. AEs ceased when the exotherm peaked, until the temperature decreased again and new AEs were observed. The cumulative number of AEs was highest in samples with a  $\Psi$ between -2.5 MPa and -3.5 MPa (Fig. 6C). At the stem apex (AE2 sensor in Fig. 1), the cumulative number of AEs was smaller than at the stem base and sometimes even missing. The onset of AEs was always later at the apex than at the stem base (Fig. 5B).

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### 338 <u>3.3 CO<sub>2</sub> efflux and acoustic emission dynamics in subsequent frost cycles</u>

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During three subsequent frost cycles (experiment 3),  $CO_2$  measurements indicated a pronounced freezing-related  $CO_2$  burst in the first cycle, a smaller one in the second, and a barely visible one in the third (Fig. 7). Unfortunately, the volume of the bursts could not be calculated due to difficulty in modeling respiration-induced  $CO_2$  efflux between the repeated frost cycles. Interestingly,  $CO_2$  efflux was also detected during thawing, whereby these thawing-induced bursts increased with consecutive temperature cycles. AEs, like in experiment 2, were only observed during freezing, whereby the number of cumulative AEs was highest inthe first cycle and decreased during consecutive cycles (Fig. 7).

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# 349 <u>3.4. Comparing CO<sub>2</sub> efflux and acoustic emission dynamics of Salix caprea</u>

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Freezing-related CO<sub>2</sub> bursts were also detected from saturated samples of S. caprea (Fig. 8A; 351 352 experiment 4). The absolute volume of the bursts was not significantly different between species, but the volume of the normalized  $CO_2$  bursts with respiration rate at  $+5^{\circ}C$  was lower 353 in S. caprea than in P. abies (P = 0.021; Fig. 4B). This can be partly explained by the lower 354 355 respiration rate of P. abies at  $+5^{\circ}$ C compared with that of S. caprea (P < 0.001) (Fig. 4B). Also, burst dynamics differed between species as the CO<sub>2</sub> bursts started earlier (already at the time 356 of ice nucleation outside the gas exchange cuvette) and were shorter in S. caprea than in P. 357 358 abies (Fig. 8). The effective CO<sub>2</sub> diffusion coefficient was clearly higher in S. caprea than in *P. abies* (P = 0.029; Fig. 4B). The dynamics and patterns of AEs during freezing were similar 359 to P. abies, with highest acoustic activity immediately after exotherm detection and a time lag 360 between the two sensors (Fig. 8). 361

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# 363 **4. Discussion**

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Performed temperature experiments on *P. abies* demonstrated a high complexity of the freezing process in xylem, with respective complex dynamics and patterns in gas bursts and AEs. Our hypotheses were partly confirmed: (i) the majority of the CO<sub>2</sub> efflux originated from the xylem; (ii) freezing-related gas bursts and AEs occurred simultaneously with freezing, indicating that the propagating ice front induced both; (iii) branch  $\Psi$  had a major influence on both gas bursts and AEs, though the gas burst was high in saturated and medium dry samples, while relevant AEs were only observed in medium dry samples; and (iv) repeated frost cycles led to decreasing gas burst volumes and AE activity. In addition, CO<sub>2</sub> bursts and AEs were demonstrated in *S. caprea*, which indicates that the observed processes are relevant in conifer as well as angiosperm species.

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Experiment 1 demonstrated that about two thirds of released CO<sub>2</sub> came from the xylem, while 376 377 only one third originated from the bark, despite the high proportion of living tissues and large intercellular spaces in the latter. In a previous study on *Pinus sylvestris* and *P. abies* (Lintunen 378 et al. 2014), a modeling approach suggested that the volume of freezing-related CO<sub>2</sub> bursts 379 380 corresponded to 70% of the total amount of dissolved CO<sub>2</sub> in the xylem. When we now consider 381 that one third of the gas burst originated from the bark, we can conclude that about 50% of the total CO<sub>2</sub> within the xylem is released via a gas burst on freezing. It is important to note that 382 xylem sap also contains other dissolved gases, with N<sub>2</sub> and O<sub>2</sub> representing the highest volumes. 383 Analyzed CO<sub>2</sub> burst dynamics, therefore, should be qualitatively equivalent to bursts of other 384 gases but differ quantitatively due to differences in concentration in xylem sap and the ambient 385 air, solubility and diffusion coefficients. The main source of CO<sub>2</sub> within the bark was most 386 likely within the intercellular spaces, which were filled with ice during freezing. This is because 387 388 living cells in hardened P. abies stems avoid lethal intracellular ice formation by extracellular freezing (Sakai & Okada 1971). The effective radial CO<sub>2</sub> diffusion coefficient at +5°C was 389 clearly higher in the detached bark than in peeled xylem and intact samples, most likely due to 390 the more porous bark tissue. The effective radial diffusion coefficient of  $CO_2$  has only rarely 391 been measured for stems and varied between  $10^{-11}$  m<sup>2</sup> s<sup>-1</sup> and  $10^{-7}$  m<sup>2</sup> s<sup>-1</sup>, depending on wood 392 structure and water status (Sorz & Hietz 2006, Spicer & Holbrook 2007, Hölttä & Kolari 2009). 393 Our calculation of  $10^{-9}$  m<sup>2</sup> s<sup>-1</sup> for *P. abies* stems (Fig. 4) falls well within this range. 394

 $CO_2$  bursts were strongly influenced by the water status of the samples (experiment 2). The 396 397 volume of normalized CO<sub>2</sub> bursts detected during freezing was lower in dehydrated samples (Fig. 6; very dry) probably due to the following reasons. First, the sap volume (because of 398 drought-induced embolism preceding frost treatment) and respective absolute volume of gases 399 400 dissolved in the sap was low, and the volumetric increase upon water-ice transition was small. Second, a lower  $\Psi$  causes lower freezing temperatures (Lintunen et al. 2018), which leads to 401 402 higher ice propagation velocities (Kitaura 1967, Langer et al. 1978, Hacker & Neuner 2007, Rauschenberger et al. 2013, Charrier et al. 2015). A low  $\Psi$  is thus associated with small CO<sub>2</sub> 403 bursts because high ice propagation velocities may not leave enough time for the gases to 404 diffuse out of the stem before being trapped in ice. Third, the effective radial diffusion 405 coefficient of CO<sub>2</sub> at +5°C was negatively related to stem  $\Psi$  (Fig. 6). This can be explained by 406 the higher portion of air in dry stems as the diffusion of gases is about 10 000 times faster in air 407 than in water (Nobel, 2005). Diffusion of CO<sub>2</sub> out of dry stems might lead to a low internal CO<sub>2</sub> 408 409 gas concentration prior to freezing and thus to small gas bursts on freezing. In very dry samples, the number of AEs was also small, probably because most conduits were already air-filled due 410 411 to drought-induced embolism. In contrast, samples between -3.2 MPa and -3.4 MPa (Fig. 6; 412 dry) showed a high number of AEs as well as pronounced gas bursts. Previous studies (Mayr et al. 2007, Mayr & Sperry 2010, Charrier et al. 2015, 2017), indicated that AEs are induced by 413 cavitation events near the moving ice front, causing a sudden release of tension. Mayr et al. 414 415 (2007) clearly demonstrated that AEs during freezing are related to xylem water under tension as samples at critical water potential showed highest acoustic activity while saturated samples 416 (or water in a straw) did hardly emit AEs. Charrier et al. (2015) found that AEs in freezing 417 samples of several tree species showed two distinct phases: the first phase during ice nucleation 418 and propagation, and the second after dissipation of the exothermal heat. They explained the 419 first phase of AEs by low water potential of ice at the ice-liquid interface, which induced 420

421 numerous and strong signals. This in accordance with our experiments, where the sensor near 422 the point of artificial ice nucleation always recorded AEs earlier than the distal sensor (Fig. 3). 423 Remarkably, the comparison of saturated and dry samples demonstrated that cavitation 424 (indicated by AEs) and gas bursts, although both induced by ice formation, are not directly 425 linked: gas bursts from dry samples were likely not solely related to cavitation events as 426 saturated samples (Fig. 6; saturated) showed nearly identical, high CO<sub>2</sub> effluxes.

427

Both freezing-related CO<sub>2</sub> bursts and AEs were detected during each of the three consecutive 428 frost cycles in experiment 3 (Fig. 7). However, the volume of the bursts and the number of AEs 429 430 strongly decreased from the first to the third cycle. The number of AEs per freezing event has already been shown to decrease in a study by Mayr et al. (2007). The authors suggested that 431 tracheids, which embolized during the first frost cycle, cannot cavitate a second time and thus 432 433 the number of cavitating conduits (and respective AEs) decreases during subsequent frost cycles. CO<sub>2</sub> bursts decreased during consecutive frost cycles because there was not probably 434 enough time for the accumulation of new CO<sub>2</sub> from tissue respiration in the xylem between the 435 frost cycles to replace CO<sub>2</sub> lost in the bursts. Also, embolized xylem portions may provide 436 437 additional diffusion pathways out of the stem, which enable gas effluxes during thawing. 438 Accordingly, CO<sub>2</sub> bursts during thawing (also see Fig. 3) increased on each subsequent frost cycle. Further studies are required to disentangle the complex interrelations of processes during 439 freezing and thawing,  $\Psi$ , and repeated frost cycles. 440

441

Freezing-induced gas bursts and AEs were also detected in *S. caprea* (Fig. 8), suggesting that simultaneous CO<sub>2</sub> bursts and AEs are a phenomenon which can be found in many species. Interestingly, neither of our study species exhibited an increase in CO<sub>2</sub> efflux before the freezing exotherm was observed. In contrast, Sperling et al. (2015) reported significant

increases in stem respiration upon response to near-freezing temperatures in several woody 446 447 species growing in temperate environments. The authors found that the increase in  $CO_2$  efflux before freezing was related to an acceleration of stem non-structural carbohydrate consumption. 448 We did not observe an effect in  $CO_2$  efflux even when *P. abies* samples were kept near (but 449 above) the freezing point for a longer time (data not shown) and thus it is unlikely that metabolic 450 activity played a role in observed gas bursts. Our experiments also demonstrated that the 451 452 majority of the CO<sub>2</sub> efflux originated from xylem tissue (which mainly consists of dead conduit cells), and that the bursts occurred simultaneously with and not before ice nucleation. However, 453 freezing-tolerant species growing at high elevation and/or latitudes differ in their temperature-454 455 dependent respiration patterns from temperate species, which might lead to interspecific 456 differences in freezing-related CO<sub>2</sub> burst dynamics and patterns.

457

### 458 **5.** Conclusion

459

Potential mechanisms underlying freezing-related patterns observed in the present study are 460 summarized in Fig. 9: ice propagation forces gases out of the freezing sap, so they either form 461 462 bubbles (which are then trapped in ice) or move away from the ice front and remain dissolved 463 in the sap. In the latter case, the gas concentration in the remaining sap increases (Sevanto et al. 2012), thus increasing the driving force for gases to escape from the conduit via diffusion 464 (leading to freezing-related gas burst). The detection of AEs during freezing indicates cavitation 465 466 events (i.e., bubble formation) near the ice-liquid interface due to high tensions induced by ice (Mayr et al. 2007, Charrier et al. 2015). In this case, cavitation would result in a sudden release 467 of tension inside the conduits and, consequently, in AEs. Accordingly, saturated samples show 468 larger gas bursts as they contain larger gas volumes dissolved in the sap but emit only a few (or 469 no) AEs as the tension in the sap is low (Fig. 9). In dry samples, higher tensions in the xylem 470

471 lead to more AEs. Very dry samples contain a higher number of embolized conduits, which472 contribute neither to gas bursts nor AEs.

474	Our study clearly indicates that processes occurring near the expanding ice front in freezing
475	xylem are responsible for observed gas burst and AE patterns. However, many aspects of the
476	freezing process in plant xylem (e.g., the small-scale spatial pattern of ice propagation, ice
477	formation through the pits, bubble formation) as well as the influence of environmental factors
478	(e.g., temperature gradients within xylem, growth conditions) and the relevance for plant life
479	exposed to freezing require further study.
480	
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482	
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- 496 A. Lintunen, S. Mayr, and T. Hölttä planned, and A. Lintunen, A. Losso and S. Mayr conducted,
- 497 the experiments. J. Aalto and T. Chan designed and built the gas measurement system together
- 498 with technician H. Laakso. T. Hölttä was responsible for the calculation of the freezing-related
- 499 burst and A. Lintunen for the statistical analysis. A. Lintunen had the main responsibility for
- 500 writing the paper, but all authors contributed to the writing process.

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### 654 Figure legends

655

Figure 1. Cuvette and measurement setup. Both ends of the branch segment (brown) were sealed 656 657 with glue (light blue) and a transparent cuvette was wrapped around the central part of the sample. The cuvette was sealed at both ends with foam (black), and gas inlet and outlet tubes 658 659 (dark blue) were connected to the cuvette. To allow constant air circulation, a battery-powered 660 mini fan (1 lpm) was inserted. The system for measuring gas exchange is presented in Fig. 2. Two acoustic emission (AE) sensors and two thermocouples (T) were attached to both sides of 661 the cuvette—AE1 and T1 at the base and AE2 and T4 at the apex. Two thermocouples were 662 663 also inserted inside the cuvette—T2 at the base and T3 at the apex. A small square  $(1 \times 2 \text{ cm})$ 664 of the bark was removed at the stem base to enable artificial freezing.

665

666 Figure 2. Schematic diagram of the gas exchange measurement system for simultaneous measurements of two samples. Circles with an internal triangle represent the pumps, and SO, 667 S1, S2, F1, and F2 are the valves. MFM refers to Mass Flow Meter. Because of using 668 pressurized air from a cylinder, the pump and buffer for the supply air were bypassed (shown 669 670 with a dashed line). The measurement consists of a stepwise program that is repeated: i) both 671 cuvettes are first flushed with supply air for 30 s and the supply air concentration is measured during the flushing. ii) The supply air line for cuvette 1 is closed and the concentration in the 672 cuvette is measured for 60s; meanwhile, cuvette 2 continues to be flushed with supply air. iii) 673 674 Both cuvettes are again flushed with supply air for 30s, and the supply air concentration is measured during the flushing. iv) The supply air line for cuvette 2 is closed and the 675 concentration in the cuvette is measured for 60s; meanwhile, cuvette 1 continues to be flushed 676 with supply air. Figure 1 shows how the cuvette was mounted on the sample. 677

Figure 3. Time series of a measured  $CO_2$  efflux and a modeled respiratory  $CO_2$  release from a peeled *Picea abies* sample (A) and detached bark (B). At the beginning of the experiment, the sample water potential was -1.5 MPa. Xylem temperatures inside the cuvette are shown (T2 and T3). The background color represents the frozen (white) and thawed (light gray) periods of the xylem sap.

684

Figure 4. A) Comparison of the average volume of freezing-related CO<sub>2</sub> bursts, freezing-related 685  $CO_2$  bursts normalized with respiration efflux at +5°C, and the effective  $CO_2$  diffusion 686 687 coefficient for peeled xylem, detached bark, and intact samples (i.e. samples including both xylem and phloem) of saturated *Picea abies*. There was no difference between peeled xylem, 688 detached bark, and intact samples in the respiration efflux at +5°C. B) Comparison of the 689 average volume of freezing-related CO<sub>2</sub> bursts, normalized CO<sub>2</sub> bursts, and respiration efflux 690 at +5°C between saturated samples of P. abies and Salix caprea. There was no difference 691 692 between species in the effective  $CO_2$  diffusion coefficient. Significant differences (P < 0.05) between groups are indicated by different letters and roman numbers. Mean  $\pm$  standard error. 693

694

Figure 5. Time series of a measured  $CO_2$  efflux and a modeled respiratory  $CO_2$  release (A) and acoustic emissions (AEs) (B) measured on a *P. abies* sample dehydrated to -3.4 MPa. Xylem temperatures inside the cuvette (T2 and T3) are shown in A, whereas xylem temperatures outside the cuvette (T1 and T4) are shown in B. The background color represents the frozen (white) and thawed (light gray) periods of the xylem sap.

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Figure 6. Comparison of the volume of freezing-related  $CO_2$  bursts normalized with respiration at +5°C (A), effective diffusion coefficient of  $CO_2$  (B), and total number of acoustic emissions (AEs) during the freezing exotherm measured at the stem base (C) for four saturated samples

704	(water potential, $\Psi$ , between -0.3 MPa and -0.5 MPa), three dry samples ( $\Psi$ between -3.2 MPa
705	and $-3.4$ MPa) and three very dry samples ( $\Psi$ lower than $-4$ MPa). Significant differences (P
706	$< 0.05$ ) between groups are indicated by different letters. Mean $\pm$ standard error.

Figure 7. Time series of  $CO_2$  efflux and acoustic emissions (AEs; see also Fig. 1) from a *Picea abies* sample during three subsequent frost cycles. At the beginning of the experiment, sample water potential was -0.8 MPa. The background color represents the frozen (white) and thawed (light gray) periods of the xylem sap, white arrows show freezing-related  $CO_2$  bursts, and gray arrows show thawing-induced  $CO_2$  bursts.

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Figure 8. Time series of CO<sub>2</sub> efflux and modeled respiratory CO<sub>2</sub> release (A) and cumulative number of acoustic emissions (AEs) (B) from a *Salix caprea* sample. At the beginning of the experiment, sample water potential was -0.3 MPa. Xylem temperatures inside the cuvette is shown in A (T2 and T3), whereas xylem temperature outside the cuvette in sample base (T1) and top (T4) is shown in B. The background color represents the frozen (white) and thawed (light gray) periods of the xylem sap.

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Figure 9. Schematic of freezing in xylem and underlying processes of gas bursts and acoustic
emissions (AEs) during ice propagation. The schematic is presented separately for a saturated
(A), dry (B) and a very dry branch (C).

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731 Fig. 1



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733 Fig. 2











737 Fig. 4



739 Fig. 5





741 Fig. 6



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743 Fig. 7







